Modulation of Supramolecular Self-Assembly of BODIPY Tectons via Halogen Bonding

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Scheme 1. Synthetic pathway of compounds 1-3 and B-1, B-2, and B-3

1. Experimental Section

1.1. Materials and Methods
All chemicals and solvents were purchased from Sigma Aldrich and TCI Chemicals used as supplied without further purification unless stated otherwise. NMR spectra were recorded on a Bruker 400 spectrometer ($^1$H, 500 MHz; $^{13}$C, 125 MHz). MALDI-TOF was performed on a Bruker Microflex LT MALDI-TOF-MS Instrument.

1.1. 2. X-ray Crystallography

Single crystal data were collected on a Bruker APEX II QUAZAR three-circle diffractometer. Indexing was performed using APEX2. Data integration and reduction was carried out with SAINT. Absorption correction was performed by multi-scan method implemented in SADABS. The structure was solved using SHELXT and then refined by full-matrix least-squares refinements on $F^2$ using the SHELXL in Olex2 Software Package. Aromatic and aliphatic C-bound H atoms were positioned geometrically and refined using a riding mode. One entire BODIPY molecule in the asymmetric unit of B-2 is disordered over two sites with occupancies 0.84:0.16. Crystal structure validations, geometrical calculations and crystal packing analysis were performed using Platon software. The molecular drawings were carried out with Mercury CSD program.

2. Synthesis

The compounds 1, 2 and 3 were synthesized and purified according to literature. Compounds B-1, B-2 and B-3 were prepared with modified methods described in literature.
Figure S1. $^1$H NMR Spectrum of Compound 1
Figure S2. $^{13}$C NMR Spectrum of Compound 1

Figure S3. Mass Spectrum of Compound 1
Figure S4. $^1$H NMR Spectrum of Compound 2
Figure S5. $^{13}$C NMR Spectrum of Compound 2

Figure S6. Mass Spectrum of Compound 2

Figure S7. $^1$H NMR Spectrum of Compound 3
Figure S8. $^{13}$C NMR Spectrum of Compound 3

Figure S9. Mass Spectrum of Compound 3
Figure S10. $^1$H NMR Spectrum of Compound B-1
Figure S11. $^{13}$C NMR Spectrum of Compound B-1

Figure S12. Mass Spectrum of Compound B-1
Figure S13. $^1$H NMR Spectrum of Compound B-2

Figure S14. $^{13}$C NMR Spectrum of Compound B-2

Figure S15. Mass Spectrum of Compound B-2
Figure S16. $^1$H NMR Spectrum of Compound B-3
Figure S17. $^{13}$C NMR Spectrum of Compound B-3

Figure S18. Mass Spectrum of Compound B-3
### Table S2: The intermolecular D-H···A interaction parameters (Å and °) for BODIPY compounds.

<table>
<thead>
<tr>
<th>D-H···A</th>
<th>Symmetry</th>
<th>d(D-H)</th>
<th>d(H···A)</th>
<th>d(D-H···A)</th>
<th>D-H···A</th>
</tr>
</thead>
<tbody>
<tr>
<td>B-2</td>
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<td></td>
<td></td>
<td></td>
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</tr>
<tr>
<td>C23-H23···F2</td>
<td>1655.01</td>
<td>0.93</td>
<td>2.588</td>
<td>3.228</td>
<td>126.43</td>
</tr>
<tr>
<td>C46-H46···F1</td>
<td>2665.03</td>
<td>0.93</td>
<td>2.506</td>
<td>3.338</td>
<td>148.10</td>
</tr>
<tr>
<td>C35-H35A···F1</td>
<td>-</td>
<td>0.96</td>
<td>2.474</td>
<td>2.755</td>
<td>96.54</td>
</tr>
<tr>
<td>C35-H35C···F1</td>
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<td>0.96</td>
<td>2.575</td>
<td>2.755</td>
<td>90.45</td>
</tr>
<tr>
<td>C15-H15···F2</td>
<td>2656.01</td>
<td>0.96</td>
<td>2.650</td>
<td>3.366</td>
<td>134.21</td>
</tr>
<tr>
<td>C47-H47···N5</td>
<td>2765.03</td>
<td>0.93</td>
<td>2.598</td>
<td>3.502</td>
<td>163.89</td>
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<tr>
<td>C45-H45···π1 (Cg2)</td>
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<td>2.554</td>
<td>3.088</td>
<td>116.70</td>
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<tr>
<td>C47-H47···π2 (Cg12)</td>
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<td>0.93</td>
<td>2.38</td>
<td>3.13(7)</td>
<td>137</td>
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### Table S3. Geometric parameters used for the determination of the π···π interactions*

