

Supporting information

Learning to Fly: Thermochemistry of Energetic Materials by Modified Thermogravimetric Analysis and Highly Accurate Quantum Chemical Calculations

Nikita V. Muravyev,^{a} Konstantin A. Monogarov,^a Igor N. Melnikov,^a Alla N. Pivkina,^a Vitaly G. Kiselev^{a,b,c}*

^a Semenov Federal Research Center for Chemical Physics RAS, 4 Kosygina Str., 119991 Moscow, Russia

^b Novosibirsk State University, 1 Pirogova Str., 630090 Novosibirsk, Russia

^c Institute of Chemical Kinetics and Combustion SB RAS, 3 Institutskaya Str., 630090 Novosibirsk, Russia

Contents

S1. Characterization of the compounds studied	S3
S2. Thermal corrections of the phase change data	S8
S4. Summary of the data fit with the modified Westwell-Trotton equation	S19
S5. Quantum chemical calculations of the gas-phase enthalpies of formation: Summary and literature data. Raw quantum chemical data: optimized geometries, electronic energies, and thermal corrections to thermodynamic potentials of all compounds under study	S20
References	S84

S1. Characterization of the compounds studied

Table S1 lists the compounds studied. They include the non-energetic species taken as references for thermochemical parameters (entries ##1-11), and the species with typical explosophoric functional groups or structural units, i.e., nitroaromatics (##12-16), aliphatic nitrocompounds (##17, 18), fluorine containing species (#19), nitramines (##20-26), caged nitramines (##27, 28), nitroesters (##29-31), diaziridine (#32), imidazoles (##33-34), pyrazoles (##35-42), pyrazines (##43, 44), triazoles (##45, 46), tetrazoles (##49-50), azasydrones (##51, 52), furazans (##53-57), furoxans (##58-63). All these compounds are comprised of CHNO atoms, except #19, which extends the proposed methodology to CHNOF compounds. Moreover, to cope with some typical difficulties in a common calorimetric experiment, we consider the two species (##47, 49) containing the water molecules in the crystal and #19, which is liquid at room temperature.

Due to the difficulties in determining of the accurate purity values for the materials with common physico-chemical tools, it was measured only for selected compounds (via DSC or NMR with the internal standard). For the rest of materials (denoted as “P” in the corresponding column) the NMR, MS spectra and the elemental analysis results were used for characterization. If the complete characterization of the species was published, the literature reference is given.

Table S1. List of investigated compounds with their purity and phase transition parameters.

#	Compound	Formula	Abbreviation	Purity / mol. %	T_m / °C		ΔH_m (T_m) / kJ mol ⁻¹	
					exp.	ref.	exp.	ref.
1	Acetamide	C ₂ H ₅ NO	ACM	99.8 ^a	80.5 ± 0.2 ^{d,f}	80.3[1] 81.8[2]	16.2 ± 0.8 ^{d,e}	15.6[1,3]
2	Carbamide	CH ₄ N ₂ O	Urea	P	133	132.9[4]	— ^c	—
3	2-Amino-3-methylbutanoic acid	C ₅ H ₁₁ NO ₂	Val	P	294 ^d	298[5]	— ^c	—
4	Benzoic acid	C ₇ H ₆ O ₂	BCA	99.9 ^a	122.1 ± 0.4	122.1 ± 0.7[4]	18.1 ± 0.6 ^d	17.3[6] 16.2[7] 18.0[8]
5	Naphthalene	C ₁₀ H ₈	NPH	99.9 ^a	80.4 ± 0.3 ^f	80.1 ± 0.7[4]	20.2 ± 0.4 ^{d,e}	19.6[9] 19.3[10]
6	Adamantane	C ₁₀ H ₁₆	ADM	P	267 ^d	269[11] 270[12] 279[13]	—	14.0[12] 10.9[14]
7	Anthracene	C ₁₄ H ₁₀	ANT	99.9 ^a	216.3 ± 0.1 ^d	217 ± 3[4]	29.3 ± 0.9 ^{d,e}	28.8[15] 29.8[16]
8	Trans-azobenzene	C ₁₂ H ₁₀ N ₂	tAB	99.9 ^a	67.8 ± 0.5 ^d	66.8[17]	27.6 ± 0.3 ^d	20.8[17]

						67.9[18]		21.3[19] 22.6[20]
9	Hypoxanthine	C ₅ H ₄ N ₄ O	HPX	P	405 ^d	–	– ^c	–
10	1,3-Dimethylxanthine	C ₇ H ₈ N ₄ O ₂	TPL	99.9 ^a	271.1 ± 0.3	269.1 for form II, 273.4 for form I [21] 270.6[22]	25.2 ± 1.8	28.2 for form II, 26.4 for form I [21] 31.2[22]
11	6,7,9,10,17,18,20,21-Octahydrodibenzo[b,k][1,4,7,10,13,16]hexaoxacyclooctadecine	C ₂₀ H ₂₄ O ₆	D18C6	99.9 ^a	162.1 ± 0.1	162.6[23] 166.9[24]	57.1 ± 1.5	55.4[24] 60.7[25] 56[23]
12	1-Azido-4-nitrobenzene	C ₆ H ₄ N ₄ O ₂	4-NPA	99.9 ^a	69.7 ± 0.4	72[26] 73.9[27]	25.1 ± 1.3	17.1 [26]
13	1-Methyl-2,4-dinitrobenzene	C ₇ H ₆ N ₂ O ₄	24-DNT	99.9 ^a	69.4 ± 0.3	67.9[28] 71.2[29]	25.3 ± 0.7	20.1[30] 20.9[10]
14	2,4-Dinitroanisole	C ₇ H ₆ N ₂ O ₅	DNAN	99.9 ^a	86.2 ± 0.6	87.5 for β-, 96.7 for α-[31] 86.8[32]	21.1 ± 0.8	20.2 [33]
15	2,4,6-Trinitrotoluene	C ₇ H ₅ N ₃ O ₆	TNT	99.8 ^b	80.4 ± 0.3[34]	80.7[35]	22.5 ± 0.5[34]	23.4[36]
16	N-Methyl-N-(2,4,6-trinitrophenyl)nitramide	C ₇ H ₅ N ₅ O ₈	Tetryl	99.9 ^a	129.0 ± 0.6	128.7- 129.1[37] 127. 4[38] 129.5[39]	25.3 ± 1.5	22.9[39] 25.9[38]
17	2,2-Dinitroethene-1,1-diamine	C ₂ H ₄ N ₄ O ₄	FOX-7	P	– ^g	–	– ^g	–
18	Tetranitroacetimidic acid	C ₂ HN ₅ O ₉	TNAA	99.5 ^b	91.0 ± 0.2[40]	91[41]	14.9 ± 1.8 ^c [40]	–
19	Bis(2-fluoro-2,2-dinitroethyl)formal	C ₅ H ₆ N ₄ O ₁₀ F ₂	FEFO	P	–	13.4[42] 14.5[43]	–	–
20	Nitroguanidine	CH ₄ N ₄ O ₂	NIGU	P	–	–	–	–
21	2,4-Dinitro-2,4-diazapentane	C ₃ H ₈ N ₄ O ₄	DNDAP	99.8 ^a	54.8 ± 0.4 ^f	54[44] 54.4[45]	22.3 ± 1.1	16.4[46]
22	1,4-Dinitropiperazine	C ₄ H ₈ N ₄ O ₄	DNPP	99.9 ^a	216.2 ± 0.3 ^{d,f}	214[46] 215[47]	35.2 ± 1.1 ^d	33.8[46]

23	1,3,5-Trinitroperhydro-1,3,5-triazine	C ₃ H ₆ N ₆ O ₆	RDX	P	204.2 ± 0.3 ^c [34]	204.0±0.1[39]	27.5 ± 0.9 ^c [34]	33.9±1.0[48]
24	Octahydro-1,3,5,7-tetranitro-1,3,5,7-tetrazocine	C ₄ H ₈ N ₈ O ₈	HMX	P	280 ^c	280.8±1.1[48]	— ^{c,g}	31.9±3.9[48]
25	1,9-diazido-2,4,6,8-tetranitro-2,4,6,8-tetraazanonane	C ₅ H ₁₀ N ₁₄ O ₈	DATN	P	177.4 ± 0.5 ^c	176-177[49] 177[50]	— ^c	—
26	cis-2,4,6,8-Tetra-nitro-1H,5H-2,4,6,8-tetraazabicyclo-[3.3.0]-octane	C ₄ H ₆ N ₈ O ₈	BC-HMX	P	—	—	—	—
27	4,10-Dinitro-2,6,8,12-tetraoxa-4,10-diazatetracyclo[5.5.0.05,9.03,11]-dodecane	C ₆ H ₆ N ₄ O ₈	TEX	99.0 ^b	—	—	—	—
28	2,4,6,8,10,12-Hexanitro-2,4,6,8,10,12-hexaazaisowurtzitane	C ₆ H ₆ N ₁₂ O ₁₂	CL-20	P	—	—	— ^g	—
29	Erythritol tetranitrate	C ₄ H ₆ N ₄ O ₁₂	ETN	99.6 ^a	61.2 ± 0.7	60±2[51]	36.9 ± 3.3 ^d	32±3[51]
30	Pentaerythritol tetranitrate	C ₅ H ₈ N ₄ O ₁₂	PETN	99.9 ^a	140.9±0.3	141.0±0.2[48] 142±2[51]	48.1±1.1	40±7[51] 53.8±3.9[48]
31	2,3-Hydroxymethyl-2,3-dinitro-1,4-butanediol tetranitrate	C ₆ H ₈ N ₆ O ₁₆	SMX	99.4 ^a	86.6 ± 0.7	85-86[52]	43.7 ± 3.0	35±6[51]
32	3,3-Pentamethylenediaziridine	C ₆ H ₁₂ N ₂	PMDA	P	105 ^{d,f}	—	28 ^d	—
33	2,4-Dinitroimidazole	C ₃ H ₂ N ₄ O ₄	24-DNI	P	—	278- 280[53] 265- 274[54]	—	—
34	4,5-Dinitroimidazole	C ₃ H ₂ N ₄ O ₄	45-DNI	P	179.9 ± 1.0 ^f	187-188[55]	23.3 ± 1.6	—
35	3,4-Dinitro-1H-pyrazole	C ₃ H ₂ N ₄ O ₄	34-DNP	99.5 ^a	85.9 ± 0.8 ^f	87[56] 89[57]	14.2 ± 0.3	—
36	3,5-Dinitro-1H-pyrazole	C ₃ H ₂ N ₄ O ₄	35-DNP	P	169.2 ± 0.3 [58]	168[56] 172[59]	20.7 ± 0.8	—
37	3,5-Dinitropyrazole-1-amine	C ₃ H ₃ N ₅ O ₄	1-ADP	99.7 ^a	112.9 ± 0.7 ^f	112[60] 113[61]	23.6 ± 2.2	—
38	5-Amino-3,4-dinitro-1H-pyrazole	C ₃ H ₃ N ₅ O ₄	5-ADP	99.5 ^b	195 ^c	—	— ^c	—
39	3,4,5-Trinitro-1H-pyrazole	C ₃ HN ₅ O ₆	TNP	99.8 ^a	186.3±0.6 ^f	182-184[62] 187[63] 186[56]	24.9±1.9	28.2[64] 28.8[63]

40	1-Methyl-3,4,5-trinitropyrazole	C ₄ H ₃ N ₅ O ₆	MTNP	99.8 ^a	90.6 ± 0.5 ^{d,f}	90.8[56] 91.3[65]	19.2 ± 1.7 ^d	17[63]
41	3,6-Dinitropyrazolo[4,3-c]pyrazole	C ₄ H ₂ N ₆ O ₄	DNPPy	P	—	—	—	—
42	1,2,9,10-Tetranitrodipyrazolo[1,5-d:5',1'-f][1,2,3,4]-tetrazine,	C ₆ N ₁₀ O ₈	TNDPT	P	195.5 ± 1.6 ^c	205 ^h [66]	— ^c	—
43	2,6-Diamino-3,5-dinitropyrazine	C ₄ H ₄ N ₆ O ₄	ANPZ	P	—	—	—	—
44	2,6-Diamino-3,5-dinitropyrazine-1-oxide	C ₄ H ₄ N ₆ O ₅	LLM-105	P	—	—	—	—
45	5-Amino-3-nitro-1H-1,2,4-triazole	C ₂ H ₃ N ₅ O ₂	ANTA	P	242.4 ^c	—	— ^c	—
46	3-Nitro-1,2,4-triazol-5-one	C ₂ H ₃ N ₅ O ₂	NTO	P	—	—	—	—
47	1H-Tetrazol-5-ylamine hydrate	CH ₃ N ₅	5-ATZ	P	205.1 ± 0.2	205[67] 203[68]	— ^c	—
48	1,5-Diaminotetrazole	CH ₄ N ₆	DAT	99.9 ^a	187.3 ± 0.8	187-188[69]	26.7 ± 1.4 ^c	—
49	5,5'-Bi-1H-tetrazole	C ₂ H ₂ N ₈	BTZ	P	—	—	—	—
50	5,5'-Bistetrazole-1,1'-diol hydrate	C ₂ H ₆ N ₈ O ₄	BTO	P	—	—	—	—
51	3-Phenyl-1,2,3,4-oxatriazolium-5-olat	C ₇ H ₅ N ₃ O ₂	Ph-Azs	P[70]	81.7 ± 0.5	—	20.8 ± 0.5	—
52	3-(4-Tolyl)-1,2,3,4-oxatriazolium-5-olat	C ₈ H ₇ N ₃ O ₂	4-Tol-Azs	P[70]	98.7 ± 1.1 ^f	—	20.5 ± 0.3	—
53	3-Amino-4-nitro-1,2,5-oxadiazole	C ₃ H ₃ N ₃ O ₄	ANF	99.4 ^a	123.8 ± 0.8 ^{d,f}	121[71]	27.6 ± 1.7 ^d	—
54	3,4-Diaminofurazan	C ₂ H ₄ N ₄ O	DAF	99.9 ^a	179.9 ± 0.3 ^d	179-180[72]	23.1 ± 0.4 ^d	—
55	3,4-Di(nitramino)furazan	C ₂ H ₂ N ₆ O ₅	DNAF	P	94.2 ^c	93[73]	— ^c	—
56	3,3'-Diamino-4,4'-azofurazan	C ₄ H ₄ N ₈ O ₂	DAAzF	P	—	325[74]	—	—
57	3,3'-Diamino-4,4'-azoxyfurazan	C ₄ H ₄ N ₈ O ₃	DAAF	P	253.9 ^c	249[74]	— ^c	—
58	3-Methyl-4-nitro-1,2,5-oxadiazole 2-oxide	C ₃ H ₃ N ₃ O ₄	34-MNF	P	60; 65	67-68[75]	21.3 ± 0.4	—
59	4-Methyl-3-nitro-1,2,5-oxadiazole 2-oxide	C ₃ H ₃ N ₃ O ₄	43-MNF	99.7 ^a	40.7 ± 0.3 ^f	—	15.1 ± 0.2	—
60	3-Cyano-4-nitrofuroxan	C ₃ N ₄ O ₄	34-CNf	99.6 ^a	48.1 ± 0.3 ^{d,f}	48-49[76]	22.8 ± 0.9 ^d	—
61	Benzotris[1,2,5]oxadiazole, 1,4,7-trioxide	C ₆ N ₆ O ₆	BTF	99.3 ^a	197.1 ± 0.4 ^{d,f}	194- 195[77] 196- 197[78] 201[79]	22.7 ± 1.6 ^{c,d}	—

62	4-nitro-(4-nitrofuran-3-yl)furoxan	C ₄ N ₆ O ₇	NNFF	P[80]	105.7 ± 1.1 ^{d,f}	–	33.0 ± 2.8	–
63	4,4'-Dinitro-3,3'-bisfuroxan	C ₄ N ₆ O ₈	DNBF	P[81]	134.5 ± 0.5 ^c	–	24.9 ± 3.3 ^c	–

^a The purity is determined by DSC.

^b The purity determined by NMR with an internal standard.

^c Melting upon decomposition.

^d High pressure DSC was used to suppress evaporation.

^e The enthalpy value was corrected to account for a slight mass loss (<10%).

^f The melting temperature for pure sample obtained from the Van't-Hoff equation [82].

^g The phase transition instead of melting observed with the following parameters:

CL-20: the temperature onset 164.3±0.5°C, the enthalpy 6.9±0.9 kJ/mol in agreement with 7.2[83] and 7.9±0.4[84] kJ/mol

HMX: the temperature onset 189.9±2.0°C, the enthalpy 8.3±0.9 kJ/mol in agreement with 9.8±0.8[38] kJ/mol

FOX-7: the temperature onset 120.5±1.2°C, the enthalpy 2.5±0.2 kJ/mol in agreement with 3.2±0.5[85] kJ/mol and the temperature onset 166.0±0.9°C, enthalpy 2.8±0.5 kJ/mol in agreement with 2.6±0.5[85] kJ/mol

^h The peak temperature was reported in the original paper, while all temperature values in present study correspond to the extrapolated onset of the peak.

S2. Thermal corrections of the phase change data

The enthalpy data obtained at elevated temperature can be corrected to the standard conditions (298 K) using several approaches. Since the heat capacity values for novel energetic materials are often unknown, semiempirical corrections were considered either with the constant $\Delta c_p^{s \rightarrow g}$:[86]

$$\Delta_s^g H_m^0 = \Delta_s^g H_m(T_{ave}) + 0.032 \cdot (T_{ave} - 298.15), \quad (\text{S1})$$

or with an estimated solid-state heat capacity by group additivity methods:[86]

$$\Delta_s^g H_m^0 = \Delta_s^g H_m(T_{ave}) + [0.75 + 0.15 \cdot c_p^s(\text{estd})] (T_{ave} - 298.15), \quad (\text{S2})$$

or with a gas-phase heat capacity only:[87]

$$\Delta_s^g H_m^0 = \Delta_s^g H_m(T_{ave}) + [0.9 + 0.176 \cdot c_p^g] (T_{ave} - 298.15), \quad (\text{S3})$$

The enthalpy of vaporization can be corrected analogously – via adjustment with a constant $\Delta c_p^{l \rightarrow g}$:[88]

$$\Delta_l^g H_m^0 = \Delta_l^g H_m(T_{ave}) + 0.054 \cdot (T_{ave} - 298.15), \quad (\text{S4})$$

or via an estimated liquid-state heat capacity:[86]

$$\Delta_l^g H_m^0 = \Delta_l^g H_m(T_{ave}) + [0.58 + 0.26 \cdot c_p^l(\text{estd})] (T_{ave} - 298.15). \quad (\text{S5})$$

For a subset of selected compounds we compared the accurate temperature corrections using Eq. (5) with accounting for temperature dependence of the solid-state heat capacity (from DSC data or from literature) and the gas-phase heat capacity (obtained by quantum chemical calculations) with the approximations (S1) - (S5).

Table S2. Thermal adjustment of the sublimation enthalpy data by various methods

Compound	Thermal correction value (second term in Eq. 5)				Error		
	exact $\Delta c_p^{s \rightarrow g}$	Equation S1	Equation S2	Equation S3	Equation S1	Equation S2	Equation S3
ACM	0.5	1.2	0.5	0.5	0.7	0.0	0.0
Urea	0.8	1.2	0.6	0.5	0.4	-0.3	-0.4
Val	4.8	6.3	5.0	5.1	1.6	0.2	0.4
BCA	2.0	2.2	1.6	1.4	0.2	-0.4	-0.6
NPH	1.4	1.1	0.9	0.7	-0.3	-0.5	-0.6
ADM	1.8	1.2	1.0	0.9	-0.6	-0.7	-0.9
ANT	5.8	4.6	4.6	4.3	-1.2	-1.2	-1.4
tAB	0.8	0.5	0.6	0.5	-0.2	-0.2	-0.2
TPL	3.0	6.8	6.4	6.8	3.9	3.4	3.9
DB18C6	5.9	3.6	7.5	7.4	-2.4	1.6	1.5

DNAN	1.2	1.3	1.4	1.4	0.2	0.3	0.2
TNT	1.9	1.1	1.4	1.3	-0.8	-0.5	-0.6
Tetryl	4.0	2.5	3.6	3.6	-1.5	-0.4	-0.5
FOX-7	5.4	5.2	4.4	4.1	-0.2	-1.0	-1.3
NIGU	2.5	3.8	2.3	2.2	1.3	-0.2	-0.4
DNPP	6.4	4.9	5.8	4.6	-1.5	-0.6	-1.8
RDX	4.1	2.4	2.7	2.4	-1.8	-1.5	-1.7
HMX	6.7	4.5	8.1	6.6	-2.2	1.4	-0.1
TEX	4.1	6.9	7.3	7.5	2.8	3.2	3.4
CL-20	19.7	4.7	10.5	9.5	-14.9	-9.2	-10.2
PETN	4.9	2.3	3.7	3.6	-2.6	-1.2	-1.3
5-ADP	4.6	4.0	3.5	3.4	-0.6	-1.1	-1.3
LLM-105	3.4	5.4	5.4	5.9	2.0	2.0	2.5
NTO	1.4	3.2	1.8	1.9	1.8	0.4	0.5
DAF	6.3	4.3	3.0	2.3	-2.0	-3.4	-4.1
DAAzF	3.1	3.3	3.4	3.4	0.2	0.3	0.3
Average deviation					-0.7	-0.4	-0.6
Absolute average deviation					1.8	1.3	1.5
Maximal absolute deviation					14.9	9.2	10.2

Table S3. Thermal corrections to the evaporation enthalpy data by various methods.

Corrections	correction			difference	
	exact $\Delta c_p^{l \rightarrow g}$	Equation S4	Equation S5	Equation S4	Equation S5
ACM	6.6	4.9	4.4	-1.7	-2.3
BCA	11.1	7.0	7.7	-4.0	-3.4
NPH	5.6	4.7	5.4	-0.9	-0.2
tAB	3.0	6.7	10.2	3.7	7.2
DB18C6	45.8	12.4	39.2	-33.4	-6.6

FEFO	5.7	1.6	3.9	-4.1	-1.7
Average deviation (AD)				-6.7	-1.2
Absolute average deviation (AAD)				6.8	3.1
Maximal absolute deviation (MAD)				33.4	7.2

S3. Summary of the experimental sublimation and vaporization enthalpies.

Table S4. Comparison of the experimentally determined sublimation and evaporation enthalpies with the literature data.

