

Two-Dimensional Stable and Ultrathin Cluster-based Metal-Organic Layers for Efficient Electrocatalytic Water Oxidation†

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Table S1. Crystallographic data and structure refinement details for $\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs}$ and Co-

CMOLs¹.

compound	Co _{0.6} Ni _{0.4} -CMOLs ^{this work}	Co-CMOLs ¹
formula	C ₁₆ H ₈ Co _{1.2} Ni _{1.8} O ₁₀	C ₁₆ H ₈ Co ₃ O ₁₀
formula weight	536.62	539.03
temperature (K)	293	293
crystal system	Monoclinic	Monoclinic
space group	<i>P</i> 2 ₁ / <i>c</i>	<i>P</i> 2 ₁ / <i>c</i>
<i>a</i> (Å)	11.7117 (14)	11.648 (2)
<i>b</i> (Å)	12.0990 (13)	12.091 (3)
<i>c</i> (Å)	11.7979 (13)	11.846 (3)
α (°)	90	90
β (°)	103.170 (12)	103.245 (4)
γ (°)	90	90
<i>V</i> (Å ³)	1627.8 (3)	1624.0 (6)
<i>Z</i>	4	4
<i>D_x</i> , g cm ⁻³	2.190	2.205
<i>Mu</i> , mm ⁻¹	12.466	3.090
F(000)	1067.2	1068
GOF on <i>F</i> ²	1.001	0.9660
<i>R</i> _{int}	0.0734	0.0425
<i>R</i> ₁ , <i>wR</i> ₂ [<i>I</i> > 2 σ (<i>I</i>)]	0.0866, 0.2141	0.0409, 0.1045
<i>R</i> ₁ , <i>wR</i> ₂ [all data]	0.1264, 0.2550	0.0601, 0.1136
CCDC no.	2068005	913356

$$R_1 = \Sigma ||F_o| - |F_c|| / \Sigma |F_o|. \quad wR_2 = [\Sigma w(F_o^2 - F_c^2)^2 / \Sigma w(F_o^2)^2]^{1/2}.$$

