

## Supporting Information

### Effect of the Substituents of New Coumarin-imidazo[1,2-*a*]heterocyclic-3-acrylate Derivatives on Nonlinear Optical Properties: A combined experimental-theoretical approach

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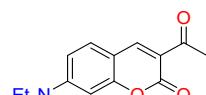
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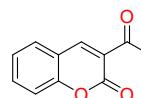
## 1. General Method

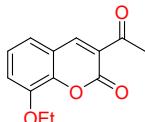
All melting points were determined on Electrothermal 9100 Melting point apparatus and were not corrected. Microwave irradiation was carried out in a Discover SP CEM microwave. Reactions were monitored by thin layer chromatography (TLC) using Sigma Aldrich TLC plates 20 x 20 locating the spots using UV light as the visualizing agent. Column chromatography was performed on silica gel 230-400 mesh. Standard work up: organic layers were dried with  $\text{Na}_2\text{SO}_4$  and concentrated in vacuum. Infrared spectra were recorded on Spectrum 100 Perkin Elmer instrument and values are reported in  $\text{cm}^{-1}$  units. UV spectra were recorded on Perkin Elmer lambda 50, and the emission spectra were recorded on Fluorescence spectrophotometer HITACHI F-7000.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were recorded on Nuclear Magnetic resonance spectrometer Bruker Avance III HD with the magnet Bruker Ascent 400 MHz or Bruker Avance III HD with the magnet Bruker Ultra Shield 500 MHz HD at 25 °. Chemical shifts are given in  $\delta$  values relative to TMS (tetramethylsilane) as internal standard. Mass spectra were recorded on Spectrometer Maxis Impact ESI-QTOF-MS, Bruker Daltonics mass spectrometer.

## 2. Synthetic procedures and spectroscopic data of compounds **1a-c**, **2a-c** and **4a-g**

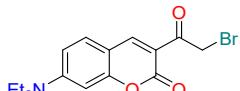
Compounds **1 a-c**, **2 a-c** and **4a-g** were synthesized using literature procedure previously.[1,2]

  
**3-acetyl-7-(diethylamino)-2H-chromen-2-one (Ia):** Yield 82%, m.p. 150 – 151 °C; RMN  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  = 8.40 (s, 1H,  $-\text{CH}=\text{C}-$ ), 7.38 (d,  $J(\text{H},\text{H})$  = 8.9 Hz, 1H, Ar), 6.62 (dd,  $J(\text{H},\text{H})$  = 8.9, 1.8 Hz, 1H, Ar), 6.44 (d,  $J(\text{H},\text{H})$  = 1.4 Hz, 1H, Ar), 3.46 (q,  $J(\text{H},\text{H})$  = 7.0 Hz, 4H,  $-\text{CH}_2\text{CH}_3$ ), 2.67 (s, 3H,  $\text{CH}_3$ ), 1.25 (t,  $J(\text{H},\text{H})$  = 7.1 Hz, 6H,  $-\text{CH}_2\text{CH}_3$ );  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  = 195.50, 160.78, 158.68, 153.00, 147.75, 131.87, 115.93, 109.87, 108.07, 96.46, 45.11, 30.51, 12.48; FT-IR (KBr)  $\nu_{\text{max}}$  = 1725  $\text{cm}^{-1}$  ( $\text{C=O}$ ), 1664  $\text{cm}^{-1}$  ( $\text{C=O}$  lactone); UV–Vis (MeOH)  $\lambda_{\text{max}}$  = 243 nm, 312 nm.

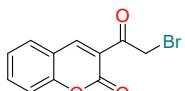
  
**3-acetyl-2H-chromen-2-one (Ib):** Yield 90%, m.p. 107 – 109 °C,  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  = 8.52 (s, 1H,  $-\text{CH}=\text{C}-$ ), 7.66 (d,  $J(\text{H},\text{H})$  = 2.8 Hz, 2H, Ar), 7.41 – 7.33 (m, 2H, Ar), 2.74 (s, 3H,  $\text{CH}_3$ );  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  = 195.5, 159.2, 155.3, 147.4, 134.3, 130.2, 125.0, 124.9, 118.3, 116.7, 30.5; FT-IR (KBr)  $\nu_{\text{max}}$  = 1741  $\text{cm}^{-1}$  ( $\text{C=O}$ ), 1678  $\text{cm}^{-1}$  ( $\text{C=O}$  lactone); UV–Vis (MeOH)  $\lambda_{\text{max}}$  = 245 nm.



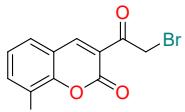
**3-acetyl-8-ethoxy-2H-chromen-2-one (1c):** Rendimiento: 95%, sólido de amarillo claro, p.f. = 133 – 136 °C, RMN  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  = 8.461(s, 1H,  $-\text{CH}=\text{C}-$ ), 7.170-7.242 (m, 3H, Ar), 4.206 (q,  $J(\text{H},\text{H})$  = 7 Hz, 2H, H-8a), 2.726 (s, 3H, H-3b), 1.518 (t,  $J(\text{H},\text{H})$  = 7 Hz, 3H, H-8b); RMN  $^{13}\text{C}$  (125 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  = 195.74, 158.97, 147.80, 146.48, 145.28, 124.89, 124.66, 121.36, 119.02, 117.21, 65.21, 30.66, 14.78; FT-IR (KBr)  $\nu_{\text{max}}$  = 1730 cm $^{-1}$  (C=O), 1716 cm $^{-1}$  (C=O lactone); UV–Vis (MeOH)  $\lambda_{\text{max}}$  = 315.



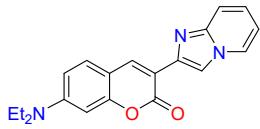
**3-(2-bromoacetyl)-7-(diethylamino)-2H-chromen-2-one (2a):** Yield 73%, m.p. 211 – 213 °C;  $^1\text{H}$  NMR (500 MHz,  $\text{DMSO}-d_6$ , 25 °C, TMS)  $\delta$  = 8.53 (s, 1H,  $-\text{CH}=\text{C}-$ ), 7.48 – 7.37 (m, 1H, Ar), 6.69 – 6.60 (m, 1H, Ar), 6.48 (d,  $J(\text{H},\text{H})$  = 2.4 Hz, 1H, Ar), 4.77 (s, 2H,  $-\text{CH}_2\text{Br}$ ), 3.48 (q,  $J(\text{H},\text{H})$  = 7.1 Hz, 4H,  $-\text{CH}_2\text{CH}_3$ ), 1.26 (t,  $J(\text{H},\text{H})$  = 7.1 Hz, 6H,  $-\text{CH}_2\text{CH}_3$ );  $^{13}\text{C}$  NMR (125 MHz,  $\text{DMSO}-d_6$ , 25 °C, TMS)  $\delta$  = 188.57, 160.55, 159.00, 153.59, 149.44, 132.22, 113.98, 112.71, 110.22, 108.44, 96.69, 45.28, 36.77, 12.45; FT-IR (KBr)  $\nu_{\text{max}}$  = 1735 cm $^{-1}$  (C=O), 1675 cm $^{-1}$  (C=O lactone); UV–Vis ( $\text{CH}_2\text{Cl}_2$ )  $\lambda_{\text{max}}$  = 245 nm, 270 nm.



**3-(2-bromoacetyl)-2H-chromen-2-one (2b):** Yield 95%, m.p. 163 – 164 °C;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  = 8.64 (1H, s,  $-\text{CH}=\text{C}-$ ), 7.71 (t,  $J(\text{H},\text{H})$  = 7.2 Hz, 2H, Ar), 7.44 – 7.35 (m, 2H, Ar), 4.76 (s, 2H,  $-\text{CH}_2\text{Br}$ );  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  = 188.90, 158.88, 155.43, 149.54, 135.12, 130.43, 125.31, 122.19, 118.15, 116.91, 35.59; FT-IR (KBr)  $\nu_{\text{max}}$  = 1727 cm $^{-1}$  (C=O), 1685 cm $^{-1}$  (C=O lactone); UV–Vis ( $\text{CH}_2\text{Cl}_2$ )  $\lambda_{\text{max}}$  = 247 nm.

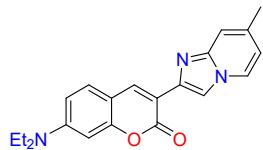


**3-(2-bromoacetyl)-8-ethoxy-2H-chromen-2-one (2c):** gray powder; yield 79%; m.p. 180 – 182 °C;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  = 8.60 (s, 1H,  $-\text{CH}=\text{C}-$ ), 7.19–7.32 (m, 3H, Ar), 4.77 (s, 2H,  $\text{CH}_2$ ), 4.23 (q,  $J(\text{H},\text{H})$  = 7.2 Hz, 2H,  $-\text{CH}_2\text{CH}_3$ ), 1.54 (t,  $J(\text{H},\text{H})$  = 7.2 Hz, 3H,  $-\text{CH}_2\text{CH}_3$ );  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  = 173.03, 162.70, 154.19, 151.42, 145.53, 127.92, 125.75, 120.01, 118.43, 115.47, 62.13, 35.10, 13.73; FT-IR (KBr)  $\nu_{\text{max}}$  = 1,722 cm $^{-1}$  (C=O), 1,692 cm $^{-1}$  (C=O lactone).



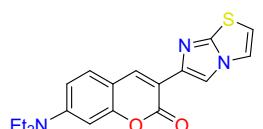
**7-(diethylamino)-3-(imidazo[1,2-a]pyridin-2-yl)-2H-chromen-2-one (4a):** Yield 92%; dark yellow powder; m.p. 167.6 – 169 °C;  $^1\text{H}$  (500 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  (ppm) = 8.65 (s, 1H,  $-\text{CH}=\text{C}-$ ), 8.44 (s, 1H,  $-\text{CH}=\text{C}-$ ), 8.11 (d,  $J(\text{H},\text{H})$  = 7.1 Hz, 1H, Ar), 7.56 (d,  $J(\text{H},\text{H})$  = 8.5 Hz, 1H, Ar), 7.41 (d,  $J(\text{H},\text{H})$  = 8.5, 1.5 Hz, 1H, Ar), 7.16 (t,  $J(\text{H},\text{H})$  = 7.5 Hz, 1H, Ar), 6.74 (t,  $J(\text{H},\text{H})$  = 7.5 Hz, 1H, Ar), 6.62 (d,  $J(\text{H},\text{H})$  = 9 Hz, 1H, Ar), 6.54 (s, 1H, Ar), 3.44 (q,  $J(\text{H},\text{H})$  = 7 Hz, 4H,  $-\text{CH}_2\text{CH}_3$ ), 1.22 (t,  $J(\text{H},\text{H})$  = 7 Hz, 6H,  $-\text{CH}_2\text{CH}_3$ ); RMN  $^{13}\text{C}$  (125 MHz,  $\text{CDCl}_3$ , TMS)  $\delta$  (ppm): 161.1, 175.8, 176.5, 164.8, 159.5, 158.8, 149.3, 145.8, 144.9, 136.6, 133.6, 132.2, 131.9, 109.3, 109.2, 97.1, 44.9, 12.6; FT-IR (KBr)  $\nu_{\text{max}}$  = 1705 cm $^{-1}$  (C=O lactone), 1619 cm $^{-1}$  (C=C Ar);

UV–Vis (MeOH)  $\lambda_{\max}$  = 421 nm; HRMS (ESI m/z) Calcd. for  $C_{20}H_{19}N_3O_2$  [M+H]<sup>+</sup> 334.1550, found 334.1564.



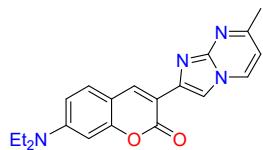
**7-(diethylamino)-3-(7-methylimidazo[1,2-a]pyridin-2-yl)-2H-chromen-2-one (4b):**

Yield 87%; dark yellow powder; m.p. 200.6 – 202°C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 8.62 (s, 1H, –CH=C–), 8.36 (s, 1H, –CH=C–), 7.98 (d, *J*(H,H) = 7 Hz, 1H, Ar), 7.41 (d, *J*(H,H) = 8.5 Hz, 1H, Ar), 7.31 (s, 1H, Ar), 6.62 (dd, *J*(H,H) = 8.5, 2 Hz, 1H, Ar), 6.59 (dd, *J*(H,H) = 7, 1.5 Hz, 1H, Ar), 6.54 (s, *J*(H,H) = 2 Hz, 1H, Ar), 3.44 (q, *J*(H,H) = 7 Hz, 4H, –CH<sub>2</sub>–CH<sub>3</sub>), 2.39 (s, 3H), 1.23 (t, *J*(H,H) = 7 Hz, 6H, –CH<sub>2</sub>–CH<sub>3</sub>); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, TMS) δ (ppm): 161.3, 155.88, 150.8, 145.7, 139.5, 139.1, 136.4, 129.7, 126.4, 115.4, 115.1, 114.2, 112.1, 109.5, 97.4, 45.2, 21.7, 12.8; FT-IR (KBr)  $\nu_{\max}$  = 1712 cm<sup>-1</sup> (C=O lactone), 1621 cm<sup>-1</sup> (C=C Ar); UV–Vis (MeOH)  $\lambda_{\max}$  = 422 nm; HRMS (ESI m/z) Calcd. for  $C_{21}H_{21}N_3O_2$  [M+H]<sup>+</sup> 348.1707, found 348.1725.



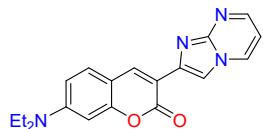
**7-(diethylamino)-3-(imidazo[2,1-b]thiazol-6-yl)-2H-chromen-2-one (4c):**

Yield 78%; dark brown powder; m.p. 173 – 175°C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 8.45 (s, 1H, –CH=C–), 8.30 (s, 1H, –CH=C–), 7.41 (d *J*(H,H) = 4.5 Hz, 1H, Ar), 7.37 (d, *J*(H,H) = 9 Hz, 1, Ar), 6.78 (d, *J*(H,H) = 4.5 Hz, 1H, Ar), 6.61 (dd, *J*(H,H) = 9, 2.5 Hz, 1H, Ar), 6.52 (s, *J*(H,H) = 2.5 Hz, 1H, Ar), 3.43 (q, *J*(H,H) = 7 Hz, 4H, –CH<sub>2</sub>–CH<sub>3</sub>), 1.22 (t, *J*(H,H) = 7 Hz, 6H, –CH<sub>2</sub>–CH<sub>3</sub>); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, TMS) δ (ppm): 161.0, 155.6, 150.5, 15.15, 141.9, 137.89, 129.4, 119.0, 114.4, 112.5, 112.3, 109.5, 109.4, 97.3, 45.1, 12.8; FT-IR (KBr)  $\nu_{\max}$  = 1710 cm<sup>-1</sup> (C=O lactone), 1618 cm<sup>-1</sup> (C=C Ar); UV–Vis (MeOH)  $\lambda_{\max}$  = 415 nm; HRMS (ESI m/z) Calcd. for  $C_{18}H_{17}N_3O_2$  [M+H]<sup>+</sup> 340.1114, found 340.1118.



**7-(diethylamino)-3-(7-methylimidazo[1,2-a]pyrimidin-2-yl)-2H-chromen-2-one (4d):**

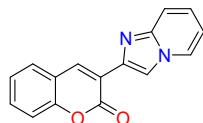
Yield 98%; yellow powder; m.p. 229.4 – 231 °C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 8.72 (s, 1H, –CH=C–), 8.26 (s, 1H, –CH=C–), 8.24 (d, *J*(H,H) = 7 Hz, 1H, Ar), 7.38 (d, *J*(H,H) = 8.5 Hz, 1H, Ar), 6.65 (d, *J*(H,H) = 7 Hz, 1H, Ar), 6.58 (d, *J*(H,H) = 9 Hz, 1H, Ar), 6.48 (s, 1H, CH<sub>3</sub>), 3.40 (q, *J*(H,H) = 7 Hz, 4H, –CH<sub>2</sub>–CH<sub>3</sub>), 2.57 (t, *J*(H,H) = 7 Hz, 6H, –CH<sub>2</sub>–CH<sub>3</sub>); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, TMS) δ (ppm) = 161.0, 160.4, 156.1, 150.9, 148.2, 140.6, 140.2, 132.8, 129.7, 113.4, 109.8, 109.4, 109.2, 97.8, 45.0, 25.1, 12.6; FT-IR (KBr)  $\nu_{\max}$  = 1714 cm<sup>-1</sup> (C=O lactone), 1620 cm<sup>-1</sup> (C=C Ar); UV–Vis (MeOH)  $\lambda_{\max}$  = 433 nm; HRMS (ESI m/z) Calcd. for  $C_{20}H_{20}N_4O_2$  [M+H]<sup>+</sup> 349.1659, found 349.1671.



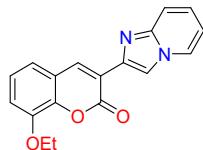
**7-(diethylamino)-3-(imidazo[1,2-a]pyrimidin-2-yl)-2H-chromen-2-one (4e):**

Yield 94%; yellow powder; m.p. 158 – 160°C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 8.78 (s, 1H, –CH=C–), 8.50 (dd, *J*(H,H) = 4, 2 Hz, 1H, Ar), 8.41 (d, *J*(H,H) = 2 Hz, 1H, Ar), 8.40 (s, 1H, –CH=C–), 7.42 (d, *J*(H,H) = 7.59 Hz, 1H, Ar), 6.82 (t, *J*(H,H) = 4, 2.5 Hz, 1H, Ar), 6.662

(dd,  $J$  (H,H) = 9, 2.5 Hz, 1H, Ar1), 6.51 (s,  $J$  (H,H) = 2Hz, 1H, Ar), 3.43 (q,  $J$  (H,H) = 7 Hz, 4H,  $-\text{CH}_2\text{--CH}_3$ ), 1.22 (t,  $J$  (H,H) = 7 Hz, 6H,  $-\text{CH}_2\text{--CH}_3$ ); RMN  $^{13}\text{C}$  (125 MHz,  $\text{CDCl}_3$ , TMS)  $\delta$  (ppm) = 161.2, 156.4, 151.2, 150.40, 148.3, 141.7, 140.9, 133.7, 130.0, 113.2, 110.4, 109.6, 109.4, 108.7, 97.3, 45.2, 12.7, FT-IR (KBr)  $\nu_{\text{max}}$  = 1702  $\text{cm}^{-1}$  (C=O lactone), 1613  $\text{cm}^{-1}$  (C=C Ar); UV–Vis (MeOH)  $\lambda_{\text{max}}$  = 436 nm; HRMS (ESI m/z) Calcd. for  $\text{C}_{19}\text{H}_{18}\text{N}_4\text{O}_2$  [M+H] $^+$  335.1503, found 335.1522.



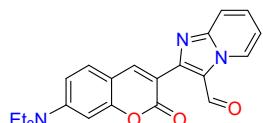
**3-(imidazo[1,2-a]pyridin-2-yl)-2H-chromen-2-one (4f):** Yield 70%; brown powder; m.p. 297 – 300 °C; RMN  $^1\text{H}$  (500 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  (ppm) = 8.76 (s, 1H,  $-\text{CH}=\text{C}-$ ), 8.53 (s, 1H,  $-\text{CH}=\text{C}-$ ), 8.13 (d,  $J$  (H,H) = 7 Hz, 1H, Ar), 7.62 (d,  $J$  (H,H) = 7.5, 1H, Ar), 7.58 (d,  $J$  (H,H) = 9.5 Hz, 1H, Ar), 7.52 (t,  $J$  (H,H) = 7.5 Hz, 1H, Ar), 7.37 (d,  $J$  (H,H) = 8 Hz, 1H, Ar), 7.31 (t,  $J$  (H,H) = 7.5 Hz, 1H, Ar), 7.21 (t,  $J$  (H,H) = 6.5 Hz, 1H, Ar), 6.79 (t,  $J$  (H,H) = 6.5 Hz, 1H, Ar); RMN  $^{13}\text{C}$  (125 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  (ppm) = 160.2, 153.3, 145.4, 138.5, 138.3, 131.63, 128.6, 126.5, 126.0, 124.9, 121.1, 120.0, 117.4, 116.7, 114.2, 112.8; FT-IR (KBr)  $\nu_{\text{max}}$  = 1726  $\text{cm}^{-1}$  (C=O lactone), 1636  $\text{cm}^{-1}$  (C=C Ar); UV–Vis (MeOH)  $\lambda_{\text{max}}$  = 345 nm; HRMS (ESI m/z) Calcd. for  $\text{C}_{16}\text{H}_{10}\text{N}_2\text{O}_2$  [M+H] $^+$  263.0815, found 263.0820.



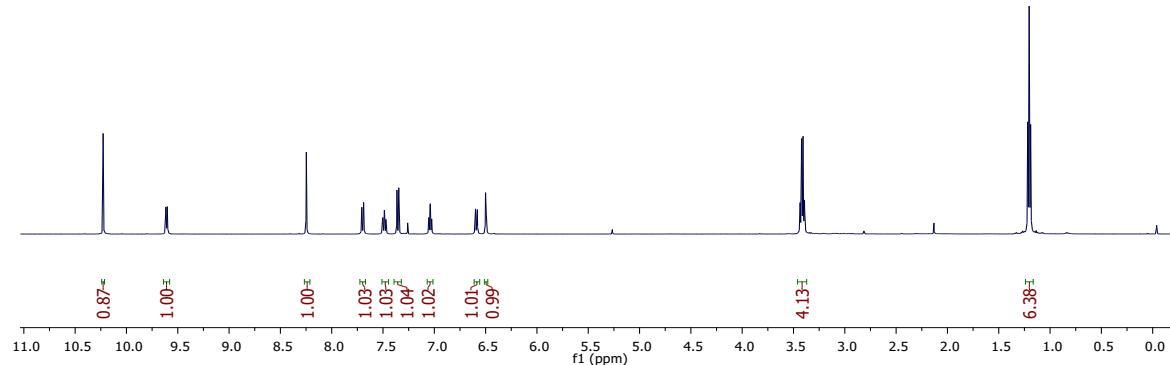
**8-ethoxy-3-(imidazo[1,2-a]pyridin-2-yl)-2H-chromen-2-one (4g):** Yield: 41%, dark brown powder; m.p. = 157 – 159 °C, RMN  $^1\text{H}$  (500 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  (ppm) = 8.75 (s, 1H,  $-\text{CH}=\text{C}-$ ), 8.57 (s, 1H,  $-\text{CH}=\text{C}-$ ), 7.14 (d,  $J$  (H,H) = 6.5 Hz, 1H, Ar), 7.59 (d,  $J$  (H,H) = 6.5 Hz, 1H, Ar), 7.20 (m, 3H, Ar), 7.07 (d, 1H, Ar), 6.78 (t,  $J$  (H,H) = 6.5 Hz, 1H, Ar), 4.21 (q,  $J$  (H,H) = 7 Hz, 2H,  $-\text{CH}_2\text{--CH}_3$ ), 1.52 (t,  $J$  (H,H) = 7 Hz, 3H,  $-\text{CH}_2\text{--CH}_3$ ); RMN  $^{13}\text{C}$  (125 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  (ppm) = 159.65, 146.40, 145.20, 143.01, 138.36, 138.33, 126.28, 125.80, 124.53, 121.09, 120.54, 119.86, 117.21, 114.56, 114.07, 112.62, 65.07, 14.90. FT-IR (KBr)  $\nu_{\text{max}}$  = 1724  $\text{cm}^{-1}$  (C=O lactone); UV–Vis (MeOH)  $\lambda_{\text{max}}$  = 340 nm; HRMS (ESI m/z) Calcd. for  $\text{C}_{18}\text{H}_{15}\text{N}_2\text{O}_3$  [M+H] $^+$  307.1077, found 307.1068.

### 3. General procedure for coumarin-imidazo[1,2-a]pyridine-3-carbaldehyde derivatives 5a-e.

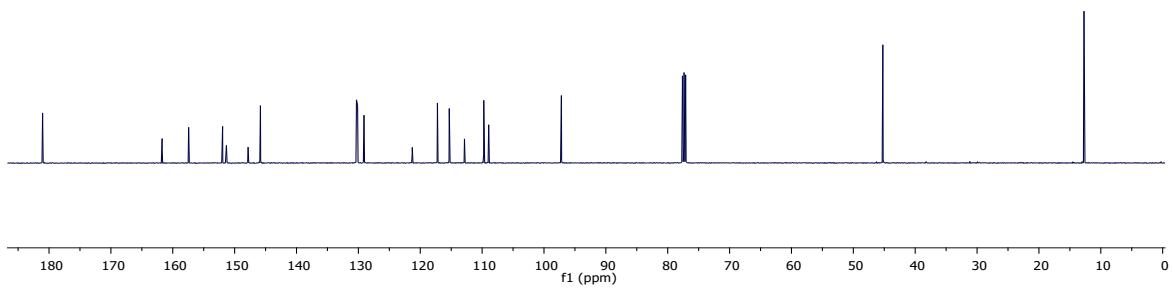
Synthesis of **5a-e** was carried out using the Vilsmeier-Haack formilation.[3] In a 50 mL round bottom flask was added **4a-e** derivatives (1mmol), then was purged with nitrogen, 15 mL of DMF was added and placed in agitation. In a second 50 mL round bottom flask with nitrogen atmosphere DMF was added and placed in an ice bath for 10 min (2.2 mmol).  $\text{POCl}_3$  were slowly added (2.4 mmol) and the mixture was kept in constant agitation in an ice bath for 15 min. Finally, the solution from the first flask was transferred to the second flask through a steel cannula and heated to 60 °C for 1 h. The reaction was treated with 10% NaOH solution and a solid product was formed. The product was filtered and purified using a chromatographic column with an Hex/AcOEt 70:30 elution system.



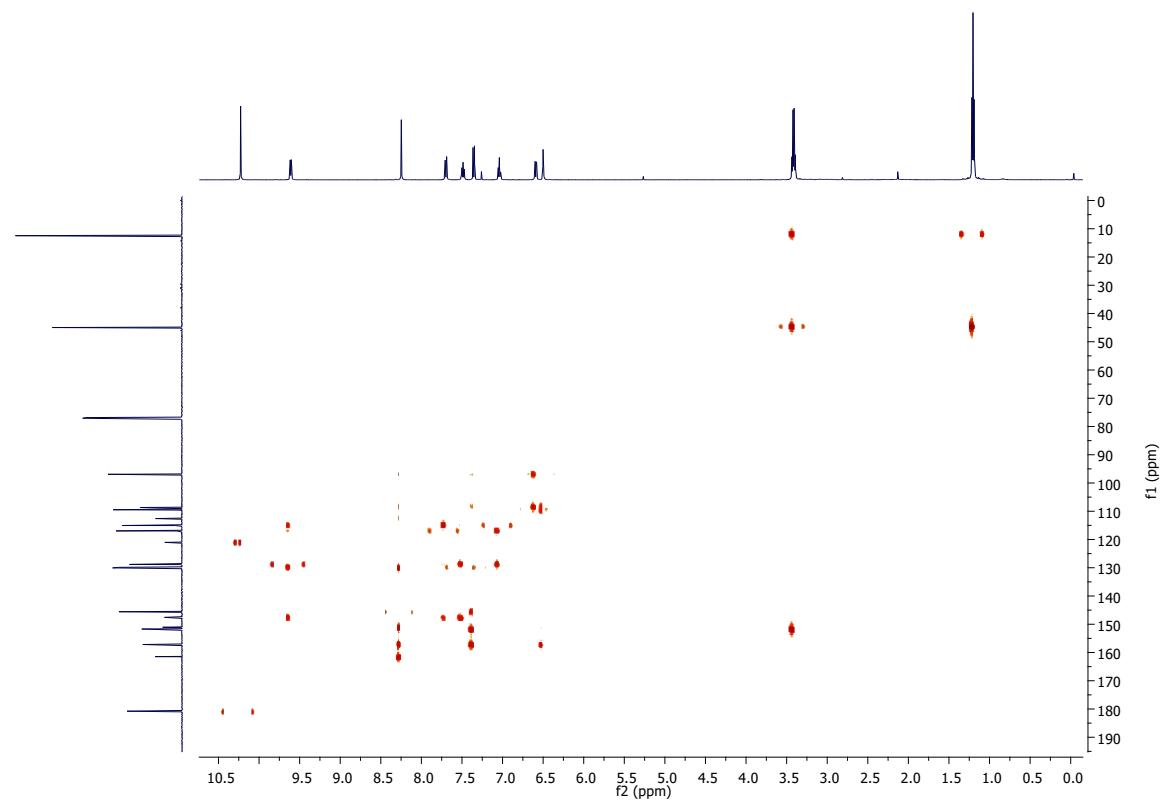
**2-(7-(diethylamino)-2-oxo-2H-chromen-3-yl)imidazo[1,2-a]pyridine-3-carbaldehyde (**5a**):** Yield 94%; yellow powder; m.p. 228 – 229.6 °C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 10.22 (s, 1H, –COH), 9.61 (d, *J* (H,H) = 6.5 Hz, 1H, Ar), 8.24 (s, 1H, –CH=C–), 7.70 (d, *J* (H,H) = 9 Hz, 1H, Ar), 7.50 (t, *J* (H,H) = 8.5–1.5 Hz, 1H, Ar), 7.36 (d, *J* (H,H) = 9 Hz, 1H, Ar), 7.05 (t, *J* (H,H) = 6.1 Hz, 1H, Ar), 6.60 (d, *J* (H,H) = 9, 2.5 Hz, 1H, Ar), 6.50 (s, *J* (H,H) = 2.5, 1H, –CH=C–), 3.43 (q, *J* (H,H) = 7.5 Hz, 4H, –CH<sub>2</sub>–CH<sub>3</sub>), 1.21 (t, *J* (H,H) = 7.5 Hz, 6H, CH<sub>3</sub>); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 181.5, 177.2, 171.7, 171.1, 167.6, 165.6, 150.1, 149.9, 148.8, 141.0, 137.0, 135.1, 132.6, 129.5, 128.7, 97.1, 45.2, 12.7; FT-IR (KBr) ν<sub>max</sub> = 2736 cm<sup>-1</sup> (H–C=O), 1698 cm<sup>-1</sup> (C=O lactone), 1647 cm<sup>-1</sup> (H–C=O); UV–Vis (MeOH) λ<sub>max</sub> = 422 nm; HRMS (ESI m/z) Calcd. for C<sub>20</sub>H<sub>18</sub>N<sub>4</sub>O<sub>3</sub> [M+H]<sup>+</sup> 362.1499, found 362.1495.



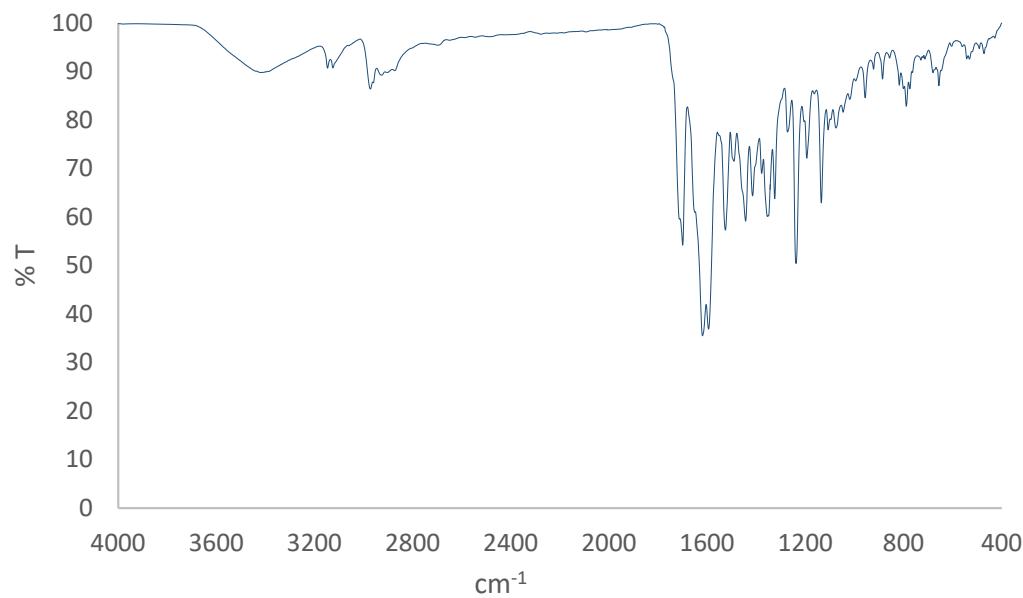
**Figure SI-1.** <sup>1</sup>H NMR spectra of **5a** on CDCl<sub>3</sub> 500 MHz



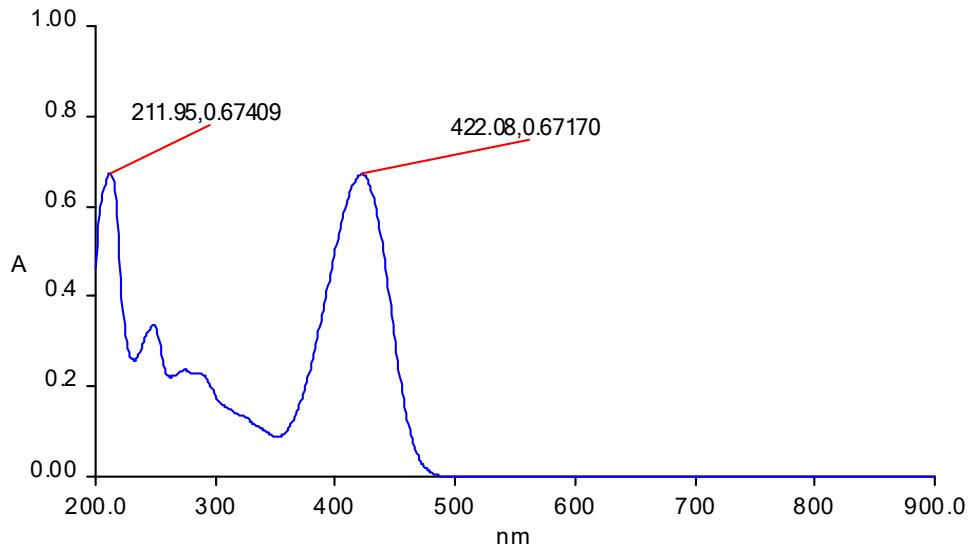
**Figure SI-2.**  $^{13}\text{C}$  NMR spectra of **5a** on  $\text{CDCl}_3$  125 MHz.



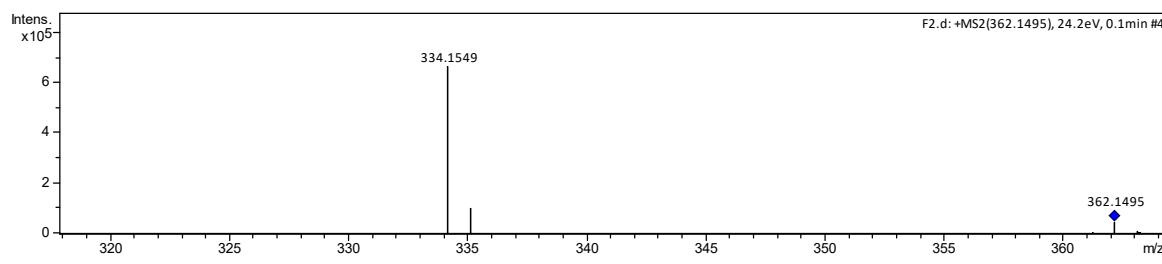
**Figure SI-3.** HMBC NMR spectra of **5a** on  $\text{CDCl}_3$



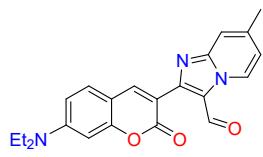
**Figure SI-4.** IR spectra for **5a** in KBr.



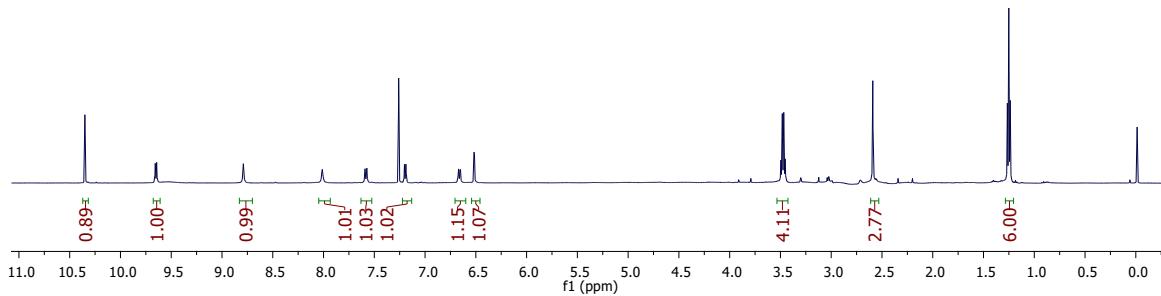
**Figure SI-5.** UV-Vis spectra of **5a** in MeOH.



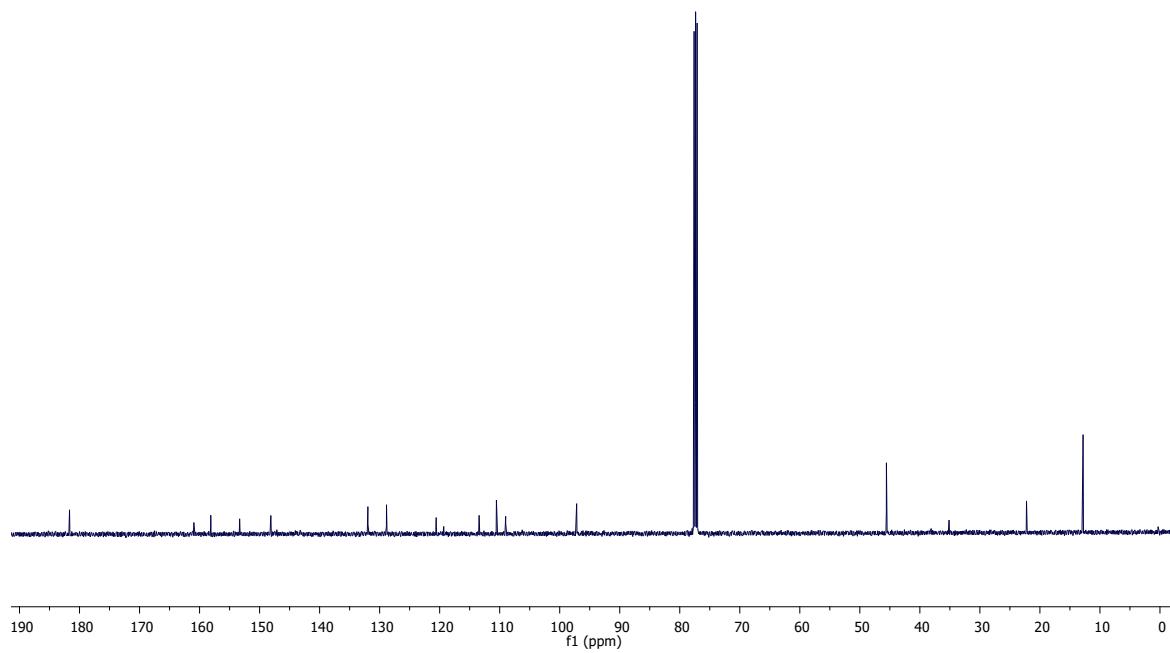
**Figure SI-6.** ESI-MS chromatogram of **5a**.



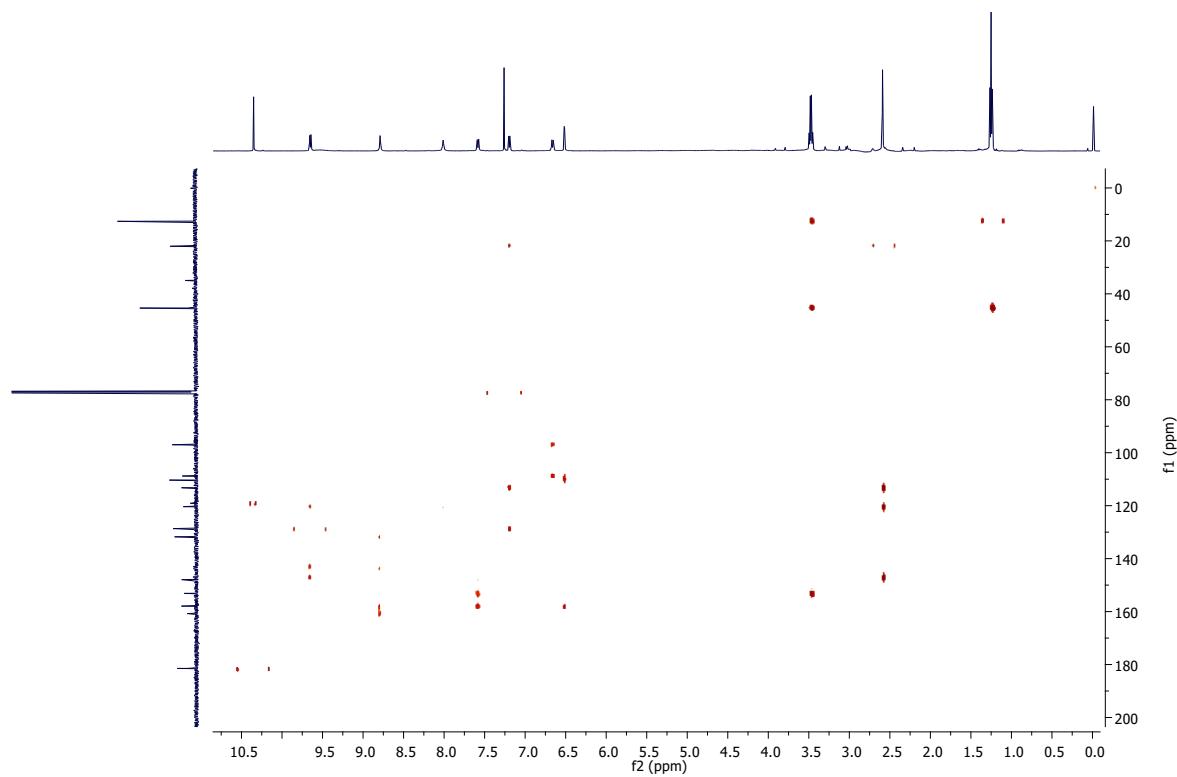
**2-(7-(diethylamino)-2-oxo-2H-chromen-3-yl)-7-methylimidazo[1,2-a]pyridine-3-carbaldehyde (**5b**):** Yield 97%; yellow powder; m.p. 174 – 176°C; RNM  $^1\text{H}$  (500 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  (ppm) = 10.35 (s, 1H,  $-\text{COH}$ ), 9.65 (d,  $J(\text{H},\text{H})$  = 7 Hz, 1H, Ar), 8.79 (s, 1H,  $-\text{CH}=\text{C}-$ ), 8.01 (s, 1H, Ar), 7.50 (d,  $J(\text{H},\text{H})$  = 9 Hz, 1H, Ar), 7.20 (d,  $J(\text{H},\text{H})$  = 7 Hz, 1H, Ar), 6.67 (dd,  $J(\text{H},\text{H})$  = 9, 2.5 Hz, 1H, Ar), 6.51 (sd,  $J(\text{H},\text{H})$  = 2 Hz, 1H, Ar), 3.49 (q,  $J(\text{H},\text{H})$  = 7 Hz, 4H,  $-\text{CH}_2-\text{CH}_3$ ), 2.59 (s, 3H,  $\text{CH}_3$ ), 1.26 (t,  $J(\text{H},\text{H})$  = 7 Hz, 6H,  $\text{CH}_3$ ); RMN  $^{13}\text{C}$  (125 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  (ppm) = 181.6, 160.9, 158.1, 153.3, 148.1, 131.9, 128.8, 120.5, 119.3, 113.4, 110.5, 109.0, 97.1, 45.5, 29.2, 12.8. FT-IR (KBr)  $\nu_{\text{max}}$  = 2687  $\text{cm}^{-1}$  ( $\text{H}-\text{C}=\text{O}$ ), 1716  $\text{cm}^{-1}$  (C=O lactone), 1600  $\text{cm}^{-1}$  ( $-\text{O}-\text{C}=\text{O}$ ); UV–Vis (MeOH)  $\lambda_{\text{max}}$  = 422 nm.