#	Compound	ΔH_{ev} (298 K) / kJ mol ⁻¹		Details	ΔH_s (298 K) / kJ mol ⁻¹				Details
		this study	literature		this study	literature	evaporation + melting	Trouton-Williams equation (4)	
1	ACM	65.6 ± 2.8	$68.6[89]^a$ $65.2[90]^a$	Argon flow, $\beta = 0.15\text{-}5 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 120^\circ\text{C}$	83.2 ± 1.6	$80.3[91]$ $78.3[2]^a$	78.7 ± 3.0	66	Argon flow, $\beta = 0.15\text{-}0.3 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 64^\circ\text{C}$
2	Urea				97.2 ± 1.9	$95.5[92]$ (tr) $96.3[93]$ (cl) $98.3[94]$ (to) $76.9[95]$ (to) $89[96]$ (ef)		77	Vacuum, $\beta = 0.1\text{-}1 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 63^\circ\text{C}$
3	Val				135.1 ± 2.2	$167.8[97]^a$		107	Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 223^\circ\text{C}$
4	BCA	75.3 ± 2.8	$73.2[98]^a$ (tr) $69.3[99]^a$ (tg) $72.4[4]^a$ (is) $76.5[100]^a$ (ma) $83[101]^a$ (ch)	Argon flow, $\beta = 1\text{-}5 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 155^\circ\text{C}$	94.0 ± 1.6	$89.7 \pm 1.0[102]$ (cl) $90 \pm 4[4]$ (re) $89.5 \pm 0.2[103]$ (re) $95.1 \pm 1.8[104]$ (ef)	88.5 ± 3.1	75	Argon flow, $\beta = 0.15\text{-}2 \text{ K min}^{-1}$, 8 runs, $\langle T \rangle = 95^\circ\text{C}$
5	NPH	52.5 ± 3.0	$53.4[14]$ (ch) $54.8[105]$ (eb) $50.3[106]$ (ma)	Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 112^\circ\text{C}$	73.4 ± 1.2	72.6 ± 0.6 [102] (re) $71 \pm 5[4]$ (re) 72.8 ± 0.8 [103] (cl)	71.5 ± 3.4	66	Argon flow, $\beta = 0.1\text{-}0.5 \text{ K min}^{-1}$, 5 runs, $\langle T \rangle = 59^\circ\text{C}$
6	ADM				62.7 ± 1.9	$59 \pm 4[4]$ (re) $59.1 \pm 0.9[107]$ (re)		102	Argon flow, $\beta = 0.1\text{-}5 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 61^\circ\text{C}$

7	ANT		77.8[108] ^a (ch) 75.1[109] ^a (ch) 77.5[16] ^a (ds)		103.0 ± 2.5	103.4 ± 2.7 [102] (re) 100.4 ± 4.2 [103] (re)		92	Argon flow, $\beta = 0.5\text{-}5$ K min^{-1} , 4 runs, $\langle T \rangle = 169^\circ\text{C}$
8	tAB	74.1 ± 1.6	72.8[19]	Argon flow, $\beta = 1\text{-}5 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 149^\circ\text{C}$	92.8 ± 1.9	93.8 ± 0.4 [4] (re)	96.6 ± 3.6	64	Vacuum, $\beta = 0.1\text{-}2 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 42^\circ\text{C}$
9	HPX				161.9 ± 2.8	162.5[110] 162.9[111] ^a		128	Argon flow, $\beta = 0.5\text{-}5$ K min^{-1} , 3 runs, $\langle T \rangle = 349^\circ\text{C}$
10	TPL	106.7 ± 2.3		Argon flow, $\beta = 1\text{-}5 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 311^\circ\text{C}$	130.8 ± 2.3	136.3[112] ^a 138.5[112] ^a 132.5[113] (fs) 129.9[22] ^a	132.0 ± 4.1	102	Argon flow, $\beta = 0.5\text{-}5$ K min^{-1} , 5 runs, $\langle T \rangle = 239^\circ\text{C}$
11	D18C6	130.9 ± 15.2	137[24] 154.1[23]	Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 254^\circ\text{C}$	167.0 ± 5.4	178.4[24] 180.2[24] 190.6[23]	168.6 ± 16.7	82	Vacuum, $\beta = 0.2\text{-}1 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 137^\circ\text{C}$
12	4-NPA	66.7 ± 3.1		Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 117^\circ\text{C}$		81[26] 79.5[114]	89.7 ± 4.4	65	
13	24-DNT	77.1 ± 3.0	75.3[115] (re) 76[89] ^a	Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 153^\circ\text{C}$	100.6 ± 2.1	94[115] (re) 96.0[116] ^a 94.5[117] ^a 83.3[118] ^a	99.8 ± 3.9	64	Vacuum, $\beta = 0.15\text{-}0.5$ K min^{-1} , 3 runs, $\langle T \rangle = 50^\circ\text{C}$
14	DNAN	86.4 ± 3.5	57.2 (comptd) [119]	Argon flow, $\beta = 0.5\text{-}2 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 172^\circ\text{C}$	107.1 ± 1.4		104.2 ± 4.4	68	Vacuum, $\beta = 0.2\text{-}1 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 67^\circ\text{C}$
15	TNT	90.7 ± 3.7	89.8[115] (re) 86.6 ± 0.4 [103] (cl)	Argon flow, $\beta = 1\text{-}10 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle =$	110.7 ± 2.6	114.1[115] (re) 113.2 ± 1.5 [120] (ef)	110.2 ± 4.1	67	Vacuum, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 60^\circ\text{C}$

			87.0 ± 1.9 [120] (ef)	192°C		106.3 ± 0.4 [103] (cl) 118.0 ± 0.8 [103] (cl)			
16	Tetryl				136.0 ± 3.1	133.1 [121] 133.8 ± 1.6 [120] (ef)		76	Vacuum, $\beta = 0.2$ - 0.5 K min $^{-1}$, up to DSC growth, 5 runs, $\langle T \rangle = 188$ °C
17	FOX-7				119.4 ± 3.3	116 [122] ^c 104 [123] ^c 152 [124] ^c		90 ^b	Vacuum, $\beta = 0.15$ - 1 K min $^{-1}$, up to DSC growth, 5 runs, $\langle T \rangle = 188$ °C
18	TNAA				101.2 ± 1.7			69	Vacuum, $\beta = 0.05$ - 0.25 K min $^{-1}$, 5 runs, $\langle T \rangle = 51$ °C
19	FEFO	93.3 ± 7.4	85.1 [125]	Vacuum, $\beta = 0.5$ - 2 K min $^{-1}$, 3 runs, $\langle T \rangle = 54$ °C				54	
20	NIGU				141.6 ± 2.4	142.7 [120] 139.7 [126]		83	Vacuum, $\beta = 0.05$ - 1 K min $^{-1}$, 4 runs, $\langle T \rangle = 145$ °C
21	DNDAP	75.0 ± 2.7		Argon flow, $\beta = 0.5$ - 2 K min $^{-1}$, 3 runs, $\langle T \rangle = 122$ °C		102.5 [127]	95.6 ± 3.8	62	
22	DNPP	78.6 ± 1.3		Argon flow, $\beta = 0.75$ - 2 K min $^{-1}$, 3 runs, $\langle T \rangle = 234$ °C	111.1 ± 1.0	112.7 ± 0.8 [128] ^a (ef) 102.9 ± 1.2 [127] (ef)	113.7 ± 2.4	92	Argon flow, $\beta = 0.75$ - 2 K min $^{-1}$, 4 runs, $\langle T \rangle = 177$ °C
23	RDX				130.4 ± 2.3	128.0 ± 1.3 [127] (cl) 127.1 [115](re) 150 [129] ^c (bu)		90	Vacuum, $\beta = 0.15$ - 0.35 K min $^{-1}$, 5 runs, $\langle T \rangle = 99$ °C
24	HMX				179.1 ± 7.0	161.0 ± 0.3 [120] (ef) 164.4 ± 1.7 [103](cl)		104	Vacuum, $\beta = 0.1$ - 0.2 K min $^{-1}$, 3 runs, $\langle T \rangle = 165$ °C (all data before

						176.1±1.7 [127](cl) 174.7[130] ^c (ef) 180[129] ^c (bu)			phase transition)
25	DATN				230.8 ±8.0	195[131] (ri)		85	Vacuum, $\beta = 0.04\text{-}0.15 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 156 \text{ }^\circ\text{C}$
26	BC-HMX				134.6 ±2.5			97 ^b	Vacuum, $\beta = 0.2\text{-}1 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 170 \text{ }^\circ\text{C}$
27	TEX				130.9 ±2.4	~130[132]		104 ^b	Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 240 \text{ }^\circ\text{C}$
28	CL-20				171.9 ±3.2	157±20[133] (po) 168[129] ^c (bu) 226.8±4.6 [127](ph)		95 ^b	Vacuum, $\beta = 0.15\text{-}0.35 \text{ K min}^{-1}$, 5 runs, $\langle T \rangle = 173 \text{ }^\circ\text{C}$
29	ETN	92.7 ±2.4		Vacuum, $\beta = 0.2\text{-}1 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 74 \text{ }^\circ\text{C}$	139.5 ±3.0	118.5[51] ^a	127.6±5.7	63	Vacuum, $\beta = 0.1\text{-}1 \text{ K min}^{-1}$, 5 runs, $\langle T \rangle = 50 \text{ }^\circ\text{C}$
30	PETN			-	147.6 ±3.5	150.8 [115] (re) 150.4±1.3[120] (ef) 150.6±1.8[134] (tg?) 149.2[51] ^a	-	78	Vacuum, $\beta = 0.4\text{-}1 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 97 \text{ }^\circ\text{C}$
31	SMX				166.3 ±3.5	106[51] ^a 164±20[135] (po)		68	Vacuum, $\beta = 0.1\text{-}0.4 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 71 \text{ }^\circ\text{C}$
32	PMDA				85.6 ±1.5			71	Argon flow, $\beta = 0.5\text{-}2 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 75 \text{ }^\circ\text{C}$
33	24-DNI				112.1 ±3.2	112.1[136]		102	Vacuum, $\beta = 0.2\text{-}1 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle =$

									134 °C
34	45-DNI				119.0 ±2.0			85	Vacuum, $\beta = 0.2\text{-}1 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 110 \text{ }^\circ\text{C}$
35	34-DNP	99.6 ±2.4		Vacuum, $\beta = 0.5\text{-}2 \text{ K min}^{-1}$, 6 runs, $\langle T \rangle = 114 \text{ }^\circ\text{C}$	113.8 ±2.3	112.5±0.8[136] (cl) 141.4[137] (tg)	114.1 ±3.3	68	Vacuum, $\beta = 0.03\text{-}0.1 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 75 \text{ }^\circ\text{C}$
36	35-DNP	89.3 ±2.0		Argon flow, $\beta = 0.2\text{-}2 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 199 \text{ }^\circ\text{C}$	107.8 ±1.3		104.0 ±2.7	83	Argon flow, $\beta = 0.1\text{-}2.5 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 145 \text{ }^\circ\text{C}$
37	1-ADP	100.1 ±4.6		Argon flow, $\beta = 1\text{-}5 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 211 \text{ }^\circ\text{C}$			119.0 ±6.8	72	
38	5-ADP				121.2 ±2.0			88	Vacuum, $\beta = 0.15\text{-}0.3 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 150 \text{ }^\circ\text{C}$
39	TNP				121.5 ±1.9			86	Vacuum, $\beta = 0.5\text{-}2 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 107 \text{ }^\circ\text{C}$
40	MTNP				104.8 ±1.7			68	Vacuum, $\beta = 0.2\text{-}2 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 70 \text{ }^\circ\text{C}$
41	DNPPy				134.2 ±4.4			112	Vacuum, $\beta = 0.5\text{-}2 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 172 \text{ }^\circ\text{C}$
42	TNDPT				122.0 ±3.8			88	Vacuum, $\beta = 0.2\text{-}1 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 170 \text{ }^\circ\text{C}$
43	ANPZ				124.7 ±2.3			119	Vacuum, $\beta = 0.1\text{-}2 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 180 \text{ }^\circ\text{C}$
44	LLM-105				132.3 ±3.9	135.6[138] (est) 163.8[137]		114	Vacuum, $\beta = 0.5\text{-}2 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 193 \text{ }^\circ\text{C}$

						103.3[139] (ri)			
45	ANTA				127.8 ±3.3			97	Vacuum, $\beta = 0.2\text{-}1 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 158^\circ\text{C}$
46	NTO				122.8 ±2.3	107.9[129]		102	Vacuum, $\beta = 0.2\text{-}1 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 126^\circ\text{C}$
47	5-ATZ				116.0 ±2.0	116.3±1.2[140] ^a	-	90	Vacuum, $\beta = 0.2\text{-}1 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 110^\circ\text{C}$
48	DAT				109.6 ±2.2			87	Vacuum, $\beta = 0.2\text{-}2 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 125^\circ\text{C}$
49	BTZ				142.5 ±2.8			101 ^b	Vacuum, $\beta = 0.3\text{-}1 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 166^\circ\text{C}$
50	BTO				141.2 ±2.7			95	Vacuum, $\beta = 0.2\text{-}1 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 178^\circ\text{C}$
51	Ph-Azs	80.5 ±3.3		Argon flow, $\beta = 0.5\text{-}2 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 152^\circ\text{C}$	109.5 ±1.7		98.2 ± 3.8	67	Vacuum, $\beta = 0.1\text{-}0.5 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 44^\circ\text{C}$
52	4-Tol-Azs	77.1 ±3.6		Argon flow, $\beta = 1\text{-}10 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 160^\circ\text{C}$	97.1 ±2.4		93.7 ± 3.9	70	Vacuum, $\beta = 0.2\text{-}2 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 66^\circ\text{C}$
53	ANF				93.3 ±2.2		90.0 ±4.7	74	Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 92^\circ\text{C}$
54	DAF	89.2 ±2.2		Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 206^\circ\text{C}$	109.4 ±4.1		103.9 ±2.6	85	Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 4 runs, $\langle T \rangle = 160^\circ\text{C}$
55	DNAF				133.3 ±2.9			69	Vacuum, $\beta = 0.1\text{-}1 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle =$

									69°C
56	DAAzF				96.4 ±3.7	103.5[141] (calc) 99.6[142] (co)		109	Vacuum, $\beta = 0.1\text{-}2 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 127^\circ\text{C}$
57	DAAF				117.1 ±2.3	117.2[141] (calc) 82.3[143]		99	Vacuum, $\beta = 0.2\text{-}2 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 153^\circ\text{C}$
58	34-MNF	67.5 ±1.8		Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 102^\circ\text{C}$			86.6 ±2.2	64	
59	43-MNF	61.8 ±2.8		Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 81^\circ\text{C}$			76.6 ±3.1	59	
60	34-CNF	72.5 ±2.6		Argon flow, $\beta = 0.5\text{-}2 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 91^\circ\text{C}$			94.0 ±3.5	60	
61	BTF	95.4 ±4.6		Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 218^\circ\text{C}$	107.1 ±0.9	175.2±2.5[79] ^a	108.9 ±6.2	88	Vacuum, $\beta = 0.2\text{-}1 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 114^\circ\text{C}$
62	NNFF	87.7 ±3.1		Argon flow, $\beta = 0.5\text{-}5 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 123^\circ\text{C}$	119.6 ±1.9			71	Vacuum, $\beta = 0.3\text{-}1 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 54^\circ\text{C}$
63	DNBF				106.5 ±2.4			77	Vacuum, $\beta = 0.5\text{-}1 \text{ K min}^{-1}$, 3 runs, $\langle T \rangle = 69^\circ\text{C}$

^aCorrected to RT with Eqs. (S1) and (S4).

^bThe onset temperature used in Eq.(4) instead of melting point.

^cCorrected to RT with Eq.(S1) and phase transition enthalpy added.

Methods:

re – the values recommended upon the analysis of the available literature

bu – corrected to RT data from the pressure-time dependence obtained in burning tests

cl – calorimetric technique.

tr – transpiration

is – isoteniscope

ma – manometry

ch – gas chromatography

tg – thermogravimetric

ds- DSC

ef - Knudsen effusion method

eb – ebulliometer

ph - supposed to be equal to the experimental value of the activation energy of the phase transition.

po – calculated by the Politzer approach

ri – calculated by the Rice approach

to – torsion-effusion method

fs – fast scanning calorimetry

S4. Summary of the data fit with the modified Westwell-Trouton equation

Residuals:

	Min	1Q	Median	3Q	Max
	-28160	-5573	1113	7456	29514

Coefficients:

	Estimate	Std. Error	t value	Pr(> t)
Nh	3268.12	494.72	6.606	4.33e-08 ***
Nn	5294.95	965.86	5.482	1.93e-06 ***
Tm	151.13	12.38	12.211	9.99e-16 ***
No	3298.10	605.52	5.447	2.18e-06 ***

Signif. codes: 0 ‘***’ 0.001 ‘**’ 0.01 ‘*’ 0.05 ‘.’ 0.1 ‘ ’ 1

Residual standard error: 12860 on 44 degrees of freedom

Multiple R-squared: 0.9895, Adjusted R-squared: 0.9885

F-statistic: 1036 on 4 and 44 DF, p-value: < 2.2e-16

S5. Quantum chemical calculations of the gas-phase enthalpies of formation: Summary and literature data. Raw quantum chemical data: optimized geometries, electronic energies, and thermal corrections to thermodynamic potentials of all compounds under study

Table S5. The results of quantum chemical calculations of $\Delta_f H_m^0(g)$ along with the literature data. All value are calculated either with W1-F12 or W2-F12 multi-level techniques, except otherwise noted. All values are given in kJ/mol.

Compound	Present study	Literature (Method/Basis)
ACM	-234.8	-238.8 ±0.8 [144] -236.5 (W3X-L) [145] -230.6 (G4MP2) [146]
Urea	-233.2	
Val	-466.6	-476.5 (G4) -475.8 (isodesmic) [147]
BCA	-296.9	-298.6 (G4MP2) [146] -294.1±1.6 [148]
NPH	146.4	150.6±1.5 (review) [149] 146.2 (W1-F12) 146.9 (G4) [150]
ADM	-135.9	-132.2±2.2 (exp) [107]
ANT	225.8	229.4±2.9 (review) [149] 225.5 (W1-F12) 224.3 (G4) [150]
tAB	407.6	405.5 ±1.3 (experiment) [19]
HPX	50.6	52.4 ±1.4 (experiment) 46 ± 4.6 (G4) [110]
TPL	-228.7	-232 ±8 (G3B3) [151]
DB18C6	—	
4-NPA	400.4	389.7 ±5.2 (experiment) [26] 393.9 (isodesmic) 384.3 (G4) [152]
24-DNT	30.5	32.5±4 (isodesmic) 31.5(recommended) [152]
DNAN	-75.9	
TNT	48.8	24.1 ±3.5 (experiment) [29] 50.1±4 (isodesmic) 23.0 (G4) [152]
Tetryl	164.2	165.7±4 (G4(MP2)+isodesmic) [152]
FOX-7	10.0	3.1±6.0 [141]
TNAA	-16.3	-38.1 (G4) -16.3 ±6.3 (G4+isodesmic) [40] -134.6 (isodesmic) [41] -174.9 [153]
FEFO	—	
NIGU	51.3	48.7±4 (G4+isodesmic) [154]
DNDA-5	41.3	51.0 ±0.8 (experiment) [127] 37.8±7 (G4+isodesmic) [154]
DNPP	84.7	58 ±3 (experiment) [128] 49.8±2.1 (experiment) [127] 64.7±5 68.3 (G4) [155]
RDX	197.9	185.8 (B3LYP/6-31G(d,p)) [156] 196.2 (G4) [155] 190.8±5 [141] 195.0 ±1.7 (experiment) [127]

HMX	259.9	247.7±7 253.9 (G4) [155] 261.9 ±2.5 (experiment) [127]
DATN	848.6	864.8 [131]
BC-HMX	318.7	
TEX	-400.0	-400.0 ±6.3 [157] -448.4 (B3LYP 6-31+G**) [158]
CL-20	548.5	-548.5 ±6.3 [157] 601.7 ±4.6 (experiment) [127] 523.1 (G4) [155] 522.8±10 [141] 691.3 (B3LYP+6-31G*) [159]
ETN	-356.7	
PETN	-395.5	-413.6 ±4[141]
SMX	-375.3 ^a	-469 -674 [160] -415.7 ± 15 [135]
PMDA	128.5	
2,4-DNI	137.2	133.5 (G4MP2-SFM) [161]
4,5-DNI	160.6	
3,4-DNP	210.3	205.0 (B3P86/6-311G (d, p))[162]
3,5-DNP	190.6	
1-ADP	305.3	243.8 (MP2/6-311++G**)[61]
5-ADP	196.5	165.7 (B3P86/6-311G (d, p))[162]
TNP	253.6	142.7 (PM3)[163]
MTNP	233.0	230.1 (B3P86/6-311G (d, p))[162] -15.1 (PM3) [163]
DNPPy	414.5	
TNDPT	897.5	847.4 (MP2/6-311++G**)[66]
ANPZ	119.9	98.3 (CBS-4M)[164]
LLM-105	112.5	108.4 (ccCA-PS3) 94.1 (G4) [138] 126.8 (CBS-4M)[164]
ANTA	201.2	
NTO	-4.2	-14 (G4) -6.5 (isodesmic)[152] -3.8(B3LYP/6-31G(d,p)//B3LYP/6-31G(d,p))[156]
5-ATZ	329.0	325.1 (W1) [165] 325.3 ±3 [141]
DAT	424.6	
BTZ	674.2	665.1 ±5[141]
BTO	669.7	689.1 (CBS-4M) [166]
Ph-Azs	175.1	

4-Tol-Azs	140.9	
ANF	220.1	
DAF	198.7	
DNAF	347.9	
DAAzF	653.3	639.3 ± 7 [141]
DAAF	616.4	605 ± 5 [141]
DNAAF	811.3 ^b	798.3 (isodesmic) [167]
34-MNF	188.2	
43-MNF	196.6	
34-CNF	421.3	
BTf	730.9	752.7 ± 4.6 (experiment)[79] 489.5 (B3LYP/6-31G*)[168]
NNFF	521.2	
DNBF	514.7	525.6 (CBS-4M) [169]

^a W1-F12 atomization energies of the reactants and DLPNO-CCSD(T) enthalpy of the isodesmic reaction SMX + C₄H₁₀ -> ETN + C₆H₁₂(NO₂)₂. ^b W1-F12 atomization energies of the reactants and DLPNO-CCSD(T) enthalpy of the isodesmic reaction DNAAF + (azoxy bisfurazan) -> 2 (azoxy bis(furazano)nitraminofurazan).

The raw computational data: B3LYP-D3BJ/def2-TZVPP geometries, zero-point vibrational energies (unscaled) and thermal corrections to thermodynamic potentials, W1-F12 and W2-F12 electronic energies. The species are named in accordance with Table 1 from the manuscript.

