# Interaction, Bond Formation or Reaction Between a Dimethylamino Group and an Adjacent Alkene or Aldehyde Group in Aromatic Systems Controlled by Remote Molecular Constraints. 

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## I. Synthesis of Compounds.

General. Solution NMR spectra were measured on a Jeol ECLIPSE 400 ECX or ECZ spectrometer at 400 MHz for ${ }^{1} \mathrm{H}$ and at 100.6 MHz for ${ }^{13} \mathrm{C}$ using $\mathrm{CDCl}_{3}$ as solvent and tetramethylsilane (TMS) as standard unless otherwise stated, and measured in p.p.m. downfield from TMS with coupling constants reported in Hz. IR spectra were recorded on a Perkin Elmer Spectrum 100 FT-IR Spectrometer using Attenuated Total Reflection sampling on solids or oils and are reported in $\mathrm{cm}^{-1}$. Mass spectra were recorded at the EPSRC Mass Spectrometry Centre at the University of Swansea. Chemical analysis data were obtained from Mr Stephen Boyer, London Metropolitan University.

## A. DIPHENYLS DERIVATIVES.

Experimental details for preparation and characterisation of 16, 17 and $\mathbf{1 8}$ have been previously described. ${ }^{\mathbf{S 1}}$

## Preparation of methyl ( $E$ )-2-cyano-3-(8'-(dimethylamino)-4',5'-diphenylnaphthalen-1'-

 yl)propenoate, 19/20.Dimethylamino-aldehyde 17 ( $75 \mathrm{mg}, 0.21 \mathrm{mmol}$ ), methyl cyanoacetate ( $0.075 \mathrm{~mL}, 0.85 \mathrm{mmol}$ ) and ethylenediamine diacetate ( $6 \mathrm{mg}, 0.03 \mathrm{mmol}$ ) were dissolved in anhydrous methanol ( 10 mL ) under nitrogen and refluxed for 24 h . The solvent was removed in vacuo and the crude product purified by flash column chromatography ( $4: 1$ hexane: ethyl acetate) to give $\mathbf{1 9 / 2 0}$ as a yellow solid ( $80 \mathrm{mg}, 87 \%$ ), m.p. $165-168^{\circ} \mathrm{C}$. ( 19 in the solid state, 20 in $\mathrm{CDCl}_{3}$ solution). $\delta \mathrm{H}$ ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}$ ): $8.95(1 \mathrm{H}, \mathrm{s}, 3-H), 7.58\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, \mathrm{Ar}-H_{1}\right), 7.39-7.51(3 \mathrm{H}$, $\left.\mathrm{m}, \mathrm{Ar}-\mathrm{H}_{3}\right)$, 6.87-6.97 (10H, m, Ar- $\mathrm{H}_{10}$ ), $3.94\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 2.77\left(6 \mathrm{H}, \mathrm{s}, 8\right.$ ' $\left.-\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta \mathrm{C}(100$ $\mathrm{MHz}, \mathrm{CDCl}_{3}, 2{ }^{\circ} \mathrm{C}$ ): $161.8(C=\mathrm{O}), 156.7$ (3-C), 148.4 ( $\left.8^{\prime}-C\right), 142.9,142.3,142.0,138.9$, $132.6,131.6,131.5,130.5,129.9,129.8,127.9,126.4,126.2,126.1,118.7$ (Ar- $C_{21}$ ), 117.4 $(\mathrm{CN}), 87.5(2-\mathrm{C}), 52.7\left(\mathrm{OCH}_{3}\right), 45.7\left(\mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}\right) ; v_{\max } / \mathrm{cm}^{-1}: 2167(\mathrm{C} \equiv \mathrm{N}), 1638(\mathrm{C}=\mathrm{O}), 1433$, 1371, 1345, 1276, 1250, 1189, 1095, 918, 849, 756, 695; Found: C, 80.53; H, 5.70; N, 6.57\%. Calc. for $\mathrm{C}_{29} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 80.53; H, 5.59; $\mathrm{N}, 6.48 \%$.

Preparation of $1-\left(1^{\prime}, 1^{\prime}-\right.$ dimethyl- $5^{\prime}, 6^{\prime}-$ diphenyl $-1^{\prime}, 2^{\prime}$-dihydrobenzo[cd]indol-1'-ium-2'-yl)-2,6-dioxocyclohexan-1-ide, 21.
Dimethylamino-aldehyde 17 ( $100 \mathrm{mg}, 0.28 \mathrm{mmol}$ ), 1,3-cyclohexandione ( $28 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) and ethylenediamine diacetate ( $5 \mathrm{mg}, 0.03 \mathrm{mmol}$ ) were dissolved in anhydrous methanol ( 10 mL ) under nitrogen and refluxed for 24 h . The solvent was removed in vacuo and the crude oil triturated with $\mathrm{Et}_{2} \mathrm{O}$, yielding a precipitate which was isolated give 21 as a yellow solid ( 35 mg , $31 \%$ ), m.p. decomp. $>150^{\circ} \mathrm{C} . \delta \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CD}_{2} \mathrm{Cl}_{2}, 24^{\circ} \mathrm{C}\right): 7.53\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 8^{\prime}-H\right)$, 7.49-7.52 ( $\left.2 \mathrm{H}, \mathrm{m}, 4^{\prime}-, 7^{\prime}-H\right), 7.23\left(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=7.3,1.4 \mathrm{~Hz}, 3^{\prime}-H\right), 7.12\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=1.4 \mathrm{~Hz}, 2^{\prime}-\right.$ H), 6.85-7.04 ( $10 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}_{10}$ ), $\left.3.62\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 3.23(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH})_{3}\right), 2.43-2.50(2 \mathrm{H}, \mathrm{m}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}$ ), 2.17-2.32 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}$ ), 1.90-2.06 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}$ ); $\delta \mathrm{C}(100$ $\left.\mathrm{MHz}, \mathrm{CD}_{2} \mathrm{Cl}_{2}, 24^{\circ} \mathrm{C}\right): 195.4 \& 193.4(C=\mathrm{O}), 146.8\left(8 \mathrm{a}^{\prime}-C\right), 141.4,140.9,140.6,137.6,137.5$, $134.1,130.8,130.0,129.6,127.5,127.3,127.1,126.5,126.0\left(\mathrm{Ar}-C_{14}\right), 119.0\left(3^{\prime}-C\right), 112.2\left(8^{\prime}-\right.$ C), $101.3\left(\mathrm{Ar}-\mathrm{C}_{1}\right), 89.8(2-\mathrm{C}), 55.5 \& 50.2\left(\mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}\right), 37.7 \& 36.7$ (3-, 5-C), 21.5 (4-C); $v_{\max } / \mathrm{cm}^{-1}: 3028,2965,2927,2868,1587,1522(\mathrm{C}=\mathrm{O}), 1444,1431,1401,1384,1349,1179$, $1120,1073,998,982,941,844,754,726,697 ;$ HRMS (ESI) calcd for $\mathrm{C}_{31} \mathrm{H}_{28} \mathrm{NO}_{2}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$: 446.2120 , found: 446.2137 .
${ }^{1} \mathrm{H}$ NMR spectrum for 21.

${ }^{13} \mathrm{C}$ NMR spectrum for 21.

(diethyl ether present)

## Preparation of N, N-Bis(4', $5^{\prime}$ 'diphenyl-8'-dimethylamino-naphthalen-1'-methylidene)-

 1,2-ethanediamine, 22.Dimethylamino aldehyde 17 ( $100 \mathrm{mg}, 0.28 \mathrm{mmol}$ ), nitromethane ( $0.05 \mathrm{~mL}, 0.85 \mathrm{mmol}$ ) and ethylenediamine diacetate ( $8 \mathrm{mg}, 0.04 \mathrm{mmol}$ ) were dissolved in anhydrous methanol ( 10 mL ) under nitrogen and stirred at room temperature for 24 h . The solvent was removed in vacuo and the crude oil triturated with $\mathrm{Et}_{2} \mathrm{O}$, and the resultant precipitate 22 was isolated as a yellow solid ( $22 \mathrm{mg}, 69 \%$ ), m.p. $218-221^{\circ} \mathrm{C} . \delta \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CD}_{2} \mathrm{Cl}_{2}, 24^{\circ} \mathrm{C}\right.$ ): $9.27(2 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{N}=\mathrm{CH})$, $7.69\left(2 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 2 \times 2^{\prime}-H\right), 7.30-7.33\left(4 \mathrm{H}, \mathrm{m}, 2 \times 3\right.$ '-, $\left.6^{\prime}-H\right), 7.26(2 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.8 \mathrm{~Hz}, 2$ x 7’-H), 6.87-6.96 (20H, m, $4 \times \mathrm{Ph}-H_{5}$ ), 4.05 ( $4 \mathrm{H}, \mathrm{s}, 1-, 2-\mathrm{H}_{2}$ ), 2.78 ( $12 \mathrm{H}, \mathrm{s}, 2 \mathrm{x} 8$ ' $\left.-\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right)$;
 134.7, 131.4, 130.9, 130.4, 130.3, 130.0, 129.9, 127.3 (Ar- $C_{34}$ ), 125.9 (2 x 2'-C), 125.8, 125.5 ( $\mathrm{Ar}-\mathrm{C}_{4}$ ), $115.6\left(2 \times 7{ }^{\prime}-\mathrm{C}\right), 62.3(1-2-\mathrm{C}), 44.9\left(2 \times \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right) ; v_{\max } / \mathrm{cm}^{-1}: 3075,2865,1632,1492$, 1440, 1390, 1354, 1276, 1183, 1146, 1114, 1036, 928, 833, 758, 695; HRMS (ESI) calcd for $\mathrm{C}_{52} \mathrm{H}_{47} \mathrm{~N}_{4}\left([\mathrm{M}+\mathrm{H}]^{+}\right): 727.3800$, found: 727.3771.

## ${ }^{1} \mathrm{H}$ NMR spectrum for 22.


${ }^{13} \mathrm{C}$ NMR spectrum for 22.


Preparation of 2-hydroxy-1,1-dimethyl-5,6-diphenyl-1,2-dihydrobenzo[cd]indol-1-ium chloride, 26.