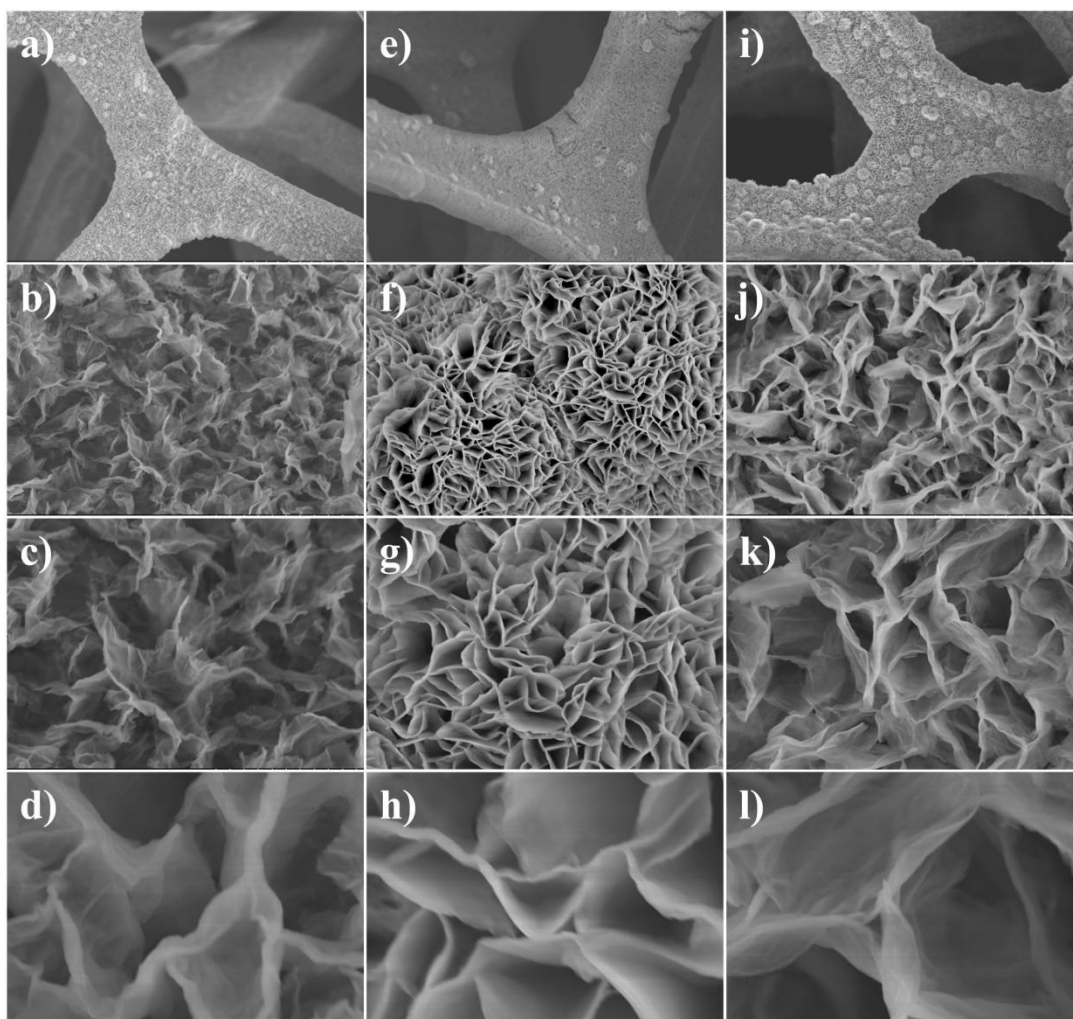


Fig. S1. SEM images of a) b), c), d) Co-CMOLs/NF-KOH/IM; e), f), g), h) Ni-CMOLs/NF-KOH/IM and i), j), k), l) $\text{Co}_{0.6}\text{Ni}_{0.4}$ -CMOLs/NF-KOH/IM.

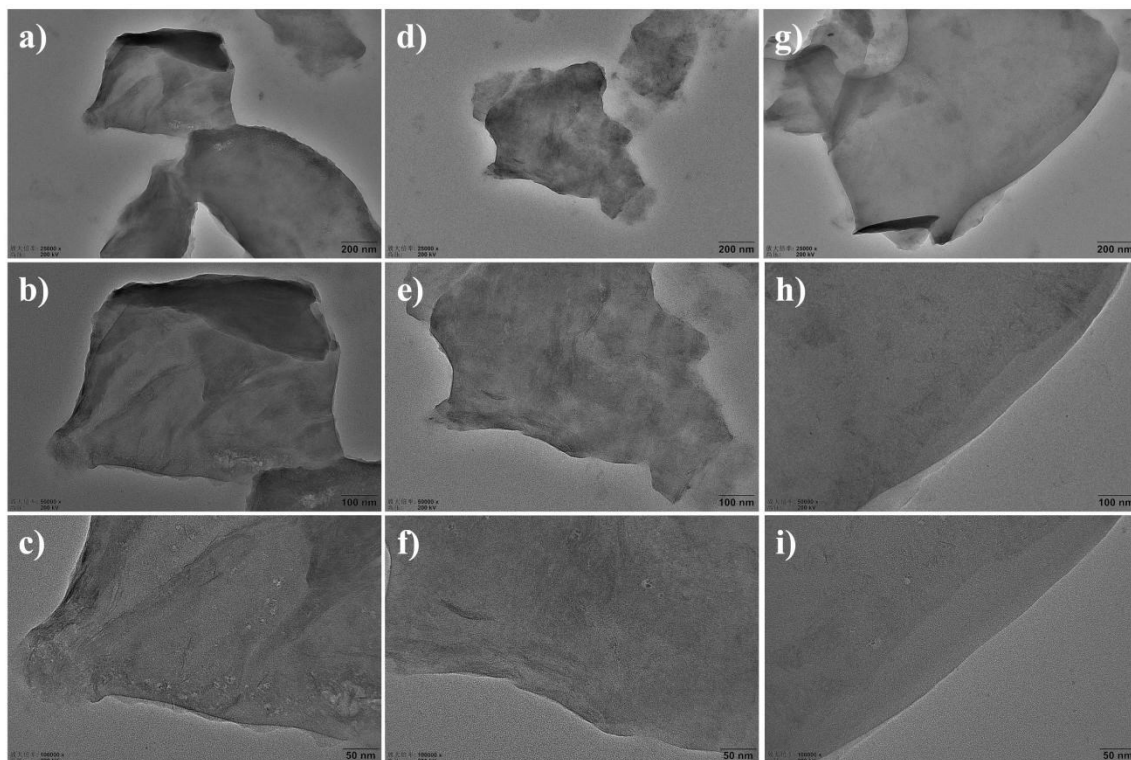


Fig. S2. TEM images of $\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs}$ (The $\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs}$ was removed from $\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs/NF}$ by ultrasonication)

