

Supporting Information for: Molecular-Scale Origins of Solution Nanostructure and Excess Thermodynamic Properties in a Water/Amphiphile Mixture

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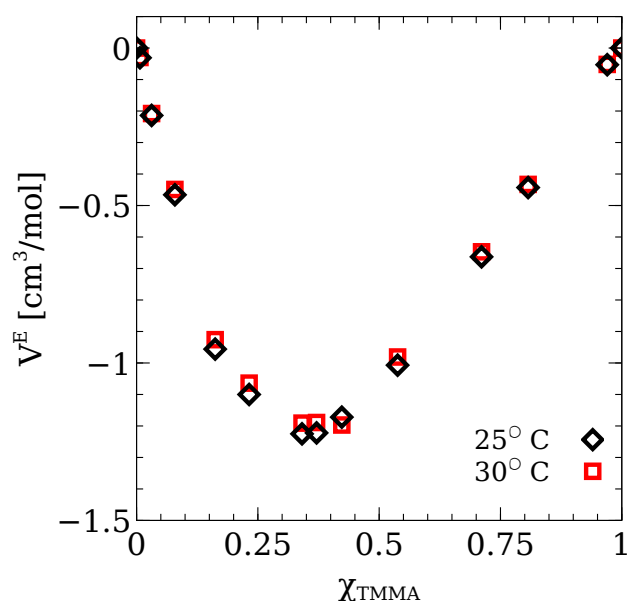


Figure S1: Comparison of experimental percent excess volume for 25° and 30° C.

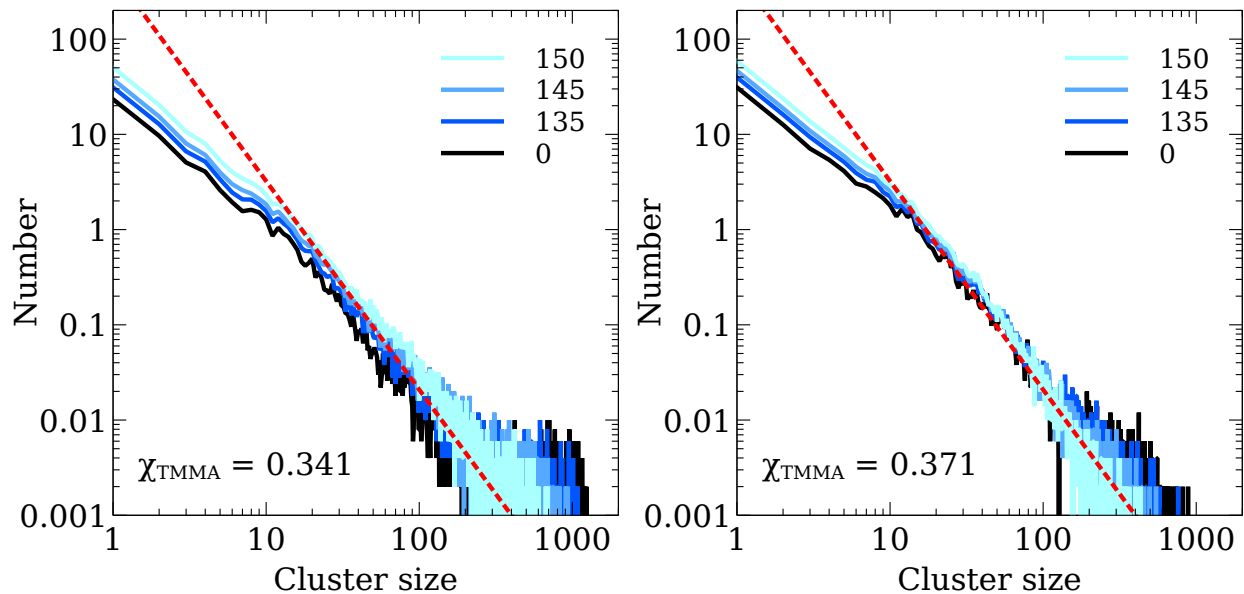


Figure S2: Water hydrogen bonded cluster distributions for $\chi_{\text{TMMA}} = 0.341$ (left) and $\chi_{\text{TMMA}} = 0.371$ (right) for different O-H...O angle cutoffs.

