

Electronic supplementary information (ESI) for:

**Mapping the role of aromatic amino acids within a blue-light sensing
LOV domain**

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Content:

Table S1, p. 2: primers for the variants of *mrad2831_4511*

Fig. S1, p. 3: sequence alignment of *Mr4511-C71S* with similar proteins and structural model

Fig. S2, p. 4: absorption and fluorescence spectra of *Mr4511* variants

Fig. S3, p. 5-6: absorption spectra in the dark and under BL

Table S2, p. 5: raw photoacoustics (PA) data for the investigated molecules

Fig. S4, p. 8: PA signals for *Mr4511-C71S/Y116W* and Arrhenius plot for the triplet decay

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Table S1. Primers for the variants of *mrad2831_4511*. See¹ for C71S and C71S/Q112W

Primer	Sequence (5' to 3')	Mr4511 variants
P1	TGGCTGACCGGCTACACCC	C71S/K57W
P2	CAGGAACCGCGTCGTTACGAA	
P3	TTTGTCCGGGCCGGTGC	C71S/Y116F
P4	GAGGGCGTTCTGGAAGGTC	
P5	CATGTCGGGCCGGTGC	C71S/Y116H
P6	GAGGGCGTTCTGGAAGGTC	
P7	TGGGTCCGGGCCGGTGC	C71S/Y116W
P8	GAGGGCGTTCTGGAAGGTCGAG	
P9	CACTTCTCGCCTCGCAGCT	C71S/Y129H
P10	GACCACCCGGCCCCGCT	
P11	TGGTTCTCGCCTCGCAGCT	C71S/Y129W
P12	GACCACCCGGCCCCGCT	
P13	CACTTCCGCTCGCAGCTCGA	C71S/F130H
P14	GTAGACCACCCGGCCCCGC	
P15	GTTCGCCTCGCAGCTCGA	C71S/F130W
P16	CAGTAGACCACCCGGCCCCG	
P17	(same as P15)	F130W
P18	(same as P16)	

		A β	B β	C α	D α	E α	
		. EEEEEE EEEEEE HHHHHHHH . . .				
		↓ ↓	↓	↓	↓	↓ ↓	
BsYtvA	25	VGVVITDPALEDNPPIVYVNQGFVQMTGYETEEILGKNCRFLQ-----					
Mr4511-C71S	34	MPMIITDPAQHDNPIVFVNDAFLKLTGYTRMEVGRNSRFLQ-----					
Crphot-LOV1	20	HTFVVADATLPDCPLVYASEG F YAMTGYGPDEVLGHNCRFLQ-----					
NcVVD-C108A	71	CALILCDLKQKDTPIVYASEAFLYMTGYSNAEVLRNARFLQSPDGMVKPKST					
AtPHOT2-LOV2	389	KNFVISDPRLPDNPIIFASDSFLELTEYSREEILGRNCRFLQ-----					
iLOV	389	KNFVITDPRLPDNPIIFASDGFLELTEYSREEILGRNCRFLQ-----					
HhBAT	151	IGISISDPDLPDYPLVYVNDAWREHTGYSVEEVLRNPRFLQ					

	F α	G β	H β	I β			
 HHHHHHHHHHHHHH EEEEEEEE EEEEEEEEEE EEEEEEEEEE . . .			
	↓ ↓	↓	↓	↓ ↓	↓ ↓	↓ ↓	
GKHTDPAEVNDNIRALQNKEPVTVQIQNYKKDGTMFWNELNIDPMEI--EDKTYFVGIQNDI					126	O34627	
GPDTEAAAVDRLLRAAIRREEDIRVDLLNRYRKDGSTF Q NALYVGPVRDEAGRVV Y FFASQLDV					137	B1M516	
GEGTDPKEVQKIRDAIKKGAECSVRLLNRYRKDGTPFWNLLTVTPIKTPDGRVSFKFGVQDV					123	Q8LPE0	
RKYVDSNTINTMRKAI DRNAEVQVEVVFNKNGQRFVNFTLMIPVRDETGEYRYSMGFQCET					185	Q1K5Y8	
GPETDQATVQKIRDAIRDQREITVQLINYTKSGKKFWNLFHLPQMRDQKGELOQYFIGVQLDG					492	P93025	
GPETDQATVQKIRDAIRDQREITVQLINYTKSGKKFWNLHLQPMRDQKGELOQYFIGV Q LDG					492		
GPGETDQATVQKIRDAIRDQREITVQLINYTKSGKKFWNLQPMRDQKGELOQYFIGV Q LDG					254	M0FIW0	

