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Crystal polymorphism and crystalline-state photochromism of a rhodium dithionite complex with *n*-methoxypropyl moieties

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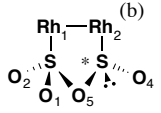
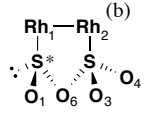
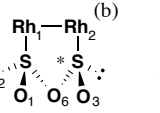
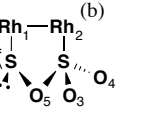
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Table S1 Percentage population of the isomers, **1**^{MPro} and **2a–d**^{MPro}, in the β -crystal^(a)

	1 ^{MPro}					2 ^{MPro} (total)
		2a ^{MPro} (R)	2b ^{MPro} (R)	2c ^{MPro} (S)	2d ^{MPro} (S)	
Sample 1	100	0	0	0	0	0
Sample 2	35	24	41	0	0	65
Sample 3	16	31	53	0	0	84
Sample 4	5	21	74	0	0	95
Sample 5	31	14	55	0	0	69
Sample 6	79	0	21	0	0	21
Sample 7	100	0	0	0	0	0

(a) All the data, except for the data of sample 1 and 7, have $\pm 4\%$ errors based on the errors of the experimental occupancy factors of the oxygen atoms. Although the crystal has mirror images of **2a–d**^{MPro}, as a set, only one mirror image in the crystal is considered in this treatment (*Angew. Chem., Int. Ed.*, 2006, **45**, 6473; *J. Am. Chem. Soc.*, 2008, **130**, 17836). (b) The four stereoisomers, **2a–d**^{MPro}, concerned with the μ -O₂SOSO unit. The Cp^{MPro} and μ -CH₂ ligands are omitted for clarity. The absolute configurations of the sulfur atoms are shown in parentheses.

The values of % for stereoisomers **2a–d**^{MPro} were calculated from the simultaneous equations based on the occupancy of the oxygen atoms determined by X-ray diffraction analysis. In the case of sample 2, the equations were as follows:

$$1.00 \text{ (occupancy of O}_1\text{)} = \mathbf{2a}^{\text{MPro}} + \mathbf{2b}^{\text{MPro}} + \mathbf{2c}^{\text{MPro}} + \mathbf{1}^{\text{MPro}}$$

$$0.59 \text{ (occupancy of O}_2\text{)} = \mathbf{2a}^{\text{MPro}} + \mathbf{2c}^{\text{MPro}} + \mathbf{2d}^{\text{MPro}} + \mathbf{1}^{\text{MPro}}$$

$$0.76 \text{ (occupancy of O}_3\text{)} = \mathbf{2b}^{\text{MPro}} + \mathbf{2c}^{\text{MPro}} + \mathbf{2d}^{\text{MPro}} + \mathbf{1}^{\text{MPro}}$$

$$1.00 \text{ (occupancy of O}_4\text{)} = \mathbf{2a}^{\text{MPro}} + \mathbf{2b}^{\text{MPro}} + \mathbf{2d}^{\text{MPro}} + \mathbf{1}^{\text{MPro}}$$

$$0.24 \text{ (occupancy of O}_5\text{)} = \mathbf{2a}^{\text{MPro}} + \mathbf{2d}^{\text{MPro}}$$

$$0.41 \text{ (occupancy of O}_6\text{)} = \mathbf{2b}^{\text{MPro}} + \mathbf{2c}^{\text{MPro}}$$

$$\mathbf{2a}^{\text{MPro}} = 0.240, \mathbf{2b}^{\text{MPro}} = 0.410, \mathbf{2c}^{\text{MPro}} = 0.000, \mathbf{2d}^{\text{MPro}} = 0.000, \mathbf{1}^{\text{MPro}} = 0.350.$$