<table>
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<tr>
<th>Rings Cg(I)-Cg(J)*</th>
<th>Symmetry</th>
<th>Cg···Cg</th>
<th>Cg(I)-Perp</th>
<th>Cg(J)-Perp</th>
<th>α</th>
<th>β</th>
<th>γ</th>
</tr>
</thead>
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<td>B-1</td>
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<td></td>
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<td></td>
<td></td>
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<td></td>
</tr>
<tr>
<td>Cg(1)···Cg(1)</td>
<td>x, 1-y, 1-z</td>
<td>4.3842(14)</td>
<td>3.8173</td>
<td>3.8172</td>
<td>0</td>
<td>29.5</td>
<td>29.5</td>
</tr>
<tr>
<td>B2</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Cg(1)···Cg(1)</td>
<td>1-x, -y, 1-z</td>
<td>4.489(5)</td>
<td>4.059(4)</td>
<td>4.060(4)</td>
<td>0.0(6)</td>
<td>25.3</td>
<td>25.3</td>
</tr>
<tr>
<td>Cg(1)···Cg(9)</td>
<td>1-x, 1-y, 1-z</td>
<td>3.946(9)</td>
<td>3.401(4)</td>
<td>3.830(8)</td>
<td>17.1(8)</td>
<td>13.9</td>
<td>30.5</td>
</tr>
<tr>
<td>Cg(2)···Cg(14)</td>
<td>1-x, 1-y, -z</td>
<td>4.13(2)</td>
<td>2.232(4)</td>
<td>3.82(2)</td>
<td>42</td>
<td>22.4</td>
<td>57.3</td>
</tr>
<tr>
<td>Cg(4)···Cg(7)</td>
<td>1-x, 1-y, 1-z</td>
<td>3.864(8)</td>
<td>3.795(4)</td>
<td>3.429(7)</td>
<td>16.7(8)</td>
<td>27.5</td>
<td>10.8</td>
</tr>
<tr>
<td>Cg(4)···Cg(11)</td>
<td>1-x, 1-y, 1-z</td>
<td>3.93(3)</td>
<td>3.772(4)</td>
<td>3.46(4)</td>
<td>12</td>
<td>28.1</td>
<td>16.1</td>
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<tr>
<td>Cg(7)···Cg(11)</td>
<td>2-x, 1-y, 1-z</td>
<td>4.55(3)</td>
<td>3.918(7)</td>
<td>4.00(3)</td>
<td>7</td>
<td>28.6</td>
<td>30.6</td>
</tr>
<tr>
<td>Cg(9)···Cg(11)</td>
<td>2-x, 1-y, -z</td>
<td>4.73(4)</td>
<td>4.319(7)</td>
<td>2.39(3)</td>
<td>48</td>
<td>59.7</td>
<td>24.1</td>
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<tr>
<td>Cg(12)···Cg(14)</td>
<td>2-x, 1-y, -z</td>
<td>4.31(5)</td>
<td>1.08(4)</td>
<td>4.07(2)</td>
<td>60</td>
<td>19.3</td>
<td>75.4</td>
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<tr>
<td>B-3</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Cg(4)···Cg(9)</td>
<td>x, y, z</td>
<td>4.062(7)</td>
<td>3.971(5)</td>
<td>3.174(4)</td>
<td>26.9(6)</td>
<td>38.6</td>
<td>12.2</td>
</tr>
<tr>
<td>Cg(9)···Cg(4)</td>
<td>x, y, z</td>
<td>4.061(7)</td>
<td>3.175(4)</td>
<td>3.970(5)</td>
<td>26.9(6)</td>
<td>38.6</td>
<td>12.2</td>
</tr>
</tbody>
</table>

* (Cg···Cg < 5.0 Å). a In B-1, Cg(1) is the centroid of the pyrrole ring involving atoms N1-C1-C2-C3-C4. In B-2, Cg(1), Cg(2), Cg(4), Cg(7), Cg(9), Cg(11), Cg(12), and Cg(14) are the centroids of the rings N1-C4, N2-C9, N3-C17, N4-C67, N9-C64, N4-C33, N5-C31, and N6-C47, respectively. In B-3, Cg(4) and Cg(9) are the centroids of the rings N3-C24 and N4-C28, respectively. b Distance between ring centroids (Å). c Perpendicular distance of Cg(I) on ring J (Å). d Perpendicular distance of Cg(J) on ring I (Å). e Dihedral Angle between planes I and J (°). f Angle
between the centroid vector \( \mathbf{C}_g(I) \cdots \mathbf{C}_g(J) \) vector and normal to plane I (°).\(^8\) Angle between the centroid vector \( \mathbf{C}_g(I) \cdots \mathbf{C}_g(J) \) vector and normal to plane J (°).