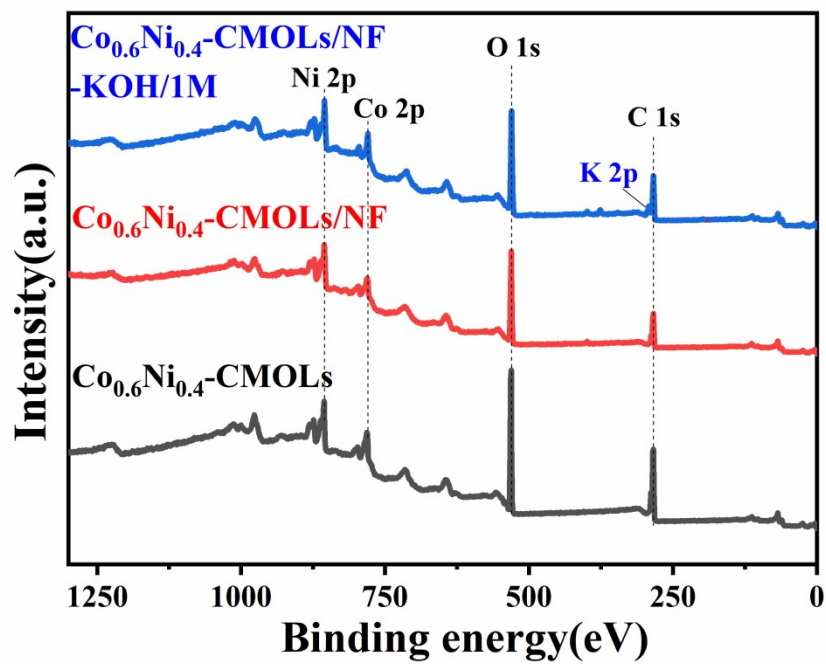


Fig. S3. XPS survey spectra for $\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs/NF-KOH/1M}$, $\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs/NF}$ and $\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs}$.

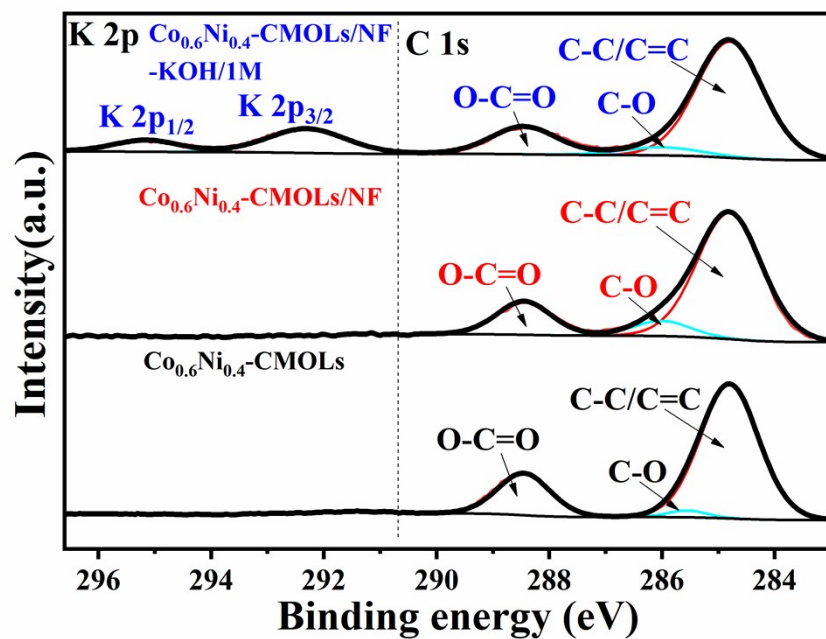


Fig. S4. The C 1s spectrum for Co_{0.6}Ni_{0.4}-CMOLs/NF-KOH/1M, Co_{0.6}Ni_{0.4}-CMOLs/NF and Co_{0.6}Ni_{0.4}-CMOLs (The K 2p shown in Co_{0.6}Ni_{0.4}-CMOLs/NF-KOH/1M is due to residual K⁺).

