

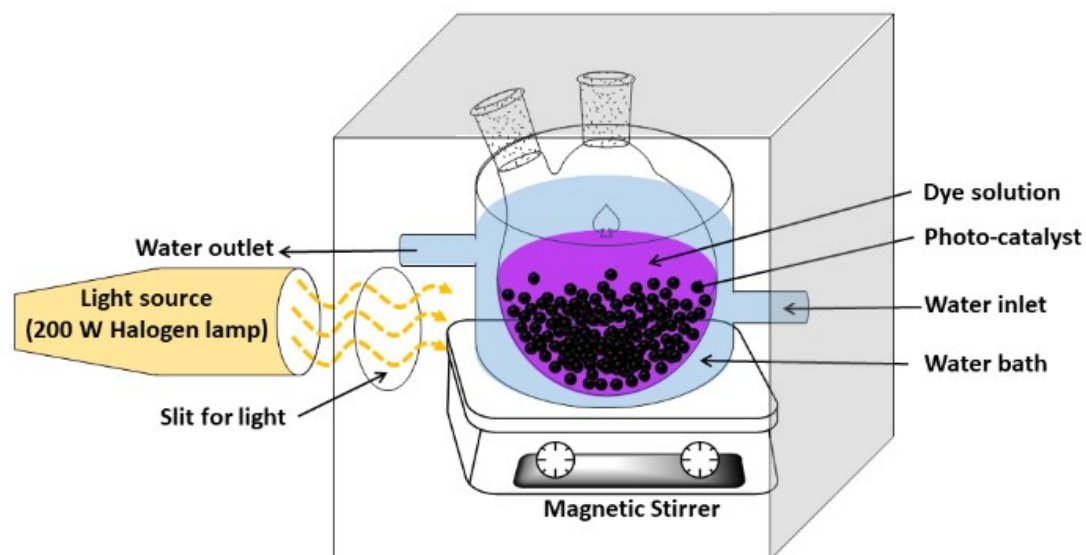
## **Electronic Supporting Information**

### **Biomaterialized Vaterite derived Efficient solid-state synthesis of pure $\text{CaMnO}_3$ Perovskite as an effective photocatalyst**

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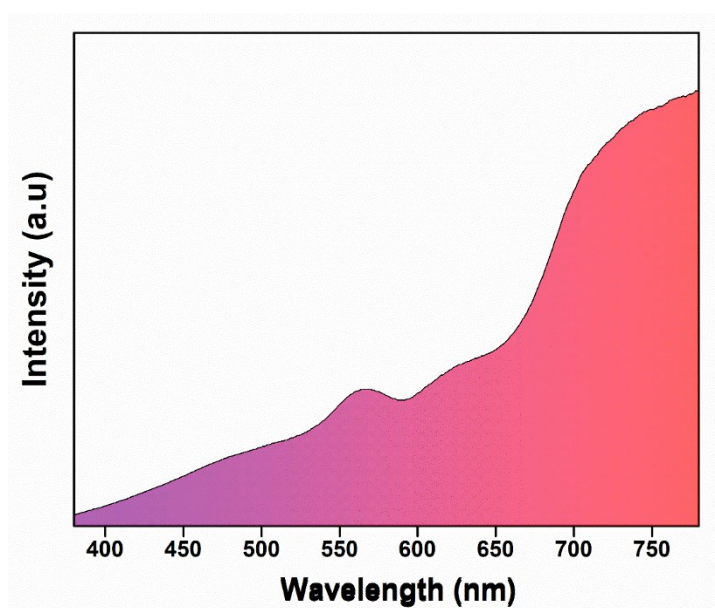
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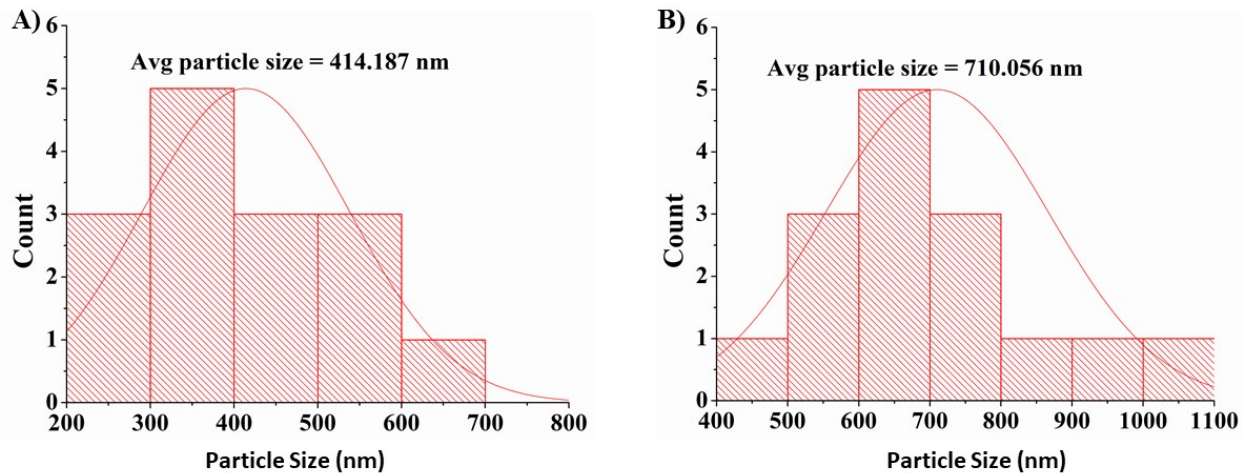