Figures

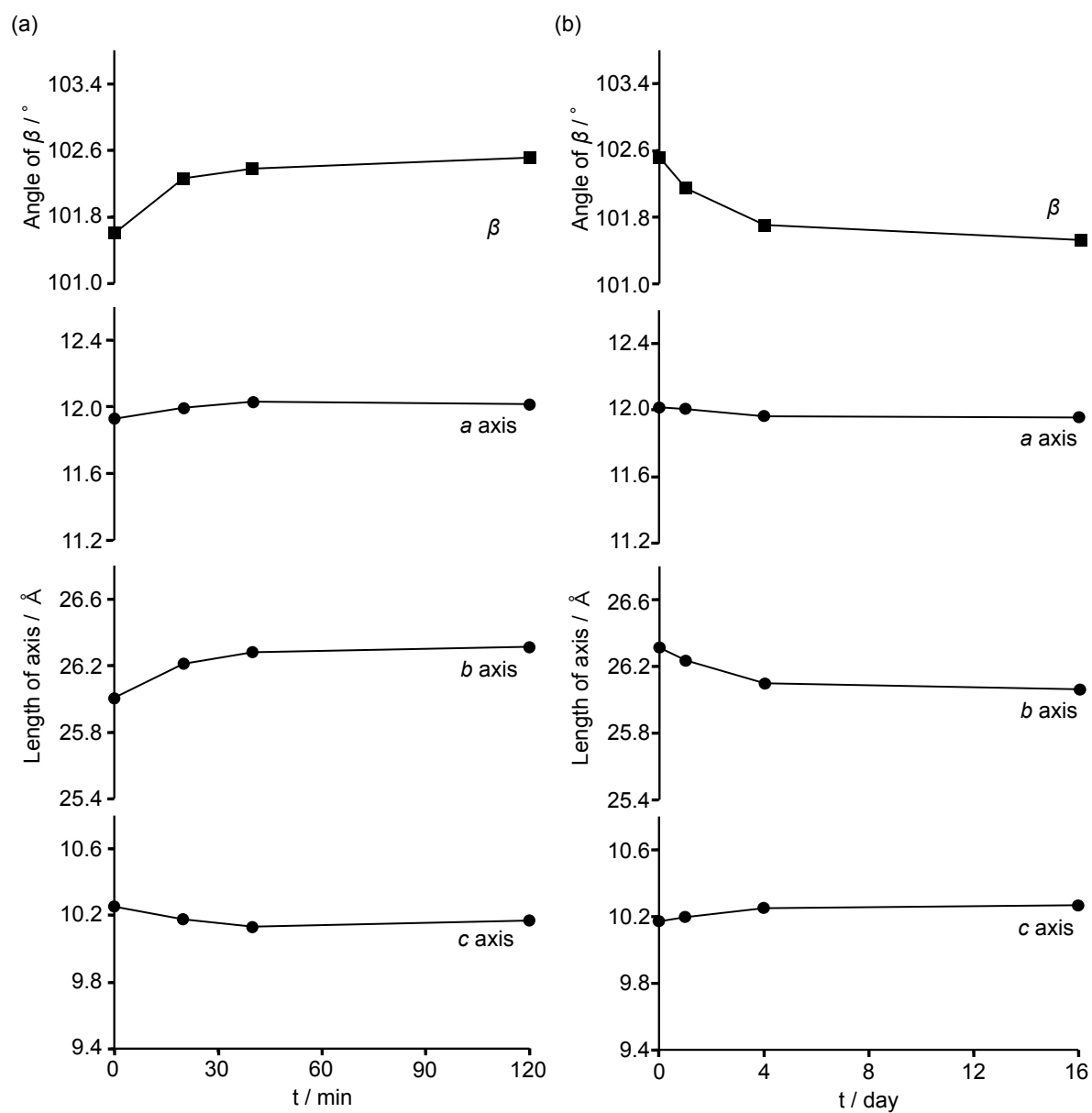


Fig. S1 (a) Changes in the unit cell parameters (β angle and *a*, *b*, and *c* axes) with (a) irradiation (samples 1–4, 0 min data: sample 1) and (b) heating time (samples 4–7, 0 day data: sample 4). X-ray diffraction data were recorded at 23 °C. Crystal size: 0.25 x 0.10 x 0.05 mm³.