ACM

Zero-point correction= 0.072408 (Hartree/Particle)
 Thermal correction to Energy= 0.077721
 Thermal correction to Enthalpy= 0.078665
 Thermal correction to Gibbs Free Energy= 0.043873

Electronic energy:

B3LYP = -209.3125834
 W1-F12 = -209.31995003
 W2-F12 = -209.32102852

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.075472	0.145704	-0.002324
2	8	0	-0.352171	1.329201	0.000092
3	6	0	1.358981	-0.346721	-0.000251
4	1	0	1.886660	0.119859	-0.830609
5	1	0	1.837704	-0.013833	0.920423
6	1	0	1.452321	-1.429241	-0.077571
7	7	0	-1.035135	-0.823209	-0.000501
8	1	0	-2.000258	-0.540754	0.007793
9	1	0	-0.814171	-1.801080	-0.001817

Urea

Zero-point correction= 0.062932 (Hartree/Particle)

Thermal correction to Energy= 0.067406

Thermal correction to Enthalpy= 0.068350

Thermal correction to Gibbs Free Energy= 0.037275

Electronic energy:

B3LYP = -225.3828217

W1-F12 = -225.39371507

W2-F12 = -225.39481251

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	0.000000	0.143468
2	8	0	0.000000	0.000000	1.357738
3	7	0	0.000000	1.160155	-0.608746
4	1	0	-0.200216	1.988011	-0.073352
5	1	0	-0.411182	1.136624	-1.526779
6	7	0	0.000000	-1.160155	-0.608746

7	1	0	0.411182	-1.136624	-1.526779
8	1	0	0.200216	-1.988011	-0.073352

Val

Zero-point correction= 0.162341 (Hartree/Particle)
 Thermal correction to Energy= 0.172007
 Thermal correction to Enthalpy= 0.172951
 Thermal correction to Gibbs Free Energy= 0.127638

Electronic energy:

B3LYP = -402.574404
 W1-F12 = -402.57033900
 W2-F12 = -402.57221050

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.271138	0.393450	-0.065944
2	1	0	1.758162	1.312634	-0.403340
3	6	0	1.261501	0.408903	1.460561
4	1	0	0.847964	-0.520300	1.859203
5	1	0	2.274708	0.509855	1.850918
6	1	0	0.665239	1.241157	1.831997
7	6	0	2.065306	-0.784328	-0.626713
8	1	0	1.665104	-1.738331	-0.288789
9	1	0	2.057357	-0.785772	-1.718621
10	1	0	3.104575	-0.720419	-0.301387
11	6	0	-0.165970	0.518443	-0.645461
12	1	0	-0.114480	0.241526	-1.707309
13	6	0	-1.153057	-0.456488	-0.030018
14	8	0	-2.178837	-0.149937	0.520867
15	8	0	-0.780683	-1.745732	-0.186656
16	1	0	-1.468683	-2.286981	0.226151
17	7	0	-0.665765	1.871665	-0.439139

18	1	0	-0.463866	2.463428	-1.231391
19	1	0	-1.663084	1.867025	-0.271693

BCA

Zero-point correction= 0.114475 (Hartree/Particle)

Thermal correction to Energy= 0.121645

Thermal correction to Enthalpy= 0.122589

Thermal correction to Gibbs Free Energy= 0.082332

Electronic energy:

B3LYP = -421.025426

W1-F12 = -421.01269127

W2-F12 = -421.01464817

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.431375	-1.717610	0.000000
2	1	0	2.422436	-2.150046	0.000000
3	6	0	1.277963	-0.340234	0.000000
4	6	0	0.000000	0.220347	0.000000
5	8	0	-1.395968	2.135370	0.000000
6	8	0	0.820598	2.464456	0.000000
7	6	0	0.310323	-2.542054	0.000000
8	6	0	-0.964574	-1.986783	0.000000
9	6	0	-1.123001	-0.608227	0.000000
10	1	0	0.430618	-3.617077	0.000000
11	1	0	-1.834973	-2.628437	0.000000
12	1	0	-2.109707	-0.170716	0.000000
13	1	0	2.133763	0.319076	0.000000
14	6	0	-0.112535	1.699090	0.000000
15	1	0	-1.356491	3.101413	0.000000

NPH

Zero-point correction= 0.145946 (Hartree/Particle)
Thermal correction to Energy= 0.152816
Thermal correction to Enthalpy= 0.153760
Thermal correction to Gibbs Free Energy= 0.116005

Electronic energy:

B3LYP = -386.0731871
W1-F12 = -386.00216480
W2-F12 = -386.00298561

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	2.422511	0.705726
2	6	0	0.000000	1.239453	1.396737
3	6	0	0.000000	0.000000	0.713642
4	6	0	0.000000	0.000000	-0.713642
5	6	0	0.000000	1.239453	-1.396737
6	6	0	0.000000	2.422511	-0.705726
7	1	0	0.000000	-1.236812	2.479353
8	1	0	0.000000	3.362991	1.240009
9	1	0	0.000000	1.236812	2.479353
10	6	0	0.000000	-1.239453	1.396737
11	6	0	0.000000	-1.239453	-1.396737
12	1	0	0.000000	1.236812	-2.479353
13	1	0	0.000000	3.362991	-1.240009
14	6	0	0.000000	-2.422511	-0.705726
15	6	0	0.000000	-2.422511	0.705726
16	1	0	0.000000	-1.236812	-2.479353
17	1	0	0.000000	-3.362991	-1.240009
18	1	0	0.000000	-3.362991	1.240009

ADM

Zero-point correction= 0.241082 (Hartree/Particle)
Thermal correction to Energy= 0.248139
Thermal correction to Enthalpy= 0.249083
Thermal correction to Gibbs Free Energy= 0.212597

Electronic energy:

B3LYP = -390.9193707
W1-F12 = -390.85144214
W2-F12 = -390.85182455

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	0.000000	1.772083
2	1	0	0.621614	0.621614	2.422078
3	1	0	-0.621614	-0.621614	2.422078
4	6	0	0.888119	-0.888119	0.888119
5	1	0	1.518891	-1.518891	1.518891
6	6	0	-0.888119	0.888119	0.888119
7	1	0	-1.518891	1.518891	1.518891
8	6	0	0.000000	1.772083	0.000000
9	1	0	0.621614	2.422078	0.621614
10	1	0	-0.621614	2.422078	-0.621614
11	6	0	1.772083	0.000000	0.000000
12	1	0	2.422078	-0.621614	-0.621614
13	1	0	2.422078	0.621614	0.621614
14	6	0	0.888119	0.888119	-0.888119
15	1	0	1.518891	1.518891	-1.518891
16	6	0	0.000000	0.000000	-1.772083
17	1	0	-0.621614	0.621614	-2.422078
18	1	0	0.621614	-0.621614	-2.422078
19	6	0	0.000000	-1.772083	0.000000
20	1	0	-0.621614	-2.422078	0.621614

21	1	0	0.621614	-2.422078	-0.621614
22	6	0	-1.772083	0.000000	0.000000
23	1	0	-2.422078	-0.621614	0.621614
24	1	0	-2.422078	0.621614	-0.621614
25	6	0	-0.888119	-0.888119	-0.888119
26	1	0	-1.518891	-1.518891	-1.518891

ANT

Zero-point correction= 0.192113 (Hartree/Particle)

Thermal correction to Energy= 0.201642

Thermal correction to Enthalpy= 0.202586

Thermal correction to Gibbs Free Energy= 0.158547

Electronic energy:

B3LYP = -539.7814526

W1-F12 = -539.68570002

W2-F12 = -539.68679622

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	3.643606	0.710558
2	6	0	0.000000	2.468408	1.401217
3	6	0	0.000000	1.217729	0.719568
4	6	0	0.000000	1.217729	-0.719568
5	6	0	0.000000	2.468408	-1.401217
6	6	0	0.000000	3.643606	-0.710558
7	6	0	0.000000	0.000000	1.398166
8	6	0	0.000000	0.000000	-1.398166
9	6	0	0.000000	-1.217729	-0.719568
10	6	0	0.000000	-1.217729	0.719568
11	6	0	0.000000	-2.468408	1.401217
12	1	0	0.000000	-2.466419	2.483718
13	6	0	0.000000	-3.643606	0.710558

14	6	0	0.000000	-3.643606	-0.710558
15	6	0	0.000000	-2.468408	-1.401217
16	1	0	0.000000	0.000000	2.481517
17	1	0	0.000000	4.585927	1.241494
18	1	0	0.000000	2.466419	2.483718
19	1	0	0.000000	2.466419	-2.483718
20	1	0	0.000000	4.585927	-1.241494
21	1	0	0.000000	0.000000	-2.481517
22	1	0	0.000000	-4.585927	1.241494
23	1	0	0.000000	-4.585927	-1.241494
24	1	0	0.000000	-2.466419	-2.483718

tAB

Zero-point correction= 0.188765 (Hartree/Particle)
 Thermal correction to Energy= 0.199755
 Thermal correction to Enthalpy= 0.200700
 Thermal correction to Gibbs Free Energy= 0.149455

Electronic energy:

B3LYP = -573.0244777

W1-F12 = -572.94482100

W2-F12 = -572.94604016

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.188045	4.519639	0.000000
2	6	0	1.099487	3.997333	0.000000
3	6	0	1.287357	2.622075	0.000000
4	6	0	0.188045	1.763973	0.000000
5	6	0	-1.108601	2.290374	0.000000
6	6	0	-1.288436	3.662265	0.000000
7	1	0	-0.338348	5.590787	0.000000
8	1	0	1.954518	4.659436	0.000000

9	1	0	2.277557	2.188081	0.000000
10	1	0	-1.949395	1.613471	0.000000
11	1	0	-2.289549	4.072379	0.000000
12	7	0	0.492373	0.383116	0.000000
13	7	0	-0.492373	-0.383116	0.000000
14	6	0	-0.188045	-1.763973	0.000000
15	6	0	1.108601	-2.290374	0.000000
16	6	0	-1.287357	-2.622075	0.000000
17	6	0	1.288436	-3.662265	0.000000
18	1	0	1.949395	-1.613471	0.000000
19	6	0	-1.099487	-3.997333	0.000000
20	1	0	-2.277557	-2.188081	0.000000
21	6	0	0.188045	-4.519639	0.000000
22	1	0	2.289549	-4.072379	0.000000
23	1	0	-1.954518	-4.659436	0.000000
24	1	0	0.338348	-5.590787	0.000000

HPX

Zero-point correction= 0.099172 (Hartree/Particle)
 Thermal correction to Energy= 0.105944
 Thermal correction to Enthalpy= 0.106888
 Thermal correction to Gibbs Free Energy= 0.067727

Electronic energy:

B3LYP = -487.4084128
 W1-F12 = -487.40981432
 W2-F12 = -487.41148885

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	7	0	1.640451	-1.042847	0.000000
2	6	0	0.000000	0.622157	0.000000
3	6	0	-0.878734	-0.456292	0.000000

4	7	0	-0.589942	-1.781682	0.000000
5	8	0	2.358831	1.131919	0.000000
6	7	0	-0.692923	1.810337	0.000000
7	7	0	-2.127451	0.100479	0.000000
8	6	0	1.418430	0.369400	0.000000
9	6	0	0.686559	-2.010858	0.000000
10	1	0	1.046714	-3.031955	0.000000
11	1	0	2.614185	-1.309535	0.000000
12	1	0	-2.994926	-0.408410	0.000000
13	6	0	-1.952762	1.466326	0.000000
14	1	0	-2.788530	2.146137	0.000000

TPL

Zero-point correction= 0.158870 (Hartree/Particle)
 Thermal correction to Energy= 0.170354
 Thermal correction to Enthalpy= 0.171298
 Thermal correction to Gibbs Free Energy= 0.120662

Electronic energy:

B3LYP = -641.360894
 W1-F12 = -641.36135072
 W2-F12 = -641.36403980

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	7	0	-0.743402	-1.470007	0.000000
2	6	0	0.000000	0.767858	0.000000
3	6	0	1.299940	0.319032	0.000000
4	7	0	1.570884	-1.020167	0.000000
5	8	0	-2.304841	0.213648	0.000000
6	7	0	0.098107	2.148904	0.000000
7	7	0	2.187231	1.340135	0.000000
8	6	0	-1.128258	-0.109604	0.000000

9	6	0	0.567490	-1.971568	0.000000
10	6	0	1.421896	2.426366	0.000000
11	1	0	1.789933	3.438933	0.000000
12	8	0	0.822990	-3.158466	0.000000
13	6	0	-1.837431	-2.441283	0.000000
14	1	0	-2.457881	-2.296362	0.881736
15	1	0	-2.457881	-2.296362	-0.881736
16	1	0	-1.402810	-3.433043	0.000000
17	1	0	2.516923	-1.364457	0.000000
18	6	0	-1.004178	3.095083	0.000000
19	1	0	-1.623278	2.950410	-0.882273
20	1	0	-1.623278	2.950410	0.882273
21	1	0	-0.593416	4.101655	0.000000

4-NPA

Zero-point correction= 0.105076 (Hartree/Particle)

Thermal correction to Energy= 0.114593

Thermal correction to Enthalpy= 0.115537

Thermal correction to Gibbs Free Energy= 0.068828

Electronic energy:

B3LYP = -600.61401

W1-F12 = -600.61748790

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.142698	0.416970	0.000000
2	1	0	2.113959	0.891997	0.000000
3	6	0	0.000000	1.195969	0.000000
4	6	0	-1.242007	0.576780	0.000000
5	7	0	-2.453102	1.404803	0.000000
6	8	0	-3.532193	0.831414	0.000000
7	8	0	-2.310989	2.618686	0.000000

8	6	0	1.033775	-0.977281	0.000000
9	6	0	-0.224493	-1.584483	0.000000
10	6	0	-1.366158	-0.806907	0.000000
11	1	0	-0.286520	-2.662638	0.000000
12	1	0	-2.348295	-1.252396	0.000000
13	1	0	0.056934	2.273011	0.000000
14	7	0	2.140864	-1.845463	0.000000
15	7	0	3.270338	-1.353768	0.000000
16	7	0	4.348541	-1.030867	0.000000

24-DNT

Zero-point correction= 0.131766 (Hartree/Particle)

Thermal correction to Energy= 0.142772

Thermal correction to Enthalpy= 0.143716

Thermal correction to Gibbs Free Energy= 0.093613

Electronic energy:

B3LYP = -680.8787573

W1-F12 = -680.90415550

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.212544	1.197388	-0.047244
2	6	0	-0.954979	-0.179976	-0.008104
3	6	0	0.326751	-0.705559	0.006674
4	6	0	1.392823	0.171093	0.008027
5	6	0	1.201858	1.543634	-0.031483
6	6	0	-0.091407	2.034070	-0.068239
7	1	0	0.484668	-1.771208	0.018006
8	1	0	2.056388	2.201849	-0.038065
9	7	0	-2.053970	-1.166154	0.014449
10	8	0	-1.827541	-2.266537	-0.458795
11	8	0	-3.113071	-0.824855	0.516373

12	7	0	2.763127	-0.369913	0.045628
13	8	0	3.682083	0.432829	0.025196
14	8	0	2.890043	-1.581392	0.095016
15	6	0	-2.580222	1.811770	-0.093883
16	1	0	-3.221392	1.321264	-0.823994
17	1	0	-3.081007	1.716170	0.868485
18	1	0	-2.500143	2.867008	-0.346671
19	1	0	-0.248424	3.102503	-0.115108

DNAN

Zero-point correction= 0.136384 (Hartree/Particle)
 Thermal correction to Energy= 0.148525
 Thermal correction to Enthalpy= 0.149469
 Thermal correction to Gibbs Free Energy= 0.096586

Electronic energy:

B3LYP = -756.1147135

W1-F12 = -756.15671416

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.168803	-0.733395	-0.018556
2	6	0	0.651856	0.577607	0.000172
3	6	0	-0.704721	0.822812	-0.003680
4	6	0	-1.576140	-0.252772	0.015682
5	6	0	-1.107504	-1.557750	0.012733
6	6	0	0.254583	-1.795033	-0.012978
7	1	0	-1.075943	1.834850	-0.016634
8	1	0	-1.812981	-2.373740	0.023970
9	7	0	1.545176	1.746925	0.029532
10	8	0	1.211177	2.709582	-0.642959
11	8	0	2.531033	1.687547	0.740596
12	7	0	-3.020675	-0.002499	0.036182

13	8	0	-3.760569	-0.974943	0.039263
14	8	0	-3.393497	1.159703	0.048894
15	1	0	0.611095	-2.811938	-0.033025
16	8	0	2.491744	-0.887926	-0.082375
17	6	0	3.041241	-2.202387	-0.131646
18	1	0	4.116261	-2.063499	-0.176143
19	1	0	2.781127	-2.766984	0.764851
20	1	0	2.701110	-2.735859	-1.020738

TNT

Zero-point correction= 0.133802 (Hartree/Particle)
 Thermal correction to Energy= 0.147509
 Thermal correction to Enthalpy= 0.148453
 Thermal correction to Gibbs Free Energy= 0.091568

Electronic energy:

B3LYP = -885.4577427

W1-F12 = -885.52480482

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.054344	-1.266422	0.000000
2	6	0	-0.025558	-0.505740	-1.176724
3	6	0	0.011003	0.876507	-1.204430
4	6	0	0.042931	1.551012	0.000000
5	6	0	0.011003	0.876507	1.204430
6	6	0	-0.025558	-0.505740	1.176724
7	1	0	0.012558	1.412350	-2.139870
8	1	0	0.012558	1.412350	2.139870
9	7	0	-0.023114	-1.169259	-2.498563
10	8	0	-0.693026	-0.653885	-3.375266
11	8	0	0.667986	-2.164950	-2.622937
12	7	0	-0.023114	-1.169259	2.498563

13	8	0	-0.693026	-0.653885	3.375266
14	8	0	0.667986	-2.164950	2.622937
15	7	0	0.102694	3.026042	0.000000
16	8	0	0.125293	3.581568	1.083737
17	8	0	0.125293	3.581568	-1.083737
18	6	0	-0.189987	-2.760432	0.000000
19	1	0	-0.717121	-3.102623	-0.884928
20	1	0	0.792877	-3.229993	0.000000
21	1	0	-0.717121	-3.102623	0.884928

Tetryl

Zero-point correction= 0.152519 (Hartree/Particle)

Thermal correction to Energy= 0.170022

Thermal correction to Enthalpy= 0.170966

Thermal correction to Gibbs Free Energy= 0.105005

Electronic energy:

B3LYP = -1145.3921458

W1-F12 = -1145.49793952

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.508001	0.130517	0.139397
2	6	0	0.125499	-1.120284	0.171903
3	6	0	1.493345	-1.260192	0.057015
4	6	0	2.268642	-0.117083	-0.024657
5	6	0	1.709521	1.144991	-0.019888
6	6	0	0.331134	1.245047	0.027735
7	1	0	1.946470	-2.238608	0.055825
8	1	0	2.327388	2.026813	-0.077795
9	7	0	-0.644016	-2.365988	0.389758
10	8	0	-0.243854	-3.360380	-0.178135
11	8	0	-1.588061	-2.307121	1.157749

12	7	0	-0.231526	2.609417	-0.080432
13	8	0	-1.197454	2.763699	-0.802787
14	8	0	0.340228	3.480968	0.549087
15	7	0	3.738014	-0.250184	-0.112946
16	8	0	4.384127	0.779484	-0.184409
17	8	0	4.193471	-1.378959	-0.106586
18	7	0	-1.892475	0.289310	0.251947
19	7	0	-2.708080	-0.338530	-0.676532
20	8	0	-2.151593	-0.969937	-1.559442
21	8	0	-3.900582	-0.184509	-0.522732
22	6	0	-2.541059	1.037878	1.318875
23	1	0	-1.762537	1.404402	1.982261
24	1	0	-3.109372	1.872487	0.916655
25	1	0	-3.200112	0.375528	1.876262

FOX-7

Zero-point correction= 0.091023 (Hartree/Particle)

Thermal correction to Energy= 0.100337

Thermal correction to Enthalpy= 0.101282

Thermal correction to Gibbs Free Energy= 0.056861

Electronic energy:

B3LYP = -598.6044501

W1-F12 = -598.66029515

W2-F12 = -598.66334834

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	0.000000	1.368590
2	6	0	0.000000	0.000000	-0.050505
3	7	0	0.231326	1.195182	-0.801368
4	8	0	0.682014	1.115405	-1.925912

5	8	0	0.000000	2.286983	-0.244575
6	7	0	-0.231326	-1.195182	-0.801368
7	8	0	-0.682014	-1.115405	-1.925912
8	8	0	0.000000	-2.286983	-0.244575
9	7	0	-0.034495	1.143654	2.063831
10	1	0	0.186501	1.140479	3.042979
11	1	0	-0.064691	2.003541	1.529422
12	7	0	0.034495	-1.143654	2.063831
13	1	0	-0.186501	-1.140479	3.042979
14	1	0	0.064691	-2.003541	1.529422

TNAA

Zero-point correction= 0.080971 (Hartree/Particle)

Thermal correction to Energy= 0.095214

Thermal correction to Enthalpy= 0.096158

Thermal correction to Gibbs Free Energy= 0.037432

Electronic energy:

B3LYP = -1027.5733512

W1-F12 = -1027.73720234

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.709575	-0.413988	0.064722
2	6	0	-0.751136	0.000188	0.001913
3	7	0	-0.921456	1.490130	-0.392216
4	8	0	-1.259127	1.723819	-1.522665
5	8	0	-0.689885	2.258637	0.505552
6	7	0	-1.441873	-0.890968	-1.059042
7	8	0	-2.505693	-1.380074	-0.785771
8	8	0	-0.799479	-0.997205	-2.076717
9	7	0	1.574278	0.535477	-0.034190
10	7	0	-1.458549	-0.146602	1.363338

11	8	0	-2.577786	0.293273	1.412048
12	8	0	-0.816724	-0.676957	2.240012
13	7	0	2.941791	0.156569	0.009930
14	8	0	3.254022	-1.037984	0.170943
15	8	0	3.721817	1.052973	-0.114641
16	8	0	0.864678	-1.697593	0.190940
17	1	0	1.855456	-1.840564	0.227837

FEFO

Zero-point correction= 0.161405 (Hartree/Particle)
 Thermal correction to Energy= 0.182030
 Thermal correction to Enthalpy= 0.182974
 Thermal correction to Gibbs Free Energy= 0.105651

Electronic energy:

B3LYP = -1365.1863583

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.800938	0.115872	0.605309
2	1	0	-2.061570	0.264751	1.649012
3	1	0	-1.148655	0.929032	0.288224
4	6	0	-3.070723	0.148123	-0.232332
5	7	0	-3.884071	1.451176	-0.033881
6	8	0	-3.435043	2.256128	0.752735
7	8	0	-4.881316	1.547902	-0.708237
8	7	0	-4.026252	-0.996774	0.185694
9	8	0	-4.446637	-0.903527	1.319858
10	8	0	-4.239388	-1.860476	-0.623695
11	9	0	-2.831946	0.037019	-1.540189
12	6	0	2.068989	-0.527747	-0.800153
13	1	0	1.845280	-0.080065	-1.774297
14	1	0	2.464952	-1.532899	-0.957071

15	6	0	3.129396	0.319258	-0.128908
16	9	0	2.767365	1.588524	0.017427
17	7	0	3.483177	-0.258128	1.267770
18	8	0	3.939660	-1.382719	1.236857
19	8	0	3.268129	0.437310	2.223192
20	7	0	4.464028	0.315970	-0.918107
21	8	0	5.341565	1.000011	-0.450013
22	8	0	4.495937	-0.345414	-1.933439
23	8	0	-1.189837	-1.153989	0.478510
24	8	0	0.957668	-0.519119	0.062240
25	6	0	-0.148992	-1.257469	-0.435670
26	1	0	-0.423853	-0.873266	-1.426206
27	1	0	0.094595	-2.320231	-0.498670

NIGU

Zero-point correction= 0.077186 (Hartree/Particle)

Thermal correction to Energy= 0.084022

Thermal correction to Enthalpy= 0.084966

Thermal correction to Gibbs Free Energy= 0.046044

Electronic energy:

B3LYP = -410.0606893

W1-F12 = -410.09054736

W2-F12 = -410.09238290

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.065026	-0.078484	0.005637
2	7	0	1.232293	1.250750	0.022966
3	1	0	2.120519	1.649210	-0.221493
4	1	0	0.379940	1.795839	-0.010786
5	7	0	2.185736	-0.848440	-0.071131
6	1	0	3.051111	-0.482619	0.285830

7	1	0	2.030346	-1.834947	0.052841
8	7	0	-0.062624	-0.769748	0.046336
9	7	0	-1.263902	-0.102441	0.003228
10	8	0	-1.329542	1.142259	0.008227
11	8	0	-2.247032	-0.813186	-0.026978

DNDAP

Zero-point correction= 0.141881 (Hartree/Particle)

Thermal correction to Energy= 0.153517

Thermal correction to Enthalpy= 0.154462

Thermal correction to Gibbs Free Energy= 0.102282

Electronic energy:

B3LYP = -639.1058288

W1-F12 = -639.14978242

W2-F12 = -639.15318517

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.179420	2.036933	0.112451
2	1	0	-1.430566	2.024623	-0.946800
3	1	0	-1.999429	1.601932	0.672153
4	1	0	-1.008665	3.065422	0.423431
5	6	0	0.000000	0.000000	1.147736
6	1	0	0.887765	-0.029486	1.770292
7	1	0	-0.887765	0.029486	1.770292
8	6	0	1.179420	-2.036933	0.112451
9	1	0	1.008665	-3.065422	0.423431
10	1	0	1.430566	-2.024623	-0.946800
11	1	0	1.999429	-1.601932	0.672153
12	7	0	0.000000	-1.226438	0.377894
13	7	0	0.000000	1.226438	0.377894
14	7	0	1.171794	1.605948	-0.212268

15	8	0	1.149390	2.627371	-0.879017
16	8	0	2.152744	0.891764	-0.020530
17	7	0	-1.171794	-1.605948	-0.212268
18	8	0	-2.152744	-0.891764	-0.020530
19	8	0	-1.149390	-2.627371	-0.879017

DNPP

Zero-point correction= 0.150196 (Hartree/Particle)

Thermal correction to Energy= 0.161011

Thermal correction to Enthalpy= 0.161956

Thermal correction to Gibbs Free Energy= 0.111574

Electronic energy:

B3LYP = -677.1899974

W1-F12 = -677.26748870

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.411377	0.641762	-1.183272
2	1	0	0.131996	1.262158	-2.030081
3	1	0	1.476278	0.415002	-1.276657
4	6	0	0.268441	-0.712233	1.257189
5	1	0	1.359566	-0.715687	1.335777
6	1	0	-0.122217	-1.242000	2.122076
7	6	0	-0.411377	-0.641762	-1.183272
8	1	0	-0.131996	-1.262158	-2.030081
9	1	0	-1.476278	-0.415002	-1.276657
10	6	0	-0.268441	0.712233	1.257189
11	1	0	-1.359566	0.715687	1.335777
12	1	0	0.122217	1.242000	2.122076
13	7	0	-0.164517	-1.394407	0.043149
14	7	0	0.164517	1.394407	0.043149
15	7	0	-0.001861	2.750607	-0.024938

16	7	0	0.001861	-2.750607	-0.024938
17	8	0	-0.411377	3.303114	0.987944
18	8	0	0.288838	3.301539	-1.078205
19	8	0	0.411377	-3.303114	0.987944
20	8	0	-0.288838	-3.301539	-1.078205

RDX (aaa conformer)

Zero-point correction= 0.141019 (Hartree/Particle)

Thermal correction to Energy= 0.153647

Thermal correction to Enthalpy= 0.154592

Thermal correction to Gibbs Free Energy= 0.100581

Electronic energy:

B3LYP = -897.8299788

W1-F12 = -897.92329127

DLPNO-CCSD(T) = -896.4187140

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	7	0	0.000000	1.395192	0.808607
2	7	0	-1.208272	-0.697596	0.808607
3	7	0	1.208272	-0.697596	0.808607
4	6	0	-1.230375	0.710358	1.172863
5	6	0	0.000000	-1.420715	1.172863
6	6	0	1.230375	0.710358	1.172863
7	1	0	2.074543	1.197738	0.705339
8	1	0	1.322862	0.763755	2.257673
9	1	0	-2.074543	1.197738	0.705339
10	1	0	0.000000	-2.395475	0.705339
11	1	0	0.000000	-1.527510	2.257673
12	1	0	-1.322862	0.763755	2.257673
13	7	0	0.000000	2.165907	-0.379262
14	8	0	-1.085081	2.489469	-0.812850

15	8	0	1.085081	2.489469	-0.812850
16	7	0	-1.875731	-1.082954	-0.379262
17	8	0	-2.698484	-0.305027	-0.812850
18	8	0	-1.613403	-2.184443	-0.812850
19	7	0	1.875731	-1.082954	-0.379262
20	8	0	2.698484	-0.305027	-0.812850
21	8	0	1.613403	-2.184443	-0.812850

RDX (aae conformer)

Zero-point correction= 0.142593 (Hartree/Particle)
 Thermal correction to Energy= 0.155143
 Thermal correction to Enthalpy= 0.156087
 Thermal correction to Gibbs Free Energy= 0.101490

Electronic energy:

B3LYP = -897.8303348

DLPNO-CCSD(T) = -896.41998406

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	7	0	0.132221	1.253084	0.000000
2	7	0	0.785558	-0.678259	1.209726
3	7	0	0.785558	-0.678259	-1.209726
4	6	0	0.756006	0.764681	1.234562
5	6	0	1.342066	-1.269770	0.000000
6	6	0	0.756006	0.764681	-1.234562
7	1	0	0.193586	1.113846	-2.087198
8	1	0	1.791644	1.112402	-1.306771
9	1	0	0.193586	1.113846	2.087198
10	1	0	1.172724	-2.337162	0.000000
11	1	0	2.410891	-1.061807	0.000000
12	1	0	1.791644	1.112402	1.306771
13	7	0	-0.242253	2.598729	0.000000

14	8	0	-0.403105	3.119892	1.087152
15	8	0	-0.403105	3.119892	-1.087152
16	7	0	-0.337703	-1.341019	1.786411
17	8	0	-1.055792	-0.673684	2.498829
18	8	0	-0.427245	-2.527701	1.565443
19	7	0	-0.337703	-1.341019	-1.786411
20	8	0	-1.055792	-0.673684	-2.498829
21	8	0	-0.427245	-2.527701	-1.565443

HMX (crown conformer)

Zero-point correction= 0.190912 (Hartree/Particle)

Thermal correction to Energy= 0.208261

Thermal correction to Enthalpy= 0.209205

Thermal correction to Gibbs Free Energy= 0.144153

Electronic energy:

B3LYP = -1197.1084085

W1-F12 = -1197.23049313

DLPNO-CCSD(T) = -1195.2254780

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.262117	1.278449	-1.065014
2	1	0	1.282494	1.093693	-2.140853
3	1	0	2.043149	1.981299	-0.803123
4	6	0	-1.262117	-1.278449	-1.065014
5	1	0	-2.043149	-1.981299	-0.803123
6	1	0	-1.282494	-1.093693	-2.140853
7	6	0	1.262117	-1.278449	-1.065014
8	1	0	2.043149	-1.981299	-0.803123
9	1	0	1.282494	-1.093693	-2.140853
10	6	0	-1.262117	1.278449	-1.065014
11	1	0	-1.282494	1.093693	-2.140853

12	1	0	-2.043149	1.981299	-0.803123
13	7	0	0.000000	-1.883775	-0.695320
14	7	0	-1.511585	0.000000	-0.436765
15	7	0	1.511585	0.000000	-0.436765
16	7	0	0.000000	1.883775	-0.695320
17	7	0	-1.895746	0.000000	0.902965
18	7	0	1.895746	0.000000	0.902965
19	7	0	0.000000	3.206973	-0.227568
20	7	0	0.000000	-3.206973	-0.227568
21	8	0	-1.083630	3.725295	-0.052902
22	8	0	1.083630	3.725295	-0.052902
23	8	0	2.043942	1.089626	1.419460
24	8	0	2.043942	-1.089626	1.419460
25	8	0	-1.083630	-3.725295	-0.052902
26	8	0	1.083630	-3.725295	-0.052902
27	8	0	-2.043942	-1.089626	1.419460
28	8	0	-2.043942	1.089626	1.419460

HMX (chair conformer)

Zero-point correction= 0.191510 (Hartree/Particle)

Thermal correction to Energy= 0.208772

Thermal correction to Enthalpy= 0.209716

Thermal correction to Gibbs Free Energy= 0.144247

Electronic energy:

B3LYP = -1197.112482

DLPNO-CCSD(T) = -1195.23082782

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.236143	0.323749	-1.181711
2	6	0	-1.236138	-0.323739	1.181696
3	6	0	-1.131717	1.201500	-0.874425

4	1	0	0.678961	-0.262065	-1.909298
5	1	0	-0.678954	0.262072	1.909283
6	1	0	2.038335	0.852093	-1.686505
7	1	0	-2.038330	-0.852083	1.686490
8	1	0	-1.521471	2.205133	-1.002220
9	1	0	-1.272735	0.651405	-1.803343
10	6	0	1.131719	-1.201496	0.874406
11	1	0	1.272737	-0.651412	1.803331
12	1	0	1.521469	-2.205132	1.002190
13	7	0	1.866903	-0.549146	-0.219207
14	7	0	0.276679	1.249630	-0.598687
15	7	0	-0.276677	-1.249619	0.598666
16	7	0	-1.866899	0.549161	0.219196
17	7	0	-3.231531	0.434530	0.018048
18	8	0	-3.653492	0.953676	-1.003814
19	8	0	-3.888966	-0.122299	0.873930
20	7	0	3.231534	-0.434515	-0.018056
21	8	0	3.888971	0.122317	-0.873934
22	8	0	3.653493	-0.953667	1.003805
23	7	0	0.759140	2.284045	0.190178
24	8	0	-0.066805	3.034820	0.674986
25	8	0	1.966241	2.321741	0.341649
26	7	0	-0.759140	-2.284028	-0.190205
27	8	0	-1.966250	-2.321786	-0.341590
28	8	0	0.066796	-3.034863	-0.674937

DATN

Zero-point correction= 0.245495 (Hartree/Particle)

Thermal correction to Energy= 0.270265

Thermal correction to Enthalpy= 0.271209

Thermal correction to Gibbs Free Energy= 0.185309

Electronic energy:

B3LYP = -1564.9775035

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	4.214140	-0.591672	0.562740
2	1	0	4.272229	-0.357880	1.620528
3	1	0	5.211367	-0.689183	0.144006
4	6	0	0.139310	0.702685	-0.156170
5	1	0	-0.217106	-0.283726	-0.431296
6	1	0	0.544098	1.197792	-1.031298
7	6	0	2.506294	1.279810	0.637348
8	1	0	2.302132	2.193056	0.094161
9	1	0	2.889054	1.516715	1.623192
10	6	0	-2.268961	0.805549	0.697623
11	1	0	-2.199480	0.343614	1.678153
12	1	0	-3.013596	1.592997	0.713732
13	7	0	1.240578	0.582550	0.787601
14	7	0	-0.997190	1.432002	0.349129
15	7	0	3.544347	0.538393	-0.056960
16	7	0	-2.620463	-0.247953	-0.230779
17	7	0	-0.917621	2.808774	0.351897
18	7	0	3.478926	0.508670	-1.448942
19	7	0	-3.262260	0.157035	-1.402569
20	7	0	1.066009	-0.251920	1.864479
21	8	0	-3.155084	1.331830	-1.696603
22	8	0	-3.839693	-0.705770	-2.038290
23	8	0	4.225191	-0.271281	-2.009631
24	8	0	2.702267	1.276541	-1.990608
25	8	0	-0.003896	-0.844581	1.928120
26	8	0	1.971311	-0.330810	2.674652
27	8	0	0.151460	3.286410	-0.005186
28	8	0	-1.886866	3.434047	0.734412
29	7	0	3.442083	-1.835982	0.431547

30	7	0	3.909457	-2.684807	-0.330621
31	7	0	4.240079	-3.542357	-0.978786
32	6	0	-2.918027	-1.583825	0.257088
33	1	0	-2.838897	-2.281878	-0.570883
34	1	0	-2.155149	-1.804350	0.996925
35	7	0	-4.215527	-1.674632	0.934682
36	7	0	-5.146118	-2.146383	0.275175
37	7	0	-6.069257	-2.573896	-0.203558

BC-HMX

Zero-point correction= 0.166186 (Hartree/Particle)
 Thermal correction to Energy= 0.182813
 Thermal correction to Enthalpy= 0.183757
 Thermal correction to Gibbs Free Energy= 0.119309

Electronic energy:

B3LYP = -1195.8942716
 W1-F12 = -1196.02804802

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	1	0	0.046783	-1.231214	-1.336921
2	7	0	-1.380803	-1.118006	0.219225
3	7	0	-1.928293	-2.361552	-0.055135
4	7	0	-1.380765	1.118048	0.219260
5	7	0	2.112384	1.816719	-0.121278
6	7	0	-1.928226	2.361611	-0.055111
7	7	0	1.011336	1.157457	0.506442
8	7	0	2.112299	-1.816812	-0.121247
9	7	0	1.011292	-1.157476	0.506459
10	8	0	3.144069	1.826167	0.511411
11	8	0	-1.163961	-3.209287	-0.467829
12	8	0	-1.163875	3.209321	-0.467818

13	8	0	-3.117669	2.476736	0.177583
14	8	0	1.897995	2.351180	-1.185971
15	8	0	1.897882	-2.351178	-1.185980
16	8	0	3.143998	-1.826315	0.511423
17	8	0	-3.117737	-2.476651	0.177574
18	6	0	-0.094534	0.780525	-0.359941
19	6	0	-0.094554	-0.780515	-0.359944
20	6	0	1.349050	-0.000008	1.336034
21	6	0	-2.307957	0.000034	0.338814
22	1	0	0.046790	1.231222	-1.336919
23	1	0	2.388853	-0.000025	1.632029
24	1	0	0.705881	0.000012	2.212786
25	1	0	-3.058300	0.000059	-0.454668
26	1	0	-2.812220	0.000027	1.300472

ETN

Zero-point correction= 0.157688 (Hartree/Particle)

Thermal correction to Energy= 0.177374

Thermal correction to Enthalpy= 0.178318

Thermal correction to Gibbs Free Energy= 0.101136

Electronic energy:

B3LYP = -1277.7555946

W1-F12 = -1277.94067950

DLPNO-CCSD(T) = -1275.819200983

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.860053	-0.513859	0.194609
2	6	0	-0.664190	0.349801	-0.163217
3	1	0	-1.789192	-1.485908	-0.288771
4	1	0	-1.934216	-0.644554	1.273399
5	6	0	0.664190	-0.349801	0.163217

6	1	0	-0.696790	0.616910	-1.217442
7	6	0	1.860053	0.513859	-0.194609
8	1	0	0.696790	-0.616910	1.217442
9	1	0	1.934216	0.644554	-1.273399
10	1	0	1.789192	1.485908	0.288771
11	8	0	-2.996762	0.205595	-0.300896
12	8	0	-0.710977	1.541921	0.631387
13	8	0	0.710977	-1.541921	-0.631387
14	8	0	2.996762	-0.205595	0.300896
15	7	0	-4.246438	-0.419783	-0.004137
16	8	0	-4.201977	-1.464170	0.589289
17	8	0	-5.176211	0.207936	-0.405817
18	7	0	-0.982372	2.779592	-0.086678
19	8	0	-1.002842	2.733262	-1.282022
20	8	0	-1.127695	3.687842	0.665239
21	7	0	0.982372	-2.779592	0.086678
22	8	0	1.002842	-2.733262	1.282022
23	8	0	1.127695	-3.687842	-0.665239
24	7	0	4.246438	0.419783	0.004137
25	8	0	4.201977	1.464170	-0.589289
26	8	0	5.176211	-0.207936	0.405817

PETN

Zero-point correction= 0.186948 (Hartree/Particle)

Thermal correction to Energy= 0.207781

Thermal correction to Enthalpy= 0.208726

Thermal correction to Gibbs Free Energy= 0.132084

Electronic energy:

B3LYP = -1317.0937455

W1-F12 = -1317.27303193

DLPNO-CCSD(T) = -1315.07923547

Center	Atomic	Atomic	Coordinates (Angstroms)
--------	--------	--------	-------------------------

Number	Number	Type	X	Y	Z
1	6	0	0.867608	0.867630	0.923858
2	1	0	0.246069	1.502191	1.551209
3	1	0	1.501954	0.246116	1.551451
4	8	0	1.681254	1.680979	0.063712
5	7	0	2.562192	2.561914	0.761606
6	8	0	2.526902	2.526855	1.961397
7	8	0	3.223033	3.222529	0.020557
8	6	0	-0.867608	0.867630	-0.923858
9	1	0	-0.246069	1.502191	-1.551209
10	1	0	-1.501954	0.246116	-1.551451
11	6	0	0.867608	-0.867630	-0.923858
12	1	0	0.246069	-1.502191	-1.551209
13	1	0	1.501954	-0.246116	-1.551451
14	6	0	-0.867608	-0.867630	0.923858
15	1	0	-1.501954	-0.246116	1.551451
16	1	0	-0.246069	-1.502191	1.551209
17	8	0	-1.681254	-1.680979	0.063712
18	8	0	1.681254	-1.680979	-0.063712
19	8	0	-1.681254	1.680979	-0.063712
20	7	0	2.562192	-2.561914	-0.761606
21	8	0	3.223033	-3.222529	-0.020557
22	8	0	2.526902	-2.526855	-1.961397
23	7	0	-2.562192	-2.561914	0.761606
24	8	0	-2.526902	-2.526855	1.961397
25	8	0	-3.223033	-3.222529	0.020557
26	7	0	-2.562192	2.561914	-0.761606
27	8	0	-3.223033	3.222529	-0.020557
28	8	0	-2.526902	2.526855	-1.961397
29	6	0	0.000000	0.000000	0.000000

SMX

Zero-point correction= 0.219298 (Hartree/Particle)

Thermal correction to Energy= 0.246214

Thermal correction to Enthalpy= 0.247158

Thermal correction to Gibbs Free Energy= 0.156395

Electronic energy:

B3LYP = -1765.5783172

DLPNO-CCSD(T) = -1762.89786073

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.617727	0.445539	1.183629
2	6	0	-0.680481	-0.223720	0.151063
3	1	0	-1.758984	-0.248095	2.012736
4	1	0	-1.185991	1.360518	1.572915
5	6	0	0.821987	-0.112597	0.627505
6	6	0	1.732945	-1.111482	-0.084288
7	1	0	1.431911	-2.132367	0.122781
8	1	0	1.720211	-0.924989	-1.154973
9	8	0	-2.895015	0.721083	0.638893
10	8	0	3.035222	-0.873983	0.459576
11	7	0	-3.284909	2.132857	0.675613
12	8	0	-2.587750	2.885663	1.290062
13	8	0	-4.296368	2.307532	0.081130
14	7	0	4.049150	-1.730483	-0.095367
15	8	0	3.687621	-2.520596	-0.922098
16	8	0	5.122214	-1.512932	0.368409
17	7	0	-0.763813	0.564096	-1.189887
18	8	0	-1.027875	1.745025	-1.112429
19	8	0	-0.531414	-0.051494	-2.205011
20	7	0	0.811795	-0.499118	2.139479
21	8	0	0.729285	-1.680296	2.405813
22	8	0	0.852071	0.410068	2.939827

23	6	0	-1.059710	-1.676659	-0.131248
24	1	0	-0.476177	-2.076630	-0.953032
25	1	0	-0.890101	-2.273324	0.762263
26	6	0	1.389512	1.319530	0.556303
27	1	0	0.649836	2.063789	0.827314
28	1	0	2.243793	1.407503	1.219868
29	8	0	1.801065	1.495880	-0.802063
30	8	0	-2.444433	-1.680264	-0.480131
31	7	0	2.017894	2.876839	-1.170784
32	8	0	1.870593	3.692645	-0.306661
33	8	0	2.328119	2.983923	-2.312678
34	7	0	-2.928335	-2.986606	-0.832009
35	8	0	-2.136913	-3.888672	-0.770217
36	8	0	-4.077187	-2.973976	-1.137042

C4H10

Zero-point correction= 0.131652 (Hartree/Particle)
 Thermal correction to Energy= 0.137428
 Thermal correction to Enthalpy= 0.138372
 Thermal correction to Gibbs Free Energy= 0.103652

Electronic energy:

B3LYP = -158.5388223
 W1-F12 = -158.49528974
 DLPNO-CCSD(T) = -158.204411

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.876958	-0.518934	0.186545
2	6	0	-0.664093	0.331020	-0.180313
3	1	0	-1.850704	-1.482178	-0.327235
4	1	0	-1.905191	-0.719859	1.259519
5	6	0	0.664093	-0.331020	0.180313