Dimethylamino-aldehyde 17 ( $75 \mathrm{mg}, 0.21 \mathrm{mmol}$ ) was dissolved in anhydrous diethyl ether ( 5 mL ) and ethereal hydrochloric acid ( $1 \mathrm{M}, 0.32 \mathrm{~mL}, 0.32 \mathrm{mmol}$ ) was added dropwise with immediate formation of a white precipitate. The solution was stirred for a further 1 h . before the solid was collected by careful filtration under a flow of nitrogen. The solid was washed with cold anhydrous diethyl ether and dried under vacuum to give $\mathbf{2 6}$ as an off-white solid (61 $\mathrm{mg}, 74 \%)$, m.p. decomp. $>150^{\circ} \mathrm{C} . \delta_{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right): 7.81(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 3-H)$,
$7.70(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 4-H), 7.63(2 \mathrm{H}, \mathrm{AB}$ system, $\mathrm{J}=7.6 \mathrm{~Hz}, 7-, 8-H), 7.31(1 \mathrm{H}, \mathrm{s}, 2-H)$, 6.86-7.03 (11H, m, OH, Ar- $H_{10}$ ), $3.84\left(3 \mathrm{H}, \mathrm{s}, \mathrm{N}\left(\mathrm{CH}_{3}\right)\right), 3.43\left(3 \mathrm{H}, \mathrm{s}, \mathrm{N}\left(\mathrm{CH}_{3}\right)\right) ; \delta_{C}(100 \mathrm{MHz}$, $\mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}$ ): 144.0 ( $8 \mathrm{a}-C$ ), 142.1, 141.6, 140.0, 139.9 ( $\mathrm{Ar}-C_{4}$ ), 134.3 (4-C), 131.7 (7-C), 131.3, 129.4, 129.1, 127.4, 126.9, 126.7 (Ar- $C_{13}$ ), 122.7 (3-C), 113.7 (8-C), 110.5 (2-C), 53.5 \& $48.7\left(\mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}: 3050,2957,1492,1466,1442,1291,1176,1133,1105,930,861$, 842, 821, 762, 697; Found: C, 77.20; H, 5.58; N, 3.58\%. Calc. for $\mathrm{C}_{25} \mathrm{H}_{22} \mathrm{NOCl}: \mathrm{C}, 77.41$; H, 5.72; N, 3.61\%.

## Preparation of 2-hydroxy-1,1-dimethyl-5,6-diphenyl-1,2-dihydrobenzo[cd]indol-1-ium monomalonate, 27.

Dimethylamino-aldehyde 17 ( $100 \mathrm{mg}, 0.28 \mathrm{mmol}$ ) and Meldrum's acid ( $82 \mathrm{mg}, 0.56 \mathrm{mmol}$ ) were dissolved in anhydrous methanol ( 10 mL ) under nitrogen and stirred at room temperature for 24 h . The solvent was removed in vacuo and the crude oil triturated with $\mathrm{Et}_{2} \mathrm{O}$, yielding a precipitate which was isolated and recrystallised from ethyl acetate to give 27 as an off-white solid ( $63 \mathrm{mg}, 49 \%$ ), m.p. $168-171^{\circ} \mathrm{C} . \delta_{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right): 7.80(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 3-$ $H), 7.72(1 \mathrm{H}, \mathrm{br}$ s, 2-H), $7.70(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=6.9 \mathrm{~Hz}, 4-H), 7.58(2 \mathrm{H}, \mathrm{m}, 7-, 8-H), 6.86-7.06(10 \mathrm{H}$, $\left.\mathrm{m}, \operatorname{Ar}-\mathrm{H}_{10}\right), 3.50\left(6 \mathrm{H}, \mathrm{s}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right), 3.31\left(2 \mathrm{H}, \mathrm{s},(\mathrm{O}=\mathrm{C})_{2} \mathrm{CH}_{2}\right) ; \delta_{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24{ }^{\circ} \mathrm{C}\right)$ : $173.4(2 \times C=0), 144.4(8 \mathrm{a}-C), 141.9,141.8,140.0,139.8\left(\mathrm{Ar}-C_{4}\right), 134.1(4-C), 132.3\left(\mathrm{Ar}-C_{1}\right)$, 131.5 (7-C), 129.4, 129.3, 127.7, 127.4, 127.3, 126.9, 126.7 (Ar-C9), 123.1 (3-C), 113.7 (8-C), $49.6\left(\mathrm{~N}^{(+)}\left(\mathrm{CH}_{3}\right)_{2}\right), 38.8\left(\mathrm{CH}_{2}\right) ; v_{\text {max }} / \mathrm{cm}^{-1}: 3051,2415 \mathrm{br}, 2117 \mathrm{br}, 1890 \mathrm{br}, 1724(\mathrm{C}=\mathrm{O}), 1588$, $1485,1465,1440,1407,1366,1260,1180,1139,1098,985,930,869,821,753,712,695$; Found: C, $73.50 ; \mathrm{H}, 5.27$; N, 2.87\%. Calc. for $\mathrm{C}_{28} \mathrm{H}_{25} \mathrm{NO}_{5}$ : C, $73.88 ; \mathrm{H}, 5.53 ; \mathrm{N}, 3.08 \%$.

## B. ACENAPTHENE DERIVATIVES.

Experimental details for preparation and characterisation of $\mathbf{2 8}$ and $\mathbf{2 9}$ have been previously described. ${ }^{\mathbf{S 1}}$

## Preparation of methyl ( $E$ )-2-cyano-3-(6'-(dimethylamino)-1', $\mathbf{2}^{\prime}$-dihydroacenaphthylen-

 5'-yl)propenoate, 30.Dimethylamino-aldehyde 28 ( $100 \mathrm{mg}, 0.44 \mathrm{mmol}$ ) was dissolved in anhydrous $\mathrm{MeOH}(10 \mathrm{~mL})$. Methyl cyanoacetate ( $0.12 \mathrm{~mL}, 1.33 \mathrm{mmol}$ ) and ethylenediamine diacetate ( $8 \mathrm{mg}, 0.04 \mathrm{mmol}$ ) were added and the deep orange solution was heated to reflux for 16 h . The solvent was
removed in vacuo to yield a crude dark orange solid which was purified by flash column chromatography (1:4 EtOAc:petrol 40-60), to give 30 as an orange solid ( $120 \mathrm{mg}, 88 \%$ ), m.p. $82-85^{\circ} \mathrm{C} . \delta \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right)$ : $9.80(1 \mathrm{H}, \mathrm{s}, 3-H), 7.92\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 4^{\prime}-H\right), 7.29$ ( $1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 3^{\prime}-H$ ), $7.26\left(2 \mathrm{H}, \mathrm{s}, 7^{\prime}-, 8^{\prime}-H\right), 3.95\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 3.31-3.438\left(4 \mathrm{H}, \mathrm{m}, 1^{\prime}-, 2^{\prime}-\right.$ H), 2.69 ( $\left.6 \mathrm{H}, \mathrm{s}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right)$ : $163.6(\mathrm{C}=\mathrm{O})$, 160.3 (3-C), 151.9 (2a'C), 147.9 ( $6^{\prime}-C$ ), 142.2 ( $8 a^{\prime}-C$ ), 140.9 ( $5^{\prime}-C$ ), 129.7 ( $4^{\prime}-C$ ), 126.7 ( $5 \mathrm{a}^{\prime}-C$ ), 125.1 ( $2 \mathrm{a}^{\prime}{ }^{\prime}-C$ ), 120.4 \& $120.3\left(7^{\prime}-, 8^{\prime}-C\right), 119.3\left(3^{\prime}-C\right), 115.9(C \equiv \mathrm{~N}), 100.0(2-\mathrm{C}), 52.9\left(\mathrm{OCH}_{3}\right), 45.3\left(\mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}\right)$, 30.9 \& 29.7 ( $1^{\prime}-, 2^{\prime}-C$ ); $v_{\max } / \mathrm{cm}^{-1}: 2922$, 2849, 2825, 2786, $2223(\mathrm{C}=\mathrm{N}), 1727(\mathrm{C}=\mathrm{O}), 1589$, 1436, 1263, 1246, 1209, 1088, 957, 842, 829, 764, 751; HRMS (ESI) calcd for $\mathrm{C}_{19} \mathrm{H}_{19} \mathrm{~N}_{2} \mathrm{O}_{2}$ $\left([\mathrm{M}+\mathrm{H}]^{+}\right): 307.1440$, found: 307.1447.
${ }^{1} \mathrm{H}$ NMR spectrum for 30.

${ }^{13} \mathrm{C}$ NMR spectrum for 30 .


## Preparation of $1^{\prime}, 2^{\prime}, 3^{\prime}, 4^{\prime}$-Tetrahydro-1',2,2-trimethyl-spiro[1,3-dioxane-5,3'-acenaphth[5,6-bc]azepine]-4,6-dione, 34.

Dimethylamino-aldehyde 28 ( $120 \mathrm{mg}, 0.53 \mathrm{mmol}$ ) was dissolved in anhydrous DMSO ( 10 mL ) under nitrogen and Meldrum's acid ( $115 \mathrm{mg}, 0.8 \mathrm{mmol}$ ) was added, the deep orange solution was stirred at room temperature. After 24 h , a precipitate had formed which was collected via filtration and dried under vacuum to give 34 as a yellow solid ( $150 \mathrm{mg}, 80 \%$ ), m.p. 198-201 ${ }^{\circ} \mathrm{C}$. $\delta \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right)$ : 7.08-7.14 (2H, m, $\left.6^{\prime}-, 9^{\prime}-H\right), 6.99\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=6.9 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right)$, $6.75\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 10^{\prime}-H\right), 3.70-3.90\left(4 \mathrm{H}, \mathrm{m}, 2^{\prime}-, 4^{\prime}-H_{2}\right), 3.25-3.35\left(4 \mathrm{H}, \mathrm{m}, 8^{\prime}-, 7^{\prime}-H_{2}\right)$, $3.08\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 1.86\left(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{CH}_{3}\right), 1.75\left(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{CH}_{3}\right) ; \delta \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, 2 \mathrm{C}^{\circ} \mathrm{C}\right)$ : $169.0(2 \times C=O), 148.4,145.5,141.2,137.9,129.0\left(\mathrm{Ar}-C_{5}\right), 128.2\left(5^{\prime}-C\right), 124.7\left(\mathrm{Ar}-C_{1}\right), 119.5$ \& 119.1 ( $6^{\prime}-, 9^{\prime}-C$ ), 110.3 (10’-C), 104.9 (2-C), 63.6 ( $3^{\prime}-C$ ), 55.3 (2’-C), 41.5 ( $4^{\prime}-C$ ), 40.8 $\left(\mathrm{NCH}_{3}\right), 30.5 \& 29.6\left(7^{\prime}-, 8^{\prime}-\mathrm{C}\right), 29.4 \& 29.0\left(2-\mathrm{CH}_{3}\right) ; v_{\max } / \mathrm{cm}^{-1}: 1772,1735(\mathrm{C}=\mathrm{O}), 1589$, $1502,1472,1448,1422,1388,1295,1276,1239,1176,1151,1129,1101,1082,1030,941$, 913, 820, 777; Found: C, 71.65 ; H, 5.99; N, 4.05\%. Calc. for $\mathrm{C}_{21} \mathrm{H}_{21} \mathrm{NO}_{4}$ : C, 71.72; H, 5.98 ; N, 3.98\%.

