Supporting Information

Five new rubidium borates with 0D cluster, 1D chain, 2D layer

and 3D framework

Qi-Ming Qiu, Ke-Ning Sun and Guo-Yu Yang*

MOE Key Laboratory of Cluster Science, School of Chemistry and Chemical Engineering, Beijing Institute of Technology, Beijing 100081, China. E-mail: ygy@bit.edu.cn; ygy@fjirsm.ac.cn, Fax: (+86)10-6891-8572

Table of Contents

Table S1 Selected bond distances (Å) and angles (°) for compound 1
Table S2 Selected bond distances (Å) and angles (°) for compound 2·····1
Table S3 Selected bond distances (Å) and angles (°) for compound 3·····2
Table S4 Selected bond distances (Å) and angles (°) for compound 4······4
Table S5 Selected bond distances (Å) and angles (°) for compound 5 5
Fig. S1 View of the supramolecular 2D structure of compound 1
Fig. S2 Structure of compounds 1 and 1a
Fig. S3 Asymmetric unit of compound 37
Fig. S4 3D supramolecular structure of compound 2 ······7
Fig. S5 Structure of compound 2a ·····7
Fig. S6 View of the Rb–O framework of 4 along the <i>b</i> axis ······ 8
Fig. S7 Structure of compounds 4, 4a and 4b ······8
Fig. S8 Structure of compound 59
Fig. S9 PXRD patterns show the comparison between the experimental value and calculated ones
for compounds 1(a), 2–3(b), 4(c) and 5(d)10
Fig. S10 TG curves for compounds 1(a), 2–3(b), 4(c) and 5(d)11
Fig. S11 IR spectra for compounds 1(a), 2–3(b), 4(c) and 5(d)12
Fig. S12 UV/Vis diffuse reflectance spectra of compounds 1(a), 2(b), 3(c), 4(d) and 5(e)13
Fig. S13 Oscilloscope traces of SHG signals for compounds 4 and LiNbO ₃ 14
Table S6 The direction and magnitude (in Debye) of the polyhedral dipole moments for
compound 1 ······14

Table S1 Selected bond distances (Å) and angles (°) for compound 1.

Rb—O1 ⁱⁱ	3.083 (5)	Na—O5 ^{vii}	2.380 (4)
Rb—O1 ^{iv}	3.128 (4)	Na—O5 ^{viii}	2.380 (4)
Rb—O2 ⁱⁱⁱ	3.110 (4)	B1—O1	1.371 (8)
Rb—O3	2.916 (4)	B1—O2	1.370 (7)
Rb—O3 ⁱ	2.976 (4)	B1—O3	1.369 (8)
Rb—O5	3.009 (4)	B2—O2 ^{ix}	1.489 (7)
Rb—O5 ^v	3.160 (4)	B2—O3	1.494 (7)
Na—O5	2.380 (4)	B2—O4	1.453 (7)
Na—O5 ⁱ	2.380 (4)	B2—O5	1.453 (7)
Na—O5 ^v	2.380 (4)	01—H1	0.8434
Na—O5 ^{vi}	2.380 (4)	O5—H5	0.8458
O3—B1—O2	122.6 (5)	O5—B2—O4	111.2 (5)
O3—B1—O1	118.4 (5)	O5—B2—O2 ^{ix}	109.6 (4)
O2—B1—O1	119.0 (5)	O4—B2—O2 ^{ix}	110.0 (5)
O2 ^{ix} —B2—O3	108.1 (5)	O5—B2—O3	108.1 (4)
O4—B2—O3	109.7 (4)		

Symmetry codes: (i) x - y, -y, -z + 1; (ii) -x + 1, -x + y + 1, -z + 1; (iii) -y + 1/3, x - y - 1/3, z - 1/3; (iv) -x + y + 1, -x + 1, z; (v) y, x, -z + 1; (vi) -x, -x + y, -z + 1; (vii) -x + y, -x, z; (viii) -y, x - y, z; (ix) -x + 2/3, -x + y + 1/3, -z + 4/3.