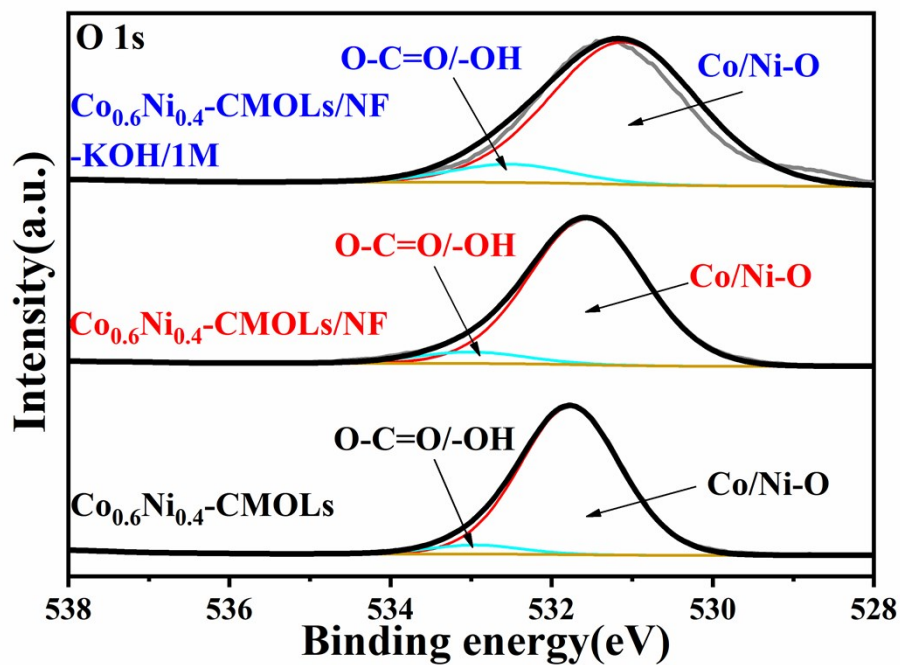


Fig. S5. The O 1s spectrum for Co_{0.6}Ni_{0.4}-CMOLs/NF-KOH/1M, Co_{0.6}Ni_{0.4}-CMOLs/NF and Co_{0.6}Ni_{0.4}-CMOLs.

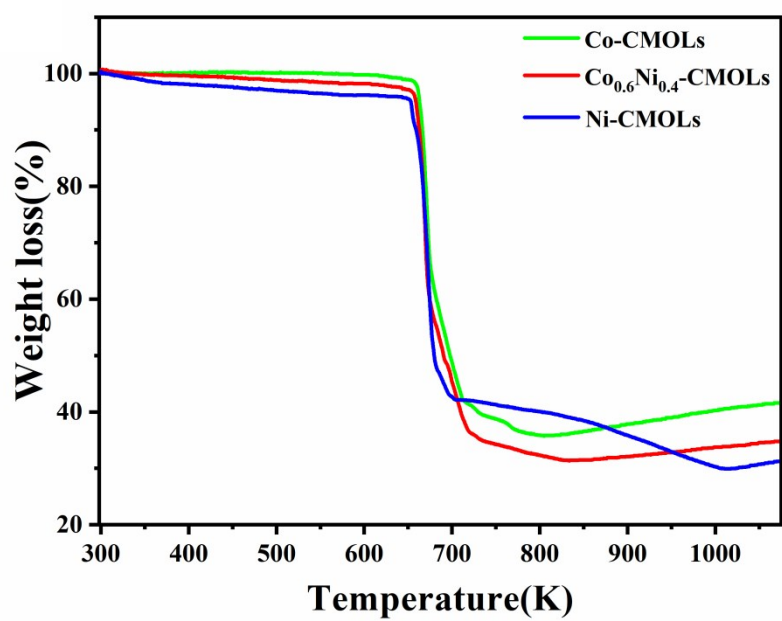


Fig.S6. TGA of Co-CMOLs, $\text{Co}_{0.6}\text{Ni}_{0.4}$ -CMOLs and Ni-CMOLs.