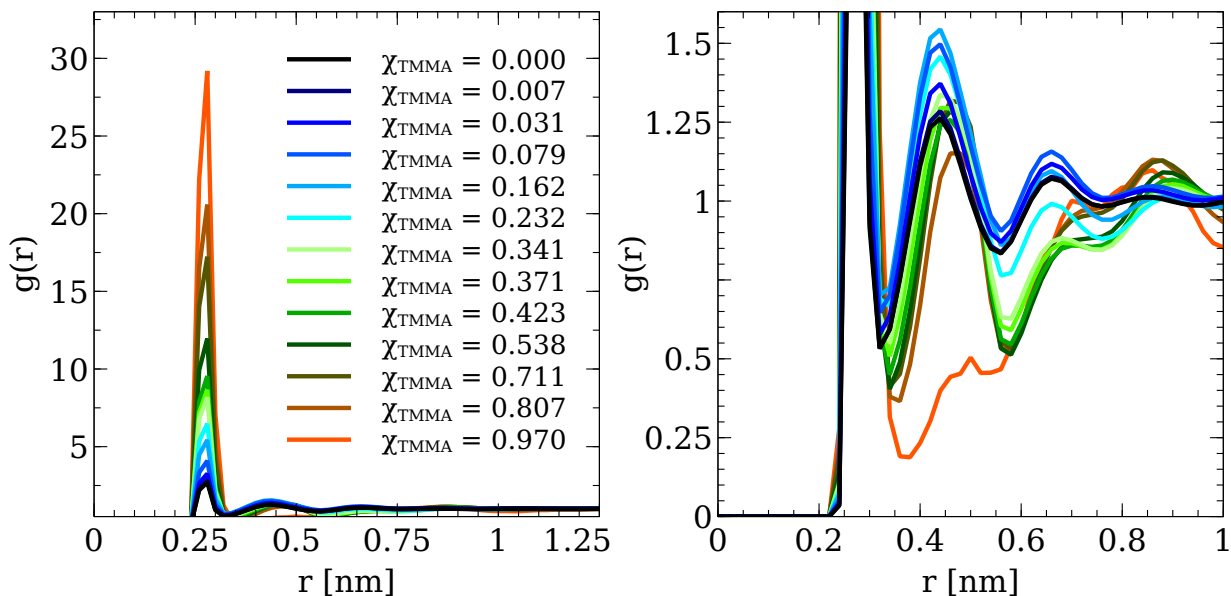


Figure S3: Water oxygen-oxygen RDFs for each simulation. Data for compositions with $\chi_{\text{TMMA}} \geq 0.423$ are extended by 100 ns for improved statistical accuracy.