	Table S2 Selected bond distances ((A) and angles (°) for compount	nd 2 .
Rb1—O1 ⁱⁱⁱ	3.086 (3)	B1—O1	1.376 (5)
Rb1—O2	2.887 (2)	B1—O2	1.337 (4)
Rb1—O3 ⁱ	3.097 (2)	B1—O3	1.376 (4)
Rb1—O4 ⁱⁱ	2.930 (2)	B2—O2	1.464 (4)
Rb1—O6 ⁱ	2.895 (3)	B2—O4	1.453 (4)
Rb1—07 ⁱⁱ	3.242 (2)	B2—O5	1.486 (4)
Rb1—O9 ^{iv}	3.183 (3)	B2—O11 ^{ix}	1.474 (4)
Rb1-010 ^{iv}	2.986 (2)	B3—O3	1.401 (4)
Rb1—O11 ⁱⁱⁱ	2.941 (2)	B3—O5	1.350 (5)
Rb2—O1 ⁱ	3.021 (3)	B3—O6	1.345 (4)
Rb2—O3 ⁱⁱⁱ	3.326 (2)	B4—O4	1.350 (4)
Rb2—O4 ^{viii}	3.018 (2)	B4—O7	1.360 (4)
Rb2—O5	2.934 (2)	B4—O8	1.387 (4)
Rb2—O6 ⁱⁱⁱ	3.316 (3)	B5—O6 ^{vii}	1.475 (4)

Cable S2 Selected bond distances (Å) and angles (°) for compound 2.

Rb2—O7 ^{viii}	3.328 (2)	B5—O7	1.483 (4)
Rb2—O9 ^{vi}	2.928 (2)	B5—O9	1.444 (4)
Rb2—O10 ^{vii}	3.008 (2)	B5—O10	1.465 (4)
Rb2—O11 ⁱⁱⁱ	3.062 (2)	B6—O8	1.402 (4)
O1—H1	0.796 (19)	B6—O10	1.363 (4)
О9—Н9	0.78 (4)	B6—O11	1.338 (4)
O2—B1—O1	121.4 (3)	O4—B4—O7	117.1 (3)
O2—B1—O3	122.7 (3)	O4—B4—O8	123.0 (3)
O1—B1—O3	115.9 (3)	O7—B4—O8	119.8 (3)
O4—B2—O5	110.1 (3)	O9—B5—O10	111.6 (3)
O2—B2—O5	111.4 (3)	O9—B5—O6 ^{vii}	109.1 (3)
O11 ⁱⁱⁱ —B2—O5	107.1 (3)	O10—B5—O6 ^{vii}	111.9 (3)
O4—B2—O2	110.8 (3)	O9—B5—O7	105.6 (3)
O4—B2—O11 ⁱⁱⁱ	110.3 (3)	O10—B5—O7	111.3 (3)
O2—B2—O11 ⁱⁱⁱ	107.0 (3)	O6 ^{vii} —B5—O7	107.1 (3)
O6—B3—O5	125.3 (3)	O11—B6—O10	125.2 (3)
O6—B3—O3	114.0 (3)	O11—B6—O8	115.6 (3)
O5—B3—O3	120.7 (3)	O10—B6—O8	119.3 (3)

Symmetry codes: (i) -x + 1, y + 1/2, -z + 1/2; (ii) -x + 1, -y + 2, -z + 1; (iii) x, y + 1, z; (iv) -x + 1, -y + 1, -z + 1; (v) x - 1, -y + 3/2, z - 1/2; (vi) x, -y + 3/2, z - 1/2; (vii) -x + 2, -y + 1, -z + 1; (viii) -x + 2, -y + 2, -z + 1; (ix) x, y - 1, z.

Table S3 Selected bond distances (Å) and angles (°) for compound 3.