6	1	0	-0.677138	0.546023	-1.253194
7	6	0	1.876958	0.518934	-0.186545
8	1	0	0.677138	-0.546023	1.253194
9	1	0	1.905191	0.719859	-1.259519
10	1	0	1.850704	1.482178	0.327235
11	1	0	-2.810741	-0.023990	-0.083306
12	1	0	-0.728990	1.300533	0.322933
13	1	0	0.728990	-1.300533	-0.322933
14	1	0	2.810741	0.023990	0.083306

C6H12(NO2)2

Zero-point correction= 0.193757 (Hartree/Particle)

Thermal correction to Energy= 0.206500

Thermal correction to Enthalpy= 0.207444

Thermal correction to Gibbs Free Energy= 0.156269

Electronic energy:

B3LYP = -646.3950873

W1-F12 = -646.42229632

DLPNO-CCSD(T) = -645.31409231

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.571160	1.496583	-1.034853
2	6	0	-0.140740	0.778711	0.104074
3	1	0	1.637458	1.530683	-0.820823
4	1	0	0.420717	1.014294	-1.995248
5	6	0	0.140740	-0.778711	0.104074
6	6	0	-0.140740	-1.469092	1.434533
7	1	0	0.464773	-1.060601	2.236051
8	1	0	-1.191646	-1.353074	1.684049
9	7	0	-1.675048	0.901436	-0.133545
10	8	0	-2.058741	1.310649	-1.210188

11	8	0	-2.405210	0.563681	0.782026
12	7	0	1.675048	-0.901436	-0.133545
13	8	0	2.405210	-0.563681	0.782026
14	8	0	2.058741	-1.310649	-1.210188
15	6	0	0.140740	1.469092	1.434533
16	1	0	-0.464773	1.060601	2.236051
17	1	0	1.191646	1.353074	1.684049
18	6	0	-0.571160	-1.496583	-1.034853
19	1	0	-0.420717	-1.014294	-1.995248
20	1	0	-0.209368	-2.518640	-1.109029
21	1	0	0.209368	2.518640	-1.109029
22	1	0	-0.079879	2.532108	1.342591
23	1	0	0.079879	-2.532108	1.342591
24	1	0	-1.637458	-1.530683	-0.820823

PMDA

Zero-point correction= 0.179448 (Hartree/Particle)

Thermal correction to Energy= 0.186623

Thermal correction to Enthalpy= 0.187567

Thermal correction to Gibbs Free Energy= 0.148394

Electronic energy:

B3LYP = -345.4819524

W1-F12 = -345.44356115

W2-F12 = -345.44408480

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.288110	-1.261780	0.235953
2	6	0	-2.080937	0.015584	-0.049563
3	6	0	-1.268763	1.264309	0.300515
4	6	0	0.078791	1.278995	-0.428676
5	6	0	0.838592	-0.002034	-0.171095

6	6	0	0.056150	-1.252946	-0.498164
7	1	0	-1.833089	2.166678	0.058900
8	1	0	-2.348172	0.043031	-1.111252
9	1	0	-3.019112	0.009147	0.508378
10	1	0	-1.111627	-1.341694	1.313070
11	1	0	-1.864210	-2.142451	-0.052707
12	1	0	-0.085777	1.379600	-1.506312
13	1	0	0.684388	2.126046	-0.103115
14	1	0	-0.115854	-1.278902	-1.578498
15	1	0	0.656213	-2.130414	-0.252215
16	1	0	-1.083514	1.285569	1.377926
17	7	0	1.695220	0.022744	0.997192
18	1	0	1.791343	-0.921993	1.362932
19	7	0	2.261731	-0.067843	-0.393476
20	1	0	2.616414	0.868312	-0.576945

24-DNI

Zero-point correction= 0.076174 (Hartree/Particle)
 Thermal correction to Energy= 0.084707
 Thermal correction to Enthalpy= 0.085651
 Thermal correction to Gibbs Free Energy= 0.041197

Electronic energy:

B3LYP = -635.494867
 W1-F12 = -635.56009325
 W2-F12 = -635.56316575

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.067080	-0.111881	0.000000
2	7	0	0.000000	0.717108	0.000000
3	6	0	0.695911	-1.440644	0.000000
4	1	0	1.274562	-2.345026	0.000000

5	6	0	-1.019566	-0.091374	0.000000
6	7	0	-0.660829	-1.406167	0.000000
7	7	0	2.435135	0.368752	0.000000
8	8	0	3.299132	-0.500986	0.000000
9	8	0	2.622581	1.566518	0.000000
10	7	0	-2.420178	0.273588	0.000000
11	8	0	-3.202845	-0.674699	0.000000
12	8	0	-2.706691	1.448199	0.000000
13	1	0	-1.311433	-2.176806	0.000000

45-DNI

Zero-point correction= 0.076107 (Hartree/Particle)

Thermal correction to Energy= 0.084814

Thermal correction to Enthalpy= 0.085758

Thermal correction to Gibbs Free Energy= 0.040478

Electronic energy:

B3LYP = -635.4867166

W1-F12 = -635.55194529

W2-F12 = -635.55502848

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.645816	0.478874	-0.029328
2	7	0	-0.872649	1.805394	-0.067285
3	6	0	0.703138	0.213526	-0.005310
4	6	0	0.323789	2.366493	-0.072310
5	7	0	1.308499	1.441593	-0.031448
6	7	0	-1.758112	-0.468614	-0.047608
7	8	0	-1.785590	-1.254328	-0.974816
8	8	0	-2.564529	-0.371042	0.854991
9	1	0	2.308677	1.571515	-0.011794
10	7	0	1.476785	-0.979523	0.090656

11	8	0	2.695851	-0.839223	0.016931
12	8	0	0.880272	-2.028169	0.246494
13	1	0	0.514961	3.425280	-0.105527

34-DNP

Zero-point correction= 0.075452 (Hartree/Particle)
 Thermal correction to Energy= 0.084284
 Thermal correction to Enthalpy= 0.085228
 Thermal correction to Gibbs Free Energy= 0.038934

Electronic energy:

B3LYP = -635.4675432

W1-F12 = -635.53259701

W2-F12 = -635.53561033

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.354282	1.493941	0.013982
2	7	0	0.331002	2.362177	0.035529
3	1	0	0.376872	3.367117	0.054127
4	1	0	2.389622	1.782185	0.000312
5	6	0	0.771734	0.242968	0.000007
6	6	0	-0.615269	0.489403	0.010349
7	7	0	-0.882522	1.772803	0.033454
8	7	0	1.466256	-1.009248	-0.072358
9	8	0	0.789963	-2.008671	-0.249331
10	8	0	2.683915	-0.975277	0.036584
11	7	0	-1.716783	-0.474622	0.051126
12	8	0	-2.412740	-0.550579	-0.940065
13	8	0	-1.838219	-1.098590	1.085972

35-DNP

Zero-point correction= 0.075446 (Hartree/Particle)

Thermal correction to Energy= 0.084071
 Thermal correction to Enthalpy= 0.085016
 Thermal correction to Gibbs Free Energy= 0.040315

Electronic energy:

B3LYP = -635.4753566
 W1-F12 = -635.53994866
 W2-F12 = -635.54292669

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.069189	-0.057812	0.000000
2	7	0	-0.587450	-1.325874	0.000000
3	1	0	-1.133329	-2.174049	0.000000
4	6	0	0.000000	0.804696	0.000000
5	6	0	1.090107	-0.077418	0.000000
6	7	0	0.735517	-1.357227	0.000000
7	7	0	2.496547	0.300336	0.000000
8	8	0	3.326865	-0.584673	0.000000
9	8	0	2.717480	1.502756	0.000000
10	7	0	-2.485233	0.186183	0.000000
11	8	0	-2.853005	1.345327	0.000000
12	8	0	-3.204408	-0.806433	0.000000
13	1	0	-0.003295	1.877512	0.000000

1-ADP

Zero-point correction= 0.092591 (Hartree/Particle)
 Thermal correction to Energy= 0.102504
 Thermal correction to Enthalpy= 0.103448
 Thermal correction to Gibbs Free Energy= 0.056004

Electronic energy:

B3LYP = -690.8221844
 W1-F12 = -690.88555275

W2-F12 = -690.88865474

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.935727	-0.255372	-0.033880
2	7	0	-0.577540	1.066508	-0.025309
3	6	0	0.232754	-0.999883	-0.023670
4	6	0	1.217654	-0.024153	-0.010692
5	7	0	0.722796	1.222560	-0.019690
6	7	0	2.649257	-0.241540	0.019117
7	8	0	3.375255	0.732842	0.057300
8	8	0	3.009311	-1.410552	0.004287
9	7	0	-2.287033	-0.715671	0.016927
10	8	0	-2.463321	-1.915194	-0.079685
11	8	0	-3.174219	0.124459	0.169460
12	1	0	0.332820	-2.067892	-0.012516
13	7	0	-1.390985	2.177300	-0.175324
14	1	0	-2.294031	1.969823	0.236643
15	1	0	-0.918548	2.957979	0.264378

5-ADP

Zero-point correction= 0.091775 (Hartree/Particle)
Thermal correction to Energy= 0.102068
Thermal correction to Enthalpy= 0.103012
Thermal correction to Gibbs Free Energy= 0.054777

Electronic energy:

B3LYP = -690.8645528

W1-F12 = -690.92793711

W2-F12 = -690.93110247

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z

1	6	0	-1.510256	-0.841756	0.036166
2	7	0	-0.857820	-2.024010	0.025499
3	1	0	-1.252750	-2.946508	0.076857
4	6	0	-0.498501	0.128946	0.025753
5	6	0	0.706698	-0.612511	0.005484
6	7	0	0.504819	-1.897581	0.008122
7	7	0	-0.706681	1.518669	-0.089676
8	8	0	0.260123	2.239555	-0.270114
9	8	0	-1.878223	1.917102	-0.013793
10	7	0	2.079694	-0.106718	0.060447
11	8	0	2.801226	-0.367570	-0.880711
12	8	0	2.375103	0.517836	1.060167
13	7	0	-2.846780	-0.689696	0.081683
14	1	0	-3.164569	0.264957	-0.007928
15	1	0	-3.448802	-1.426557	-0.240258

TNP

Zero-point correction= 0.077434 (Hartree/Particle)
 Thermal correction to Energy= 0.088875
 Thermal correction to Enthalpy= 0.089820
 Thermal correction to Gibbs Free Energy= 0.037794

Electronic energy:

B3LYP = -840.0388949

W1-F12 = -840.14418816

W2-F12 = -840.14846355

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.085713	-0.612046	0.000000
2	7	0	-0.617159	-1.882590	0.000000
3	1	0	-1.179218	-2.721518	0.000000

4	6	0	0.011341	0.216899	0.000000
5	6	0	1.097553	-0.672529	0.000000
6	7	0	0.703463	-1.937848	0.000000
7	7	0	0.013353	1.679327	0.000000
8	8	0	0.011341	2.211264	1.088321
9	8	0	0.011341	2.211264	-1.088321
10	7	0	2.510583	-0.324740	0.000000
11	8	0	2.745768	0.873321	0.000000
12	8	0	3.317171	-1.228228	0.000000
13	7	0	-2.494779	-0.334351	0.000000
14	8	0	-2.824708	0.834524	0.000000
15	8	0	-3.231926	-1.311022	0.000000

MTNP

Zero-point correction= 0.104922 (Hartree/Particle)

Thermal correction to Energy= 0.117971

Thermal correction to Enthalpy= 0.118915

Thermal correction to Gibbs Free Energy= 0.063383

Electronic energy:

B3LYP = -879.3715799

W1-F12 = -879.46730380

W2-F12 = -879.47174409

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.792580	-0.712659	0.000000
2	7	0	1.742020	0.269165	0.000000
3	6	0	-0.439008	-0.090708	0.000000
4	6	0	-0.126804	1.269248	0.000000
5	7	0	1.184462	1.469733	0.000000
6	7	0	-1.756406	-0.726287	0.000000
7	8	0	-2.236971	-0.957065	1.087972

8	8	0	-2.236971	-0.957065	-1.087972
9	7	0	-1.054987	2.386814	0.000000
10	8	0	-2.236971	2.079630	0.000000
11	8	0	-0.593809	3.508178	0.000000
12	7	0	1.076545	-2.123097	0.000000
13	8	0	0.110561	-2.862428	0.000000
14	8	0	2.250712	-2.463105	0.000000
15	6	0	3.201010	0.169883	0.000000
16	1	0	3.536756	-0.362463	0.884639
17	1	0	3.536756	-0.362463	-0.884639
18	1	0	3.565971	1.190874	0.000000

DNPPy

Zero-point correction= 0.093215 (Hartree/Particle)
 Thermal correction to Energy= 0.103715
 Thermal correction to Enthalpy= 0.104660
 Thermal correction to Gibbs Free Energy= 0.056292

Electronic energy:

B3LYP = -783.1141741
 W1-F12 = -783.18238765
 W2-F12 = -783.18566511

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	7	0	1.333642	1.042678	0.000000
2	6	0	0.019098	0.694093	0.000000
3	6	0	-0.019098	-0.694093	0.000000
4	7	0	-1.333642	-1.042678	0.000000
5	7	0	-2.131747	0.031249	0.000000
6	7	0	2.131747	-0.031249	0.000000
7	6	0	1.333642	-1.088155	0.000000
8	6	0	-1.333642	1.088155	0.000000

9	7	0	-1.821220	2.442384	0.000000
10	8	0	-3.015393	2.648655	0.000000
11	8	0	-0.935836	3.297473	0.000000
12	7	0	1.821220	-2.442384	0.000000
13	8	0	3.015393	-2.648655	0.000000
14	8	0	0.935836	-3.297473	0.000000
15	1	0	1.733470	1.966020	0.000000
16	1	0	-1.733470	-1.966020	0.000000

ANPZ

Zero-point correction= 0.115263 (Hartree/Particle)
 Thermal correction to Energy= 0.126874
 Thermal correction to Enthalpy= 0.127819
 Thermal correction to Gibbs Free Energy= 0.077171

Electronic energy:

B3LYP = -784.4244385
 W1-F12 = -784.47765122
 W2-F12 = -784.48123233

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	1.158121	1.241696
2	6	0	0.000000	1.131169	-0.193852
3	6	0	0.000000	-1.131169	-0.193852
4	6	0	0.000000	-1.158121	1.241696
5	7	0	0.000000	2.342805	-1.018148
6	8	0	0.000000	2.229561	-2.221903
7	8	0	0.000000	3.426414	-0.418516
8	7	0	0.000000	-2.342805	-1.018148
9	8	0	0.000000	-3.426414	-0.418516
10	8	0	0.000000	-2.229561	-2.221903
11	7	0	0.000000	2.277250	1.968257

12	1	0	0.000000	3.165917	1.494189
13	1	0	0.000000	2.193144	2.969500
14	7	0	0.000000	0.000000	-0.838578
15	7	0	0.000000	0.000000	1.901960
16	7	0	0.000000	-2.277250	1.968257
17	1	0	0.000000	-2.193144	2.969500
18	1	0	0.000000	-3.165917	1.494189

LLM-105

Zero-point correction= 0.119203 (Hartree/Particle)
 Thermal correction to Energy= 0.131586
 Thermal correction to Enthalpy= 0.132531
 Thermal correction to Gibbs Free Energy= 0.080511

Electronic energy:

B3LYP = -859.6204907

W1-F12 = -859.69129033

W2-F12 = -859.69536880

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	1.200019	0.978531
2	6	0	0.000000	1.130469	-0.436446
3	6	0	0.000000	-1.130469	-0.436446
4	6	0	0.000000	-1.200019	0.978531
5	7	0	0.000000	2.352898	-1.248689
6	8	0	0.000000	2.247182	-2.451844
7	8	0	0.000000	3.425496	-0.633876
8	7	0	0.000000	-2.352898	-1.248689
9	8	0	0.000000	-3.425496	-0.633876
10	8	0	0.000000	-2.247182	-2.451844
11	7	0	0.000000	2.265408	1.765140
12	1	0	0.000000	3.178600	1.343765

13	1	0	0.000000	2.078026	2.758234
14	7	0	0.000000	-2.265408	1.765140
15	1	0	0.000000	-2.078026	2.758234
16	1	0	0.000000	-3.178600	1.343765
17	7	0	0.000000	0.000000	1.644070
18	7	0	0.000000	0.000000	-1.086721
19	8	0	0.000000	0.000000	2.941340

ANTA

Zero-point correction= 0.077885 (Hartree/Particle)
 Thermal correction to Energy= 0.085428
 Thermal correction to Enthalpy= 0.086372
 Thermal correction to Gibbs Free Energy= 0.044796

Electronic energy:

B3LYP = -502.3319039

W1-F12 = -502.37112720

W2-F12 = -502.37310281

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.390155	0.106711	-0.001599
2	7	0	1.823438	-0.187885	0.003707
3	7	0	-0.038864	1.343687	-0.015752
4	7	0	-1.381818	1.149263	-0.018095
5	6	0	-1.667342	-0.178807	-0.000234
6	7	0	-0.543973	-0.868844	0.013083
7	8	0	2.584413	0.762019	0.036493
8	8	0	2.138403	-1.363015	-0.025595
9	1	0	-1.999275	1.941385	-0.071614
10	7	0	-2.929554	-0.697365	-0.057101
11	1	0	-2.962557	-1.690076	0.111642
12	1	0	-3.662189	-0.182760	0.402886

NTO

Zero-point correction= 0.065825 (Hartree/Particle)

Thermal correction to Energy= 0.072892

Thermal correction to Enthalpy= 0.073836

Thermal correction to Gibbs Free Energy= 0.033702

Electronic energy:

B3LYP = -522.2186539

W1-F12 = -522.27081546

W2-F12 = -522.27333393

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	0.405289	0.000000
2	7	0	0.810218	1.602539	0.000000
3	7	0	-1.288370	0.395443	0.000000
4	7	0	-1.596082	-0.925137	0.000000
5	6	0	-0.489431	-1.772699	0.000000
6	7	0	0.553688	-0.839203	0.000000
7	8	0	-0.446266	-2.974953	0.000000
8	8	0	2.020741	1.402458	0.000000
9	8	0	0.250682	2.677685	0.000000
10	1	0	1.536339	-1.052290	0.000000
11	1	0	-2.557177	-1.220267	0.000000

5-ATZ

Zero-point correction= 0.063068 (Hartree/Particle)

Thermal correction to Energy= 0.067975

Thermal correction to Enthalpy= 0.068919

Thermal correction to Gibbs Free Energy= 0.035120

Electronic energy:

B3LYP = -313.7557239

W1-F12 = -313.76663149

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.621357	-0.043508	-0.003522
2	7	0	-1.485948	0.574839	0.004407
3	7	0	-0.120171	-1.130869	0.009688
4	7	0	-1.411133	-0.699429	0.008601
5	7	0	-0.197292	1.024435	-0.005354
6	1	0	0.000338	2.008284	-0.059131
7	7	0	1.987750	0.003468	-0.077526
8	1	0	2.443019	0.743010	0.432364
9	1	0	2.416062	-0.897350	0.069181

DAT

Zero-point correction= 0.079926 (Hartree/Particle)

Thermal correction to Energy= 0.086085

Thermal correction to Enthalpy= 0.087029

Thermal correction to Gibbs Free Energy= 0.050269

Electronic energy:

B3LYP = -369.1086825

W1-F12 = -369.12026832

W2-F12 = -369.12106271

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.224681	-0.715635	0.005140
2	7	0	-0.203895	0.631730	0.010662
3	7	0	1.103356	1.040522	0.003121
4	7	0	1.798890	-0.029376	-0.006850
5	7	0	1.019212	-1.149906	-0.005384

6	7	0	-1.363340	-1.453563	0.065972
7	1	0	-2.208461	-0.989698	-0.224160
8	1	0	-1.267533	-2.415170	-0.212430
9	7	0	-1.302055	1.471692	-0.014877
10	1	0	-1.327152	2.017143	0.839472
11	1	0	-1.213949	2.103845	-0.802222

BTZ

Zero-point correction= 0.073894 (Hartree/Particle)

Thermal correction to Energy= 0.080780

Thermal correction to Enthalpy= 0.081725

Thermal correction to Gibbs Free Energy= 0.042034

Electronic energy:

B3LYP = -515.5413664

W1-F12 = -515.56636622

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.009975	0.719456	0.000000
2	7	0	0.672316	2.787262	0.000000
3	7	0	-1.077572	1.488415	0.000000
4	7	0	-0.614951	2.763199	0.000000
5	7	0	1.077572	1.507326	0.000000
6	6	0	0.009975	-0.719456	0.000000
7	7	0	1.077572	-1.488415	0.000000
8	7	0	-1.077572	-1.507326	0.000000
9	7	0	0.614951	-2.763199	0.000000
10	7	0	-0.672316	-2.787262	0.000000
11	1	0	-2.053756	-1.255529	0.000000
12	1	0	2.053756	1.255529	0.000000

BTO

Zero-point correction= 0.080937 (Hartree/Particle)

Thermal correction to Energy= 0.089917

Thermal correction to Enthalpy= 0.090861

Thermal correction to Gibbs Free Energy= 0.046544

Electronic energy:

B3LYP = -665.9311065

W1-F12 = -665.98965514

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.012039	0.711494	0.000000
2	7	0	0.703994	2.766652	0.000000
3	7	0	-1.073829	1.498533	0.000000
4	7	0	-0.597014	2.760046	0.000000
5	7	0	1.073829	1.492424	0.000000
6	6	0	0.012039	-0.711494	0.000000
7	7	0	1.073829	-1.498533	0.000000
8	7	0	-1.073829	-1.492424	0.000000
9	7	0	0.597014	-2.760046	0.000000
10	7	0	-0.703994	-2.766652	0.000000
11	8	0	2.381718	1.131362	0.000000
12	8	0	-2.381718	-1.131362	0.000000
13	1	0	2.371471	0.148506	0.000000
14	1	0	-2.371471	-0.148506	0.000000

Ph-Azs

Zero-point correction= 0.118288 (Hartree/Particle)

Thermal correction to Energy= 0.127054

Thermal correction to Enthalpy= 0.127998

Thermal correction to Gibbs Free Energy= 0.082104

Electronic energy:

B3LYP = -584.6296639

W1-F12 = -584.63103737

W2-F12 = -584.63319377

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.051753	2.522444	0.000000
2	7	0	0.978235	1.513353	0.000000
3	7	0	-1.223854	1.170458	0.000000
4	7	0	0.000000	0.677564	0.000000
5	8	0	0.394192	2.715039	0.000000
6	8	0	-1.803212	3.443566	0.000000
7	6	0	0.242871	-0.734506	0.000000
8	6	0	-0.849059	-1.591001	0.000000
9	6	0	1.550893	-1.205715	0.000000
10	6	0	-0.618241	-2.959814	0.000000
11	1	0	-1.847244	-1.183009	0.000000
12	6	0	1.760520	-2.575463	0.000000
13	1	0	2.378853	-0.514192	0.000000
14	6	0	0.680445	-3.453293	0.000000
15	1	0	-1.458715	-3.639242	0.000000
16	1	0	2.771538	-2.957394	0.000000
17	1	0	0.853007	-4.520531	0.000000