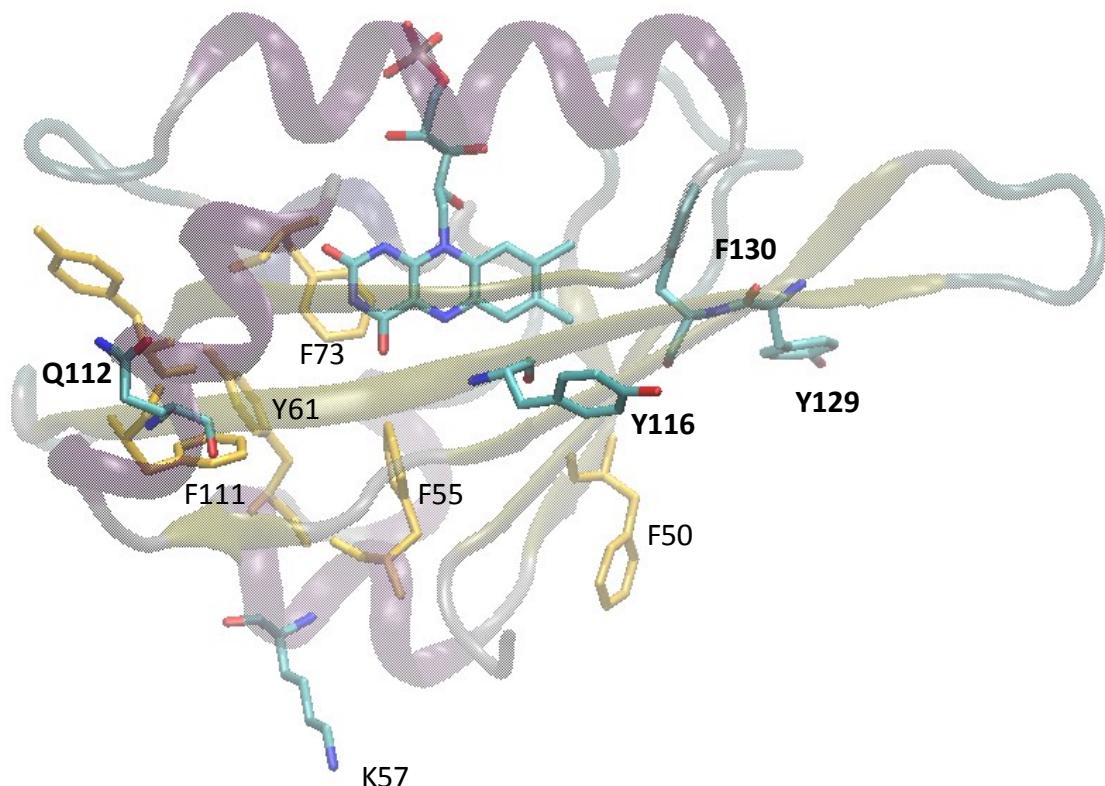


Fig. S1. Top, sequence alignment (performed with the CLUSTAL Omega tool at the European Bioinformatics Institute, EMBL-EBI, using default parameters) of the LOV core of *Mr4511* with other LOV domains mentioned in the text; the secondary structure elements are derived from the structure of *BsYtvA* (PDB access code 2pr5), indicated as E (yellow, β -strands) and H (red, α -helices). Residues interacting with the chromophore are indicated with arrows. UniProt accession codes are given at the end of each sequence. The site of the reactive cysteine in photosensing LOV domains is shown in white on blue background; mutated residues mentioned in the text are evidenced in cyan; *Bs* = *Bacillus subtilis*; *Nc* = *Neurospora crassa*; *Cr* = *Chlamydomonas reinhardtii*; *At* = *Arabidopsis thaliana*; *Hh* = *Halorubrum hochstetnium*. Bottom, structural model of *Mr4511*-LOV (same as in Fig. 2a but 90° rotated); residues labeled in bold have been mutated in this work. Structural modeling was with SWISS-MODEL in the automated mode,² rendering with VMD.³