## Preparation of $\mathbf{1 '}^{\prime}, \mathbf{2}^{\prime}, \mathbf{3}^{\prime}, \mathbf{4}^{\prime}$-Tetrahydro-1'-methyl-spiro[cyclohexan-1,3'-acenaphth[5,6-bc]azepine]-2,6-dione, 35.

Dimethylamino-aldehyde 28 ( $100 \mathrm{mg}, 0.44 \mathrm{mmol}$ ) was dissolved in anhydrous DMSO ( 10 mL ) under nitrogen and 1,3-cyclohexanedione ( $88 \mathrm{mg}, 0.79 \mathrm{mmol}$ ) was added, and the deep orange solution was stirred at room temperature. After $24 \mathrm{~h}, \mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ was added to produce a pale orange precipitate. The precipitate was collected via filtration and dried under vacuum to give 35 as a pale orange solid ( $84 \mathrm{mg}, 59 \%$ ), m.p. decomp. $>155^{\circ} \mathrm{C} . \delta \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right.$ ): $7.08\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.8 \mathrm{~Hz}, 9^{\prime}-H\right), 7.04\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=6.9 \mathrm{~Hz}, 6^{\prime}-H\right), 6.94\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=6.9 \mathrm{~Hz}, 5^{\prime}-H\right)$, $6.78\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.8 \mathrm{~Hz}, 10^{\prime}-H\right), 3.63-3.72\left(4 \mathrm{H}, \mathrm{m}, 2^{\prime}-, 4^{\prime}-H_{2}\right)$, 3.22-3.33 (4H, m, 7’-, $8^{\prime}-H_{2}$ ), $3.03\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 2.62-3.00\left(4 \mathrm{H}, \mathrm{m}, 3-, 5-H_{2}\right), 2.10-2.25\left(1 \mathrm{H}, \mathrm{m}, 4-H_{\alpha}\right), 1.75-1.89(1 \mathrm{H}, \mathrm{m}, 4-$ $H_{\beta}$ ); $\delta \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right): 207.5(2 \times \mathrm{C}=\mathrm{O}), 149.6,145.4,141.2,137.1,129.4\left(\mathrm{Ar}-C_{5}\right)$, $127.0\left(5^{\prime}-C\right), 125.3\left(\mathrm{Ar}_{-}\right), 119.4 \& 119.3\left(6^{\prime}-, 9^{\prime}-C\right), 109.9\left(10^{\prime}-C\right), 73.7\left(3^{\prime}-C\right), 57.8\left(2^{\prime}-C\right)$, $40.8\left(4^{\prime}-C\right), 40.6\left(\mathrm{NCH}_{3}\right), 37.9(3-, 5-C), 30.7 \& 29.7\left(7^{\prime}, 8^{\prime}-C\right), 18.5(4-C) ; v_{\max } / \mathrm{cm}^{-1}: 1718$, 1684 (C=O), 1593, 1504, 1470, 1431, 1317, 1272, 1190, 1108, 1079, 1004, 834, 767; Found: C, 78.90; H, 6.65; N, 4.37\%. Calc. for $\mathrm{C}_{21} \mathrm{H}_{21} \mathrm{NO}_{2}$ : C, 78.90; H, 6.58; N, 4.38\%.

## Preparation of $\mathbf{1}^{\prime}, \mathbf{2}^{\prime}, 3^{\prime}, 4^{\prime}$-Tetrahydro-1'-methyl-spiro[cyclopentan-1,3'-acenaphth $[5,6-$ bc]azepine]-2,5-dione, 36.

Dimethylamino-aldehyde $\mathbf{2 8}$ ( $100 \mathrm{mg}, 0.44 \mathrm{mmol}$ ) was dissolved in anhydrous DMSO ( 10 mL ) under nitrogen and 1,3 -cyclopentanedione ( $65 \mathrm{mg}, 0.67 \mathrm{mmol}$ ) was added, the deep orange solution was stirred at room temperature for 24 h to give a pale orange solution, $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ was added to produce a pale orange precipitate. The precipitate was dissolved by the addition of DCM ( 30 mL ) and the organic layer separated. The aqueous solution was extracted further with DCM ( $2 \times 30 \mathrm{~mL}$ ). The combined organics were washed with $\mathrm{H}_{2} \mathrm{O}(30 \mathrm{~mL})$ and brine ( 30 mL ), dried over $\mathrm{MgSO}_{4}$ and filtered. The solvent was removed in vacuo to give a crude orange solid which was purified by flash column chromatography (1:2 EtOAc:petrol 40-60), to give 36 as an off-white solid ( $54 \mathrm{mg}, 40 \%$ ), m.p. $180-183^{\circ} \mathrm{C} . \delta \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right.$ ): 7.07$7.11\left(2 \mathrm{H}, 2\right.$ overlapping $\left.\mathrm{d}, 6^{\prime}-, 9^{\prime}-H\right), 6.91\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=6.9 \mathrm{~Hz}, 5^{\prime}-H\right), 6.72\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 10^{\prime}-\right.$ H), $3.41\left(2 \mathrm{H}, \mathrm{br} \mathrm{s}, 4^{\prime}-H_{2}\right), 3.36\left(2 \mathrm{H}, \mathrm{s}, 2^{\prime}-H_{2}\right), 3.25-3.34\left(4 \mathrm{H}, \mathrm{m}, 7{ }^{\prime}-, 8^{\prime}-H_{2}\right), 3.05(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{NCH}_{3}\right), 2.91-3.01\left(2 \mathrm{H}, \mathrm{m}, 3-\right.$ or $\left.4-\mathrm{H}_{2}\right), 2.70-2.81\left(2 \mathrm{H}, \mathrm{m}, 3-\right.$ or $\left.4-\mathrm{H}_{2}\right) ; \delta \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$, $24^{\circ} \mathrm{C}$ ): $213.6(2 \times \mathrm{x}=\mathrm{O}), 148.6,145.3,141.1,137.5,128.8\left(\mathrm{Ar}-C_{5}\right), 127.6\left(5^{\prime}-\mathrm{C}\right), 124.9$ (Ar$\left.C_{1}\right), 119.3 \& 119.1\left(6^{\prime}-, 9^{\prime}-C\right), 110.3\left(10^{\prime}-C\right), 63.5\left(3^{\prime}-C\right), 59.0\left(2^{\prime}-C\right), 41.1\left(\mathrm{~N} C H_{3}\right), 38.9\left(4^{\prime}-\right.$ C), 34.7 (3-, 4-C), $30.5 \& 29.5\left(7^{\prime}-, 8^{\prime}-C\right) ; v_{\max } / \mathrm{cm}^{-1}: 2917,1707(\mathrm{C}=\mathrm{O}), 1591,1502,1470$, 1416, 1285, 1271, 1149, 1101, 1077, 1037, 1000, 943, 834, 805, 766; HRMS (ESI) calcd for $\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{NO}_{2}\left([\mathrm{M}+\mathrm{H}]^{+}\right): 306.1494$, found: 306.1486 .
${ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{3 6}$.



## C. FLUORENE DERIVATIVES.

Preparation of 4,5-diiodo-9,9-dimethyl-9H-fluorene, 43.
9, 9-Dimethylfluorene $\mathbf{4 2}^{\mathrm{S} 2}(1.00 \mathrm{~g}, 5.15 \mathrm{mmol})$ was stirred in anhydrous TMEDA ( 3.09 mL , 20.62 mmol ) whilst $n$-BuLi ( 1.6 M in hexanes, $12.89 \mathrm{~mL}, 20.62 \mathrm{mmol}$ ) was steadily added, producing an orange solution which was heated to $60^{\circ} \mathrm{C}$ for $5 \mathrm{~h} .{ }^{\mathrm{S} 3}$ The deep brown solution was diluted with anhydrous THF ( 100 mL ), cooled to $-78^{\circ} \mathrm{C}$ and iodine ( $13.09 \mathrm{~g}, 51.50 \mathrm{mmol}$ ) added. The reaction was allowed to warm to room temperature and stirred for 16 h . The deep brown solution was quenched with sat. sodium thiosulfate solution $(150 \mathrm{~mL})$ and stirred for 10 min. The aqueous layer was washed with EtOAc ( $3 \times 40 \mathrm{~mL}$ ) and the combined organic layers washed with sat. sodium thiosulfate solution ( 3 x 40 mL ), $\mathrm{H}_{2} \mathrm{O}(40 \mathrm{~mL})$, brine ( 40 mL ) and dried over $\mathrm{MgSO}_{4}$. The solvent was removed in vacuo to give a crude thick brown oil which was first purified by flash column chromatography (hexanes), to give two close running bands ( $R_{f}: 0.32$ and 0.27 ). The faster band contained a mono-iodinated-9,9-dimethyl-fluorene ( $\delta \mathrm{H}$ ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}$ ): $8.82\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=6.8 \mathrm{~Hz}, \operatorname{Ar}-H_{1}\right), 7.83\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=6.8 \mathrm{~Hz} \mathrm{Ar}-\mathrm{H}_{1}\right)$, 7.41-7.46 (4H, m, Ar- $H_{4}$ ), $7.00\left(1 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.6 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H}_{1}\right), 1.48(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{CH}), \delta \mathrm{C}(100$ $\mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}$ ): 156.6, 154.1, 140.3, 139.0, 128.2, 128.1, 126.2, 122.4, 122.3 ( $\mathrm{Ar}-\mathrm{C}_{11}$ ), 88.2 (C-I), 46.4 (9-C), $27.3\left(2 \times \mathrm{CH}_{3}\right)$ ). The slower band contained starting material 42 and the diiodo compound 43. Increasing the polarity of the solvent eluted further materials. The mixture of $\mathbf{4 2}$ and $\mathbf{4 3}$ co-eluted in most solvent combinations, however they just separated ( $R_{f}$ : 0.54 (42), 0.49 (43)) in hexane/ethyl acetate $59 / 5$, though to gain a reasonable amount of material for further reactions it was necessary to accept some $\mathbf{4 2}$ as impurity. The product was
obtained as a pale yellow solid ( $570 \mathrm{mg}, 25 \%$ ), m.p. $50-53^{\circ} \mathrm{C} . \delta \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right)$ : $7.93(2 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.8 \mathrm{~Hz}, 3,6-H), 7.40(2 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.2 \mathrm{~Hz}, 1,8-H), 7.04(2 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.7 \mathrm{~Hz}, 2,7-$ H), $1.43\left(6 \mathrm{H}, \mathrm{s}, 9-\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, 2{ }^{\circ} \mathrm{C}\right): 156.9,142.7\left(\mathrm{Ar}-\mathrm{C}_{4}\right), 140.8(3,6-\mathrm{C})$, $128.9(2,7-C), 121.5(1,8-C), 87.3(4,5-C), 47.1(9-\mathrm{C}), 27.7\left(9-\left(\mathrm{CH}_{3}\right)_{2}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}: 2957,2920$, 2857, 1442, 1392, 1075, 931, 779, 762, 732, 661. Crystal structure determined (p. 29). This synthesis was not optimised.