	Table 55 Selected bond distances	(A) and angles (¹) for compou	ind 3 .
Rb—O1	2.853 (6)	B1—O1	1.451 (10)
Rb—O2 ⁱⁱ	2.988 (5)	B1—O2	1.467 (10)
Rb—O3 ⁱⁱⁱ	3.318 (5)	B1—O3	1.487 (10)
Rb—O4 ^v	3.055 (5)	B1—O8 ^x	1.464 (10)
Rb—O6 ⁱⁱⁱ	3.020 (6)	B2—O2	1.340 (11)
Rb—O7 ⁱ	2.887 (5)	B2—O4	1.345 (11)
Rb—O7 ⁱⁱⁱ	3.357 (6)	B2—O5	1.402 (10)
Rb—O8 ^v	3.309 (6)	B3—O3	1.356 (11)
Rb—O9 ^v	3.302 (6)	B3—O5	1.390 (10)
Rb-011 ^{iv}	3.030 (7)	B3—O6	1.347 (10)
Ba—O1 ^{iv}	3.085 (6)	B4—O4 ^{ix}	1.463 (10)
Ba—O2 ^{iv}	2.914 (5)	B4—O6	1.446 (11)
Ba—O3 ^{viii}	3.230 (5)	B4—O7	1.497 (10)

Ba—O4 ^{vii}	2.837 (6)	B4—O10	1.466 (11)
Ba—O6 ^{viii}	2.864 (6)	В5—07	1.342 (11)
Ba—O8 ^{vi}	2.818 (6)	В5—О8	1.341 (11)
Ba—O9 ^{vi}	3.000 (6)	В5—О9	1.403 (11)
Ba—O10	2.804 (5)	В6—О9	1.365 (11)
Ba—O11 ^{vii}	3.006 (6)	B6—O10	1.347 (11)
O1—H1	0.8146	B6—O11	1.374 (11)
O11—H11	0.8190		
O1—B1—O8 ^x	109.2 (7)	O6—B4—O4 ^{vii}	111.0 (7)
O1—B1—O2	111.7 (6)	O6—B4—O10	111.1 (7)
O8 ^x —B1—O2	112.6 (6)	O4 ^{vii} —B4—O10	106.8 (7)
O1—B1—O3	105.2 (6)	O6—B4—O7	109.7 (7)
O8x—B1—O3	106.8 (6)	O4 ^{vii} —B4—O7	107.4 (6)
O2—B1—O3	111.1 (6)	O10—B4—O7	110.9 (7)
O2—B2—O4	125.4 (7)	O8—B5—O7	125.6 (8)
O2—B2—O5	119.9 (7)	O8—B5—O9	113.3 (8)
O4—B2—O5	114.7 (8)	O7—B5—O9	121.1 (8)
O6—B3—O3	117.5 (7)	O10—B6—O9	123.0 (8)
O6—B3—O5	123.0 (8)	O10—B6—O11	120.5 (8)
O3—B3—O5	119.5 (7)	O9—B6—O11	116.4 (7)

Symmetry codes: (i) x, -y + 1/2, z - 1/2; (ii) -x, y - 1/2, -z + 1/2; (iii) -x, y + 1/2, -z + 1/2; (iv) -x + 1, -y + 1, -z + 1; (v) x, -y + 3/2, z - 1/2; (vi) -x + 1, y - 1/2, -z + 3/2; (vii) x, y - 1, z; (viii) -x + 1, -y, -z + 1; (ix) x, y + 1, z; (x) -x, -y + 1, -z + 1.

Table S4 Selected bond distances (Å) and angles (°) for compound 4.