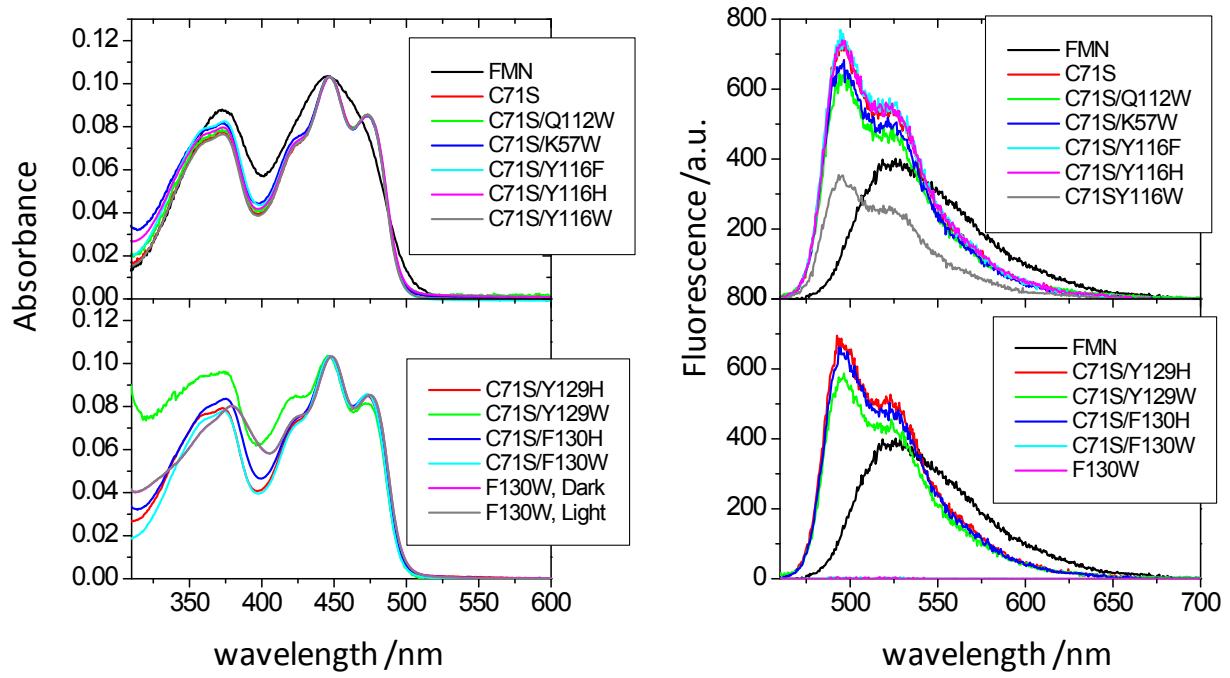


Fig. S2 Left, absorption spectra of the studied proteins; note in the bottom panel that the F130W variant does not show any difference in the absorption spectrum under dark or blue light conditions; right, fluorescence spectra taken at matched absorbance at $\lambda_{ex} = 450$ nm. Besides C71S/F130W and F130W that show negligible fluorescence, the sole variant having a sharply lower fluorescence is C71S/Y116W (top panel). Proteins C71S, C71S/Q112W, C71S/K57W, C71S/Y116W and C71S/F130W were dissolved in buffer NaPi 10 mM, NaCl 100 mM, pH = 8; all other variants in buffer KPi 50 mM, 300 mM NaCl, pH = 8. C71S and C71S/Y116W were studied in both buffers and gave consistent results.

