

**Supplementary Information**

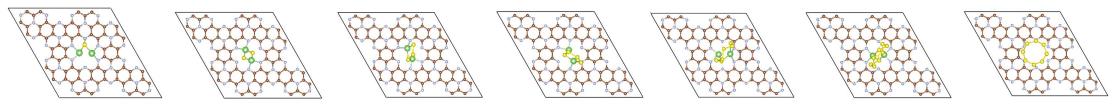
**C<sub>9</sub>N<sub>4</sub> and C<sub>2</sub>N<sub>6</sub>S<sub>3</sub> Monolayers as Promising Anchoring Materials for Lithium-Sulfur Batteries: Weakening the Shuttle Effects *via* Optimizing Lithium Bonds**

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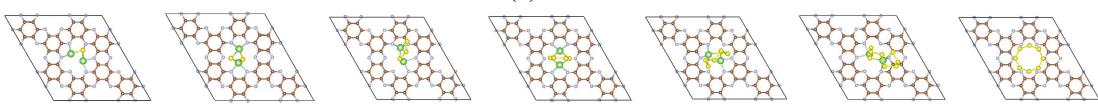
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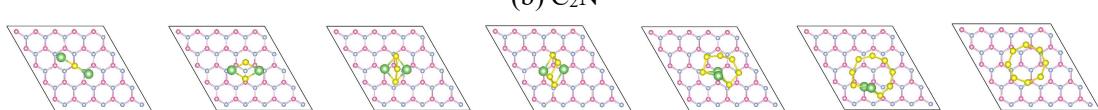
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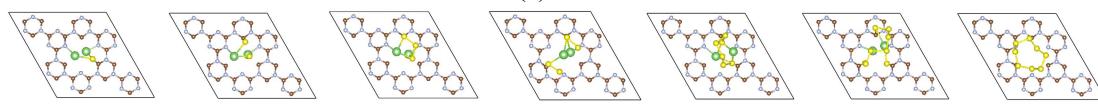
(a)  $\text{C}_9\text{N}_4$