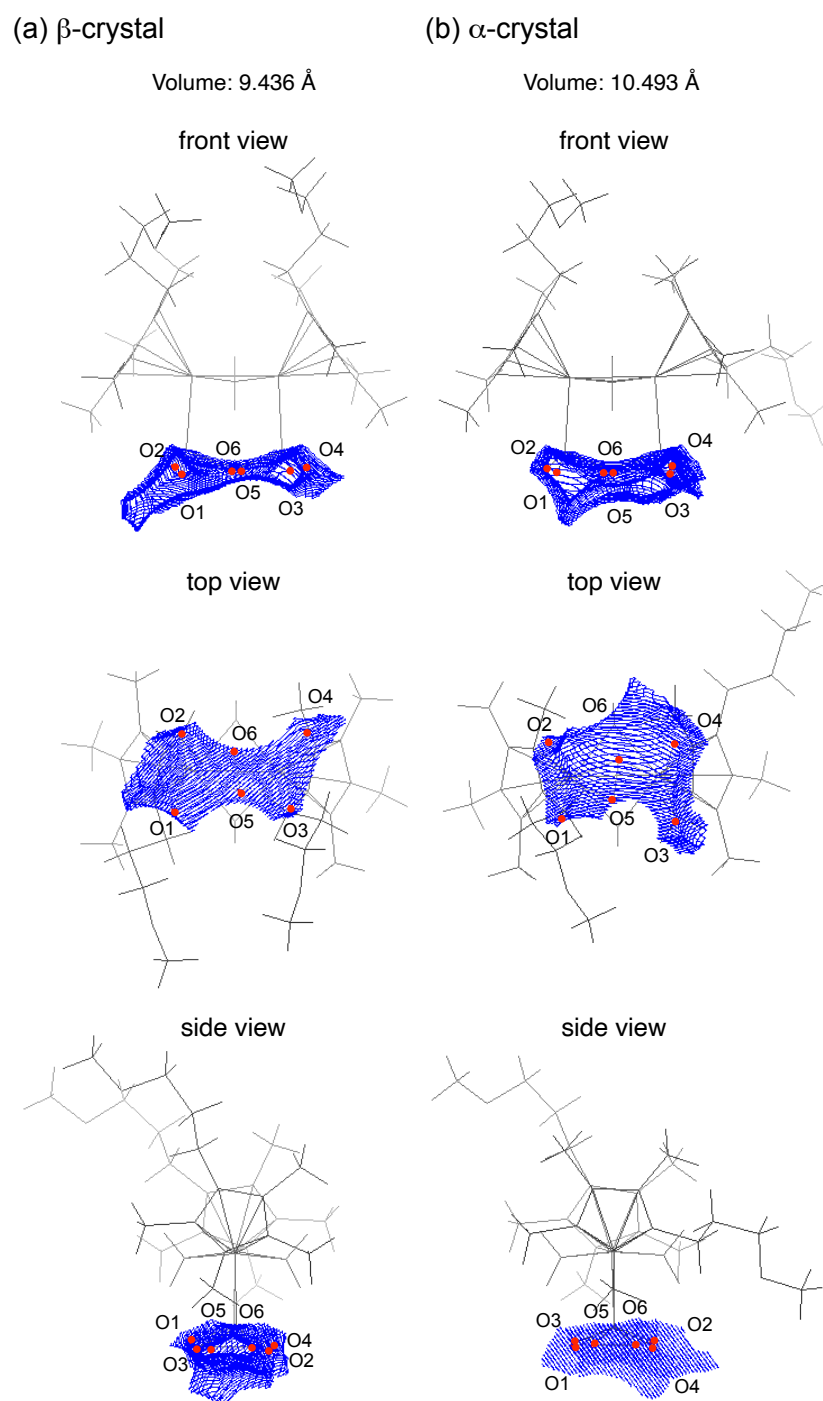


Fig. S2 Cavities (front, top and side views) in sample 4 (2^{MPro}) of the (a) β - and (b) α -crystals. The contours are drawn in sections separated by 0.10 Å.