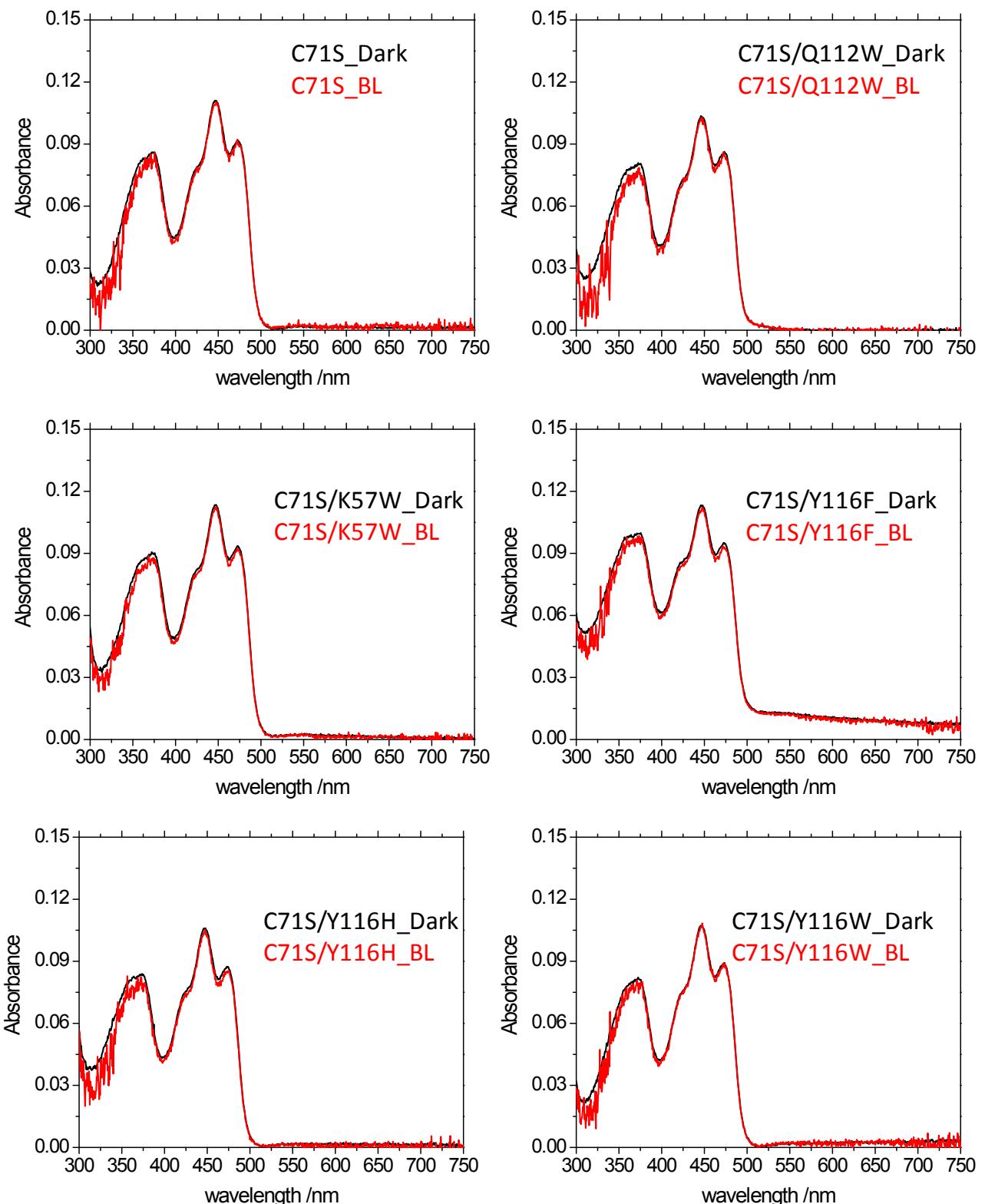


Fig. S3 (continues next page) Steady state absorption spectra of the investigated *Mr4511* variants investigated in this work in the dark (dark lines) and under BL from LED455 operating at ca. 3 mW; in this latter case 1 ml of sample (at ca. 8 μ M) was kept illuminated from above the cuvette for the entire duration of the measurement (7 minutes).