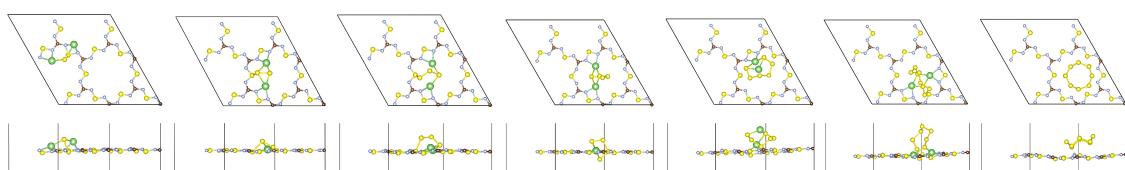
(b)  $\text{C}_2\text{N}$



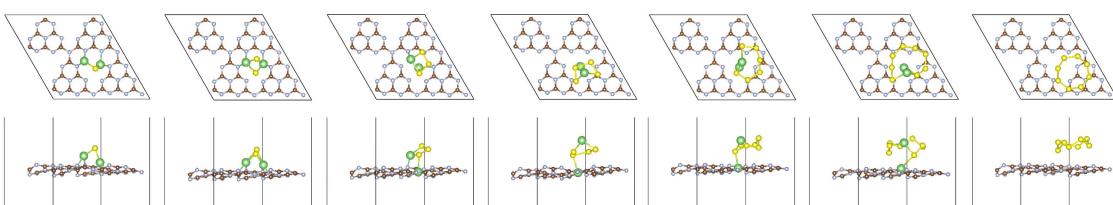
(c) BN



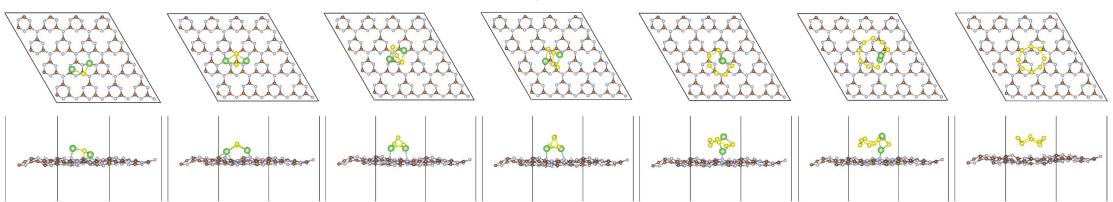
(d) CTF



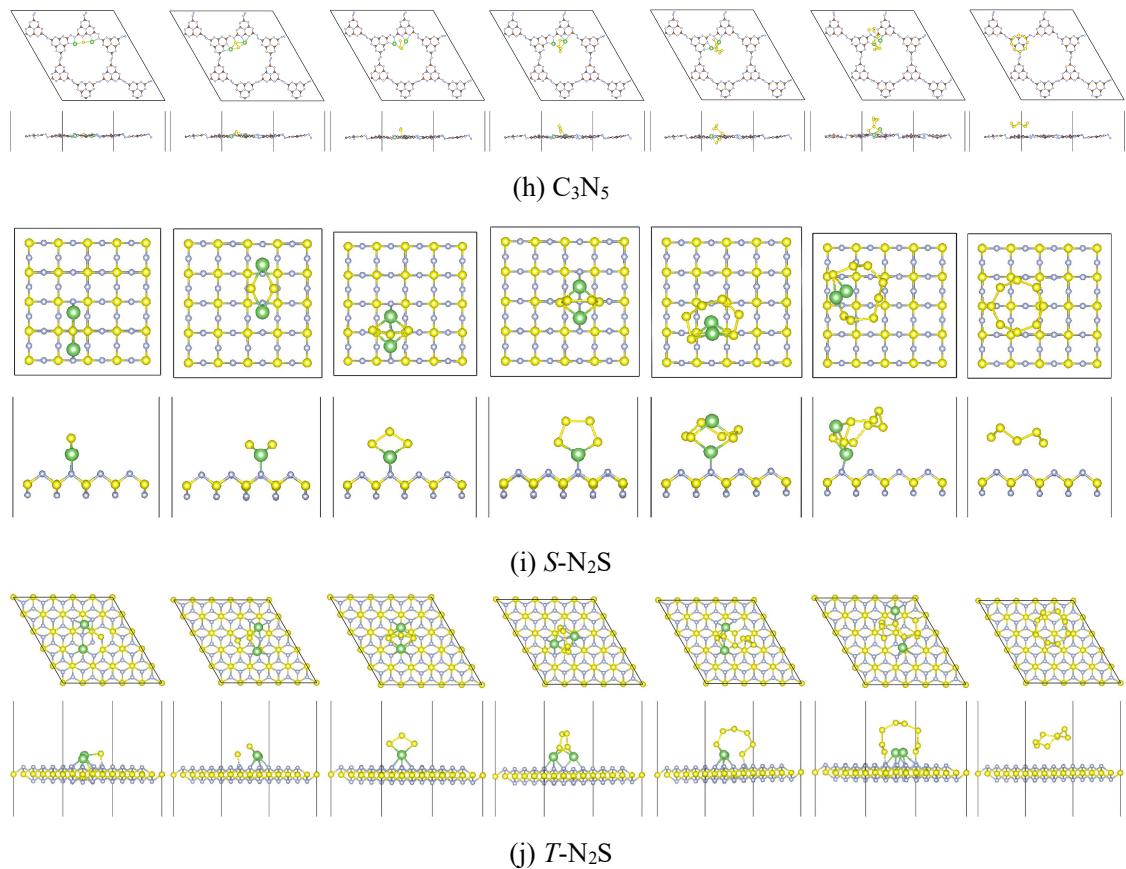
(e)  $\text{C}_2\text{N}_6\text{S}_3$



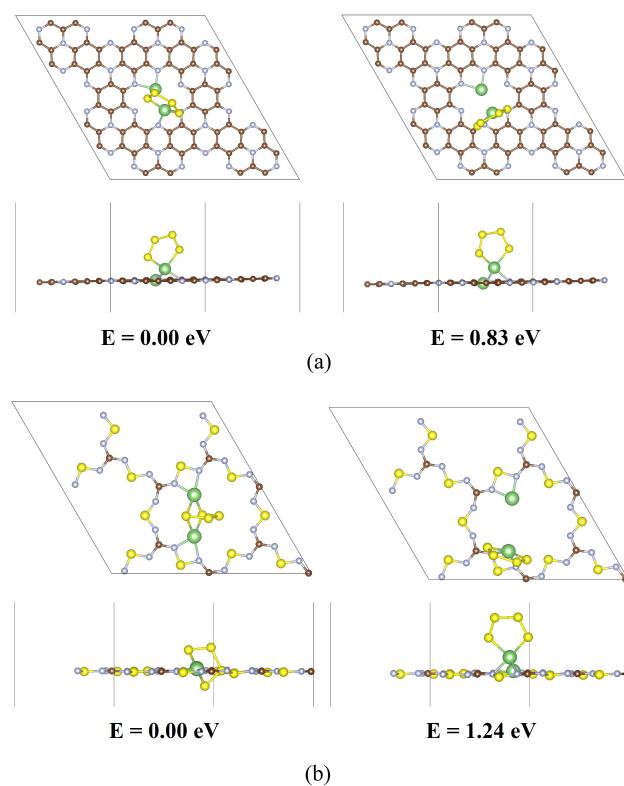
(f)  $g\text{-}\text{C}_3\text{N}_4$



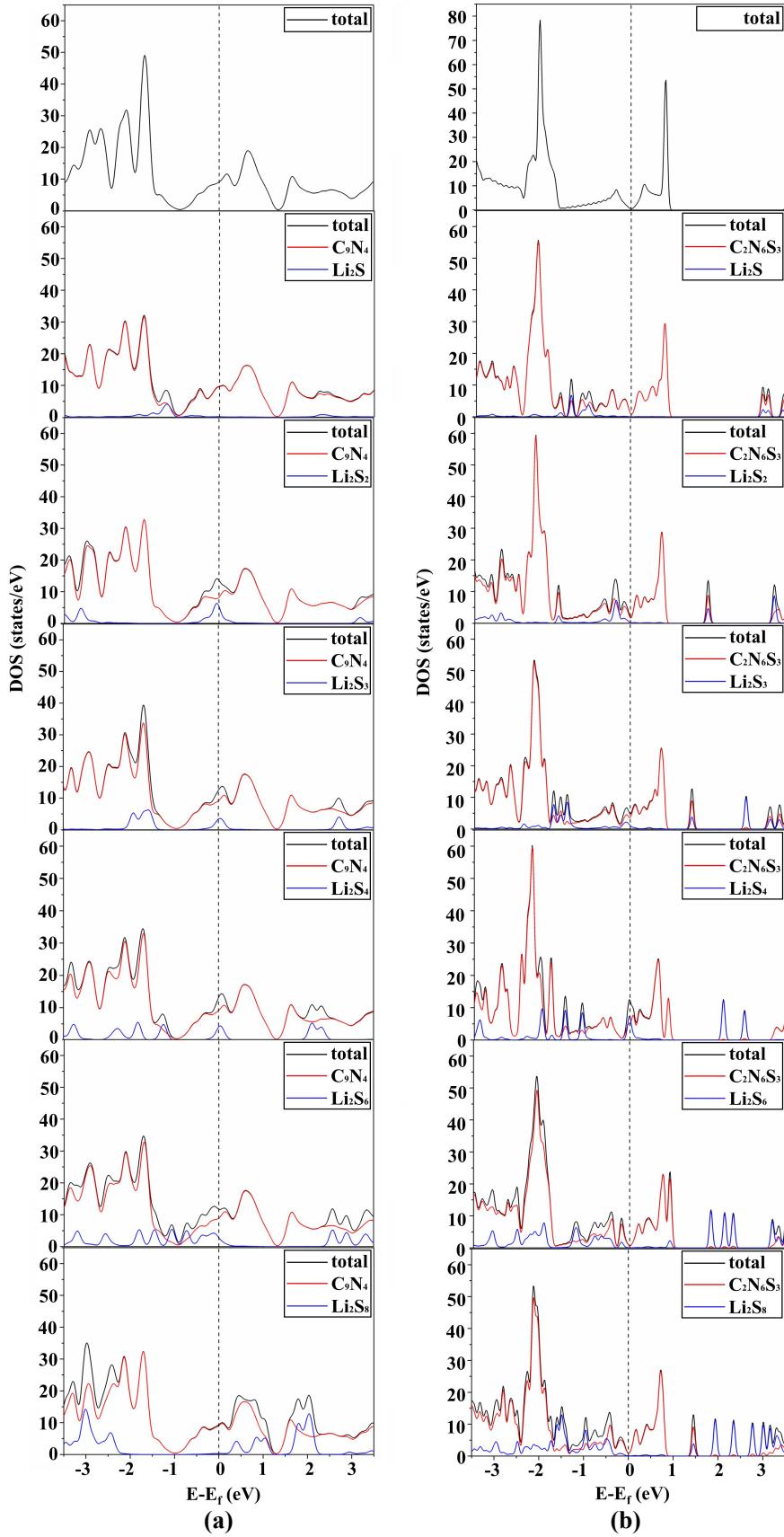
(g)  $p\text{-}\text{C}_3\text{N}_4$



