

Supporting Information

Facile formation of Fe-doped NiCoP hollow nanocage as bifunctional electrocatalysts for overall water splitting

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Table S1. HER performances comparison of recently reported representative electrocatalysts in alkaline medium (1.0 M KOH, at 10 mA cm⁻²)

catalyst	Overpotential (mV)	Reference
NiCoFeP	147	This work
MoS ₂ -NiS ₂ /NGF	172	1
Co/ β -Mo ₂ C@NCNTs	170	2
EBP@NG	190	3
p-WCx NWs/CC	122	4
N-Ni ₃ S ₂ /NF	110	5
Ni _{0.9} Fe _{0.1} PS ₃ @MXene	196	6
amorphous CoP/NF	143	7
MoS ₂ /NiS	244	8
Co(OH) ₂ @NCNTs@NF	170	9
Mo-doped NiCoP	76	10
Ni ₃ S ₂ nanosheet arrays/NF	223	11
Fe-Ni ₂ P/MoS _x /NF	112	12
Co@CNTs@DSCNCs	214	13
Co(OH) ₂ @P-NiCo-LDH	226	14
CFC-CNT-CoO _x /CoP	108	15

Table S2. OER performances comparison of recently reported representative electrocatalysts in alkaline medium (1.0 M KOH, at 10 mA cm⁻²)

catalyst	Overpotential (mV)	Reference
NiCoFeP	235	This work
Ni ₃ N-NiMoN	277	16
FeCo-FeCoP@C@NCCs	280	17
amorphous CoP/NF	284	7
CP/CTS/Co-S	190	18
Ni ₃ S ₂ @NGCLs/NF	271	19
MoS ₂ -NiS ₂ /NGF	370	1
Ni ₃ S ₂ @MoS ₂ /FeOOH	260	20
Co ₆ W ₆ C@NC	286	21
CoS ₂ HNSs	290	22
S-CoWP@(S,N)-C	280	23
FeNiB/FeNi foam-700	272	24
Ni ₆ Fe ₂ W-LDH@CC	264	25
CoMoN _x -500 NSAs/NF	231	26
CdP ₂ -CDs-CoP	285	27
CoFeP/CoP/CC	240	28

Table S3. Overall water splitting performances of our sample compared with recently reported representative electrocatalysts in alkaline electrolytes

catalyst	Potential (V) at 10 mA cm⁻²	Reference
NiCoFeP	1.60	This work
CoTe ₂ @NCNTFs	1.67	29
MoS ₂ -NiS ₂ /NGF	1.64	1
Ni ₃ S ₂ @NGCLs/NF	1.55	19
NiS/Ni ₂ P	1.67	30
CoP/NCNHP	1.64	31
Fe-CoP HTPAs	1.59	32
CoTeNR/NF	1.64	33
A-NiSe ₂ P	1.62	34
FeCo-FeCoP@C	1.64	17
CoxP	1.72	35
CoNi ₂ S ₄ /Ni ₃ S ₂ @NF	1.65	36
Co ₉ S ₈ /Cu ₂ S/CF	1.6	37
Fe-Ni ₃ P ₄ /NiFeOH	1.55	38
CuO@CoZn-LDH/CF	1.55	39

