Supporting Information

In-situ cleavage and rearrangement synthesis of an easy-obtained and high stable Cu(II)-based MOF for efficient heterogeneous catalysis of carbon dioxide conversion

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Figure S1 Asymmetric unit of **1**. Selective bond distance (Å) and angle (°) in **1**. Cu(1)-O(1) 1.940(4), Cu(1)-O(2) 1.945(4), Cu(1)-O(3) 2.296(5), Cu(1)-N(1) 2.018(5), O(1)-Cu(1)-O(2) 175.36(19), O(1)-Cu(1)-N(1) 92.1(2), O(2)-Cu(1)-N(1) 90.3(2), O(1)-Cu(1)-O3(1) 91.28(17), O(2)-Cu(1)-O(3) 84.56(17).



Figure S2 The coordination environment and connected mode of Cu chains.



Figure S3 Structure of 1 showing the three-dimensional frameworks with one-dimensional channels.



Figure S4 (*a*) The SEM image of 1; (*b-d*) the element mapping image of 1 for Cu, O, and N, respectively.



Figure S5 Isosteric adsorption enthalpy of CO_2 on 1.



Figure S6 PXRD patterns of fresh 1 (red) and after desolvation (black).



Figure S7. Comparisons of FT-IR spectra of 1, 1 after immersed in styrene oxide and free styrene oxide.



Scheme S1. Proposed cleavage and rearrangement process of the ligand *N*,*N*'-bis(4-picolinoyl)hydrazine into isonicotinate moiety.

Entry	MOF	Amount MOF / TBABr	T / ºC	P / atm	t / h	Yield / %	Ref.
1	Cu-NTTA	5 µmol / 1.5 mol%	100	10	8	56.3	39
2	Cu-MOF	2 µmol / 1.5 mol%	100	10	12	>99	40
3	JLU-Liu20	0.25 mol% / 5 mol%	80	1	48	72	41
4	Cu(TPA)	5 wt% / 5 mol%	70	1	10	93	42
5	Cu-MOF	0.4 mol% / 0.2 mol%	70	1	16	90	43
6	1	5 mol% / 0.5 mol%	100	5	12	>99	This work

Table S1. Comparisons of CO₂ cycloaddition catalysis based on Cu-MOFs heterogeneous catalysts.



Figure S8 GC spectrum of cycloaddition product 1b.



Figure S9 GC spectrum of cycloaddition product 2b.



Figure S10 GC spectrum of cycloaddition product 3b.



Figure S11 GC spectrum of cycloaddition product 4b.



Figure S12 GC spectrum of cycloaddition product 5b.