## ${ }^{1} \mathrm{H}$ NMR spectrum of 43 .


${ }^{13} \mathrm{C}$ NMR spectrum of 43 .


Preparation of 4-(dimethylamino)-9, 9-dimethyl-9H-fluorene-5-carbaldehyde, 44.
The $d i$-iodo-fluorene 43 ( $0.4 \mathrm{~g}, 0.90 \mathrm{mmol}$ ) was dissolved in anhydrous $\mathrm{Et}_{2} \mathrm{O}(15 \mathrm{~mL})$ under nitrogen and cooled to $-78^{\circ} \mathrm{C} . n-\mathrm{BuLi}(1.6 \mathrm{M}$ in hexanes, $0.62 \mathrm{~mL}, 0.99 \mathrm{mmol}$ ) was steadily added, and the orange solution was stirred at $-78^{\circ} \mathrm{C}$ for 2 h . Anhydrous DMF $(0.36 \mathrm{~mL}, 4.63$ mmol ) was added and the reaction was allowed to warm to room temperature. After 16 h . the resulting green/yellow solution was quenched with $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL})$ and stirred for 10 min . The aqueous solution was washed with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{~mL})$ and the combined organic layers washed with $\mathrm{H}_{2} \mathrm{O}(40 \mathrm{~mL})$, brine $(40 \mathrm{~mL})$ and dried over $\mathrm{MgSO}_{4}$. The solvent was removed in vacuo to give a yellow/green oil which was purified by flash column chromatography (5:95 EtOAc/petrol 40-60), to give 44 as a yellow solid ( $100 \mathrm{mg}, 42 \%$ ), m.p. $80-83^{\circ} \mathrm{C} . \delta \mathrm{H}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right): 10.86(1 \mathrm{H}, \mathrm{s}, \mathrm{CHO}), 7.61(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=7.3,0.9 \mathrm{~Hz}, 6-H), 7.52(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=7.3$, $0.9 \mathrm{~Hz}, 8-H), 7.34(2 \mathrm{H}, \mathrm{m}, 2-, 7-H), 7.17(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 1-H), 7.12(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.8 \mathrm{~Hz}, 3-$ H), 2.63 ( $\left.6 \mathrm{H}, \mathrm{s}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.47\left(6 \mathrm{H}, \mathrm{s}, 9-\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right): 192.5(\mathrm{C}=\mathrm{O})$, 156.1, 154.0 ( $\mathrm{Ar}-C_{2}$ ), 150.1 (4-C), 138.1 ( $\mathrm{Ar}-C_{1}$ ), 133.5 (5-C), 131.8 ( $\left.\mathrm{Ar}-C_{1}\right), 129.5 \& 127.1$ (2-,7-C), 126.3 (6-C), 125.8 (8-C), 118.3 (3-C), 118.1 (1-C), $\left.\left.47.0(9-C), 43.5\left(\mathrm{~N}_{(~(~}^{(1)}\right)_{3}\right)_{2}\right), 27.4$ $\left(9-\left(\mathrm{CH}_{3}\right)_{2}\right) ; v_{\max } / \mathrm{cm}^{-1}: 2950,2834,2786,1677$ (C=O), 1581, 1481, 1455, 1381, 1317, 1289, 1224, 1200 1181, 1118, 985, 795, 764, 726; HRMS (ESI) calcd for $\mathrm{C}_{18} \mathrm{H}_{20} \mathrm{NO}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$: 266.1545 , found: 266.1538 .
${ }^{1} \mathrm{H}$ NMR spectrum of 44.

${ }^{13} \mathrm{C}$ NMR spectrum of 44 .


Preparation of 2-cyano-2-(4'-dimethylamino-9',9'-dimethyl-9H-fluoren-5'-yl)propenenitrile, 45.
Dimethylamino-aldehyde 44 ( $100 \mathrm{mg}, 0.38 \mathrm{mmol}$ ) was dissolved in anhydrous $\mathrm{MeOH}(15 \mathrm{~mL})$. Malononitrile ( $30 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) and ethylenediamine diacetate ( $10 \mathrm{mg}, 0.06 \mathrm{mmol}$ ) were added and the solution was heated to reflux for 2 h . The solvent was removed in vacuo to yield a crude orange solid which was purified by flash column chromatography (3:97 EtOAc:petrol $40-60$ ), to give $\mathbf{4 5}$ as an orange solid ( $70 \mathrm{mg}, 59 \%$ ), m.p. $122-125^{\circ} \mathrm{C} . \delta \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24\right.$ $\left.{ }^{\circ} \mathrm{C}\right): 9.07(1 \mathrm{H}, \mathrm{s}, \mathrm{C} H), 7.67\left(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=7.8 \mathrm{~Hz}, 6^{\prime}-H\right), 7.56\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 8^{\prime}-H\right), 7.39(2 \mathrm{H}$, $\left.\mathrm{m}, 2^{\prime}-, 7^{\prime}-H\right), 7.22\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.8 \mathrm{~Hz}, 1^{\prime}-H\right), 7.19\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.8 \mathrm{~Hz}, 3^{\prime}-H\right), 2.68(6 \mathrm{H}, \mathrm{s}$,
$\left.\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.48\left(6 \mathrm{H}, \mathrm{s}, 9^{\prime}-\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, 2{ }^{\circ}{ }^{\circ} \mathrm{C}\right): 163.8$ (3-C), 156.1, 154.6, 149.1, 138.3, 131.6 ( $\mathrm{Ar}-C_{5}$ ), 129.9 ( $2^{\prime}-$ or $7^{\prime}-C$ ), $127.8\left(6^{\prime}-C\right), 127.4$ ( $2^{\prime}-$ or $7^{\prime}-C$ ), 126.3 ( $\mathrm{Ar}-$ $C_{1}$ ), 126.1 ( $8^{\prime}-C$ ), $118.9\left(3^{\prime}-C\right), 118.8\left(1^{\prime}-C\right), 114.6(2 \times C \equiv \mathrm{~N}), 112.9\left(C(\mathrm{CN})_{2}\right), 46.9\left(9^{\prime}-C\right)$, $44.2\left(\mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}\right), 27.1\left(9{ }^{\prime}-\left(\mathrm{CH}_{3}\right)_{2}\right) ; v_{\max } / \mathrm{cm}^{-1}: 2942,2862,2223(\mathrm{C} \equiv \mathrm{N}), 1582,1481,1459,1314$, 1183, 1043, 984, 879, 797, 764, 730; Found: C, 80.27; H, 6.23; N, 13.36\%. Calc. for $\mathrm{C}_{21} \mathrm{H}_{19} \mathrm{~N}_{3}$ : C, 80.41 ; H, 6.06; N, 13.40\%.

## Preparation of $\boldsymbol{E}$-2-(4'-Dimethylamino-9',9'-dimethyl-9H-fluorene-5'-yl)-1-nitroethene,

46. 

Dimethylamino-aldehyde 44 ( $55 \mathrm{mg}, 0.21 \mathrm{mmol}$ ) was dissolved in anhydrous $\mathrm{MeOH}(5 \mathrm{~mL})$. Nitromethane ( $0.03 \mathrm{~mL}, 0.56 \mathrm{mmol}$ ) and ethylenediamine diacetate ( $5 \mathrm{mg}, 0.03 \mathrm{mmol}$ ) were added and the solution was heated to reflux for 24 h . The solvent was removed in vacuo to yield a crude orange solid which was purified by flash column chromatography (5:95 EtOAc:petrol 40-60), to give 46 as an orange solid ( $40 \mathrm{mg}, 63 \%$ ), m.p. $99-102^{\circ} \mathrm{C} . \delta \mathrm{H}(400$ $\mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}$ ): $9.93(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=13.7 \mathrm{~Hz}, 2-H), 7.45-7.52\left(3 \mathrm{H}, \mathrm{m}, 1-, 6^{\prime}-, 8^{\prime}-H\right), 7.29-$ $7.39\left(2 \mathrm{H}, \mathrm{m}, 2^{\prime}-, 7^{\prime}-H\right), 7.10-7.18\left(2 \mathrm{H}, \mathrm{m}, 1^{\prime}-, 3^{\prime}-H\right), 2.72\left(6 \mathrm{H}, \mathrm{s}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.46\left(6 \mathrm{H}, \mathrm{s}, 9^{\prime}-\right.$ $\left.\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, 2{ }^{\circ} \mathrm{C}\right): 156.7,155.2,150.1\left(\mathrm{Ar}-\mathrm{C}_{3}\right), 141.7(3-\mathrm{C}), 140.2,135.0$ $2-C), 131.1\left(\mathrm{Ar}-C_{2}\right), 129.5 \& 127.0\left(2^{\prime}-, 7{ }^{\prime}-C\right), 126.6,126.2,125.1$ ( $\left.\mathrm{Ar}-C_{3}\right), 117.9 \& 117.2$ (1'-, 3'-C), $46.3\left(9^{\prime}-C\right), 43.8\left(\mathrm{~N}^{\prime}\left(\mathrm{CH}_{3}\right)_{2}\right), 27.7\left(9^{\prime}-\left(\mathrm{CH}_{3}\right)_{2}\right) ; v_{\max } / \mathrm{cm}^{-1}: 2957,2860,1623,1582$, 1545, 1500, 1481, 1332, 1259, 1090, 985, 969, 799, 732; HRMS (ESI) calcd for $\mathrm{C}_{19} \mathrm{H}_{21} \mathrm{~N}_{2} \mathrm{O}_{2}$ $\left([\mathrm{M}+\mathrm{H}]^{+}\right): 309.1603$, found: 309.1594.
${ }^{1} \mathrm{H}$ NMR spectrum of 46.