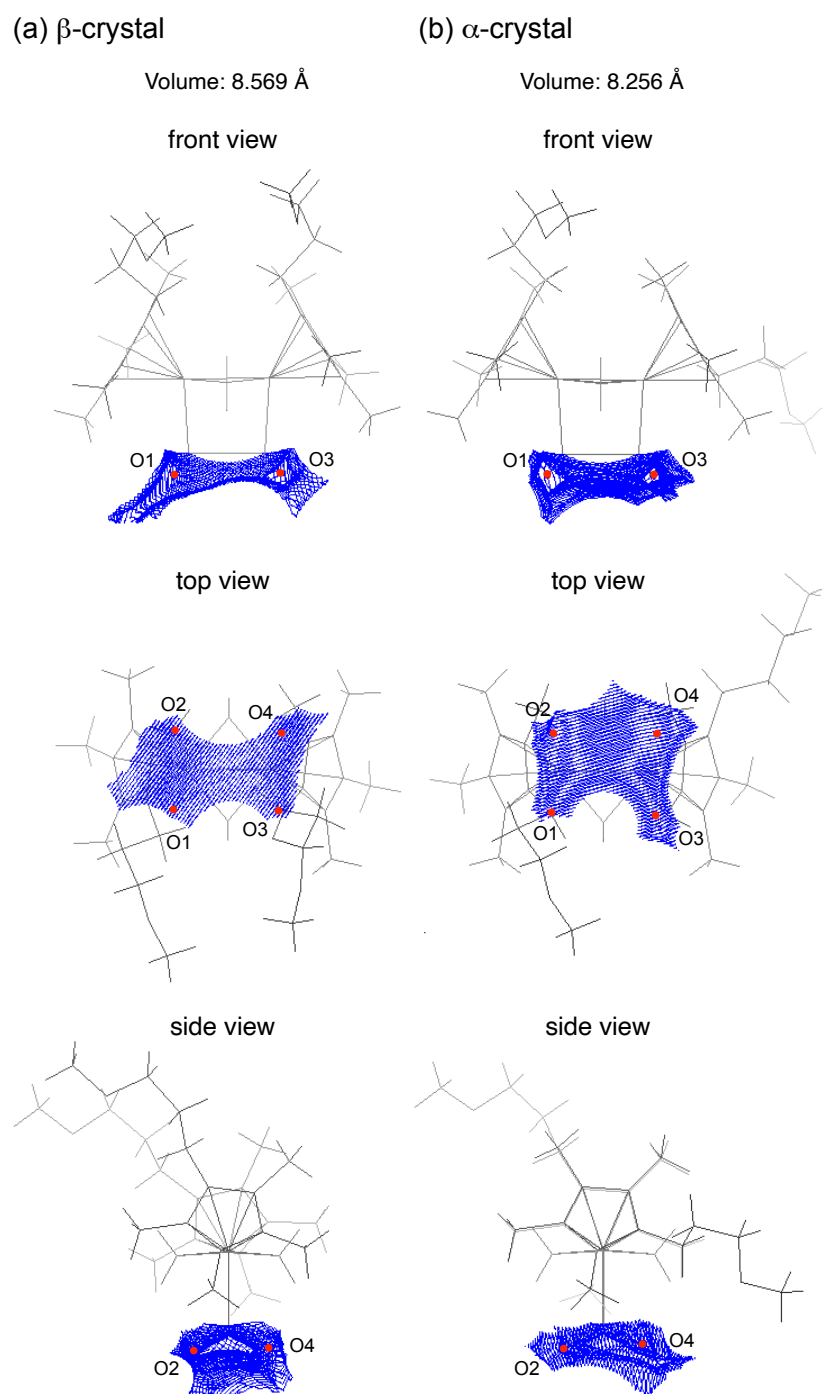


Fig. S3 Cavities (front, top and side views) in sample 1 (1^{MPro}) of the (a) β - and (b) α -crystals. The contours are drawn in sections separated by 0.10 Å.