Tol-Azs

Zero-point correction= 0.145045 (Hartree/Particle)

Thermal correction to Energy= 0.155734

Thermal correction to Enthalpy= 0.156678

Thermal correction to Gibbs Free Energy= 0.105981

Electronic energy:

B3LYP = -623.9300195

W1-F12 = -623.96032703

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	3.159197	-0.416976	0.003214
2	7	0	1.707800	1.322982	0.000542
3	7	0	1.883798	-0.898563	0.000708
4	7	0	1.119679	0.178342	-0.000245
5	8	0	3.013430	1.031933	0.002233
6	8	0	4.229482	-0.936545	0.003507
7	6	0	-0.310977	0.088027	-0.001691
8	6	0	-0.899854	-1.168051	-0.005192
9	6	0	-1.078830	1.248684	-0.003325
10	6	0	-2.285156	-1.256357	-0.008340
11	1	0	-0.279535	-2.050728	-0.007560
12	6	0	-2.457797	1.133703	-0.006452
13	1	0	-0.603969	2.217902	-0.004406
14	6	0	-3.087359	-0.115806	-0.006706
15	1	0	-2.747192	-2.234808	-0.014125
16	1	0	-3.058476	2.034107	-0.010830
17	6	0	-4.587578	-0.217918	0.013204
18	1	0	-5.042711	0.496898	-0.673098
19	1	0	-4.922895	-1.216686	-0.262028
20	1	0	-4.977325	-0.000943	1.010809

ANF

Zero-point correction= 0.064986 (Hartree/Particle)

Thermal correction to Energy= 0.072228

Thermal correction to Enthalpy= 0.073172

Thermal correction to Gibbs Free Energy= 0.032759

Electronic energy:

B3LYP = -522.1313267

W1-F12 = -522.18429138

W2-F12 = -522.18662923

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	7	0	-0.399843	-1.512225	0.013474
2	6	0	-0.948597	0.661566	-0.012223
3	6	0	0.101506	-0.313363	-0.005824
4	8	0	-1.723255	-1.342687	0.020516
5	7	0	-2.083778	0.004094	0.006910
6	7	0	1.523739	-0.074013	0.001471
7	8	0	2.279237	-1.018164	-0.049137
8	8	0	1.851188	1.111308	0.063798
9	7	0	-0.838049	2.003100	-0.072881
10	1	0	-1.658431	2.559929	0.086531
11	1	0	0.069147	2.400499	0.097524

DAF

Zero-point correction= 0.079131 (Hartree/Particle)
 Thermal correction to Energy= 0.085284
 Thermal correction to Enthalpy= 0.086228
 Thermal correction to Gibbs Free Energy= 0.049486

Electronic energy:

B3LYP = -372.9392806
 W1-F12 = -372.95526847
 W2-F12 = -372.95645727

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.231554	-0.717313	0.000554
2	6	0	0.232208	0.717082	-0.000590
3	7	0	-0.997289	-1.135845	0.000758
4	8	0	-1.778941	0.000760	0.000469

5	7	0	-0.996240	1.136716	-0.001169
6	7	0	1.351195	-1.523594	0.069227
7	1	0	1.142611	-2.506602	-0.024934
8	1	0	2.094449	-1.239753	-0.552811
9	7	0	1.352587	1.522372	-0.069260
10	1	0	1.144782	2.505645	0.023949
11	1	0	2.095337	1.238481	0.553364

DNAF

Zero-point correction= 0.084625 (Hartree/Particle)
 Thermal correction to Energy= 0.095204
 Thermal correction to Enthalpy= 0.096148
 Thermal correction to Gibbs Free Energy= 0.047456

Electronic energy:

B3LYP = -782.0629457

W1-F12 = -782.15339585

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.042924	0.717420	0.962926
2	6	0	0.042924	-0.717420	0.962926
3	7	0	-0.064128	1.121556	2.201552
4	8	0	0.000000	0.000000	2.968867
5	7	0	0.064128	-1.121556	2.201552
6	7	0	0.000000	1.655704	-0.052384
7	1	0	0.212843	2.613962	0.184917
8	7	0	0.000000	-1.655704	-0.052384
9	1	0	-0.212843	-2.613962	0.184917
10	7	0	-0.875540	1.576959	-1.136368
11	8	0	-0.928723	2.570844	-1.826177
12	8	0	-1.460638	0.525258	-1.289766
13	7	0	0.875540	-1.576959	-1.136368

14	8	0	0.928723	-2.570844	-1.826177
15	8	0	1.460638	-0.525258	-1.289766

DAAzF

Zero-point correction= 0.113970 (Hartree/Particle)
 Thermal correction to Energy= 0.125634
 Thermal correction to Enthalpy= 0.126578
 Thermal correction to Gibbs Free Energy= 0.076102

Electronic energy:

B3LYP = -743.4078607
 W1-F12 = -743.44465733

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	7	0	-1.453287	2.456132	-0.050502
2	7	0	-0.546269	0.317300	-0.004094
3	7	0	0.546269	-0.317300	-0.004094
4	7	0	1.453287	-2.456132	-0.050502
5	6	0	0.783137	2.523712	0.024927
6	6	0	-0.783137	-2.523712	0.024927
7	6	0	-0.393885	1.687040	-0.006057
8	6	0	0.393885	-1.687040	-0.006057
9	7	0	2.066269	2.118624	0.111137
10	1	0	2.257568	1.140190	-0.015847
11	1	0	2.800384	2.794043	-0.003363
12	7	0	0.393885	3.772231	-0.009191
13	8	0	-1.002586	3.713212	-0.053183
14	8	0	1.002586	-3.713212	-0.053183
15	7	0	-2.066269	-2.118624	0.111137
16	1	0	-2.257568	-1.140190	-0.015847
17	1	0	-2.800384	-2.794043	-0.003363
18	7	0	-0.393885	-3.772231	-0.009191

DAAF

Zero-point correction= 0.119410 (Hartree/Particle)

Thermal correction to Energy= 0.131463

Thermal correction to Enthalpy= 0.132407

Thermal correction to Gibbs Free Energy= 0.079994

Electronic energy:

B3LYP = -818.6159881

W1-F12 = -818.67080523

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	7	0	-2.631574	-1.321278	-0.287926
2	7	0	-0.390923	-0.543388	-0.234240
3	7	0	0.292997	0.519008	-0.043320
4	7	0	2.311461	1.530076	-0.435399
5	6	0	-2.513626	0.864593	0.202123
6	6	0	2.648241	-0.568824	0.262173
7	6	0	-1.797241	-0.341655	-0.104398
8	6	0	1.670727	0.436427	-0.098857
9	7	0	-2.017953	2.083237	0.497434
10	1	0	-1.045445	2.245136	0.298640
11	1	0	-2.653258	2.860913	0.521431
12	7	0	-3.790501	0.566240	0.187693
13	8	0	-3.847118	-0.791694	-0.112455
14	8	0	3.618490	1.250037	-0.339651
15	7	0	2.444619	-1.794872	0.815449
16	1	0	1.620009	-2.274687	0.482976
17	1	0	3.273914	-2.360621	0.903062
18	7	0	3.835733	-0.048957	0.090940
19	8	0	0.026023	-1.676398	-0.536239

34-MNF

Zero-point correction= 0.079466 (Hartree/Particle)
Thermal correction to Energy= 0.088149
Thermal correction to Enthalpy= 0.089093
Thermal correction to Gibbs Free Energy= 0.044562

Electronic energy:

B3LYP = -581.2789124
W1-F12 = -581.33433187
W2-F12 = -581.33730925

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	0.562771	0.000000
2	6	0	-0.136516	-0.846341	0.000000
3	7	0	1.230911	0.972603	0.000000
4	8	0	1.998711	-0.121481	0.000000
5	7	0	1.105645	-1.315578	0.000000
6	7	0	-1.089539	1.533169	0.000000
7	8	0	-0.804904	2.709470	0.000000
8	8	0	-2.215221	1.054413	0.000000
9	6	0	-1.307727	-1.758349	0.000000
10	1	0	-1.928564	-1.583642	0.877222
11	1	0	-1.928564	-1.583642	-0.877222
12	1	0	-0.959200	-2.788424	0.000000
13	8	0	1.615496	-2.407917	0.000000

34-MNF

Zero-point correction= 0.079613 (Hartree/Particle)
Thermal correction to Energy= 0.088085
Thermal correction to Enthalpy= 0.089029
Thermal correction to Gibbs Free Energy= 0.045297

Electronic energy:

B3LYP = -581.2766975

W1-F12 = -581.33107150

W2-F12 = -581.33409331

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	0.132988	0.000000
2	6	0	-0.691557	-1.111380	0.000000
3	7	0	1.317808	-0.108332	0.000000
4	8	0	1.430745	-1.544315	0.000000
5	7	0	0.168532	-2.085653	0.000000
6	8	0	2.319714	0.545654	0.000000
7	7	0	-0.530314	1.462992	0.000000
8	8	0	0.246700	2.398462	0.000000
9	8	0	-1.754188	1.538248	0.000000
10	6	0	-2.153714	-1.370655	0.000000
11	1	0	-2.621874	-0.924091	0.875657
12	1	0	-2.621874	-0.924091	-0.875657
13	1	0	-2.320557	-2.444971	0.000000

34-CNF

Zero-point correction= 0.050100 (Hartree/Particle)

Thermal correction to Energy= 0.058863

Thermal correction to Enthalpy= 0.059807

Thermal correction to Gibbs Free Energy= 0.014835

Electronic energy:

B3LYP = -634.1967524

W1-F12 = -634.26511485

W2-F12 = -634.26807240

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z

1	6	0	-0.115135	-0.712835	0.000000
2	6	0	0.000000	0.701210	0.000000
3	7	0	-1.340085	-1.120915	0.000000
4	8	0	-2.121183	-0.028084	0.000000
5	7	0	-1.256997	1.172601	0.000000
6	7	0	0.993645	-1.668069	0.000000
7	8	0	0.717205	-2.845409	0.000000
8	8	0	2.103155	-1.163675	0.000000
9	6	0	1.102658	1.573121	0.000000
10	7	0	1.978016	2.320622	0.000000
11	8	0	-1.767576	2.249837	0.000000

BTF

Zero-point correction= 0.086722 (Hartree/Particle)
 Thermal correction to Energy= 0.099321
 Thermal correction to Enthalpy= 0.100265
 Thermal correction to Gibbs Free Energy= 0.047555

Electronic energy:

B3LYP = -1008.4731726

W1-F12 = -1008.56090455

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	1.416690	0.000000
2	6	0	-1.249528	0.726628	0.000000
3	6	0	-1.227001	-0.708444	0.000000
4	6	0	-0.004595	-1.445476	0.000000
5	6	0	1.226956	-0.708607	0.000000
6	6	0	1.253981	0.718701	0.000000
7	7	0	-0.213745	-2.731929	0.000000
8	7	0	2.481887	-1.133838	0.000000
9	7	0	2.472840	1.180517	0.000000

10	7	0	-0.258867	2.716079	0.000000
11	7	0	-2.259028	1.550994	0.000000
12	7	0	-2.222961	-1.582222	0.000000
13	8	0	-1.743141	2.804212	0.000000
14	8	0	-1.557037	-2.911656	0.000000
15	8	0	3.299796	0.107800	0.000000
16	8	0	3.036786	-2.196750	0.000000
17	8	0	-3.420890	-1.530768	0.000000
18	8	0	0.384515	3.727892	0.000000

NNFF

Zero-point correction= 0.079347 (Hartree/Particle)

Thermal correction to Energy= 0.092632

Thermal correction to Enthalpy= 0.093576

Thermal correction to Gibbs Free Energy= 0.035993

Electronic energy:

B3LYP = -1007.4629609

W1-F12 = -1007.58855748

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	7	0	0.556437	1.629179	-0.714607
2	7	0	-0.550431	-1.571494	-1.309883
3	6	0	1.907934	0.090089	0.065421
4	6	0	-1.880608	-0.406847	-0.003161
5	6	0	0.591850	0.339058	-0.384216
6	6	0	-0.571158	-0.512048	-0.550324
7	7	0	2.662626	1.140011	0.032383
8	8	0	1.907583	2.147531	-0.439614
9	8	0	-1.801102	-2.114352	-1.222169
10	7	0	-2.614083	-1.389303	-0.422023
11	7	0	-2.385019	0.586079	0.938820

12	8	0	-3.574902	0.612168	1.150392
13	8	0	-1.529883	1.300702	1.438295
14	7	0	2.413405	-1.170651	0.608555
15	8	0	3.611098	-1.334807	0.623883
16	8	0	1.554338	-1.938509	1.010008
17	8	0	-0.275714	2.373736	-1.148174

DNBF

Zero-point correction= 0.083177 (Hartree/Particle)

Thermal correction to Energy= 0.097451

Thermal correction to Enthalpy= 0.098395

Thermal correction to Gibbs Free Energy= 0.040222

Electronic energy:

B3LYP = -1082.6645102

W1-F12 = -1082.80078235

1	7	0	1.240654	1.206632	0.878017
2	7	0	-1.240654	-1.206632	0.878017
3	6	0	-0.596630	1.824535	-0.149138
4	6	0	0.596630	-1.824535	-0.149138
5	6	0	0.132717	0.707531	0.323126
6	6	0	-0.132717	-0.707531	0.323126
7	7	0	-0.001753	2.951052	0.073497
8	8	0	1.151824	2.664487	0.702286
9	8	0	-1.151824	-2.664487	0.702286
10	7	0	0.001753	-2.951052	0.073497
11	7	0	1.861338	-1.783017	-0.881500
12	8	0	2.444310	-2.826253	-1.058682
13	8	0	2.199885	-0.672452	-1.260194
14	7	0	-1.861338	1.783017	-0.881500
15	8	0	-2.444310	2.826253	-1.058682
16	8	0	-2.199885	0.672452	-1.260194
17	8	0	-2.199885	-0.744020	1.424837

18 8 0 2.199885 0.744020 1.424837

References

- [1] H.-H. Emons, R. Naumann, K. Jahn, H.-J. Flammersheim, Thermal properties of acetamide in the temperature range from 298 K to 400 K, *Thermochim. Acta.* 104 (1986) 127–137. [https://doi.org/10.1016/0040-6031\(86\)85191-7](https://doi.org/10.1016/0040-6031(86)85191-7).
- [2] M. Davies, A.H. Jones, G.H. Thomas, The lattice energies of the straight-chain primary amides, *Trans. Faraday Soc.* 55 (1959) 1100. <https://doi.org/10.1039/tf9595501100>.
- [3] H.G.M. De Wit, C.G. De Kruif, J.C. Van Miltenburg, Thermodynamic properties of molecular organic crystals containing nitrogen, oxygen, and sulfur II. Molar heat capacities of eight compounds by adiabatic calorimetry, *J. Chem. Thermodyn.* 15 (1983) 891–902. [https://doi.org/10.1016/0021-9614\(83\)90095-2](https://doi.org/10.1016/0021-9614(83)90095-2).
- [4] P.J. Linstrom, W.G. Mallard, eds., NIST Chemistry WebBook, NIST Standard Reference Database Number 69, National Institute of Standards and Technology, Gaithersburg MD, n.d. <http://webbook.nist.gov> (accessed September 14, 2016).
- [5] M.J. O’Neil, ed., The Merck index: an encyclopedia of chemicals, drugs, and biologicals, 14th ed, Merck, Whitehouse Station, N.J, 2006.
- [6] B.L. Sharma, R. Kant, R. Sharma, S. Tandon, Deviations of binary organic eutectic melt systems, *Mater. Chem. Phys.* 82 (2003) 216–224. [https://doi.org/10.1016/S0254-0584\(03\)00199-8](https://doi.org/10.1016/S0254-0584(03)00199-8).
- [7] P. Pacor, Applicability of the du pont 900 DTA apparatus in quantitative differential thermal analysis, *Anal. Chim. Acta.* 37 (1967) 200–208. [https://doi.org/10.1016/S0003-2670\(01\)80660-7](https://doi.org/10.1016/S0003-2670(01)80660-7).
- [8] G.T. Furukawa, R.E. McCoskey, G.J. King, Calorimetric properties of benzoic acid from 0 to 410K, *J. Res. Natl. Bur. Stand.* 47 (1951) 256–261.
- [9] K. Khimeche, A. Dahmani, Solid–Liquid Equilibria of Naphthalene + Alkanediamine Mixtures, *J. Chem. Eng. Data.* 51 (2006) 382–385. <https://doi.org/10.1021/je0502851>.
- [10] D.J. David, Determination of Specific Heat and Heat of Fusion by Differential Thermal Analysis. Study of Theory and Operating Parameters., *Anal. Chem.* 36 (1964) 2162–2166. <https://doi.org/10.1021/ac60217a038>.
- [11] B.J. Mair, Muthu. Shamaengar, N.C. Krouskop, F.D. Rossini, Isolation of Adamantane from Petroleum, *Anal. Chem.* 31 (1959) 2082–2083. <https://doi.org/10.1021/ac60156a063>.
- [12] G.J. Kabo, A.V. Blokhin, M.B. Charapennikau, A.G. Kabo, V.M. Sevruk, Thermodynamic properties of adamantine and the energy states of molecules in plastic crystals for some cage hydrocarbons, *Thermochim. Acta.* 345 (2000) 125–133. [https://doi.org/10.1016/S0040-6031\(99\)00393-7](https://doi.org/10.1016/S0040-6031(99)00393-7).
- [13] J.T.S. Andrews, R.E. Carpenter, T.M. Martinko, R.C. Fort, T.A. Flood, M.G. Adlington, Transition and Fusion Thermodynamics of Heteroadamantanes, *Mol. Cryst. Liq. Cryst.* 41 (1978) 257–261. <https://doi.org/10.1080/00268947808070312>.
- [14] J. Chickos, D. Hesse, S. Hosseini, G. Nichols, P. Webb, Sublimation enthalpies at 298.15K using correlation gas chromatography and differential scanning calorimetry measurements, *Thermochim. Acta.* 313 (1998) 101–110. [https://doi.org/10.1016/S0040-6031\(97\)00432-2](https://doi.org/10.1016/S0040-6031(97)00432-2).
- [15] Z. Lisicki, M.E. Jamróz, (Solid + liquid) equilibria in (polynuclear aromatic+ tertiary amide) systems, *J. Chem. Thermodyn.* 32 (2000) 1335–1353. <https://doi.org/10.1006/jcht.2000.0685>.

- [16] A. Rojas, E. Orozco, Measurement of the enthalpies of vaporization and sublimation of solids aromatic hydrocarbons by differential scanning calorimetry, *Thermochim. Acta.* 405 (2003) 93–107. [https://doi.org/10.1016/S0040-6031\(03\)00139-4](https://doi.org/10.1016/S0040-6031(03)00139-4).
- [17] J.R. Donnelly, L.A. Drewes, R.L. Johnson, W.D. Munslow, K.K. Knapp, G.W. Sovocool, Purity and heat of fusion data for environmental standards as determined by differential scanning calorimetry, *Thermochim. Acta.* 167 (1990) 155–187. [https://doi.org/10.1016/0040-6031\(90\)80476-F](https://doi.org/10.1016/0040-6031(90)80476-F).
- [18] A. Cingolani, G. Berchiesi, Thermodynamic properties of organic compounds: Note I. A DSC study of phase transitions in aliphatic dicarboxylic acids, *J. Therm. Anal.* 6 (1974) 87–90. <https://doi.org/10.1007/BF01911490>.
- [19] W.V. Steele, R.D. Chirico, S.E. Knipmeyer, A. Nguyen, N.K. Smith, Thermodynamic Properties and Ideal-Gas Enthalpies of Formation for Butyl Vinyl Ether, 1,2-Dimethoxyethane, Methyl Glycolate, Bicyclo[2.2.1]hept-2-ene, 5-Vinylbicyclo[2.2.1]hept-2-ene, *trans* -Azobenzene, Butyl Acrylate, Di-*tert*-butyl Ether, and Hexane-1,6-diol, *J. Chem. Eng. Data.* 41 (1996) 1285–1302. <https://doi.org/10.1021/je960117w>.
- [20] F.-W. Schulze, H.-J. Petrick, H.K. Cammenga, H. Klinge, Thermodynamic Properties of the Structural Analogues Benzo[c]cinnoline, Trans-azobenzene, and Cis-azobenzene, *Z. Für Phys. Chem.* 107 (1977) 1–19. <https://doi.org/10.1524/zpch.1977.107.1.001>.
- [21] E. Suzuki, K. Shimomura, K. Sekiguchi, Thermochemical study of theophylline and its hydrate., *Chem. Pharm. Bull. (Tokyo)*. 37 (1989) 493–497. <https://doi.org/10.1248/cpb.37.493>.
- [22] J.G. Fokkens, J.G.M. van Amelsfoort, C.J. de Blaey, C.G. de Kruif, J. Wilting, A thermodynamic study of the solubility of theophylline and its hydrate, *Int. J. Pharm.* 14 (1983) 79–93. [https://doi.org/10.1016/0378-5173\(83\)90116-3](https://doi.org/10.1016/0378-5173(83)90116-3).
- [23] T. Sánchez-Bulás, O. Cruz-Vásquez, J. Hernández-Obregón, A. Rojas, Enthalpies of fusion, vaporisation and sublimation of crown ethers determined by thermogravimetry and differential scanning calorimetry, *Thermochim. Acta.* 650 (2017) 123–133. <https://doi.org/10.1016/j.tca.2016.03.013>.
- [24] G. Nichols, J. Orf, S.M. Reiter, J. Chickos, G.W. Gokel, The vaporization enthalpies of some crown and polyethers by correlation gas chromatography, *Thermochim. Acta.* 346 (2000) 15–28. [https://doi.org/10.1016/S0040-6031\(99\)00405-0](https://doi.org/10.1016/S0040-6031(99)00405-0).
- [25] A. Bianchi, J. Giusti, P. Paoletti, Analysis by differential scanning calorimetry of the KNCS/dibenzo-18-crown-6 system: Phase diagram and enthalpy changes, *Thermochim. Acta.* 90 (1985) 109–114. [https://doi.org/10.1016/0040-6031\(85\)87088-X](https://doi.org/10.1016/0040-6031(85)87088-X).
- [26] A. Finch, P.J. Gardner, A.J. Head, X. Wu, The enthalpy of formation of 4-nitrophenyl azide, *Thermochim. Acta.* 298 (1997) 191–194. [https://doi.org/10.1016/S0040-6031\(97\)00192-5](https://doi.org/10.1016/S0040-6031(97)00192-5).
- [27] A.E. Fogelzang, V.Yu. Egorshev, V.P. Sinditsky, M.D. Dutov, Combustion of nitro derivatives of azidobenzenes and benzofuroxans, *Combust. Flame.* 87 (1991) 123–135. [https://doi.org/10.1016/0010-2180\(91\)90162-5](https://doi.org/10.1016/0010-2180(91)90162-5).
- [28] D.-R. Hwang, M. Tamura, T. Yoshida, N. Tanaka, F. Hosoya, Estimation of Δ_f of nitro derivatives of benzene and toluene using AM1 and DSC, *J. Energ. Mater.* 8 (1990) 85–98. <https://doi.org/10.1080/07370659008017247>.
- [29] C. Lenchitz, R.W. Velicky, G. Silvestro, L.P. Schlosberg, Thermodynamic properties of several nitrotoluenes, *J. Chem. Thermodyn.* 3 (1971) 689–692. [https://doi.org/10.1016/S0021-9614\(71\)80090-3](https://doi.org/10.1016/S0021-9614(71)80090-3).
- [30] W.E. Acree, Thermodynamic properties of organic compounds: enthalpy of fusion and melting point temperature compilation, *Thermochim. Acta.* 189 (1991) 37–56. [https://doi.org/10.1016/0040-6031\(91\)87098-H](https://doi.org/10.1016/0040-6031(91)87098-H).