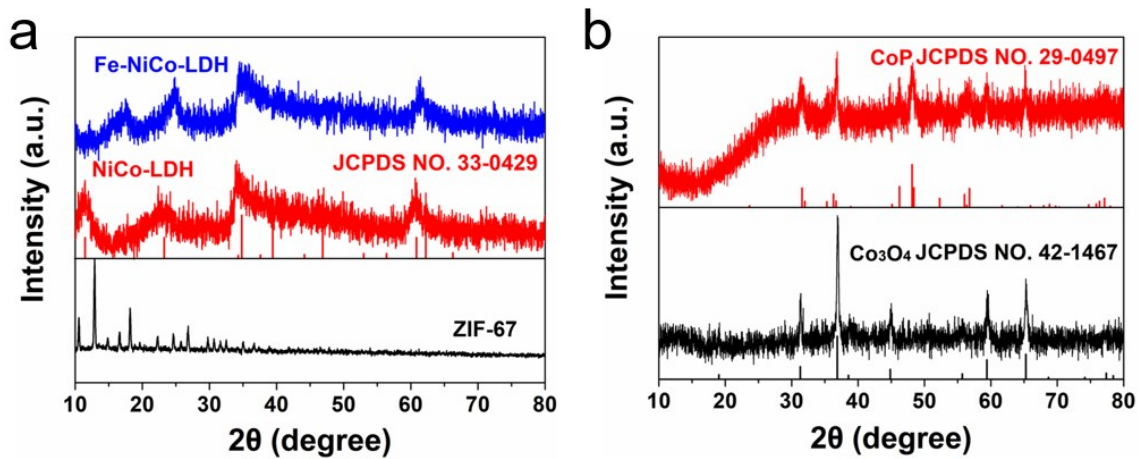


Fig. S1. (a) XRD pattern of the as-prepared ZIF-67, NiCo-LDH and Fe-NiCo-LDH samples; (b) XRD pattern of the as-prepared Co₃O₄ and CoP samples.

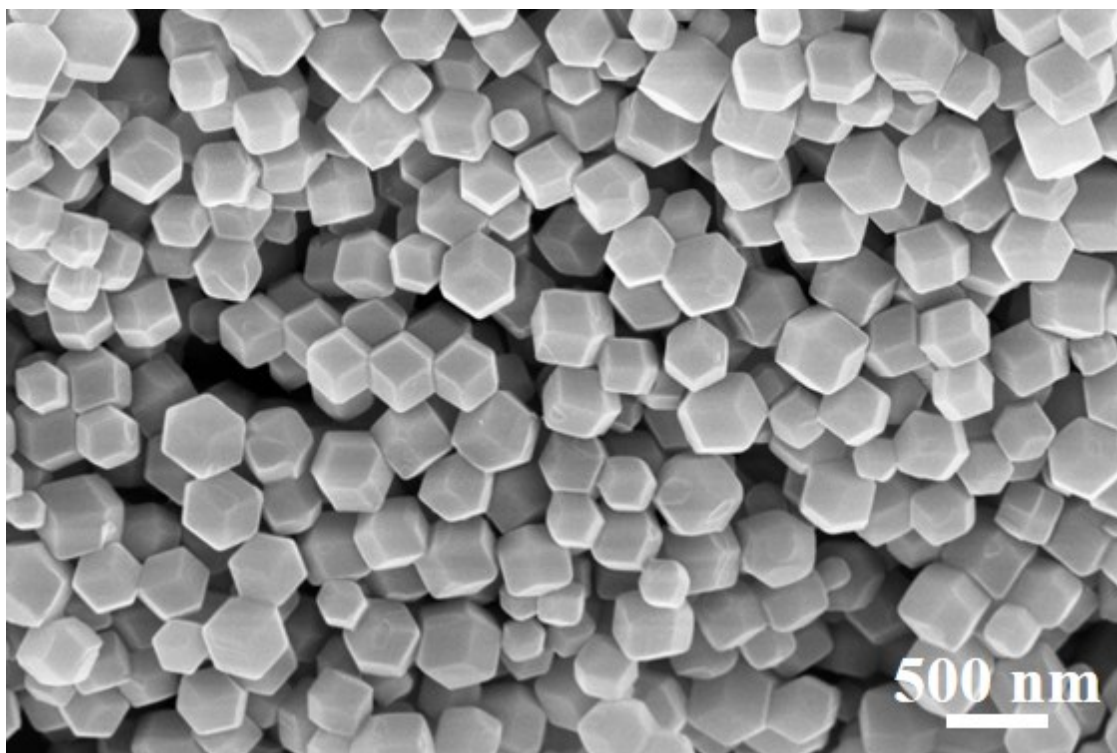


Fig. S2. SEM image of ZIF-67.

