Supplementary Information

## A Solid-liquid Two-phase Precipitation Method for Growth of Fullerene (C<sub>60</sub>) Nanowires

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**Fig. S1.** Streak camera images of the time-resolved photoluminescence (TRPL) of (a) C60 nanowires and (b) pristine C60 powder.



**Fig. S2.** The growth of  $C_{60}$  nanowires in m-xylene/IPA system for 24 h with a fixed  $C_{60}$  solution-to-IPA volume ratio of 1:2 and  $C_{60}$  concentration of 1 mg/mL, 2 mg/mL, 3 mg/mL and ~5 mg/mL, respectively, through the SLTPP method (top) and the LLIP method (bottom).



**Fig. S3.** The growth of  $C_{60}$  nanowires in m-xylene/IPA system for a week with a fixed  $C_{60}$  solution-to-IPA volume ratio of 1:2 and  $C_{60}$  concentration of 1 mg/mL, 2 mg/mL, 3 mg/mL and ~5 mg/mL, respectively, through the SLTPP method (top) and the LLIP method (bottom).



**Fig. S4.** The growth of  $C_{60}$  nanowires in m-xylene/IPA system for 24 h with a fixed  $C_{60}$  concentration of 2 mg/mL and  $C_{60}$  solution-to-IPA volume ratio of 1:6, 1:4, 1:2 1:1, and 2:1, respectively, through the SLTPP method (top) and the LLIP method (bottom).



**Fig. S5.** The growth of  $C_{60}$  nanowires in m-xylene/IPA system for a week with a fixed  $C_{60}$  concentration of 2 mg/mL and  $C_{60}$  solution-to-IPA volume ratio of 1:6, 1:4, 1:2 1:1, and 2:1, respectively, through the SLTPP method (top) and the LLIP method (bottom).

**Table S1.**  $C_{60}$  nanowire samples prepared by the SLTPP and the LLIP methods for comparison using  $C_{60}$ /m-xylene/IPA systems with different  $C_{60}$  concentrations in m-xylene. Yellow precipitates correspond to  $C_{60}$  nanowires while the dark precipitates correspond to needle-like larger  $C_{60}$  crystals.

SAMPLE	SYNTHESI	CONCENTRATIO	PRESENCE OF C <sub>60</sub> AFTER 2		PRESENCE OF C <sub>60</sub> AFTER 24	
ID	S METHOD	N OF C <sub>60</sub> (mg/ml)	HOUR	RS	HOURS (VISUAL	
			(VISUAL APPEARANCE)		APPEARANCE)	
			COLOUR &	COLOUR	COLOUR &	COLOUR
			AMOUNT OF	OF	AMOUNT OF	OF
			PRECIPITATES	SOLUTION	PRECIPITATES	SOLUTION
S1c	SLTPP	1	Yellow	Purple	Yellow/Abundant	Colourless
S2c	SLTPP	2	Yellow	Purple	Yellow/Abundant	Colourless
S3c	SLTPP	3	Yellow	Purple	Yellow/Abundant	Colourless
S4c	SLTPP	~5.2	Yellow/Abundant	Nearly	Yellow/Abundant	Colourless
		(Saturated)		colourless		
L1c	LLIP	1	Brown-	Purple	Yellow/Very few	Purple
			black/Very few		Black/A few	
L2c	LLIP	2	Brown-	Purple	Yellow/Very few	Purple
			black/Very few		Black/A few	
L3c	LLIP	3	Brown-black	Purple	Yellow and black	Nearly
					/Abundant	colourless
L4c	LLIP	~5.2	Yellow and black	Nearly	Yellow and black	Colourless
		(Saturated)	/Abundant	colourless	/Abundant	

**Table S2.**  $C_{60}$  nanowire samples prepared through both the SLTPP and the LLIP methods using  $C_{60}$ /mxylene/IPA system with varied solvent volume ratios. Yellow precipitates correspond to  $C_{60}$  nanowires while the dark precipitates correspond to needle-like larger  $C_{60}$  crystals.

SAMPLE	SYNTHESI	M-XYLENE	PRESENCE OF C <sub>60</sub> AFTER 2 HOURS		PRESENCE OF C <sub>60</sub> AFTER 24 HOURS (VISUAL	
ID	S METHOD	TO IPA				
		VOLUME	(VISUAL APPEARANCE)		APPEARANCE)	
		RATIO	COLOUR &	COLOUR OF	COLOUR &	COLOUR OF
			AMOUNT OF	SOLUTION	AMOUNT OF	SOLUTION
			PRECIPITATES		PRECIPITATES	
S1r	SLTPP	1:6	Yellow/Abundant	Colourless	Yellow/Abundant	Colourless
S2r	SLTPP	1:4	Yellow	Purple	Yellow/Abundant	Colourless
S3r	SLTPP	1:2	Yellow	Purple	Yellow/Abundant	Nearly
						colourless
S4r	SLTPP	1:1	Yellow	Purple	Yellow/Abundant	Light purple
S5r	SLTPP	2:1	Yellow-brown	Purple	Yellow-brown	Light purple
					/Abundant	
L1r	LLIP	1:6	Yellow	Purple	Yellow/Abundant	Colourless
L2r	LLIP	1:4	Brown	Purple	Brown/Abundant	Colourless
L3r	LLIP	1:2	Dark brown	Purple	Dark brown	Purple
L4r	LLIP	1:1	Dark brown	Dark purple	Dark brown	Dark purple
L5r	LLIP	2:1	Dark brown	Dark purple	Dark brown	Dark purple