A high contrast mechanochromic luminescent diacetylene-linked bisbenzothiadiazole derivative

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S1. General Information

Proton and carbon NMR spectra. Proton and carbon NMR spectra were recorded on Bruker AC 300 spectrometer (¹H 300MHz, ¹³C 75 MHz). Samples were recorded as solution in deuterated NMR solvents as stated and chemical shifts (δ) are quoted in parts per million, referenced to residual solvent.

Linear Spectroscopy: UV-vis in solution studies were carried out on a PerkinElmer Lamba XLS+ spectrometer. UV-is in solid state were carried out on a Shimadzu UV-2401 PC spectrophotometer. Fluorescence spectra were recorded on a Varian Cary Eclipse spectrophotometer.

Cyclic Voltammetry Measurements. Cyclic voltammetry (CV) experiments were performed on a Bioanalytical Systems Inc. (BASI) Epsilon electrochemical workstation in a threeelectrode cell at room temperature under nitrogen atmosphere. Electrochemical measurements were carried out in CH_2Cl_2 solution (c = 1 x 10⁻³ M) containing 0.1 M tetra-n-butylammonium hexafluorophosphate (TBAPF₆) of supporting electrolyte at a scan rate 100 mV/s. A three electrode setup was used including a platinum working electrode, a Ag/AgCl (3 M NaCl) reference electrode, and a platinum wire auxiliary electrode.

X-ray structure determinations: Crystal of **4** showing well defined faces were mounted Bruker Kappa Apex II (X8 APEX) diffractometer equipped with a Mo INCOATED microsource. Diffraction data were collected exploring over a hemisphere of the reciprocal space in a combination of φ and ω scans to reach a resolution of 0.86 Å, using a Bruker APEXII software suite (APEX2; Bruker-AXS: Madison, WI, 2006). The structures were solved by the Multan and Fourier methods. Most of the calculations were carried out with APEXII software for data collection and reduction, and OLEX2 for structure solution and refinements.

S2 Synthesis and characterization of 3-5

Synthesis of 4-ethynyl-7-(4-nonylphenyl)benzo[c][1,2,5]thiadiazole (3).

A mixture of **2** (200 mg, 0.48 mmol), CuI (4.6 mg, 0.024mmol) and Pd(dppf)₂Cl₂ (17.6 mg, 0.024mmol) in 6 ml of a 1:1 mixture triethylamine:THF was degassed, and then, ethynyl trimethylsilane (0.73ml, 0.52 mmol) was added. The solution was irradiated with an Anton Paar microwave irradiator (CEM) at 120°C (80W) for 120 min. After cooling to room temperature, the mixture was diluted with CH₂Cl₂, washed with water, and dried (MgSO₄); the solvent was then evaporated and the residue was purified by chromatography with CH₂Cl₂/hexane (1:3) to give a yellow solid 4-(4-nonylphenyl)-7-((trimethylsilyl)ethynyl)benzo[*c*][1,2,5]thiadiazole (177 mg, 85%).

¹H NMR (300 MHz, CDCl₃) δ 7.86 (d, J = 8.2 Hz, 2H), 7.83 (d, J = 7.4 Hz, 1H), 7.64 (d, J = 7.4 Hz, 1H), 7.34 (d, J = 8.2 Hz, 2H), 2.69 (t, J = 7.7 Hz, 2H), 1.86 – 1.56 (m, 2H), 1.51 – 1.16 (m, 12H), 1.00 – 0.80 (m, 3H), 0.35 (s, 9H).¹³C NMR (75 MHz,CDCl₃) δ 144.0, 135.2, 134.4, 134.1, 129.3, 128.9, 127.2, 115.5, 101.6, 100.7, 35.9, 32.0, 31.5, 29.9, 29.7, 29.5, 22.8, 14.2, 0.1. UV (CH₂Cl₂, 25 °C) λ_{max} (log ε) 275(4.98), 392(4.63). FAB MS m/z 434.71 (M⁺); HRMS (FAB) calcd for C₂₆H₃₄N₂SSi: 434.2212, found: 434.2205.