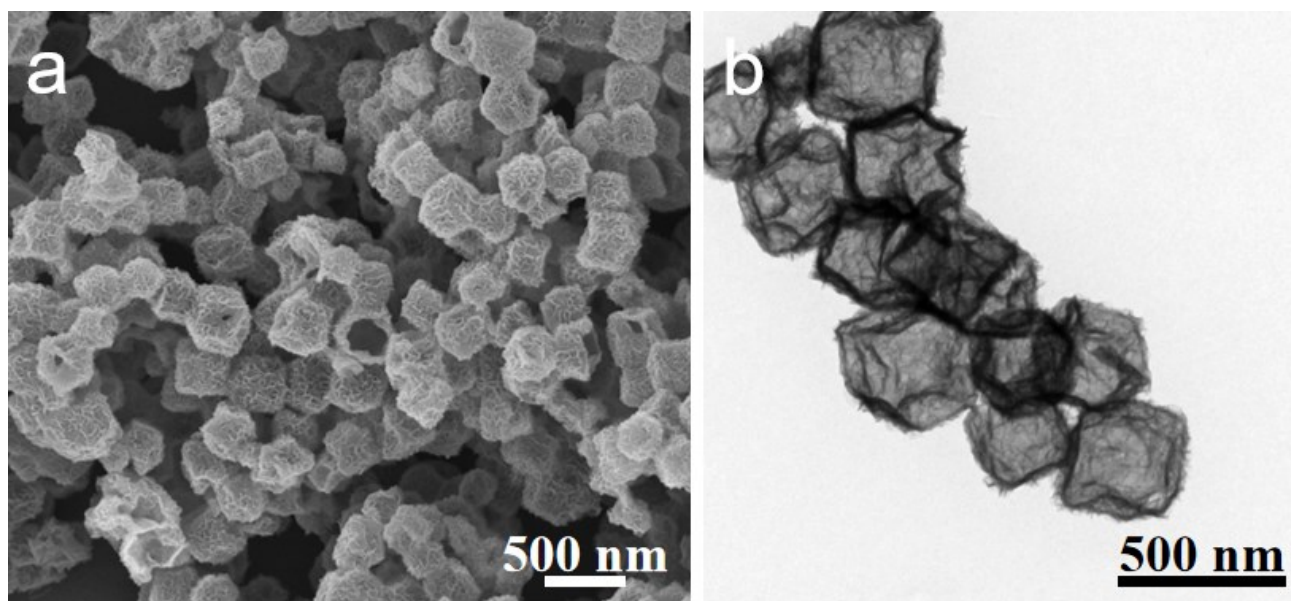


Fig. S3. (a, b) SEM and TEM images of NiCo-LDH.

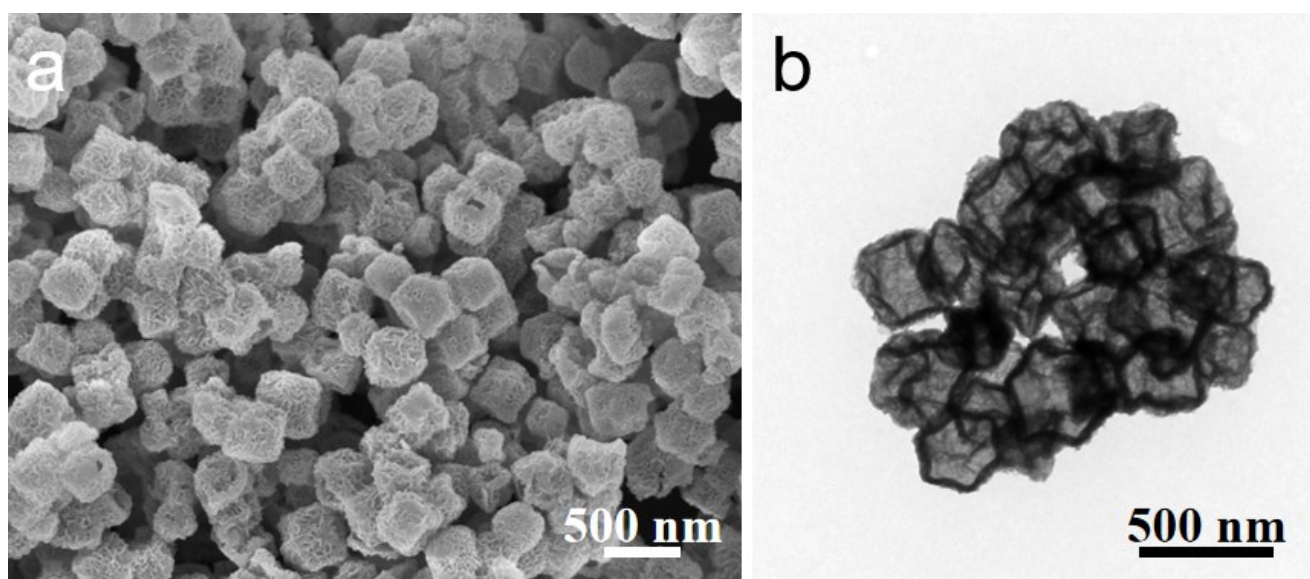


Fig. S4. (a, b) SEM and TEM images of Fe-NiCo-LDH.