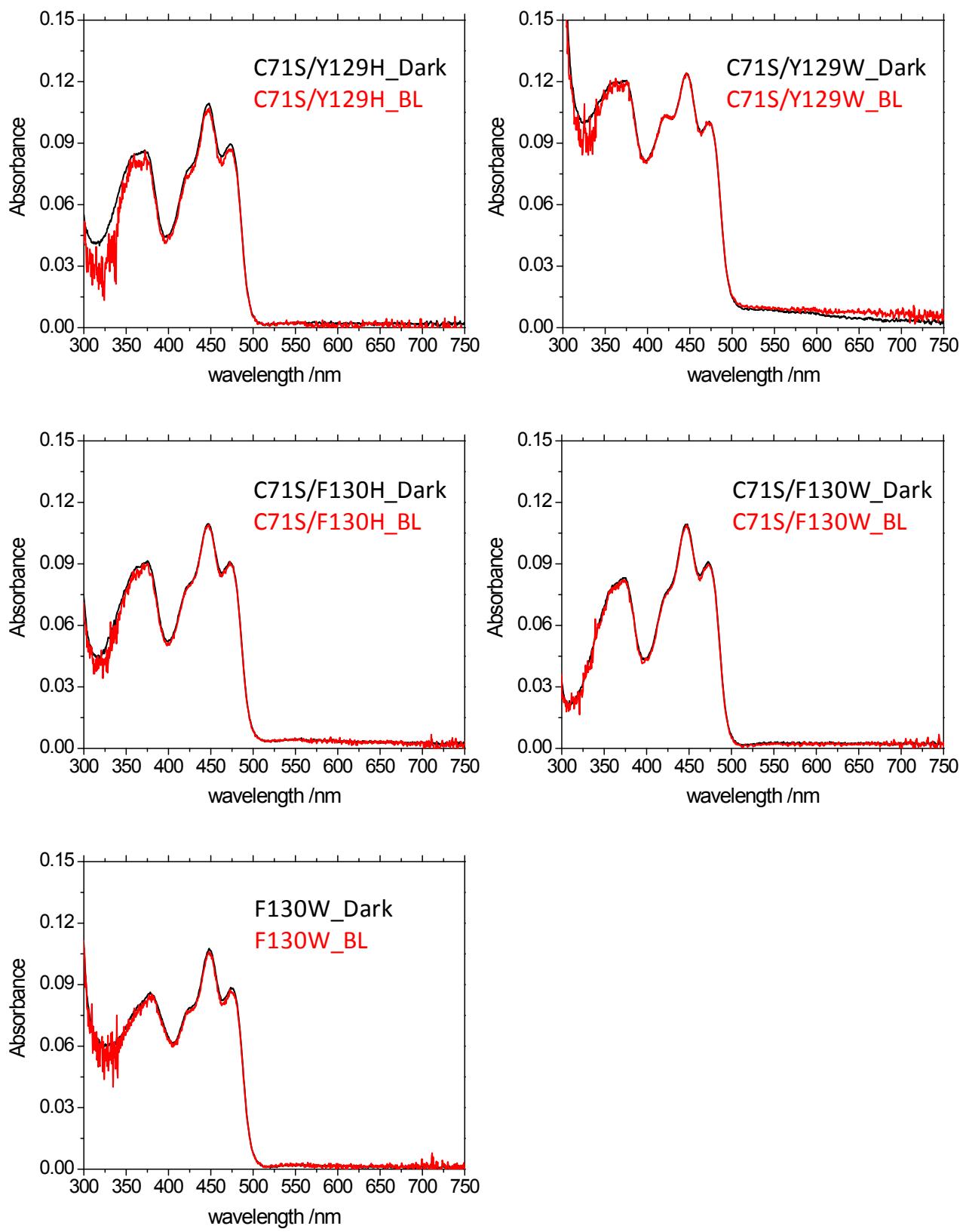
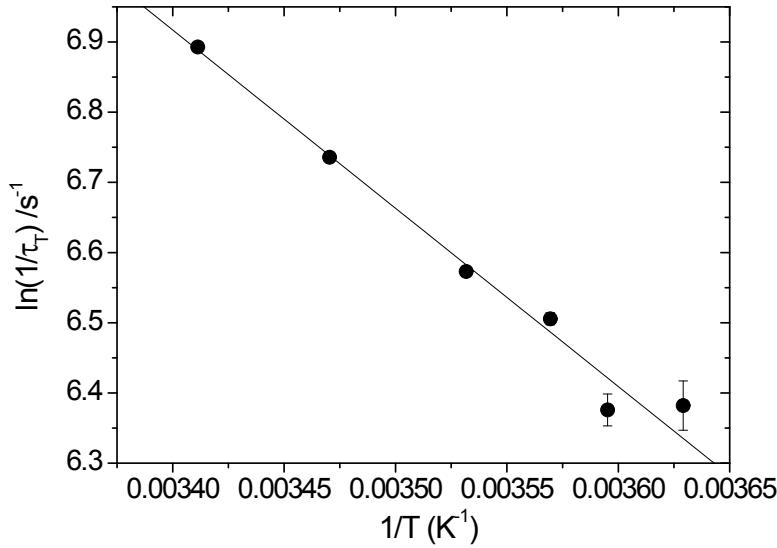
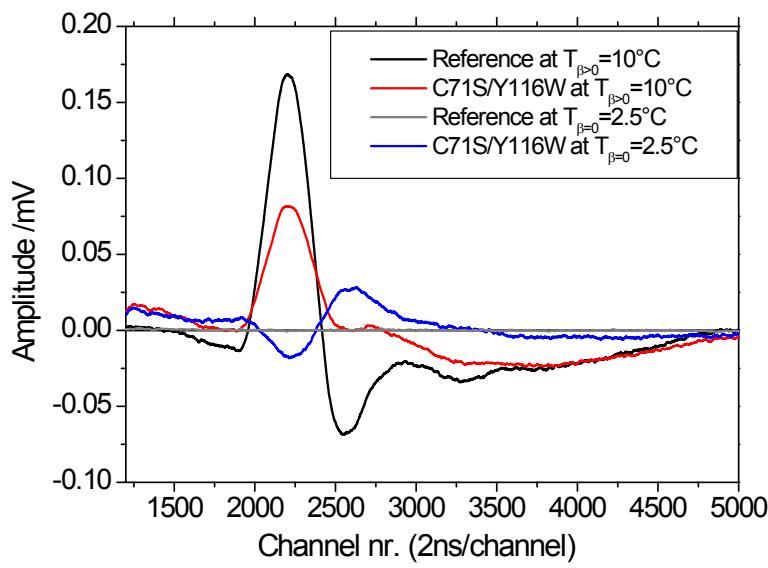