A mixture of 4-(4-nonylphenyl)-7-((trimethylsilyl)ethynyl)benzo[*c*][1,2,5]thiadiazole (100mg, 0.23mmol) and KF (138mg, 3.45mmol) was stirred for 12h at room temperature in 8 ml of a 1:1 mixture of THF:MeOH. Then the mixture was diluted with CH₂Cl₂, washed with water, and dried (MgSO₄); the solvent was then evaporated to give a yellow solid (**3**) (75 mg, 90%) .¹H NMR (300 MHz, CDCl₃) δ 7.89 (d, *J* = 7.2 Hz, 2H), 7.84 (d, *J* = 7.8 Hz, 1H), 7.67 (d, *J* = 7.3 Hz, 1H), 7.35 (d, *J* = 7.9 Hz, 2H), 3.61 (s, 1H), 2.69 (t, *J* = 7.7 Hz, 2H), 2.03 (t, *J* = 7.1 Hz, 2H), 1.41 – 1.17 (m, 12H), 0.88 (t, *J* = 6.5 Hz, 3H). ¹³C NMR (75 MHz,CDCl₃) δ 144.2, 139.4, 135.8, 134.2, 129.3, 129.0, 127.2, 123.7, 114.2, 83.6, 79.7, 36.0, 34.0, 32.1, 31.5, 29.8, 29.7, 29.5, 29.3, 29.1, 22.8, 14.2. UV (CH₂Cl₂, 25 °C) λ_{max} (log ε) 267(3.34), 383(3.05). FAB MS m/z 362.53 (M⁺); HRMS (FAB) calcd for C₂₃H₂₆N₂S: 362.1817, found: 362.1825.

Synthesis of 7,7'-bis(4-nonylphenyl)-4,4'-bibenzo[c][1,2,5]thiadiazole. (4)

A mixture of **2** (150mg, 0.37mmol), Pd(OAc)₄ (4.1 mg, 0.02 mmol), K₂CO₃ (54.6 mg, 0.4mmol) and polyethilene glycol (PEG 4000, 359 mg, 0,09 mmol) in 2 ml of a dimethylformamide was stirred for 24h at 120°C. Then the mixture was diluted with Diethyl ether, washed with water, and dried (MgSO₄); the solvent was then evaporated and the residue was purified by chromatography with CH₂Cl₂/Hexane (1:2) to give a yellow solid **4** (53 mg, 45%). ¹H NMR (300 MHz, CDCl₃) δ 8.41 (d, *J* = 7.4 Hz, 2H), 7.93 (d, *J* = 6.9 Hz, 4H), 7.90

(d, J = 6.1 Hz, 2H), 7.39 (d, J = 8.1 Hz, 4H), 2.71 (t, J = 7.8 Hz, 4H), 1.78 – 1.60 (m, 4H), 1.45 – 1.26 (m, 24H), 0.90 (d, J = 6.4 Hz, 6H). ¹³C NMR (75 MHz, CDCl₃) δ 154.31, 154.15, 143.65, 134.64, 134.43, 131.18, 129.20, 128.80, 128.53, 127.65, 35.87, 31.94, 31.50, 29.60, 29.46, 29.39, 22.72, 14.18. FAB MS m/z 674.35 (M⁺); HRMS (FAB) calcd for C₄₂H₅₁N₄S₂: 675.3550, found: 675.3548 (M+1)

Synthesis of 1,4-bis(7-(4-nonylphenyl)benzo[c][1,2,5]thiadiazol-4-yl)buta-1,3-diyne. (5)

A mixture of **5** (100mg, 0.27mmol) and Cu₂(OAc)₄•H₂O (40.4mg, 0.20mmol) in 10ml of a 1:1 mixture of Piperidine: CH₂Cl₂ was stirred for 12h at room temperature. Then the mixture was diluted with CH₂Cl₂, washed with water, and dried (MgSO₄); the solvent was then evaporated and the residue was purified by chromatography with CH₂Cl₂ /Hexane (1:3) to give a yellow solid **5** (57 mg, 58%).¹H NMR (300 MHz, CDCl₃) δ 7.95 (d, *J* = 7.4 Hz, 1H), 7.88 (d, *J* = 8.0 Hz, 2H), 7.70 (d, *J* = 7.4 Hz, 1H), 7.37 (d, *J* = 8.0 Hz, 2H), 2.70 (t, *J* = 7.7 Hz, 2H), 1.82 – 1.60 (m, 2H), 1.44 – 1.18 (m, 12H), 1.00 – 0.78 (m, 3H).¹³C NMR (75 MHz, CDCl₃) δ 155.8, 153.3, 144.4, 136.3, 135.0, 134.2, 129.3, 129.0, 127.3, 114.0, 80.2, 80.0, 36.0, 32.0, 31.5, 29.7, 29.5, 22.8, 14.3. MALDI MS m/z 722.4 (M⁺)HRMS (MALDI) calcd for C₄₉H₅₀N₄S₂: 722.3471, found: 722.3472.



S3. Copy of the ¹H NMR and ¹³C NMR spectra of compounds 3-5

Figure S2. Copy of ¹³C NMR(75MHz, CDCl₃) experimental spectra for 3.