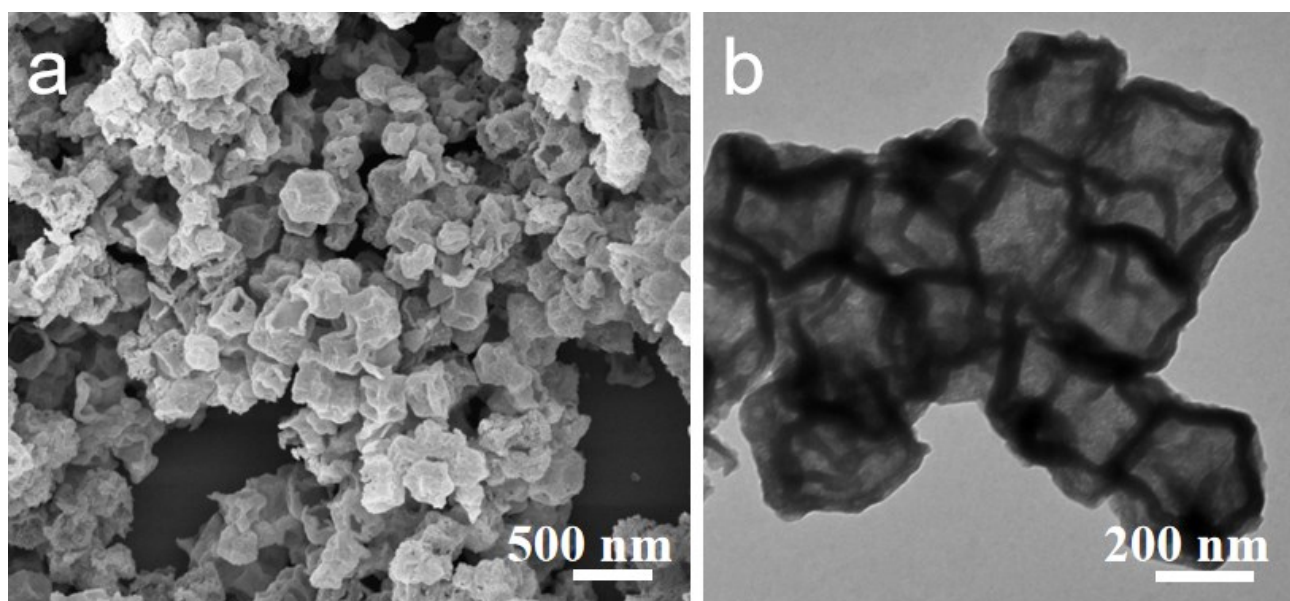


Fig. S5. (a, b) SEM and TEM images of NiCoP.

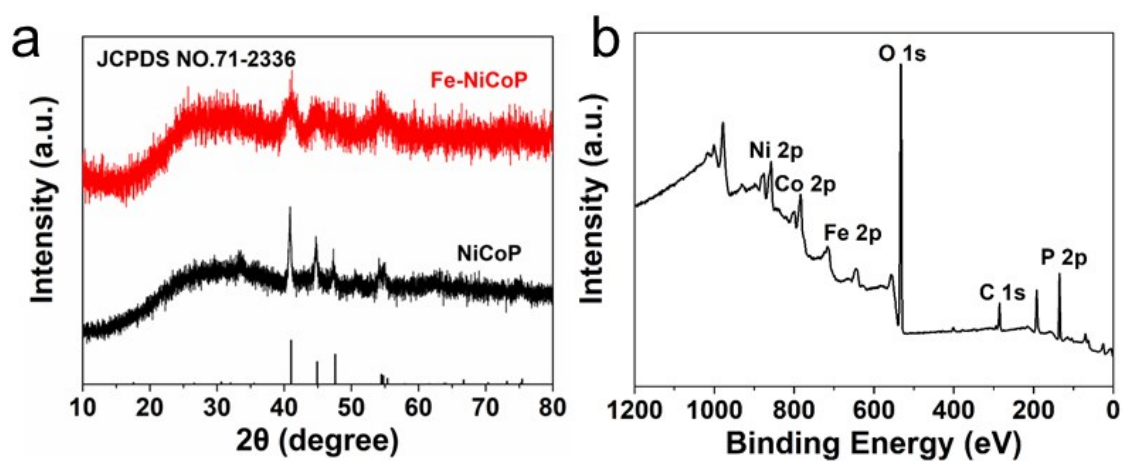


Fig. S6. (a) XRD pattern of the as-prepared NiCoP and Fe-NiCoP samples; (b) XPS survey spectra of Fe-NiCoP sample.