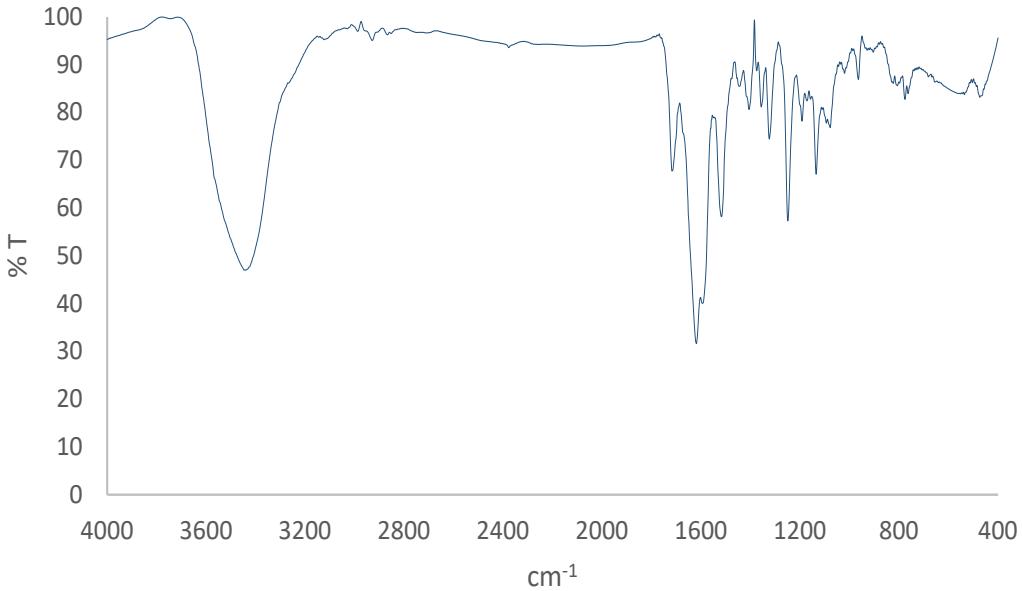
**Figure SI-7.**  $^1\text{H}$  NMR spectra of **5b** on  $\text{CDCl}_3$  500 MHz



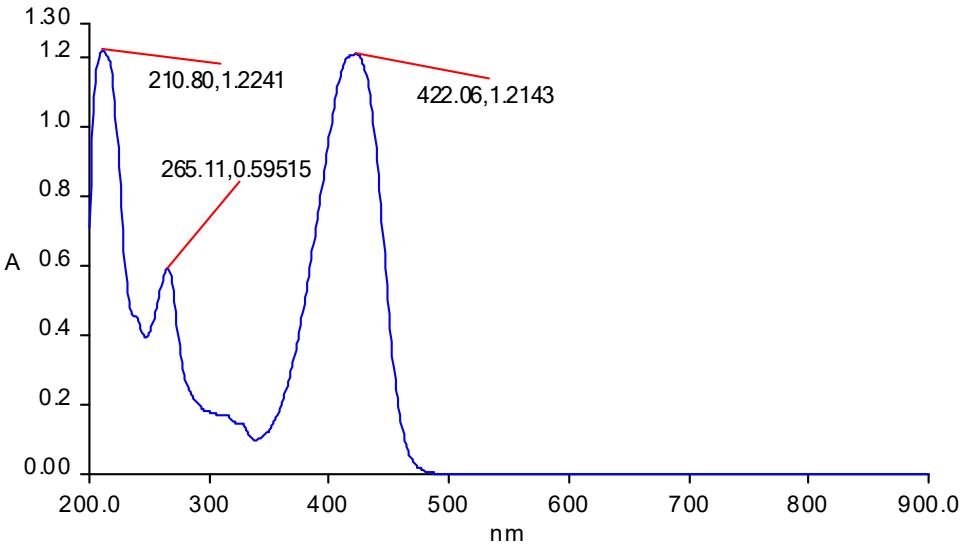
**Figure SI-8.**  $^{13}\text{C}$  NMR spectra of **5b** on  $\text{CDCl}_3$  125 MHz.



**Figure SI-9.** HMBC NMR spectra of **5b** on  $\text{CDCl}_3$

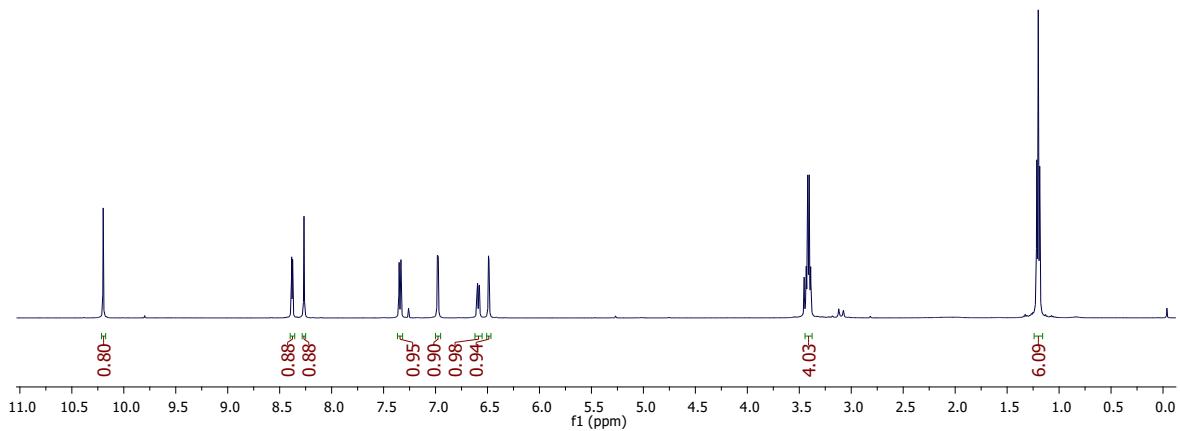


**Figure SI-10.** IR spectra for **5b** in KBr.

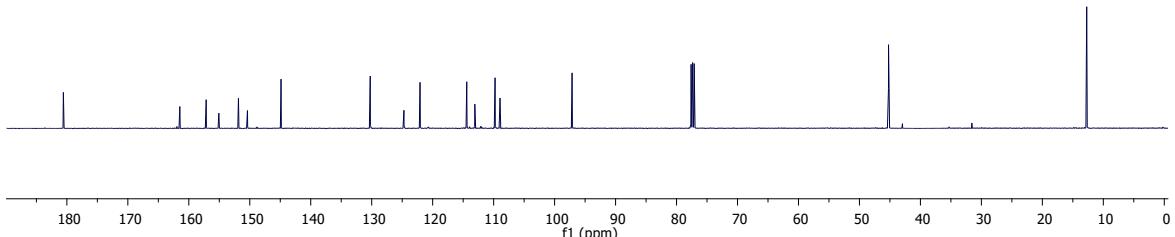


**Figure SI-11.** UV-Vis spectra of **5b** in MeOH.

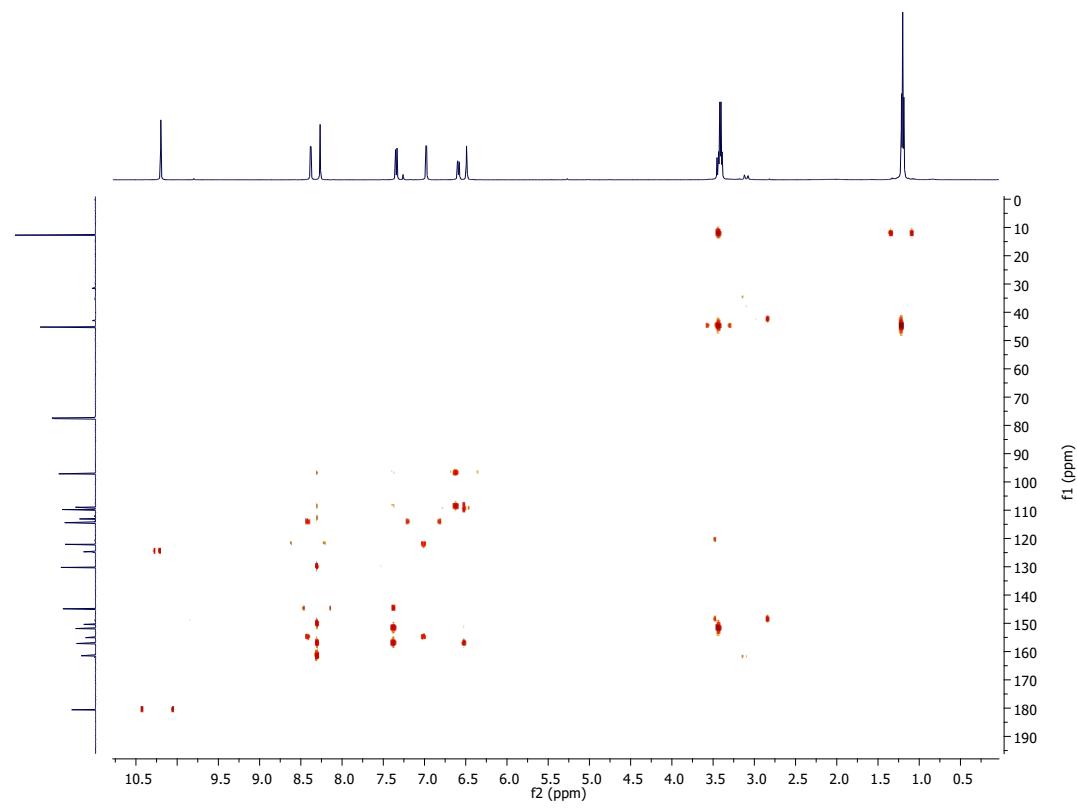
**6-(7-(diethylamino)-2-oxo-2H-chromen-3-yl)imidazo[2,1-b]thiazole-5-carbaldehyde (**5c**):** Yield 60%; brown powder; m.p. 194 – 196°C; <sup>1</sup>H RMN (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 10.20 (s, 1H, –COH), 8.38 (d, *J* (H,H) = 4.5 Hz, 1H, –CH=C–), 8.26 (s, 1H, –CH=C–), 7.35 (d, *J* (H,H) = 9 Hz, 1H, Ar), 6.98 (d, *J* (H,H) = 4.5 Hz, 1H, –CH=C–), 6.60 (dd, *J* (H,H) = 9, 2 Hz, 1H, Ar), 6.49 (sd, *J* = 2 Hz, 1H, Ar), 3.43 (q, *J* (H,H) = 7 Hz, 4H, –CH<sub>2</sub>–CH<sub>3</sub>), 1.21 (t, *J* (H,H) = 7 Hz, 6H, CH<sub>3</sub>); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 180.5, 161.4, 157.1, 155.0, 151.8, 150.3, 144.8, 130.2, 124.7, 122.0, 114.4, 113.0, 109.7, 108.9, 97.1, 45.2, 12.7; FT-IR (KBr) ν<sub>max</sub> = 2743 cm<sup>-1</sup> (H–C=O), 1699 cm<sup>-1</sup> (C=O lactone), 1619 cm<sup>-1</sup> (H–C=O); UV-Vis (MeOH) λ<sub>max</sub> = 425 nm; HRMS (ESI m/z) Calcd. for C<sub>19</sub>H<sub>17</sub>N<sub>3</sub>O<sub>3</sub>S [M+H]<sup>+</sup> 368.1063, found 368.1062.



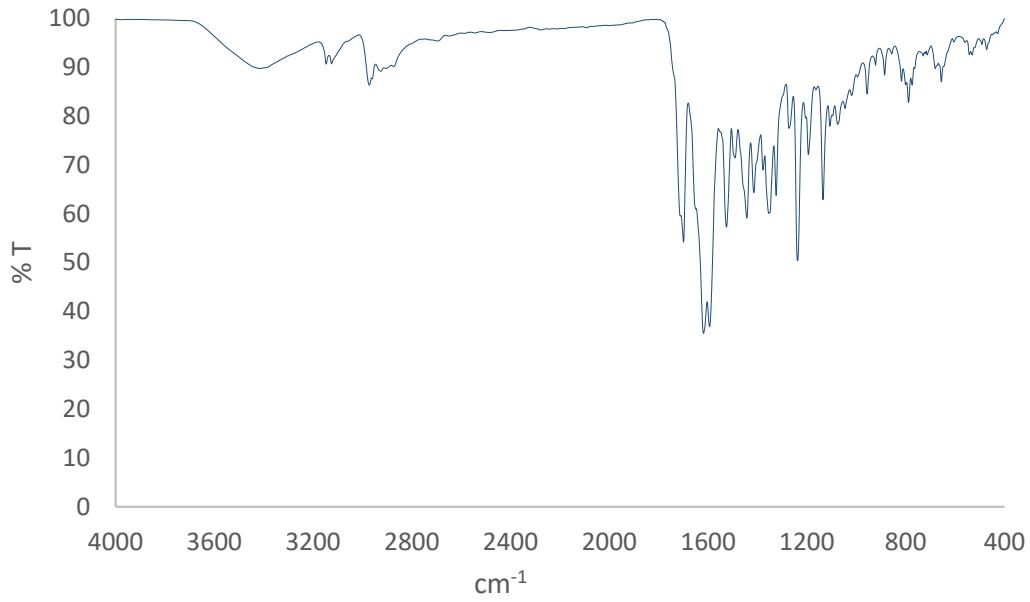
**Figure SI-12.** <sup>1</sup>H NMR spectra of **5c** on  $\text{CDCl}_3$  500 MHz



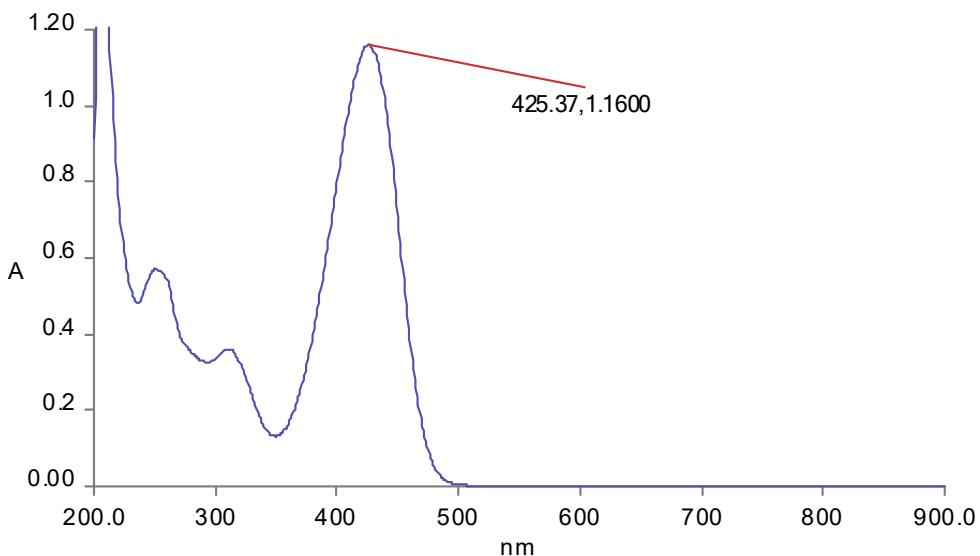
**Figure SI-13.** <sup>13</sup>C NMR spectra of **5b** on  $\text{CDCl}_3$  125 MHz.



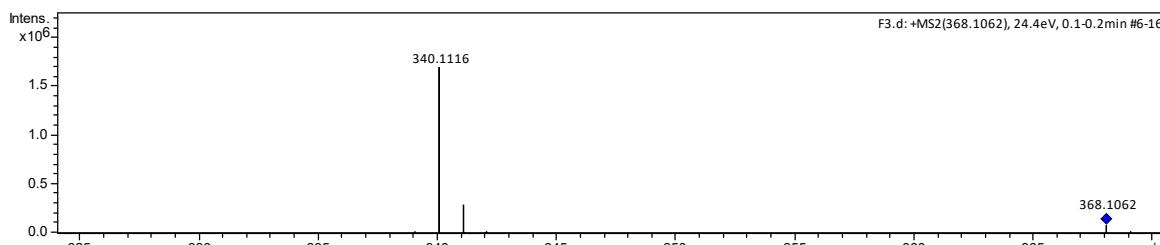
**Figure SI-14.** HMBC NMR spectra of **5c** on  $\text{CDCl}_3$



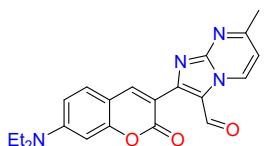
**Figure SI-15.** IR spectra for **5c** in KBr.



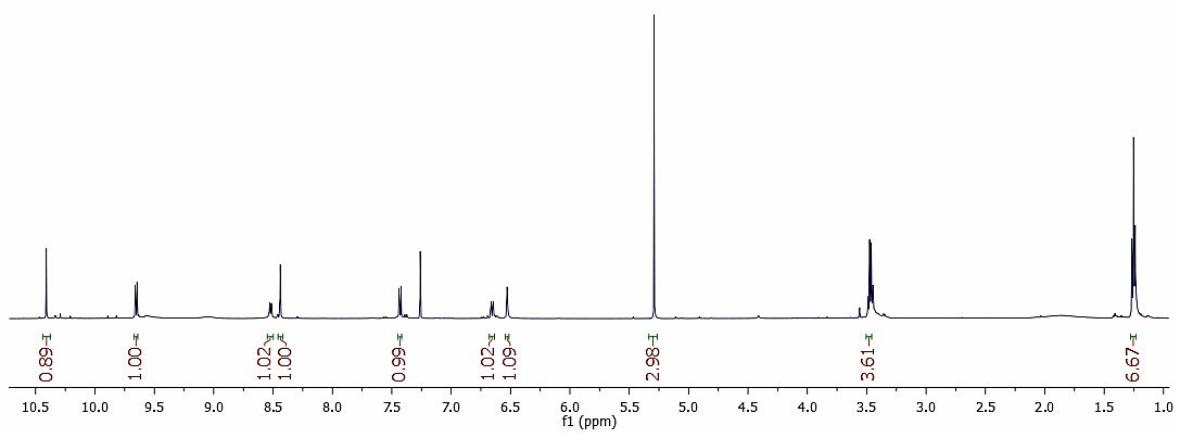
**Figure SI-16.** UV-Vis spectra of **5c** in MeOH.



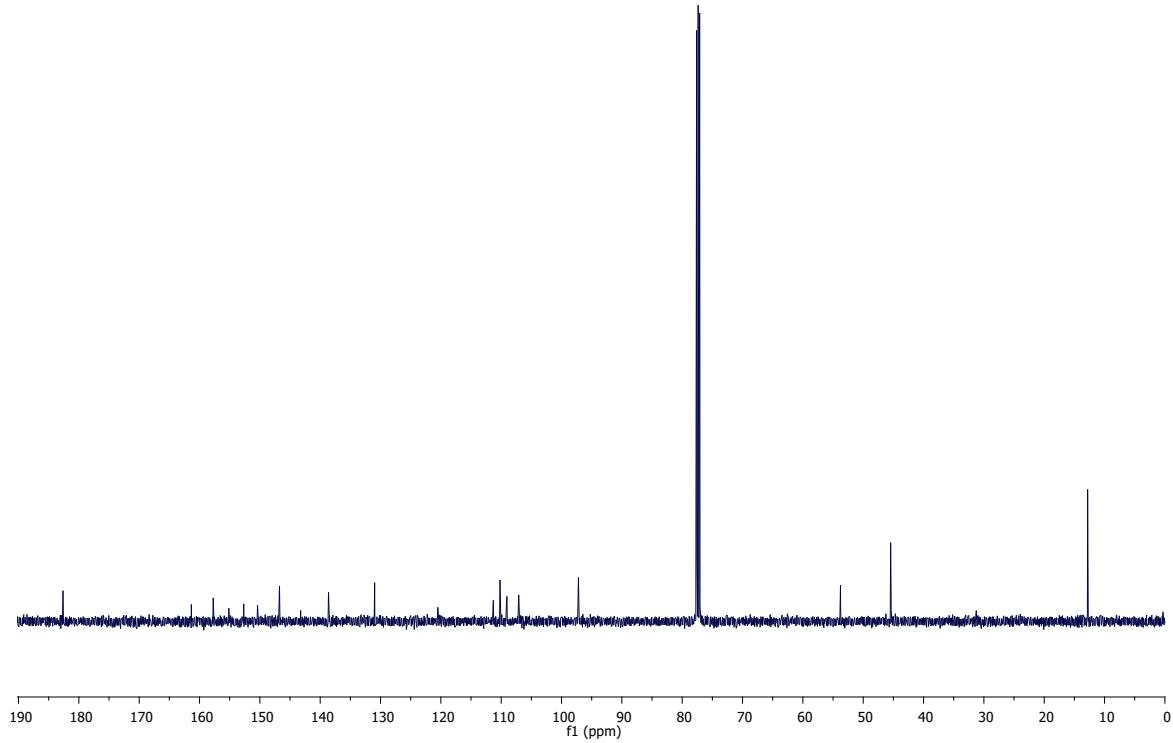
**Figure SI-17.** ESI-MS chromatogram of **5c**.



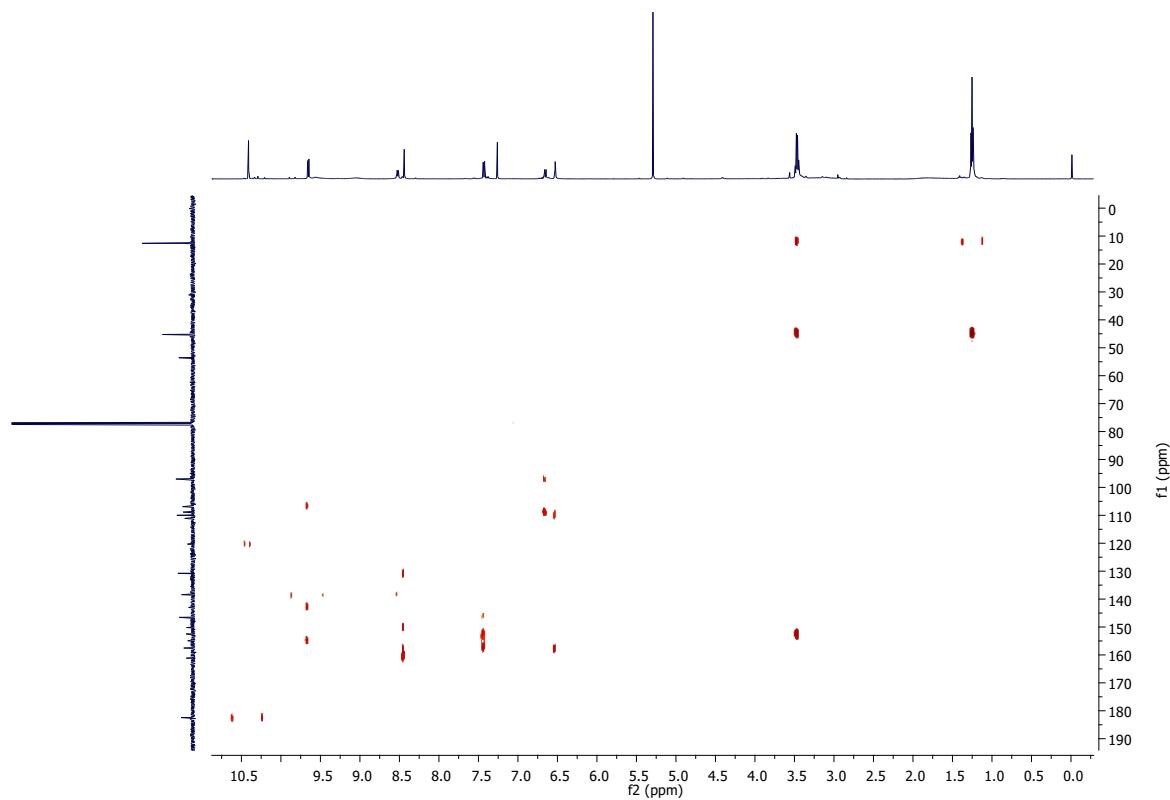
**2-(7-(diethylamino)-2-oxo-2H-chromen-3-yl)-7-methylimidazo[1,2-a]pyrimidine-3-carbaldehyde (**5d**):** Yield 76%; orange powder; m.p. 778-779.3°C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 10.41 (s, 1H, -COH), 9.66 (d, *J* (H,H) = 7.5 Hz, 1H, Ar), 8.52 (d, *J* (H,H) = 7.5 Hz, 1H, Ar), 8.43 (s, 1H, -CH=C-), 7.44 (d, *J* (H,H) = 9 Hz, 1H, Ar), 6.66 (dd, *J* (H,H) = 9, 2.5 Hz, 1H, Ar), 6.52 (sd, *J* = 2.5 Hz, 1H, Ar), 3.49 (q, *J* (H,H) = 7.5 Hz, 4H, -CH<sub>2</sub>-CH<sub>3</sub>), 2.16 (s, 3H, CH<sub>3</sub>), 1.26 (t, *J* (H,H) = 7.5 Hz, 6H, CH<sub>2</sub>); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 182.6, 161.3, 161.3, 157.7, 152.6, 146.7, 138.6, 130.9, 111.3, 110.2, 109.0, 107.1, 97.2, 53.7, 45.4, 19.8; FT-IR (KBr) ν<sub>max</sub> = 1711 cm<sup>-1</sup> (C=O lactone), 1619 cm<sup>-1</sup> (H-C=O); UV-Vis (MeOH) λ<sub>max</sub> = 431 nm; HRMS (ESI m/z) Calcd. for C<sub>21</sub>H<sub>20</sub>N<sub>4</sub>O<sub>3</sub> [M+H]<sup>+</sup> 376.1635, found 376.1638.



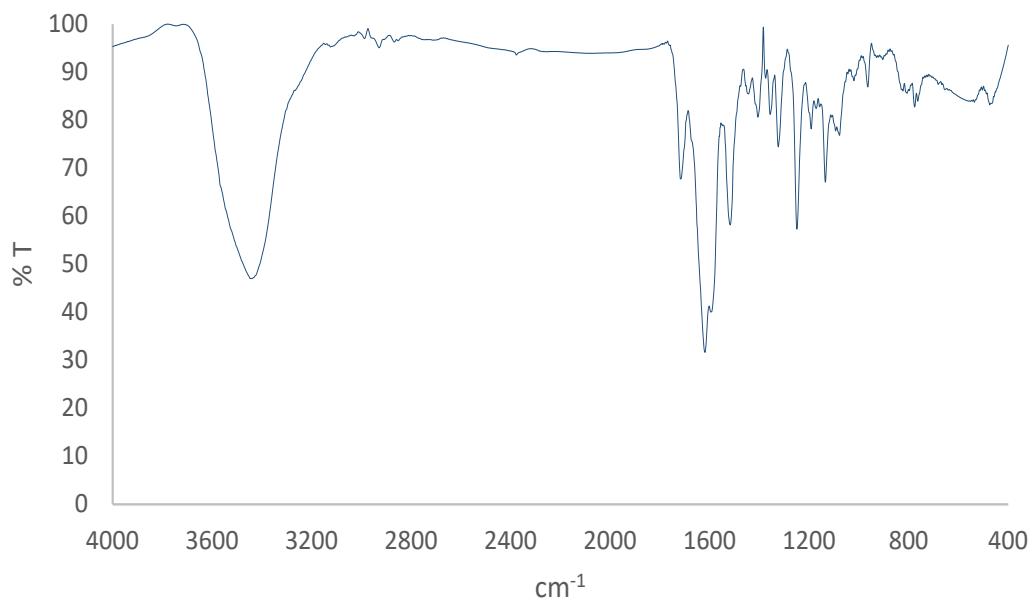
**Figure SI-18.** <sup>1</sup>H NMR spectra of **5d** on  $\text{CDCl}_3$  500 MHz



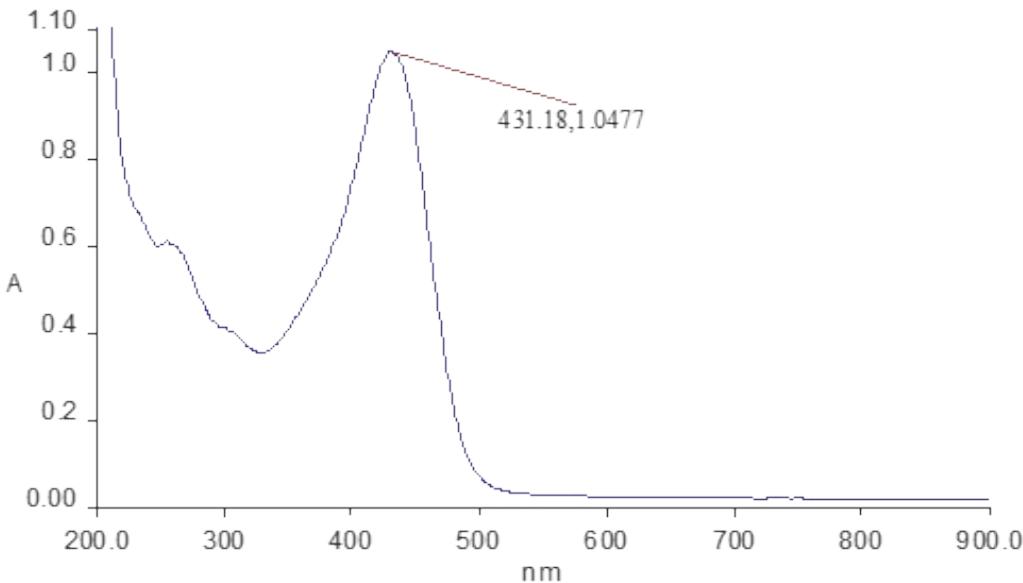
**Figure SI-19.** <sup>13</sup>C NMR spectra of **5d** on  $\text{CDCl}_3$  125 MHz.



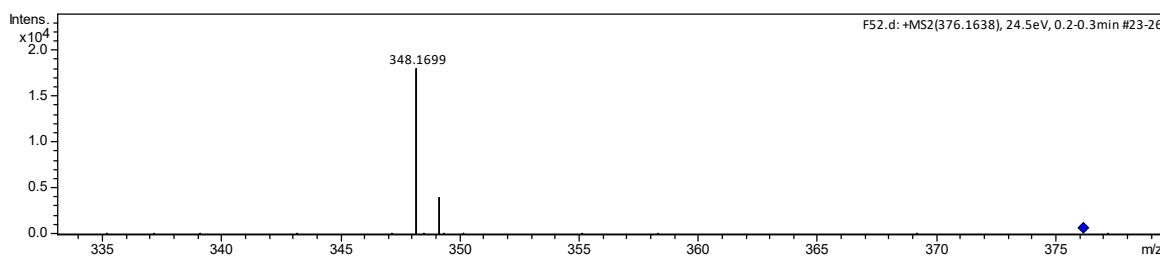
**Figure SI-20.** HMBC NMR spectra of **5d** on  $\text{CDCl}_3$



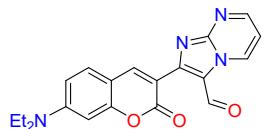
**Figure SI-21.** IR spectra for **5b** in KBr.



**Figure SI-22.** UV-Vis spectra of **5d** in MeOH.



**Figure SI-23.** ESI-MS chromatogram of **5e**.



**2-(7-(diethylamino)-2-oxo-2H-chromen-3-yl)imidazo[1,2-a]pyrimidine-3-carbaldehyde (**5e**):** Yield 63%; orange powder; m.p. 263 – 264.5°C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 10.43 (s, 1H, -COH), 9.93 (dd, *J*(H,H) = 7, 2 Hz, 1H, Ar), 8.76 (d, *J*(H,H) = 2 Hz, 1H, Ar), 8.57 (s, 1H, -CH=C-), 7.43 (d, *J*(H,H) = 9 Hz, 1H, Ar), 7.13 (m, 1H, Ar), 6.65 (dd, *J*(H,H) = 9, 2.5 Hz, 1H, Ar), 6.54 (sd, *J*(H,H) = 2.5 Hz, 1H, Ar), 3.48 (q, *J*(H,H) = 7 Hz, 4H, -CH<sub>2</sub>-CH<sub>3</sub>), 1.26 (t, *J*(H,H) = 7 Hz, 6H, CH<sub>3</sub>); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 182.8, 161.7, 157.7, 154.2, 152.3, 151.7, 150.4, 147.0, 136.9, 130.8, 119.3, 112.3, 111.3, 110.1, 109.2, 97.2, 45.3, 12.8. FT-IR (KBr) ν<sub>max</sub> = 2700 cm<sup>-1</sup> (H-C=O), 1714 cm<sup>-1</sup> (C=O lactone), 1618 cm<sup>-1</sup> (H-C=O); UV-Vis (MeOH) λ<sub>max</sub> = 431 nm.

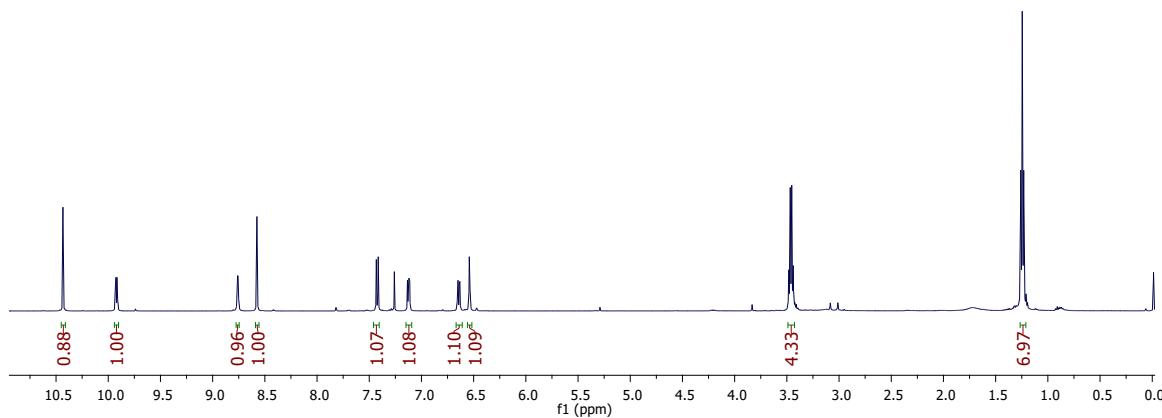


Figure SI-24.  $^1\text{H}$  NMR spectra of **5e** on  $\text{CDCl}_3$  500 MHz

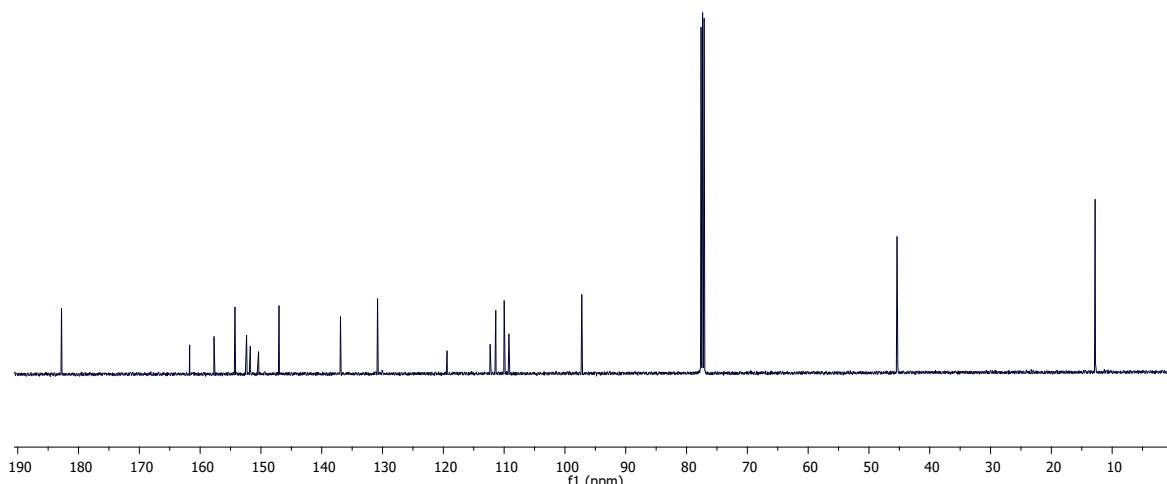
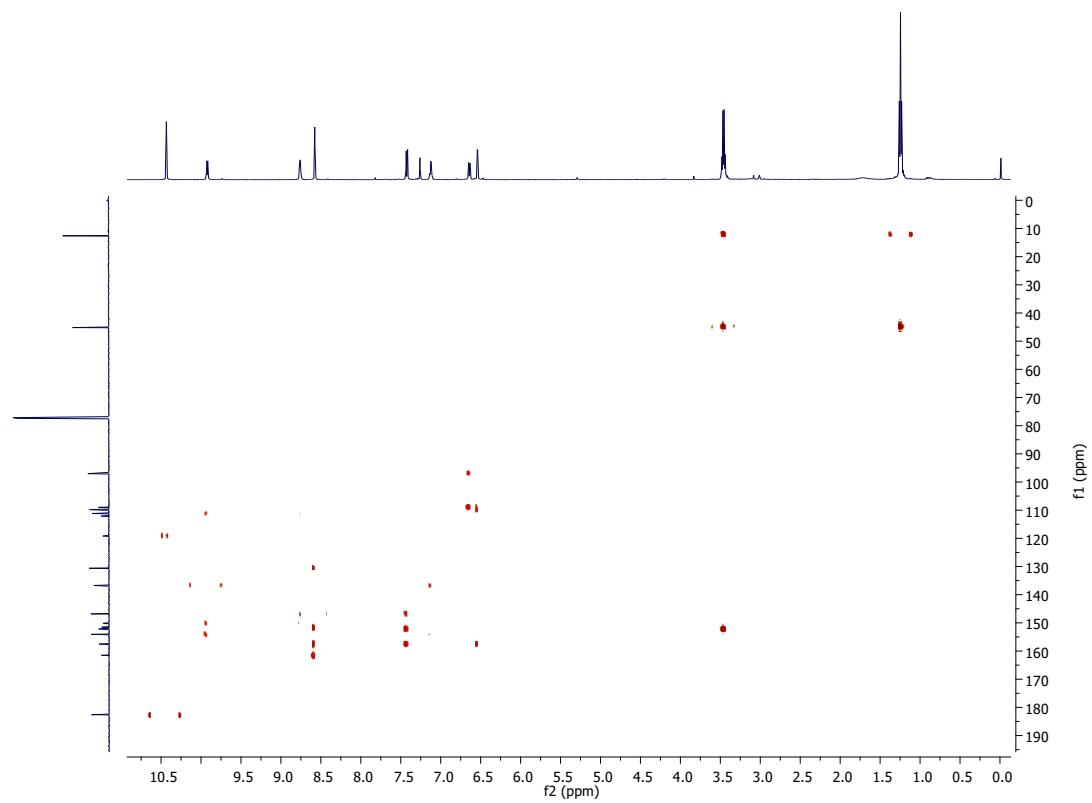
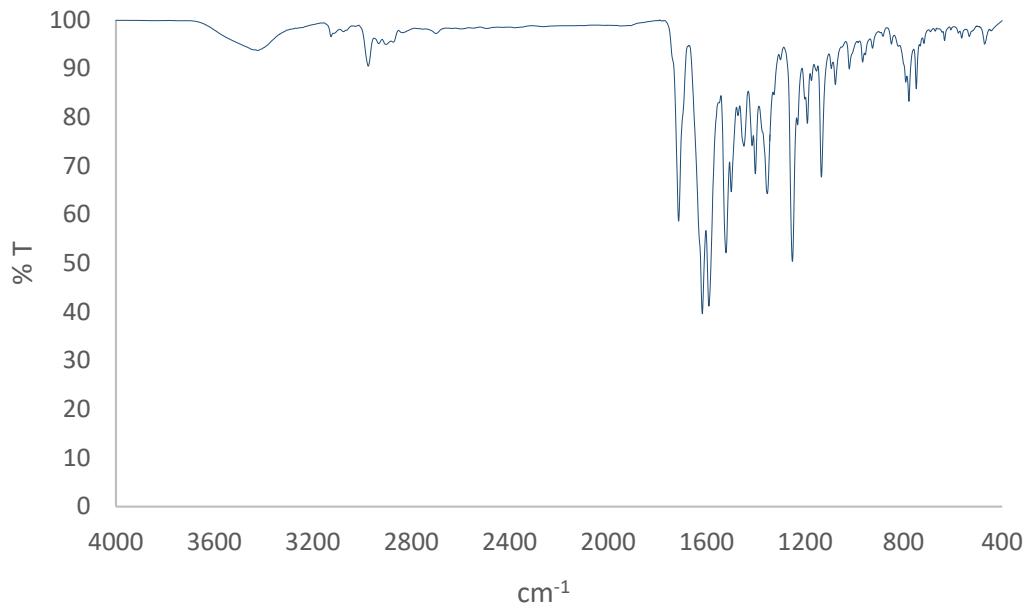


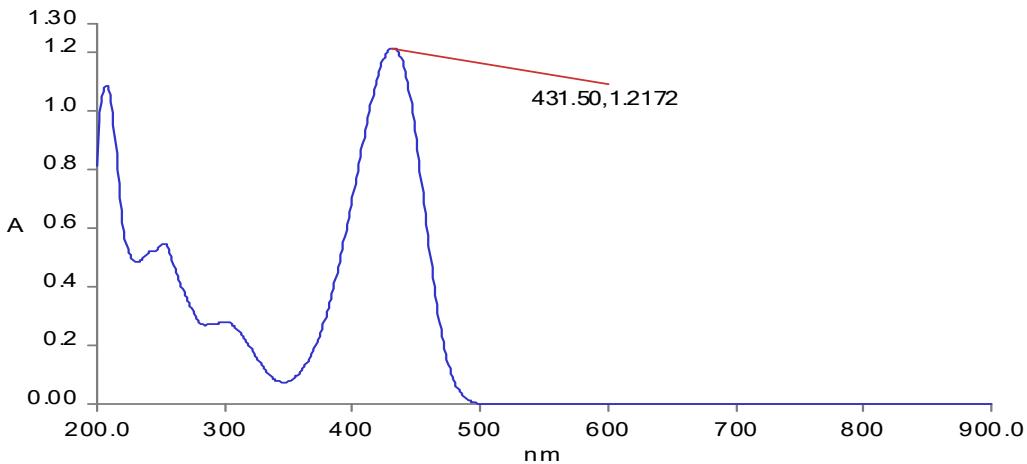
Figure SI-25.  $^{13}\text{C}$  NMR spectra of **5e** on  $\text{CDCl}_3$  125 MHz.