		· · · · · · · · · · · · · · · · · · ·	
Rb1—O1	3.141 (2)	B1—01	1.363 (4)
Rb1—O2	2.909 (2)	B1—O2	1.374 (5)
Rb1—O3 ^{vii}	2.889 (3)	B1—O3	1.361 (4)
Rb1—O5 ^{ix}	3.380 (3)	B2—O1 ^{xi}	1.463 (4)
Rb1—O6 ⁱ	2.736 (2)	B2—O2	1.467 (4)
Rb1—O6 ⁱⁱⁱ	2.841 (2)	B2—O4	1.492 (5)
Rb1—O8 ⁱⁱ	3.419 (3)	B2—O5	1.457 (5)
Rb1—O9 ^{viii}	3.056 (3)	B3—O3	1.483 (4)
Rb1—O9 ^x	3.479 (3)	B3—O5	1.431 (4)
Rb2—O1 ^v	3.304 (2)	B3—O6	1.496 (5)
Rb2—O2 ^v	2.919 (2)	B3—O7	1.514 (4)
Rb2—O3 ^{vi}	2.957 (3)	B4—O4 ⁱ	1.340 (5)
Rb2—O4	2.807 (3)	B4—O6	1.355 (5)
Rb2—O5	3.258 (3)	B4—O8	1.408 (5)
Rb2—07	3.067 (3)	B5—O7	1.347 (5)
Rb2—O7 ^{iv}	2.895 (2)	B5—O8	1.391 (5)
Rb2—O8 ⁱ	2.939 (3)	B5—O9	1.358 (5)
Rb2—O9	3.428 (3)	О9—Н9	0.818 (14)
O3—B1—O1	124.8 (3)	O3—B3—O7	106.7 (3)
O3—B1—O2	121.3 (3)	O5—B3—O7	112.6 (3)
O1—B1—O2	113.9 (3)	O6—B3—O7	106.3 (3)
O1 ^{xii} —B2—O4	107.9 (3)	O4 ⁱ —B4—O6	125.6 (3)
O1 ^{xii} —B2—O2	110.4 (3)	O4 ⁱ —B4—O8	115.5 (3)
O2—B2—O4	109.0 (3)	O6—B4—O8	118.9 (3)
O5—B2—O1 ^{xii}	110.9 (3)	O7—B5—O9	123.8 (3)
O5—B2—O2	110.8 (3)	O7—B5—O8	121.5 (3)
O5—B2—O4	107.7 (3)	O9—B5—O8	114.6 (3)
O3—B3—O6	108.6 (3)		

Symmetry codes: (i) -x + 3/2, y, z - 1/2; (ii) -x + 3/2, y - 1, z - 1/2; (iii) -x + 1, -y + 2, z - 1/2; (iv) x - 1/2, -y + 1, z; (v) x, y - 1, z; (vi) x + 1/2, -y + 1, z; (vii) -x + 1, -y + 2, z + 1/2; (viii) -x + 3/2, y + 1, z + 1/2; (ix) -x + 3/2, y, z + 1/2; (x) -x + 1, -y + 1, z + 1/2; (xi) x - 1/2, -y + 2, z; (xii) x + 1/2, -y + 2, z.

Table S5 Selected bond distances (Å) and angles (°) for compound 5.

Na—O1 ⁱ	2.4093 (12)	B1—O3	1.361 (2)
Na—O3 ⁱ	2.3203 (14)	B2—O1	1.4571 (19)
Rb—O1	3.5750 (12)	B2—O2 ⁱⁱ	1.4466 (19)
Rb—O2 ⁱⁱⁱ	2.9149 (11)	B2—O3 ^{iv}	1.452 (2)
B1—O1	1.378 (2)	B2—O4	1.5313 (16)
B1—O2	1.356 (2)		
O2—B1—O3	123.49 (15)	O2v—B2—O3 ^{vi}	112.63 (13)
O2—B1—O1	122.20 (14)	O2 ^v —B2—O1	109.59 (13)
O3—B1—O1	114.31 (16)	O3 ^{vi} —B2—O1	110.55 (12)
O1—B2—O4	107.88 (12)	O3 ^{vi} —B2—O4	107.25 (13)
O2 ^v —B2—O4	108.79 (11)		

Symmetry codes: (i) -x + 1, -y + 1, -z + 1; (ii) -z + 1/2, -x + 1, y - 1/2; (iii) -x + 1/2, -y + 1, z - 1/2; (iv) y - 1/2, z, -x + 1/2; (v) -y + 1, z + 1/2, -x + 1/2; (vi) -z + 1/2, x + 1/2, y.



Fig. S1 View of the supramolecular 2D structure of compound 1 along the *c* axis. Color code for atom: Na: blue, Rb: teal. Color code for polyhedra: BO₄: red, BO₃: green, similarly hereinafter.