Fig. S3 (continues from previous page) Steady state absorption spectra of the investigated *Mr4511* variants investigated in this work in the dark (dark lines) and under BL from LED455 (see caption of fig. S2 for details).

Table S2. Raw PA data for the investigated proteins and free FMN

Sample	α_1 ($\tau_1 < 20\text{ns}$)	α_2	$\Delta V_1/\text{ml Einstein}^{-1}$	$\Delta V_2/\text{ml Einstein}^{-1}$	$\tau_2/\mu\text{s}$ (20°C)
FMN	0.40 ± 0.01	N.D.	-1.70 ± 0.01	N.D.	3.5 ± 0.2
Mr4511-					
C71S	0.32 ± 0.02	N.D.	-1.02 ± 0.02	N.D.	> 10
C71S/Q112W	0.30 ± 0.02	N.D.	-0.92 ± 0.02	N.D.	> 10
C71S/K57W	0.30 ± 0.02	N.D.	-0.91 ± 0.02	N.D.	> 10
C71S/Y116F	0.23 ± 0.02	N.D.	-1.33 ± 0.02	N.D.	> 10
C71S/Y116H	0.23 ± 0.02	N.D.	-1.25 ± 0.02	N.D.	> 10
C71S/Y116W	0.44 ± 0.01	0.33 ± 0.01	-1.14 ± 0.02	$+2.85 \pm 0.05$	1.02 ± 0.05
C71S/Y129H	0.28 ± 0.01	N.D.	-1.67 ± 0.02	N.D.	> 10
C71S/Y129W	0.32 ± 0.02	N.D.	-1.15 ± 0.01	N.D.	> 10
C71S/F130H	0.52 ± 0.02	N.D.	-2.27 ± 0.02	N.D.	> 10
C71S/F130W	0.99 ± 0.02	N.D.	~ 0	N.D.	N.D.
F130W	0.97 ± 0.01	N.D.	-0.36 ± 0.02	N.D.	N.D.



$$\ln \frac{1}{\tau_T} = -\frac{E_a}{R} \frac{1}{T} + \ln A$$

Fig. S4. Top, PA signals at $T_{\beta=0} = 2.5 \text{ } ^\circ\text{C}$ and at $T_{\beta=0} = 10 \text{ } ^\circ\text{C}$ for C71S/Y116W. At $T_{\beta=0}$ the signal for the reference compound (new coccine) is a zero line; note the shift of the sample signal with respect to the reference, indicative of the resolved kinetics; Bottom, Arrhenius plot for triplet decay in C71S/Y116W, as derived from the deconvolution of PA data between 3 °C and 20 °C ($\tau_2=\tau_T$). From the linear fit, an activation energy $E_a = 21 \text{ kJ/mol}$ and a pre-exponential factor $A = 5.7 \times 10^6 \text{ s}^{-1}$ was determined.

Note that the signal of the reference (H^R) and of the sample (H^S) are given by, respectively:⁴

$$H^R = K \left(E_\lambda \frac{\beta}{c_p \rho} \right)$$

and

$$H^S = K \left(Q \frac{\beta}{c_p \rho} + \Delta V \right)$$

where E_λ is the molar excitation energy, β is the volume expansion coefficient, c_p is the heat capacity at constant pressure, and ρ is the mass density of the solvent; K is an instrumental constant. Therefore at $T_{\beta=0}$, where heat transport is zero, H^R vanishes (for the reference all absorbed energy is released as heat within few ps) while H^S is solely due to volume changes of non thermal origin (ΔV), because the heat $Q = 0$

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