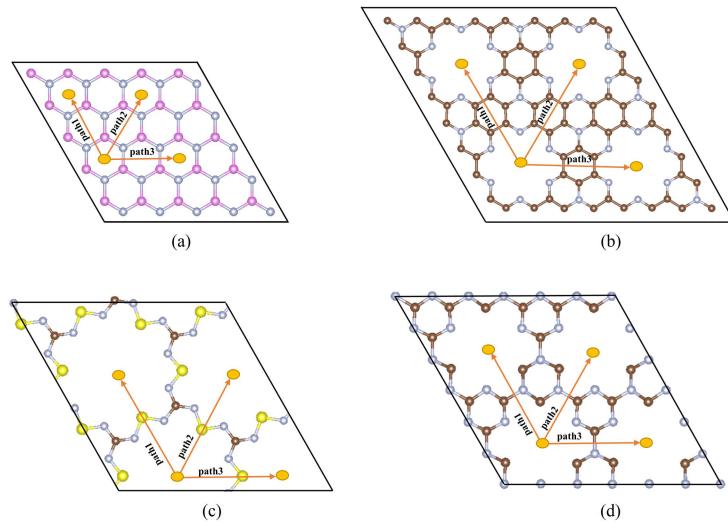
**Fig. S1** The most stable configurations of  $\text{Li}_2\text{S}_n$  and  $\text{S}_8$  on the ten examined 2D AMs.



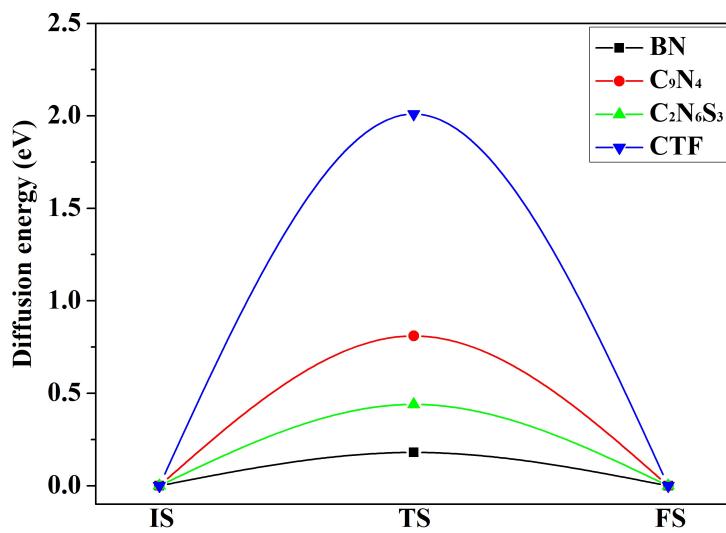
**Fig. S2** Top and sides views of the “intact” and decomposed  $\text{Li}_2\text{S}_4$  on the  $\text{C}_9\text{N}_4$  (a) and  $\text{C}_2\text{N}_6\text{S}_3$  (b), as well as their relative energies.



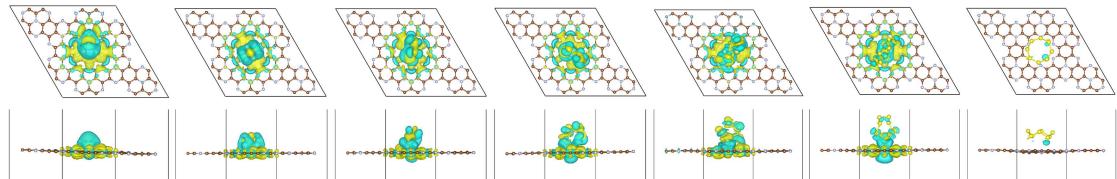