**Scheme S1:** Illustration of the photocatalytic device set up.

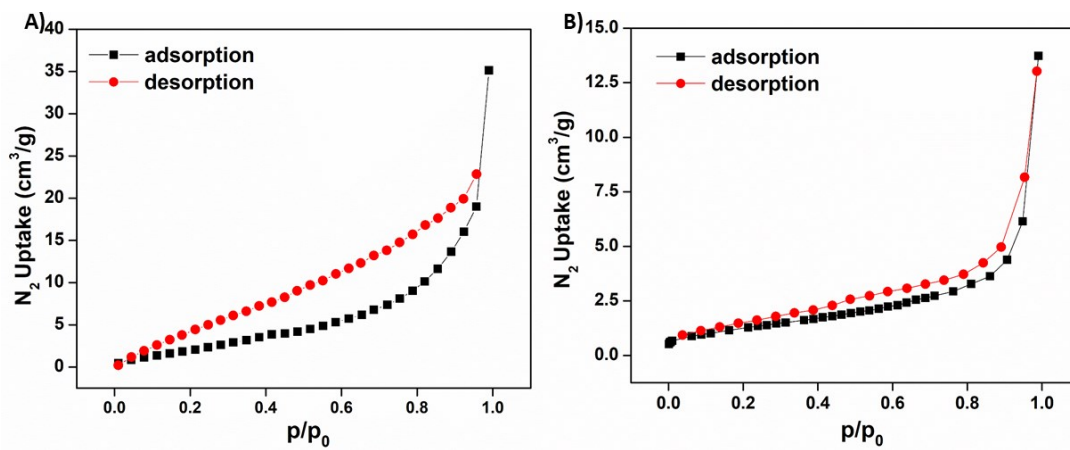
The wavelength spectrum of the light source (200 W Halogen lamp) was obtained from Spectroradiometer CS-200 (Konica Minolta). The experiment was performed by keeping the light source at a distance of 2 m from the spectroradiometer. The intensity of the light was found out to be 11507.13 cd/m<sup>2</sup> and the wavelength spectrum was obtained in the range of 380-780 nm, Fig. S1.



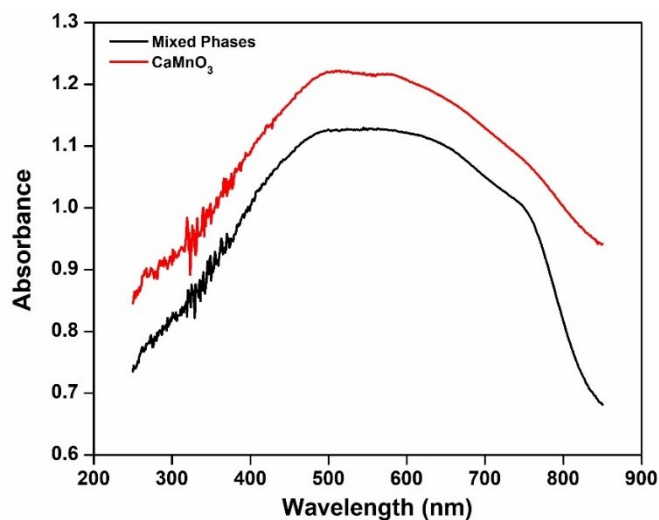
**Figure S1:** Wavelength spectrum of the 200 W Halogen lamp.