- [31] H. Takahashi, R. Tamura, Low temperature phase transition induced biaxial negative thermal expansion of 2,4-dinitroanisole, *CrystEngComm.* 17 (2015) 8888–8896. <https://doi.org/10.1039/C5CE00029G>.
- [32] F.M. Beringer, A. Brierley, M. Drexler, E.M. Gindler, C.C. Lumpkin, Diaryliodonium Salts. II. The Phenylation of Organic and Inorganic Bases ^{1,2}, *J. Am. Chem. Soc.* 75 (1953) 2708–2712. <https://doi.org/10.1021/ja01107a047>.
- [33] X. Xing, F. Zhao, S. Ma, K. Xu, L. Xiao, H. Gao, T. An, R. Hu, Specific Heat Capacity, Thermal Behavior, and Thermal Hazard of 2,4-Dinitroanisole, *Propellants Explos. Pyrotech.* 37 (2012) 179–182. <https://doi.org/10.1002/prep.201000077>.
- [34] N.V. Muravyev, K.A. Monogarov, A.A. Bragin, I.V. Fomenkov, A.N. Pivkina, HP-DSC study of energetic materials. Part I. Overview of pressure influence on thermal behavior, *Thermochim. Acta.* 631 (2016) 1–7. <https://doi.org/10.1016/j.tca.2016.03.018>.
- [35] E.G. Kayser, Analysis methods for explosive materials—I. polynitro compounds, *J. Energ. Mater.* 1 (1983) 251–273. <https://doi.org/10.1080/07370658308010621>.
- [36] G. Edwards, The vapour pressure of 2 : 4 : 6-trinitrotoluene, *Trans. Faraday Soc.* 46 (1950) 423. <https://doi.org/10.1039/tf9504600423>.
- [37] W.E. Garner, C.L. Abernethy, Heats of Combustion and Formation of Nitro-Compounds. Part I. Benzene, Toluene, Phenol and Methylaniline Series, *Proc. R. Soc. Math. Phys. Eng. Sci.* 99 (1921) 213–235. <https://doi.org/10.1098/rspa.1921.0036>.
- [38] P.G. Hall, Thermal decomposition and phase transitions in solid nitramines, *Trans. Faraday Soc.* 67 (1971) 556. <https://doi.org/10.1039/tf9716700556>.
- [39] G. Krien, H.H. Licht, J. Zierath, Thermochemische untersuchungen an nitraminen, *Thermochim. Acta.* 6 (1973) 465–472. [https://doi.org/10.1016/0040-6031\(73\)85078-6](https://doi.org/10.1016/0040-6031(73)85078-6).
- [40] V.G. Kiselev, N.V. Muravyev, K.A. Monogarov, P.S. Gribanov, A.F. Asachenko, I.V. Fomenkov, C.F. Goldsmith, A.N. Pivkina, N.P. Gritsan, Toward reliable characterization of energetic materials: interplay of theory and thermal analysis in the study of the thermal stability of tetrinitroacetimidic acid (TNAA), *Phys. Chem. Chem. Phys.* 20 (2018) 29285–29298. <https://doi.org/10.1039/C8CP05619F>.
- [41] T.T. Vo, D.A. Parrish, J.M. Shreeve, Tetrinitroacetimidic Acid: A High Oxygen Oxidizer and Potential Replacement for Ammonium Perchlorate, *J. Am. Chem. Soc.* 136 (2014) 11934–11937. <https://doi.org/10.1021/ja5074036>.
- [42] L.V. Kustova, E.P. Kirpichev, Yu.I. Rubtsov, V.V. Avdonin, A.G. Koperin, L.T. Eremenko, Standard Enthalpies of Formation of Certain Nitro Compounds, *Bull. Acad. Sci. USSR Div. Chem. Sci.* (1981) 1830–1836.
- [43] B.M. Dobratz, LLNL Explosives Handbook, University of California, Livermore, CA, USA, 1985.
- [44] V.A. Tartakovsky, A.S. Ermakov, N.V. Sigai, O.N. Varfolomeeva, Synthesis of N,N'-dialkylmethylenebis(nitramines) from N-alkylsulfamates, *Russ. Chem. Bull.* 49 (2000) 1079–1081. <https://doi.org/10.1007/BF02494897>.
- [45] D. Spitzer, S. Braun, M.R. Schäfer, F. Ciszek, Comparative Crystallization Study of Several Linear Dinitramines in Nitrocellulose-Based Gels, *Propellants Explos. Pyrotech.* 28 (2003) 58–64. <https://doi.org/10.1002/prep.200390009>.
- [46] S. Zeman, Some predictions in the field of the physical thermal stability of nitramines, *Thermochim. Acta.* 302 (1997) 11–16. [https://doi.org/10.1016/S0040-6031\(96\)03101-2](https://doi.org/10.1016/S0040-6031(96)03101-2).
- [47] A. Stolovy, A.I. Namenson, J.B. Aviles, E.C. Jones, J.M. Kidd, Thermal initiation of high explosives by electron beam heating, *J. Energ. Mater.* 5 (1987) 181–238. <https://doi.org/10.1080/07370658708012352>.

- [48] S.K. Bhattacharia, B.L. Weeks, C.-C. Chen, Melting Behavior and Heat of Fusion of Compounds that Undergo Simultaneous Melting and Decomposition: An investigation with HMX, *J. Chem. Eng. Data.* 62 (2017) 967–972. <https://doi.org/10.1021/acs.jced.6b00769>.
- [49] R.A. Henry, W.P. Norris, 1,9-Diazido-2,4,6,8-tetranitro-2,4,6,8-tetrazanonane, US4362583A, 1982.
- [50] Yu.M. Burov, G.M. Nazin, G.B. Manelis, Retardation of monomolecular reactions in the solid phase, *Russ. Chem. Bull.* 48 (1999) 1250–1254. <https://doi.org/10.1007/BF02495284>.
- [51] J.C. Oxley, J.L. Smith, J.E. Brady, A.C. Brown, Characterization and Analysis of Tetranitrate Esters, *Propellants Explos. Pyrotech.* 37 (2012) 24–39. <https://doi.org/10.1002/prep.201100059>.
- [52] D.E. Chavez, M.A. Hiskey, D.L. Naud, D. Parrish, Synthesis of an Energetic Nitrate Ester, *Angew. Chem. Int. Ed.* 47 (2008) 8307–8309. <https://doi.org/10.1002/anie.200803648>.
- [53] K.C. Agrawal, K.B. Bears, R.K. Sehgal, J.N. Brown, P.E. Rist, W.D. Rupp, Potential radiosensitizing agents. Dinitroimidazoles, *J. Med. Chem.* 22 (1979) 583–586. <https://doi.org/10.1021/jm00191a025>.
- [54] L. Minier, R. Behrens, S. Bulusu, Solid-Phase Thermal Decomposition of 2,4-Dinitroimidazole (2,4-DNI), *MRS Proc.* 418 (1995) 111. <https://doi.org/10.1557/PROC-418-111>.
- [55] S.S. Novikov, L.I. Khmel'nitskii, O.V. Lebedev, V.V. Sevast'yanova, L.V. Epishina, Nitration of imidazoles with various nitrating agents, *Chem. Heterocycl. Compd.* 6 (1970) 465–469. <https://doi.org/10.1007/BF00478393>.
- [56] V.V. Dubikhin, G.M. Nazin, V.G. Prokudin, Z.G. Aliev, I.A. Vatsadze, S.A. Shevelev, I.L. Dalinger, Kinetics and mechanism of thermal decomposition of nitropyrazoles, *Russ. Chem. Bull.* 64 (2015) 127–131. <https://doi.org/10.1007/s11172-015-0830-9>.
- [57] J. Ritums, C. Oscarson, M. Liljedahl, P. Goede, K. Dudek, U. Heiche, Evaluation of 3(5),4-dinitropyrazole (DNP) as a New Melt Cast Matrix, in: *Proc. 45th Int. Conf. Fraunhofer ICT, Karlsruhe, 2014:* pp. 1–12.
- [58] M.V. Gorn, K.A. Monogarov, I.L. Dalinger, I.N. Melnikov, V.G. Kiselev, N.V. Muravyev, Pressure DSC for energetic materials. Part 2. Switching between evaporation and thermal decomposition of 3,5-dinitropyrazole, *Thermochim. Acta.* 690 (2020) 178697. <https://doi.org/10.1016/j.tca.2020.178697>.
- [59] S. Ek, N.V. Latypov, Four Syntheses of 4-Amino-3,5-dinitropyrazole: Four Syntheses of 4-Amino-3,5-dinitropyrazole, *J. Heterocycl. Chem.* 51 (2014) 1621–1627. <https://doi.org/10.1002/jhet.1725>.
- [60] X. Zhao, C. Qi, L. Zhang, Y. Wang, S. Li, F. Zhao, S. Pang, Amination of Nitroazoles — A Comparative Study of Structural and Energetic Properties, *Molecules.* 19 (2014) 896–910. <https://doi.org/10.3390/molecules19010896>.
- [61] P. Yin, J. Zhang, C. He, D.A. Parrish, J.M. Shreeve, Polynitro-substituted pyrazoles and triazoles as potential energetic materials and oxidizers, *J. Mater. Chem. A.* 2 (2014) 3200. <https://doi.org/10.1039/c3ta15057g>.
- [62] I.L. Dalinger, G.P. Popova, I.A. Vatsadze, T.K. Shkineva, S.A. Shevelev, Synthesis of 3,4,5-trinitropyrazole, *Russ. Chem. Bull.* 58 (2009) 2185–2185. <https://doi.org/10.1007/s11172-009-0301-2>.
- [63] G. Hervé, C. Roussel, H. Graindorge, Selective preparation of 3,4,5-trinitro-1H-pyrazole: a stable all-carbon-nitrated arene, *Angew. Chem. Int. Ed.* 49 (2010) 3177–3181. <https://doi.org/10.1002/anie.201000764>.

- [64] A. Bragin, A. Pivkina, N. Muravyev, K. Monogarov, O. Gryzlova, T. Shkineva, I. Dalinger, Thermal Decomposition of Nitropyrazoles, *Phys. Procedia*. 72 (2015) 358–361. <https://doi.org/10.1016/j.phpro.2015.09.110>.
- [65] P.J. Samuels, R. Damavarapu, H. Grau, K. Spangler, K. Caflin, E. Wrobel, Characterization of MTNP (1-methyl-3,4,5-trinitro-1,2-pyrazole), (2018). https://ndiastorage.blob.core.usgovcloudapi.net/ndia/2018/imem/20074_Samuels_Presentation.pdf.
- [66] Y. Tang, D. Kumar, J.M. Shreeve, Balancing Excellent Performance and High Thermal Stability in a Dinitropyrazole Fused 1,2,3,4-Tetrazine, *J. Am. Chem. Soc.* 139 (2017) 13684–13687. <https://doi.org/10.1021/jacs.7b08789>.
- [67] A.I. Lesnikovich, O.A. Ivashkevich, S.V. Levchik, A.I. Balabanovich, P.N. Gaponik, A.A. Kulak, Thermal decomposition of aminotetrazoles, *Thermochim. Acta*. 388 (2002) 233–251. [https://doi.org/10.1016/S0040-6031\(02\)00027-8](https://doi.org/10.1016/S0040-6031(02)00027-8).
- [68] S.V. Levchik, O.A. Ivashkevich, A.I. Balabanovich, A.I. Lesnikovich, P.N. Gaponik, L. Costa, Thermal decomposition of aminotetrazoles: Part 1. 5-Aminotetrazole, *Thermochim. Acta*. 207 (1992) 115–130. [https://doi.org/10.1016/0040-6031\(92\)80129-K](https://doi.org/10.1016/0040-6031(92)80129-K).
- [69] M. Tremblay, SPECTRES INFRAROUGES DE COMPOSÉS RICHES EN AZOTE, *Can. J. Chem.* 43 (1965) 1154–1157. <https://doi.org/10.1139/v65-153>.
- [70] I.L. Dalinger, O.V. Serushkina, N.V. Muravyev, D.B. Meerov, E.A. Miroshnichenko, T.S. Kon'kova, K.Yu. Suponitsky, M.V. Vener, A.B. Sheremetev, Azasydnone – novel “green” building block for designing high energetic compounds, *J. Mater. Chem. A*. 6 (2018) 18669–18676. <https://doi.org/10.1039/C8TA06895J>.
- [71] T. Klapötke, P. Schmid, J. Stierstorfer, Crystal Structures of Furazanes, *Crystals*. 5 (2015) 418–432. <https://doi.org/10.3390/crust5040418>.
- [72] A. Gunasekaran, J.H. Boyer, Dense energetic compounds of C, H, N, and O atoms. III. 5-[4-nitro-(1,2,5)oxadiazolyl]-5H-[1,2,3]triazolo[4,5-c][1,2,5]oxadiazole, *Heteroat. Chem.* 4 (1993) 521–524. <https://doi.org/10.1002/hc.520040519>.
- [73] Y. Tang, J. Zhang, L.A. Mitchell, D.A. Parrish, J.M. Shreeve, Taming of 3,4-Di(nitramino)furan, *J. Am. Chem. Soc.* 137 (2015) 15984–15987. <https://doi.org/10.1021/jacs.5b09831>.
- [74] G.D. Solodyuk, M.D. Boldyrev, B.V. Gidaspov, V.D. Nikolaev, ChemInform Abstract: OXIDATION OF 3,4-DIAMINOFURAZAN BY SOME PEROXIDE REAGENTS, *Chem. Informationsdienst*. 12 (1981). <https://doi.org/10.1002/chin.198136127>.
- [75] A.D. Nikolaeva, Yu.N. Matyushin, V.I. Pepekin, V.S. Smelov, V.V. Bulidorov, T.I. Bulidorova, A.Ya. Apin, Synthesis and study of detonation properties of 3-methyl-4-nitrofuroxan, *Bull. Acad. Sci. USSR Div. Chem. Sci.* 21 (1972) 927–928. <https://doi.org/10.1007/BF00854509>.
- [76] O.A. Rakitin, V.A. Ogurtsov, E.A. Khaibullina, T.I. Godovikova, L.I. Khmel'nitskii, Reaction of furoxanenitrolic acids with nitrogen tetroxide, *Chem. Heterocycl. Compd.* 29 (1993) 1099–1103. <https://doi.org/10.1007/BF00534399>.
- [77] A.S. Bailey, J.R. Case, 4:6-dinitrobenzofuroxan, nitrobenzodifuroxan and benzotrifuroxan: A new series of complex-forming reagents for aromatic hydrocarbons, *Tetrahedron*. 3 (1958) 113–131. [https://doi.org/10.1016/0040-4020\(58\)80003-4](https://doi.org/10.1016/0040-4020(58)80003-4).
- [78] D.L. Ornellas, The heat and products of detonation in a calorimeter of CNO, HNO, CHNF, CHNO, CHNOF, and CHNOSi explosives, *Combust. Flame*. 23 (1974) 37–46. [https://doi.org/10.1016/S0010-2180\(74\)80025-8](https://doi.org/10.1016/S0010-2180(74)80025-8).
- [79] Yu.N. Matyushin, V.I. Pepekin, V.P. Lebedev, V.V. Chironov, L.M. Kostikova, Y.O. Inozemtcev, T.S. Pivina, A.B. Sheremetev, Thermochemical properties and quantum-chemical parameters of benzotrifurazan and its N-oxides, in: Proc. 30th Int. Annu. Conf. Fraunhofer ICT, Pfingstal: Fraunhofer ICT, 1999: p. 77.1-77.9.

- [80] L.L. Fershtat, I.V. Ovchinnikov, M.A. Epishina, A.A. Romanova, D.B. Lempert, N.V. Muravyev, N.N. Makhova, Assembly of Nitrofurazan and Nitrofuroxan Frameworks for High-Performance Energetic Materials, *ChemPlusChem.* 82 (2017) 1315–1319. <https://doi.org/10.1002/cplu.201700340>.
- [81] A.A. Larin, A.V. Shaferov, M.A. Epishina, I.N. Melnikov, N.V. Muravyev, I.V. Ananyev, L.L. Fershtat, N.N. Makhova, Pushing the Energy-Sensitivity Balance with High-Performance Bifuroxans, *ACS Appl. Energy Mater.* 3 (2020) 7764–7771. <https://doi.org/10.1021/acsaem.0c01162>.
- [82] ASTM E928-03, Standard Test Method for Purity by Differential Scanning Calorimetry, ASTM International, West Conshohocken, PA, 2005. 10.1520/E0928-08R14.
- [83] S. Löbbecke, M.A. Bohn, A. Pfeil, H.H. Krause, Thermal behavior and stability of HNIW (CL-20), in: Proc. 29th Int. Annu. Conf. Fraunhofer ICT, Pfingstal: Fraunhofer ICT, 1998: p. 145.1-145.15.
- [84] R. Turcotte, M. Vachon, Q.S.M. Kwok, R. Wang, D.E.G. Jones, Thermal study of HNIW (CL-20), *Thermochim. Acta.* 433 (2005) 105–115. <https://doi.org/10.1016/j.tca.2005.02.021>.
- [85] A.K. Burnham, R.K. Weese, R. Wang, Q.S.M. Kwok, D.E.G. Jones, Solid-Solid Phase Transition Kinetics of FOX-7, in: Proc. 2005 NATAS Annu. Conf., 2005: pp. 1–8.
- [86] J.S. Chickos, S. Hosseini, D.G. Hesse, J.F. Liebman, Heat capacity corrections to a standard state: a comparison of new and some literature methods for organic liquids and solids, *Struct. Chem.* 4 (1993) 271–278. <https://doi.org/10.1007/BF00673701>.
- [87] J.S. Chickos, A. Gavezzotti, Sublimation Enthalpies of Organic Compounds: A Very Large Database with a Match to Crystal Structure Determinations and a Comparison with Lattice Energies, *Cryst. Growth Des.* 19 (2019) 6566–6576. <https://doi.org/10.1021/acs.cgd.9b01006>.
- [88] N.V. Sidgwick, Some Physical Properties of The Covalent Link in Chemistry, Cornell University Press, London, 1933.
- [89] R.M. Stephenson, S. Malanowski, Handbook of the Thermodynamics of Organic Compounds, Springer Netherlands, Dordrecht, 1987. <http://link.springer.com/10.1007/978-94-009-3173-2> (accessed May 5, 2016).
- [90] D.R. Stull, Vapor Pressure of Pure Substances. Organic and Inorganic Compounds, *Ind. Eng. Chem.* 39 (1947) 517–540. <https://doi.org/10.1021/ie50448a022>.
- [91] I. Wadsö, G.L. Borgen, G.O. Sørensen, G. Olsen, G. Jansen, Thermochemical Properties of Diacetamide, N-Butyldiacetamide and N-Phenyldiacetamide., *Acta Chem. Scand.* 19 (1965) 1079–1087. <https://doi.org/10.3891/acta.chem.scand.19-1079>.
- [92] V.N. Emel'yanenko, G.J. Kabo, S.P. Verevkin, Measurement and Prediction of Thermochemical Properties: Improved Increments for the Estimation of Enthalpies of Sublimation and Standard Enthalpies of Formation of Alkyl Derivatives of Urea, *J. Chem. Eng. Data.* 51 (2006) 79–87. <https://doi.org/10.1021/je050230z>.
- [93] D. Zaitsau, G.J. Kabo, A.A. Kozyro, V.M. Sevruk, The effect of the failure of isotropy of a gas in an effusion cell on the vapor pressure and enthalpy of sublimation for alkyl derivatives of carbamide, *Thermochim. Acta.* 406 (2003) 17–28. [https://doi.org/10.1016/S0040-6031\(03\)00231-4](https://doi.org/10.1016/S0040-6031(03)00231-4).
- [94] H.G.M. De Wit, J.C. Van Miltenburg, C.G. De Kruif, Thermodynamic properties of molecular organic crystals containing nitrogen, oxygen, and sulphur 1. Vapour pressures and enthalpies of sublimation, *J. Chem. Thermodyn.* 15 (1983) 651–663. [https://doi.org/10.1016/0021-9614\(83\)90079-4](https://doi.org/10.1016/0021-9614(83)90079-4).