${ }^{13} \mathrm{C}$ NMR spectrum of 46


Preparation of $1^{\prime}, 8^{\prime}, 8^{\prime}$-Trimethyl- $\mathbf{1}^{\prime}, 2^{\prime}, 3^{\prime}, \mathbf{4}^{\prime}$-tetrahydro- $\mathbf{8}^{\prime} \mathbf{H}$-spiro[cyclopentane-1,3'-fluoreno[4,5-bcd] azocine]-2,5-dione, 47.

Dimethylamino-aldehyde 44 ( $100 \mathrm{mg}, 0.38 \mathrm{mmol}$ ) was dissolved in anhydrous DMSO ( 10 mL ) under nitrogen and 1,3 -cyclopentandione ( $44 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) was added, and the deep orange solution was stirred at room temperature. After 24 h , the solution was diluted with water (15 mL ) and extracted with EtOAc ( $3 \times 20 \mathrm{~mL}$ ). The combined organics were washed with brine ( $1 \times 20 \mathrm{~mL}$ ), dried over $\mathrm{MgSO}_{4}$ and filtered. The solvent was removed in vacuo to yield a crude brown oil which was purified by flash column chromatography (10:90 EtOAc:petrol 40-60), to give 47 as an orange solid ( $53 \mathrm{mg}, 41 \%$ ), m.p. $60-63^{\circ} \mathrm{C} . \delta \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right.$ ): 7.30 $\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 7^{\prime}-H\right), 7.23\left(1 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.8 \mathrm{~Hz}, 10^{\prime}-H\right), 7.12\left(1 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.3 \mathrm{~Hz}, 6^{\prime}-H\right), 6.95$ $\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.8 \mathrm{~Hz}, 9^{\prime}-H\right), 6.85\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=8.2 \mathrm{~Hz}, 5^{\prime}-H\right), 6.77\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 11^{\prime}-H\right), 4.26$ $\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=13.7 \mathrm{~Hz}, 4^{\prime}-H_{\alpha}\right), 3.86\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=15.1 \mathrm{~Hz}, 2^{\prime}-H_{\alpha}\right), 3.16\left(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=13.1,1.4 \mathrm{~Hz}, 4^{\prime}-\right.$ $H_{\beta}$ ), $3.02\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 2.75-2.90\left(4 \mathrm{H}, \mathrm{m}, 3-, 4-H_{2}\right), 2.57(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=13.7,1.9 \mathrm{~Hz}, 2-H), 1.47$ ( $3 \mathrm{H}, \mathrm{s}, 8^{\prime}-\mathrm{CH}_{3}$ ), $1.40\left(3 \mathrm{H}, \mathrm{s}, 8^{\prime} \mathrm{CH}_{3}\right) ; ~ \delta \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right): 215.3 \& 212.0(2 \times \mathrm{C}=\mathrm{O})$, 156.7, 153.3, 148.7, $139.8\left(\mathrm{Ar}-C_{4}\right), 130.7\left(5^{\prime}-C\right), 128.9\left(\mathrm{Ar}-C_{1}\right), 128.3\left(10^{\prime}-C\right), 126.7\left(\mathrm{Ar}-C_{1}\right)$, 126.1 ( $6^{\prime}-C$ ), 121.6 ( $7^{\prime}-C$ ), $113.8\left(9^{\prime}-C\right), 112.6$ ( $11^{\prime}-C$ ), 57.8 ( $3^{\prime}-C$ ), 57.1 ( $2^{\prime}-C$ ), $46.2\left(8^{\prime}-C\right)$, $40.4\left(\mathrm{NCH}_{3}\right), 37.7\left(4^{\prime}-C\right), 36.1 \& 35.1(3-, 4-C), 28.2 \& 27.7\left(2 \times 8^{\prime}-\mathrm{CH}_{3}\right) ; v_{\max } / \mathrm{cm}^{-1}: 2957$, 2920, 2862, 1714 (C=O), 1584, 1485, 1442, 1420, 1280, 1168, 1153, 1129, 987, 907, 790, 728;
${ }^{1} \mathrm{H}$ NMR spectrum of 47.

${ }^{13} \mathrm{C}$ NMR spectrum of 47 .


## Preparation of $\mathbf{1}^{\prime}, \mathbf{2 , 2 , 8 ^ { \prime }}, 8^{\prime}$-Pentamethyl-1', $\mathbf{2}^{\prime}, 3^{\prime}, 4^{\prime}$-tetrahydro- $\mathbf{8}^{\prime} \boldsymbol{H}$-spiro $[1,3$-dioxane-5,3'-fluoreno[4,5-bcd]azocine]-4,6-dione, 48.

Dimethylamino-aldehyde 44 ( $143 \mathrm{mg}, 0.54 \mathrm{mmol}$ ) was dissolved in anhydrous $\mathrm{MeOH}(10 \mathrm{~mL})$. Meldrum's acid ( $109 \mathrm{mg}, 0.76 \mathrm{mmol}$ ) and ethylenediamine diacetate ( $9 \mathrm{mg}, 0.05 \mathrm{mmol}$ ) were added and the reaction heated to reflux for 5 h , with a precipitate forming after 15 min . The precipitate was filtered, washed with cold MeOH and dried in vacuo to give 48 as a pale yellow
solid ( $150 \mathrm{mg}, 71 \%$ ), m.p. $204-207^{\circ} \mathrm{C} . \delta \mathrm{H}\left(400 \mathrm{MHz}\right.$, acetone- $\left.\mathrm{d}_{6}, 24^{\circ} \mathrm{C}\right): 7.34(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.8$ $\left.\mathrm{Hz}, 7^{\prime}-H\right), 7.20\left(1 \mathrm{H}, \mathrm{t}, \mathrm{J}=8.2 \mathrm{~Hz}, 10^{\prime}-H\right), 7.12\left(1 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.8 \mathrm{~Hz}, 6^{\prime}-H\right), 7.00(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3$ $\left.\mathrm{Hz}, 9^{’}-H\right), 6.94\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 5^{\prime}-H\right), 6.87\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=8.2 \mathrm{~Hz}, 11^{\prime}-H\right), 4.51(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=13.3$ $\left.\mathrm{Hz}, 4^{\prime}-H_{\alpha}\right), 4.12\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=15.1 \mathrm{~Hz}, 2^{\prime}-H_{\alpha}\right), 3.71\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=15.6 \mathrm{~Hz}, 4^{\prime}-H_{\beta}\right), 3.04(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=$ $\left.13.7 \mathrm{~Hz}, 2^{\prime}-H_{\beta}\right), 2.96\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 1.81\left(6 \mathrm{H}, \mathrm{m}, 2-\left(\mathrm{CH}_{3}\right)_{2}\right), 1.39\left(6 \mathrm{H}, \mathrm{m}, 8^{\prime}-\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta \mathrm{C}(100$ MHz , acetone- $\mathrm{d}_{6}, 2{ }^{\circ} \mathrm{C}$ ): $168.9 \& 166.2(\mathrm{C}=\mathrm{O}), 156.6,152.9,148.9,139.4$ (Ar-C4), 131.5 (5’C), $129.7\left(\mathrm{Ar}-C_{1}\right), 128.4\left(10^{\prime}-C\right), 126.2\left(\mathrm{Ar}-C_{1}\right), 126.1\left(6^{\prime}-C\right), 121.4$ (7'-C), $114.0\left(9^{\prime}-C\right), 112.7$ (11'-C), 105.0 (2-C), 60.8 (2’-C), 50.1 ( $\left.3^{\prime}-C\right), 45.8\left(8^{\prime}-C\right), 40.3\left(\mathrm{NCH}_{3}\right), 39.1\left(4^{\prime}-C\right), 28.6$ \& $28.3\left(2-\left(\mathrm{CH}_{3}\right)_{2}\right), 27.8 \& 27.0\left(8^{\prime}-\left(\mathrm{CH}_{3}\right)_{2}\right) ; v_{\max } / \mathrm{cm}^{-1}: 2965,2868,1778,1735(\mathrm{C}=\mathrm{O}), 1571$, 1477, 1436, 1379, 1278, 1258, 1198, 1026, 948, 795, 740; Found: C, 73.49; H, 6.36; N, 3.86\%. Calc. for $\mathrm{C}_{24} \mathrm{H}_{25} \mathrm{NO}_{4}$ : C, 73.57; H, 6.39; N, 3.58\%.

## Preparation of 3-Benzoyl-3-nitro-1,2,3,4-tetrahydro-1,8,8-trimethyl-8H-fluoreno[4,5bcd]azocine, 49.

Dimethylamino-aldehyde $44(55 \mathrm{mg}, 0.21 \mathrm{mmol})$ was dissolved in anhydrous $\mathrm{MeOH}(10 \mathrm{~mL})$. Benzoyl-nitromethane ( $69 \mathrm{mg}, 0.42 \mathrm{mmol}$ ) and ethylenediamine diacetate ( $5 \mathrm{mg}, 0.03 \mathrm{mmol}$ ) were added and the solution was refluxed for 4 h . The solvent was removed in vacuo to yield a crude yellow oil which was purified by flash column chromatography (5:95 EtOAc:petrol $40-60$ ), to give 49 as an orange solid ( $65 \mathrm{mg}, 76 \%$ ), m.p. $172-175^{\circ} \mathrm{C} . \delta \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24\right.$ $\left.{ }^{\circ} \mathrm{C}\right): 7.87(2 \mathrm{H}, \mathrm{dd}, \mathrm{J}=8.2,0.91 \mathrm{~Hz}$, ortho- $\mathrm{Ph}-H), 7.65(1 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.8 \mathrm{~Hz}$, para-Ph-H), $7.51(2 \mathrm{H}$, $\mathrm{t}, \mathrm{J}=7.8 \mathrm{~Hz}$, meta-Ph-H), 7.20-7.30 (2H, m, 7-, 10-H), $6.98(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=7.3,0.9 \mathrm{~Hz}, \mathrm{Ar}-H)$, $6.95(1 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.3 \mathrm{~Hz}, 6-H), 6.83(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=8.2 \mathrm{~Hz}, \operatorname{Ar}-H), 6.30(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=7.8,0.9 \mathrm{~Hz}, 5-$ $H), 4.89\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=13.3 \mathrm{~Hz}, 2-H_{\alpha}\right), 4.60\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=16.0 \mathrm{~Hz}, 4-H_{\alpha}\right), 3.90(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=16.0,2.3$ $\left.\mathrm{Hz}, 4-H_{\beta}\right), 3.60\left(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=13.3,2.3 \mathrm{~Hz}, 2-H_{\beta}\right), 2.81\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 1.44\left(3 \mathrm{H}, \mathrm{s}, 8-\mathrm{C}\left(\mathrm{CH} H_{3}\right)\right)$, $1.40\left(3 \mathrm{H}, \mathrm{s}, 8-\mathrm{C}\left(\mathrm{CH}_{3}\right)\right) ; \delta \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, 24^{\circ} \mathrm{C}\right): 189.4(\mathrm{C}=\mathrm{O}), 156.8$ ( $8 \mathrm{a}-\mathrm{C}$ ), 153.5 (7aC), 148.4 (11a-C), 139.9 ( $\mathrm{Ar}-C_{1}$ ), 134.1 ( $p-\mathrm{Ph}-C$ ), 129.7 (6-C), 129.2 ( $m-\mathrm{Ph}-C$ ), 128.7 ( $\mathrm{Ar}-C_{1}$ ), 128.6 (o-Ph-C), 128.5 ( $\mathrm{Ar}-C_{I}$ ), 126.5 (6-C), 126.4, 126.0, 120.0, 114.3, 112.3 ( $\mathrm{Ar}-C_{5}$ ), 94.0 (3C), $62.6(2-\mathrm{C}), 46.4(8-\mathrm{C}), 39.7\left(\mathrm{NCH}_{3}\right), 38.2(4-\mathrm{C}), 28.1 \& 27.5\left(8-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{2}\right) ; v_{\max } / \mathrm{cm}^{-1}: 2980$, 2821, 1686 (C=O), 1533, 1485, 1444, 1425, 1321, 1269, 1241, 1200, 1179, 1159, 1000, 922, 786, 728, 695; HRMS (ESI) calcd for $\mathrm{C}_{26} \mathrm{H}_{25} \mathrm{~N}_{2} \mathrm{O}_{3}\left([\mathrm{M}+\mathrm{H}]^{+}\right): 413.1865$, found: 413.1854.
${ }^{1} \mathrm{H}$ NMR spectrum of 49 .