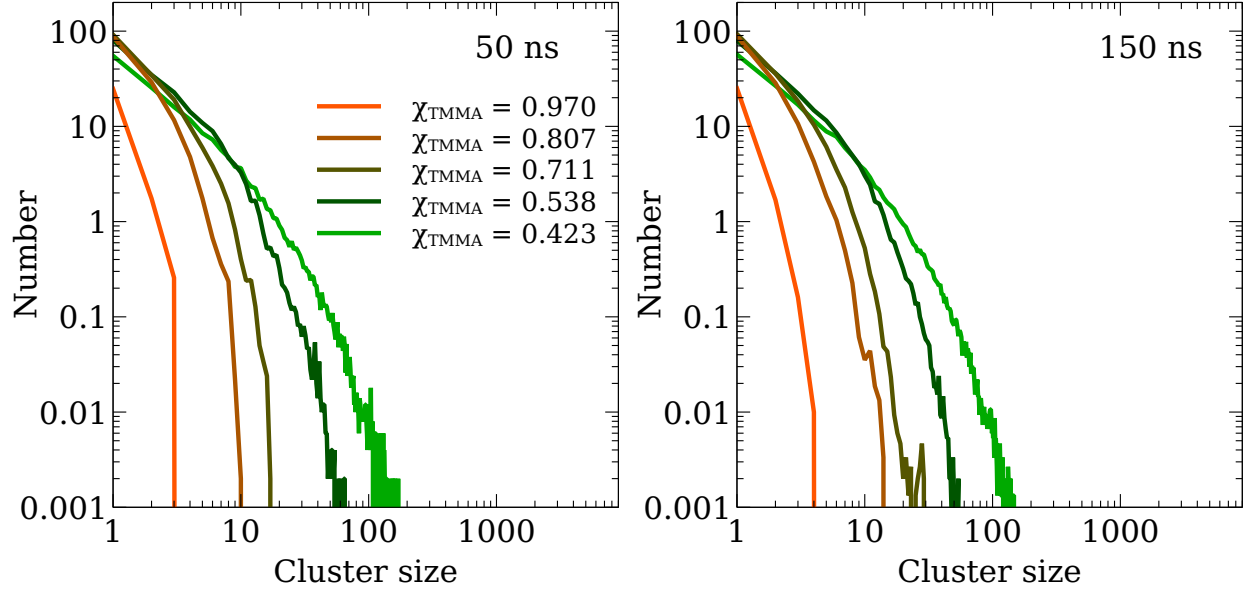


Figure S4: The water cluster size distributions for the TMMA-rich compositions are plotted for the 50 ns production trajectory (left panel) and for extended trajectories with an additional 100 ns (right panel).

The high-energy incident photons provide an adequate q range both for normalizing intensities and sufficient peak resolution in $g(r)$ after Fourier transform. The total scattering function, $S(q)$, was obtained from the experimental data using weighting factors determined based on relative atomic concentrations and appropriate form factors, as described in detail elsewhere.¹⁻³ The total experimental pair correlation functions, $g(r)$, was obtained from $S(q)$ by Fourier transform,

$$g(r) = 1 + \frac{1}{2\pi^2\rho_0 r} \int q(S(q) - 1) \sin(qr) dq, \quad (1)$$

where ρ_0 is the average atomic density, and plotted in Figure S2.

Scattering intensities, $I(q)$, are computed as described by Walter et al.⁴ for the periodic simulation box with

$$I(q) = \sum_i^N \sum_{i \neq j}^N f_i(q) f_j(q) e^{-iqr_{ij}}, \quad (2)$$

where r_{ij} is the distance between atoms i and j and $f_i(q)$ is the atomic form factor for atom

i determined by Cromer-Mann expansions.⁵ Intensities were normalized by a scalar factor for comparison to experimental data.