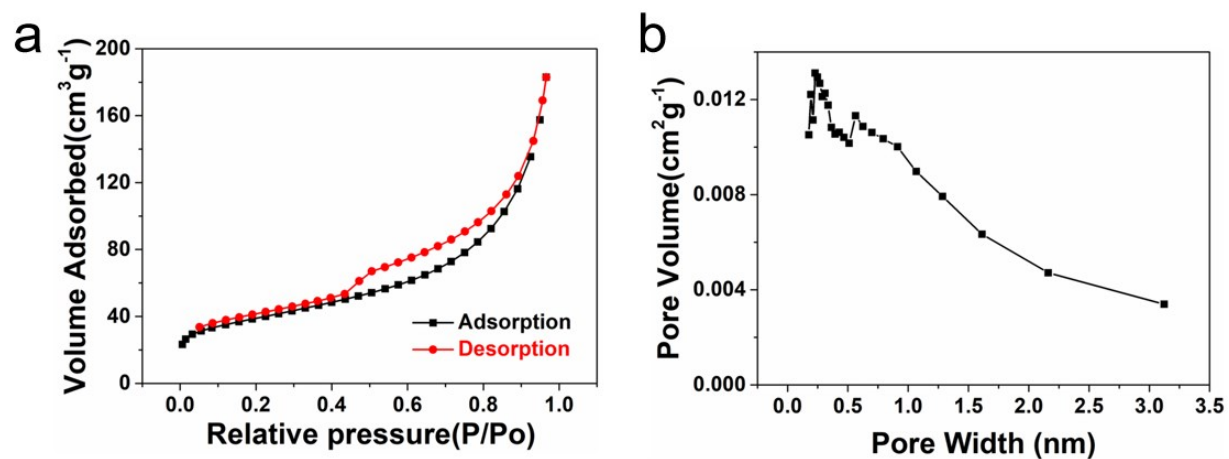


Fig. S7. (a,b) N₂ adsorption-desorption isotherms and pore size distribution of the as-prepared Fe-NiCoP.