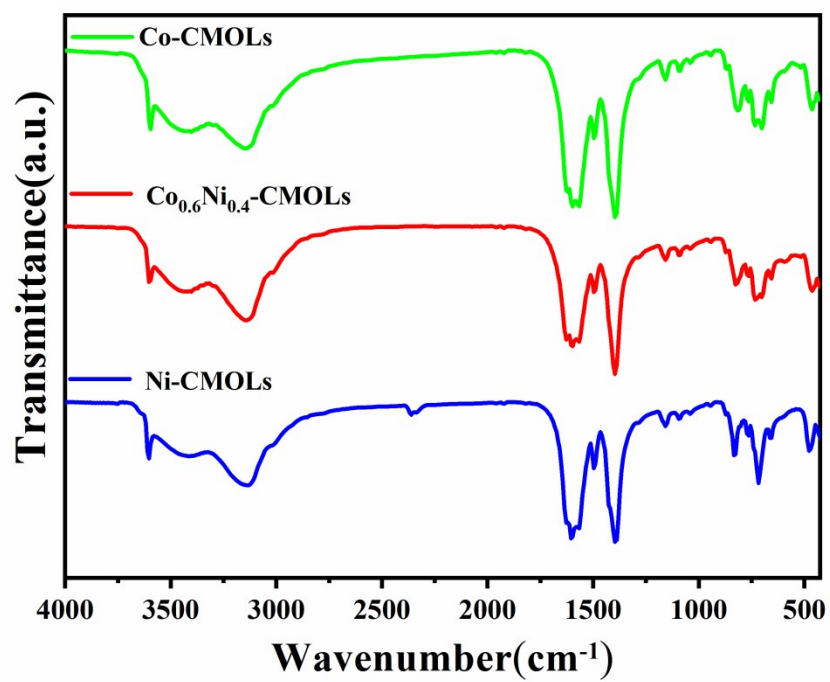


Fig. S7. IR of Co-CMOLs, $\text{Co}_{0.6}\text{Ni}_{0.4}$ -CMOLs and Ni-CMOLs.

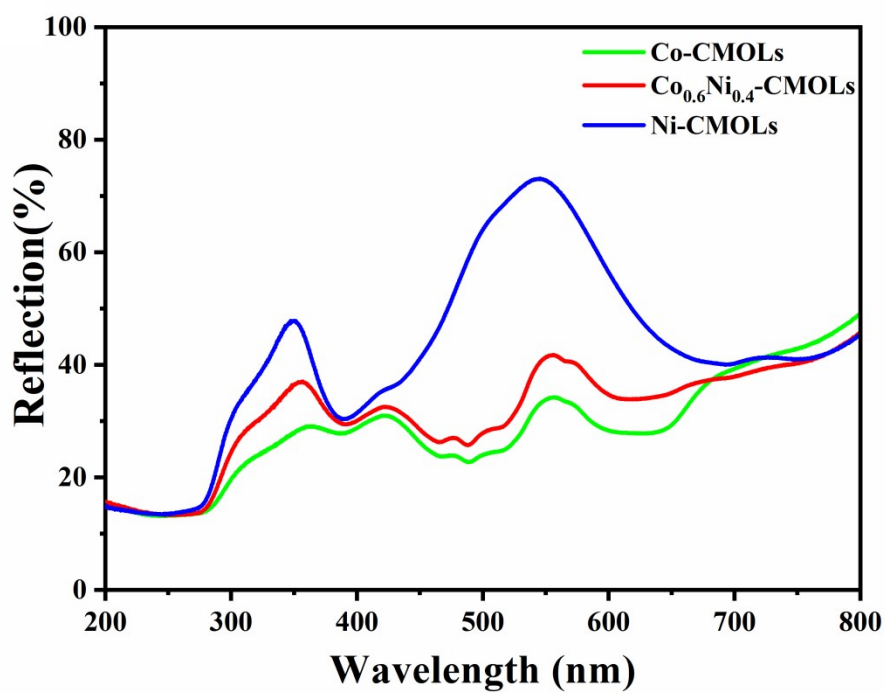


Fig. S8. UV-visible diffuse reflectance spectra of Co-CMOLs, Co_{0.6}Ni_{0.4}-CMOLs and Ni-CMOLs.