**Fig. S3** Density of states for  $\text{C}_9\text{N}_4$  (a) and  $\text{C}_2\text{N}_6\text{S}_3$  (b) before and after adsorption of  $\text{Li}_2\text{S}_n$  ( $n = 1, 2, 3, 4, 6, 8$ ).



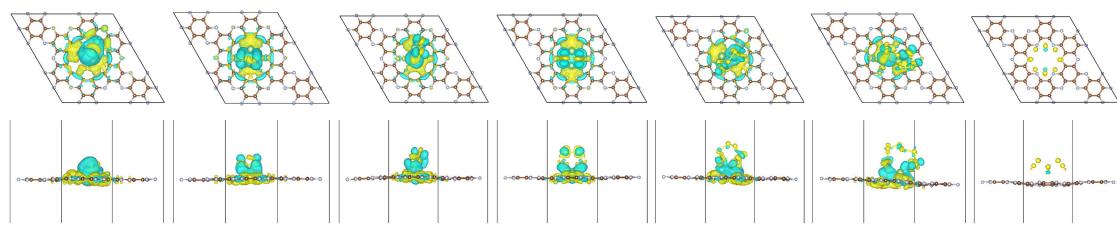
**Fig. S4** Diffusion paths for  $\text{Li}_2\text{S}_6$  on the BN (a)  $\text{C}_9\text{N}_4$  (b)  $\text{C}_2\text{N}_6\text{S}_3$  (c) and CTF (d) monolayers.