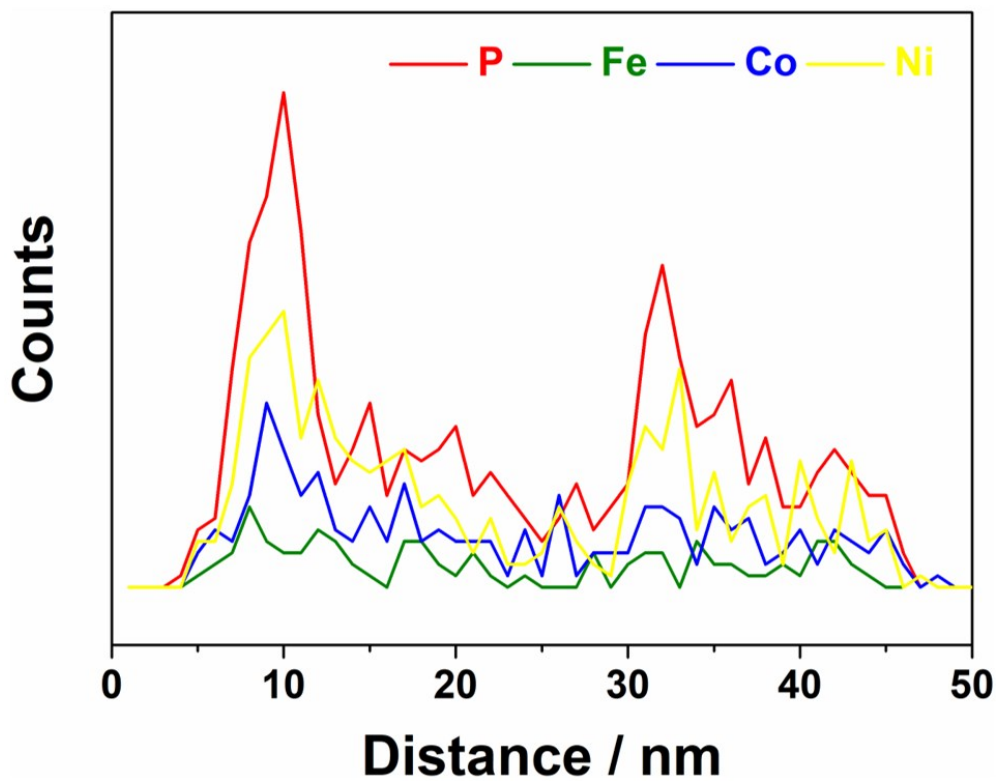


Fig. S8. EDX cross-sectional line scan profiles of the Fe-NiCoP sample.

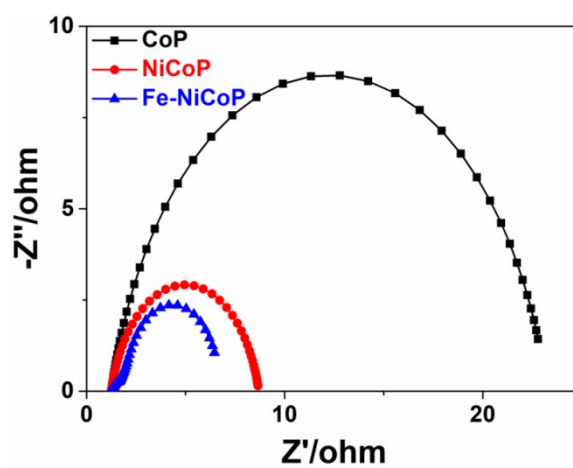


Fig. S9. Nyquist plots of CoP, NiCoP, and Fe-NiCoP samples.

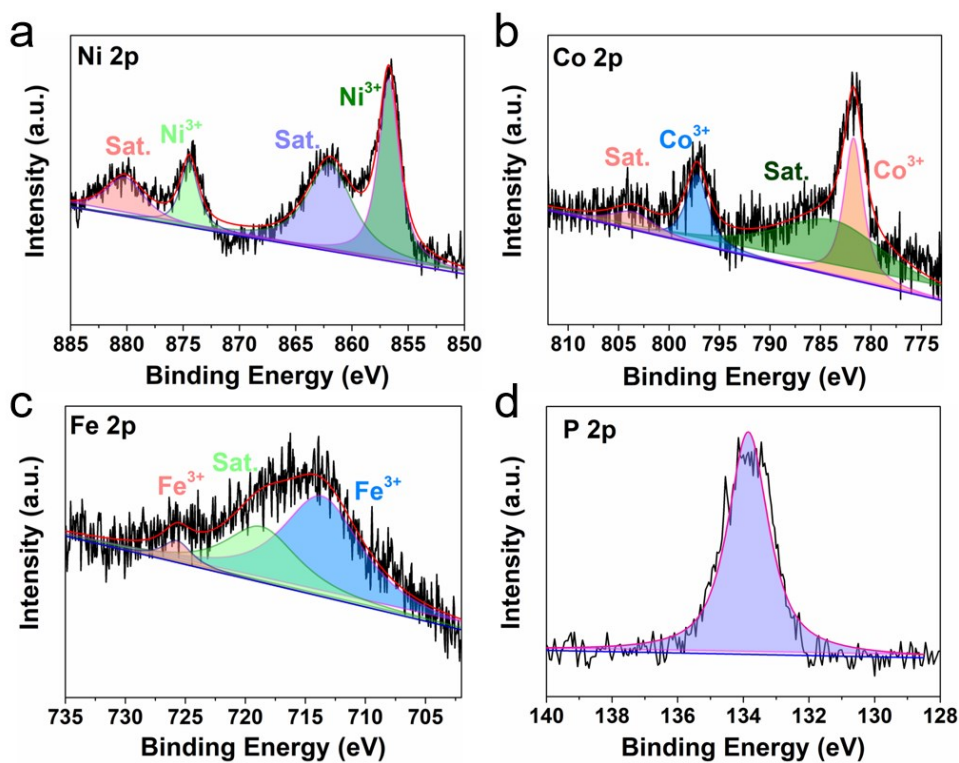


Fig. S10. XPS spectra of the Fe-NiCoP after OER test: (a) Ni 2p, (b) Co 2p, (c) Fe 2p, and (d) P 2p.