(some evidence two non-interconverting conformers).
${ }^{13} \mathrm{C}$ NMR spectrum of 49.


Preparation of 3-Nitro-1,2,3,4-tetrahydro-1,8,8-trimethyl-8H-fluoreno[4,5-bcd]azocine,
50.

Recrystallisation of the fused azocine 49 from a DCM/hexanes solution gave two separate crystal types. Separation of the two crystal systems under a microscope gave two samples; the first the starting azocine $\mathbf{4 9}$ and the second a decomposition product $\mathbf{5 0}$ as pale yellow plates, m.p. $115-118^{\circ} \mathrm{C}$. $v_{\max } / \mathrm{cm}^{-1}: 2957,2862,1584,1537,1474,1454,1438,1358,1269,1177,984$,

793, 730; HRMS (ESI) calcd for $\mathrm{C}_{19} \mathrm{H}_{21} \mathrm{~N}_{2} \mathrm{O}_{2}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$: 309.1603, found: 309.1595. Crystal structure was determined (p. 29 and main article).

## II. X-Ray Crystallography.

## General.

Low temperature ( 150 K ) X-ray diffraction data ( $\mathrm{MoK} \alpha$ for all compounds but for 22 and 44 which were measured with $\mathrm{CuK}_{\alpha}$ ) were measured on an Rigaku Oxford Diffraction Xcalibur diffractometer equipped with a Sapphire detector and an 700 series Cryostream low temperature system using the CrysAlis-Pro software package. ${ }^{S 4}$ Structures were solved and refined using the SHELXS and SHELXL suite of programs ${ }^{55}$ using the XSEED interface ${ }^{\text {S6 }}$ or OLEX ${ }^{2} .{ }^{\text {S7 }}$ Molecular illustrations were made with Mercury, ${ }^{58}$ and geometric analysis with PLATON. ${ }^{59}$ Data are deposited at the Cambridge Crystallographic Data Centre with code numbers CCDC: 2069090-2069106 and 2069108.

## A. DIPHENYL DERIVATIVES.

Table S1. Crystallographic data for dimethylamino-diphenyl-naphthalene derivatives $\mathbf{1 6 - 1 9}$, and 21-22 and salts 26-27.

|  | $\mathbf{1 6}$ |  | $\mathbf{1 7}$ | $\mathbf{1 8}^{\mathbf{s 1}}$ |
| :--- | :--- | :--- | :--- | :--- |


|  | 21 | 22 | 26 | 27 |
| :---: | :---: | :---: | :---: | :---: |
| Formula | $\begin{aligned} & \mathrm{C}_{31} \mathrm{H}_{27} \mathrm{NO}_{2} . \\ & \mathrm{CHCl}_{3} \end{aligned}$ | $\mathrm{C}_{52} \mathrm{H}_{46} \mathrm{~N}_{4}$ | $\begin{aligned} & \mathrm{C}_{25} \mathrm{H}_{22} \mathrm{NO} . \\ & \mathrm{Cl}^{2} . \mathrm{CH}_{2} \mathrm{Cl}_{2} \end{aligned}$ | $\begin{aligned} & \mathrm{C}_{25} \mathrm{H}_{22} \mathrm{NO} . \\ & \mathrm{C}_{3} \mathrm{H}_{3} \mathrm{O}_{4} \end{aligned}$ |
| Formula <br> Weight | 564.90 | 726.93 | 472.81 | 455.49 |
| Crystal System | Monoclinic | Orthorhombic | Orthorhombic | Triclinic |
| Space group | I2/a | Pbca | $P 2.22_{1}$ | $P-1$ |
| $a[\AA]$ | 19.4622(12) | 9.8042(2) | 6.1138(3) | 6.0395(3) |
| $\boldsymbol{b}_{[\boldsymbol{A}}$ ] | 9.9340 (8) | 15.4073(4) | 11.1707(5) | 13.8019(8) |
| $c[\AA]$ | 29.6756(15) | 52.7317(11) | 33.8350(17) | 14.6137(9) |
| $\alpha\left[{ }^{\circ}\right]$ | 90 | 90 | 90 | 67.400(5) |
| $\beta\left[{ }^{\circ}\right]$ | 93.325(5) | 90 | 90 | 86.108(4) |
| $\gamma\left[{ }^{\circ}\right]$ | 90 | 90 | 90 | 81.742(4) |
| $V\left[\AA^{3}\right]$ | 5727.7(6) | 7965.5(3) | 2310.78(19) | 1112.84(12) |
| $\boldsymbol{Z}$ | 8 | 8 | 4 | 2 |
| $\rho\left[\mathrm{g} \mathrm{cm}^{-3}\right]$ | 1.310 | 1.212 | 1.359 | 1.359 |
| $T[K]$ | 150 | 150 | 150 | 150 |
| $\lambda(\AA)$ | 0.71073 | 1.54184 | 0.71073 | 0.71073 |
| $\mu\left(\mathrm{mm}^{-1}\right)$ | 0.350 | 0.542 | 0.415 | 0.093 |
| unique refl. | 5880 | 7645 | 5702 | 6361 |
| Refl, $\mathrm{I}>\mathbf{2} \boldsymbol{\sigma} \mathrm{I}$ | 2821 | 5584 | 4042 | 4061 |
| $R_{1}$ | 0.0805 | 0.0524 | 0.0594 | 0.0644 |
| $w R_{2}$ | 0.1734 | 0.1423 | 0.1217 | 0.1563 |
| $\Delta \rho(\mathrm{r})\left[\mathrm{e} \AA^{-3}\right]$ | 0.40/-0.37 | 0.21/-0.19 | 0.49/-0.64 | 0.44/-0.29 |
| Crystallisation solvent | $\mathrm{CHCl}_{3}$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ <br> $/ n$-hexane | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | $\mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ |

Table S2. Phenyl Group Orientations for peri-Diphenylnaphthalene Derivatives.


|  | $\alpha^{\prime} /{ }^{\circ}$ | $\beta^{\prime \prime} /{ }^{\circ}$ | $\delta^{\prime \prime} /{ }^{\circ}$ | $\varepsilon^{\prime \prime}{ }^{\circ}$ | Phenyl/ Naphthyl planes/ ${ }^{0}$ | Phenyl/ Phenyl plane ${ }^{\circ}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Parent ${ }^{\text {S10 }}$ | 114.8(2) | 125.8(2) | 124.4(2) | 115.8(2) | $\begin{aligned} & \hline 66.48(10) / \\ & 67.10(10) \end{aligned}$ | 20.49(12) |
| 16 | 116.12(13) | 125.11(13) | 125.80(14) | 115.16(14) | $\begin{aligned} & 59.36(4) / \\ & 57.01(4) \end{aligned}$ | 27.08(7) |
| 17 | 116.5(2) | 124.6(2) | 125.1(2) | 116.0(2) | $\begin{aligned} & 59.23(7) \\ & 56.71(7) \end{aligned}$ | 20.69(10) |
| 18 | 115.44(12) | 125.88(12) | 124.50(12) | 116.02(12) | $\begin{aligned} & 62.22(3) \\ & 63.21(3) \end{aligned}$ | 21.10(6) |
| 19 | 116.75(9) | 123.75(9) | 123.10(9) | 118.05(9) | $\begin{aligned} & 60.12(4) \\ & 62.98(5) \end{aligned}$ | 16.97(6) |
| 21 | 116.2(3) | 124.9(3) | 123.6(3) | 117.9(3) | $\begin{aligned} & 57.51(14) \\ & 53.06(13) \end{aligned}$ | 26.26(18) |
| 22 | 115.75(18) | 125.08(17) | 124.91(17) | 115.84(19) | $\begin{aligned} & 57.08(8) \\ & 48.78(8) \end{aligned}$ | 30.50(10) |
|  | 116.89(17) | 123.87(17) | 124.85(17) | 115.96(18) | $\begin{aligned} & 56.71(8) \\ & 56.66(9) \end{aligned}$ | 34.21(11) |
| 26 | 115.4(3) | 125.6(3) | 124.6(4) | 116.3(4) | $\begin{aligned} & 61.90(18) \\ & 53.69(16) \end{aligned}$ | 25.1(2) |
| 27 | 116.98(16) | 123.93(16) | 124.96(16) | 115.84(16) | $\begin{aligned} & 55.78(8) \\ & 61.87(9) \\ & \hline \end{aligned}$ | 22.78(11) |

Parent $=1,8$-diphenylnaphthalene, room temperature measurement.

## Crystal packing diagrams of 16, 17 and 19.

There is a common packing motif with pairs of molecules oriented so that the two phenyls lie opposite a naphthalene.





Figure S1. Crystal packing in 16, and an isolated pair of molecules of $\mathbf{1 6}$ (top), crystal packing of $\mathbf{1 7}$ (middle) and crystal packing of $\mathbf{1 9}$ (with some molecules along the $b$ axis omitted for clarity) (bottom).