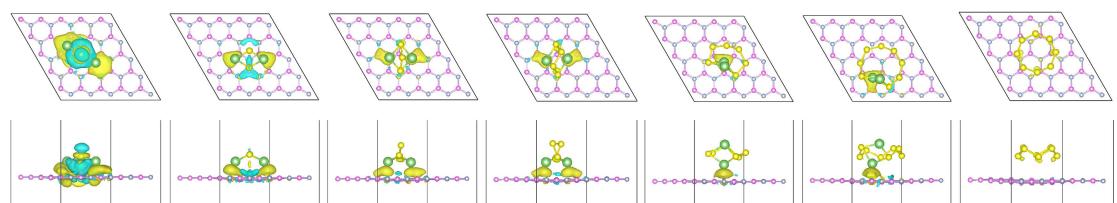
**Fig. S5** Migration energy barrier of  $\text{Li}_2\text{S}_6$  on BN,  $\text{C}_9\text{N}_4$ ,  $\text{C}_2\text{N}_6\text{S}_3$  and CTF.



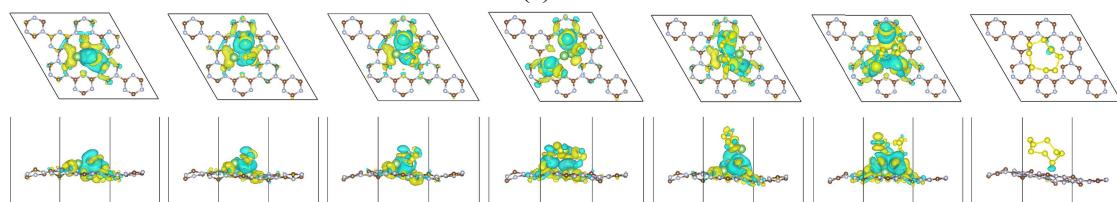
(a)  $C_9N_4$



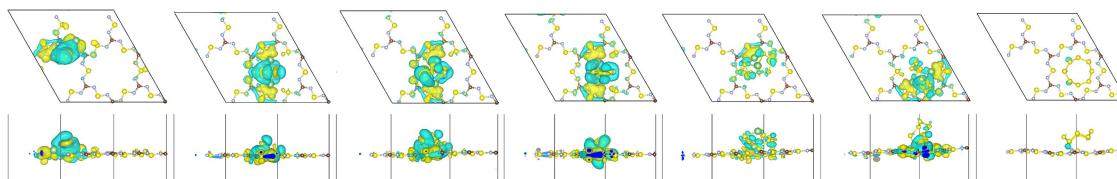
(b)  $C_2N$



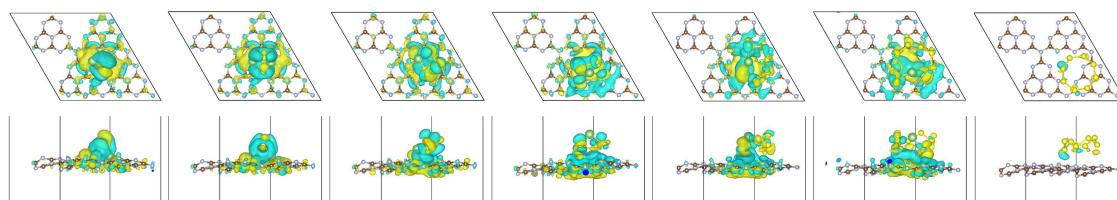
(c) BN



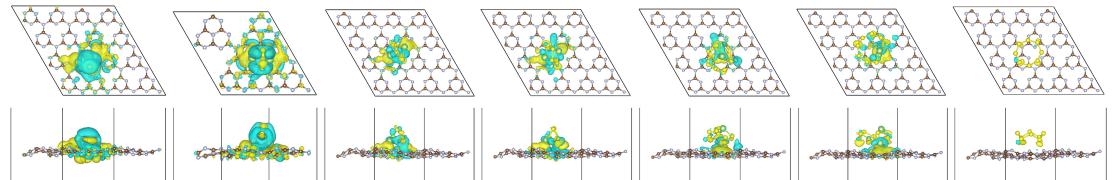
(d) CTF



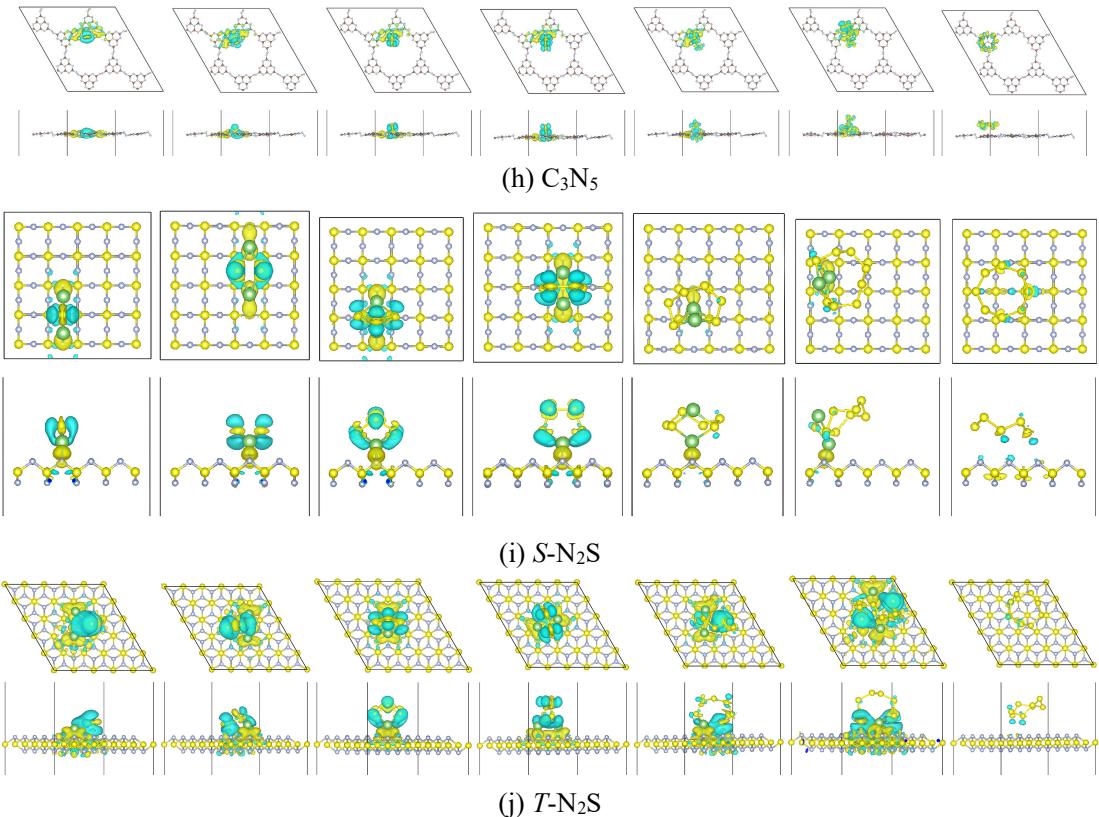
(e)  $C_2N_6S_3$



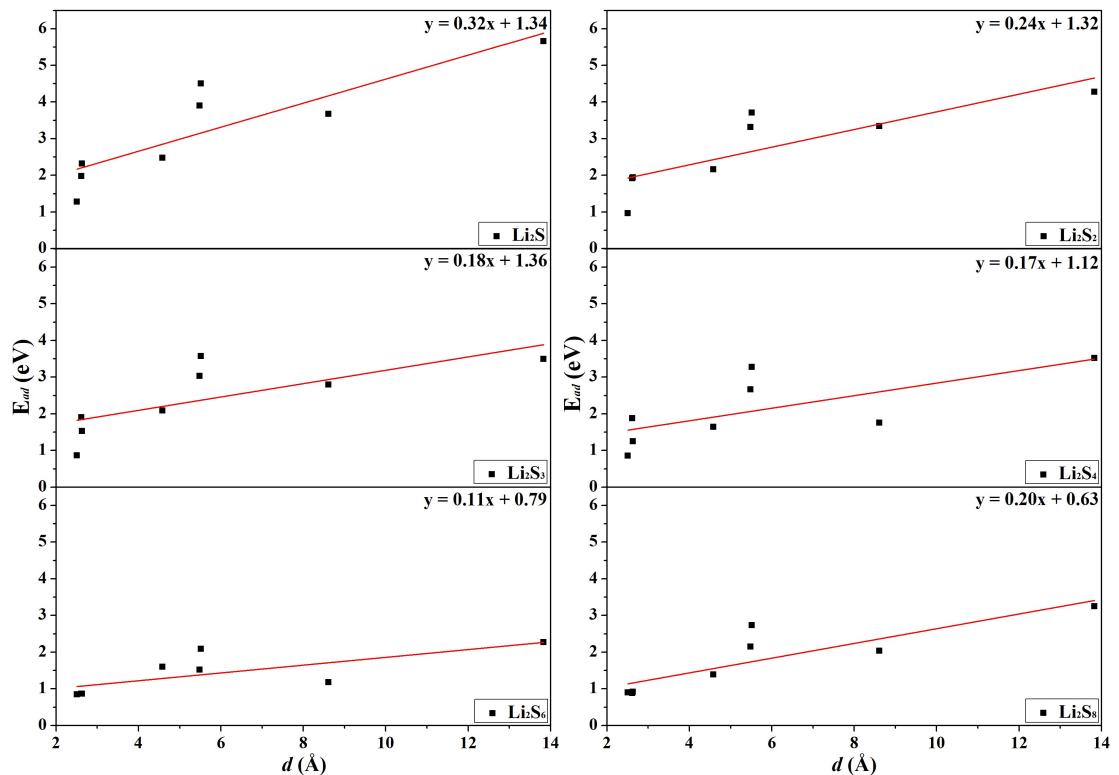
(f)  $g\text{-}C_3N_4$



(g)  $p\text{-}C_3N_4$



**Fig. S6** Charge density difference diagrams of the  $\text{Li}_2\text{S}_n$  and  $\text{S}_8$  on ten examined 2D AMs.



**Fig. S7** Linear fitting between the adsorption energy of  $\text{Li}_2\text{S}_n$  ( $n = 1\sim 4, 6, 8$ ) and the pore size of the AMs. The sequence of the AMs is the same as Fig. 6b except CTF and  $\text{T}-\text{N}_2\text{S}$ .