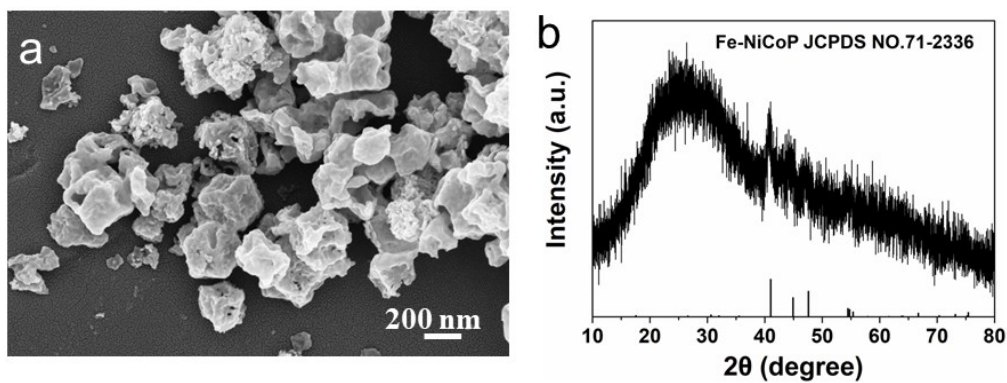


Fig. S11. (a) SEM and (b) XRD pattern for Fe-NiCoP after HER test.

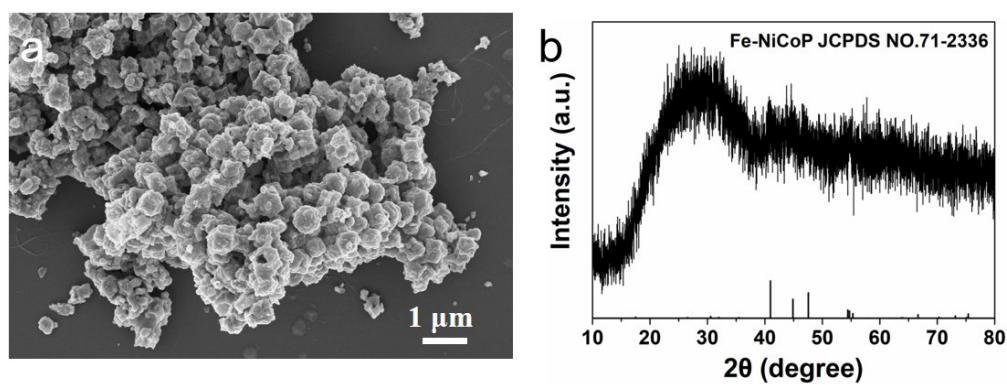


Fig. S12. (a) SEM and (b) XRD pattern for Fe-NiCoP after OER test.

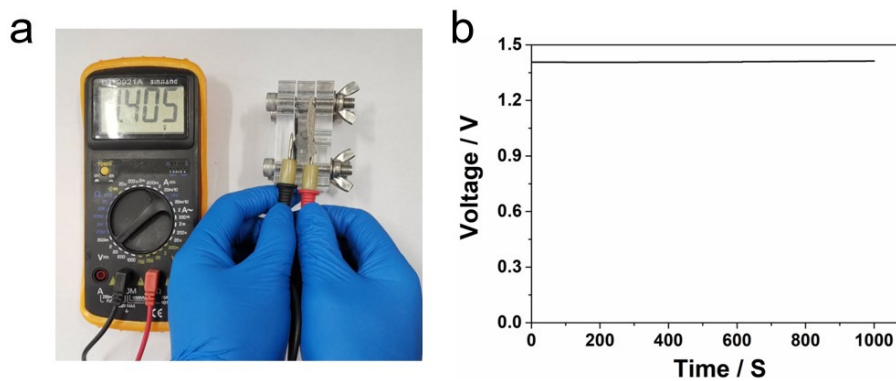


Fig. S13. (a,b) Open-circuit voltage plot of the Fe-NiCoP + Pt/C based Zn-air battery.

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