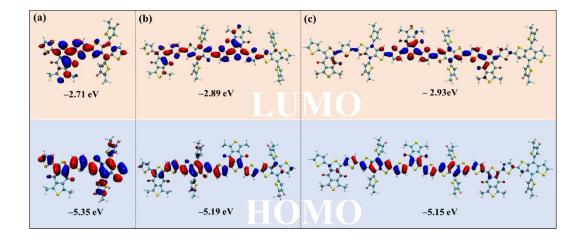
Electronic Supplementary Material (ESI) for Physical Chemistry Chemical Physics. This journal is © the Owner Societies 2021

## **Supporting Information**

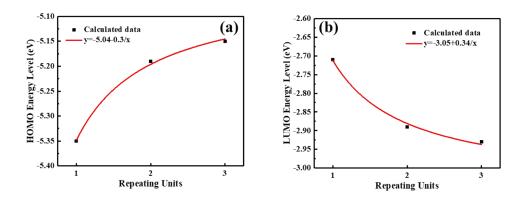
## Replacing Cyano (−C≡N) Group to Design Environmentally Friendly Fused-Ring Electron Acceptors

Chuang Yao  $^{1\ast},$  Yezi Yang  $^1,$  Lei Li  $^1,$  Maolin Bo  $^1,$  Cheng Peng  $^1,$  Zhongkai Huang  $^1$  & Jinshan Wang  $^{2\ast}$ 

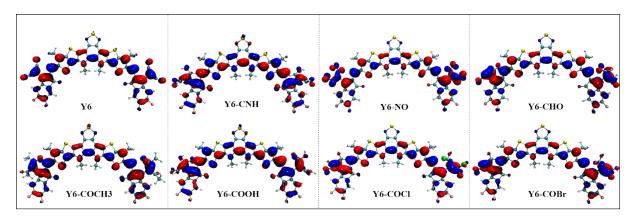


**Figure S1**. Spatial distribution and energy levels of HOMO and LUMO for one repeating unit (a), two repeating units (b), and three repeating units of PBDB-TF.

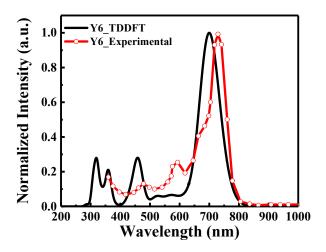
Key Laboratory of Extraordinary Bond Engineering and Advance Materials Technology (EBEAM) of Chongqing, School of Materials Science and Engineering, Yangtze Normal University, Chongqing 408100, P.R. China
School of Materials Science and Engineering, Yancheng Institute of Technology, Yancheng 224051, P. R. China



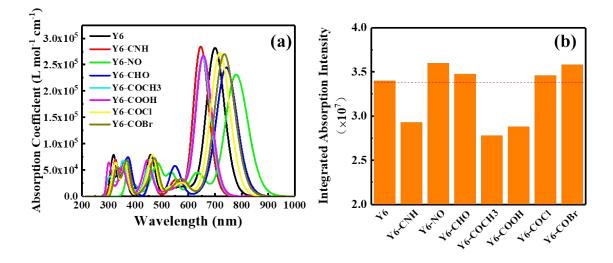
**Figure S2**. Evolution of the energy levels of HOMO (a) and LUMO (b) for one, two, and three repeating units of PBDB-TF. The red line represents the data fitting curve.



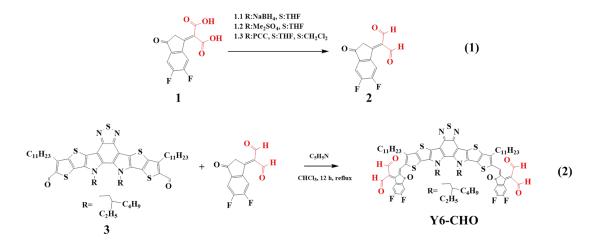
**Figure S3** Spatial distribution of LUMO for Y6, Y6-CNH, Y6-NO, Y6-CHO, Y6-COCH3, Y6-COOH, Y6-COCl, and Y6-COBr.



**Figure S4**. TD-DFT calculated normalized absorption spectrum and experimental result of Y6 in chloroform based on PBE0/def2-SVPD.



**Figure S5** TD-DFT calculated absorption spectra (a) and integrated absorption intensity (b) of Y6, Y6-CNH, Y6-NO, Y6-CHO, Y6-COCH3, Y6-COOH, Y6-COCl, and Y6-COBr in chloroform based on PBE0/def2-SVPD.



**Figure S6** Possible synthetic routes of Y6-CHO.