**Table S1** The N content ( $\eta$ , in %), pore size ( $d$ , in Å), and number of bridge N atoms in one pore ( $m$ ) for each considered AM.

	$\eta$	$d$	$m$
C <sub>9</sub> N <sub>4</sub>	30.77	5.51	6
C <sub>2</sub> N	33.33	5.49	6
BN	50.00	2.51	0
CTF	50.00	5.29	3
C <sub>2</sub> N <sub>6</sub> S <sub>3</sub>	54.54	8.61	6
<i>g</i> -C <sub>3</sub> N <sub>4</sub>	57.14	4.58	6
<i>p</i> -C <sub>3</sub> N <sub>4</sub>	57.14	2.63	3
C <sub>3</sub> N <sub>5</sub>	62.50	13.83	18
<i>S</i> -N <sub>2</sub> S	66.67	2.62	2
<i>T</i> -N <sub>2</sub> S	66.67	2.63	0

**Table S2** Average length of the formed S–C bonds ( $d_{S-C}$ , in Å) between the Li<sub>2</sub>S<sub>*n*</sub> and the ten AMs. Data in parentheses denote the number of S–C bonds.

	Li <sub>2</sub> S	Li <sub>2</sub> S <sub>2</sub>	Li <sub>2</sub> S <sub>3</sub>	Li <sub>2</sub> S <sub>4</sub>	Li <sub>2</sub> S <sub>6</sub>	Li <sub>2</sub> S <sub>8</sub>