Figure S4. Copy of ¹³C NMR(75 MHz, CDCl₃) experimental spectra for 4.



Figure S6. Copy of ¹³C NMR(75 MHz, CDCl₃) experimental spectra for 5.

S4. Single crystal X-ray structure determination

Crystal Structure Report of 4.

A yellow plate-like specimen of $C_{42}H_{50}N_4S_2$, approximate dimensions 0.015 mm x 0.323 mm x 0.372 mm, was used for the X-ray crystallographic analysis. The X-ray intensity data were measured.

The total exposure time was 21.50 hours. The frames were integrated with the Bruker SAINT software package using a narrow-frame algorithm. The integration of the data using a triclinic unit cell yielded a total of 134454 reflections to a maximum θ angle of 25.35° (0.83 Å resolution), of which 34850 were independent (average redundancy 3.858, completeness = 99.5%, R_{int} = 12.62%, R_{sig} = 18.12%) and 11926 (34.22%) were greater than $2\sigma(F^2)$. The final cell constants of <u>a</u> = 7.7495(2) Å, <u>b</u> = 33.5689(8) Å, <u>c</u> = 39.6305(9) Å, α $= 68.5122(14)^{\circ}, \beta = 88.3620(15)^{\circ}, \gamma = 84.7067(16)^{\circ}, \text{ volume} = 9551.9(4)^{\circ} \text{Å}^{3}, \text{ are based upon}$ the refinement of the XYZ-centroids of 5792 reflections above 20 $\sigma(I)$ with 4.941° < 20 < 42.54°. Data were corrected for absorption effects using the multi-scan method (SADABS). The ratio of minimum to maximum apparent transmission was 0.899. The calculated and maximum transmission coefficients (based minimum on crystal size) are 0.9380 and 0.9970.

The final anisotropic full-matrix least-squares refinement on F^2 with 2171 variables converged at R1 = 7.71%, for the observed data and wR2 = 23.41% for all data. The goodness-of-fit was 0.955. The largest peak in the final difference electron density synthesis was 0.253 e⁻/Å³ and the largest hole was -0.379 e⁻/Å³ with an RMS deviation of 0.060 e⁻/Å³. On the basis of the final model, the calculated density was 1.173 g/cm³ and F(000), 3620 e⁻.

Table S1. Sample and crystal data for 4

Identification code	4	
Chemical formula	$C_{42}H_{50}N_4S_2$	
Formula weight	674.98 g/mol	
Temperature	296(2) K	
Wavelength	0.71073 Å	
Crystal size	0.015 x 0.323 x 0.372 mm	
Crystal habit	yellow plate	
Crystal system	triclinic	
Space group	P -1	
Unit cell dimensions	a = 7.7495(2) Å	$\alpha = 68.5122(14)^{\circ}$
	b = 33.5689(8) Å	$\beta = 88.3620(15)^{\circ}$
	c = 39.6305(9) Å	$\gamma = 84.7067(16)^{\circ}$
Volume	9551.9(4) Å ³	•
Z	10	
Density (calculated)	1.173 g/cm ³	
Absorption coefficient	0.173 mm ⁻¹	
F(000)	3620	
Table S2. Data colle	ction and structure	refinement for 4.

Theta range for data 1.00 to 25.35°

collection			
Index ranges	-9<=h<=9, -40<=k<=40, -47<=l<=47		
Reflections collected	134454		
Independent reflections	34850 [R(int) = 0.1262]		
Coverage of independent reflections	99.5%		
Absorption correction	multi-scan		
Max. and min. transmission	0.9970 and 0.9380		
Refinement method	Full-matrix least-squares on F ²		
Refinement program	SHELXL-2014/7 (Sheldrick, 2014)		
Function minimized	$\Sigma w (F_o^2 - F_c^2)^2$		
Data / restraints / parameters	34850 / 0 / 2171		
Goodness-of-fit on F ²	0.955		
Δ/σ_{max}	0.001		
Final R indices	11926 data; I>2σ(I)	R1 = 0.0771, wR2 = 0.1569	
	all data	R1 = 0.2474, wR2 = 0.2341	
Weighting scheme	w=1/[$\sigma^2(F_o^2)$ +(0.0903P) ²] where P=(F_o^2 +2 F_c^2)/3		
Largest diff. peak and hole	0.253 and -0.379	eÅ-3	
R.M.S. deviation from mean	0.060 eÅ ⁻³		



Figure S7. Five crystallographic independent molecules in the asymmetric unit cell of 4.

S4 PXRD patterns of 5



Figure S8. PXRD experimental pattern of 5a.



Figure S9. PXRD experimental pattern of 5β.