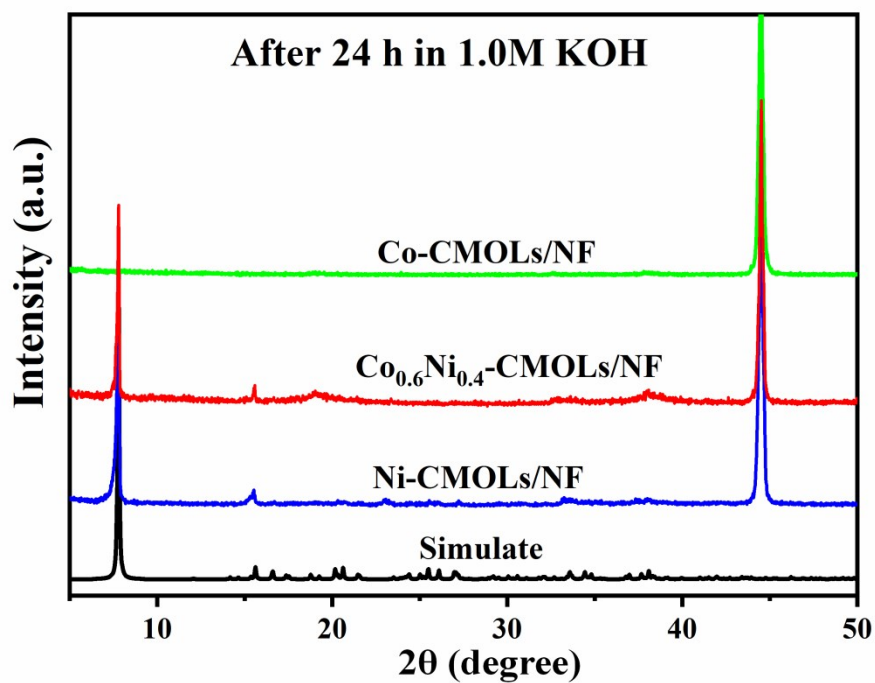


Fig. S9. PXRD patterns of three CMOLs/NF immersed in 1.0 M KOH for 24 h.

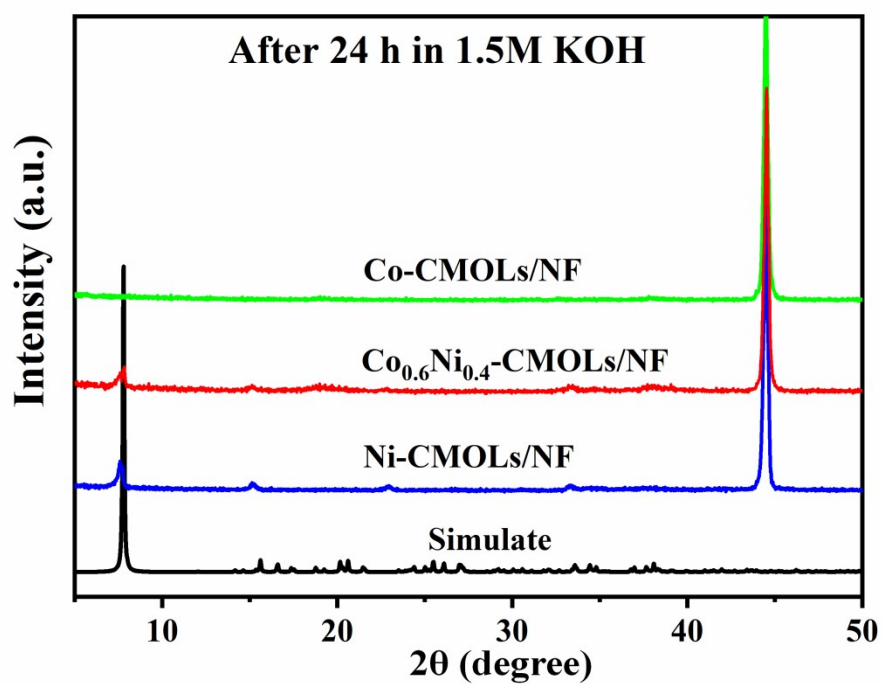


Fig. S10. PXRD patterns of three CMOLs/NF immersed in 1.5M KOH for 24 h.

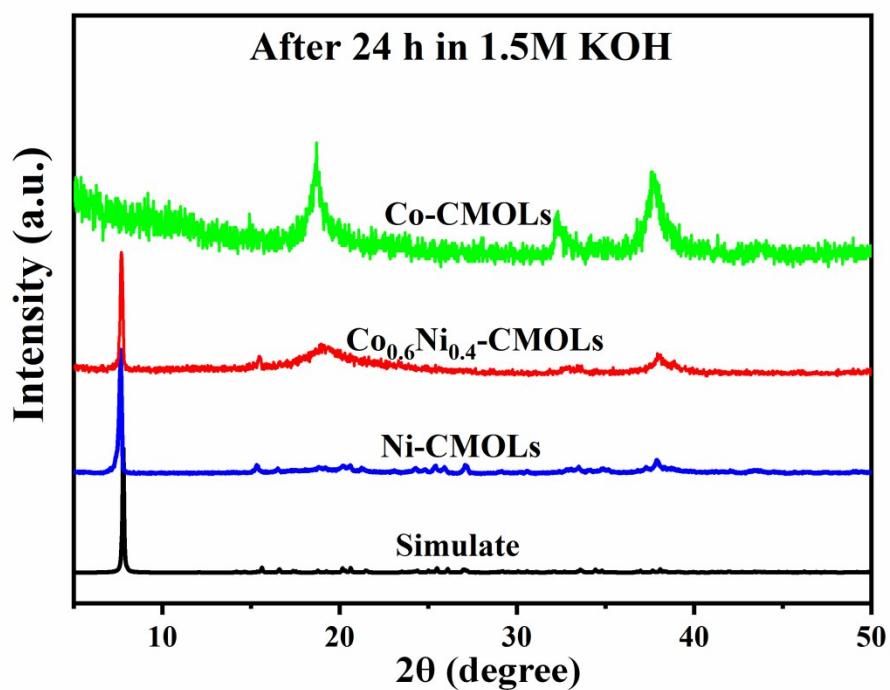


Fig. S11. PXRD patterns of three CMOLs immersed in 1.5M KOH for 24 h.