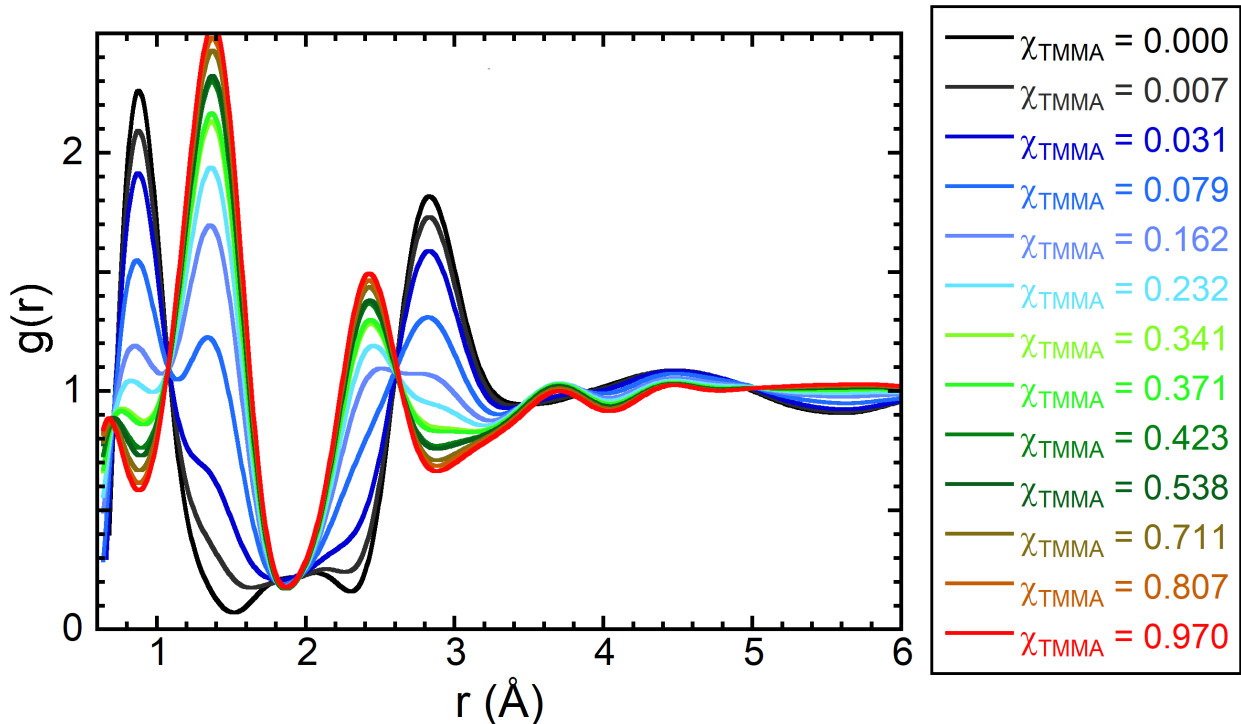


Figure S5: Pair distribution function generated by the Fourier transform of the experimental X-ray scattering data.

Simulation Lennard-Jones parameters for TMMA are provided in Table S1, with mixed terms derived using Lorentz-Berthelot mixing rules. The bonded potential parameters for TMMA are given in the GROMACS .itp topology file at the end of this document.

Table 1: Lennard-Jones parameters for TMMA atoms.

Atom name	σ_{ii} [nm]	ϵ_{ii} [kJ/mol]
H1	0.2422	0.0870
HC	0.2600	0.0870
CT	0.3398	0.4510
C	0.3315	0.4134
N	0.3181	0.6845
O	0.3048	0.6121

References

- (1) Soderholm, L.; Skanthakumar, S.; Gorman-Lewis, D.; Jensen, M.; Nagy, K. Characterizing solution and solid-phase amorphous uranyl silicates. *Geochimica et Cosmochimica Acta* **2008**, *72*, 140–150.
- (2) Egami, T.; Billinge, S. *Underneath the Bragg Peaks: Structural Analysis of Complex Materials*; Pergamon Materials Series; Elsevier Science, 2003.
- (3) Als-Nielsen, J.; McMorrow, D. *Elements of Modern X-ray Physics*; Wiley, 2011.
- (4) Walter, N. P.; Jaiswal, A.; Cai, Z.; Zhang, Y. LiquidLib: A comprehensive toolbox for analyzing classical and ab initio molecular dynamics simulations of liquids and liquid-like matter with applications to neutron scattering experiments. *Computer Physics Communications* **2018**, *228*, 209–218.
- (5) Cromer, D. T.; Mann, J. B. X-ray scattering factors computed from numerical Hartree–Fock wave functions. *Acta Crystallographica Section A: Crystal Physics, Diffraction, Theoretical and General Crystallography* **1968**, *24*, 321–324.