Fig. S2 Structure of compounds **1** (a) and **1a** (b). Symmetry codes: (i) 1 + x - y, 2 - y, 1 - z. The arrangement of supramolecular structure of compounds **1** (c) and **1a** (d). View of the 0D Na–O–Rb cluster of **1** (e) and the 2D Na–O–Ca layer of **1a** (f).



Fig. S3 Asymmetric unit of compound 3. Symmetry codes: (i) x, -1 + y, z; (ii) -x, 1 - y, 1 - z.



Fig. S4 3D supramolecular structure of compound **2** along the *b* axis connected by O9–H9···O7 hydrogen bonds (2.354 Å, 3.041 Å, 148°, dashed lines) with the charge balancing Rb^+ ions.



Fig. S5 (a) 3D framework of compound 2a along the *c* axis. (b) 2D layer of 2a assuming O11 atoms are protonated. (c) 1D chain of 2a assuming O6 atoms are protonated.



Fig. S6 View of the Rb–O framework of 4 along the *b* axis.



Fig. S7 Asymmetric unit of compounds **4** (a), **4a** (b) and **4b** (c). The 2D layers with 3- and 9-MR in **4**(d), 8- and 10-MR in **4a**(e), 3- and 9-MR in **4b**(f). The arrangements of **4** (g), **4a** (h) and **4b** (i).



Fig. S8 (a) View of the structure of compound **5**, symmetry codes: (i) 0.5 - z, 1 - x, -0.5 + y; (ii) 1 - y, 0.5 + z, 0.5 - x; (iii) 0.5 - z, 0.5 + x, y; (iv) 1 - y, 0.5 + z, 0.5 - x. (b) View of the linkage of {B₆O₁₃} cluster units, in which the {B₆O₁₃} clusters are shown as pink nodes. (c) 3D framework viewed along the *a* axis.



Fig. S9 PXRD patterns show the comparison between the experimental value and calculated ones for compounds 1(a), 2–3(b), 4(c) and 5(d).



Fig. S10 TG curves for compounds 1(a), 2–3(b), 4(c) and 5(d).



Fig. S11 IR spectra for compounds 1(a), 2–3(b), 4(c) and 5(d).



Fig. S12 UV/Vis diffuse reflectance spectra of compounds 1(a), 2(b), 3(c), 4(d) and 5(e). Inset: plots of $\alpha/S vs. E$.



Fig. S13 Oscilloscope traces of SHG signals for compounds 4 and LiNbO₃ in the same particle size of 150–212 μ m.

Specie	а	b	С	Magnitude
Compound 1				
NaO_6	0×2	0×2	± 0.02	0.02
	0×4	0×4	0×4	0
	0	0	0	
RbO ₇	-4.84×3	4.75 × 3	-2.82×3	7.35
	-4.75×3	-9.59×3	-2.82×3	11.11
	-9.59×3	-4.75×3	2.82×3	11.11
	4.84×3	9.59 × 3	2.82×3	11.11
	4.75×3	-4.84×3	2.82×3	7.35
	9.59 × 3	4.84×3	-2.82×3	11.11
	0	0	0	
BO_3	-0.36×2	0.45×2	-0.29×2	0.65
	0.45×2	-0.09×2	-0.29×2	0.54
	-0.09×2	-0.36×2	-0.29×2	0.47
	-0.45×2	0.36×2	0.29×2	0.65
	0.36×2	0.09×2	0.29×2	0.47
	0.09×2	-0.45×2	0.29×2	0.54
	0.09	-0.45	0.33	0.56
	0.36	0.09	0.33	0.50
	-0.45	0.36	0.33	0.66
	-0.36	0.45	-0.33	0.66
	0.45	-0.09	-0.33	0.56
	-0.09	-0.36	-0.33	0.50
	0	0	0	
Unit Cell	0	0	0	0

Table S6 The direction and magnitude (in Debye) of the polyhedral dipole moments for compound $\mathbf{1}$