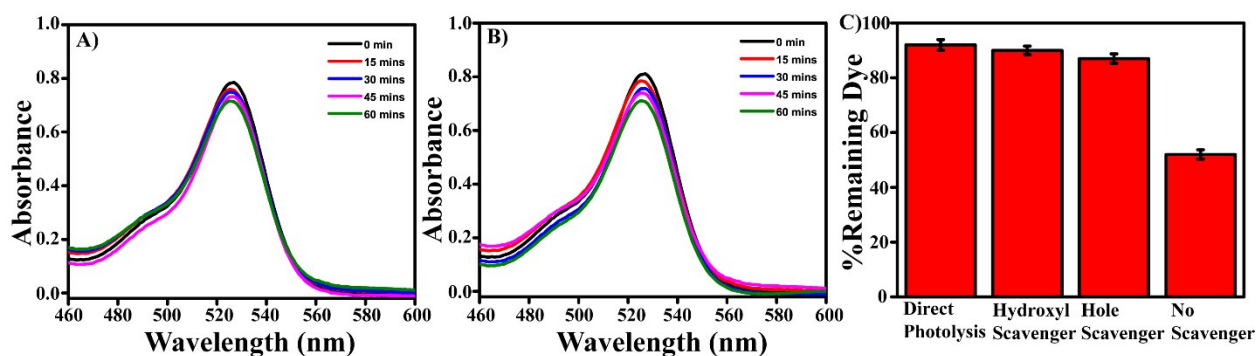
**Figure S2:** Average particle size of (A) pure CaMnO<sub>3</sub> particles; (B) Mixed phase CMOs.