## B. ACENAPTHENE DERIVATIVES.

Table S3. Crystallographic data for the acenaphthene derivatives 28-30 and 34-36.

|  | 28 | $29^{\text {S1 }}$ | 30 |
| :---: | :---: | :---: | :---: |
| Formula | $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{NO}$ | $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{~N}_{3}$ | $\mathrm{C}_{19} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{2}$ |
| Formula weight | 225.28 | 273.33 | 306.35 |
| Crystal system | Monoclinic | Triclinic | Triclinic |
| Space group | $P 2{ }_{1} / \mathrm{c}$ | $P-1$ | $P-1$ |
| $\boldsymbol{a}$ [£] | 9.1836(6) | 9.8149(4) | 8.7457(5) |
| $\boldsymbol{b}[\mathbf{\AA}]$ | $9.4806(5)$ | 9.9853(4) | 9.3638(7) |
|  | 13.2584(8) | 15.1587(6) | $11.2739(7)$ |
| $\alpha\left[{ }^{\circ}\right]$ | 90 | 84.234(3) | 108.887(6) |
| $\boldsymbol{\beta}\left[{ }^{\circ}\right]$ | 99.767(7) | 78.977(4) | 91.171(5) |
| $\gamma\left[{ }^{\circ}\right]$ | 90 | 80.518(3) | 113.671(6) |
| $V\left[\AA^{3}\right]$ | 1137.62(12) | 1434.64(10) | 787.91(10) |
| $Z$ | 4 | 4 | 2 |
| $\rho\left[\mathrm{g} \mathrm{cm}^{-3}\right]$ | 1.315 | 1.265 | 1.291 |
| $\boldsymbol{T}[K]$ | 150 | 150 | 150 |
| $\lambda(\AA)$ | 0.71073 | 0.71073 | 0.71073 |
| $\mu\left(\mathrm{mm}^{-1}\right)$ | 0.082 | 0.077 | 0.085 |
| unique refl. | 2341 | 5911 | 3915 |
| Refl, $\mathrm{I}>\mathbf{2 \sigma I}$ | 1961 | 3884 | 2990 |
| $R_{1}$ | 0.0487 | 0.0575 | 0.0506 |
| $w R_{2}$ | 0.1119 | 0.1157 | 0.1360 |
| $\Delta \rho(\mathrm{r})\left[\mathrm{e} \AA^{-3}\right]$ | 0.21/-0.21 | 0.23/-0.25 | 0.30/-0.21 |
| Crystallisation solvent | $\mathrm{CDCl}_{3}$ | $\mathrm{CH}_{3} \mathrm{OH}$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2} /$ <br> $n$-hexane |


|  | 34 | 35 | 36 |
| :---: | :---: | :---: | :---: |
| Formula | $\mathrm{C}_{21} \mathrm{H}_{21} \mathrm{NO}_{4}$ | $\mathrm{C}_{21} \mathrm{H}_{21} \mathrm{NO}_{2}$ | $\mathrm{C}_{20} \mathrm{H}_{19} \mathrm{NO}_{2}$ |
| Formula | 351.39 | 319.39 | 305.36 |
| weight |  |  |  |
| Crystal | Monoclinic | Monoclinic | Monoclinic |
| system |  |  |  |
| Space group | $P 2_{1} / n$ | $P 2{ }_{1} / \mathrm{c}$ | $P 2_{1} / c$ |
| $\boldsymbol{a}[\AA]$ | 10.4257(3) | 15.2740(6) | 10.7625(5) |
| $b[\AA]$ | 8.8243(3) | 10.3218(4) | 32.3176(10) |
| $c[\AA]$ | 18.7805(7) | 10.2175(3) | 8.6130(3) |
| $\alpha\left[{ }^{\circ}\right]$ | 90 | 90 | 90 |
| $\boldsymbol{\beta}\left[^{\circ}\right]$ | 101.546(3) | 93.356(3) | 98.945(4) |
| $\gamma\left[{ }^{\circ}\right]$ | 90 | 90 | 90 |
| $V\left[\AA^{\mathbf{3}}\right]$ | 1692.83(10) | 1608.08(10) | 2959.3(2) |
| $Z$ | 4 | 4 | 8 |
| $\rho\left[\mathrm{g} \mathrm{cm}^{-3}\right]$ | 1.379 | 1.319 | 1.371 |
| $\boldsymbol{T}[\mathrm{K}]$ | 150 | 150 | 150 |
| $\lambda(\AA)$ | 0.71073 | 0.71073 | 0.71073 |
| $\mu\left(\mathrm{mm}^{-1}\right)$ | 0.096 | 0.084 | 0.088 |
| unique refl. | 3489 | 3307 | 6829 |
| Refl, $\mathrm{I}>\mathbf{2 \sigma I}$ | 2736 | 2578 | 4795 |
| $\mathrm{R}_{1}$ | 0.0552 | 0.0514 | 0.1040 |
| $w R_{2}$ | 0.1132 | 0.1111 | 0.2072 |
| $\Delta \rho(\mathrm{r})\left[\mathrm{e} \AA^{-3}\right]$ | 0.22/-0.24 | 0.18/-0.26 | 0.53/-0.29 |
| Crystallisation | $\mathrm{CH}_{2} \mathrm{Cl}_{2} /$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2} /$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2} /$ |
| Solvent | n-hexane | $n$-hexane | $n$-hexane |

Table S4. Selected geometric details for the azepine ring for spiro acenaphtho-azepines 34-36.

$34 \mathrm{X}, \mathrm{X}=(\mathrm{C}(=\mathrm{O}) \mathrm{O})_{2} \mathrm{CMe}_{2}$
$35 \mathrm{X}, \mathrm{X}=\left(\mathrm{C}(=\mathrm{O}) \mathrm{CH}_{2}\right)_{2} \mathrm{CH}_{2}$
$36 \mathrm{X}, \mathrm{X}=\left(\mathrm{C}(=\mathrm{O}) \mathrm{CH}_{2}\right)_{2}$

|  | $\mathbf{a} / \AA$ | $\mathbf{b} / \AA$ | $\mathbf{c} / \AA$ | $\boldsymbol{\Sigma}^{\mathbf{a} / \mathbf{0}}$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathbf{3 4}$ | $1.457(2)$ | $1.549(2)$ | $1.560(2)$ | $349.5(3)$ |
| $\mathbf{3 5}$ | $1.470(2)$ | $1.525(2)$ | $1.562(2)$ | $343.6(3)$ |
| $\mathbf{3 6}$ | $1.453(4)$ | $1.551(4)$ | $1.529(4)$ | $346.9(9)$ |
|  | $1.447(5)$ | $1.551(5)$ | $1.520(5)$ | $348.3(9)$ |

$\Sigma^{\mathbf{a}}=$ Sum of bond angles at nitrogen.

|  | $\boldsymbol{\eta} /{ }^{\mathbf{0}}$ | $\boldsymbol{\kappa} /{ }^{\mathbf{0}}$ | $\boldsymbol{\lambda} /{ }^{\circ}$ | $\boldsymbol{\mu} /{ }^{\mathbf{o}}$ |
| :--- | :---: | :---: | :---: | :---: |
| $\mathbf{3 4}$ | $115.56(14)$ | $113.02(14)$ | $109.01(14)$ | $116.19(14)$ |
| $\mathbf{3 5}$ | $113.57(13)$ | $114.02(13)$ | $109.23(13)$ | $115.66(13)$ |
| $\mathbf{3 6}$ | $114.4(3)$ | $113.3(3)$ | $109.4(3)$ | $117.0(3)$. |
|  | $113.8(3)$ | $113.3(3)$ | $110.0(3)$ | $117.2(3)$ |

Table S5: Selected angles around the acenaphthene ring, and deviations of selected atoms in the azepine ring from the acenaphthene plane for the spiro acenaphtho-azepines 34-36.


$$
\begin{aligned}
& 34 \mathrm{X}, \mathrm{X}=(\mathrm{C}(=\mathrm{O}) \mathrm{O})_{2} \mathrm{CMe}_{2} \\
& 35 \mathrm{X}, \mathrm{X}=\left(\mathrm{C}(=\mathrm{O}) \mathrm{CH}_{2}\right)_{2} \mathrm{CH}_{2} \\
& 36 \mathrm{X}, \mathrm{X}=\left(\mathrm{C}(=\mathrm{O}) \mathrm{CH}_{2}\right)_{2}
\end{aligned}
$$

|  | $\boldsymbol{\alpha} / \mathbf{0}$ | $\boldsymbol{\beta} / \mathbf{0}$ | $\boldsymbol{\varphi} /{ }^{\circ}$ | $\boldsymbol{\delta} / \mathbf{0}$ | $\boldsymbol{\varepsilon} /{ }^{\mathbf{o}}$ | $\boldsymbol{\psi} /{ }^{\mathbf{o}}$ |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{3}$ | $122.86(1$ | $118.57(1$ | $127.35(1$ | $120.79(1$ | $119.73(1$ | $112.07(1$ |
| $\mathbf{4}$ | $6)$ | $5)$ | $7)$ | $6)$ | $6)$ | $7)$ |
| $\mathbf{3}$ | $122.76(1$ | $118.53(1$ | $127.14(1$ | $121.27(1$ | $119.75(1$ | $111.81(1$ |
| $\mathbf{5}$ | $5)$ | $4)$ | $5)$ | $5)$ | $5)$ | $5)$ |
| $\mathbf{3}$ | $123.2(3)$ | $118.3(3)$ | $127.3(3)$ | $121.8(3)$ | $119.6(3)$ | $111.8(3)$ |
| $\mathbf{6}$ |  |  |  |  |  |  |
|  | $121.6(2)$ | $120.2(3)$ | $126.1(3)$ | $121.9(3)$ | $118.8(4)$ | $111.6(4)$ |

Deviations from the acenaphthene plane $/ \AA$.

|  | $\mathbf{N}$ | $\mathbf{C}_{\mathbf{A}}$ | $\mathbf{C}_{\boldsymbol{B}}$ | $\mathbf{C}_{\mathbf{C}}$ |
| :--- | ---: | ---: | ---: | ---: |
| $\mathbf{3 4}$ | $0.303(2)$ | $1.319(2)$ | $0.731(3)$ | $-0.383(2)$ |
| $\mathbf{3 5}$ | $0.249(2)$ | $1.354(2)$ | $0.919(2)$ | $-0.246(2)$ |
| $\mathbf{3 6}$ | $0.242(4)$ | $1.254(4)$ | $0.670(4)$ | $-0.389(4)$ |
|  | $0.183(4)$ | $1.163(5)$ | $0.551(5)$ | $-0.450(4)$ |