**Figure SI-26.** HMBC NMR spectra of **5e** on  $\text{CDCl}_3$



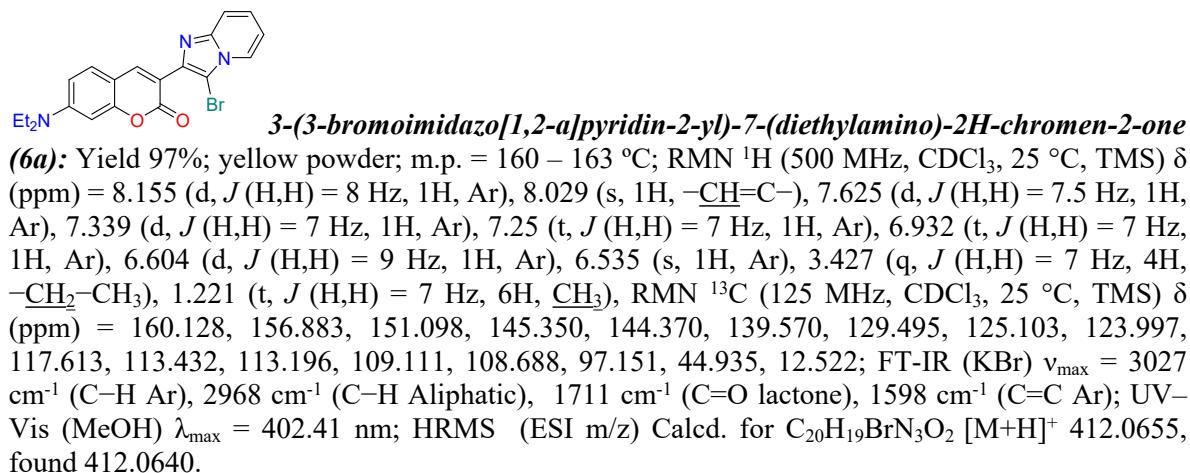
**Figure SI-27.** IR spectra for **5e** in  $\text{KBr}$ .



**Figure SI-28.** UV-Vis spectra of **5e** in MeOH.

#### 4. General procedure for 3-bromo-2-(cumarin-3-il)imidazo[1,2-a]pyridine derivatives **6a-h**

In a 15 mL round bottom flask, **4a-h** derivatives (0.6 mmol) and acetic acid was added (4 mL), the solution was maintained in agitation and heating at 50 °C for 5 minutes. Later, Br<sub>2</sub> were added (0.7 mmol) and a brown precipitate was generated. The solid was filtered, dissolved in dichloromethane and washed with a saturated solution of NaHCO<sub>3</sub>. The organic phase was concentrated and purified by chromatographic column in Hex/AcOEt 80:20 elution system.



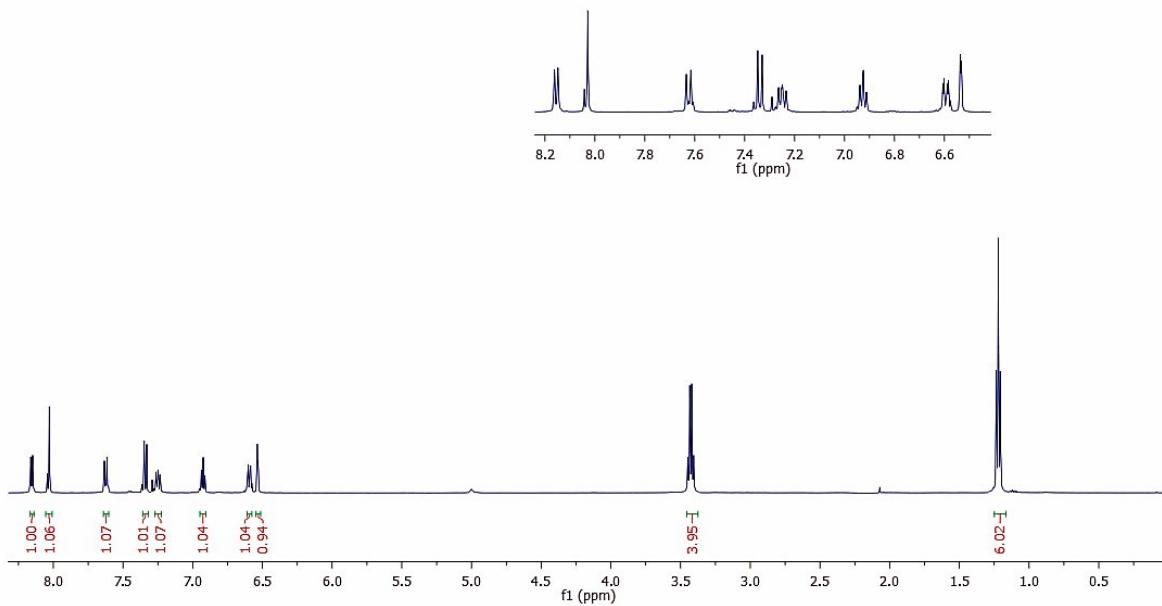


Figure SI-29.  $^1\text{H}$  NMR spectra of **6a** on  $\text{CDCl}_3$  500 MHz

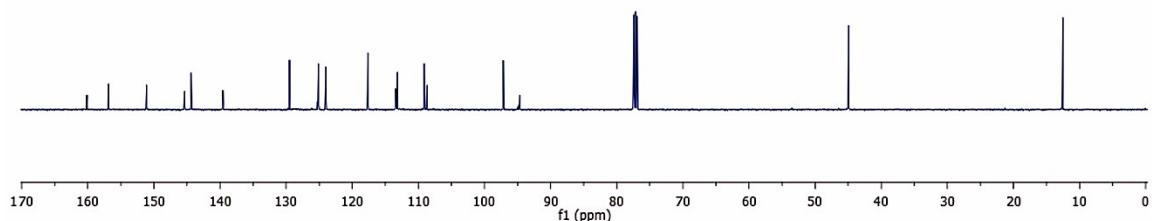


Figure SI-30.  $^{13}\text{C}$  NMR spectra of **6a** on  $\text{CDCl}_3$  125 MHz.

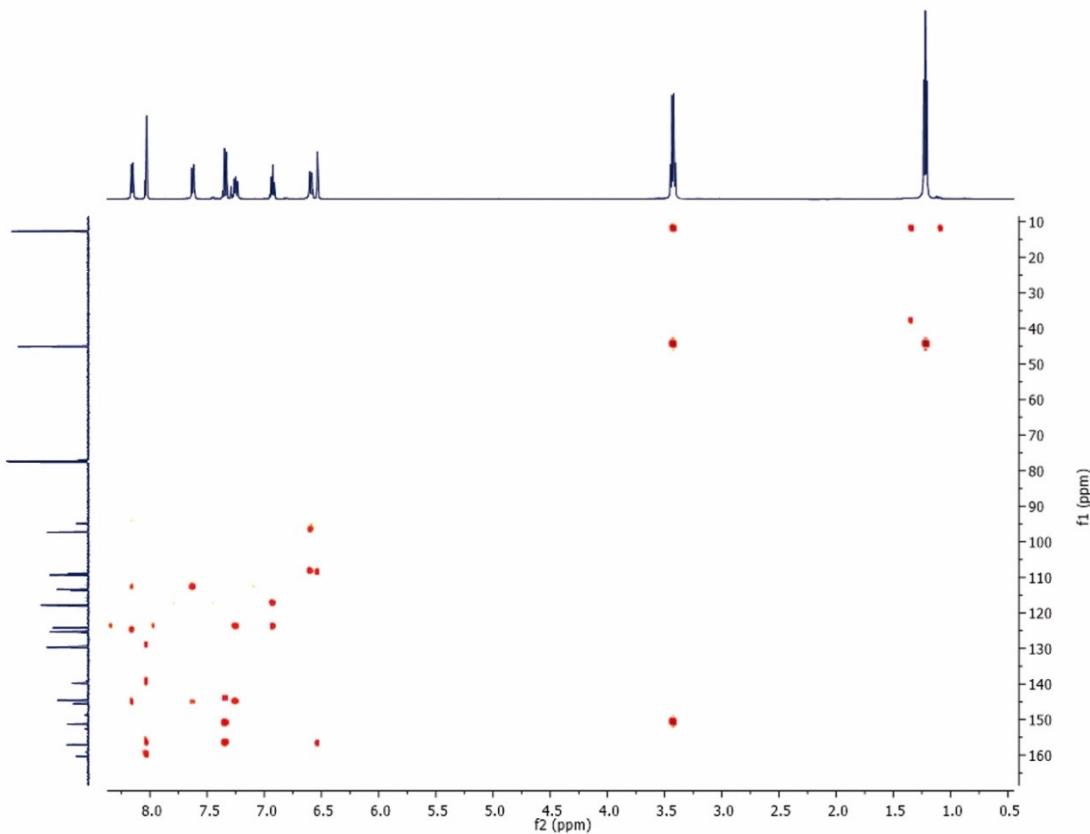


Figure SI-31. HMBC NMR spectra of **6a** on  $\text{CDCl}_3$

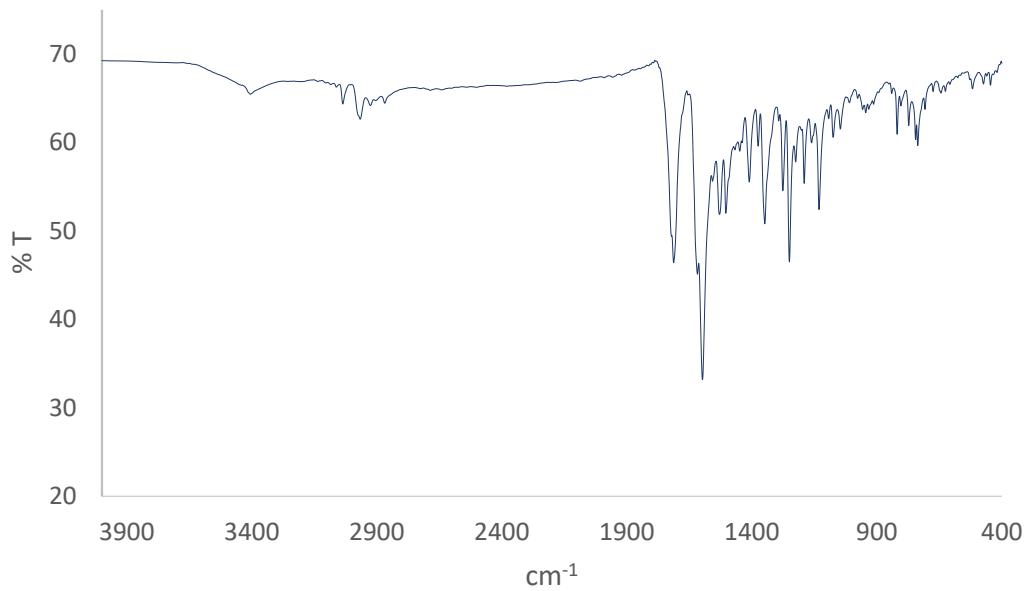
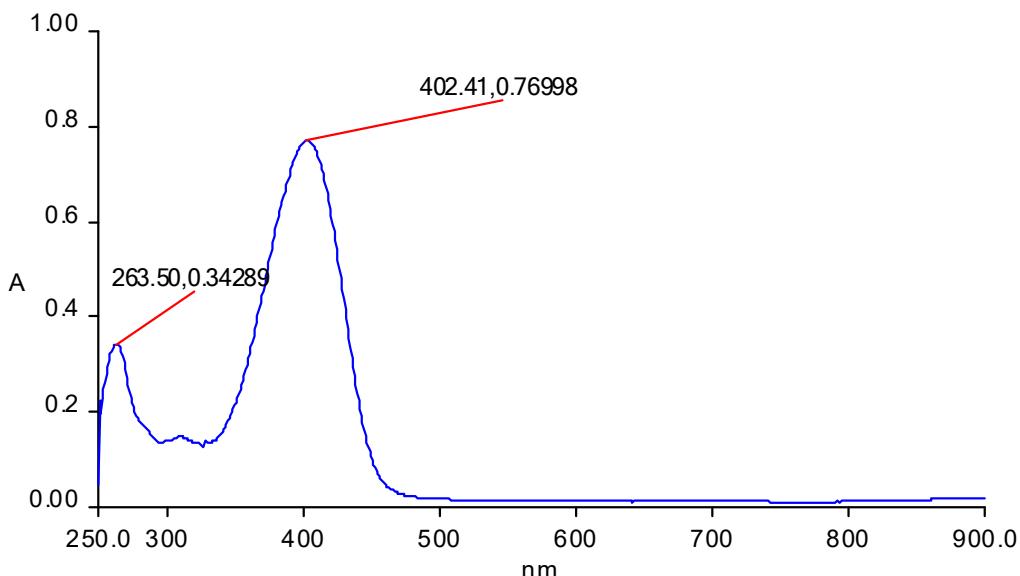
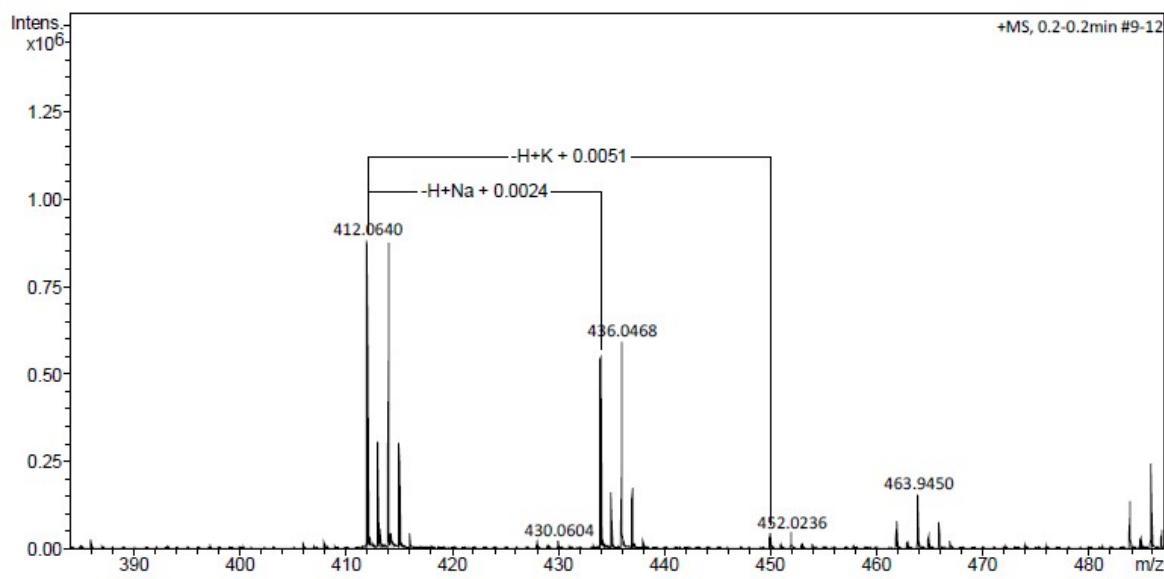


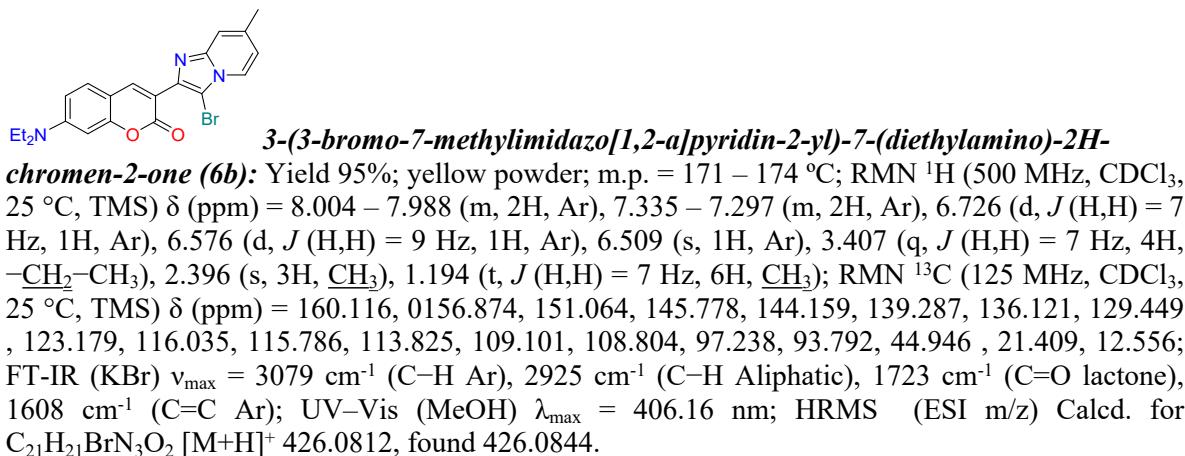
Figure SI-32. IR spectra for **6a** in KBr.

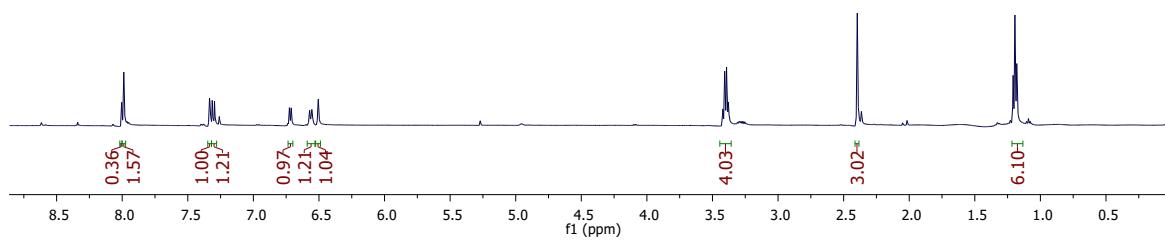


**Figure SI-33.** UV-Vis spectra of **6a** in MeOH.

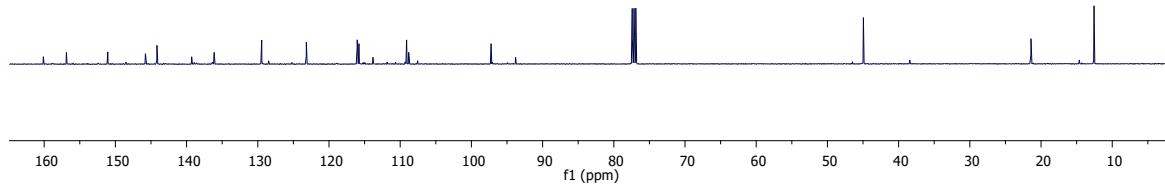


**Figure SI-34.** ESI-MS chromatogram of **6a**.

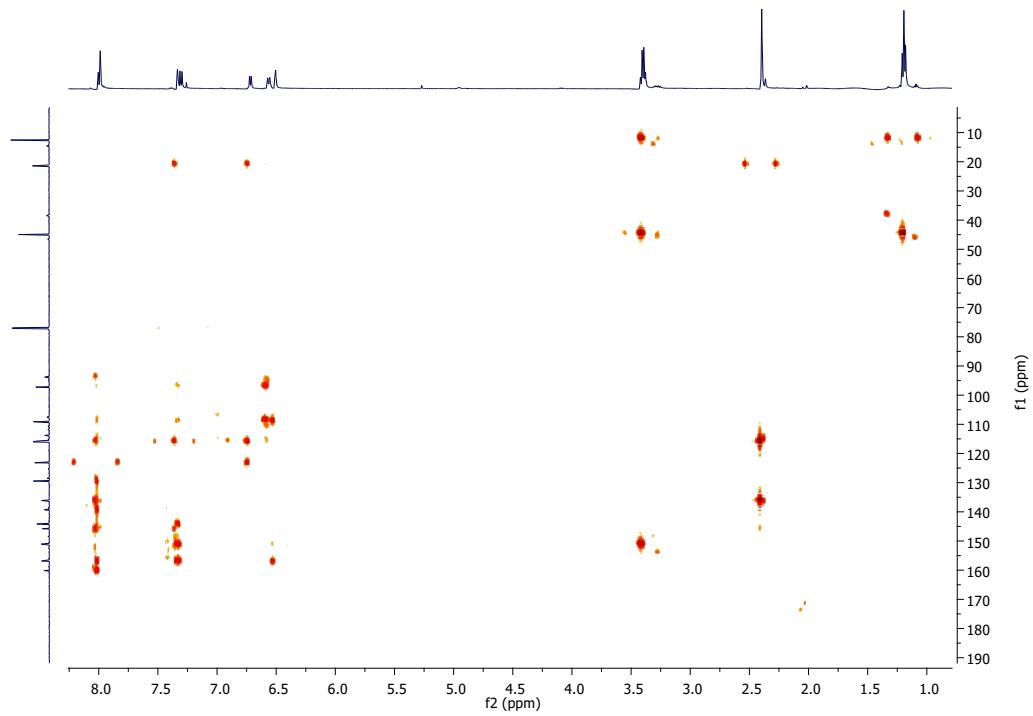




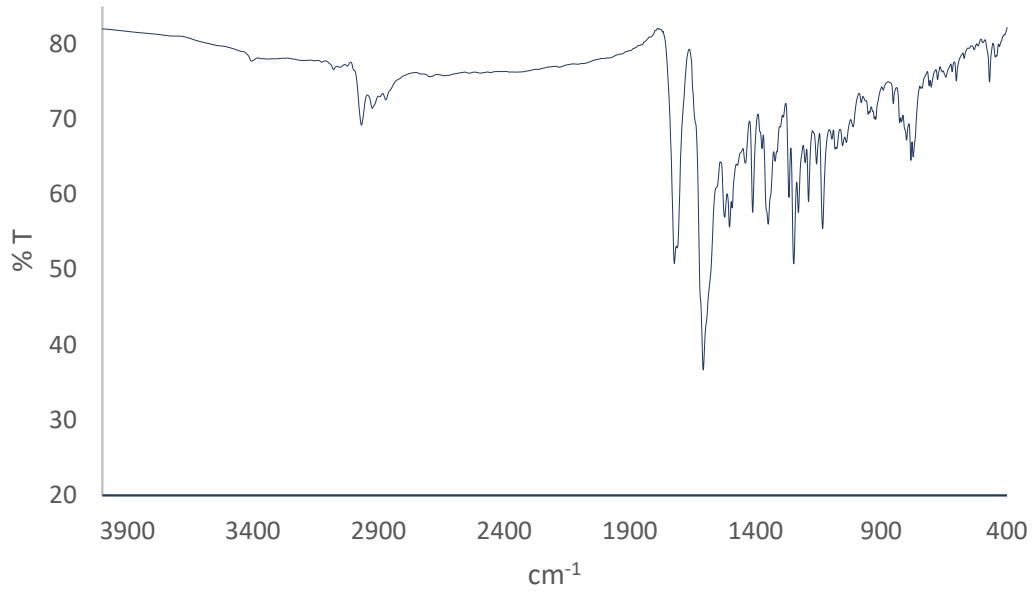
**Figure SI-35.** <sup>1</sup>H NMR spectra of **6b** on CDCl<sub>3</sub> 500 MHz



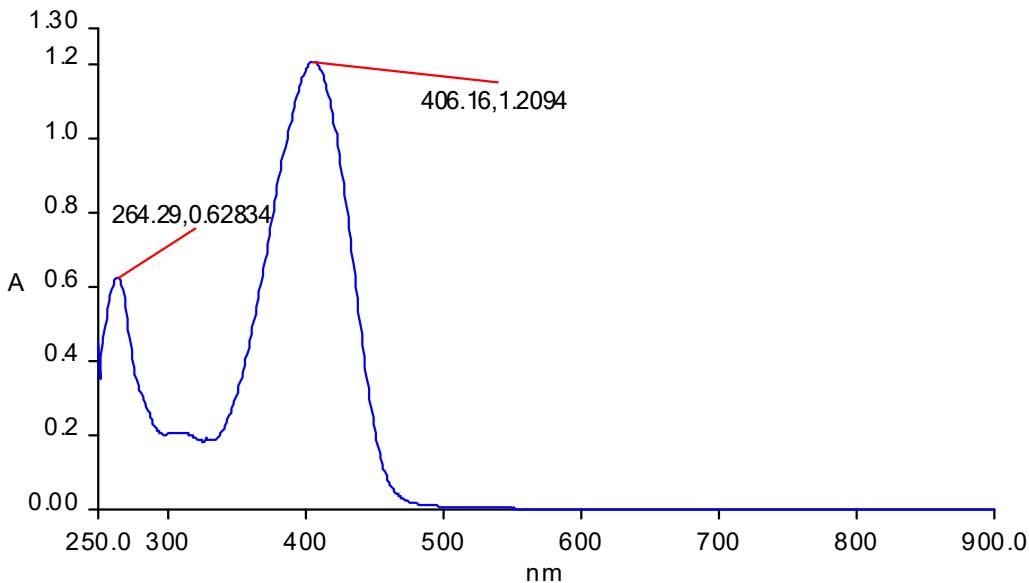
**Figure SI-36.** <sup>13</sup>C NMR spectra of **6b** on CDCl<sub>3</sub> 125 MHz.



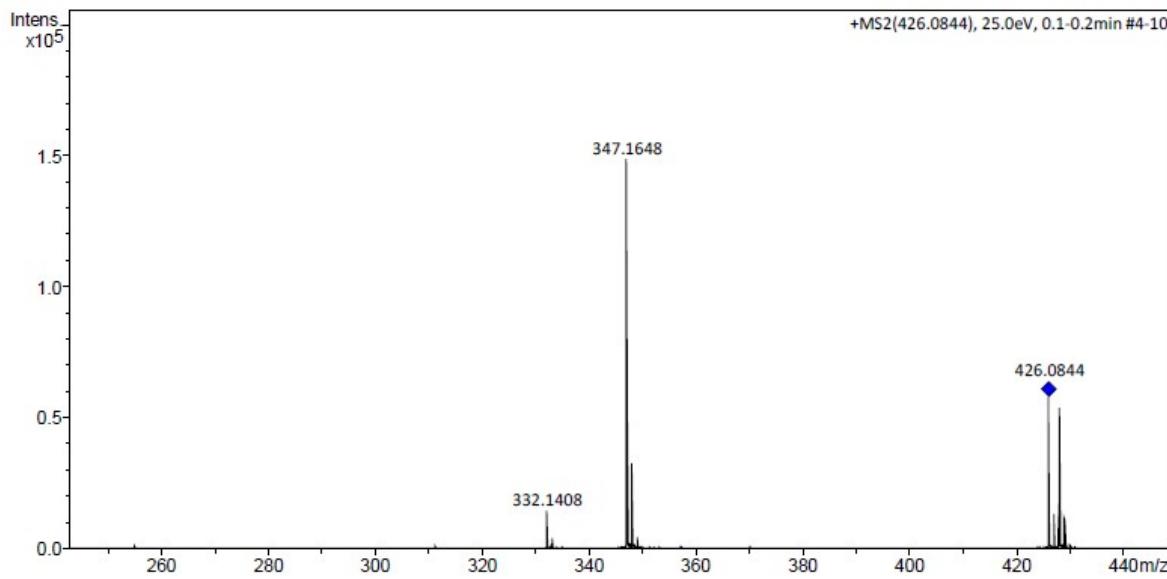
**Figure SI-37.** HMBC NMR spectra of **6b** on  $\text{CDCl}_3$



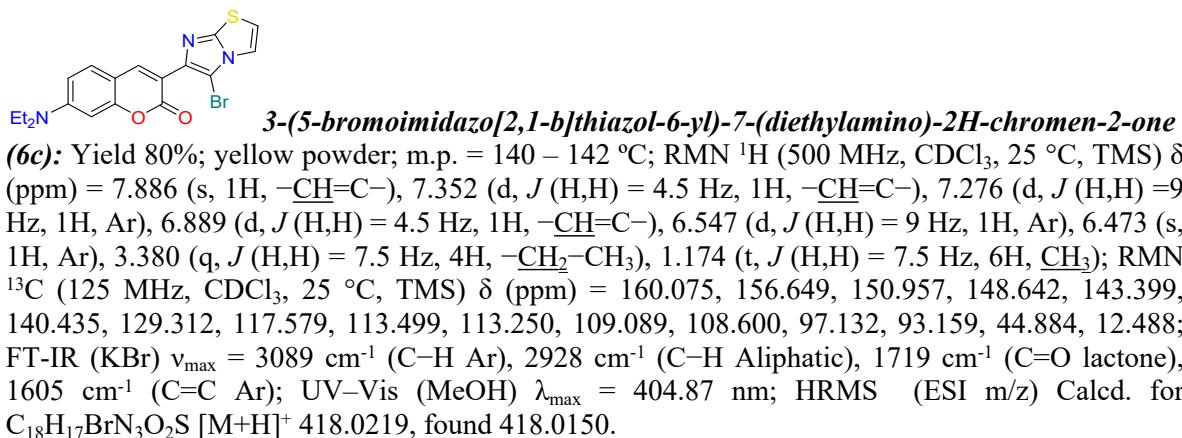
**Figure SI-38.** IR spectra for **6b** in KBr.

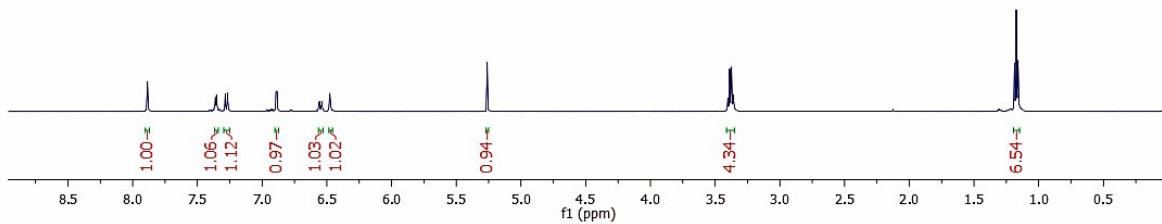


**Figure SI-39.** UV-Vis spectra of **6b** in MeOH.

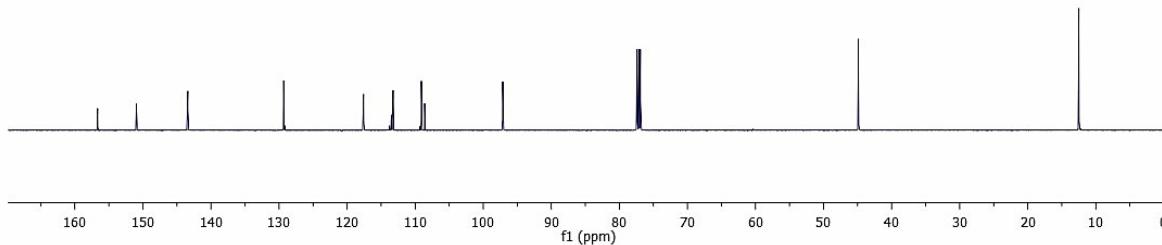


**Figure SI-40.** ESI-MS chromatogram of **6b**.

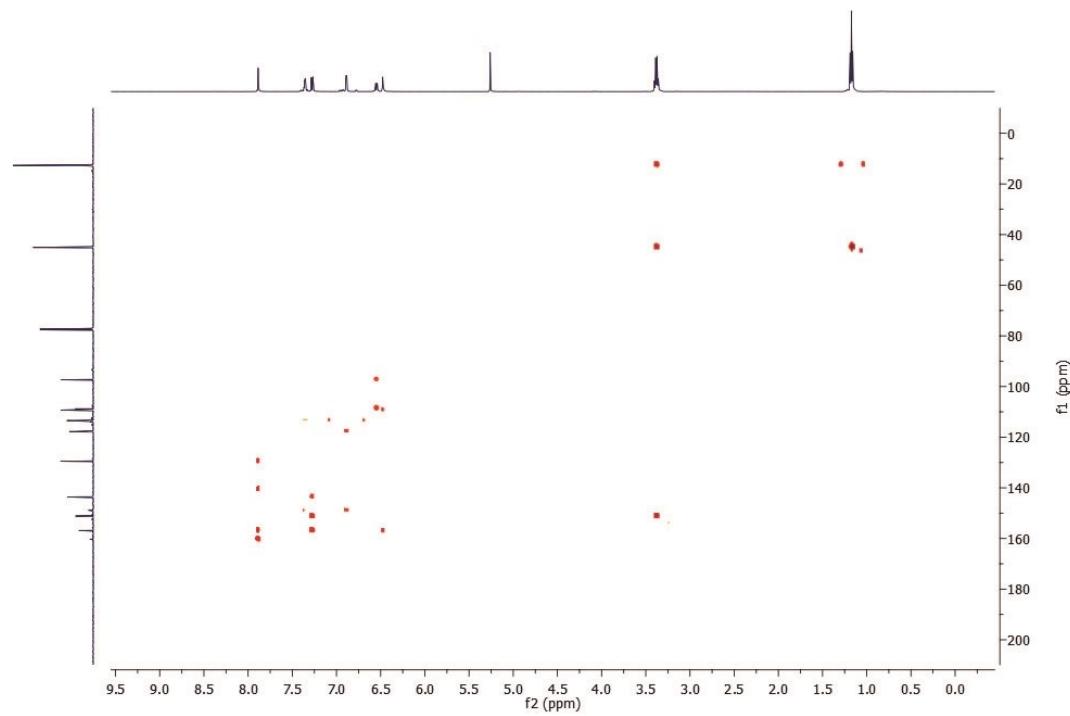




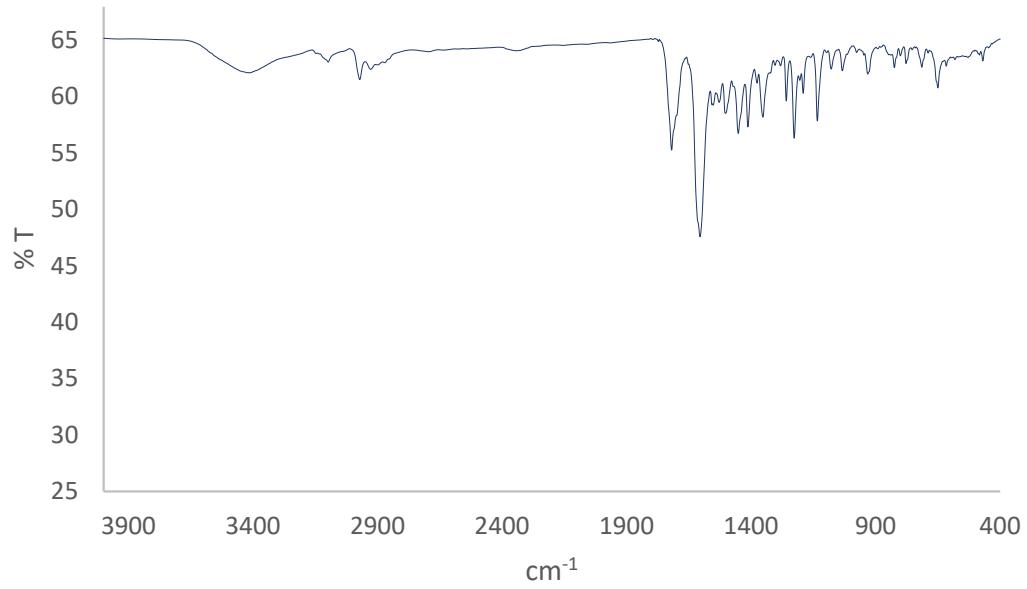
**Figure SI-41.**  $^1\text{H}$  NMR spectra of **6c** on  $\text{CDCl}_3$  500 MHz



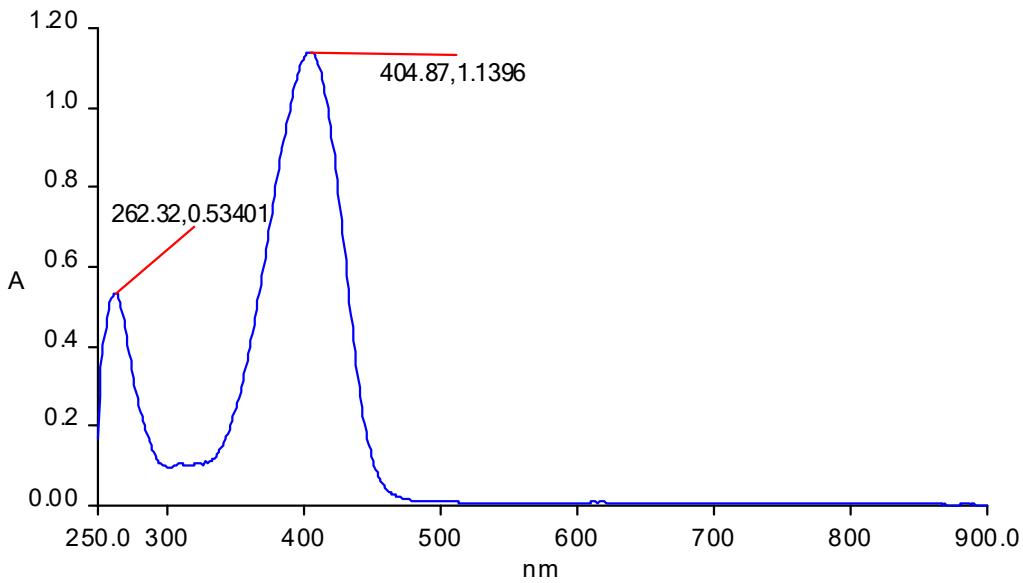
**Figure SI-42.**  $^{13}\text{C}$  NMR spectra of **6c** on  $\text{CDCl}_3$  125 MHz.



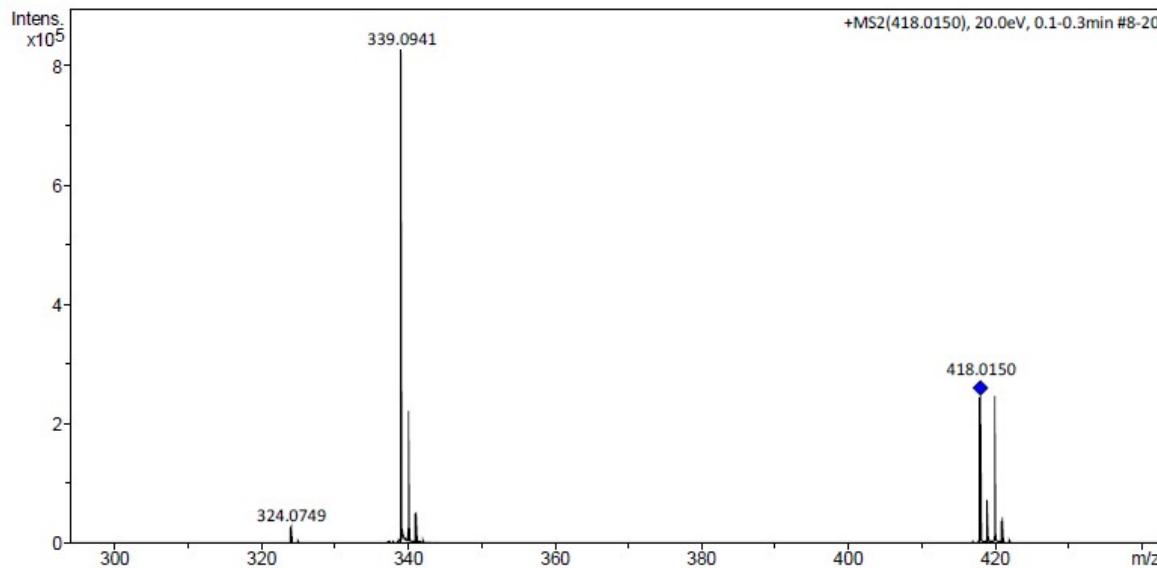
**Figure SI-43.** HMBC NMR spectra of **6c** on  $\text{CDCl}_3$



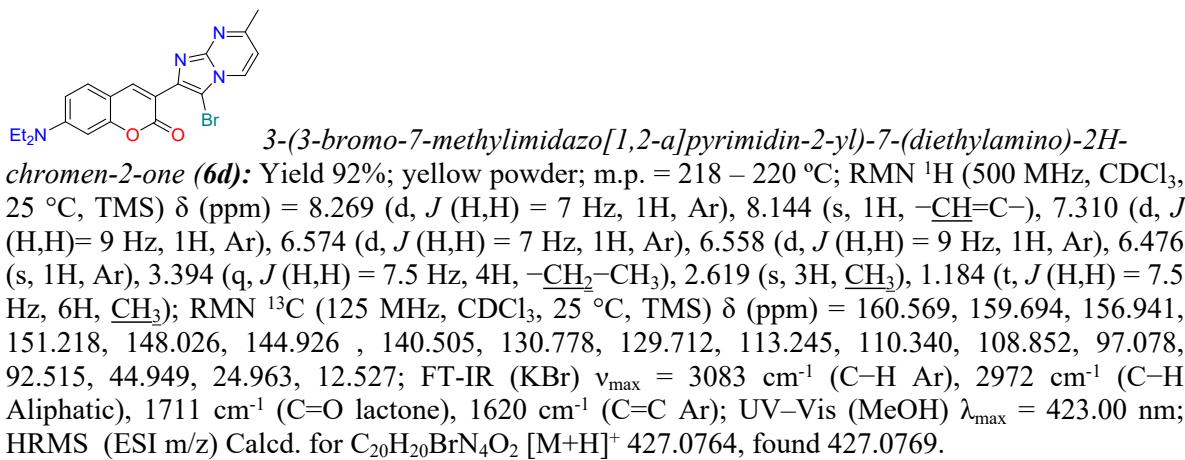
**Figure SI-44.** IR spectra for **6c** in KBr.