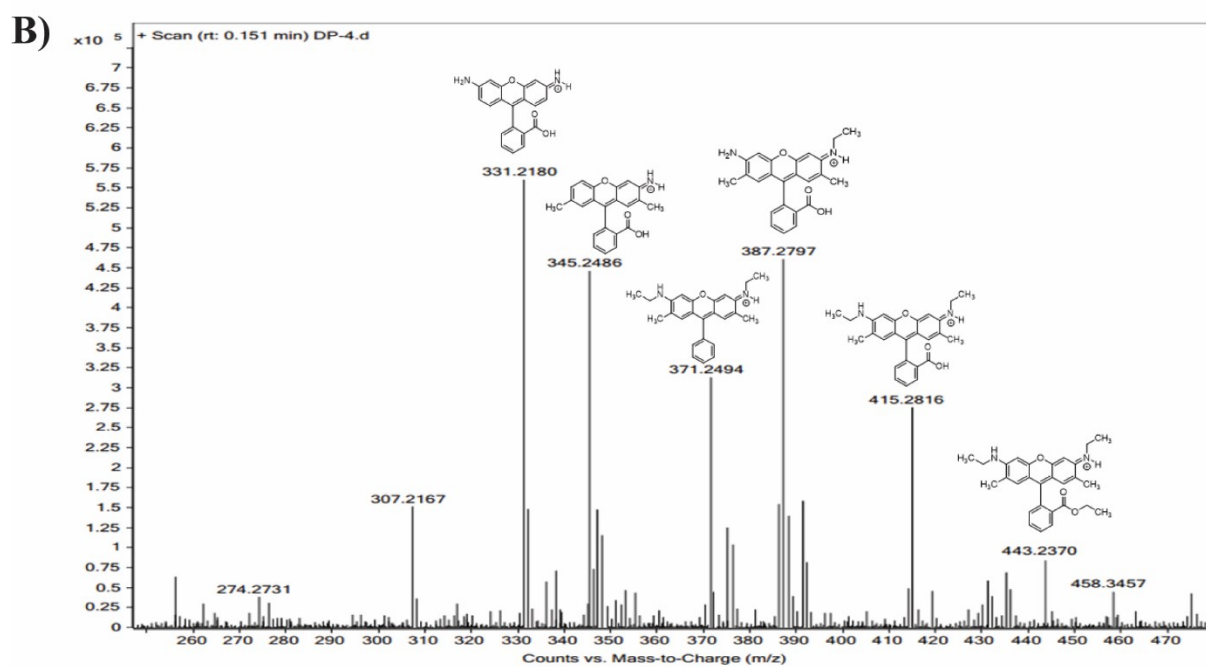
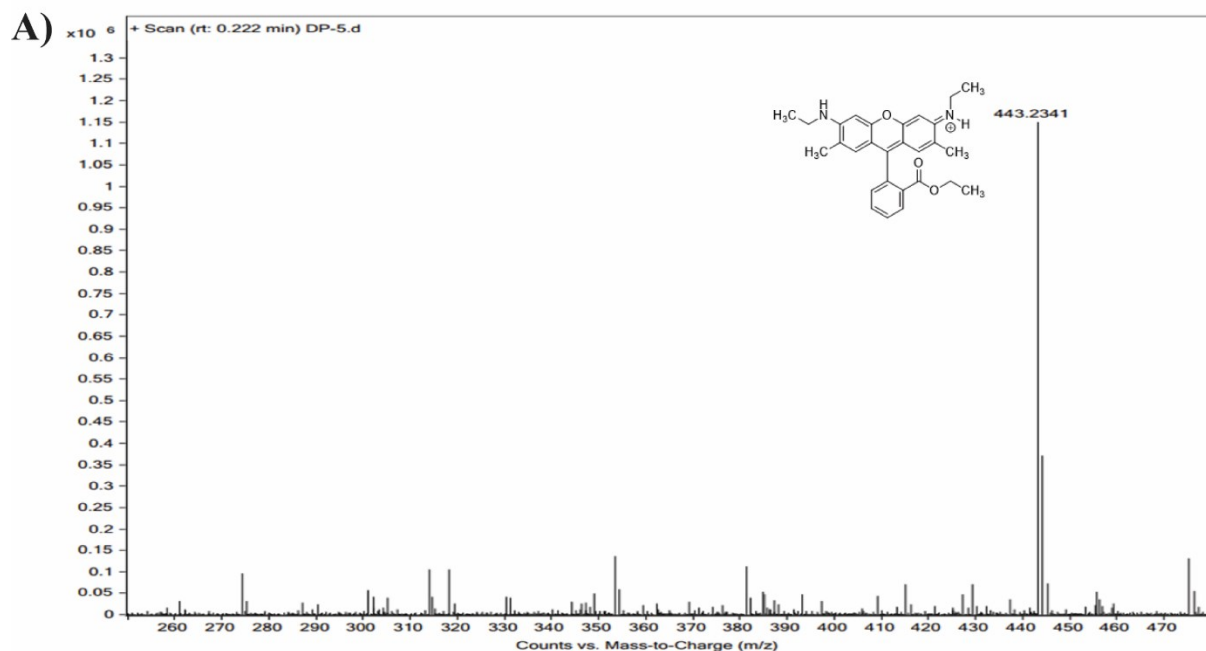
**Figure S3:** Nitrogen adsorption-desorption BET isotherm (A) CaMnO<sub>3</sub>; (B) Mixed Phase.



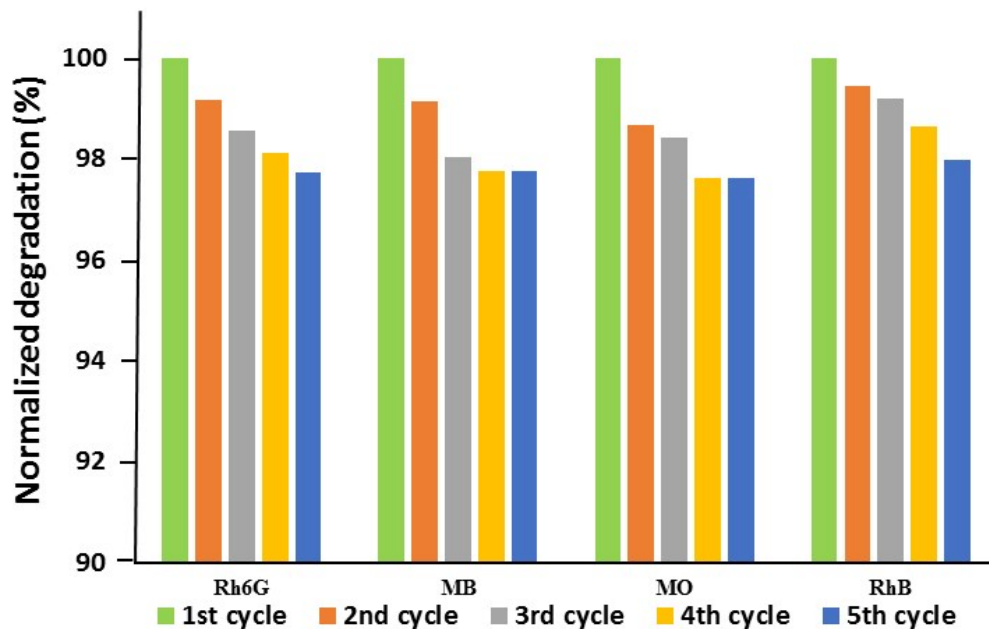
**Figure S4:** UV/visible diffuse reflectance spectra of  $\text{CaMnO}_3$  and the mixed phase CMOs.