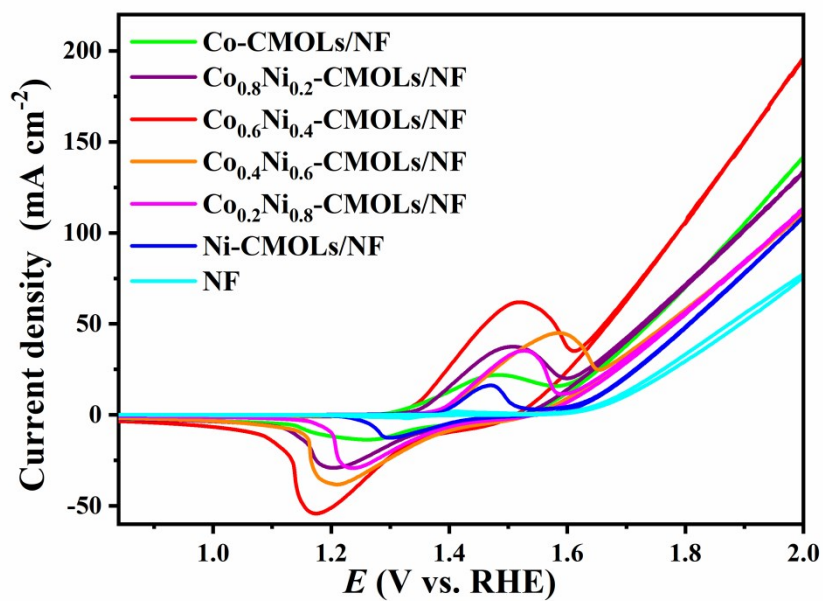


Fig. S12. CV curves of Co-CMOLs/NF, Ni-CMOLs/NF, $\text{Co}_x\text{Ni}_{1-x}$ -CMOLs/NF and blank NF in 1.0 M KOH for OER without iR correction.

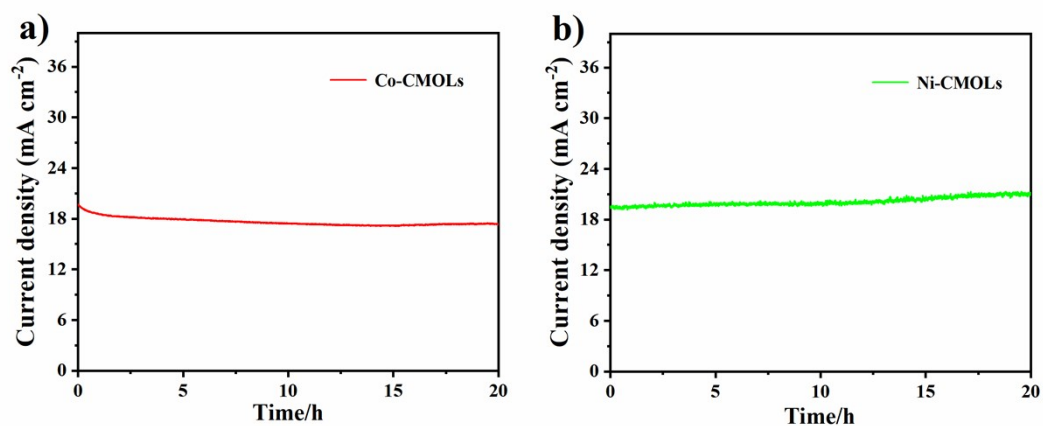


Fig. S13. a) Chronoamperometry acquired with Co-CMOLs under a constant potential for 20 h, b) Chronoamperometry acquired with Ni-CMOLs under a constant potential for 20 h.