- [95] D. Ferro, G. Barone, G. Della Gatta, V. Piacente, Vapour pressures and sublimation enthalpies of urea and some of its derivatives, *J. Chem. Thermodyn.* 19 (1987) 915–923. [https://doi.org/10.1016/0021-9614\(87\)90038-3](https://doi.org/10.1016/0021-9614(87)90038-3).
- [96] K. Suzuki, S. Onishi, T. Koide, S. Seki, Vapor Pressures of Molecular Crystals. XI. Vapor Pressures of Crystalline Urea and Diformylhydrazine. Energies of Hydrogen Bonds in these Crystals, *Bull. Chem. Soc. Jpn.* 29 (1956) 127–131. <https://doi.org/10.1246/bcsj.29.127>.
- [97] H.J. Svec, D.D. Clyde, Vapor Pressures of Some α -Amino Acids., *J. Chem. Eng. Data.* 10 (1965) 151–152. <https://doi.org/10.1021/je60025a024>.
- [98] N. Matsubara, T. Kuwamoto, Vapor pressure measurements in carrier gas containing ligand vapor using the transpiration technique, *Thermochim. Acta.* 83 (1985) 193–202. [https://doi.org/10.1016/0040-6031\(85\)87003-9](https://doi.org/10.1016/0040-6031(85)87003-9).
- [99] R. Pena, J.P. Ribet, J.L. Maurel, L. Valat, F. Lacoulonche, A. Chauvet, Sublimation and vaporisation processes of S(–) efroxan hydrochloride, *Thermochim. Acta.* 408 (2003) 85–96. [https://doi.org/10.1016/S0040-6031\(03\)00321-6](https://doi.org/10.1016/S0040-6031(03)00321-6).
- [100] S. Klosky, L.P.L. Woo, R.J. Flanigan, THE VAPOR-PRESSURE CURVE OF BENZOIC ACID, *J. Am. Chem. Soc.* 49 (1927) 1280–1284. <https://doi.org/10.1021/ja01404a017>.
- [101] J.S. Chickos, S. Hosseini, D.G. Hesse, Determination of vaporization enthalpies of simple organic molecules by correlations of changes in gas chromatographic net retention times, *Thermochim. Acta.* 249 (1995) 41–62. [https://doi.org/10.1016/0040-6031\(95\)90670-3](https://doi.org/10.1016/0040-6031(95)90670-3).
- [102] J.S. Chickos, W.E. Acree Jr., Enthalpies of Sublimation of Organic and Organometallic Compounds. 1910–2001, *J. Phys. Chem. Ref. Data.* 31 (2002) 537–698. <https://doi.org/10.1063/1.1475333>.
- [103] Yu.A. Lebedev, E.A. Miroshnichenko, *Thermochemistry of vapor formation of organic compounds*, Nauka, Moscow, 1981.
- [104] E. Kaisersberger, W. Hädrich, W.-D. Emmerich, Measurement of low vapour pressures according to the Knudsen effusion method, *Thermochim. Acta.* 95 (1985) 331–336. [https://doi.org/10.1016/0040-6031\(85\)85294-1](https://doi.org/10.1016/0040-6031(85)85294-1).
- [105] R.D. Chirico, S.E. Knipmeyer, A. Nguyen, W.V. Steele, The thermodynamic properties to the temperature 700 K of naphthalene and of 2,7-dimethylnaphthalene, *J. Chem. Thermodyn.* 25 (1993) 1461–1494. <https://doi.org/10.1006/jcht.1993.1148>.
- [106] C.G. de Kruif, T. Kuipers, J.C. van Miltenburg, R.C.F. Schaake, G. Stevens, The vapour pressure of solid and liquid naphthalene, *J. Chem. Thermodyn.* 13 (1981) 1081–1086. [https://doi.org/10.1016/0021-9614\(81\)90006-9](https://doi.org/10.1016/0021-9614(81)90006-9).
- [107] A.B. Bazyleva, A.V. Blokhin, G.J. Kabo, M.B. Charapennikau, V.N. Emel'yanenko, S.P. Verevkin, V. Diky, Thermodynamic Properties of Adamantane Revisited, *J. Phys. Chem. B.* 115 (2011) 10064–10072. <https://doi.org/10.1021/jp204792b>.
- [108] Y.D. Lei, R. Chankalal, A. Chan, F. Wania, Supercooled Liquid Vapor Pressures of the Polycyclic Aromatic Hydrocarbons, *J. Chem. Eng. Data.* 47 (2002) 801–806. <https://doi.org/10.1021/je0155148>.
- [109] D.A. Hinckley, T.F. Bidleman, W.T. Foreman, J.R. Tuschall, Determination of vapor pressures for nonpolar and semipolar organic compounds from gas chromatographic retention data, *J. Chem. Eng. Data.* 35 (1990) 232–237. <https://doi.org/10.1021/je00061a003>.
- [110] V.N. Emel'yanenko, D.H. Zaitsev, S.P. Verevkin, Thermochemical Properties of Xanthine and Hypoxanthine Revisited, *J. Chem. Eng. Data.* 62 (2017) 2606–2609. <https://doi.org/10.1021/acs.jced.7b00085>.
- [111] A.B. Teplitskii, I.K. Yanson, Effect of substituents on the heat of sublimation of nucleic acid nitrogenous bases, *Biofizika.* 20 (1975) 189–193.
- [112] U.J. Griesser, M. Szelagiewicz, U.Ch. Hofmeier, C. Pitt, S. Cianferani, Vapor Pressure and Heat of Sublimation of Crystal Polymorphs, *J. Therm. Anal. Calorim.* 57 (1999) 45–60. <https://doi.org/10.1023/A:1010188923713>.

- [113] A. Abdelaziz, D.H. Zaitsau, A. Buzyurov, A.A. Minakov, S.P. Verevkin, C. Schick, Fast scanning calorimetry: Sublimation thermodynamics of low volatile and thermally unstable compounds, *Thermochim. Acta.* 676 (2019) 249–262. <https://doi.org/10.1016/j.tca.2019.05.008>.
- [114] E.F.C. Byrd, B.M. Rice, Improved Prediction of Heats of Formation of Energetic Materials Using Quantum Mechanical Calculations, *J. Phys. Chem. A.* 110 (2006) 1005–1013. <https://doi.org/10.1021/jp0536192>.
- [115] H. Östmark, S. Wallin, H.G. Ang, Vapor Pressure of Explosives: A Critical Review, *Propellants Explos. Pyrotech.* 37 (2012) 12–23. <https://doi.org/10.1002/prep.201100083>.
- [116] P.A. Pella, Generator for producing trace vapor concentrations of 2,4,6-trinitrotoluene, 2,4-dinitrotoluene, and ethylene glycol dinitrate for calibrating explosives vapor detectors, *Anal. Chem.* 48 (1976) 1632–1637. <https://doi.org/10.1021/ac50005a054>.
- [117] A. Freedman, P.L. Kebabian, Z. Li, W.A. Robinson, J.C. Wormhoudt, Apparatus for determination of vapor pressures at ambient temperatures employing a Knudsen effusion cell and quartz crystal microbalance, *Meas. Sci. Technol.* 19 (2008) 125102. <https://doi.org/10.1088/0957-0233/19/12/125102>.
- [118] L. Rittfeldt, Determination of Vapor Pressure of Low-Volatility Compounds Using a Method To Obtain Saturated Vapor with Coated Capillary Columns, *Anal. Chem.* 73 (2001) 2405–2411. <https://doi.org/10.1021/ac010015v>.
- [119] D.S. Viswanath, T.K. Ghosh, V.M. Boddu, Emerging Energetic Materials: Synthesis, Physicochemical, and Detonation Properties, Springer Netherlands, Dordrecht, 2018. <https://doi.org/10.1007/978-94-024-1201-7>.
- [120] R.B. Cundall, T. Frank Palmer, C.E.C. Wood, Vapour pressure measurements on some organic high explosives, *J. Chem. Soc. Faraday Trans. 1 Phys. Chem. Condens. Phases.* 74 (1978) 1339. <https://doi.org/10.1039/f19787401339>.
- [121] Hwang D, Yoshizawa, F, Tamura, M., T. Yoshida, Determination of the enthalpies of phase change for some of energetic materials by using the DSC method under reduced pressure, *Anzen Kogaku (Journal of Japan Society for Safety Engineering)*. 29 (1990) 168–174.
- [122] V.P. Sinditskii, A.I. Levshenkov, V.Yu. Egorshev, V.V. Serushkin, Study on Combustion and Thermal Decomposition of 1,1-Diamino-2,2-dinitroethylene (FOX-7), in: Proc. 30th Int. Pyrotech. Semin., IPSUSA Seminars, 2003: pp. 299–311.
- [123] A.A. Kushtaev, D.Y. Nekrasov, N.N. Kondakova, Sintez i nekotorye svojstva dejterirovannogo 1,1-diamino-2,2-dinitrojetilena, in: Usp. V Spec. Him. Him. Tehnol., Moscow, 2015: pp. 142–147.
- [124] N.N. Volkova, A.V. Shastin, Z.G. Aliev, V.G. Prokudin, Osobennosti sublimacii 1,1-diamino-2,2-dinitrojetilena i ego proizvodnyh, in: Usp. V Spec. Him. Him. Tehnol., Moscow, 2015: pp. 142–147.
- [125] L.M.G. Minier, R. Behrens, Thermal decomposition mechanisms of bis(2-fluoro-2,2-dinitroethyl) formal (FEFO) and bis(2-fluoro-2,2-dinitroethyl) difluoroformal (DFF) from simultaneous thermogravimetric modulated beam mass spectrometry (STMBMS) measurements, *Propellants Explos. Pyrotech.* 22 (1997) 23–33. <https://doi.org/10.1002/prep.19970220107>.
- [126] A.M. Astakhov, K.P. Dyugaev, A.A. Kuzubov, V.A. Nasluzov, A.D. Vasiliev, É.S. Buka, Theoretical studies of the structure of nitrimines. I. Structure of 2-nitroguanidine and its alkyl derivatives, *J. Struct. Chem.* 50 (2009) 201–211. <https://doi.org/10.1007/s10947-009-0030-2>.
- [127] E.A. Miroshnichenko, T.S. Kon'kova, Yu.N. Matyushin, Ya.O. Inozemtsev, Bond dissociation energies in nitramines, *Russ. Chem. Bull.* 58 (2009) 2015–2019. <https://doi.org/10.1007/s11172-009-0275-0>.

- [128] V.I. Pepekin, Yu.N. Matyushin, Yu.A. Lebedev, Thermochemistry of N-nitro- and N -nitrosoamines of the alicyclic series, Bull. Acad. Sci. USSR Div. Chem. Sci. 23 (1974) 1707–1710. <https://doi.org/10.1007/BF00923193>.
- [129] V.P. Sinditskii, V.Yu. Egorshev, V.V. Serushkin, A.I. Levshenkov, M.V. Berezin, S.A. Filatov, S.P. Smirnov, Evaluation of decomposition kinetics of energetic materials in the combustion wave, Thermochim. Acta. 496 (2009) 1–12. <https://doi.org/10.1016/j.tca.2009.07.004>.
- [130] J.W. Taylor, R.J. Crookes, Vapour pressure and enthalpy of sublimation of 1,3,5,7-tetranitro-1,3,5,7-tetra-azacyclo-octane (HMX), J. Chem. Soc. Faraday Trans. 1 Phys. Chem. Condens. Phases. 72 (1976) 723. <https://doi.org/10.1039/f19767200723>.
- [131] J. Yang, G. Wang, X. Gong, X. Wei, Comparing the performance of aliphatic azido nitramines, nitrate esters, and nitro compounds: theoretical design and investigation, Can. J. Chem. 93 (2015) 1232–1238. <https://doi.org/10.1139/cjc-2015-0266>.
- [132] K. Gańczyk, A. Zygmunt, T. Gołofit, Thermal properties of TEX decomposition or sublimation, J. Therm. Anal. Calorim. 125 (2016) 967–975. <https://doi.org/10.1007/s10973-016-5476-9>.
- [133] O.V. Dorofeeva, M.A. Suntsova, Enthalpy of formation of CL-20, Comput. Theor. Chem. 1057 (2015) 54–59. <https://doi.org/10.1016/j.comptc.2015.01.015>.
- [134] L.M. Kostikova, E.A. Miroshnichenko, Y.O. Inozemtcev, Yu.N. Matyushin, Efficient Energies of Interaction of Functional Groups and Energies of Dissociation Bonds in Alkylnitrates, in: Proc. 32th Int. Annu. Conf. ICT, Karlsruhe, Germany, 2001: p. 104.1-104.9.
- [135] M.A. Suntsova, (PhD Thesis) Prediction of enthalpy of formation of novel nitrogen-based high-energy substances using the quantum chemical computations, Lomonosov Moscow State University, 2016. <http://www.chem.msu.ru/rus/theses/2016/2016-10-10-suntsova/fulltext.pdf>.
- [136] E.A. Miroshnichenko, V.P. Lebedev, L.M. Kostikova, V.P. Vorob'eva, A.B. Vorob'ev, Y.O. Inozemtcev, Enthalpy characteristics of nitro- and fluorine-nitro derivatives heterocyclic compounds, in: Proc. 34th Int. Annu. Conf. Fraunhofer ICT, Pfinztal: Fraunhofer ICT, 2003: pp. 1–7.
- [137] D. Price, GrIMEx: Development of a Novel, Green IM Comp B Replacement, (2016). <https://ndiastorage.blob.core.usgovcloudapi.net/ndia/2016/IMEM/18770price.pdf>.
- [138] M.R. Manaa, I.-F.W. Kuo, L.E. Fried, First-principles high-pressure unreacted equation of state and heat of formation of crystal 2,6-diamino-3, 5-dinitropyrazine-1-oxide (LLM-105), J. Chem. Phys. 141 (2014) 064702. <https://doi.org/10.1063/1.4891933>.
- [139] N. Cheng, Q. Gan, Q. Yu, X. Zhang, R. Li, S. Qian, C. Feng, Initial Mechanisms for the Unimolecular Thermal Decomposition of 2,6-Diamino-3,5-dinitropyrazine-1-oxide, Molecules. 24 (2018) 125. <https://doi.org/10.3390/molecules24010125>.
- [140] A.A. Kozyro, V.V. Simirskii, A.P. Krasulin, V.M. Sevruk, G.Y. Kabo, M.L. Frenkel, P.N. Gaponik, Y.V. Grigor'ev, Thermodynamic properties of tetrazolederivatives in different aggregation states, Zh Fiz Khim. 64 (1990) 656–661.
- [141] M.A. Suntsova, O.V. Dorofeeva, Prediction of enthalpies of sublimation of high-nitrogen energetic compounds: Modified Politzer model, J. Mol. Graph. Model. 72 (2017) 220–228. <https://doi.org/10.1016/j.jmgm.2017.01.013>.
- [142] V.P. Sinditskii, M.C. Vu, A.B. Sheremetev, N.S. Alexandrova, Study on thermal decomposition and combustion of insensitive explosive 3,3'-diamino-4,4'-azofurazan (DAAzF), Thermochim. Acta. 473 (2008) 25–31. <https://doi.org/10.1016/j.tca.2008.04.004>.
- [143] V.P. Sinditskii, M.C. Vu, V. Shelaputina, A.B. Sheremetev, N.S. Alexandrova, Study on thermal decomposition and combustion of insensitive explosives 3,3'-diamino-4,4'-azofurazan (DAAzF) and 3,3'-diamino-4,4'-azoxyfurazan (DAAF), in: Theory Pract. Energ. Mater. Proc 7th Int Autumn Sem Propellants Explos. Pyrotech., Science Press, Beijing, Xi'an, China, 2007: pp. 422–428.

- [144] D.S. Barnes, G. Pilcher, Enthalpies of combustion of ethanamide, propanamide, and butanamide, *J. Chem. Thermodyn.* 7 (1975) 377–382. [https://doi.org/10.1016/0021-9614\(75\)90176-7](https://doi.org/10.1016/0021-9614(75)90176-7).
- [145] B. Chan, L. Radom, W2X and W3X-L: Cost-Effective Approximations to W2 and W4 with kJ mol^{-1} Accuracy, *J. Chem. Theory Comput.* 11 (2015) 2109–2119. <https://doi.org/10.1021/acs.jctc.5b00135>.
- [146] B. Narayanan, P.C. Redfern, R.S. Assary, L.A. Curtiss, Accurate quantum chemical energies for 133 000 organic molecules, *Chem. Sci.* 10 (2019) 7449–7455. <https://doi.org/10.1039/C9SC02834J>.
- [147] O.V. Dorofeeva, O.N. Ryzhova, Gas-Phase Enthalpies of Formation and Enthalpies of Sublimation of Amino Acids Based on Isodesmic Reaction Calculations, *J. Phys. Chem. A.* 118 (2014) 3490–3502. <https://doi.org/10.1021/jp501357y>.
- [148] J.P. Guthrie, P.A. Cullimore, Effect of the acyl substituent on the equilibrium constant for hydration of esters, *Can. J. Chem.* 58 (1980) 1281–1294. <https://doi.org/10.1139/v80-201>.
- [149] M.V. Roux, M. Temprado, J.S. Chickos, Y. Nagano, Critically Evaluated Thermochemical Properties of Polycyclic Aromatic Hydrocarbons, *J. Phys. Chem. Ref. Data.* 37 (2008) 1855–1996. <https://doi.org/10.1063/1.2955570>.
- [150] A. Karton, J.M.L. Martin, Explicitly correlated W_n theory: W1-F12 and W2-F12, *J. Chem. Phys.* 136 (2012) 124114. <https://doi.org/10.1063/1.3697678>.
- [151] N.T. Cuong, T.B. Tai, V.T.T. Ha, M.T. Nguyen, Thermochemical parameters of caffeine, theophylline, and xanthine, *J. Chem. Thermodyn.* 42 (2010) 437–440. <https://doi.org/10.1016/j.jct.2009.10.006>.
- [152] M.A. Suntsova, O.V. Dorofeeva, Use of G4 Theory for the Assessment of Inaccuracies in Experimental Enthalpies of Formation of Aromatic Nitro Compounds, *J. Chem. Eng. Data.* 61 (2016) 313–329. <https://doi.org/10.1021/acs.jcd.5b00558>.
- [153] X. Zhang, X. Gong, A computational study on new oxidizers as replacements for ammonium perchlorate: tetranitroacetimidic acid and tetranitroacetamide, *Can. J. Chem.* 95 (2017) 199–206. <https://doi.org/10.1139/cjc-2016-0468>.
- [154] M.A. Suntsova, O.V. Dorofeeva, Use of G4 Theory for the Assessment of Inaccuracies in Experimental Enthalpies of Formation of Aliphatic Nitro Compounds and Nitramines, *J. Chem. Eng. Data.* 59 (2014) 2813–2826. <https://doi.org/10.1021/je500440y>.
- [155] M.A. Suntsova, O.V. Dorofeeva, Comment on “Use of G4 Theory for the Assessment of Inaccuracies in Experimental Enthalpies of Formation of Aliphatic Nitro Compounds and Nitramines,” *J. Chem. Eng. Data.* 60 (2015) 1532–1533. <https://doi.org/10.1021/acs.jcd.5b00159>.
- [156] A. Osmont, L. Catoire, I. Gökalp, V. Yang, Ab initio quantum chemical predictions of enthalpies of formation, heat capacities, and entropies of gas-phase energetic compounds, *Combust. Flame.* 151 (2007) 262–273. <https://doi.org/10.1016/j.combustflame.2007.05.001>.
- [157] V.G. Kiselev, C.F. Goldsmith, Accurate Prediction of Bond Dissociation Energies and Barrier Heights for High-Energy Caged Nitro and Nitroamino Compounds Using a Coupled Cluster Theory, *J. Phys. Chem. A.* 123 (2019) 4883–4890. <https://doi.org/10.1021/acs.jpca.9b01506>.
- [158] X.L. Zeng, X.H. Ju, H.X. Gao, Theoretical Study of 4,10-Dinitro-2,6,8,12-tetraoxa-4,10-diaza-tetracyclododecane (TEX), *Adv. Mater. Res.* 554–556 (2012) 1618–1623. <https://doi.org/10.4028/www.scientific.net/AMR.554-556.1618>.
- [159] V.D. Ghule, P.M. Jadhav, R.S. Patil, S. Radhakrishnan, T. Soman, Quantum-Chemical Studies on Hexaazaisowurtzitanes, *J. Phys. Chem. A.* 114 (2010) 498–503. <https://doi.org/10.1021/jp9071839>.

- [160] T.M. Klapötke, B. Krumm, F.X. Steemann, K.-D. Umland, Bis(1,3-dinitratorprop-2-yl) Nitramine, a New Sensitive Explosive Combining a Nitrate Ester with a Nitramine, *Z. Für Anorg. Allg. Chem.* 636 (2010) 2343–2346. <https://doi.org/10.1002/zaac.201000229>.
- [161] M.A.M. Rashid, S.G. Cho, T.H. Choi, C.H. Choi, Heat of formation predictions of various nitro-substituted azoles by G4MP2-SFM scheme, *Theor. Chem. Acc.* 134 (2015) 126. <https://doi.org/10.1007/s00214-015-1733-4>.
- [162] Z. Yu, E.R. Bernstein, Sensitivity and Performance of Azole-Based Energetic Materials, *J. Phys. Chem. A.* 117 (2013) 10889–10902. <https://doi.org/10.1021/jp4054007>.
- [163] S. Comte, G. Daquin, Synthesis of Di- and Trinitropyrazoles, in: Proc. 43rd Int. Annu. Conf. Fraunhofer ICT, Pfinztal: Fraunhofer ICT, 2012: p. 5.1-5.7.
- [164] E. Gökçinär, T.M. Klapötke, A.J. Bellamy, Computational study on 2,6-diamino-3,5-dinitropyrazine and its 1-oxide and 1,4-dioxide derivatives, *J. Mol. Struct. THEOCHEM.* 953 (2010) 18–23. <https://doi.org/10.1016/j.theochem.2010.04.015>.
- [165] V.G. Kiselev, P.B. Cheblakov, N.P. Gritsan, Tautomerism and Thermal Decomposition of Tetrazole: High-Level ab Initio Study, *J. Phys. Chem. A.* 115 (2011) 1743–1753. <https://doi.org/10.1021/jp112374t>.
- [166] N. Fischer, T.M. Klapötke, M. Reymann, J. Stierstorfer, Nitrogen-Rich Salts of 1 *H*,1' *H*-5,5'-Bitetrazole-1,1'-diol: Energetic Materials with High Thermal Stability, *Eur. J. Inorg. Chem.* 2013 (2013) 2167–2180. <https://doi.org/10.1002/ejic.201201192>.
- [167] J. Zhang, J.M. Shreeve, 3,3'-Dinitroamino-4,4'-azoxyfurazan and Its Derivatives: An Assembly of Diverse N–O Building Blocks for High-Performance Energetic Materials, *J. Am. Chem. Soc.* 136 (2014) 4437–4445. <https://doi.org/10.1021/ja501176q>.
- [168] P. Gao, Structures and properties of the isomers of hexanitrosobenzene: A DFT study, *J. Mol. Struct. THEOCHEM.* 767 (2006) 119–130. <https://doi.org/10.1016/j.theochem.2006.05.007>.
- [169] D. Fischer, T.M. Klapötke, J. Stierstorfer, Synthesis and Characterization of Diaminobisfuroxane: Synthesis and Characterization of Diaminobisfuroxane, *Eur. J. Inorg. Chem.* 2014 (2014) 5808–5811. <https://doi.org/10.1002/ejic.201402960>.