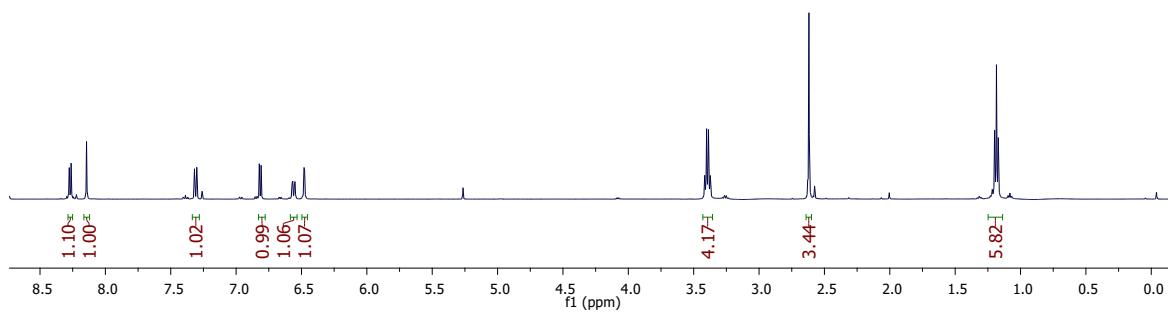


**Figure SI-45.** UV-Vis spectra of **6c** in MeOH.

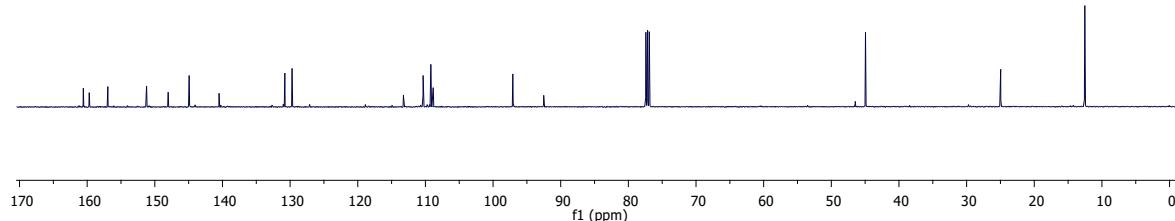


**Figure SI-46.** ESI-MS chromatogram of **6c**.

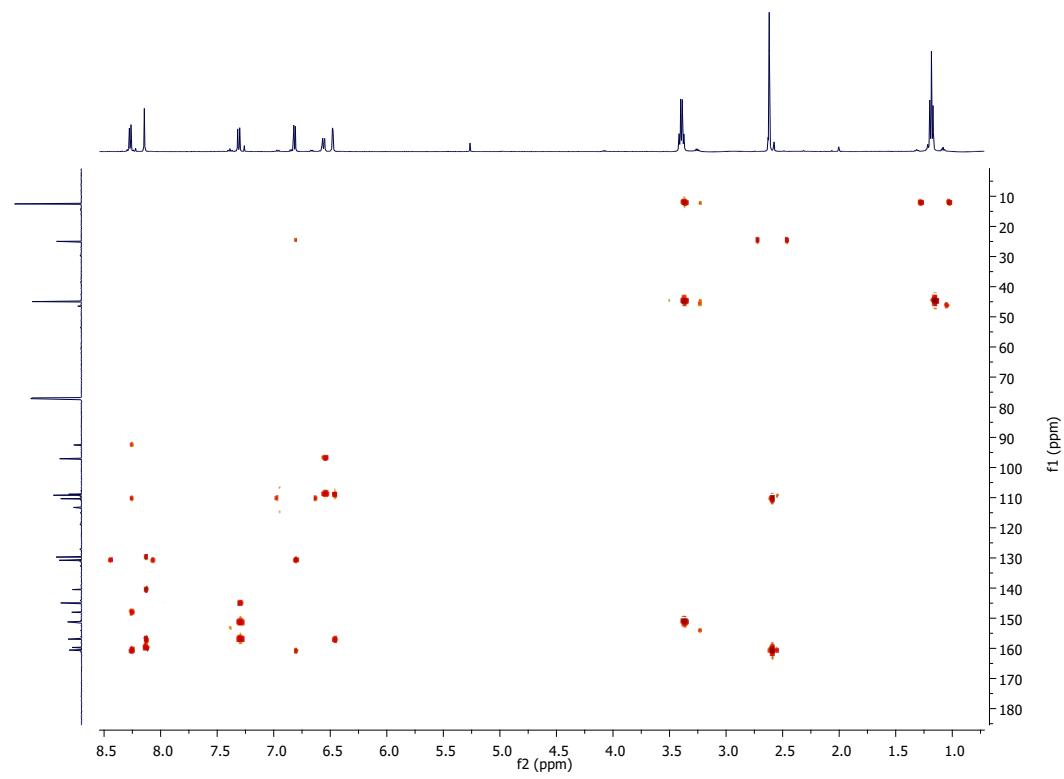




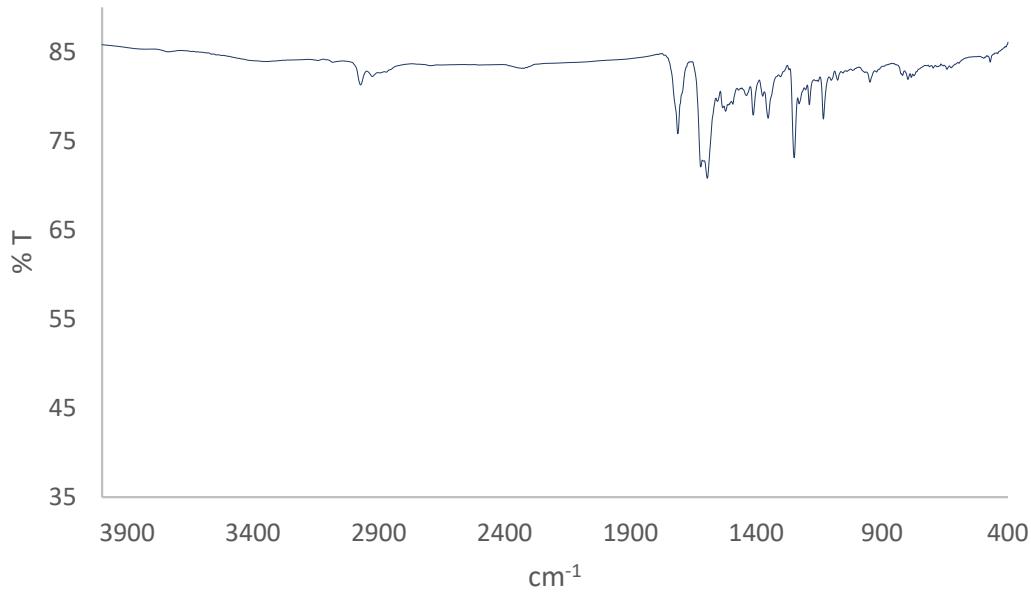
**Figure SI-47.** <sup>1</sup>H NMR spectra of **6d** on  $\text{CDCl}_3$  500 MHz



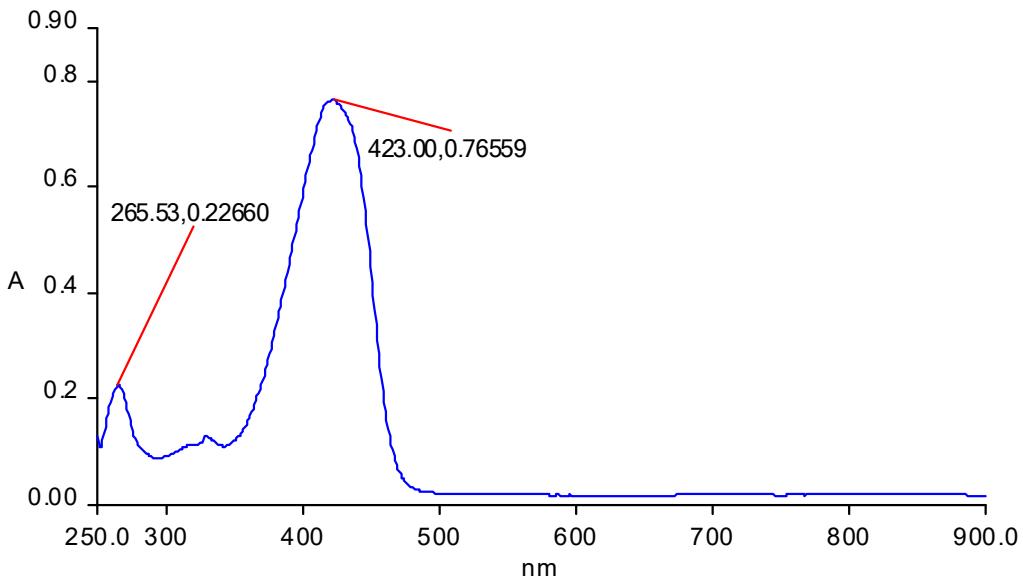
**Figure SI-48.** <sup>13</sup>C NMR spectra of **6d** on  $\text{CDCl}_3$  125 MHz.



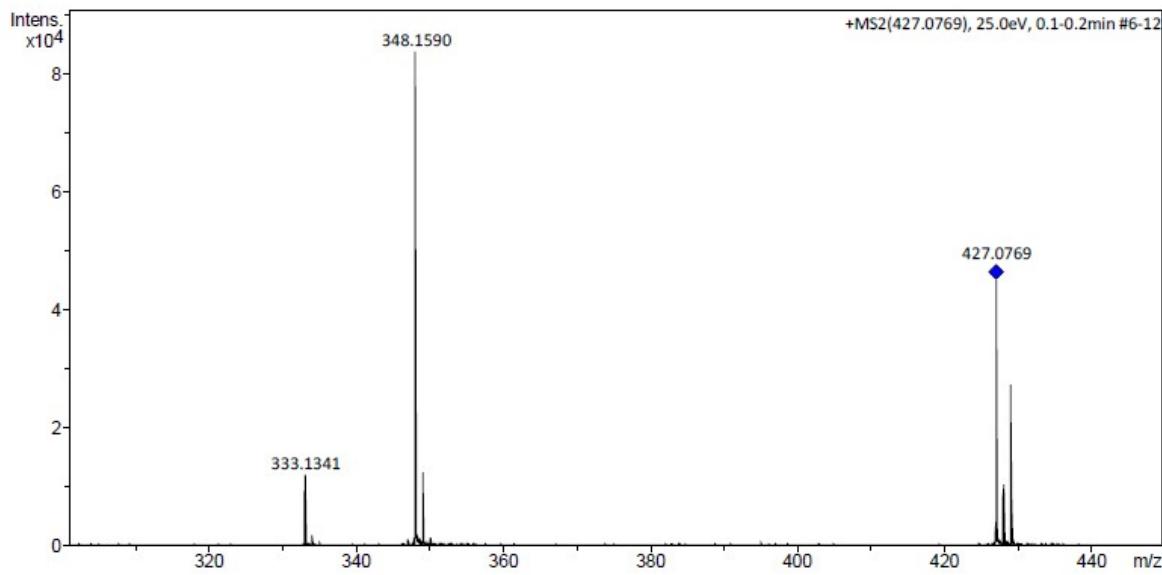
**Figure SI-49.** HMBC NMR spectra of **6d** on  $\text{CDCl}_3$



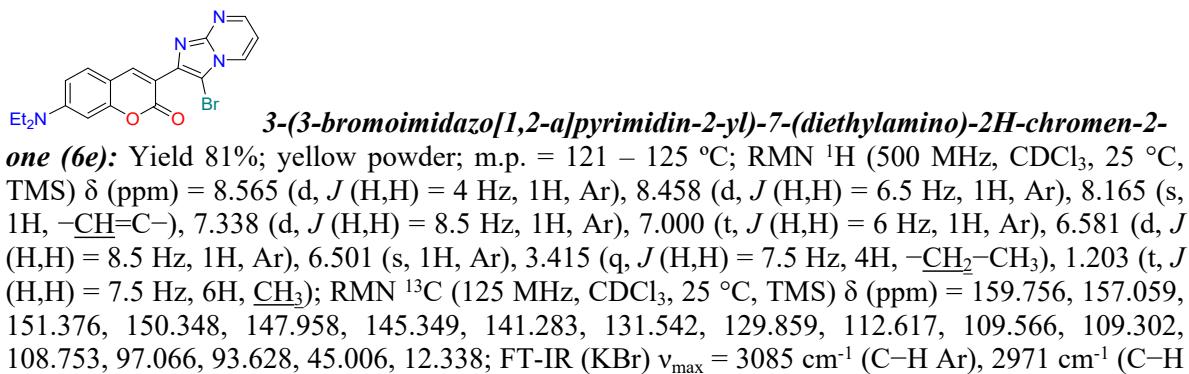
**Figure SI-50.** IR spectra for **6d** in KBr.



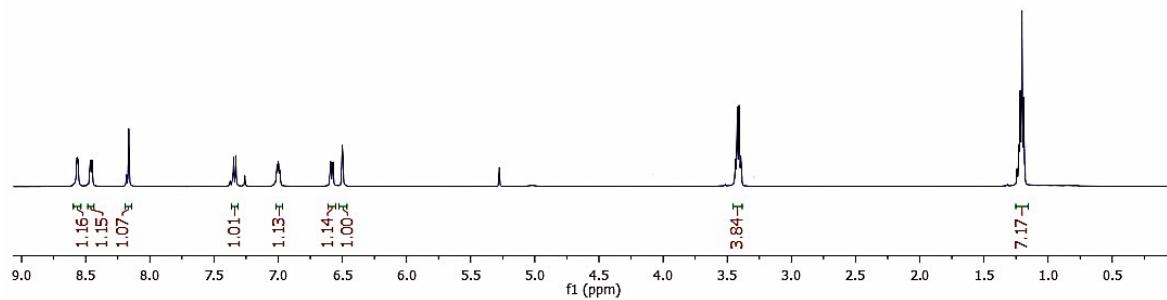
**Figure SI-51.** UV-Vis spectra of **6d** in MeOH.



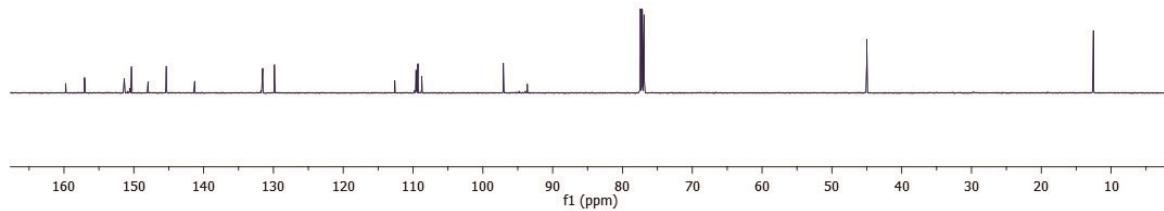
**Figure SI-52.** ESI-MS chromatogram of **6d**.



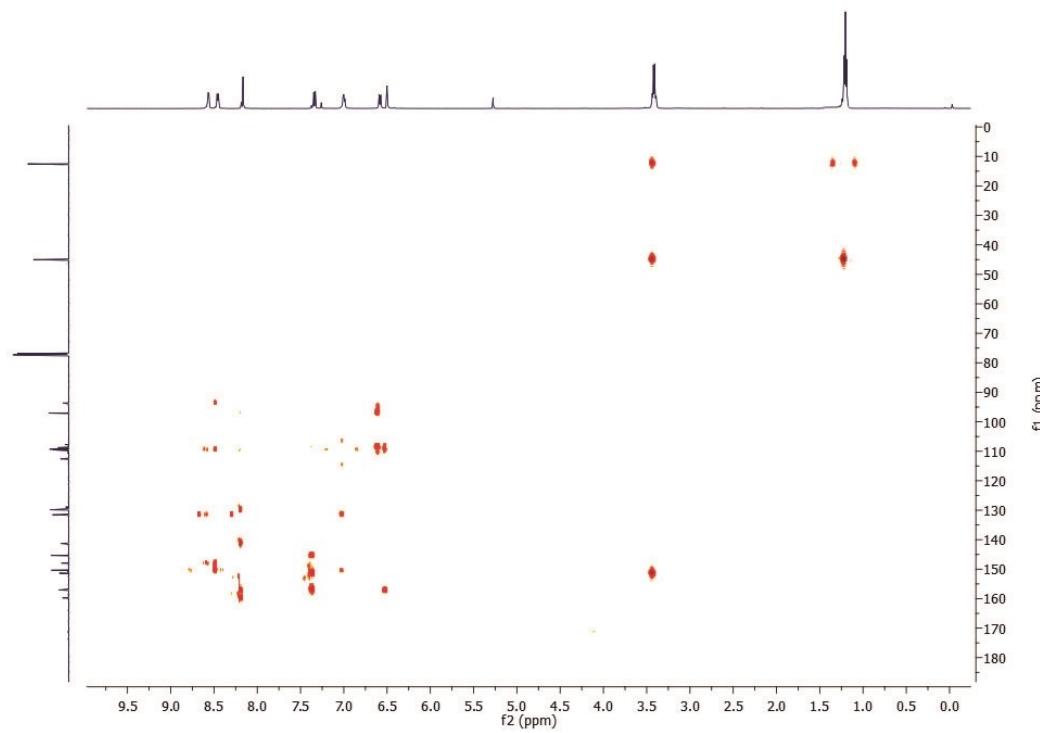
Aliphatic), 1717 cm<sup>-1</sup> (C=O lactone), 1598 cm<sup>-1</sup> (C=C Ar); UV–Vis (MeOH)  $\lambda_{\text{max}} = 408.14$  nm; HRMS (ESI m/z) Calcd. for C<sub>19</sub>H<sub>18</sub>BrN<sub>4</sub>O<sub>2</sub> [M+H]<sup>+</sup> 413.0608, found 413.0614.



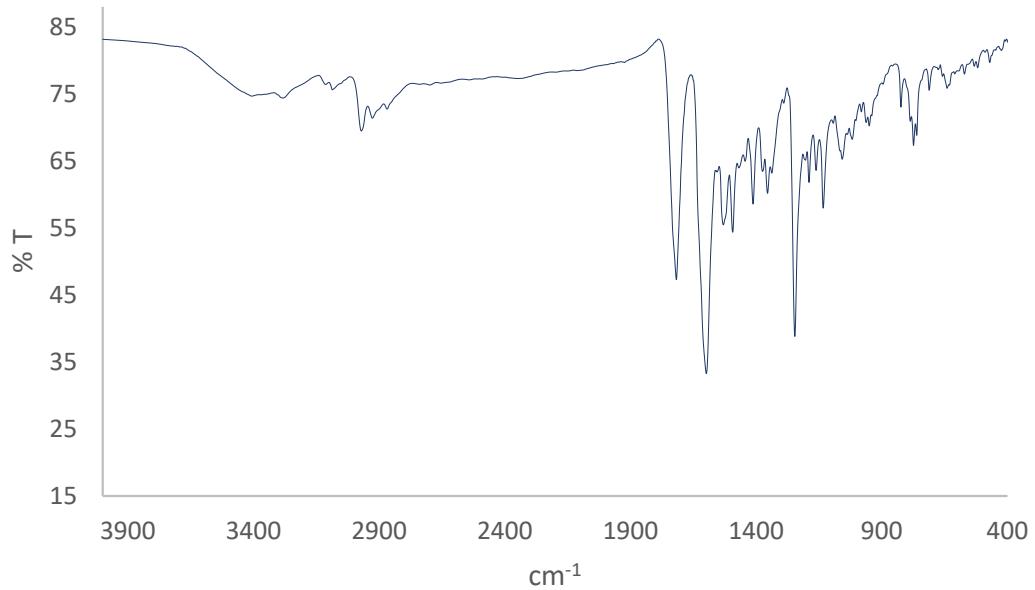
**Figure SI-53.** <sup>1</sup>H NMR spectra of **6e** on CDCl<sub>3</sub> 500 MHz



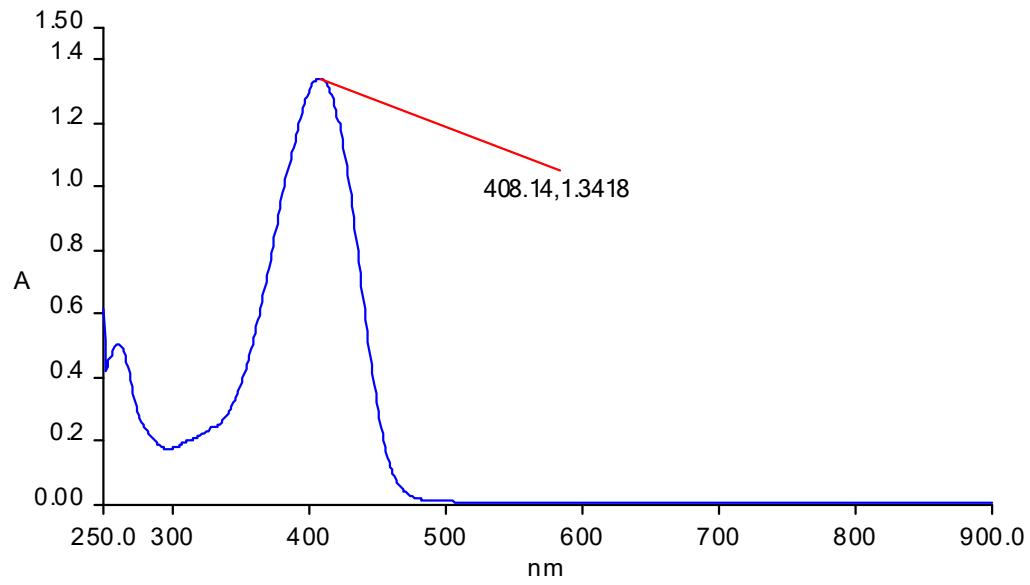
**Figure SI-54.**  $^{13}\text{C}$  NMR spectra of **6e** on  $\text{CDCl}_3$  125 MHz.



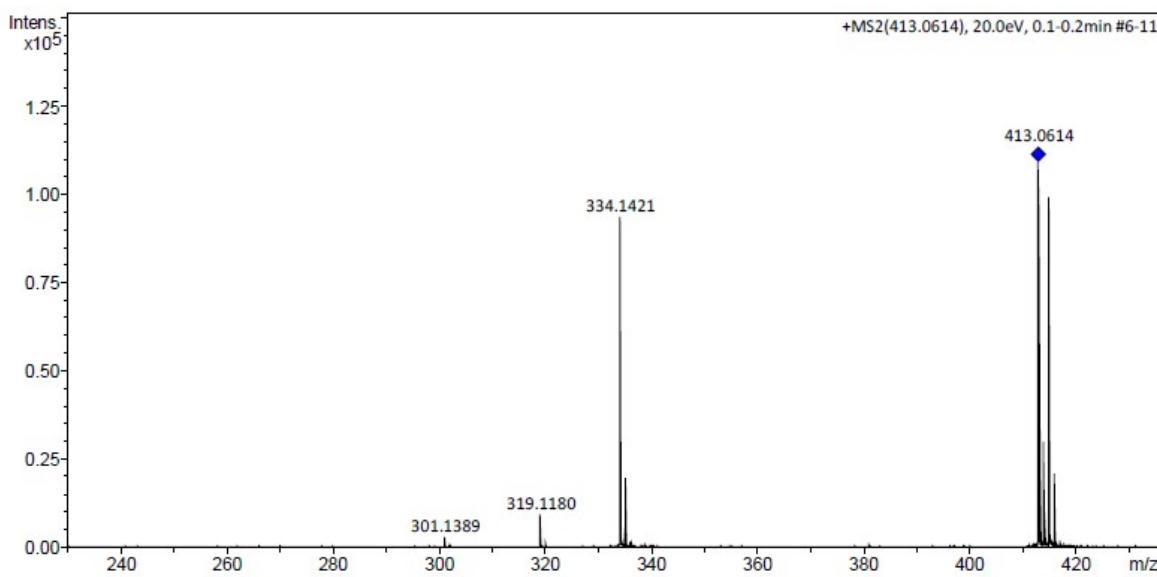
**Figure SI-55.** HMBC NMR spectra of **6e** on  $\text{CDCl}_3$



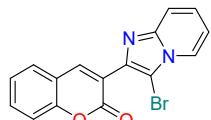
**Figure SI-56.** IR spectra for **6e** in KBr.



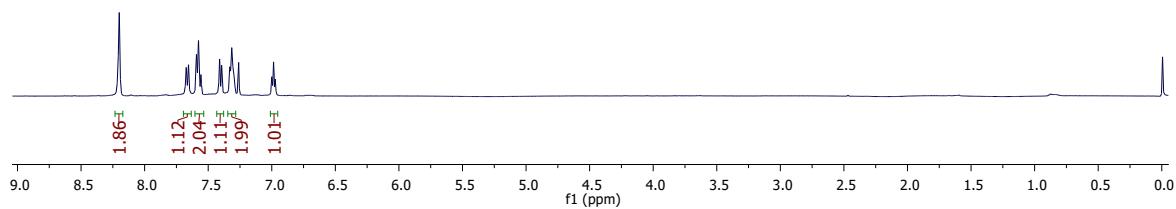
**Figure SI-57.** UV-Vis spectra of **6e** in MeOH.



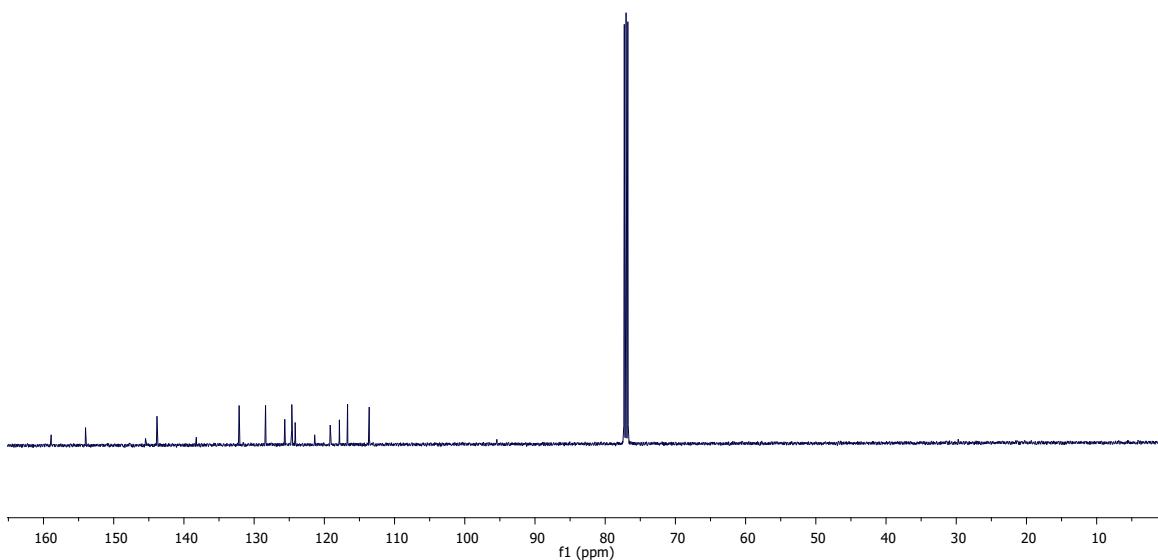
**Figure SI-58.** ESI-MS chromatogram of **6e**.



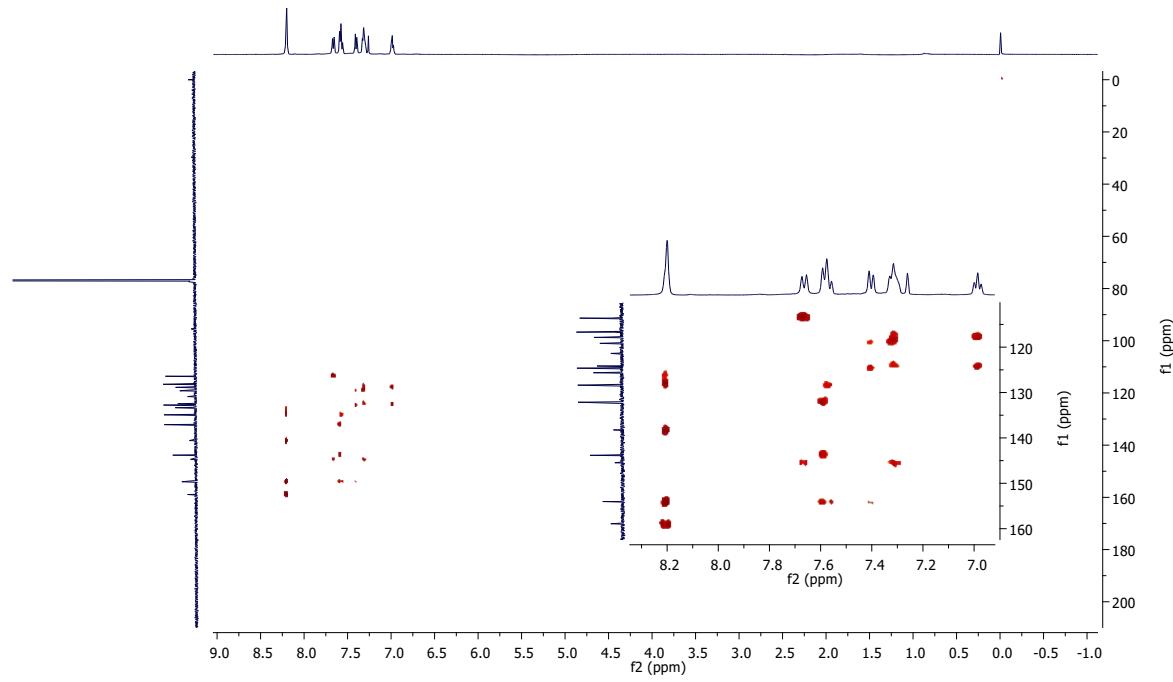
**3-(3-bromoimidazo[1,2-a]pyridin-2-yl)-2H-chromen-2-one (6f):** Yield 98%; brown powder; m.p. = 195 – 197 °C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 8.200 (m, 2H, Ar), 7.664 (d, *J* (H,H) = 7 Hz, 1H, Ar), 7.592 – 7.557 (m, 2H, Ar), 7.419 (d, *J* (H,H) = 9 Hz, 1H, Ar), 7.320 – 7.310 (m, 2H, Ar), 6.986 (t, *J* (H,H) = 7 Hz, 1H, Ar); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 158.909, 154.018, 145.469, 143.842, 138.235, 132.138, 128.376, 125.639, 100, 124.156, 121.382, 119.166, 117.860, 116.691, 113.609, 95.432; FT-IR (KBr) ν<sub>max</sub> = 3040 cm<sup>-1</sup> (C–H Ar), 2922 cm<sup>-1</sup> (C–H Aliphatic), 1722 cm<sup>-1</sup> (C=O lactone), 1632 cm<sup>-1</sup> (C=C Ar); UV–Vis (MeOH) λ<sub>max</sub> = 335.14 nm; HRMS (ESI m/z) Calcd. for C<sub>16</sub>H<sub>10</sub>BrN<sub>2</sub>O<sub>2</sub> [M+H]<sup>+</sup> 340.9920, found 340.9923.



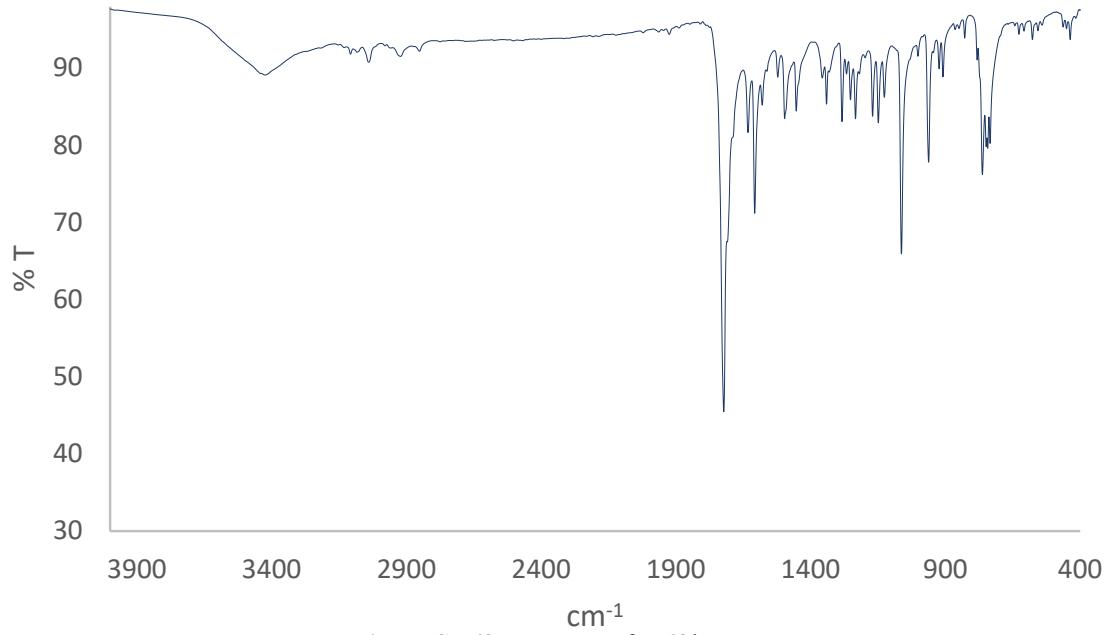
**Figure SI-59.**  $^1\text{H}$  NMR spectra of **6f** on  $\text{CDCl}_3$  500 MHz



**Figure SI-60.**  $^{13}\text{C}$  NMR spectra of **6f** on  $\text{CDCl}_3$  125 MHz.



**Figure SI-61.** HMBC NMR spectra of **6f** on  $\text{CDCl}_3$



**Figure SI-62.** IR spectra for **6f** in KBr.

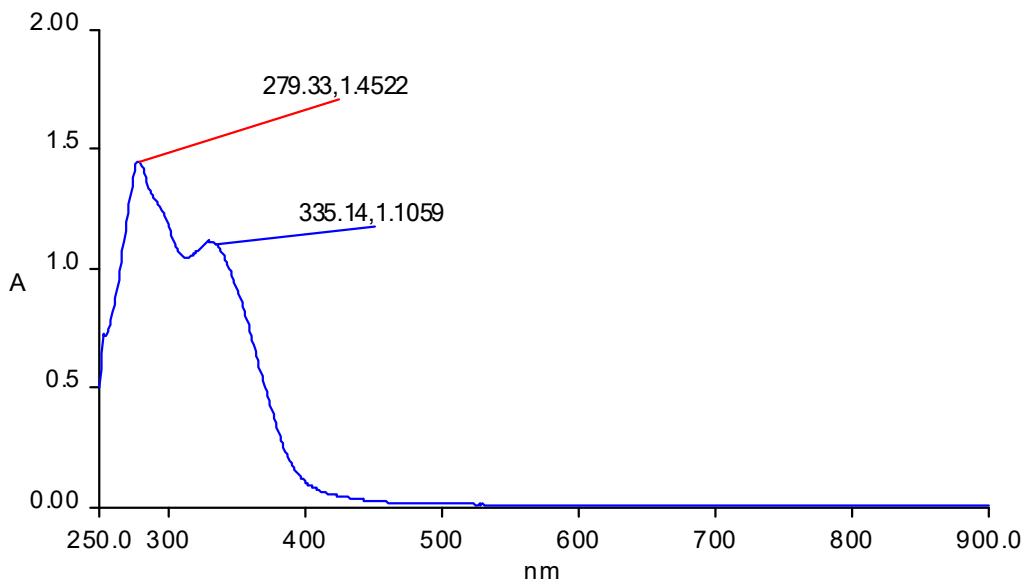


Figure SI-63. UV-Vis spectra of **6f** in MeOH.

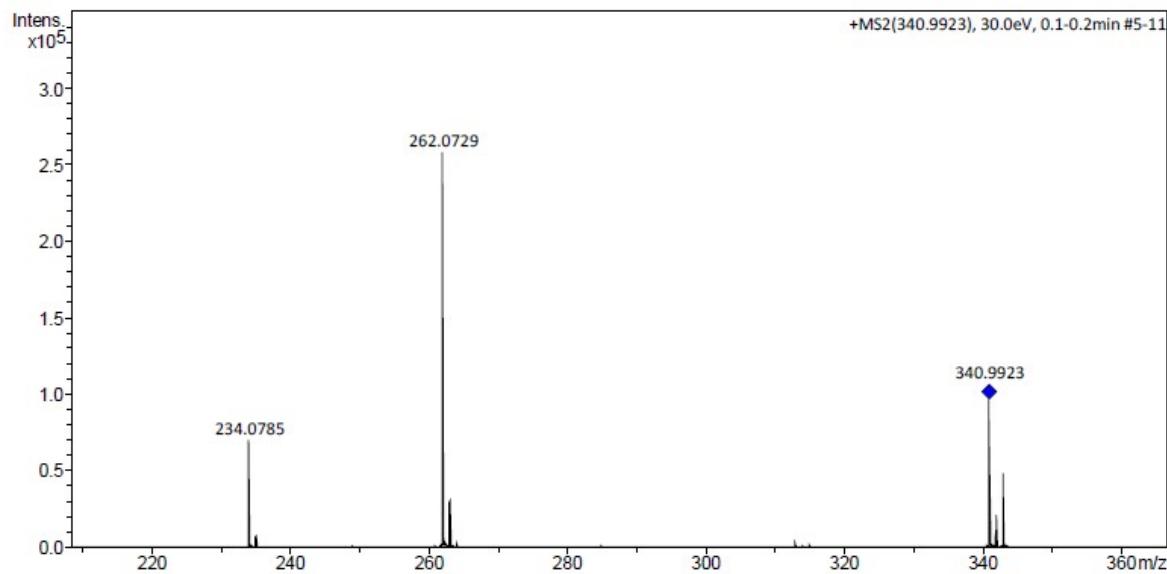
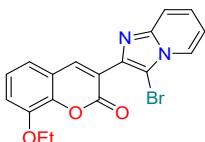
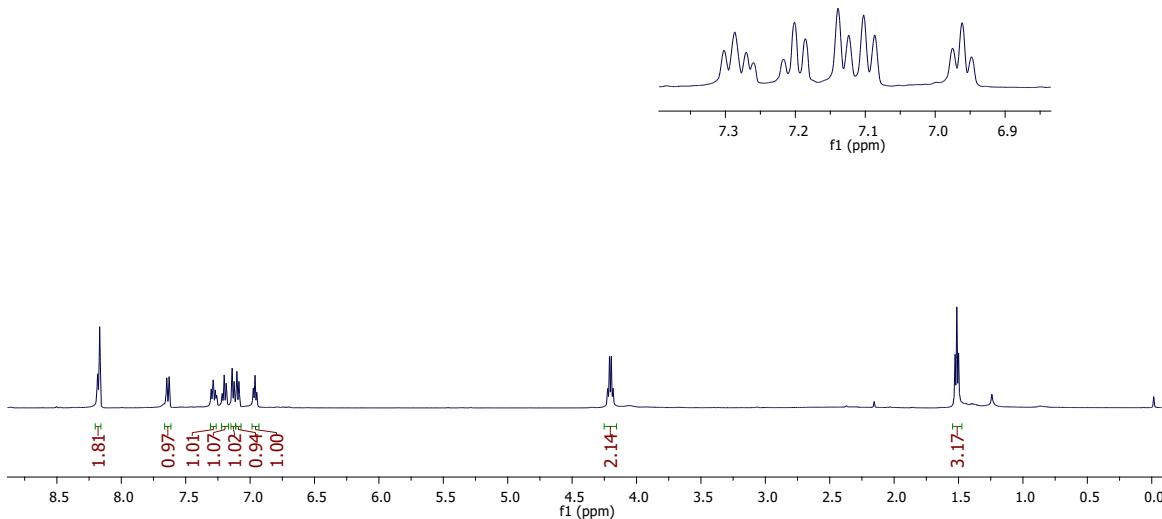


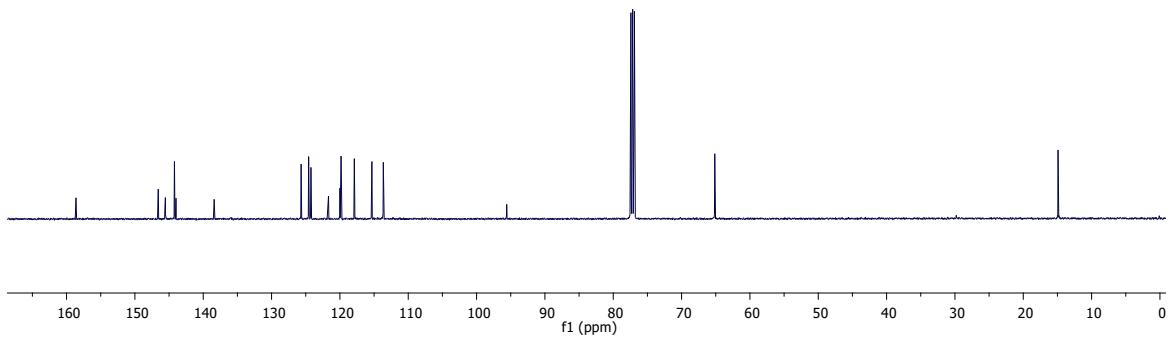
Figure SI-64. ESI-MS chromatogram of **6f**.