## C. FLUORENE DERIVATIVES.

Table S6. Crystallographic data for the fluorene derivatives 43-45, 48 and 50.

|  | 43 | 44 | 45 | 48 | 50 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Formula | $\mathrm{C}_{15} \mathrm{H}_{12} \mathrm{I}_{2}$ | $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{NO}$ | $\mathrm{C}_{21} \mathrm{H}_{19} \mathrm{~N}_{3}$ | $\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{2}$ | $\mathrm{C}_{24} \mathrm{H}_{25} \mathrm{NO}_{4}$ |
| Formula | 446.05 | 265.34 | 313.39 | 308.37 | 391.45 |
| weight |  |  |  |  |  |
| Crystal system | Monoclinic | Tetragonal | Monoclinic | Monoclinic | Monoclinic |
| Space group | $P 2{ }_{1} / \mathrm{c}$ | I4, cd | I2/a | $P 2_{1}$ | $P 2{ }_{1} / n$ |
| $\boldsymbol{a}$ [ $\AA$ ] | 8.4109(2) | 24.79560 (10) | 18.4256(4) | 9.7431(3) | 10.9539(3) |
| $\boldsymbol{b}[\mathbf{\AA}]$ | 9.6735(3) | 24.79560 (10) | 11.2921(3) | 6.9459(2) | 9.5493(3) |
| $c[\AA]$ | 17.0655(5) | 9.47440 (10) | 16.4939(4) | 12.1403(4) | 19.3902(6) |
| $\alpha\left[^{\circ}\right]$ | 90 | 90 | 90 | 90 | 90 |
| $\boldsymbol{\beta}\left[^{\circ}{ }^{\text {] }}\right.$ | 90.496(2) | 90 | 99.223(2) | 105.257(4) | 104.652(3) |
| $\gamma\left[{ }^{\circ}\right]$ | 90 | 90 | 90 | 90 | 90 |
| $V\left[\AA^{3}\right]$ | 1388.45(7) | 5825.07(8) | 3387.42(14) | 792.63(4) | 1962.30(11) |
| $\boldsymbol{Z}$ | 4 | 16 | 8 | 2 | 4 |
| $\rho\left[\mathrm{g} \mathrm{cm}^{-3}\right]$ | 2.134 | 1.210 | 1.229 | 1.292 | 1.325 |
| $T[K]$ | 150 | 150 | 150 | 150 | 150 |
| $\lambda(\AA)$ | 0.71073 | 1.54184 | 0.71073 | 0.71073 | 0.71073 |
| $\mu\left(\mathrm{mm}^{-1}\right)$ | 4.506 | 0.579 | 0.074 | 0.085 | 0.090 |
| unique refl. | 2837 | 2152 | 3936 | 4532 | 3165 |
| Refl, $\mathrm{I}>2 \boldsymbol{\sigma} \mathrm{I}$ | 2477 | 2108 | 3251 | 3645 | 2904 |
| $R_{1}(\mathrm{I}>2 \boldsymbol{\sigma})$ | 0.0289 | 0.0314 | 0.0478 | 0.0462 | 0.0510 |
| $w R_{2}$ | 0.0646 | 0.0803 | 0.1188 | 0.0961 | 0.1109 |


| $\Delta \rho(\mathbf{r})\left[\mathbf{e} \AA^{-3}\right]$ | $0.96 /-0.53$ | $0.13 /-0.17$ | $0.29 /-0.23$ | $0.20 /-0.20$ | $0.28 /-0.23$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| Crystallisation <br> solvent | $\mathrm{EtOAc} /$ Petrol | $\mathrm{CH}_{2} \mathrm{Cl}_{2} / n$ - | $\mathrm{CDCl}_{3} /$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | $\left(\mathrm{CH}_{3}\right)_{2} \mathrm{CO}$ |
|  | $(40-60)$ | hexane | $n$-hexane | /n-hexane |  |

## Crystal Structure of 4,5-diiodo-9,9-dimethylfluorene 43.

The diiodo compound $\mathbf{4 3}$ crystallises in space group $P 2_{1} / c$ with four molecules in the unit cell. The molecule is strongly distorted to increase the $\mathrm{I} \cdots \mathrm{I}$ separation to 3.6392 (4) $\AA$ (Fig. S2). This is achieved in three ways: (a) the C-I bonds are displaced apart in the plane of the fluorene, with the two bonds displaced from their symmetrical positions by 5.9 and $7.1^{\circ}$, (b) the iodo groups are displaced out of their respective phenyl ring planes by 0.614 and $0.598 \AA$ and (c) the fluorene is twisted about the bond connecting the phenyl rings, by $21.4(8)^{\circ}$. The C-I bonds are 2.106(4) and 2.114(4) $\AA$ long. Similar distortions is seen in the 4,5-diiodophenanthrene $\mathbf{S 1},{ }^{\text {, } 10} 1,12$-diiodo-triphenylene $\mathbf{S 2},{ }^{\text {S11 }}$ and a polybenzenoid system $\mathbf{S 3}{ }^{\text {S12 }}: \mathrm{I} \cdots \mathrm{I}$ separations S1: $3.602 / 3.610 \AA, \mathbf{S 2}: 3.679 / 3.687 \AA$, S3: $3.636 \AA$, but with a larger twist about the bond between the relevant rings: S1: $33.7 / 34.6^{\circ}, \mathbf{S 2}: 33.5^{\circ}, \mathbf{S 3}: 40.3^{\circ}$.

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S1

S2




Figure S2: Two views of the molecular structure of $\mathbf{4 3}$ (top) and its crystal packing arrangement (below).

Table S7. Selected molecular geometry for fluoreno-azocines $\mathbf{4 8}$ and $\mathbf{5 0}$.

$48 \mathrm{X}=\mathrm{Y}=-.(\mathrm{C}=\mathrm{O}) \mathrm{O}_{2} \mathrm{CMe}_{2}$
$50 \mathrm{X}=\mathrm{NO}_{2}, Y=\mathrm{H}$.

|  | $\mathbf{a} / \AA$ | $\mathbf{b} / \AA$ | $\mathbf{c} / \AA$ | $\boldsymbol{\Sigma}^{\mathbf{a} / \mathbf{0}}$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathbf{4 8}$ | $1.454(2)$ | $1.565(2)$ | $1.561(2)$ | $350.6(2)$ |
| $\mathbf{5 0}$ | $1.460(4)$ | $1.525(4)$ | $1.506(4)$ | $348.8(4)$ |

$\Sigma^{\mathbf{a}}=$ Sum of bond angles at nitrogen.

|  | $\boldsymbol{\eta} / \mathbf{0}$ | ${ }^{\circ} /{ }^{\mathbf{o}}$ | $\boldsymbol{\lambda} /{ }^{\circ}$ | $\boldsymbol{\mu} /{ }^{\mathbf{o}}$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathbf{4 8}$ | $118.26(13)$ | $113.75(13)$ | $108.14(12)$ | $116.44(12)$ |
| $\mathbf{5 0}$ | $116.1(2)$ | $116.1(2)$ | $112.0(2)$ | $112.1(2)$ |

Table S8. Selected angles around the fluorene ring, and deviations of selected atoms in the azocine ring from the fluorene plane for the spiro fluoreno-azepines $\mathbf{4 8}$ and $\mathbf{5 0}$.


|  | $\boldsymbol{\alpha} / \mathbf{0}$ | $\boldsymbol{\beta} / \mathbf{0}$ | $\boldsymbol{\gamma} / \mathbf{0}$ | $\boldsymbol{\gamma}^{\boldsymbol{\prime} / \mathbf{0}}$ | $\boldsymbol{\delta} / \mathbf{0}$ | $\boldsymbol{\varepsilon} / \mathbf{0}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
|  |  |  |  |  |  |  |
| $\mathbf{4 8}$ | $120.14(14)$ | $122.97(14)$ | $133.50(14)$ | $133.35(14)$ | $124.02(14)$ | $117.80(14)$ |
| $\mathbf{5 0}$ | $121.5(2)$ | $121.0(2)$ | $133.4(2)$ | $134.0(2)$ | $124.2(3)$ | $117.6(2)$ |

Deviations from the fluorene plane / $\AA$

|  | $\mathbf{N}$ | $\mathbf{C}_{\mathbf{A}}$ | $\mathbf{C}_{\boldsymbol{B}}$ | $\mathbf{C}_{\mathbf{C}}$ |
| :--- | :--- | ---: | :--- | ---: |
| $\mathbf{4 8}$ | $0.389(3)$ | $1.458(4)$ | $0.953(4)$ | $-0.306(4)$ |
| $\mathbf{5 0}$ | $0.4379(19)$ | $1.428(2)$ | $0.829(2)$ | $-0.391(2)$ |

## References.

S1. J. C. Bristow, I. Naftalin, S. V. A. Cliff, S. Yang, M. Carravetta, R. Stern, I. Heinmaa and J. D. Wallis, CrystEngComm, 2020, 22, 6783-6795.

S2. N. Lardiés, I. Romeo, E. Cerrada, M. Laguna and P. J. Skabara, Dalton Trans., 2007, 5329-5338.

S3. V. D. B. Bonifácio, J. Morgado and U. Scherf, Synlett, 2010, 1333-1336.
S4. CrysAlisPro, Agilent Technologies, Version 1.171.35.15 (release 03-08-2011 CrysAlis171.NET).

S5. G. M. Sheldrick, Acta Crystallogr. Sect. A, 2008, 64, 112-122; Acta Crystallogr. Sect. C, 2015, 71, 3-8.

S6. L. J. Barbour, "X-Seed - A software tool for supramolecular crystallography" $J$. Supramol. Chem. 2001, 1, 189-191.

S7. OLEX ${ }^{2}$ O.V. Dolomanov, L.J. Bourhis, R.J. Gildea, J.A.K. Howard and H. Pushmann, J. Appl. Cryst., 2009, 42, 339-341.
S8. C.F. Macrae, P.R. Edgington, P. McCabe, E. Pidcock, G.P. Shields, R. Taylor, M. Towler and J. van de Streek, J. Appl. Crystallogr. 2006, 39, 453-457.

S9. PLATON, A Multipurpose Crystallographic Took, A. L. Spek, University of Utrecht, The Netherlands, https://www.platonsoft.nl/platon/.
S10. H. Bock, M. Sievert and Z. Havlas, Chem. Eur. J., 1998, 4, 677-685.
S11. A. J. Ashe, III, J. W. Kampf and P. M. Savia, J. Org. Chem., 1990, 55, 5558-5559.
S12. Y. Shi, H. Qian, Y. Li, W. Yue and Z. Wang, Org. Lett., 2008, 10, 2337-2340.