; gaff2 TMMAL

[moleculetype]

; Name nrexcl

TMMAL 3

[atoms]

;AT.NUM	TYPE	RESID	RESNAME	PDB-NAME	IGRP	CHRG	MASS
1	CT	1	TMMAL	C1	1	0.0185	12.0100
2	H1	1	TMMAL	H1	1	0.0481	1.0080

3	H1	1	TMMAL	H2	1	0.0481	1.0080
4	H1	1	TMMAL	H3	1	0.0481	1.0080
5	CT	1	TMMAL	C2	2	0.0061	12.0100
6	H1	1	TMMAL	H4	2	0.0519	1.0080
7	H1	1	TMMAL	H5	2	0.0519	1.0080
8	H1	1	TMMAL	H6	2	0.0519	1.0080
9	CT	1	TMMAL	C3	3	-0.2247	12.0100
10	HC	1	TMMAL	H7	3	0.0942	1.0080
11	HC	1	TMMAL	H8	3	0.0942	1.0080
12	CT	1	TMMAL	C4	4	0.0178	12.0100
13	H1	1	TMMAL	H9	4	0.0482	1.0080
14	H1	1	TMMAL	H10	4	0.0482	1.0080
15	H1	1	TMMAL	H11	4	0.0482	1.0080
16	CT	1	TMMAL	C5	5	0.0061	12.0100
17	H1	1	TMMAL	H12	5	0.0518	1.0080
18	H1	1	TMMAL	H13	5	0.0518	1.0080
19	H1	1	TMMAL	H14	5	0.0518	1.0080
20	N	1	TMMAL	N1	6	-0.2418	14.0100
21	C	1	TMMAL	C6	6	0.5072	12.0100
22	O	1	TMMAL	O1	6	-0.5715	16.0000
23	C	1	TMMAL	C7	7	0.5072	12.0100
24	O	1	TMMAL	O2	7	-0.5715	16.0000
25	N	1	TMMAL	N2	7	-0.2418	14.0100

[bonds]

;	I1	I2	TYPE	r0(nm)	K(kJ/nm ² /mol)
	1	2	1	0.1097	157284.929

1	3	1	0.1097	157284.929
1	20	1	0.1462	110361.368
1	4	1	0.1097	157284.929
5	6	1	0.1097	157284.929
5	7	1	0.1097	157284.929
5	8	1	0.1097	157284.929
5	20	1	0.1462	110361.368
9	10	1	0.1097	157284.929
9	21	1	0.1524	101763.248
9	23	1	0.1524	101763.248
9	11	1	0.1097	157284.929
12	13	1	0.1097	157284.929
12	14	1	0.1097	157284.929
12	25	1	0.1462	110361.368
12	15	1	0.1097	157284.929
16	17	1	0.1097	157284.929
16	18	1	0.1097	157284.929
16	19	1	0.1097	157284.929
16	25	1	0.1462	110361.368
20	21	1	0.1379	149038.265
21	22	1	0.1218	273035.289
23	24	1	0.1218	273035.289
23	25	1	0.1379	149038.265

[angles]

;	I1	I2	I3	TYPE	theta(deg)	K(kJ/rad ² /mol)
	1	20	5	1	115.64	271.45792

1	20	21	1	120.69	273.01437
2	1	3	1	108.46	162.34757
2	1	20	1	108.88	257.50010
2	1	4	1	108.46	162.34757
3	1	20	1	108.88	257.50010
3	1	4	1	108.46	162.34757
4	1	20	1	108.88	257.50010
5	20	21	1	120.69	273.01437
6	5	7	1	108.46	162.34757
6	5	8	1	108.46	162.34757
6	5	20	1	108.88	257.50010
7	5	8	1	108.46	162.34757
7	5	20	1	108.88	257.50010
8	5	20	1	108.88	257.50010
9	21	20	1	115.18	352.56895
9	21	22	1	123.20	353.76557
9	23	24	1	123.20	353.76557
9	23	25	1	115.18	352.56895
10	9	21	1	108.77	198.36762
10	9	23	1	108.77	198.36762
10	9	11	1	107.58	163.00864
11	9	21	1	108.77	198.36762
11	9	23	1	108.77	198.36762
12	25	16	1	115.64	271.45792
12	25	23	1	120.69	273.01437
13	12	14	1	108.46	162.34757
13	12	25	1	108.88	257.50010