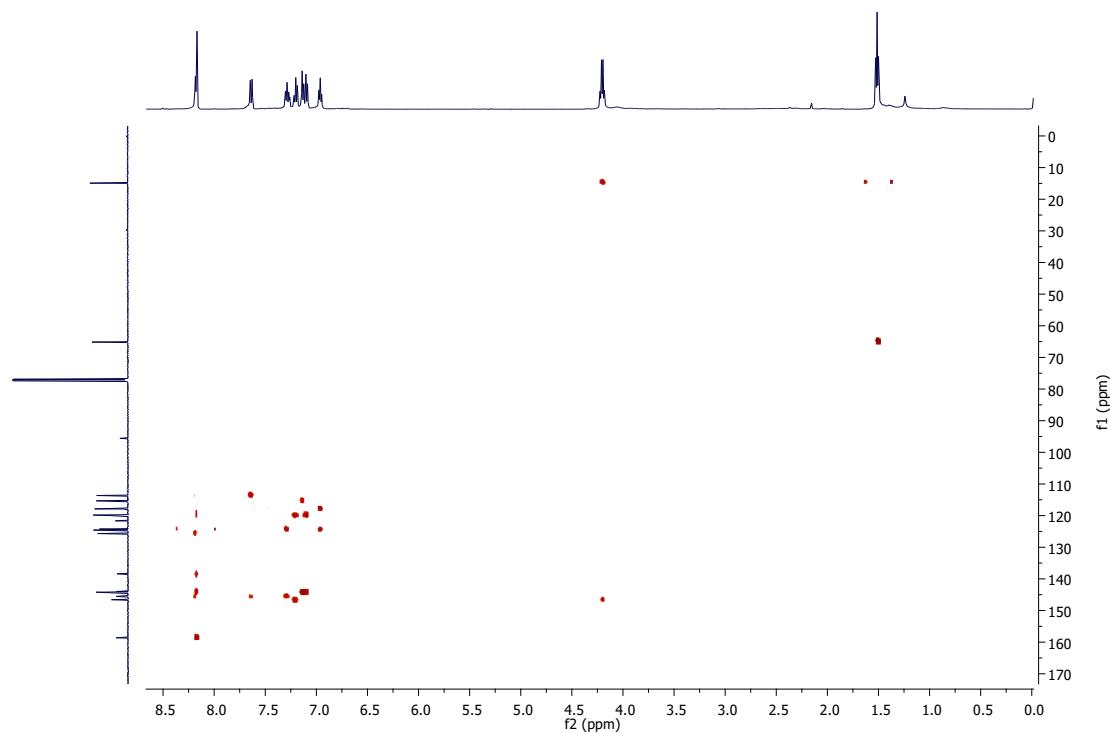
**3-(3-bromoimidazo[1,2-a]pyridin-2-yl)-8-ethoxy-2H-chromen-2-one (**6g**):** Yield 56%; dark brown powder; m.p. = 158 – 161 °C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 8.184 – 8.168 (m, 2H, Ar), 7.638 (sd, *J* (H,H) = 9 Hz, 1H, Ar), 7.287 (d, *J* (H,H) = 8.5 Hz, 1H, Ar), 7.201 (t, *J* (H,H) = 7.5 Hz, 1H, Ar), 7.132 (d, *J* (H,H) = 8 Hz, 1H, Ar), 7.094 (d, *J* (H,H) = 8 Hz, 1H, Ar), 6.962 (t, *J* (H,H) = 7 Hz, 1H, Ar), 4.203 (q, *J* (H,H) = 7.5 Hz, 1H, –CH<sub>2</sub>–CH<sub>3</sub>), 1.513 (t, *J* (H,H) = 7.5 Hz, 1H, CH<sub>3</sub>); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 158.620, 146.580, 145.542, 144.212, 143.985, 138.401, 125.674, 124.574, 124.124, 121.668, 119.997, 119.829, 117.889, 115.319, 113.649, 95.590, 65.144, 14.911; FT-IR (KBr) ν<sub>max</sub> = 3040 cm<sup>-1</sup> (C–H Ar), 2922 cm<sup>-1</sup> (C–H Aliphatic), 1722 cm<sup>-1</sup> (C=O lactone), 1632 cm<sup>-1</sup> (C=C Ar); UV–Vis (MeOH) λ<sub>max</sub> = 340.05 nm; HRMS (ESI m/z) Calcd. for C<sub>18</sub>H<sub>14</sub>BrN<sub>2</sub>O<sub>3</sub> [M+H]<sup>+</sup> 385.0182, found 385.0214.



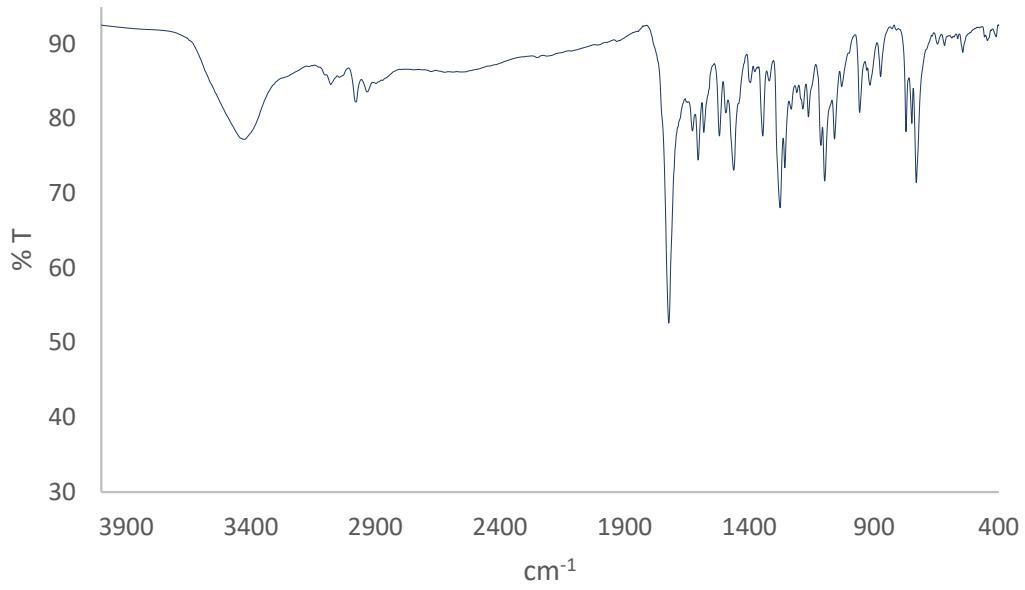
**Figure SI-65.** <sup>1</sup>H NMR spectra of **6g** on CDCl<sub>3</sub> 500 MHz



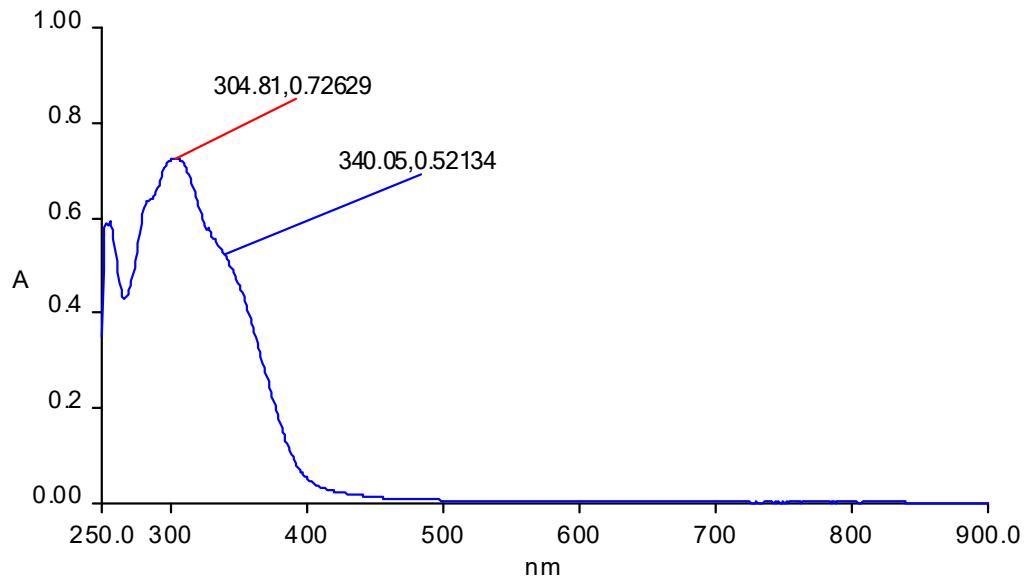
**Figure SI-66.**  $^{13}\text{C}$  NMR spectra of **6g** on  $\text{CDCl}_3$  125 MHz.



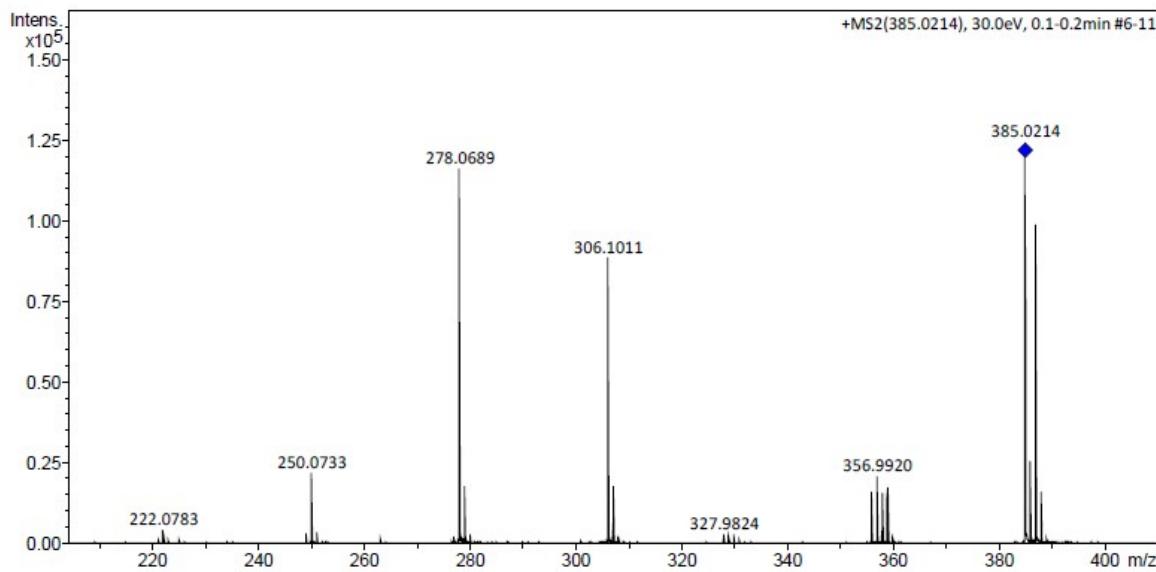
**Figure SI-67.** HMBC NMR spectra of **6g** on  $\text{CDCl}_3$



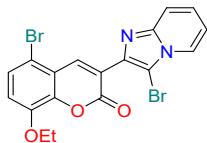
**Figure SI-68.** IR spectra for **6g** in KBr.



**Figure SI-69.** UV-Vis spectra of **6g** in MeOH.

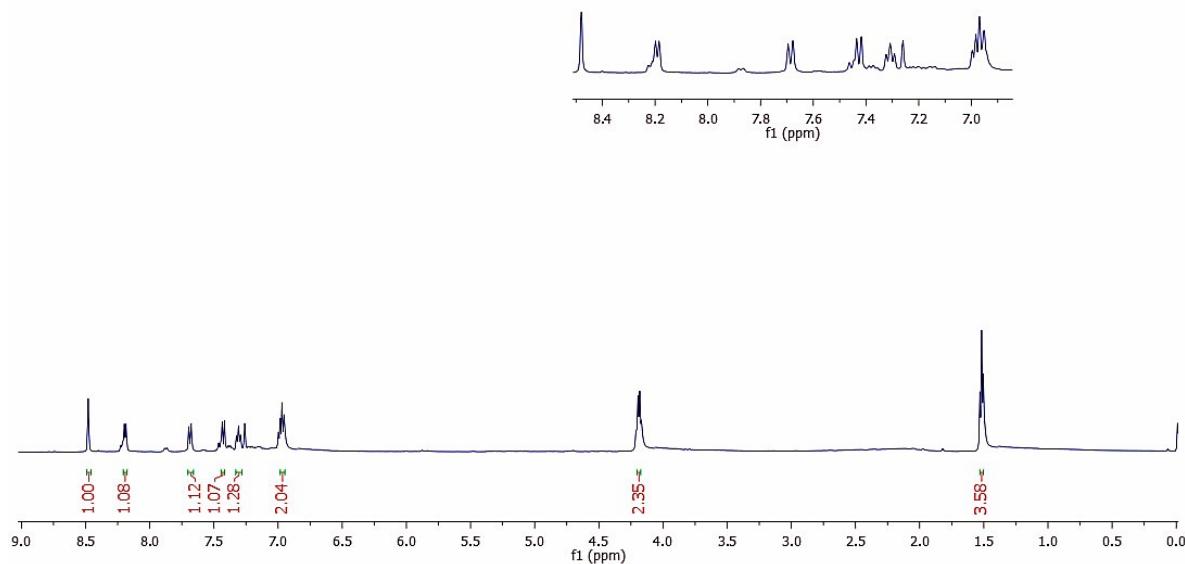


**Figure SI-70.** ESI chromatogram of **6g**.

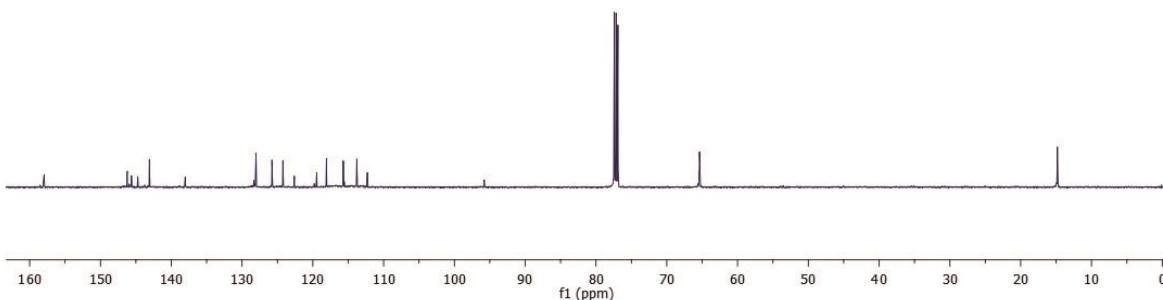


**5-bromo-3-(3-bromoimidazo[1,2-a]pyridin-2-yl)-8-ethoxy-2H-chromen-2-one**

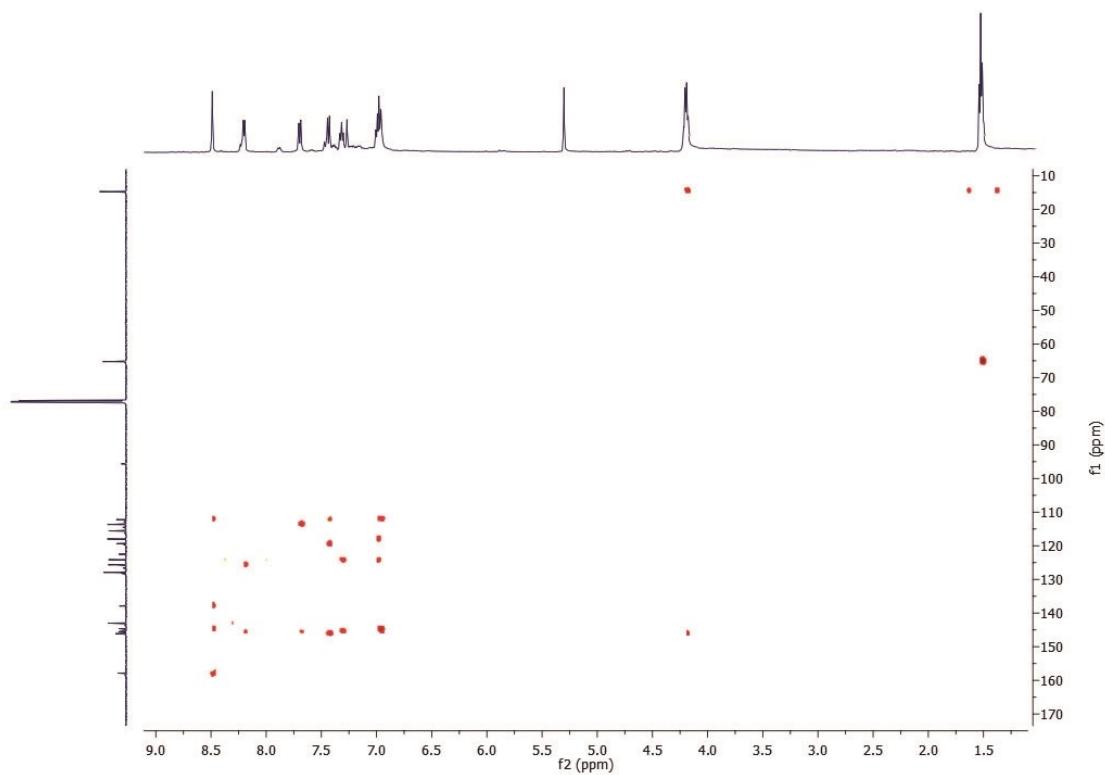
**(6h):** Yield 89%; reddish brown; m.p. = 165 – 167 °C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 8.480 (s, 1H, –CH<sub>2</sub>–CH<sub>3</sub>), 8.192 (d, *J* (H,H) = 6.5 Hz, 1H, Ar), 7.676 (d, *J* (H,H) = 9 Hz, 1H, Ar), 7.427 (d, *J* (H,H) = 9 Hz, 1H, Ar), 7.308 (t, *J* (H,H) = 7 Hz, 1H, Ar), 6.983 – 6.952 (m, 2H, H-6, Ar), 4.188 (q, *J* (H,H) = 7.5 Hz, 1H, –CH<sub>2</sub>–CH<sub>3</sub>), 1.517 (t, *J* (H,H) = 7.5 Hz, 1H, CH<sub>3</sub>); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 157.960, 146.215, 145.619, 144.729, 143.090, 138.033, 128.023, 125.776, 124.232, 122.606, 119.450, 118.080, 115.698, 113.785, 112.302, 95.780, 65.374, 14.807; FT-IR (KBr) ν<sub>max</sub> = 29.24 cm<sup>-1</sup> (C–H Aliphatic), 1728 cm<sup>-1</sup> (C=O lactone), 1573 cm<sup>-1</sup> (C=C Ar); UV–Vis (MeOH) λ<sub>max</sub> = 316.850 nm; HRMS (ESI m/z) Calcd. for C<sub>18</sub>H<sub>13</sub>Br<sub>2</sub>N<sub>2</sub>O<sub>3</sub> [M+H]<sup>+</sup> 461.9215, found 462.9295.



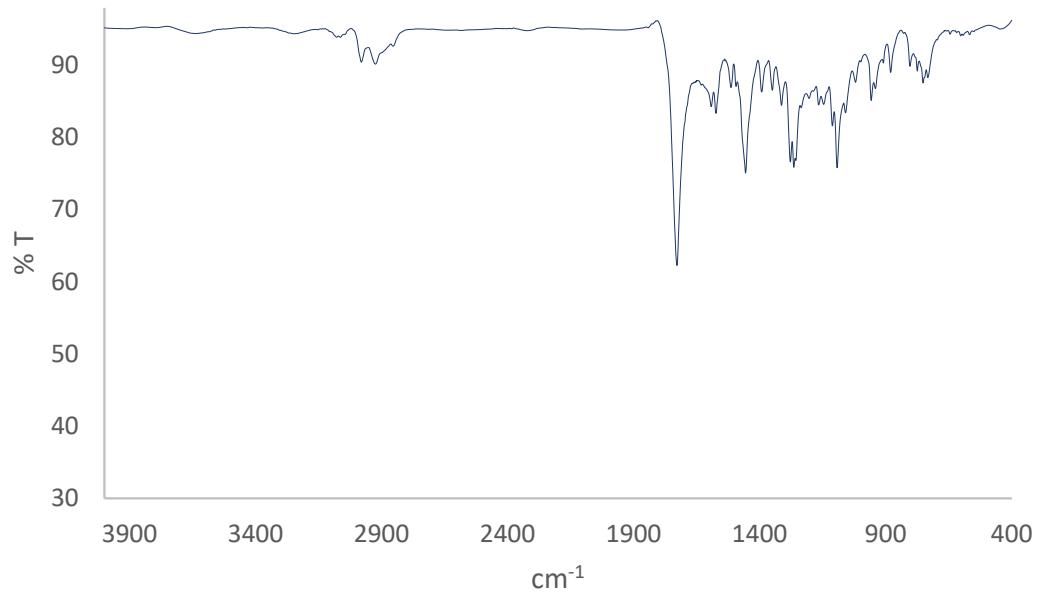
**Figure SI-71.** <sup>1</sup>H NMR spectra of **6h** on  $\text{CDCl}_3$  500 MHz



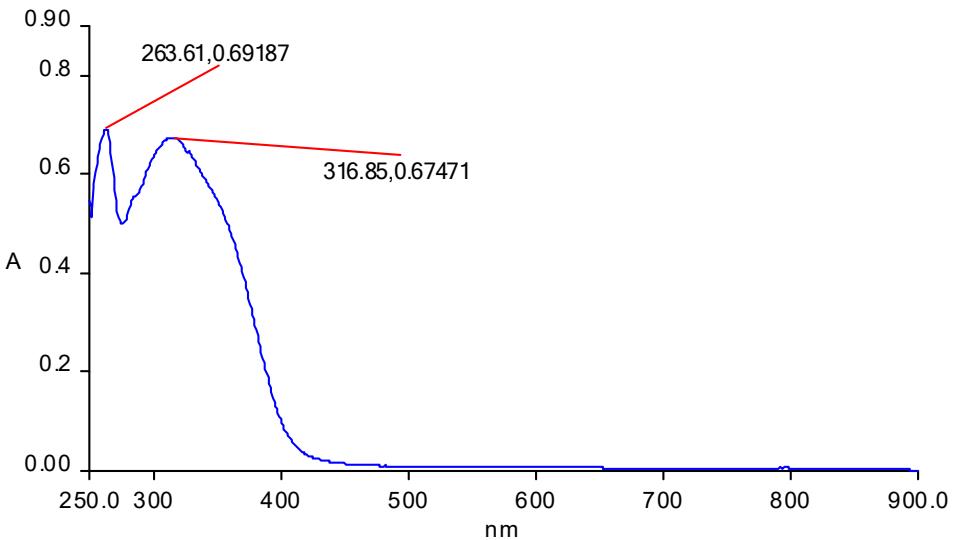
**Figure SI-72.** <sup>13</sup>C NMR spectra of **6h** on  $\text{CDCl}_3$  125 MHz.



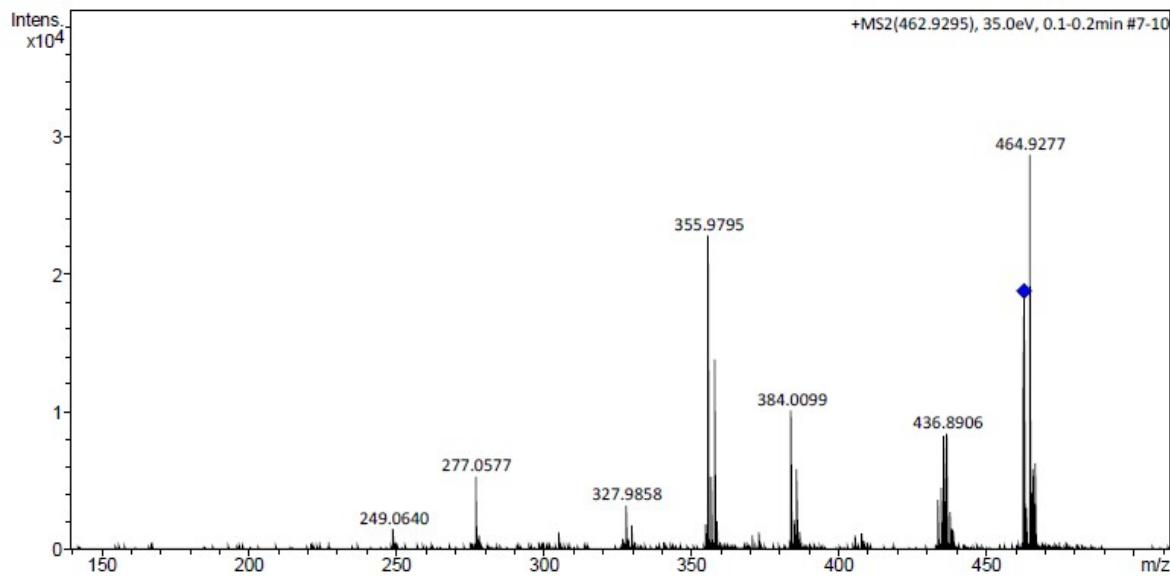
**Figure SI-73.** HMBC NMR spectra of **6h** on  $\text{CDCl}_3$



**Figure SI-74.** IR spectra for **6h** in KBr.



**Figure SI-75.** UV-Vis spectra of **6h** in MeOH.



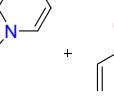
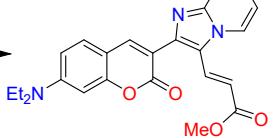
**Figure SI-76.** ESI-MS chromatogram of **6h**.

## 5. General procedure of coumarin-imidazo[1,2-*a*]pyridine-3-acrylates derivatives **7a-h**

### Method A: Alkenylation by C–H bond activation.

In a 15 mL round bottom flask **4a** (0.5 mmol), catalyst (15% mol), phase transfer agent (TA, 1 mmol), and base (1 mmol) were added according to the reagents shown in Table 1, the flask was sealed and purged with nitrogen, 5 mL of DMF and 2 mmol of methyl acrylate were injected into the flask. The flask was kept at 100 °C and shaken constantly for 7 days. The reaction crude was treated by adding distilled water, which caused the precipitation of the medium, the solid was filtered and redissolved in dichloromethane, the solution was filtered again through a buchner funnel packed with celite, the collected solution was washed with three volumes of NaHCO<sub>3</sub> saturated solution, the organic phase was concentrated and the solid was purified by chromatographic column in Hex/AcOEt 80:20 elution system.

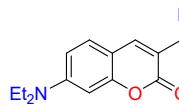
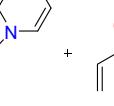
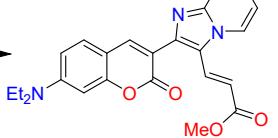
**Table S1.** Condition optimization for C–H bond activation.

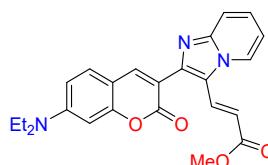
		+ 	DMF, 100 °C		
Entry	Catalyst	Base	TA <sup>a</sup>	t (days)	(%) R
1	Pd(OAc) <sub>2</sub> (15%)			2	30
2	PdCl <sub>2</sub> (15%)			7	22
3	Palladium Dichloro-N,N,N',N'-Tetramethylethylenediamine (II) (15%)	NaHCO <sub>3</sub>	Hexadecyltrimethylammonium Bromide (C <sub>19</sub> H <sub>42</sub> BrN)	7	18
4		CaCO <sub>3</sub>		7	18
5		t-BuOK		7	15
6	Pd(OAc) <sub>2</sub> (15%)	NaHCO <sub>3</sub>	Tetrabutylammonium acetate	7	-----
7			-----	7	-----

<sup>a</sup>TA : Transfer agent**Method B: Heck reaction**

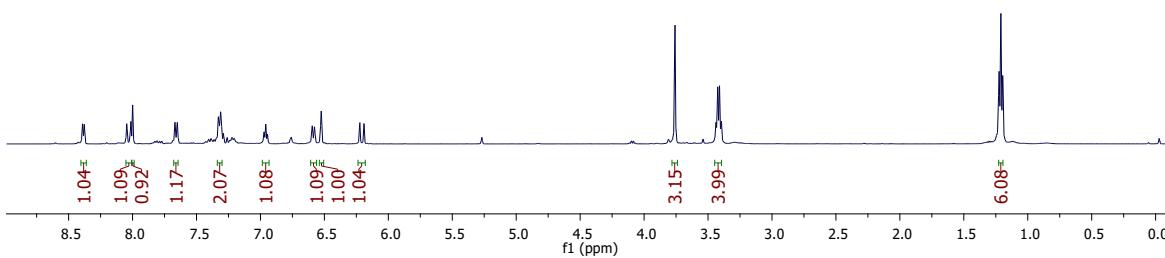
In a 15 mL round bottom flask, **6a-j** (0.5 mmol), catalyst (15% mol), BINAP (30% mol), base (1 mmol) and AgOAc (1 mmol) were added. Table 2 shows the reagents used. The flask was sealed and purged with nitrogen, then solvent (5 mL) and methyl acrylate (2 mmol) were injected, the flask was kept at 100 °C and stirred (2-7 days). The reaction was treated by adding distilled water which caused the precipitation of the medium, the solid was filtered and redissolved in dichloromethane, the collected solution was washed with three volumes of NaHCO<sub>3</sub> saturated solution, the organic phase was concentrated and purified by chromatographic column in Hex/AcOEt 80:20 elution system.

**Table S2.** Condition optimization for Heck reaction.

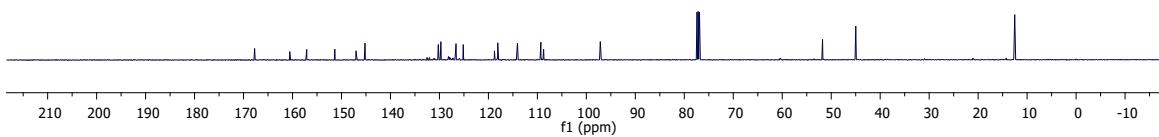
		+ 	Conditions				
Entry	Catalyst	Solvent	Base	Additive	T (°C)	%R	t (days)
1	Pd(OAc) <sub>2</sub> (15%)	DMF		----		70	2
2					100	78	2
3		Dioxane				85	2
4		THF			66	60	7
5	PdCl <sub>2</sub> (15%)		NaHCO <sub>3</sub>	Ag(OAc) <sub>2</sub>		80	7
6	Palladium Dichloro-N,N,N',N'-Tetramethylethylenediamine (II) (15%)	Dioxane			100	66	7
7	Pd(OAc) <sub>2</sub> (15%)		CaCO <sub>3</sub>			85	2
8	Pd(OAc) <sub>2</sub> (15%)		t-BuOK			67	2
9	Pd(OAc) <sub>2</sub> (15%)		CaCO <sub>3</sub>			87	2



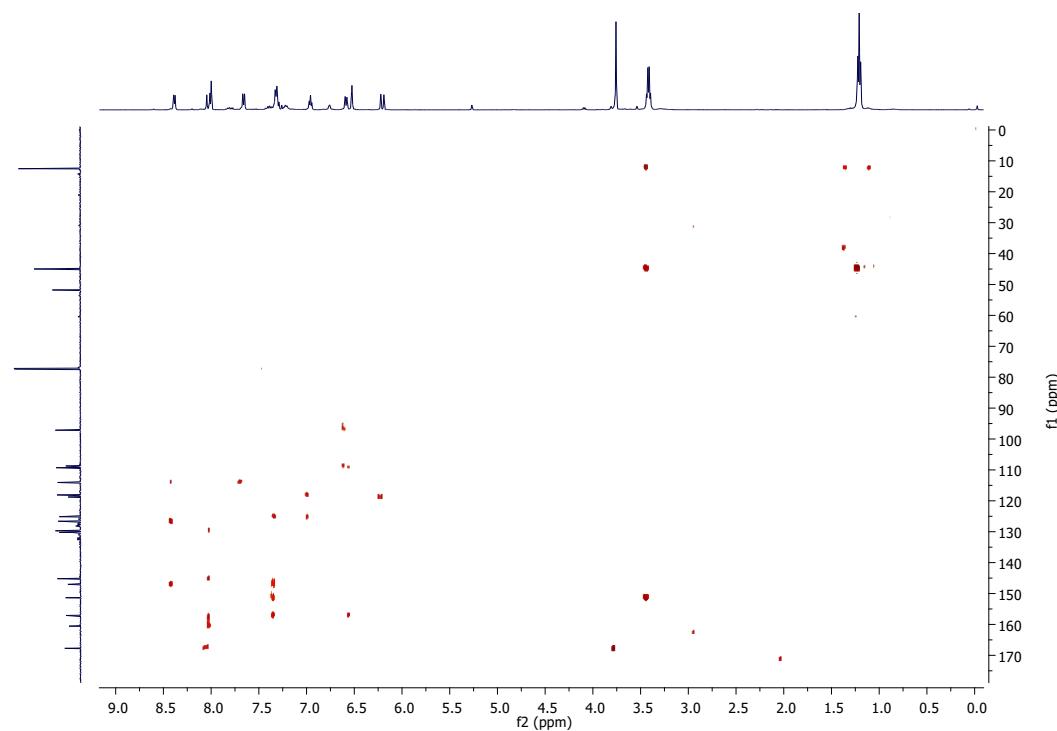
**Methyl (E)-3-(2-(7-(diethylamino)-2-oxo-2H-chromen-3-yl)imidazo[1,2-a]pyridin-3-yl)acrylate (7a):** Yield 87%; yellow powder; m.p. = 219 – 223 °C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 8.391 (d, *J* (H,H) = 6.5 Hz, 1H, Ar), 8.013 – 7.999 (m, 2H, Ar), 7.659 (d, *J* (H,H) = 9 Hz, 1H, Ar), 7.329 – 7.311 (m, 2H, Ar), 6.965 (d, *J* (H,H) = 7 Hz, 1H, Ar), 6.587 (d, *J* (H,H) = 9 Hz, 1H, Ar), 6.525 (s, 1H, Ar), 7.206 (d, *J* (H,H) = 16 Hz, 1H, –CH=C–), 3.758 (s, 1H, CH<sub>3</sub>), 3.417 (q, *J* (H,H) = 7.5 Hz, 4H, –CH<sub>2</sub>–CH<sub>3</sub>), 1.210 (t, *J* (H,H) = 7.5 Hz, 6H, CH<sub>3</sub>); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, TMS) δ (ppm): CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 167.734, 160.559, 157.095, 151.356, 147.035, 145.191, 145.160, 130.231, 129.705, 126.609, 125.141, 118.770, 118.095, 114.177, 114.067, 113.992, 109.283, 108.727, 97.157, 51.772, 44.985, 12.532; FT-IR (KBr)  $\nu_{\text{max}}$  = 3020 cm<sup>-1</sup> (C–H Ar), 2969 cm<sup>-1</sup> (C–H Aliphatic), 1714 cm<sup>-1</sup> (C=O lactone), 1619 cm<sup>-1</sup> (–O–C=O), 1595 cm<sup>-1</sup> (C=C Ar); UV–Vis (MeOH)  $\lambda_{\text{max}}$  = 415.15 nm; HRMS (ESI m/z) Calcd. for C<sub>24</sub>H<sub>24</sub>N<sub>3</sub>O<sub>4</sub> [M+H]<sup>+</sup> 418.1761, found 418.1770.



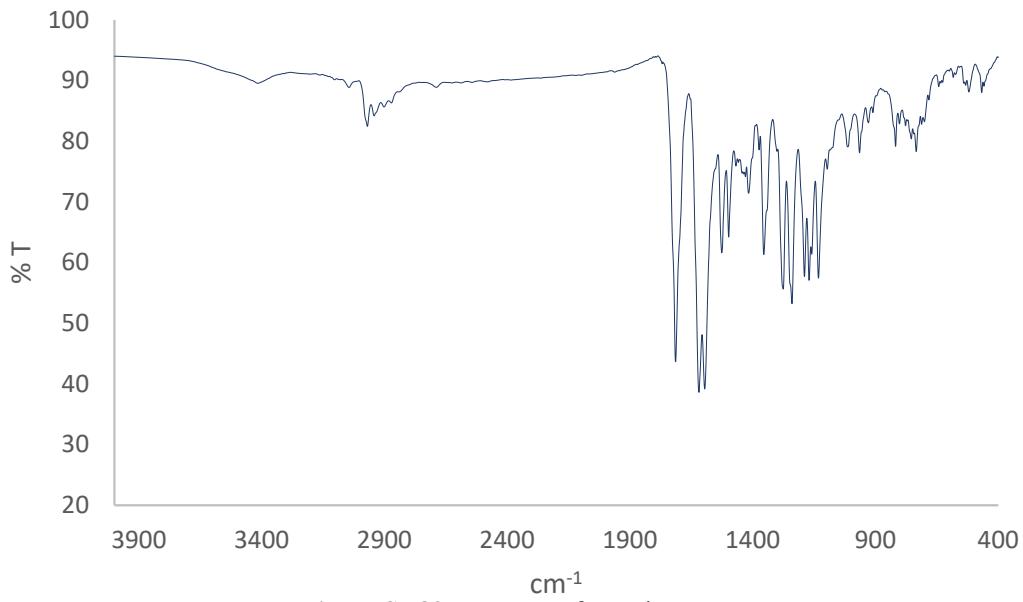
**Figure SI-77.** <sup>1</sup>H NMR spectra of 7a on CDCl<sub>3</sub> 500 MHz



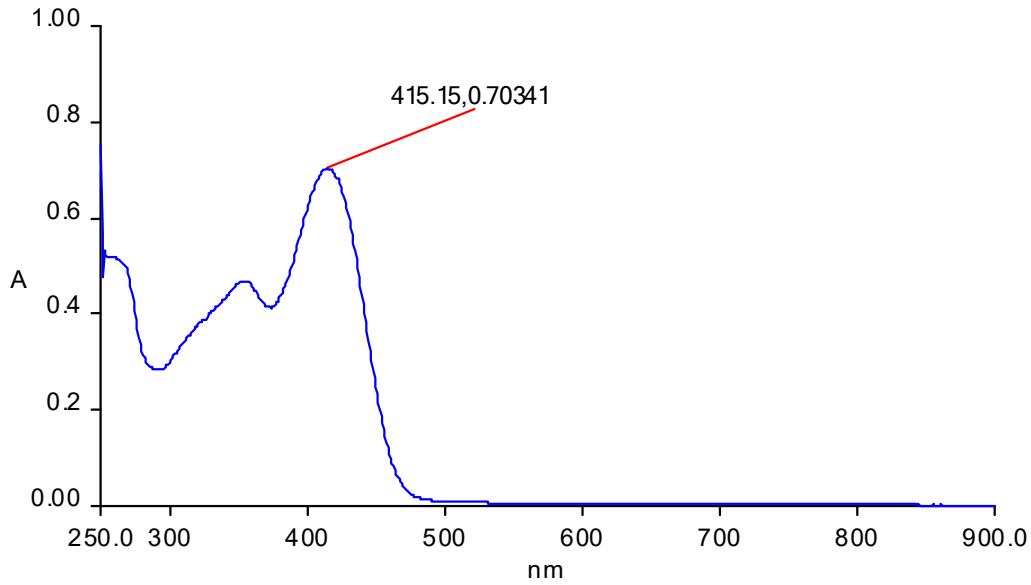
**Figure SI-78.** <sup>13</sup>C NMR spectra of 7a on CDCl<sub>3</sub> 125 MHz.



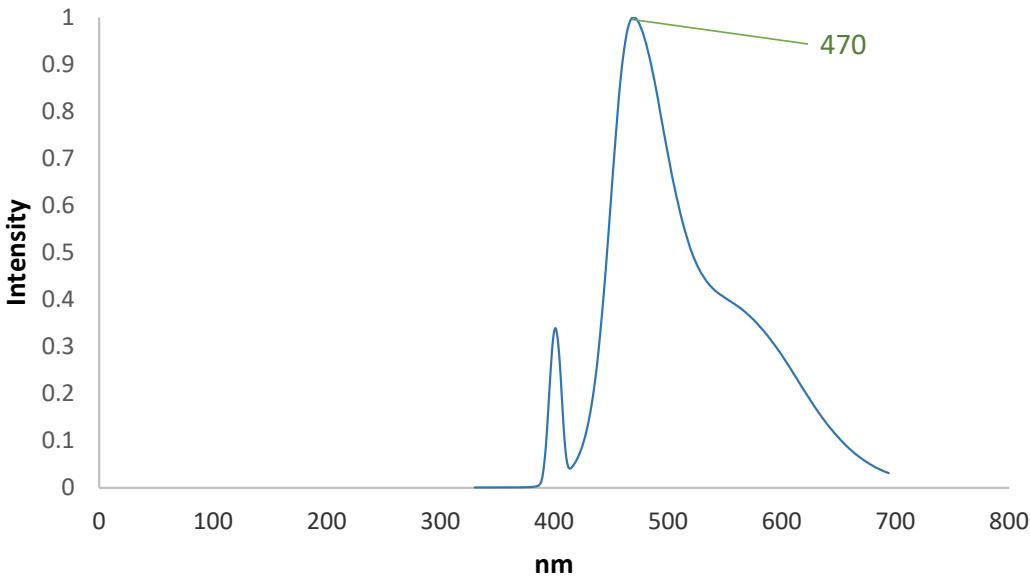
**Figure SI-79.** HMBC NMR spectra of 7a on CDCl<sub>3</sub>



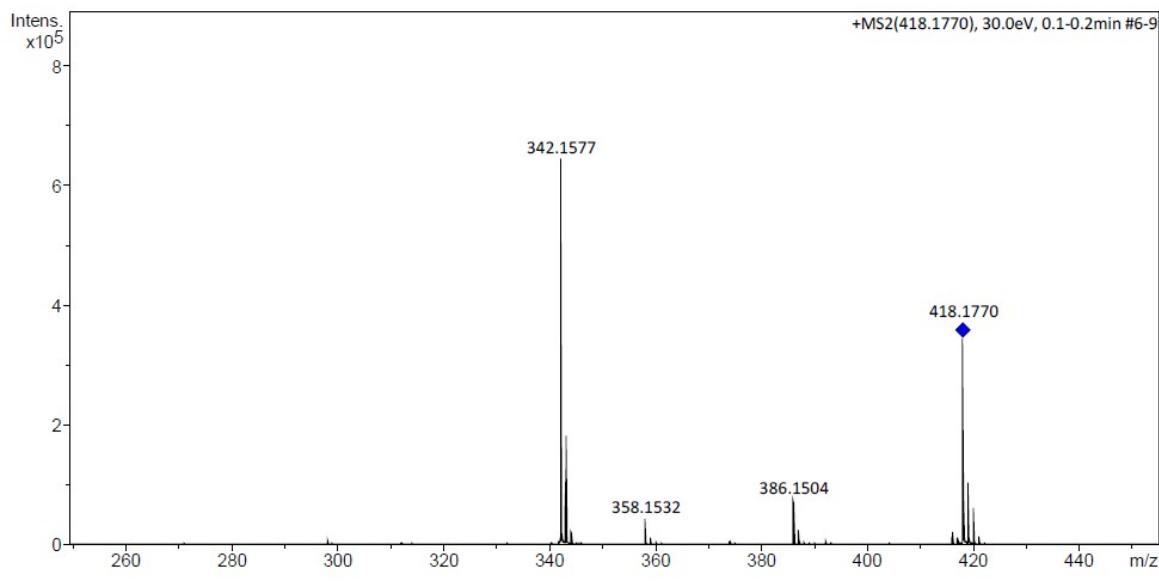
**Figure SI-80.** IR spectra for **7a** in KBr.



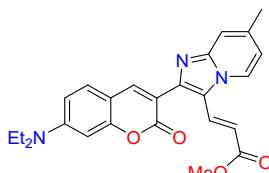
**Figure SI-81.** UV-Vis spectra of **7a** in MeOH.