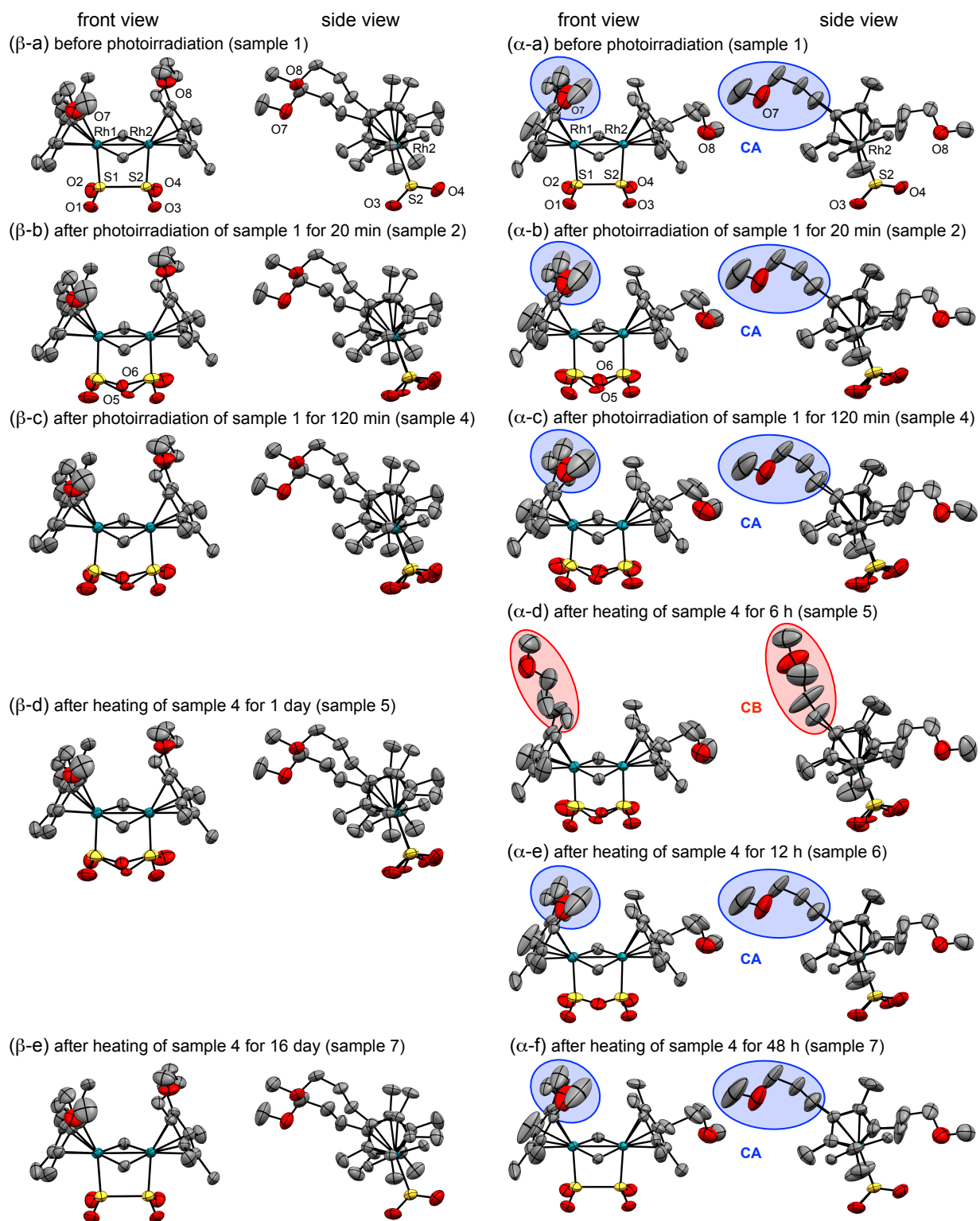


Fig. S4 ORTEP drawings of 1^{MPro} (50% probability ellipsoids, front and side views) in samples of (β -a–e) the β -crystal and (α -a–f) α -crystal. The hydrogen atoms are omitted for clarity. Data of α -crystal: *Dalton Trans.*, 2022, **51**, 48.