13	12	15	1	108.46	162.34757
14	12	25	1	108.88	257.50010
14	12	15	1	108.46	162.34757
15	12	25	1	108.88	257.50010
16	25	23	1	120.69	273.01437
17	16	18	1	108.46	162.34757
17	16	19	1	108.46	162.34757
17	16	25	1	108.88	257.50010
18	16	19	1	108.46	162.34757
18	16	25	1	108.88	257.50010
19	16	25	1	108.88	257.50010
20	21	22	1	123.05	476.18523
21	9	23	1	111.63	273.73402
24	23	25	1	123.05	476.18523

[pairs]

;	I1	I4
	1	6
	1	7
	1	8
	1	9
	1	22
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	4	5

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[dihedrals]

;	I1	I2	I3	I4	TYPE	theta(deg)	K(kJ/rad ² /mol)	N
	1	20	5	6	9	0.0000	0.0000	2
	1	20	5	7	9	0.0000	0.0000	2
	1	20	5	8	9	0.0000	0.0000	2
	1	20	21	9	9	180.0000	1.0878	2
	1	20	21	9	9	0.0000	2.0920	1
	1	20	21	22	9	180.0000	10.4600	2
	2	1	20	5	9	0.0000	0.0000	2
	2	1	20	21	9	0.0000	0.0000	2
	3	1	20	5	9	0.0000	0.0000	2
	3	1	20	21	9	0.0000	0.0000	2
	4	1	20	5	9	0.0000	0.0000	2
	4	1	20	21	9	0.0000	0.0000	2
	5	20	21	9	9	180.0000	1.0878	2
	5	20	21	9	9	0.0000	2.0920	1
	5	20	21	22	9	180.0000	10.4600	2
	6	5	20	21	9	0.0000	0.0000	2
	7	5	20	21	9	0.0000	0.0000	2

8	5	20	21	9	0.0000	0.0000	2
9	23	25	12	9	180.0000	1.0878	2
9	23	25	12	9	0.0000	2.0920	1
9	23	25	16	9	180.0000	1.0878	2
9	23	25	16	9	0.0000	2.0920	1
10	9	21	20	9	0.0000	0.0000	2
10	9	21	22	9	0.0000	3.4727	1
10	9	21	22	9	180.0000	0.1674	3
10	9	23	24	9	0.0000	3.4727	1
10	9	23	24	9	180.0000	0.1674	3
10	9	23	25	9	0.0000	0.0000	2
11	9	21	20	9	0.0000	0.0000	2
11	9	21	22	9	0.0000	3.4727	1
11	9	21	22	9	180.0000	0.1674	3
11	9	23	24	9	0.0000	3.4727	1
11	9	23	24	9	180.0000	0.1674	3
11	9	23	25	9	0.0000	0.0000	2
12	25	16	17	9	0.0000	0.0000	2
12	25	16	18	9	0.0000	0.0000	2
12	25	16	19	9	0.0000	0.0000	2
12	25	23	24	9	180.0000	10.4600	2
13	12	25	16	9	0.0000	0.0000	2
13	12	25	23	9	0.0000	0.0000	2
14	12	25	16	9	0.0000	0.0000	2
14	12	25	23	9	0.0000	0.0000	2
15	12	25	16	9	0.0000	0.0000	2
15	12	25	23	9	0.0000	0.0000	2

16	25	23	24	9	180.0000	10.4600	2
17	16	25	23	9	0.0000	0.0000	2
18	16	25	23	9	0.0000	0.0000	2
19	16	25	23	9	0.0000	0.0000	2
20	21	9	23	9	0.0000	0.0000	2
21	9	23	24	9	0.0000	5.6902	1
21	9	23	24	9	180.0000	0.7531	3
21	9	23	25	9	0.0000	0.0000	2
22	21	9	23	9	0.0000	5.6902	1
22	21	9	23	9	180.0000	0.7531	3
20	5	1	21	4	180.0000	4.6024	2
21	20	9	22	4	180.0000	43.9320	2
23	24	9	25	4	180.0000	43.9320	2
25	16	12	23	4	180.0000	4.6024	2

[system]

TMMAL

[molecules]

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