**Figure SI-82.** Emission spectra of **7a** in MeOH.

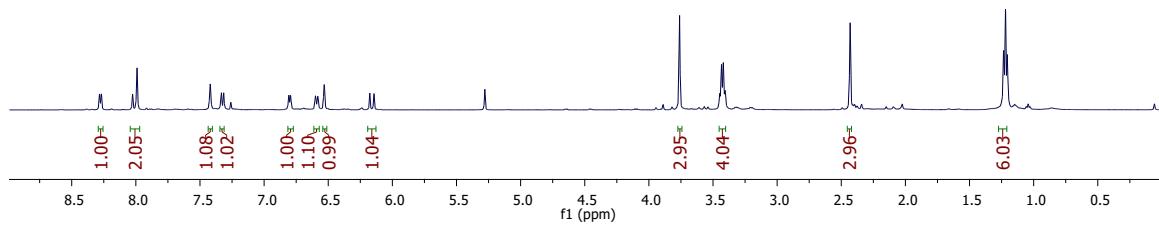


**Figure SI-83.** ESI-MS chromatogram of **7a**.

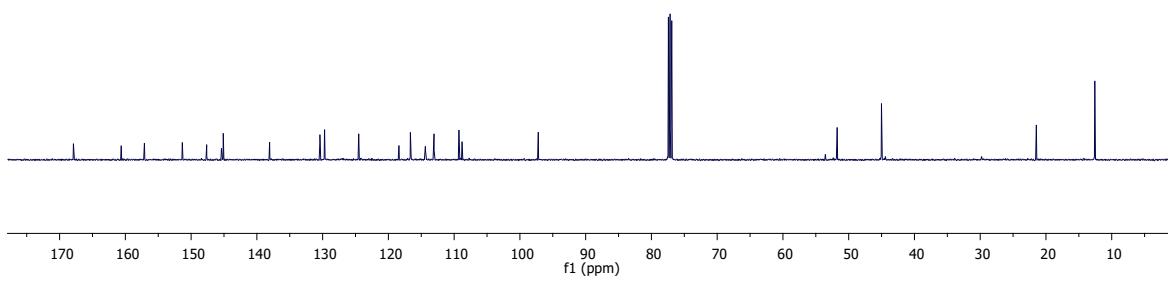


**Methyl (E)-3-(2-(7-(diethylamino)-2-oxo-2H-chromen-3-yl)-7-methylimidazo[1,2-a]pyridin-3-yl)acrylate (7b):** Yield 85%; yellow powder; m.p. = 188 – 190 °C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 8.276 (d, *J* (H,H) = 7 Hz, 1H, Ar), 8.026 – 7.991 (m, 1H, Ar), 7.420 (s, 1H, Ar), 7.323 (d, *J* (H,H) = 8 Hz, 1H, Ar), 8.000 (d, *J* (H,H) = 6.5 Hz, 1H, Ar), 6.600 (d, *J* (H,H) = 9 Hz, 1H, Ar), 6.532 (s, 1H, Ar), 6.159 (d, *J* (H,H) = 16 Hz, 1H, –CH=C–), 3.761 (s, 3H, CH<sub>3</sub>), 3.426 (q, *J* (H,H) = 7 Hz, 4H, –CH<sub>2</sub>–CH<sub>3</sub>), 2.431 (s, 3H, CH<sub>3</sub>), 1.220 (t, *J* (H,H) = 7.5 Hz, 6H, H-7b'); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 167.900, 160.632, 157.099, 151.338, 147.631, 145.387, 145.111, 138.060, 130.413, 129.705,

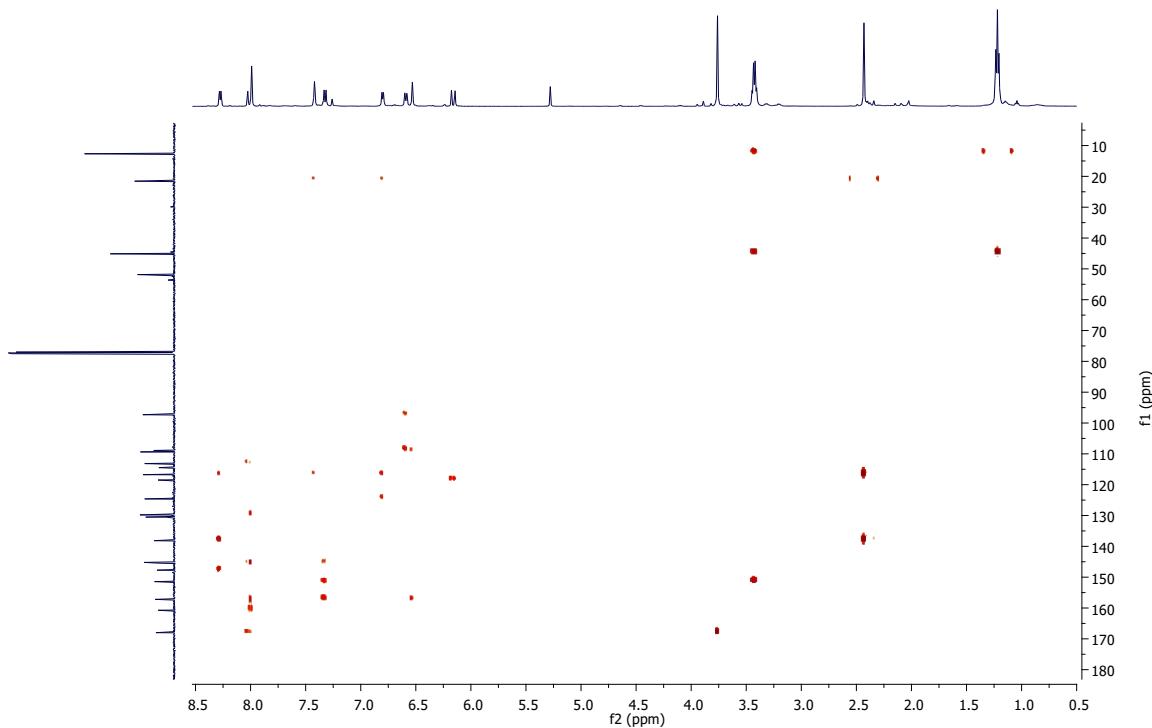
124.513, 118.396, 116.651, 116.575, 114.370, 113.056, 109.274, 108.7996, 97.207, 51.854, 45.119, 21.562, 12.670; FT-IR (KBr)  $\nu_{\text{max}} = 2925 \text{ cm}^{-1}$  (C–H Aliphatic), 1708  $\text{cm}^{-1}$  (C=O lactone), 1619  $\text{cm}^{-1}$  ( $-\text{O}-\text{C}=\text{O}$ ), 1608  $\text{cm}^{-1}$  (C=C Ar); UV–Vis (MeOH)  $\lambda_{\text{max}} = 415.81 \text{ nm}$ ; HRMS (ESI m/z) Calcd. for  $\text{C}_{25}\text{H}_{26}\text{N}_3\text{O}_4$  [M+H]<sup>+</sup> 432.1918, found 432.1919.



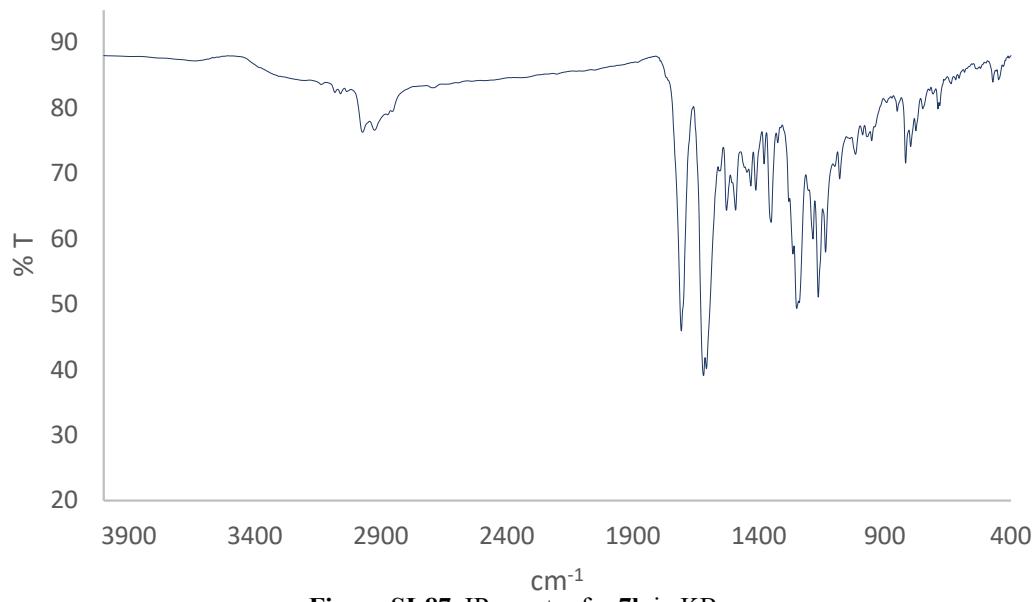
**Figure SI-84.**  $^1\text{H}$  NMR spectra of **7b** on  $\text{CDCl}_3$  500 MHz



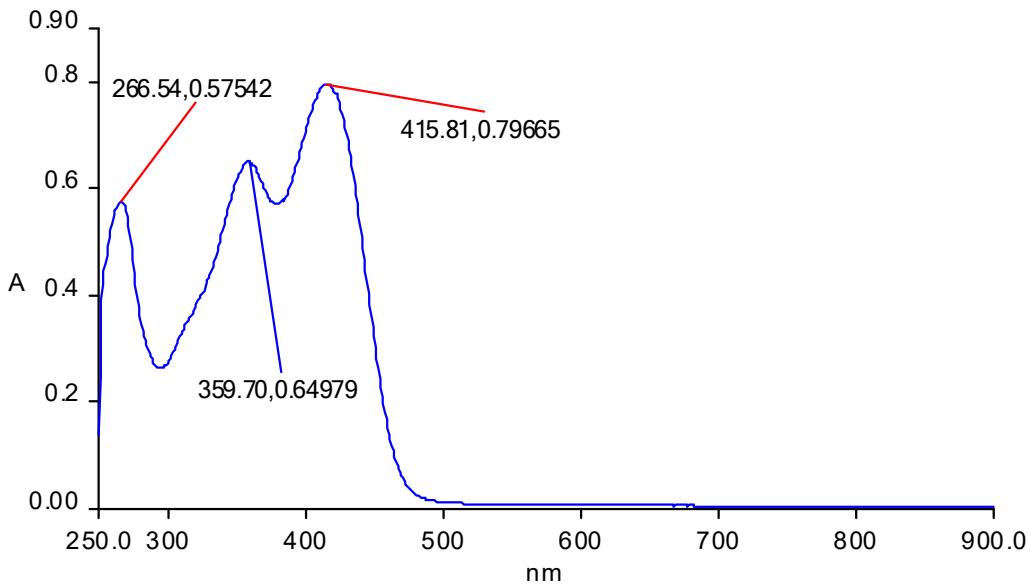
**Figure SI-85.**  $^{13}\text{C}$  NMR spectra of **7b** on  $\text{CDCl}_3$  125 MHz.



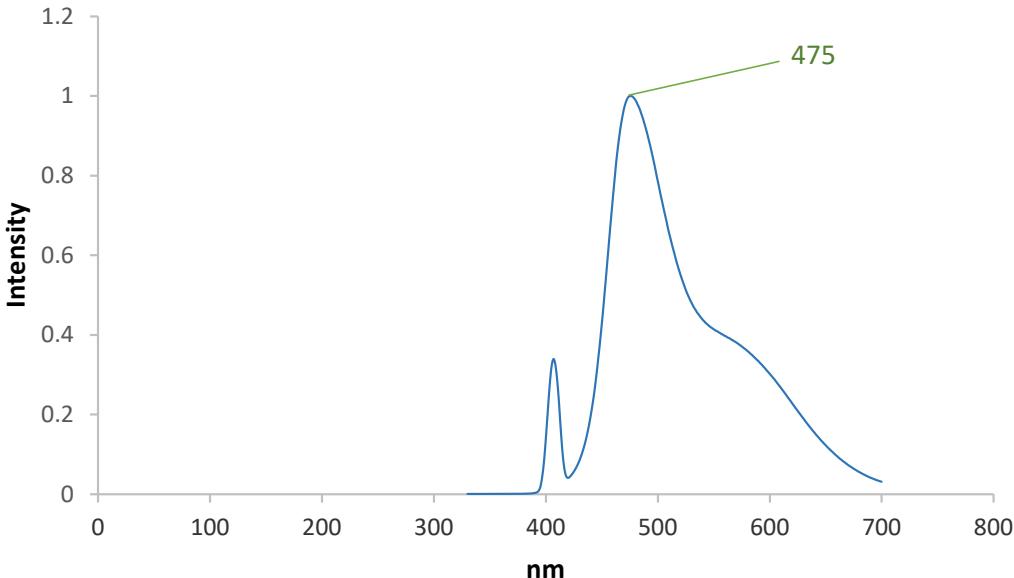
**Figure SI-86.** HMBC NMR spectra of **7b** on  $\text{CDCl}_3$



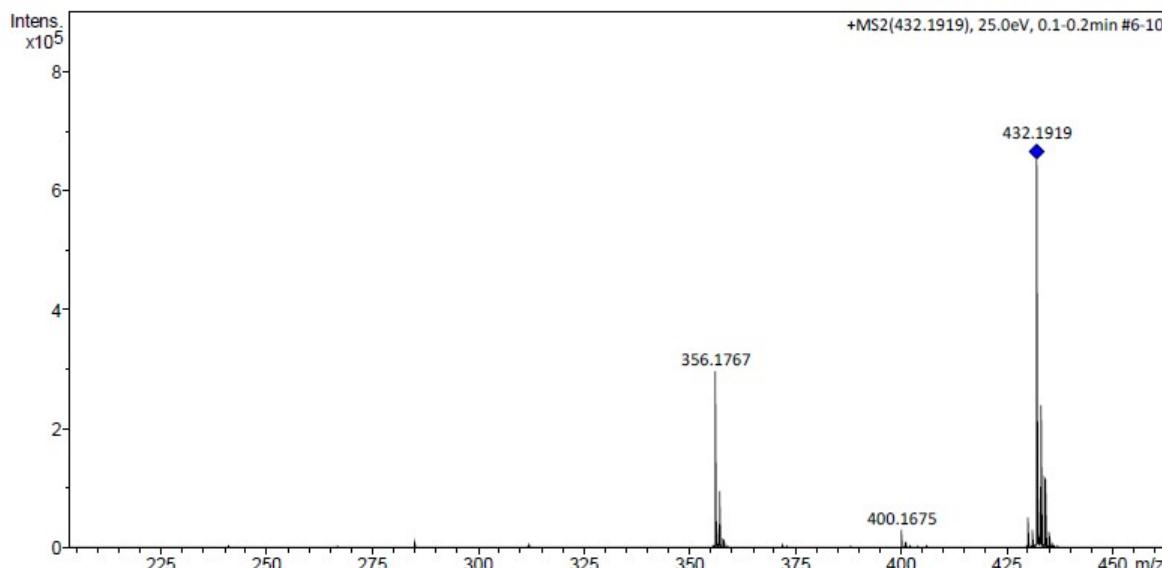
**Figure SI-87.** IR spectra for **7b** in KBr.



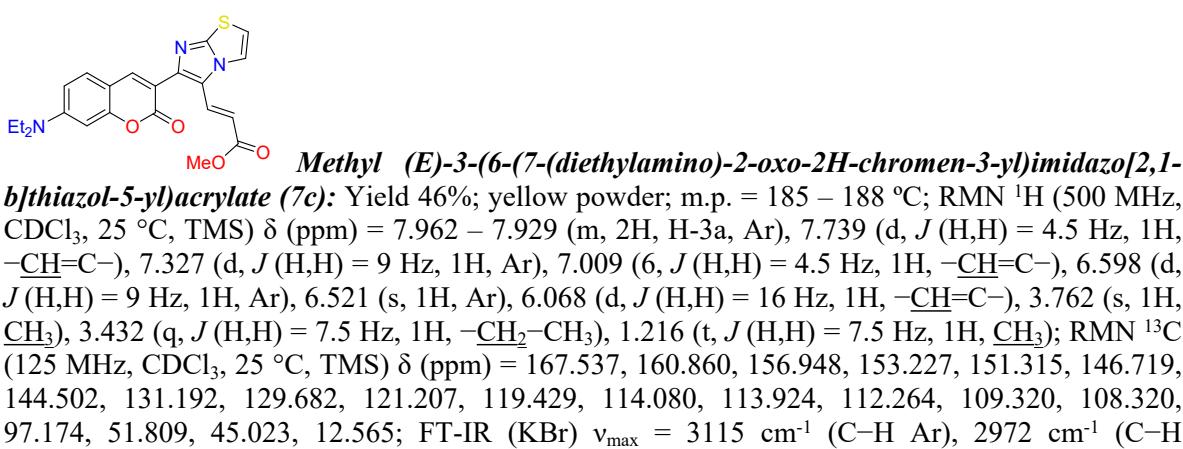
**Figure SI-88.** UV-Vis spectra of **7a** in MeOH.



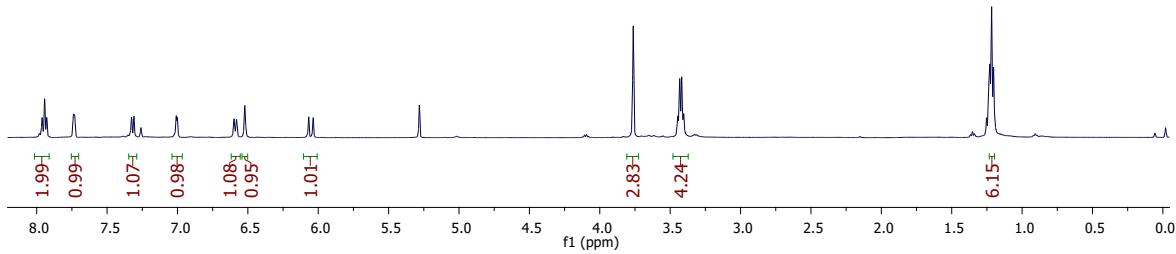
**Figure SI-89.** Emission spectra of **7b** in MeOH.



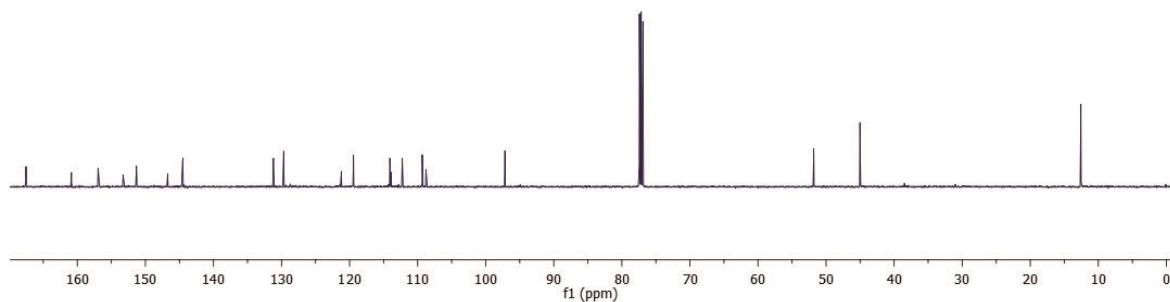
**Figure SI-90.** ESI-MS chromatogram of **7a**.



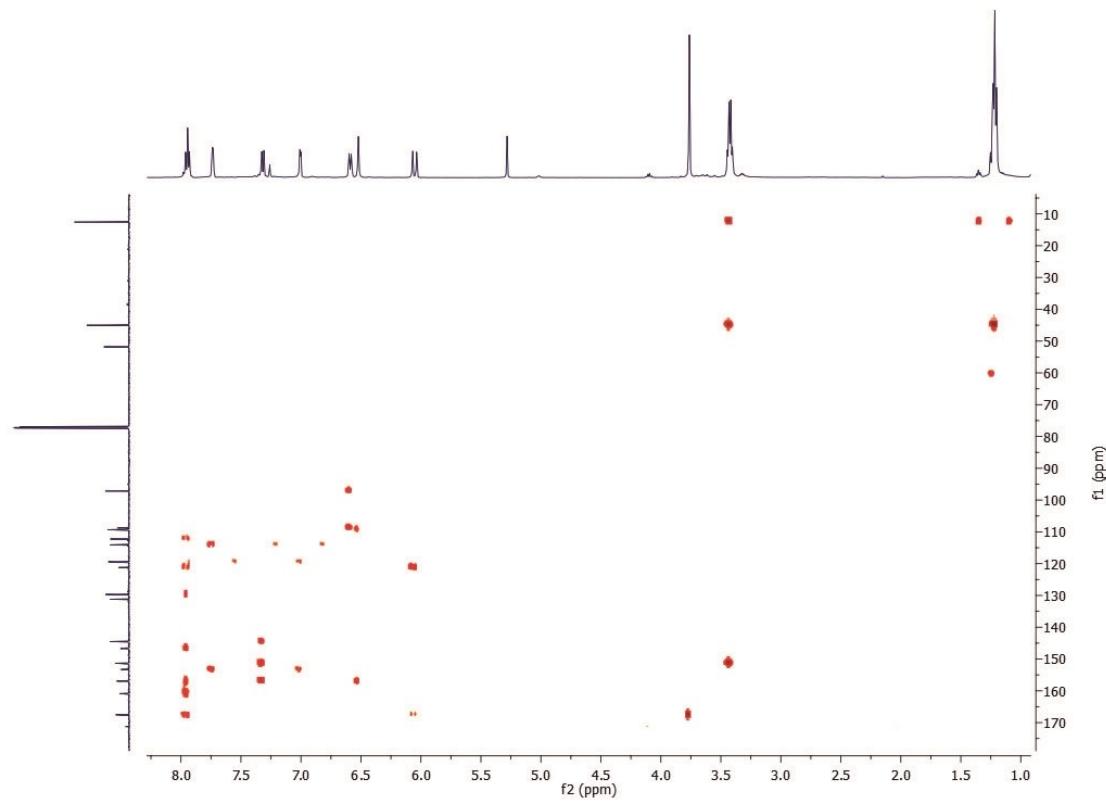
Aliphatic), 1709 cm<sup>-1</sup> (C=O lactone), 1617 cm<sup>-1</sup> (-O-C=O), 1595 cm<sup>-1</sup> (C=C Ar); UV-Vis (MeOH)  $\lambda_{\text{max}} = 416.21$  nm; HRMS (ESI m/z) Calcd. for C<sub>22</sub>H<sub>22</sub>N<sub>3</sub>O<sub>4</sub>S [M+H]<sup>+</sup> 424.1326, found 424.1332.



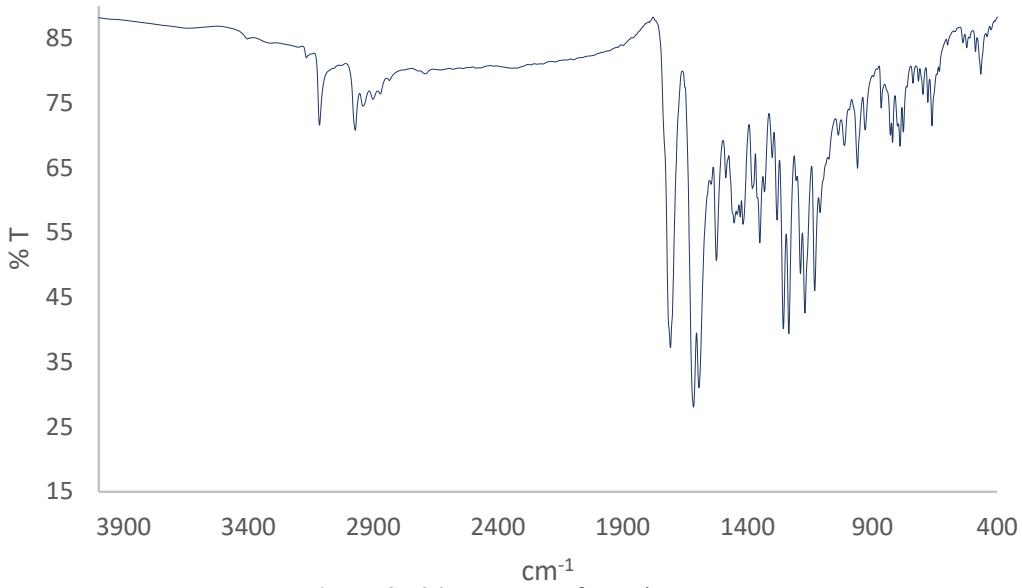
**Figure SI-91.** <sup>1</sup>H NMR spectra of **7c** on CDCl<sub>3</sub> 500 MHz



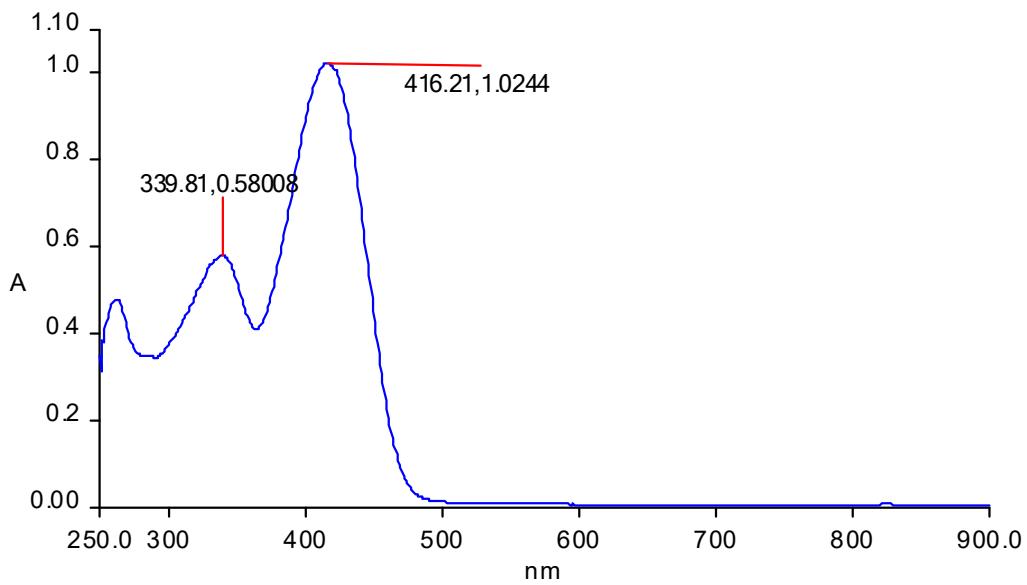
**Figure SI-92.**  $^{13}\text{C}$  NMR spectra of  $7\text{c}$  on  $\text{CDCl}_3$  125 MHz.



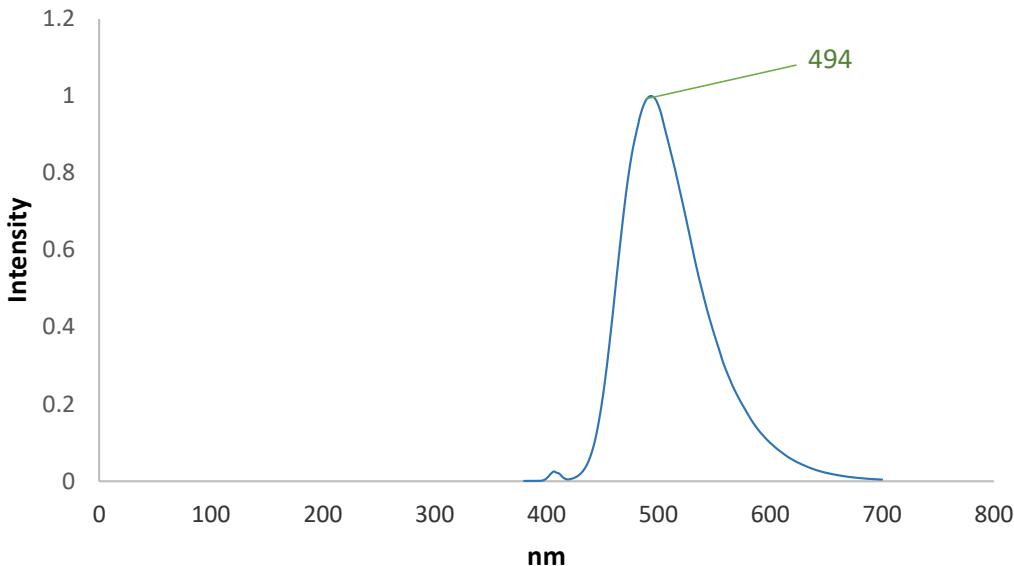
**Figure SI-93.** HMBC NMR spectra of  $7\text{c}$  on  $\text{CDCl}_3$



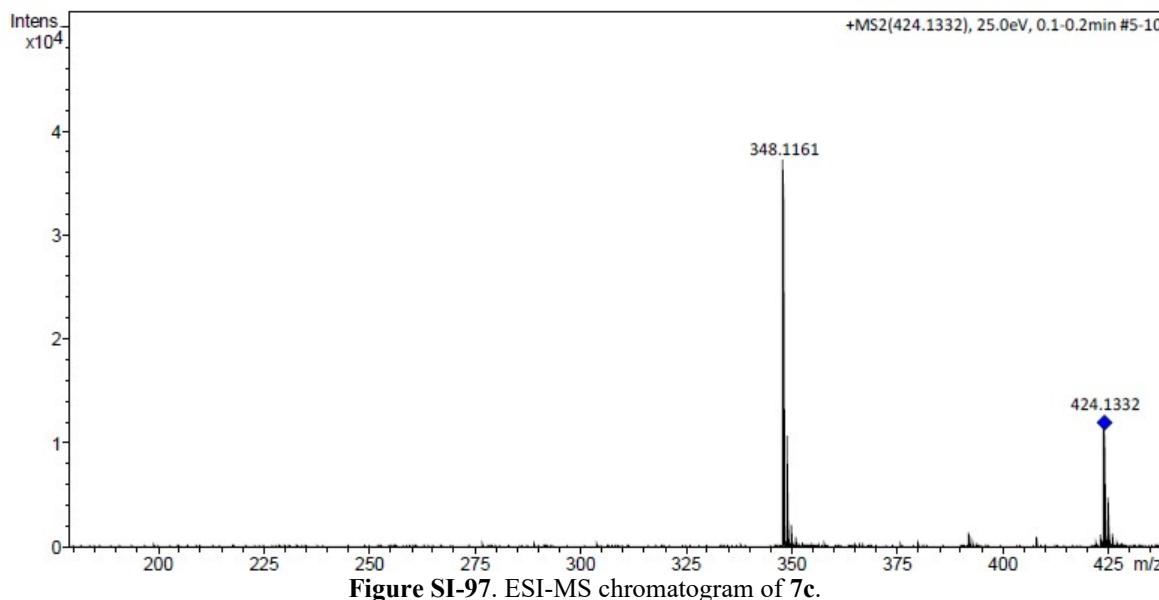
**Figure SI-94.** IR spectra for **7c** in KBr.



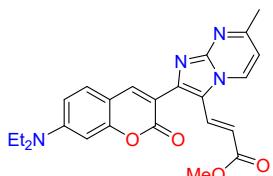
**Figure SI-95.** UV-Vis spectra of **7c** in MeOH.



**Figure SI-96.** Emission spectra of **7c** in MeOH.

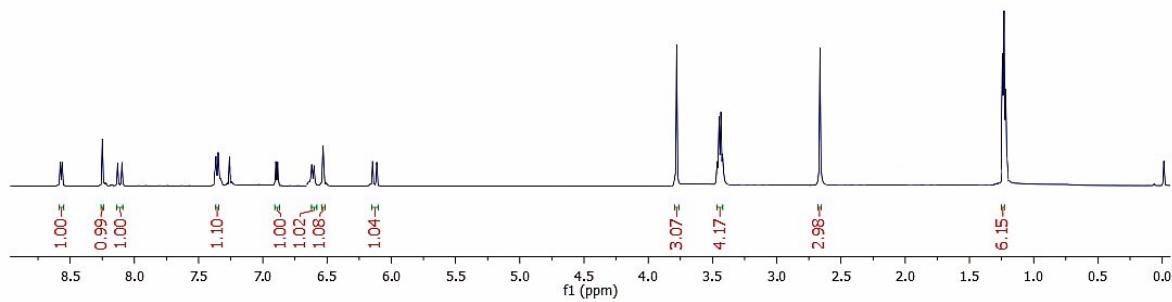


**Figure SI-97.** ESI-MS chromatogram of **7c**.

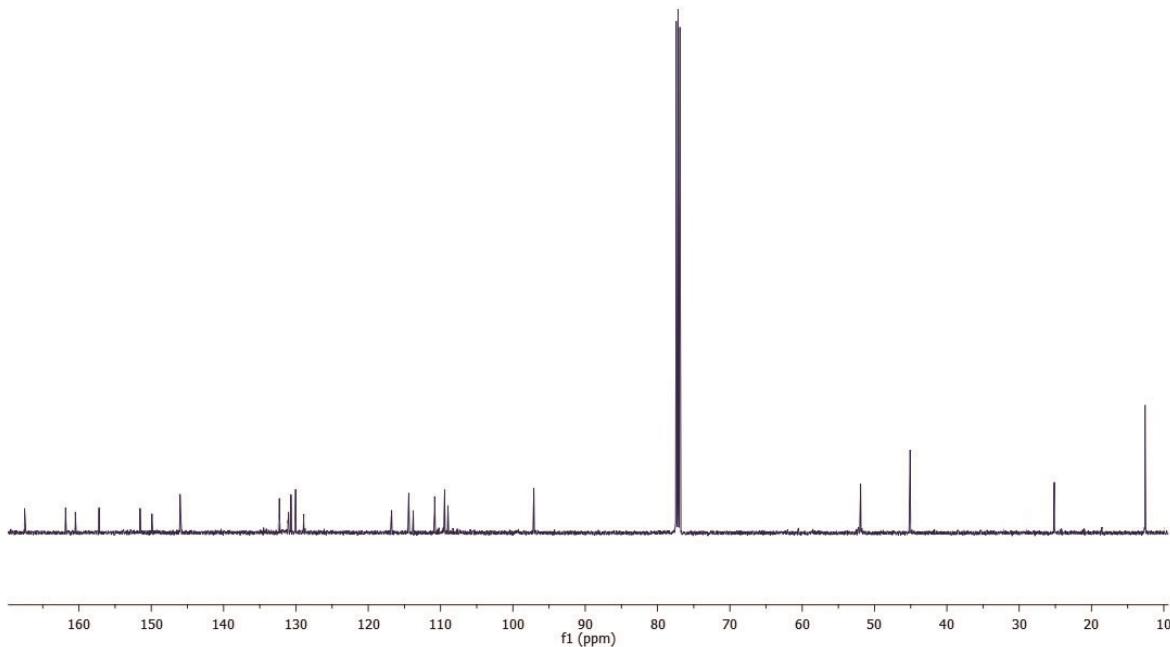


**Methyl (E)-3-(2-(7-(diethylamino)-2-oxo-2H-chromen-3-yl)-7-methylimidazo[1,2-a]pyrimidin-3-yl)acrylate (7d):** Yield 65%; yellow powder; m.p. = 229 – 231 °C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 8.567 (d, *J* (H,H) = 6 Hz, 1H, Ar), 8.249 (s, 1H, –CH=C–), 8.113 (d, *J* (H,H) = 16.5 Hz, 1H, –CH=C–), 7.357 (d, *J* (H,H) = 8.5 Hz, 1H, Ar), 6.892 (t, *J* (H,H) = 7 Hz, 1H, Ar), 6.620 (d, *J* (H,H) = 9 Hz, 1H, Ar), 6.531 (s, 1H, Ar), 6.129 (d, *J* (H,H) = 16.5 Hz, 1H, –CH=C–), 3.779 (s, 3H, CH<sub>3</sub>), 3.442 (q, *J* (H,H) = 7 Hz, 4H, –CH<sub>2</sub>–CH<sub>3</sub>), 2.663 (s, 3H, CH<sub>3</sub>), 1.231 (t, *J* (H,H) = 7 Hz, 6H, CH<sub>3</sub>); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 167.511, 161.84, 160.494, 157.221, 151.546, 149.904, 146.012, 145.938, 132.288, 130.688, 130.044, 116.767, 114.395, 113.772, 110.816, 109.434, 108.975, 97.105, 51.922, 45.082,

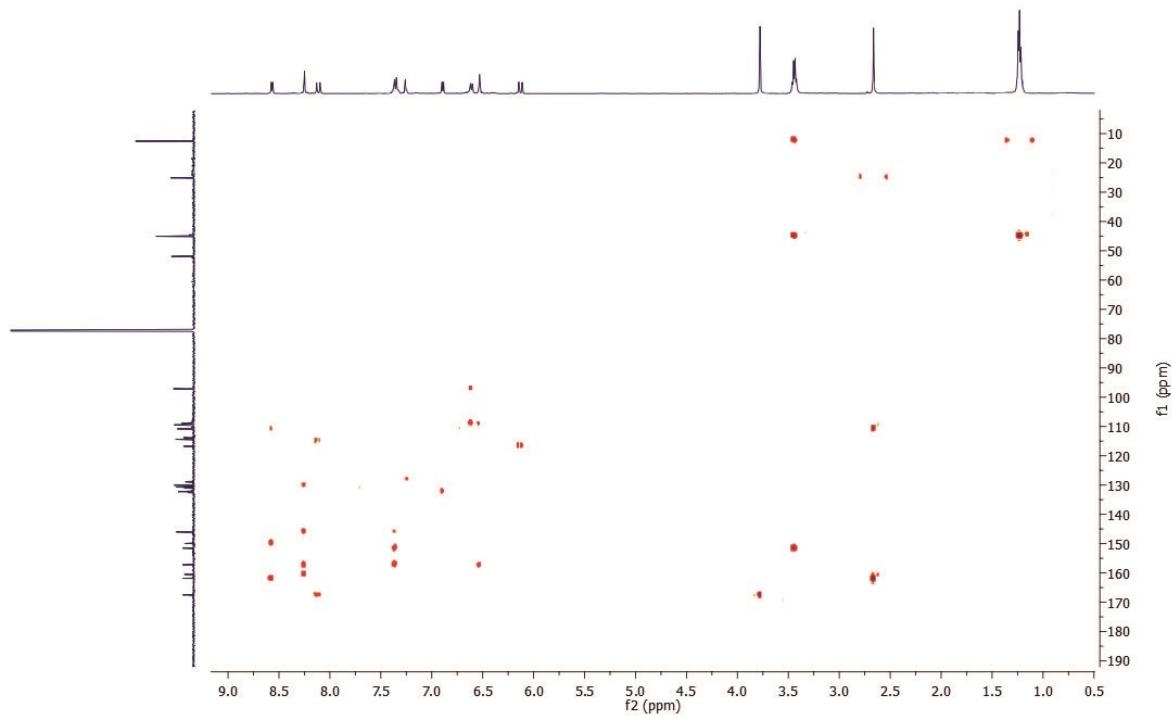
25.145, 12.593; FT-IR (KBr)  $\nu_{\text{max}} = 3046 \text{ cm}^{-1}$  (C–H Ar), 2963  $\text{cm}^{-1}$  (C–H Aliphatic), 1719  $\text{cm}^{-1}$  (C=O lactone), 1621  $\text{cm}^{-1}$  ( $-\text{O}-\text{C}=\text{O}$ ), 1593  $\text{cm}^{-1}$  (C=C Ar); UV–Vis (MeOH)  $\lambda_{\text{max}} = 423.07 \text{ nm}$ ; HRMS (ESI m/z) Calcd. for  $\text{C}_{24}\text{H}_{25}\text{N}_4\text{O}_4$  [ $\text{M}+\text{H}]^+$  433.1870, found 433.1872.



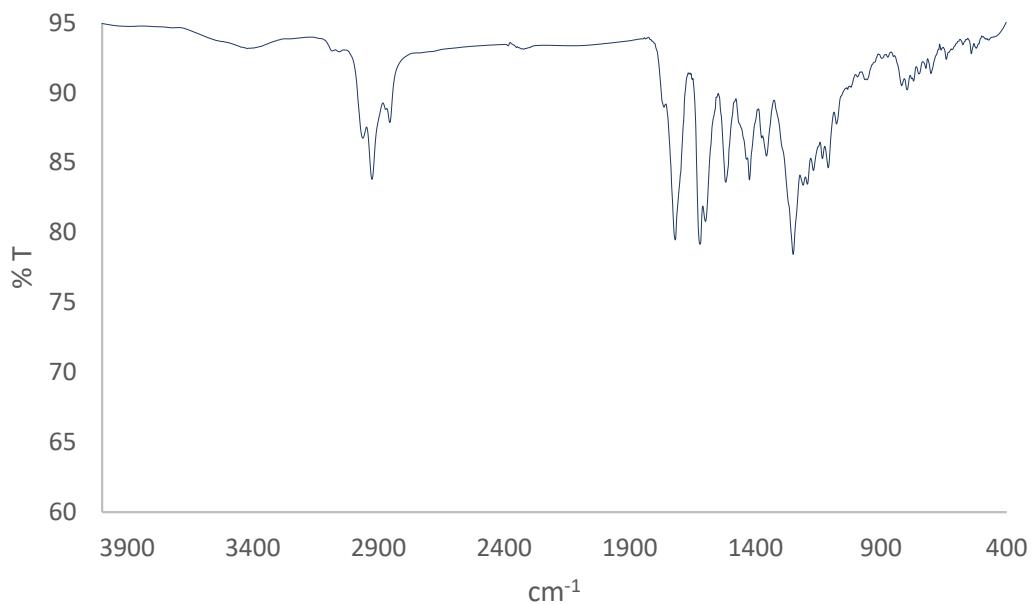
**Figure SI-98.**  $^1\text{H}$  NMR spectra of **7d** on  $\text{CDCl}_3$  500 MHz



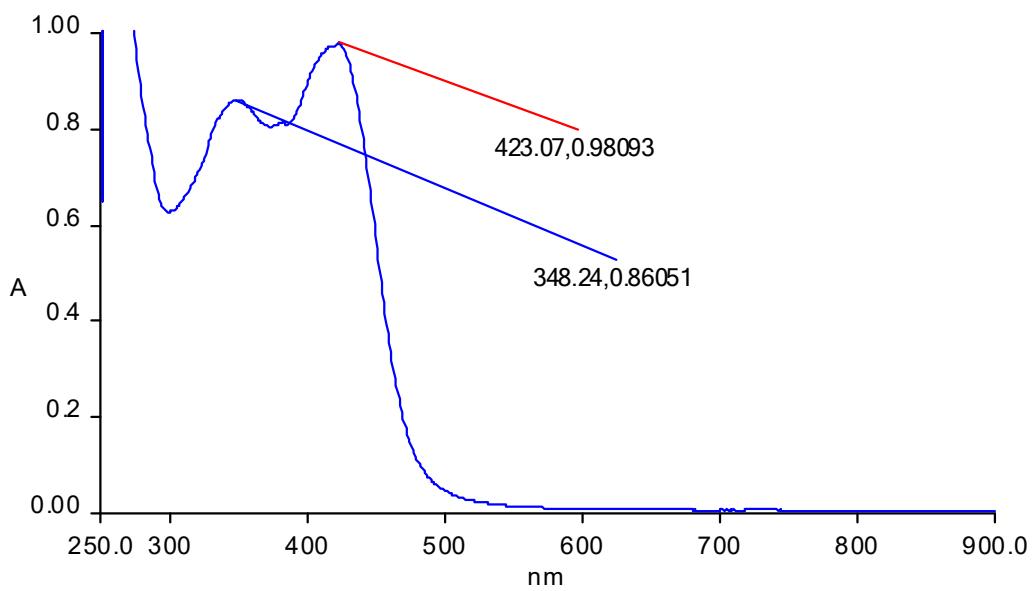
**Figure SI-99.**  $^{13}\text{C}$  NMR spectra of **7d** on  $\text{CDCl}_3$  125 MHz.