C <sub>9</sub> N <sub>4</sub>	--	--	--	--	--	--
	--	--	--	--	--	--
C <sub>2</sub> N	--	--	--	--	--	--
	--	--	--	--	--	--
BN	--	--	--	--	--	--
	--	--	--	--	--	--
CTF	1.69 (1)	1.77 (1)	1.83 (1)	1.65 (2)	1.73 (2)	1.74 (3)
C <sub>2</sub> N <sub>6</sub> S <sub>3</sub>	--	--	--	--	--	--
	--	--	--	--	--	--
<i>g</i> -C <sub>3</sub> N <sub>4</sub>	--	--	--	--	--	--
	--	--	--	--	--	--
<i>p</i> -C <sub>3</sub> N <sub>4</sub>	2.02 (1)	--	--	--	--	--
	--	--	--	--	--	--
C <sub>3</sub> N <sub>5</sub>	--	--	--	--	--	--
	--	--	--	--	--	--
<i>S</i> -N <sub>2</sub> S	--	--	--	--	--	--
	--	--	--	--	--	--
<i>T</i> -N <sub>2</sub> S	--	--	--	--	--	--
	--	--	--	--	--	--

**Table S3** Li...S distances of the broken Li–S bonds ( $d_{\text{Li-S}}$ , in Å). Data in parentheses denote the number of broken Li–S bonds.

