

Electronic Supporting Information (ESI)

Template-free formation of one-dimensional mesoporous ZnMn₂O₄ tube-in-tube nanofibers towards Lithium-ion batteries as anode materials

Jingxuan Wei,[†] Senyang Xu,[†] Zhaolin Tan, Linrui Hou,* Changzhou Yuan*

School of Materials Science & Engineering, University of Jinan, Jinan, 250022, P.R.

China

Corresponding authors:

E-mail: mse_houlr@ujn.edu.cn (*Prof. L.R. Hou*)

mse_yuancz@ujn.edu.cn; ayuancz@163.com (*Prof. C.Z. Yuan*)

[†] These authors equally contributed to the work.

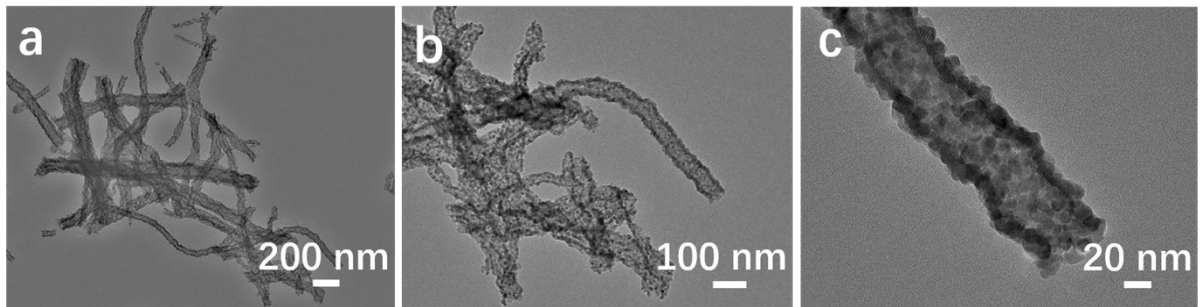


Fig. S1 (a-c) TEM images of the ZMO NTs.

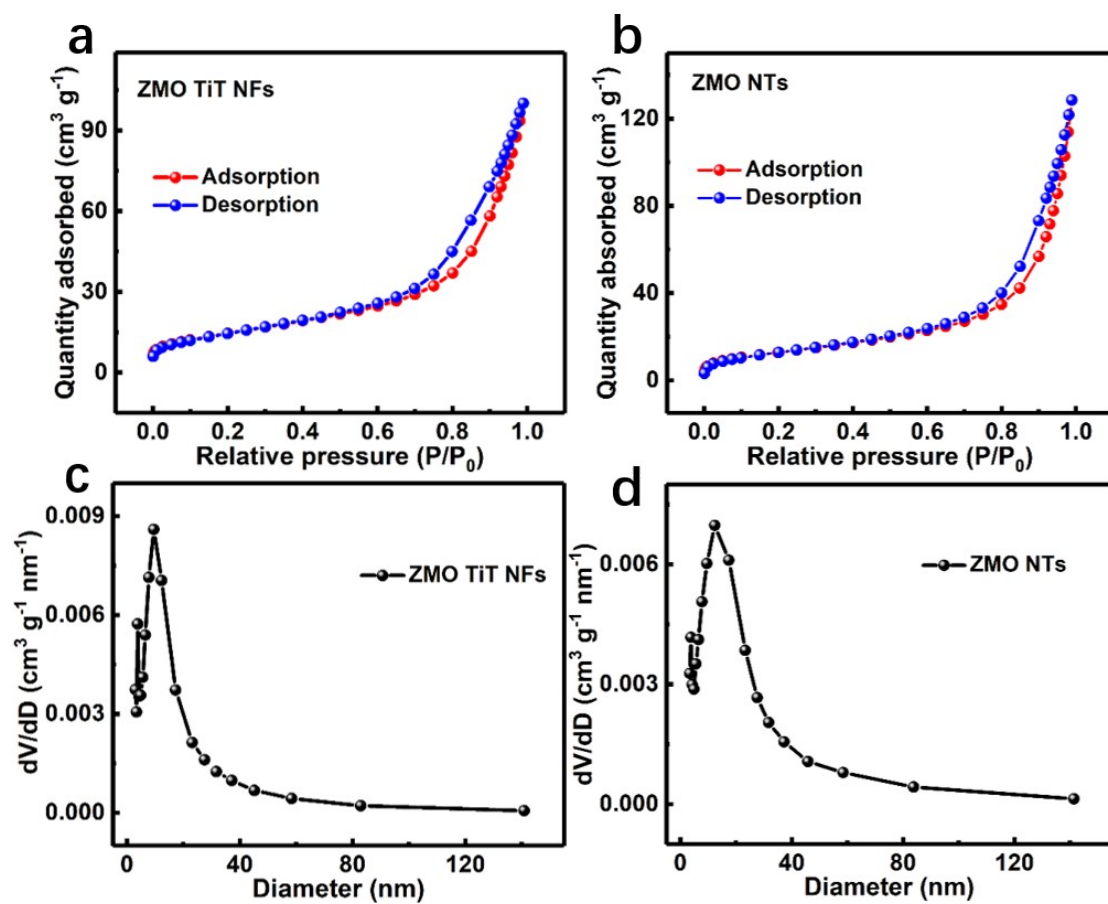


Fig. S2 (a, b) Nitrogen adsorption/desorption isotherms and (c, d) pore-size distribution plots of (a, c) ZMO TiT NFs and (b, d) ZMO NTs.