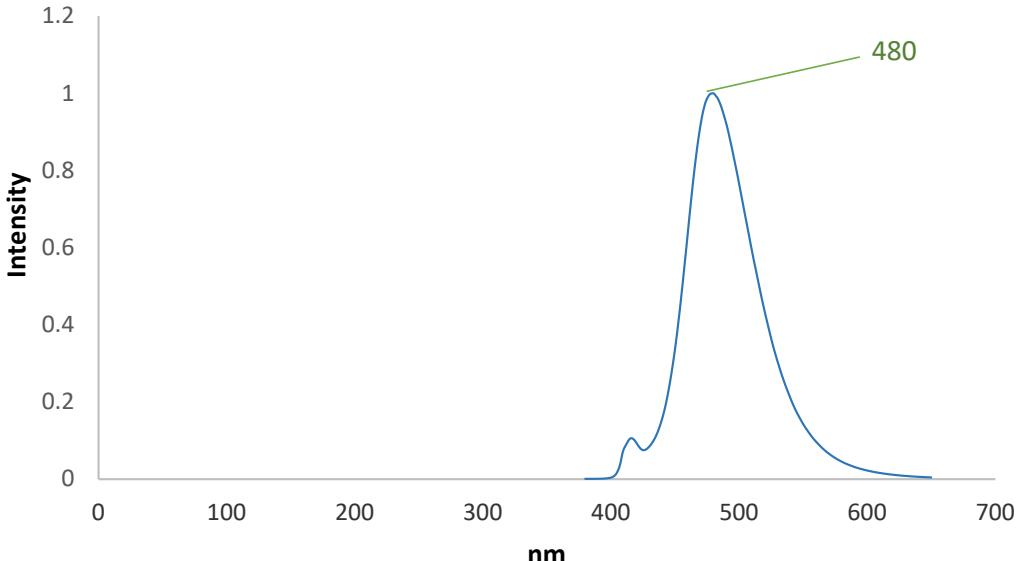
**Figure SI-100.** HMBC NMR spectra of **7d** on  $\text{CDCl}_3$



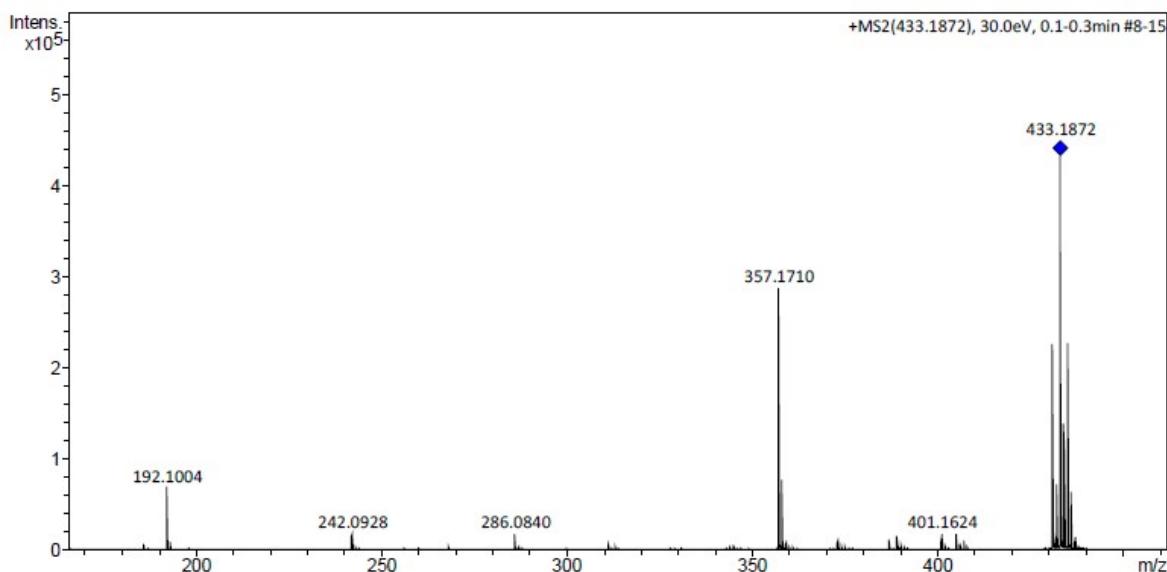
**Figure SI-101.** IR spectra for **7d** in KBr.



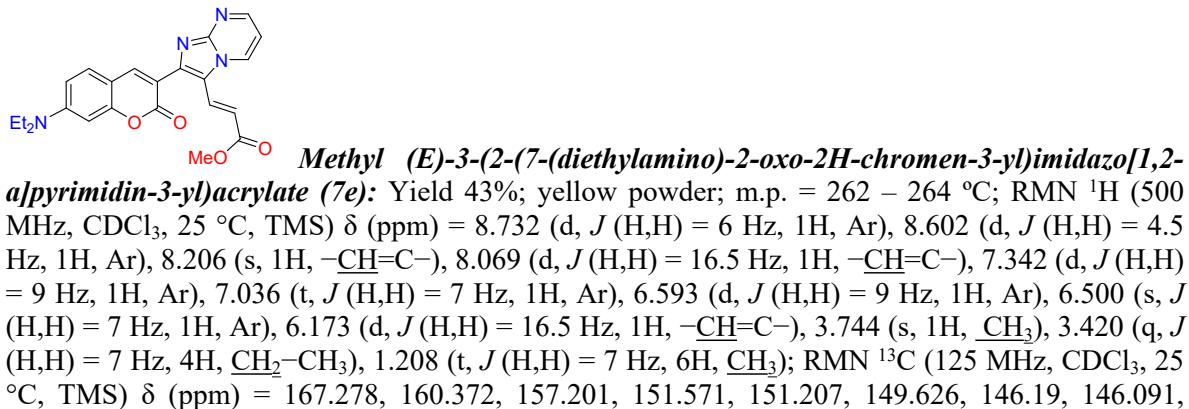
**Figure SI-102.** UV-Vis spectra of **7d** in MeOH.



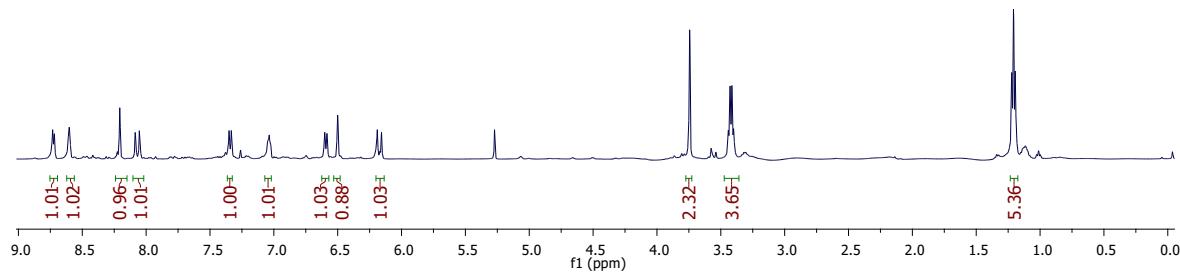
**Figure SI-103.** Emission spectra of **7d** in MeOH.



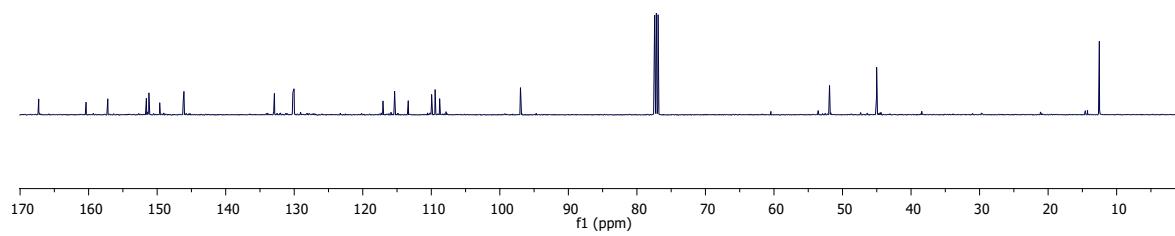
**Figure SI-104.** ESI-MS chromatogram of **7d**.



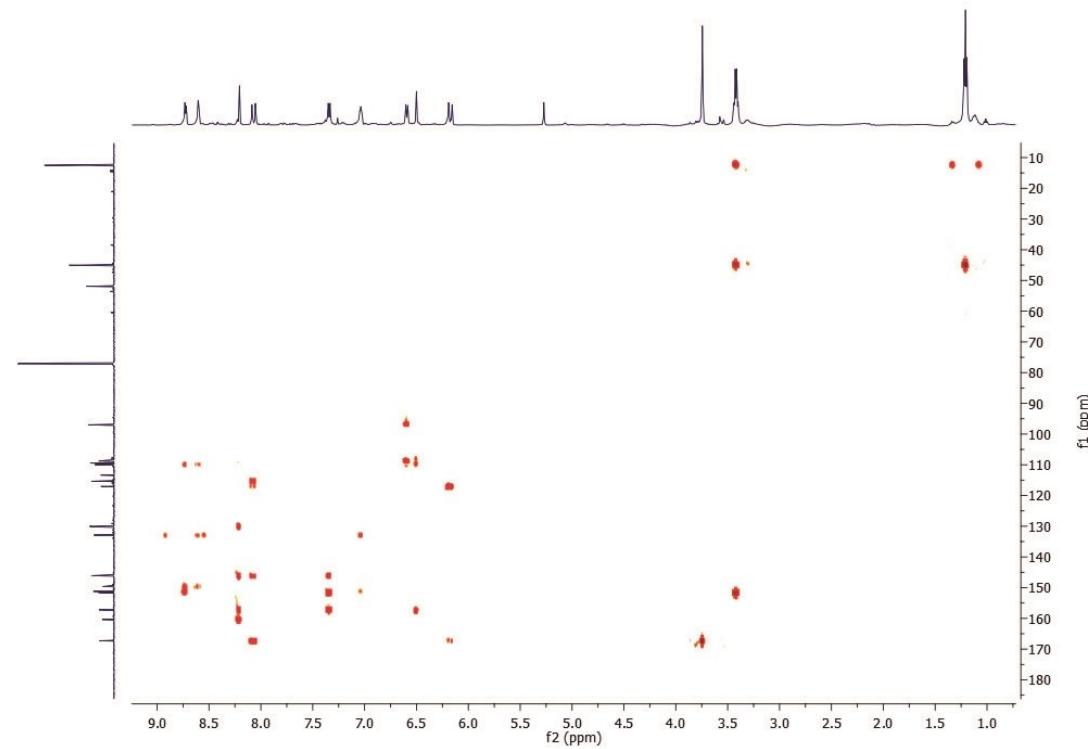
132.891, 130.199, 130.019, 117.050, 115.356, 113.361, 109.953, 109.441, 108.783, 96.988, 51.895, 45.023, 12.525; FT-IR (KBr)  $\nu_{\text{max}}$  = 3067 cm<sup>-1</sup> (C–H Ar), 2968 cm<sup>-1</sup> (C–H Aliphatic), 1711 cm<sup>-1</sup> (C=O lactone), 1616 cm<sup>-1</sup> (–O–C=O), 1588 cm<sup>-1</sup> (C=C Ar); UV–Vis (MeOH)  $\lambda_{\text{max}}$  = 422.98 nm; HRMS (ESI m/z) Calcd. for C<sub>23</sub>H<sub>23</sub>N<sub>4</sub>O<sub>4</sub> [M+H]<sup>+</sup> 419.1714, found 419.1625.



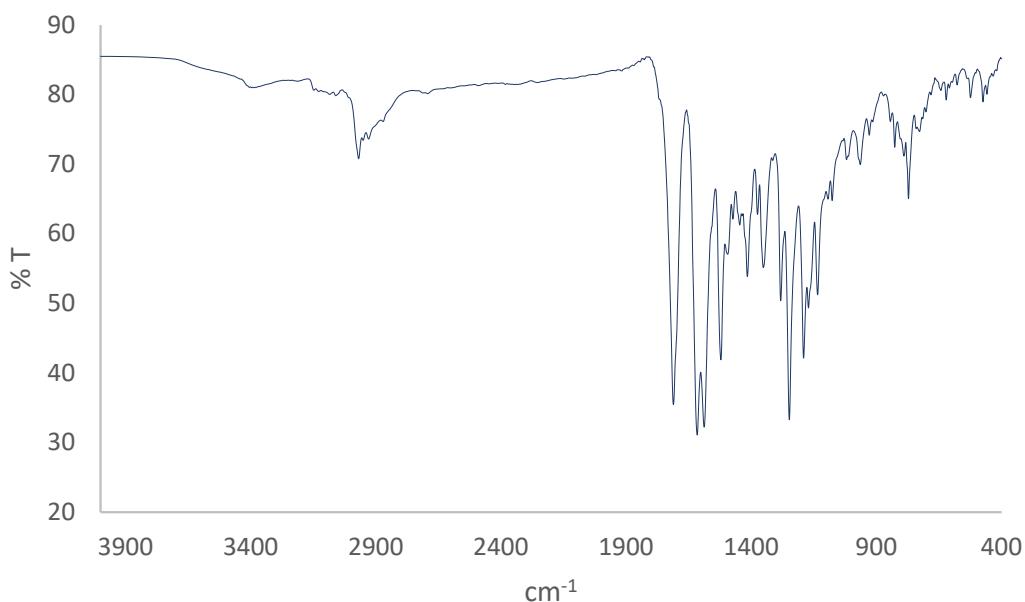
**Figure SI-105.** <sup>1</sup>H NMR spectra of 7e on CDCl<sub>3</sub> 500 MHz



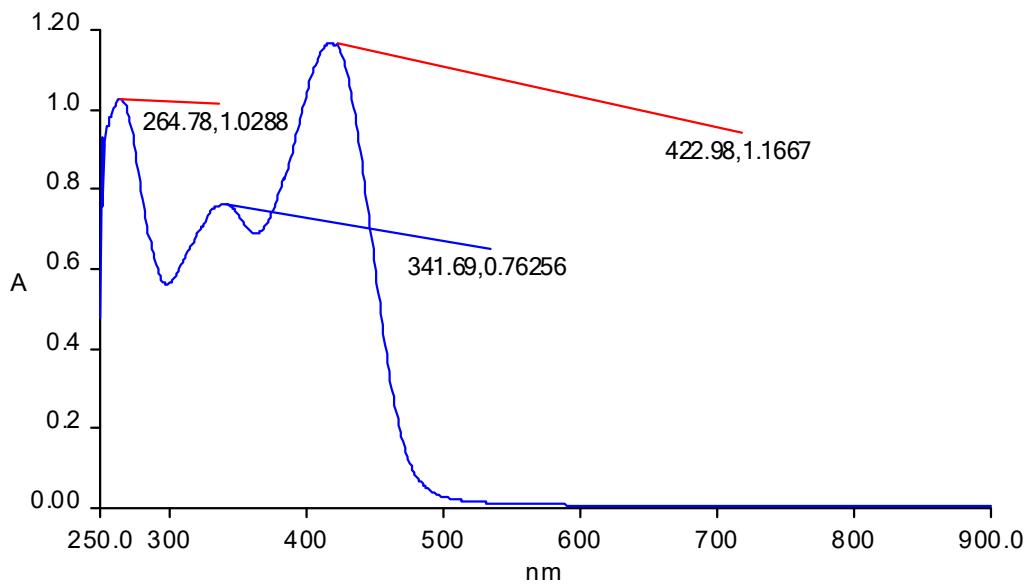
**Figure SI-106.**  $^{13}\text{C}$  NMR spectra of **7e** on  $\text{CDCl}_3$  125 MHz.



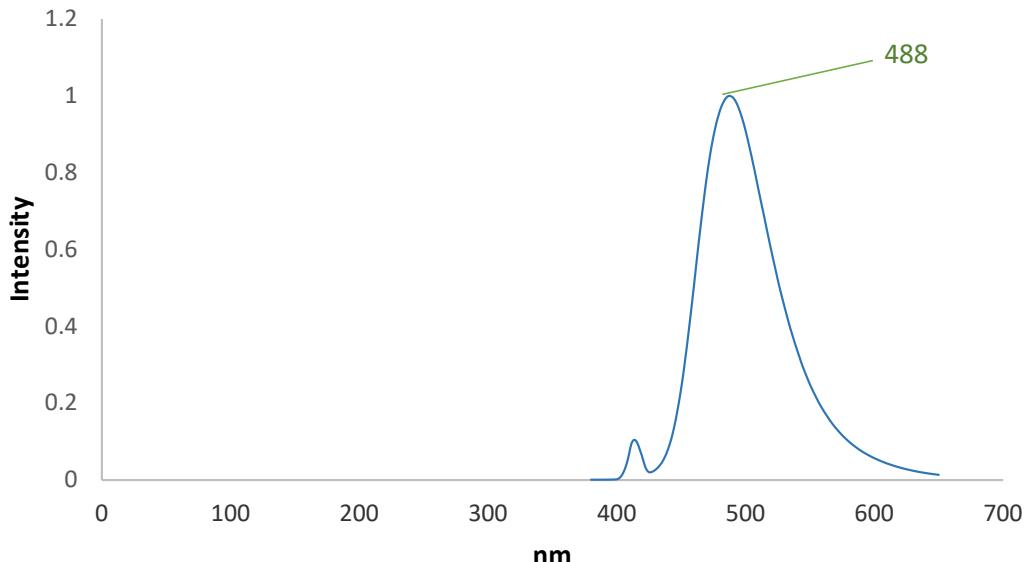
**Figure SI-107.** HMBC NMR spectra of **7e** on  $\text{CDCl}_3$



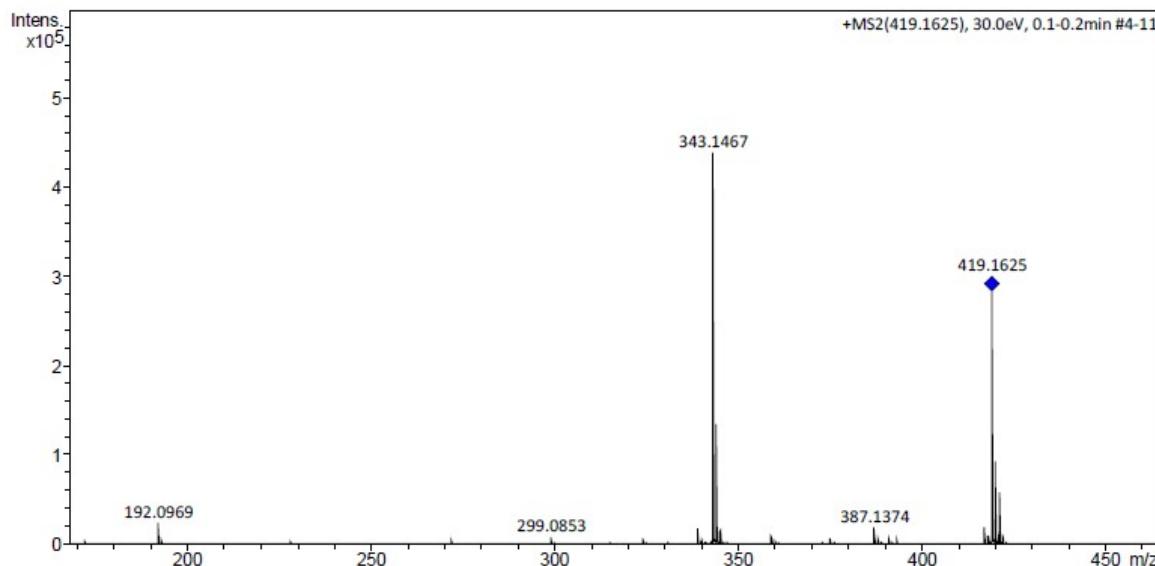
**Figure 108.** IR spectra for **7e** in KBr.



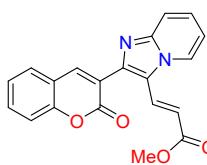
**Figure SI-109.** UV-Vis spectra of **7e** in MeOH.



**Figure SI-110.** Emission spectra of **7e** in MeOH.

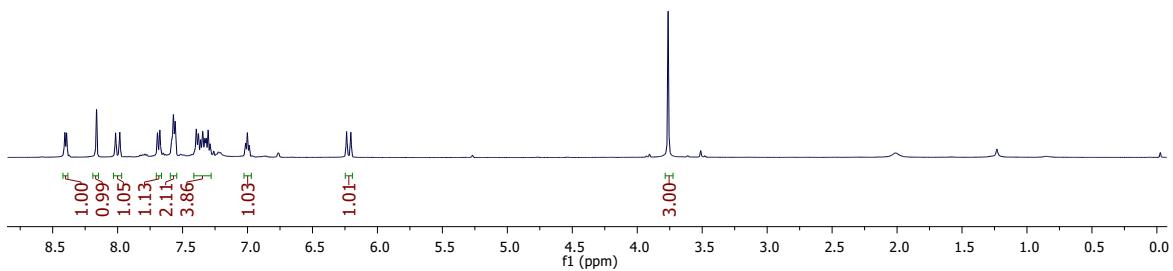


**Figure SI-111.** ESI-MS chromatogram of **7e**.

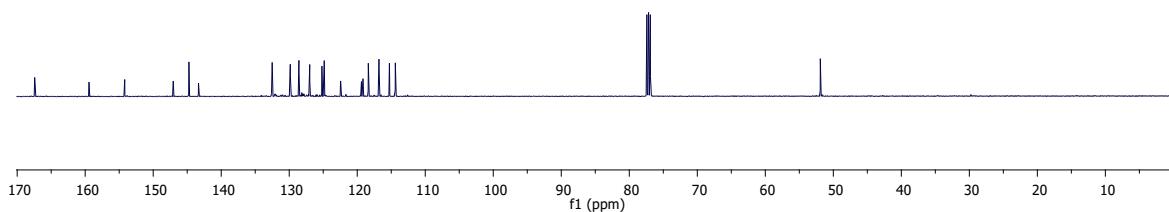


**Methyl (E)-3-(2-(2-oxo-2H-chromen-3-yl)imidazo[1,2-a]pyridin-3-yl)acrylate (7f):** Yield 81%; light brown powder; m.p. = 208 – 210 °C; RMN <sup>1</sup>H (500 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 8.399 (d, *J* (H,H) = 7 Hz, 1H, Ar), 8.162 (s, 1H, –CH=C–), 7.999 (d, *J* (H,H) = 16 Hz, 1H, –CH=C–), 7.684 (d, *J* (H,H) = 9 Hz, 1H, Ar), 7.571 (m, 2H, H-7', Ar), 7.395 – 7.303 (m, 3H, Ar), 7.001 (t, *J* (H,H) = 7 Hz, 1H, Ar), 6.221 (d, *J* (H,H) = 16 Hz, 1H, –CH=C–), 3.762 (s, 1H, CH<sub>3</sub>); RMN <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>, 25 °C, TMS) δ (ppm) = 167.425, 159.441, 154.189, 147.052, 144.749, 143.339, 132.505, 129.849, 128.572, 126.983, 125.198, 124.850, 122.441, 119.378, 119.153, 118.373, 116.802, 115.267, 114.399, 51.909; FT-IR (KBr) ν<sub>max</sub> = 3071 cm<sup>-1</sup> (C=H Ar), 2940 cm<sup>-1</sup> (C=H Aliphatic), 1724 cm<sup>-1</sup> (C=O lactone), 1705 cm<sup>-1</sup> (–O–C=O), 1623 cm<sup>-1</sup> (C=C Ar);

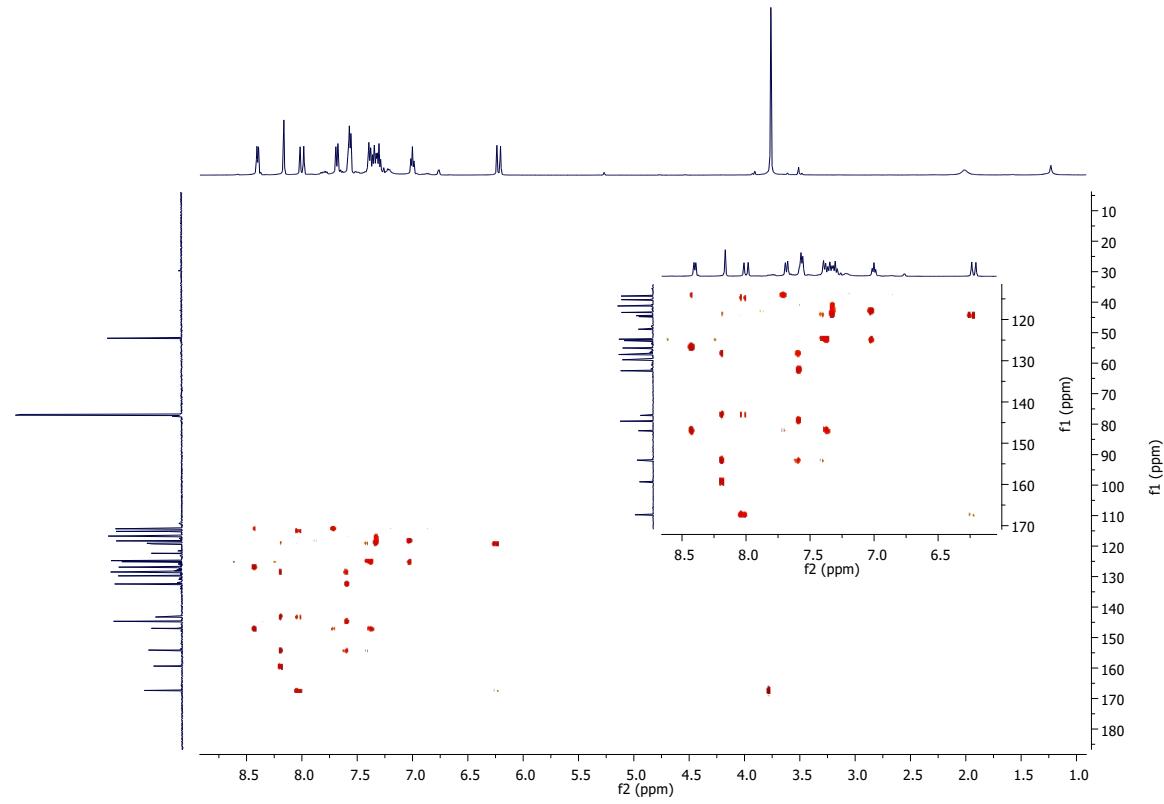
UV–Vis (MeOH)  $\lambda_{\text{max}} = 330.17$  nm; HRMS (ESI m/z) Calcd. for  $\text{C}_{20}\text{H}_{15}\text{N}_2\text{O}_4$  [M+H]<sup>+</sup> 347.1026, found 347.1032.



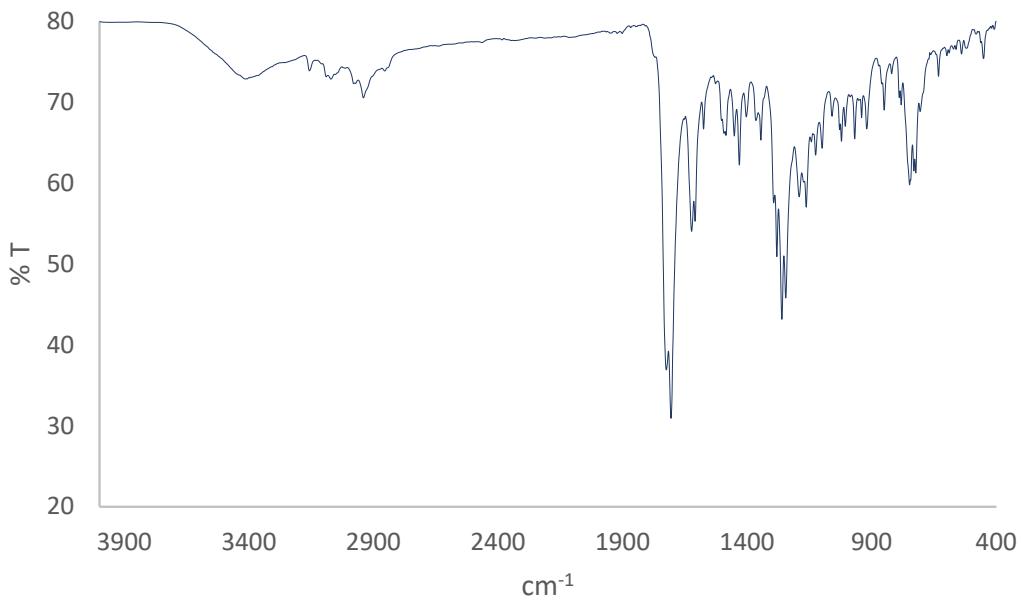
**Figure SI-112.** <sup>1</sup>H NMR spectra of 7f on  $\text{CDCl}_3$  500 MHz



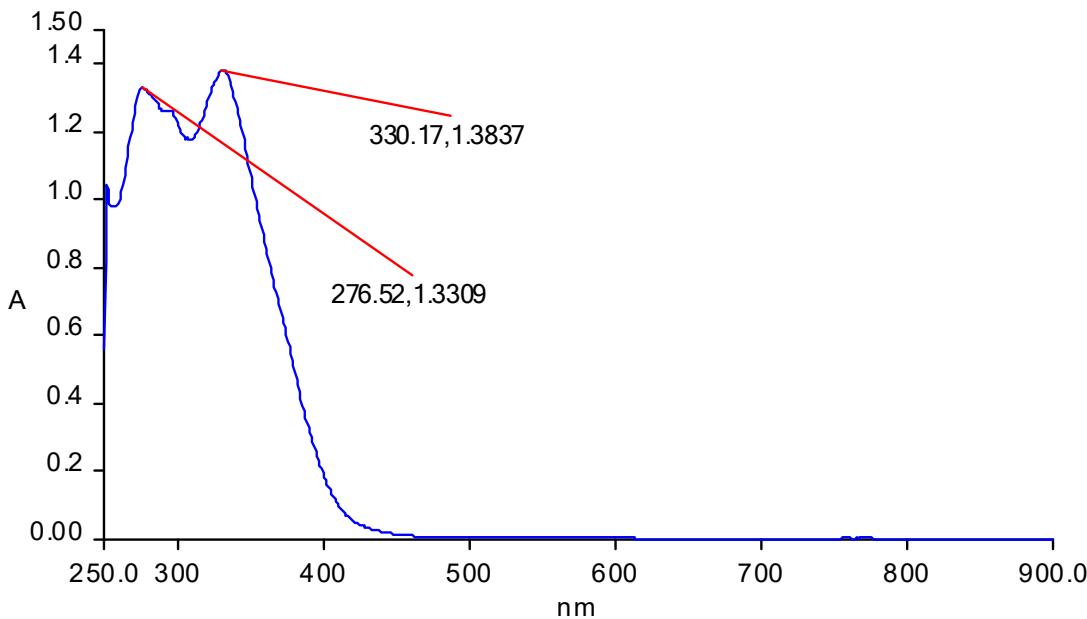
**Figure SI-113.**  $^{13}\text{C}$  NMR spectra of **7f** on  $\text{CDCl}_3$  125 MHz.



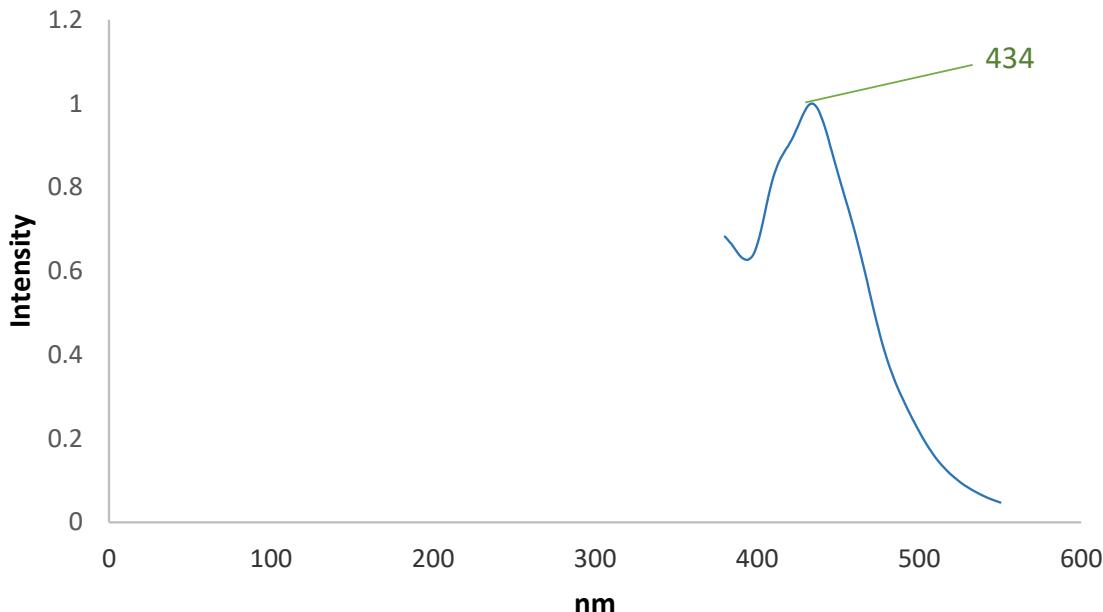
**Figure SI-114.** HMBC NMR spectra of **7f** on  $\text{CDCl}_3$



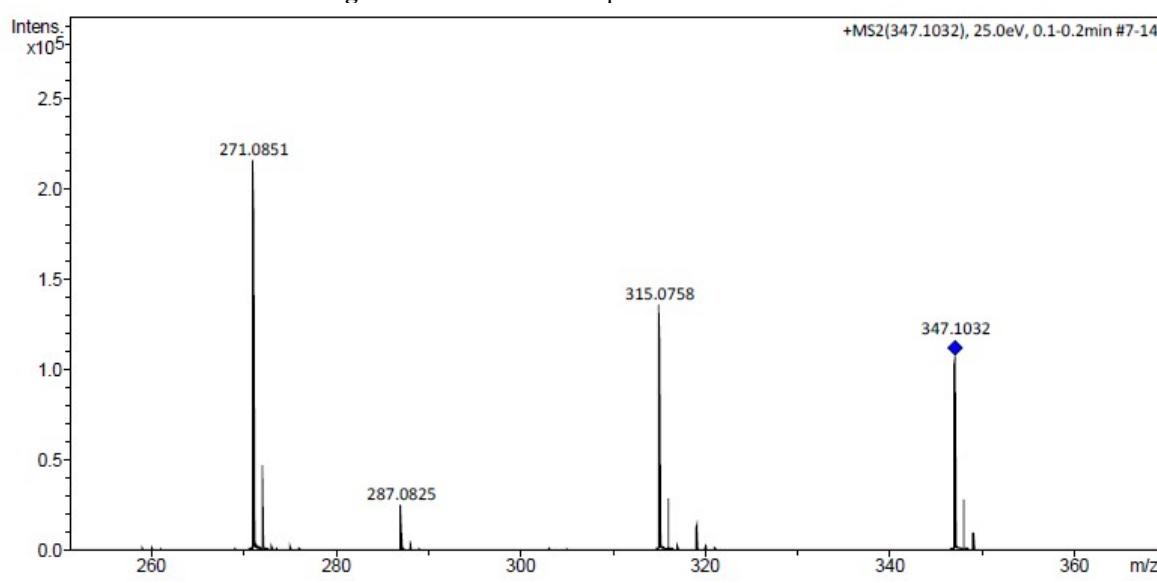
**Figure SI-115.** IR spectra for **7f** in KBr.



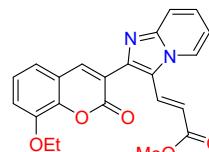
**Figure SI-116.** UV-Vis spectra of **7f** in MeOH.



**Figure SI-117.** Emission spectra of **7f** in MeOH.

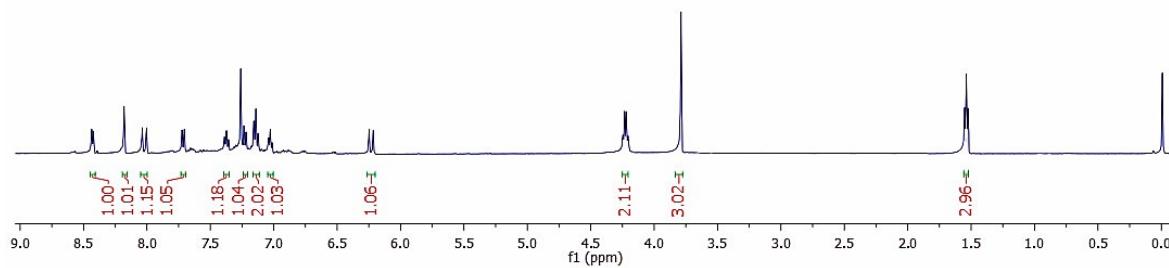


**Figure SI-118.** ESI-MS chromatogram of **7f**.

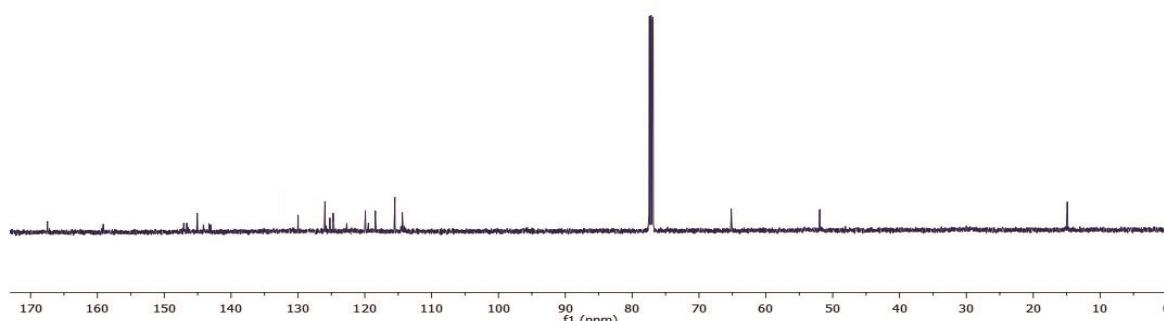


**Methyl (E)-3-(2-(8-ethoxy-2-oxo-2H-chromen-3-yl)imidazo[1,2-a]pyridin-3-yl)acrylate (7g):** Yield 71%; dark brown powder; m.p. = 137 – 141 °C; RMN  $^1\text{H}$  (500 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  (ppm) = 8.428 (d,  $J$  (H,H) = 6.5 Hz, 1H, Ar), 8.180 (s, 1H,  $-\text{CH}=\text{C}-$ ), 8.019 (d,  $J$  (H,H) = 16 Hz, 1H,  $-\text{CH}=\text{C}-$ ), 7.713 (d,  $J$  (H,H) = 9 Hz, 1H, Ar), 7.373 (d,  $J$  (H,H) = 7 Hz, 1H, Ar), 7.227 (d,  $J$  (H,H) = 8 Hz, 1H, Ar), 7.157 – 7.140 (m, 2H, Ar), 7.027 (t,  $J$  (H,H) = 7 Hz, 1H, Ar), 7.231 (d,  $J$  (H,H) = 16 Hz, 1H,  $-\text{CH}=\text{C}-$ ), 4.226 (q,  $J$  (H,H) = 7 Hz, 2H,  $\text{CH}_2-\text{CH}_3$ ), 3.788 (s, 1H,  $\text{CH}_3$ ), 1.538 (t,  $J$  (H,H) = 7 Hz, 3H,  $\text{CH}_3$ ); RMN  $^{13}\text{C}$  (125 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  (ppm) = 167.460, 159.120, 147.080, 146.631, 145.059, 144.059, 143.303, 129.954, 126.950, 125.220,

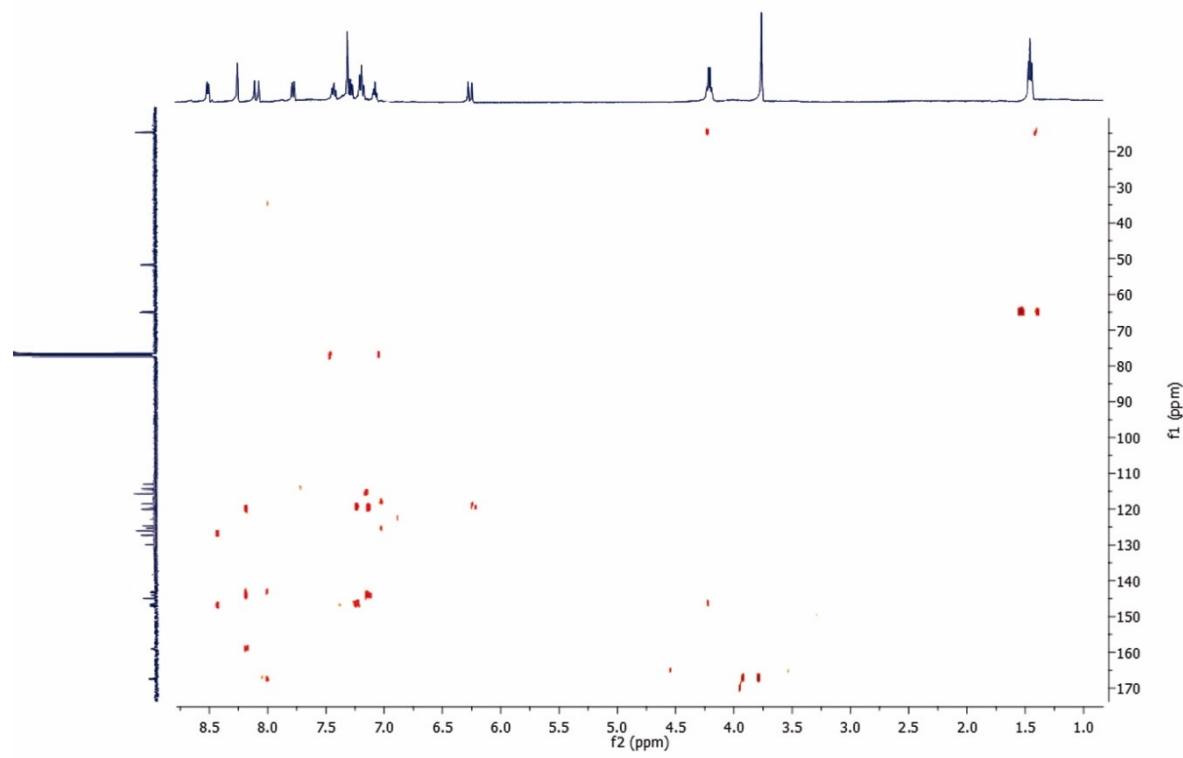
124.732, 122.701, 119.957, 119.914, 119.463, 118.408, 115.499, 114.322, 65.173, 51.940, 14.909; FT-IR (KBr)  $\nu_{\text{max}}$  = 2926 cm<sup>-1</sup> (C—H Aliphatic), 1720 cm<sup>-1</sup> (C=O lactone), 1621 cm<sup>-1</sup> (—O—C=O); UV–Vis (MeOH)  $\lambda_{\text{max}}$  = 314.90 nm; HRMS (ESI m/z) Calcd. for C<sub>22</sub>H<sub>19</sub>N<sub>2</sub>O<sub>5</sub> [M+H]<sup>+</sup> 391.1288, found 391.1299.