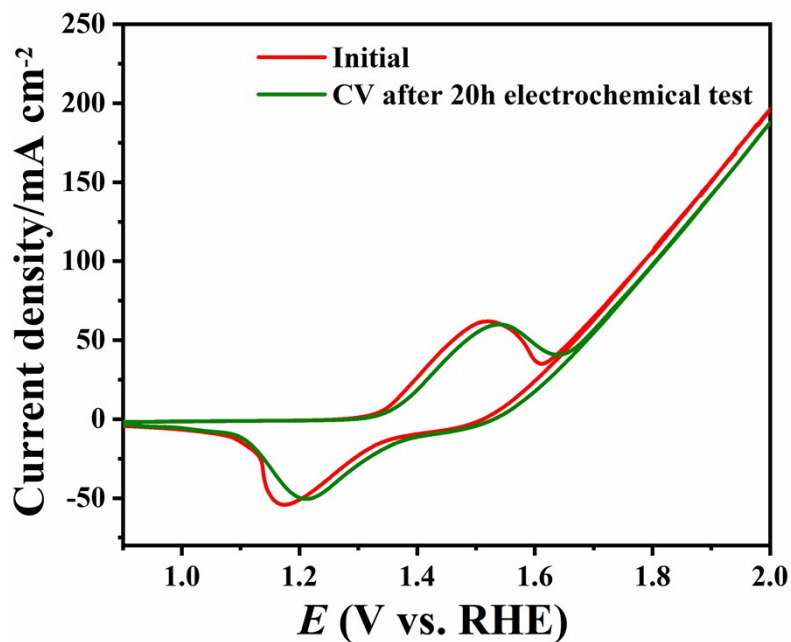


Fig. S14. Initial CV and CV after after 20 h electrochemical test.

Table S2: ICP analysis for $\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs}$.

$\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs}$	Sampling quality (g)	Volume (mL)	Coefficient of dilution	Instrument readings (mg/L)	Molar ratio
Co	0.0114	25	50	1.244	0.409
Ni	0.0114	25	50	1.791	0.591

Table S3: XPS high-resolution spectrum of $\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs}$, $\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs/NF}$ and $\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs/NF-KOH/1M}$.

$\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs}$	FWHM eV	Area (P) CPS.eV	Atomic %
Co 2p	3.08	173031.74	40.51
Ni 2p	2.52	276214.97	59.49
$\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs/NF}$	FWHM eV	Area (P) CPS.eV	Atomic %
Co 2p	4.04	111457.02	33.60
Ni 2p	2.90	239475.26	66.40
$\text{Co}_{0.6}\text{Ni}_{0.4}\text{-CMOLs/NF-KOH/1M}$	FWHM eV	Area (P) CPS.eV	Atomic %
Co 2p	3.48	109448.03	32.43
Ni 2p	2.78	247881.31	67.57

Table S4: OER performance of pristine MOFs comparison between recently reported studies and this work.²

MOFs	Over potential	Tafel slope	Electrolyte	Ref.
FeNi-DOBDC-3	270 ($\eta_{j=50}$)	49	1.0 M KOH	3
Co _{0.6} Fe _{0.4} -MOF-74	280	56	1.0 M KOH	4
Fe/Ni _{2.4} -MIL-53	244	48.7	1.0 M KOH	5
Fe/Ni _{2.4} /Co _{0.4} -MIL-53	219	53.5	1.0 M KOH	5
(Ni ₂ Co ₁) _{0.925} Fe _{0.075} -MOF-NF	257	41.3	1.0 M KOH	6
HE-MOF-RT	245	54	1.0 M KOH	7
PCN-Fe ₂ Co-Fe ₂ Ni	271	67.7	1.0 M KOH	8
UTSA-16	408	77	1.0 M KOH	9
CUMSs-ZIF-67	320	53.7	1.0 M KOH	10
CoBDC-Fe _{0.17}	219	61	1.0 M KOH	11
NiFe-UMNs	260	30	1.0 M KOH	12
Ni-Fe-NS	221	56	1.0 M KOH	13
Fe _{0.1} -Ni-MOF/NF	264 ($\eta_{j=100}$)	69.8	1.0 M KOH	14
LM-160-12	274	46.7	1.0 M KOH	15
NiCo-BDC BMNSs	230	61	1.0 M KOH	16
CoFe-MOF-OH	310	41	1.0 M KOH	17
Fe ₁ Ni ₂ -BDC	260	35	1.0 M KOH	18
Co ₃ Fe-MOF	280	38	1.0 M KOH	19
A _{2.7} B-MOF-FeCo _{1.6}	288	39	1.0 M KOH	20
Co/Cu-MOF (3)	395	94	1.0 M KOH	21
Co _{0.6} Ni _{0.4} -CMOLs/NF	320	95	1.0 M KOH	This work

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