	Li <sub>2</sub> S	Li <sub>2</sub> S <sub>2</sub>	Li <sub>2</sub> S <sub>3</sub>	Li <sub>2</sub> S <sub>4</sub>	Li <sub>2</sub> S <sub>6</sub>	Li <sub>2</sub> S <sub>8</sub>
Li <sub>2</sub> S <sub>n</sub> @C <sub>9</sub> N <sub>4</sub>	--	--	4.38 (1)	4.61 (1)	4.34 (1)	--
Li <sub>2</sub> S <sub>n</sub> @C <sub>2</sub> N	--	--	4.61 (1)	--	3.62 (1)	4.78 (1)
Li <sub>2</sub> S <sub>n</sub> @BN	--	--	--	--	--	--
Li <sub>2</sub> S <sub>n</sub> @CTF	--	3.02 (1)	3.77 (1)	4.05 (1)	--	3.37 (2)
Li <sub>2</sub> S <sub>n</sub> @C <sub>2</sub> N <sub>6</sub> S <sub>3</sub>	--	--	3.68 (1)	--	--	--
Li <sub>2</sub> S <sub>n</sub> @g-C <sub>3</sub> N <sub>4</sub>	--	--	4.47 (1)	3.38 (1)	4.45 (1)	4.02 (1)
Li <sub>2</sub> S <sub>n</sub> @p-C <sub>3</sub> N <sub>4</sub>	--	--	--	--	--	--
Li <sub>2</sub> S <sub>n</sub> @C <sub>3</sub> N <sub>5</sub>	--	--	3.10 (2)	2.88 (1)	3.20 (2)	4.62 (2)
Li <sub>2</sub> S <sub>n</sub> @S-N <sub>2</sub> S	--	--	--	--	--	--
Li <sub>2</sub> S <sub>n</sub> @T-N <sub>2</sub> S	2.99 (1)	3.27 (2)	--	--	3.02 (1)	3.62 (3)

**Table S4** Average length of the formed S–N bond ( $d_{\text{S-N}}$ , in Å) between the Li<sub>2</sub>S<sub>n</sub> and the ten AMs. Data in parentheses denote the number of S–N bonds.

	Li <sub>2</sub> S	Li <sub>2</sub> S <sub>2</sub>	Li <sub>2</sub> S <sub>3</sub>	Li <sub>2</sub> S <sub>4</sub>	Li <sub>2</sub> S <sub>6</sub>	Li <sub>2</sub> S <sub>8</sub>
C <sub>9</sub> N <sub>4</sub>	1.81 (1)	-- --	-- --	-- --	-- --	-- --
C <sub>2</sub> N	1.78 (1)	-- --	-- --	-- --	-- --	-- --
BN	--	--	--	--	--	--
CTF	--	--	--	--	--	--
C <sub>2</sub> N <sub>6</sub> S <sub>3</sub>	--	--	--	--	--	--
g-C <sub>3</sub> N <sub>4</sub>	--	--	--	--	--	--
p-C <sub>3</sub> N <sub>4</sub>	--	--	--	--	--	--
C <sub>3</sub> N <sub>5</sub>	1.75 (1)	1.87 (1)	-- --	-- --	-- --	1.83 (1)
S-N <sub>2</sub> S	1.63 (1)	-- --	-- --	-- --	-- --	-- --
T-N <sub>2</sub> S	1.62 (1)	1.64 (1)	-- --	1.61 (1)	-- --	1.60 (2)

**Table S5** Diffusion barriers ( $E_a$ , in eV) of  $\text{Li}_2\text{S}_6$  along three paths on BN,  $\text{C}_9\text{N}_4$ ,  $\text{C}_2\text{N}_6\text{S}_3$  and CTF monolayers.

$\text{Li}_2\text{S}_6$	BN	$\text{C}_9\text{N}_4$	$\text{C}_2\text{N}_6\text{S}_3$	CTF
Path1	0.21	0.89	0.51	2.25
Path2	0.18	0.81	0.44	2.01
Path3	0.20	0.85	0.49	2.13

**Table S6** Variation of adsorption energies ( $E_{ad}$ , in eV), charger transfer ( $\Delta q$ , in  $|e|$ ) and average length of Li–S/N bond ( $d_{\text{Li–S}}$ ,  $d_{\text{Li–N}}$ , in Å). Data in parentheses denote the number of S–N bonds.

$@\text{Li}_2\text{S}_8$	$E_{ad}$	$\Delta q$	$d_{\text{Li–S}}$	$d_{\text{Li–N}}$
non-Li-trapped $\text{C}_9\text{N}_4$	2.73	0.75	2.69 (4)	2.03 (4)
Li-trapped $\text{C}_9\text{N}_4$	1.87	0.08	2.44 (4)	2.04 (2)
non-Li-trapped $\text{C}_2\text{N}_6\text{S}_3$	2.03	0.75	2.46 (4)	2.08 (4)
Li-trapped $\text{C}_2\text{N}_6\text{S}_3$	1.46	0.17	2.35 (4)	2.11 (2)