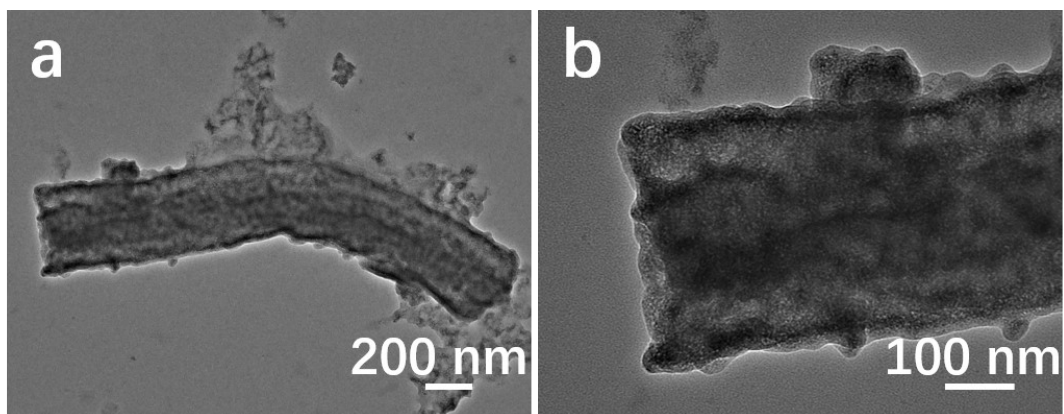


Fig. S3 (a, b) TEM images of the ZMO TiT NFs electrode after 300 cycles at 0.5 A g⁻¹.

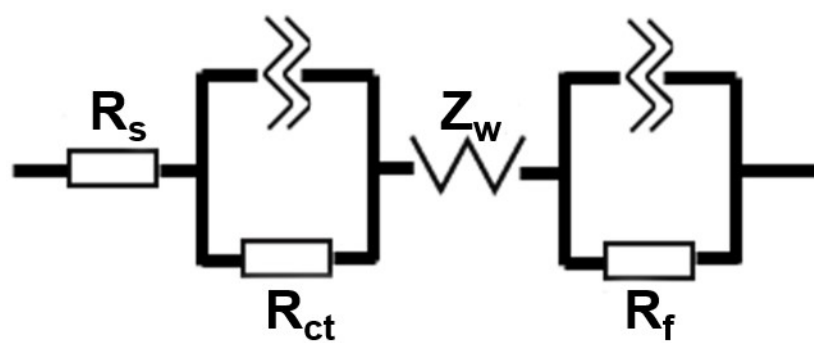


Fig. S4 Equivalent circuit model for Nyquist plots in Fig. 4f.

Table S1 Corresponding EIS fitted data for the ZMO TiT NFs and ZMO NTs

Electrode	R_s (ohm)	R_{ct} (ohm)	R_f (ohm)
ZMO TiT NFs	1.3	151.7	4.1
ZMO NTs	1.6	204.6	6.2

Table S2 Comparison in cycling performance of the ZMO TiT NFs with other ZMO-based anodes

Sample	Current density (mA g ⁻¹)	Cycle number	Reversible capacity (mAh g ⁻¹)	Ref.
ZMO TiT NFs	500 2000	200 1000	938.9 564.6	This work
ZnMn ₂ O ₄ microtubules	200 500	300 300	750.4 535.3	S1
Nano-ZnMn ₂ O ₄	100 1000	90 1200	716 500	S2
Yolk-shell MnO@ZnMn ₂ O ₄ /N-C nanorods	1000	200	595	S3
ZnMn ₂ O ₄ micro-belts	500 1000	150 350	731 372	S4
ZnMn ₂ O ₄ nano-peanuts	2000	200	516	S5
MWCNT/ZnMn ₂ O ₄	1600	1000	527	S6
Porous ZnMn ₂ O ₄ /biocarbon microsphere	500 1000	150 650	820 550	S7
Core-shell ZnMn ₂ O ₄ nanosheets@carbon nanotubes	50 1000	100 200	803 595	S8
Loaf-like ZnMn ₂ O ₄	500	100	517	S9
ZnMn ₂ O ₄ hollow microspheres	400	100	607	S10
ZnMn ₂ O ₄ microspheres	100	100	602	S11
ZnMn ₂ O ₄ /AC	100	50	714	S12
ZnMn ₂ O ₄ nanoparticles	200	160	745	S13
ZnMn ₂ O ₄ /N-doped graphene	500	200	747	S14

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