Supplementary Materials

Cobalt(II) coordination polymers constructed from bis(N-pyrid-3ylmethyl)adipoamide and polycarboxylic acids: reversible structural transformation upon proton delivery and removal

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Fig. S1. (a) Simulated and (b) experimental PXRD patterns of 1.



Fig. S2. (a) Simulated and (b) experimental PXRD patterns of 2.



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Fig. S3. (a) Simulated and (b) experimental PXRD patterns of 3.



Fig. S4. (a) Simulated and (b) experimental PXRD patterns of 4.



Complex	1	2	3	4
^a N-HO	2.23 (175)	2.48 (138)	2.16 (175)	2.01 (176)
	~ /		2.40 (141)	
^b N-HO		2.10 (151)		
		2.07 (156)		
°N_H0		2.07 (150)		2/11(112)
N-II0				2.71(112)
				2.30(136)
"N-HO				2.29 (158)
°O-HO	2.38 (170)	1.90 (147)	1.83 (151)	
		1.82 (162)	2.43 (123)	
		2.00 (130)	1.79 (165)	
fO-HO	2.02 (177)	1.98 (156)	1.98 (177)	
	()	2 00 (166)		
80-H0	2 21 (152)	1.85(173)		2 56 (157)
	2.21(102)	1.03(175)		2.50 (157)
	2.00 (122)	2.02(103)		
ⁿ O-HO	1.92 (171)	1.97 (161)		1.88 (162)
		2.28 (160)		
ⁱ O-HO	1.80 (165)			
^j O-HO		1.87 (157)		
		2.16 (129)		
		1 96 (143)		
ko u o		1.90(143)		
О		2.09(174)		
'О-НО		2.44 (131)		
		1.69 (169)		

Table S1. H---O distances (Å) of hydrogen bonds with angles (°) in parenthesis for 1 - 4.

a from the amine hydrogen atoms of L to the carboxylate oxygen atoms;

b from the amine hydrogen atoms of L to the cocrystallized water oxygen atoms;

c from the amine hydrogen atoms of 5-NH₂-IPA²⁻ to the carboxylate oxygen atoms;

d from the amine hydrogen atoms of 5-NH₂-IPA²⁻ to the cocrystallized water oxygen atoms;

e from the coordinated water molecules to the carboxylate oxygen atoms

f from the coordinated water molecules to the amide oxygen atoms

g from the cocrystallized water molecules to the carboxylate oxygen atoms

h from the cocrystallized water molecules to the amide oxygen atoms

i from the carboxylate oxygen atoms to the cocrystallized water molecules

j from the coordinated water molecules to the cocrystallized water molecules

k from the cocrystallized water molecules to the coordinated water molecules

1 from the cocrystallized water molecules to the cocrystallized water molecules

Fig. S5. Hydrogen bonds of complex 1.



Fig. S6. Hydrogen bonds of complex 2.



Fig. S7. Hydrogen bonds of complex 3.



Fig. S8. Hydrogen bonds of complex 4.



Fig. S9. The high resolution XPS spectra of Co 2p regions of (a) $Co(CH_3COO)_2 \cdot 4H_2O$, (b) complex 1 and (c) complex 2.



Complex	Weight loss of solvent °C (calc/found), %	Weight loss of ligand °C (calc/found), %
1	2 H ₂ O 130 ~ 175 (7.92/7.45)	0.5 L + 1,3,5-HBTC ²⁻ 275 - 740 (79.57/79.81)
2	8 H ₂ O ~ 120 (15.51/15.31)	1.5 L + 1,3,5-BTC ³⁻ 250 - 600 (74.90/73.25)
3	$\begin{array}{c} 2 \ \mathrm{H_2O} \\ 105 \sim 190 \\ (5.61/5.56) \end{array}$	L + 5-tert-IPA ²⁻ 275 - 695 (84.86/84.68)
4	H ₂ O ~ 200 (4.29/2.56)	$L + 5-NH_2-IPA^{2-}$ 380 - 700 (81.65/82.58)

 Table S2. Thermal properties of complexes 1 - 4.

Fig. S10. The TGA curve for complex 1.



Fig. S11. The TGA curve for complex 2.



Fig. S12. The TGA curve for complex 3.



Fig. S13. The TGA curve for complex 4.



Fig. S14. PXRD patterns of complex **1** immersed in aqueous solutions with different pH values.



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Fig. S15. PXRD patterns of complex **2** immersed in aqueous solutions with different pH values.



Fig. S16. PXRD patterns of complex **1**. (a) simulation, (b) as-synthesized and (c) heated at 175 °C for 1 h.



Fig. S17. PXRD patterns of complex **2**. (a) simulation, (b) as-synthesized and (c) heated at 120 °C for 1 h.



Fig. S18. PXRD patterns of complex **1**. (a) simulation, (b) as-synthesized, (c) heated at 175 °C for 1 h and then immersed in (d) water, (e) MeOH, (f) EtOH, (g) THF and (h) DMF.



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Fig. S19. PXRD patterns of complex **2**. (a) simulation, (b) as-synthesized, (c) heated at 120 °C for 1 h and then immersed in (d) water, (e) MeOH, (f) EtOH, (g) THF and (h) DMF.



Fig. S20. PXRD patterns of complex **1** immersed in a variety of organic solvents.



Fig. S21. PXRD patterns of complex **2** immersed in a variety of organic solvents.



Fig. S22. PXRD patterns for **2**. (a) simulation, (b) as-synthesized, (c) heated in water at 120 °C for 2 days and (d) simulation of **1**.



Fig. S23. PXRD patterns of **2**. (a) simulation, (b) as-synthesized and heated at 120 °C for 2 days in HCl solutions with pH values of (c) 1, (d) 3 and (e) 5 and (f) simulation of **1**.



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