**Hirshfeld surface analysis**

**Comparison of intermolecular interactions in B-1, B-2, and B-3**

In order to get a better insight into the intermolecular interactions in BODIPY structures (B-1, B-2, and B-3), Hirshfeld surfaces\(^5\) incorporating two-dimensional (2D) fingerprint plots\(^6\) using Crystal Explorer\(^7\) program was used. The normalized contact distance \( d_{\text{norm}} \) surface, which expressed in terms of distances to the surface from the nuclei inside and outside the Hirshfeld surface \( d_i \) and \( d_e \), respectively) and the vDW radii of the atoms, defined as Eq. 1 gives identification of the regions of particular importance to intermolecular interactions.\(^8,9\) The 2D fingerprint plots, which were derived from the combination of \( d_i \) and \( d_e \), were used for quantifying the intermolecular contacts in the crystal.

\[
d_{\text{norm}} = \frac{d_i - r_{\text{vdw}}}{r_{\text{vdw}}} + \frac{d_e - r_{\text{vdw}}}{r_{\text{vdw}}}
\]

Equation (1)

Full fingerprint plots and their resolved fingerprint plots showing the percentage contributions to the total Hirshfeld surface area in B-1, B-2, and B-3 are given in Figure S19, S20, and S21. Also, Figure S22 shows the relative percentage contributions of different intermolecular contacts contributing to the Hirshfeld surface. The 2D fingerprint plots exhibit that the most dominant interaction in B-1, B-2 and B-3 is the H/H interaction contributing to the total Hirshfeld surface with the values equal to 45%, 51.9%, and 49.7%, respectively. Unlike B-2 and B-3, according to the 2D fingerprint plot B-1, the second most contribution to the total Hirshfeld surface in B-1 is from I/H···H/I interactions appearing as distinct spikes, which constitute 20.4%. The C···H/H···C contacts representing C-H···π interactions comprise 12.9%, 15.6%, and 17.3% of the total
Hirshfeld surfaces for B-1, B-2, and B-3, respectively. As for the non-classical C-H⋯F hydrogen bonding interactions, these contacts are identified with the ratio of 3.5% (B-1), 12.9% (B-2), and 11.9% (B-3) of F/H⋯H/F in 2D fingerprint plots, and have an important role in the stabilization of 3D supramolecular network of B-2 and B-3. With regard to the C-I⋯N XBs interactions, the I⋯N/N⋯I intermolecular interactions that represent XBs comprise 3.2% and 3.0% of the total Hirshfeld surfaces for B-1 and B-2, respectively. On the other hand, B-3 exhibits the I⋯F/F⋯I XBs interactions with the minor ratio of 1.4%. B-1 and B-3 do not show the close π–π stacking interactions (the C⋯C intermolecular interactions is smaller than 1% of the total Hirshfeld surfaces for both molecules) and also no exhibits the adjacent red and blue triangles on the shape index surface and also do not exhibit large flat region on the curvedness mapped on the Hirshfeld surface, as shown in Figure S23. For B-2, the C⋯C intermolecular interactions comprise 3.5% of the total Hirshfeld surface represent the aromatic π–π stacking ranging from 3.864(8) Å to 4.73(4) Å, as indicated in Table S3, between BODIPY molecules in crystal packing.
Figure S19. Full fingerprint plots and the resolved fingerprint plots showing the percentage contributions to the total Hirshfeld surface area in B-1.
Figure S20. The weak C-H···π_{BODIPY} interactions in B-2.
Figure S21. The $\pi_{\text{BODIPY}} \cdots \pi_{\text{BODIPY}}$ interactions in B-2.
Figure S22. Full fingerprint plots and the resolved fingerprint plots showing the percentage contributions to the total Hirshfeld surface area in B-2.
Figure S23. Full fingerprint plots and the resolved fingerprint plots showing the percentage contributions to the total Hirshfeld surface area in B-3.
Figure S24. Relative percentage contributions of different intermolecular contacts contributing to the Hirshfeld surface in B-1, B-2, and B-3.
Figure S25. Perspective view of shape index (left) and curvedness (right)-mapped Hirshfeld surfaces in B-1, B-2, and B-3.
Figure S26. The $\pi_{\text{BODIPY}} \cdots \pi_{\text{BODIPY}}$ interactions in B-3.
Table S4. All BODIPY compounds containing I···F XB contacts in the Cambridge Structural Database (CSD) system.

<table>
<thead>
<tr>
<th>CSD Ref. Code</th>
<th>REPRESENTATION</th>
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<th>Ref.</th>
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<td>1D ZIG-ZAG CHAIN</td>
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<td>DANCAY</td>
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<td>1D ZIG-ZAG CHAIN</td>
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Figure S27. Electrostatic potential maps of XB dimer of B-1, B-2(I, II), and B-3 (I, II) along with the corresponding electrostatic potential values (kcal/mol).
Table S5. Molecular orbital plots of the HOMOs and LUMOs of B-1, B-2, and B-3.

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<tr>
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Figure S28. Centers of masses (shown as yellow spheres) in XB dimers of B-1, B-2, and B-3.

REFERENCES


7 M. J. Turner, J. J. McKinnon, S. K. Wolff, D. J. Grimwood, P. R. Spackman,