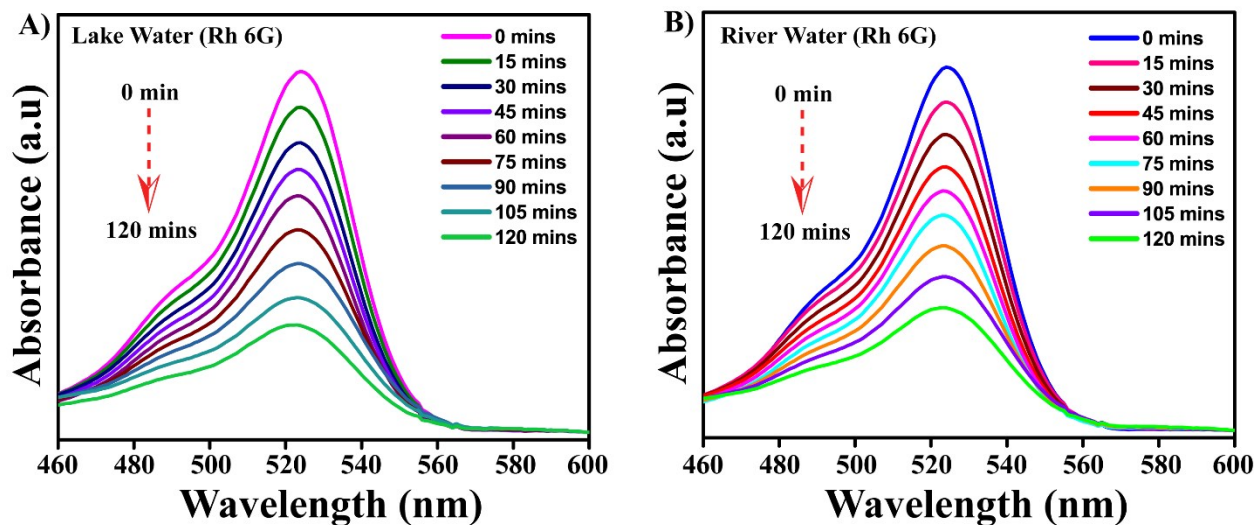
**Figure S5:** UV/Vis absorption spectra of the photolytic degradation of Rhodamine 6G catalyzed by  $\text{CaMnO}_3$  in the presence (A)  $\cdot\text{OH}$  scavenger; (B)  $\text{h}^+$  scavenger; and (C) Percentage degradation of Rhodamine 6G in the presence and absence of the scavengers.



**Figure S6:** Mass spectrum of Rh 6G (A) before light irradiation; (B) after 2 hours of light irradiation.



**Figure S7:** Normalized dye-degradation percentages of the different dyes upon recyclability and reusability.



**Figure S8:** UV/Vis absorption spectra of the photolytic degradation of Rhodamine 6G catalyzed by  $\text{CaMnO}_3$  performed in (A) Lake water and (B) River water.

**Table S1:** Photocatalytic activity of calcium manganese oxides towards the degradation of Rhodamine 6G

<b>Reference</b>	<b>Photocatalyst</b>	<b>Time (mins)</b>	<b>Percentage of Degradation (%)</b>
<i>J. Phys. Chem., C</i> 2014, <b>118</b> , <b>41</b> , 24127–24135	CaMn <sub>3</sub> O <sub>6</sub>	240	82
<i>J. Solid State Chem.</i> , 2020, <b>288</b> , 121390	CaMnO <sub>3</sub>	180	62
	CaMn <sub>2</sub> O <sub>4</sub>	180	61
	Ca <sub>2</sub> Mn <sub>3</sub> O <sub>8</sub>	180	32
<b>This work</b>	<b>CaMnO<sub>3</sub></b>	<b>120</b>	<b>72</b>