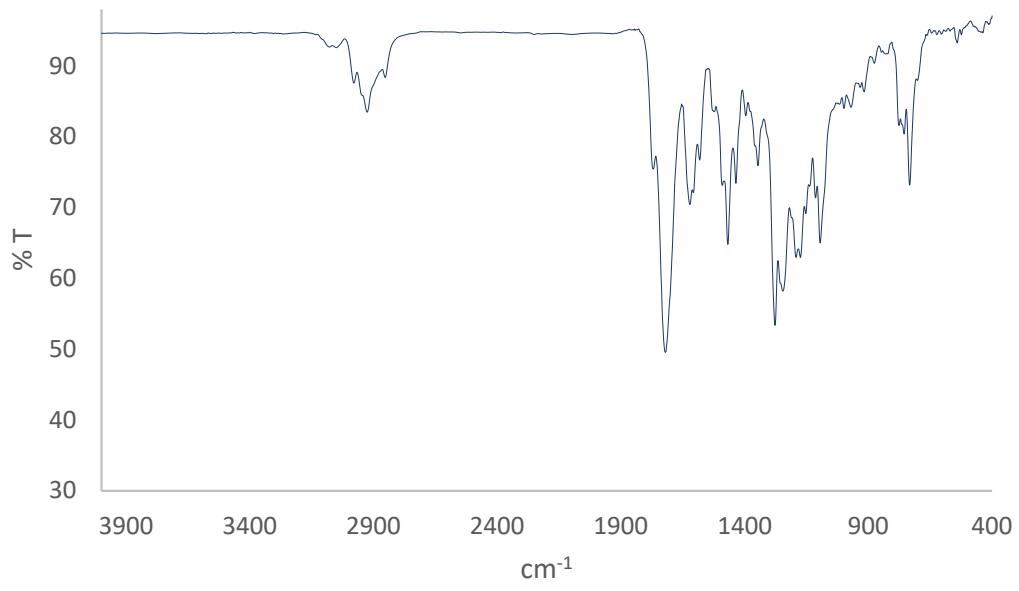
**Figure SI-119.** <sup>1</sup>H NMR spectra of 7g on CDCl<sub>3</sub> 500 MHz



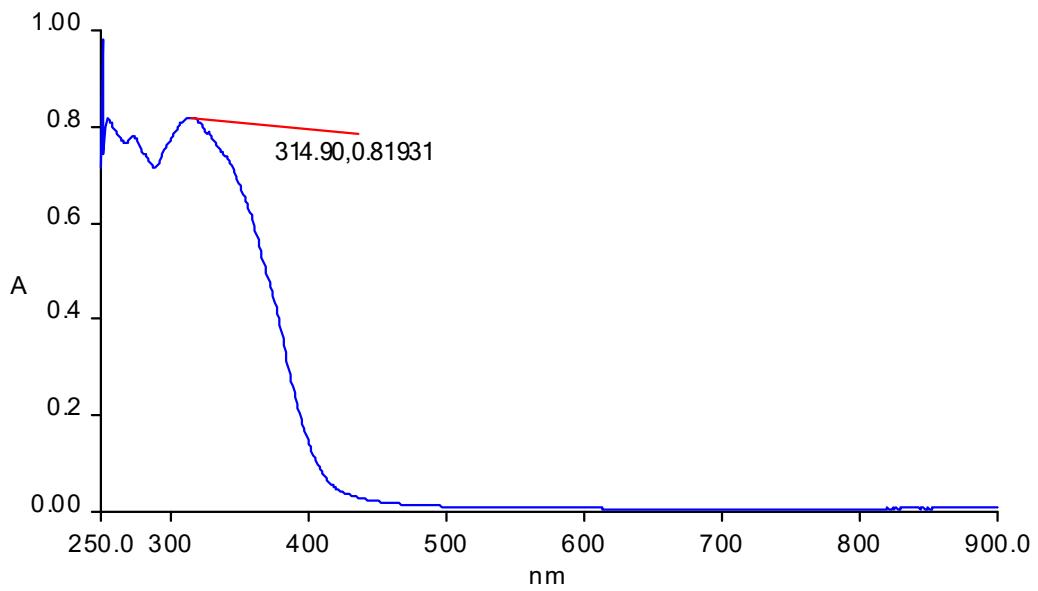
**Figure SI-120.** <sup>13</sup>C NMR spectra of 7g on CDCl<sub>3</sub> 125 MHz.



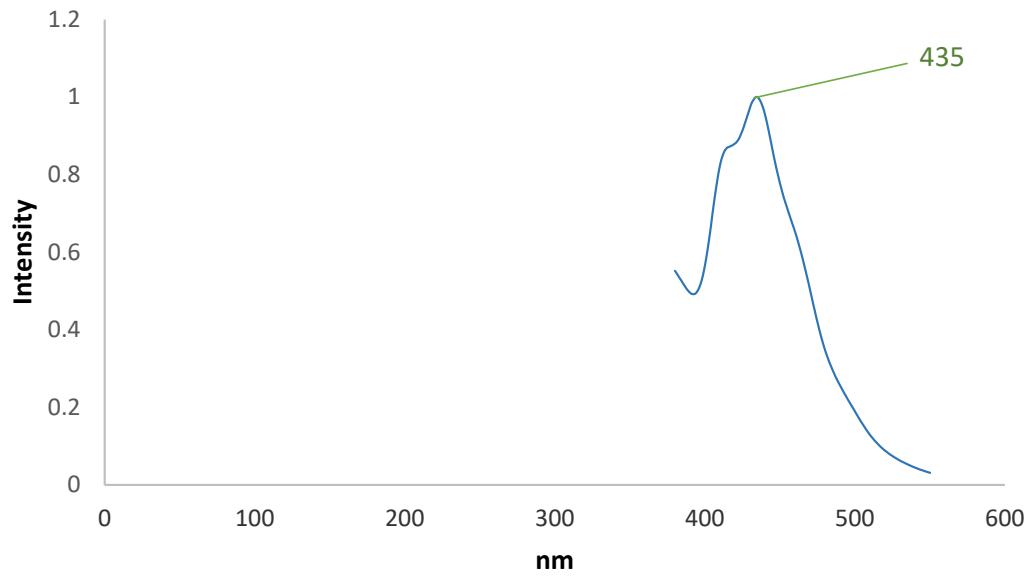
**Figure SI-121.** HMBC NMR spectra of **7g** on  $\text{CDCl}_3$



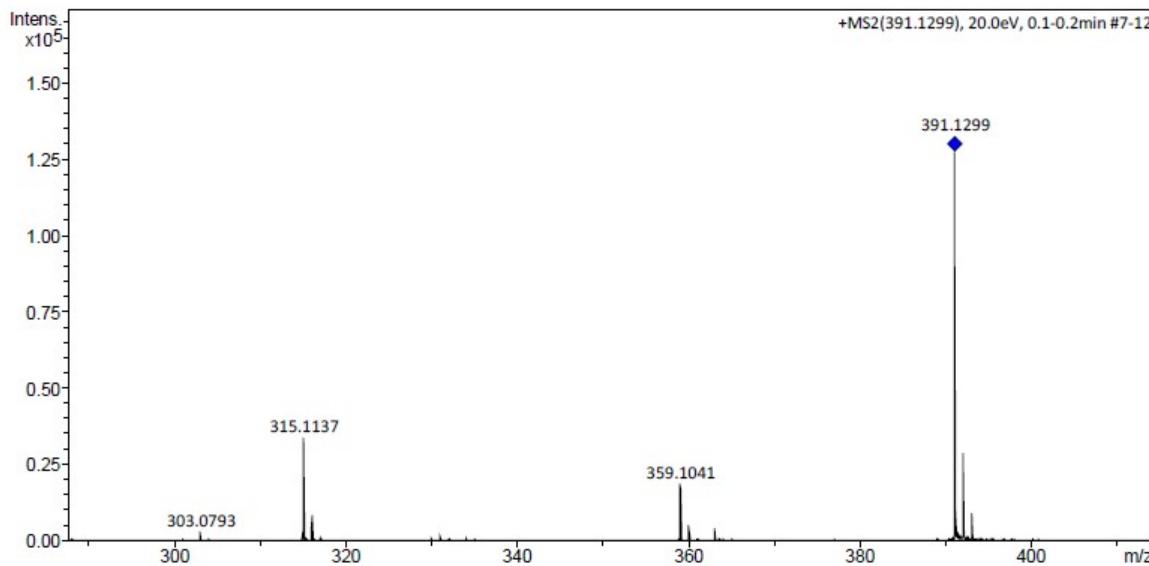
**Figure SI-122.** IR spectra for **7g** in  $\text{KBr}$ .



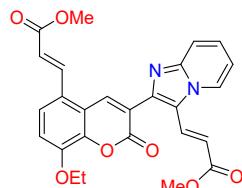
**Figure SI-123.** UV-Vis spectra of **7g** in MeOH.



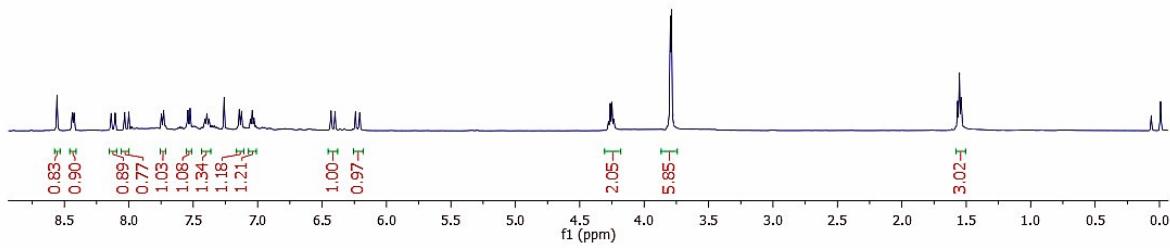
**Figure SI-124.** Emission spectra of **7g** in MeOH.



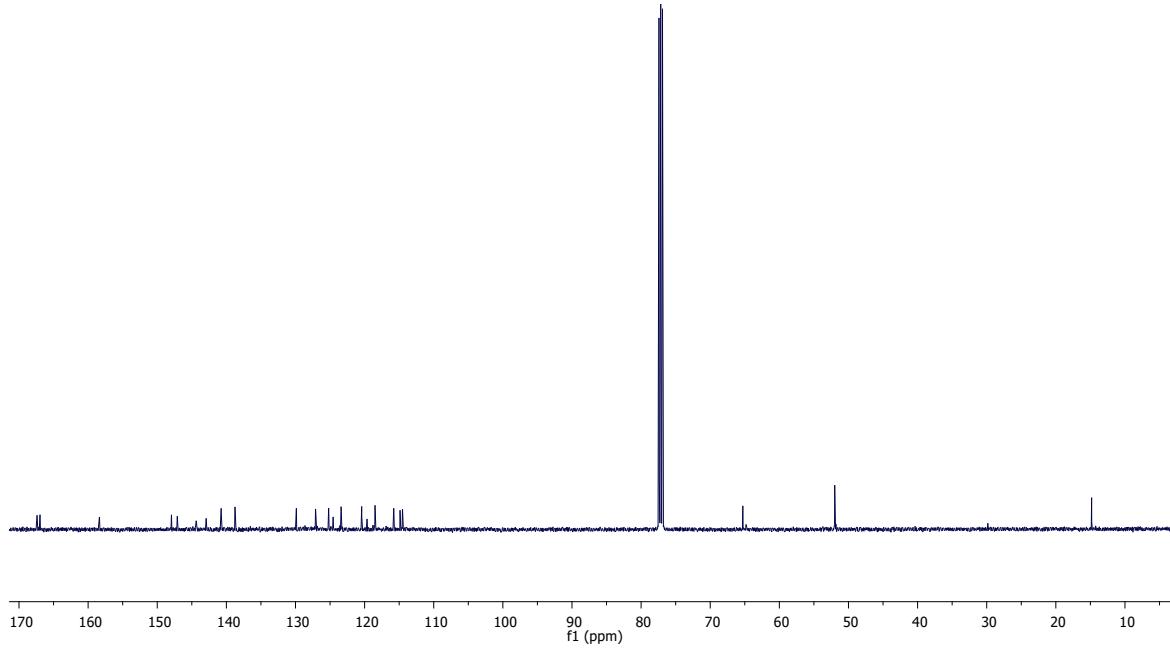
**Figure SI-125.** ESI-MS chromatogram of **7g**.



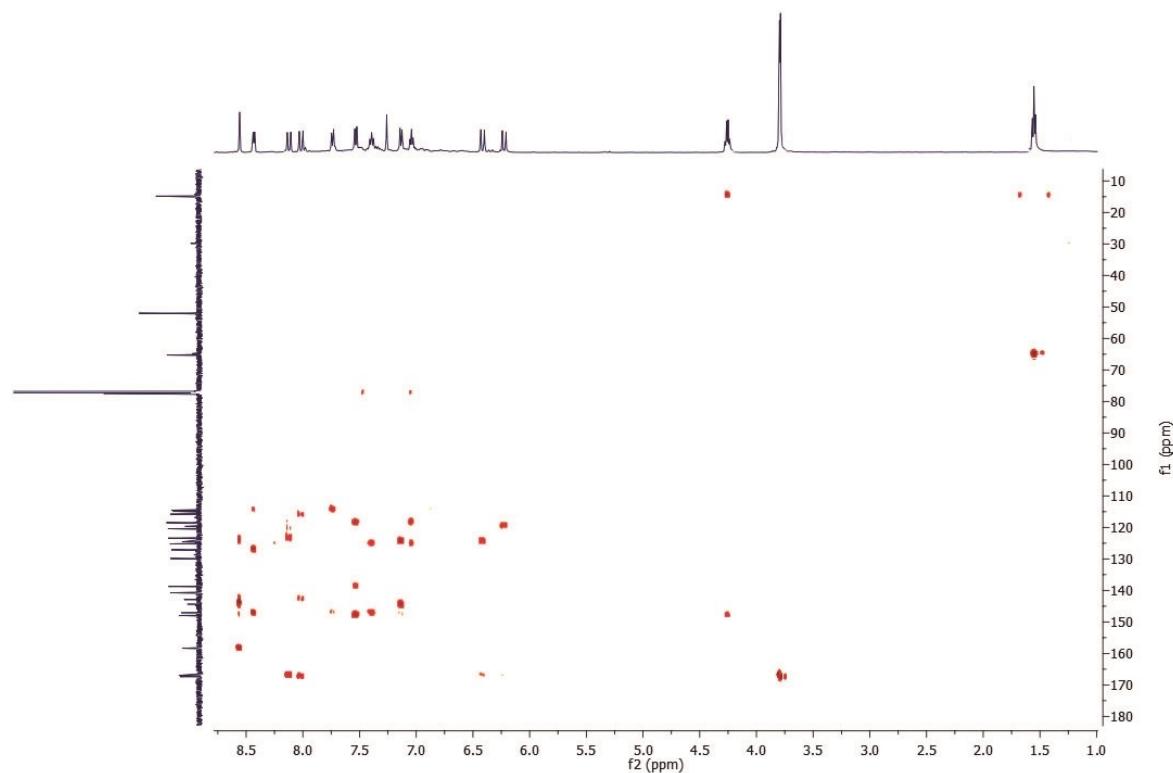
**Methyl (E)-3-(8-ethoxy-3-((E)-3-methoxy-3-oxoprop-1-en-1-yl)imidazo[1,2-a]pyridin-2-yl)-2-oxo-2H-chromen-5-yl)acrylate (7h):** Yield 61%; dark brown powder; m.p. = 179 – 181 °C; RMN  $^1\text{H}$  (500 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  (ppm) = 8.557 (s, 1H,  $-\text{CH}=\text{C}-$ ), 8.432 (d,  $J(\text{H},\text{H})$  = 7 Hz, 1H, Ar), 8.121 (d,  $J(\text{H},\text{H})$  = 15.5 Hz, 1H,  $-\text{CH}=\text{C}-$ ), 8.015 (d,  $J(\text{H},\text{H})$  = 16.5 Hz, 1H,  $-\text{CH}=\text{C}-$ ), 7.729 (d,  $J(\text{H},\text{H})$  = 9 Hz, 1H, Ar), 7.532 (d,  $J(\text{H},\text{H})$  = 8.5 Hz, 1H, Ar), 7.393 (t,  $J(\text{H},\text{H})$  = 8 Hz, 1H, Ar), 7.134 (d,  $J(\text{H},\text{H})$  = 8.5 Hz, 1H, Ar), 7.042 (t,  $J(\text{H},\text{H})$  = 7 Hz, 1H, Ar), 6.414 (d,  $J(\text{H},\text{H})$  = 15.5 Hz, 1H,  $-\text{CH}=\text{C}-$ ), 6.224 (d,  $J(\text{H},\text{H})$  = 16.5 Hz, 1H,  $-\text{CH}=\text{C}-$ ), 4.256 (q,  $J(\text{H},\text{H})$  = 7 Hz, 2H,  $\text{CH}_2-\text{CH}_3$ ), 3.796 (s, 3H,  $\text{CH}_3$ ), 3.788 (s, 3H,  $\text{CH}_3$ ), 1.553 (t,  $J(\text{H},\text{H})$  = 7 Hz, 3H,  $\text{CH}_3$ ); RMN  $^{13}\text{C}$  (125 MHz,  $\text{CDCl}_3$ , 25 °C, TMS)  $\delta$  (ppm) = 123.428, 167.384, 166.963, 158.365, 147.943, 147.103, 144.360, 142.923, 140.950, 138.753, 129.872, 127.096, 125.203, 124.553, 123.406, 120.420, 119.650, 118.514, 118.493, 115.786, 114.875, 114.517, 65.281, 51.988, 14.823; FT-IR (KBr)  $\nu_{\text{max}}$  = 3045 cm $^{-1}$  (C–H Ar), 2978 cm $^{-1}$  (C–H Aliphatic), 1718 cm $^{-1}$  (C=O lactone), 1631 cm $^{-1}$  ( $-\text{O}-\text{C}=\text{O}$ ); UV–Vis (MeOH)  $\lambda_{\text{max}}$  = 328.33 nm. HRMS (ESI m/z) Calcd. for  $\text{C}_{26}\text{H}_{22}\text{N}_2\text{O}_7$  [ $\text{M}+\text{H}]^+$  475.1500, found 475.1526.



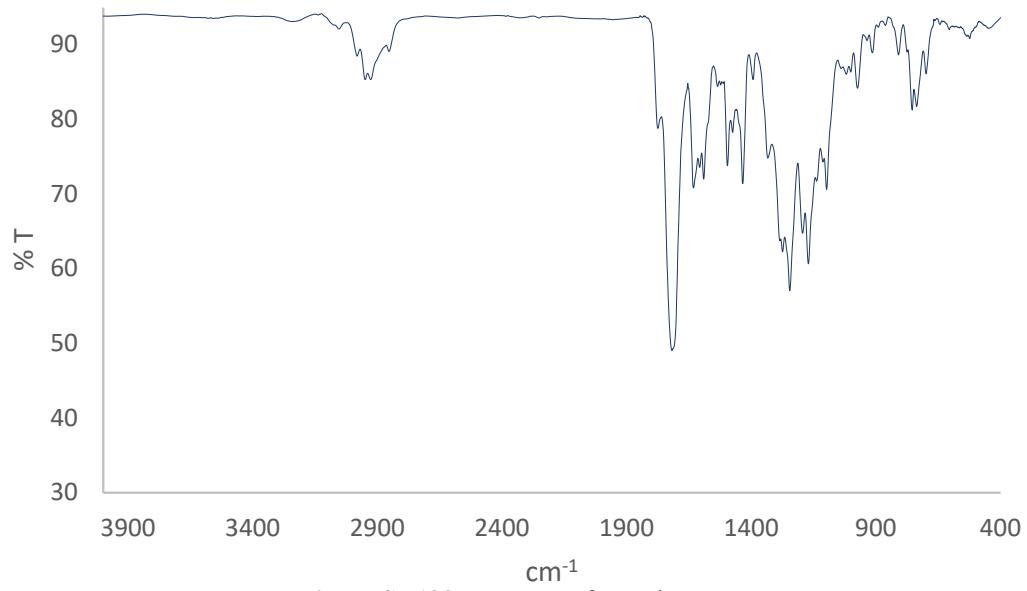
**Figure SI-126.**  $^1\text{H}$  NMR spectra of **7h** on  $\text{CDCl}_3$  500 MHz



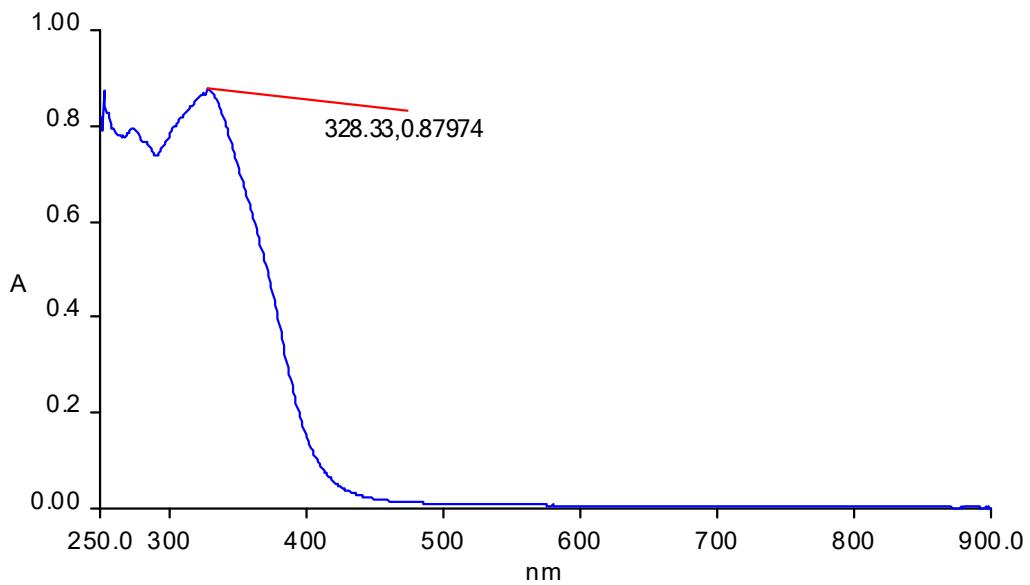
**Figure SI-127.**  $^{13}\text{C}$  NMR spectra of **7h** on  $\text{CDCl}_3$  125 MHz.



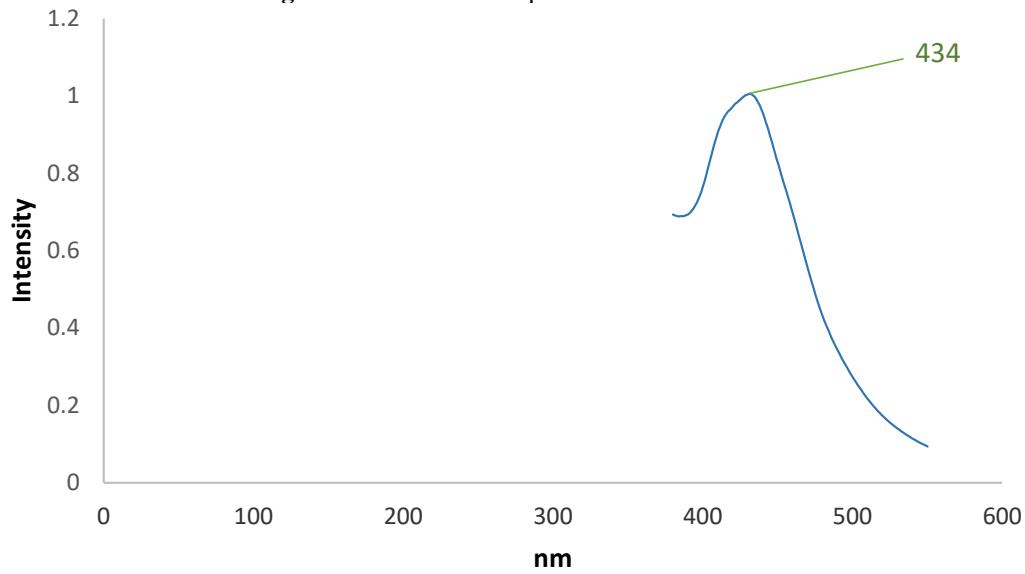
**Figure SI-128.** HMBC NMR spectra of **7h** on  $\text{CDCl}_3$



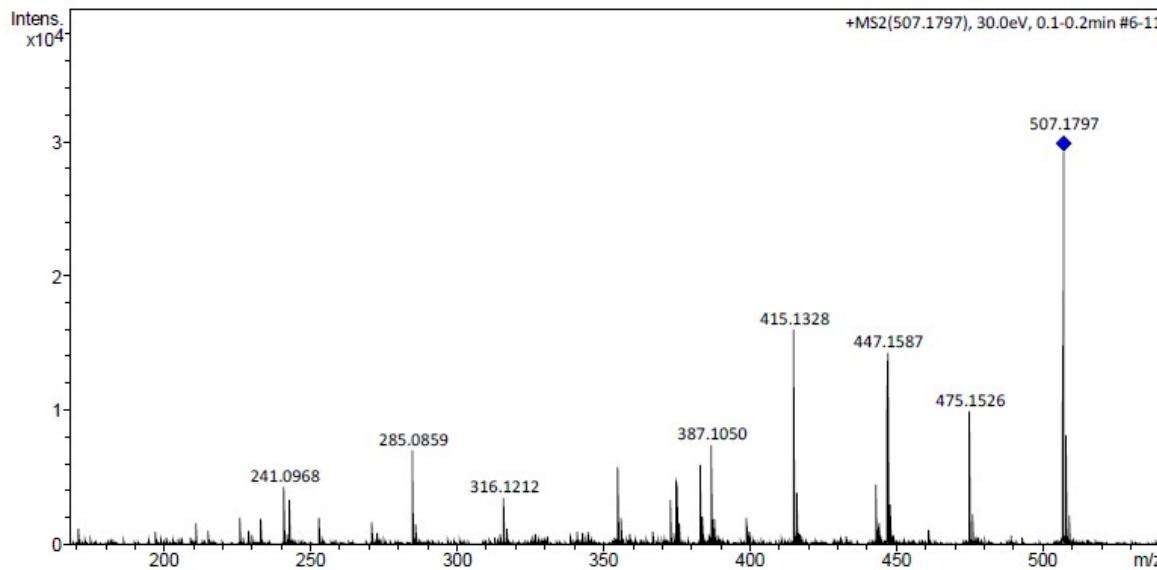
**Figure SI-129.** IR spectra for **7h** in KBr.



**Figure SI-130.** UV-Vis spectra of **7h** in MeOH.



**Figure SI-131.** Emission spectra of **7h** in MeOH.



**Figure SI-132.** ESI chromatogram of **7h**.

## 6. Nonlinear optical properties

**Table S3.** Non-linear refractive index values of coumarin-imidazo[1,2-*a*]heterocyclic-3-acrylate **7a-h**.

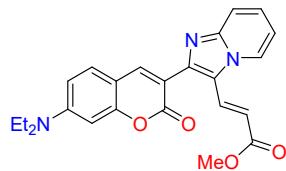
Power (mW)	Nonlinear refraction index $n_2$ (W/cm <sup>2</sup> )									
	4a	5a	7a	7b	7c	7d	7e	7f	7g	7h
1	-7.403e <sup>-7</sup>									
3	-1.22e <sup>-7</sup>									
4	-7.7218e <sup>-8</sup>									
5	-6.332e <sup>-8</sup>			-1.994e <sup>-9</sup>						
7	-1.179e <sup>-8</sup>			-1.616e <sup>-8</sup>						
10				-7.514e <sup>-9</sup>	-2.622e <sup>-9</sup>	-9.499e <sup>-10</sup>	-1.324e <sup>-08</sup>			-2.188e <sup>-09</sup>
15				-2.486e <sup>-9</sup>	-4.133e <sup>-9</sup>	-2.215e <sup>-9</sup>	-4.557e <sup>-10</sup>	-3.252e <sup>-09</sup>		-1.198e <sup>-09</sup>
20				-1.507e <sup>-9</sup>		-1.210e <sup>-9</sup>	-8.254e <sup>-10</sup>	-1.656e <sup>-09</sup>		-6.373e <sup>-10</sup>
25			-2.1424e <sup>-10</sup>	-1.184e <sup>-09</sup>						-4.795e <sup>-10</sup>
30			-1.1188e <sup>-10</sup>	-8.832e <sup>-10</sup>	-9.553e <sup>-10</sup>	-8.475e <sup>-10</sup>			-2.267e <sup>-10</sup>	-3.688e <sup>-10</sup>
35			-2.6887e <sup>-10</sup>	-7.719e <sup>-10</sup>	-5.532e <sup>-10</sup>					-1.634e <sup>-10</sup>
40			-1.824e <sup>-10</sup>		-5.721e <sup>-10</sup>			-1.569e <sup>-10</sup>	-1.290e <sup>-10</sup>	-3.146e <sup>-10</sup>
45			-1.621e <sup>-10</sup>					-1.254e <sup>-10</sup>	-1.491e <sup>-10</sup>	
50								-1.112e <sup>-10</sup>	-1.243e <sup>-10</sup>	
55										-9.896e <sup>-11</sup>

## 7.- Theoretical calculations

*Computational Details.*

The stationary points of all molecules were obtained by Berny geometry optimization using the Density Functional Theory (DFT). To test the performance of the level of theory against the experimental absorption-emission data, four DFT functionals were used: B3LYP [4,5], CAM-B3LYP [6], PBE [7] (PBE/PBE) and M062XR [8] with the basis set 6-311++g(d,p) [9], the Polarizable Continuum Model (PCM) [10,11]. with methanol as solvent was used, all stationary points were confirmed with a frequency analysis. The theoretical characterization of the optical properties were performed using the Time Dependent Density Functional Theory (TD-DFT) [12]. We have performed vertical excitation of all the molecules using TD-DFT at the corresponding theory level of the respective optimization calculation. In addition, the relaxation of the brighter excited states in the gas phase was performed using TD-DFT at corresponding level to that of its fundamental geometry stationary point. Similarly, the theoretical calculation of polarizability and hyperpolarizability was made. All calculations were done using the Gaussian09 software [13].

#### **8.- Equilibrium Geometry in Ground State, Equilibrium Geometry in Excited states, Excitation energies and oscillator strength of compound 7a (B3LYP)**



**7a**

B3LYP

Standard orientation:

Center	Atomic Number	Atomic Number	Type	Coordinates (Angstroms)		
				X	Y	Z
1	6	0	4.597339	-0.346524	-0.178995	
2	6	0	3.583912	0.318825	0.554517	
3	6	0	2.268361	-0.084203	0.434986	
4	6	0	1.862884	-1.146831	-0.394390	
5	6	0	2.877632	-1.806859	-1.123080	
6	6	0	4.194480	-1.428834	-1.025876	
7	1	0	3.805590	1.130209	1.231006	
8	1	0	2.603404	-2.625927	-1.778838	
9	1	0	4.927516	-1.955642	-1.618172	
10	6	0	0.485667	-1.464123	-0.444542	
11	6	0	-0.454689	-0.770254	0.277149	
12	6	0	-0.021404	0.291737	1.168878	
13	8	0	1.335559	0.589138	1.177980	
14	1	0	0.158558	-2.280675	-1.078849	
15	6	0	-3.027309	-0.388957	0.112442	
16	6	0	-3.497975	-2.586482	0.120210	
17	7	0	-2.166051	-2.505354	0.208726	
18	6	0	-1.866769	-1.183860	0.224975	
19	7	0	-4.077286	-1.315349	0.067458	
20	6	0	-4.325147	-3.724454	0.097533	

21	6	0	-5.688427	-3.562735	0.040586
22	1	0	-6.340033	-4.427247	0.022118
23	6	0	-6.247793	-2.261400	0.019436
24	1	0	-7.319826	-2.120225	-0.003066
25	6	0	-5.441514	-1.157981	0.037674
26	1	0	-5.827184	-0.153799	0.054233
27	1	0	-3.858535	-4.700019	0.131391
28	7	0	5.906144	0.029228	-0.085610
29	6	0	6.985196	-0.716984	-0.746046
30	1	0	7.879597	-0.594592	-0.131336
31	1	0	6.753502	-1.782982	-0.727618
32	6	0	7.274185	-0.248563	-2.175503
33	1	0	6.406162	-0.384561	-2.824586
34	1	0	7.549601	0.809045	-2.192309
35	1	0	8.106368	-0.821646	-2.593407
36	6	0	6.323831	1.199206	0.696113
37	1	0	7.224350	1.596096	0.222435
38	1	0	5.568170	1.981576	0.606294
39	6	0	6.610079	0.887014	2.168433
40	1	0	6.939113	1.793533	2.683595
41	1	0	5.721413	0.508379	2.678208
42	1	0	7.400998	0.138515	2.262169
43	8	0	-0.712994	0.946098	1.925777
44	6	0	-3.097529	1.024236	-0.071681
45	6	0	-4.130102	1.822256	-0.439024
46	6	0	-4.001493	3.270839	-0.613564
47	8	0	-4.919373	3.987816	-0.983967
48	8	0	-2.773355	3.755437	-0.329057
49	6	0	-2.585766	5.175471	-0.485894
50	1	0	-1.552327	5.363071	-0.204557
51	1	0	-2.756959	5.471473	-1.521500
52	1	0	-3.263138	5.724856	0.168668
53	1	0	-5.121531	1.462274	-0.668467
54	1	0	-2.149062	1.517457	0.096685

-----  
Rotational constants (GHZ):    0.2218649    0.0648972    0.0538852

#### Excitation energies and oscillator strengths:

Excited State 1: Singlet-A    2.9247 eV 423.92 nm f=0.6876 <S\*\*2>=0.000  
110 ->111    0.69871

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -1393.85529913

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A    3.1591 eV 392.46 nm f=0.2270 <S\*\*2>=0.000  
109 ->111    -0.27309  
110 ->112    0.64086

Excited State 3: Singlet-A    3.4174 eV 362.81 nm f=0.3052 <S\*\*2>=0.000  
109 ->111    0.60016

		109 ->112	0.22260
		110 ->112	0.27381
Excited State 4:	Singlet-A	3.6365 eV	340.94 nm f=0.2581 <S**2>=0.000
		109 ->111	-0.19047
		109 ->112	0.59655
		110 ->113	0.30221
Excited State 5:	Singlet-A	3.7336 eV	332.08 nm f=0.1596 <S**2>=0.000
		109 ->111	0.12601
		109 ->112	-0.29031
		109 ->113	-0.14217
		110 ->113	0.60006
Excited State 6:	Singlet-A	4.0594 eV	305.42 nm f=0.0272 <S**2>=0.000
		107 ->111	-0.11886
		107 ->112	0.10333
		108 ->111	-0.25097
		108 ->112	0.10133
		109 ->113	0.59225
		110 ->113	0.19319
Excited State 7:	Singlet-A	4.0893 eV	303.19 nm f=0.0118 <S**2>=0.000
		107 ->111	-0.37913
		107 ->112	-0.12616
		108 ->111	0.43196
		108 ->112	0.15045
		110 ->114	0.33212
Excited State 8:	Singlet-A	4.2893 eV	289.05 nm f=0.0787 <S**2>=0.000
		107 ->111	0.20125
		108 ->111	0.46235
		109 ->113	0.22319
		110 ->114	-0.40788
Excited State 9:	Singlet-A	4.4064 eV	281.37 nm f=0.0643 <S**2>=0.000
		107 ->111	0.51700
		109 ->113	0.10946
		110 ->114	0.42094
Excited State 10:	Singlet-A	4.4980 eV	275.64 nm f=0.0617 <S**2>=0.000
		108 ->112	0.65434
		109 ->113	-0.15040
Excited State 11:	Singlet-A	4.6091 eV	269.00 nm f=0.0004 <S**2>=0.000
		105 ->111	0.57057
		105 ->112	-0.37659
Excited State 12:	Singlet-A	4.6305 eV	267.76 nm f=0.1150 <S**2>=0.000
		107 ->111	-0.12069
		107 ->112	0.65544

		109 ->113	-0.10713
		110 ->114	0.10348
Excited State 13:	Singlet-A	4.7459 eV	261.25 nm f=0.0167 <S**2>=0.000
		110 ->115	0.55740
		110 ->116	0.34334
		110 ->117	0.16316
		110 ->118	0.11489
Excited State 14:	Singlet-A	4.8314 eV	256.62 nm f=0.0367 <S**2>=0.000
		107 ->113	0.11523
		108 ->113	0.13886
		109 ->115	0.13746
		109 ->116	-0.10647
		110 ->115	-0.34197
		110 ->116	0.50094
		110 ->117	0.18859
Excited State 15:	Singlet-A	4.8430 eV	256.01 nm f=0.0034 <S**2>=0.000
		104 ->111	0.53853
		104 ->112	0.32439
		104 ->113	-0.14132
		106 ->111	0.20195
Excited State 16:	Singlet-A	4.9228 eV	251.86 nm f=0.1442 <S**2>=0.000
		106 ->111	0.11497
		108 ->113	-0.12714
		110 ->116	-0.22092
		110 ->117	0.60734
Excited State 17:	Singlet-A	4.9826 eV	248.83 nm f=0.0119 <S**2>=0.000
		109 ->114	0.15247
		109 ->116	0.11492
		110 ->118	0.62705
		110 ->120	0.11290
Excited State 18:	Singlet-A	5.0062 eV	247.66 nm f=0.0010 <S**2>=0.000
		106 ->111	0.19133
		109 ->114	0.63399
		110 ->118	-0.12571
Excited State 19:	Singlet-A	5.0462 eV	245.70 nm f=0.0689 <S**2>=0.000
		103 ->111	0.26245
		106 ->111	-0.37355
		108 ->113	0.37938
		109 ->114	0.21346
		110 ->116	-0.11629
		110 ->117	0.13217
Excited State 20:	Singlet-A	5.1038 eV	242.93 nm f=0.0258 <S**2>=0.000
		102 ->111	0.10510

103 ->111      0.52201  
 105 ->112      -0.10536  
 108 ->113      -0.31813  
 109 ->115      -0.12730  
 109 ->116      0.12885  
 110 ->116      0.12105

SavETr: write IOETrn= 770 NScale= 10 NData= 16 NLR=1 LETran= 370.

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	4.713993	-0.266004	-0.108189
2	6	0	3.643156	0.555039	0.345878
3	6	0	2.351388	0.085758	0.310142
4	6	0	2.005743	-1.204021	-0.168585
5	6	0	3.087034	-2.016732	-0.617565
6	6	0	4.385604	-1.575014	-0.589771
7	1	0	3.812320	1.545085	0.741805
8	1	0	2.867446	-3.008204	-0.996329
9	1	0	5.162049	-2.229278	-0.957170
10	6	0	0.654532	-1.582992	-0.189927
11	6	0	-0.368421	-0.705257	0.250444
12	6	0	0.025970	0.571237	0.837500
13	8	0	1.370600	0.920278	0.781920
14	1	0	0.373117	-2.555581	-0.567012
15	6	0	-2.992746	-0.424924	0.171754
16	6	0	-3.268043	-2.670565	0.093500
17	7	0	-1.960009	-2.497881	0.116732
18	6	0	-1.730889	-1.145132	0.221194
19	7	0	-3.954642	-1.444656	0.143841
20	6	0	-4.015955	-3.872116	0.052720
21	6	0	-5.383133	-3.814199	0.101841
22	1	0	-5.968037	-4.725269	0.072932
23	6	0	-6.045115	-2.559595	0.213696
24	1	0	-7.122252	-2.507123	0.292854
25	6	0	-5.325437	-1.400235	0.238454
26	1	0	-5.779907	-0.433122	0.366745
27	1	0	-3.476858	-4.808276	-0.006437
28	7	0	6.002679	0.181332	-0.087665
29	6	0	7.136585	-0.677869	-0.452473
30	1	0	7.996061	-0.326455	0.121553
31	1	0	6.939603	-1.696852	-0.118651
32	6	0	7.466297	-0.645916	-1.949730
33	1	0	6.631378	-1.013691	-2.549551
34	1	0	7.704622	0.368503	-2.277595
35	1	0	8.335516	-1.279464	-2.143586

36	6	0	6.346821	1.554845	0.301387
37	1	0	7.260130	1.816524	-0.236335
38	1	0	5.572930	2.235022	-0.056039
39	6	0	6.568402	1.721886	1.809766
40	1	0	6.850409	2.756126	2.022559
41	1	0	5.663605	1.488777	2.374563
42	1	0	7.371326	1.070511	2.162289
43	8	0	-0.685733	1.372034	1.409415
44	6	0	-3.245467	0.950118	-0.027541
45	6	0	-4.395621	1.590280	-0.441195
46	6	0	-4.490019	3.022298	-0.631611
47	8	0	-5.485973	3.600179	-1.068191
48	8	0	-3.364322	3.706266	-0.282071
49	6	0	-3.396813	5.131605	-0.458869
50	1	0	-2.424385	5.488517	-0.125978
51	1	0	-3.553093	5.388442	-1.507943
52	1	0	-4.188624	5.578581	0.144406
53	1	0	-5.302010	1.071603	-0.716091
54	1	0	-2.385304	1.579054	0.143149

-----  
Rotational constants (GHZ): 0.2250089 0.0635153 0.0517698

#### Excitation energies and oscillator strengths:

Excited State 1: Singlet-A 2.4729 eV 501.37 nm f=1.4063 <S\*\*2>=0.000  
 110 ->111 0.70410

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -1393.86578184

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A 2.9807 eV 415.95 nm f=0.0278 <S\*\*2>=0.000  
 109 ->111 -0.38970  
 110 ->112 0.58164

Excited State 3: Singlet-A 3.1268 eV 396.52 nm f=0.5942 <S\*\*2>=0.000  
 109 ->111 0.57982  
 110 ->112 0.39249

Excited State 4: Singlet-A 3.5456 eV 349.68 nm f=0.1535 <S\*\*2>=0.000  
 109 ->112 0.67461  
 110 ->113 0.18019

Excited State 5: Singlet-A 3.6383 eV 340.77 nm f=0.1601 <S\*\*2>=0.000  
 108 ->111 0.15513  
 109 ->112 -0.18544  
 109 ->113 -0.10925  
 110 ->113 0.64474

Excited State 6: Singlet-A 3.8330 eV 323.46 nm f=0.0202 <S\*\*2>=0.000  
                   107 ->111 -0.28106  
                   108 ->111  0.61067  
                   110 ->113 -0.15208

Excited State 7: Singlet-A 3.9491 eV 313.96 nm f=0.0055 <S\*\*2>=0.000  
                   107 ->111 -0.57770  
                   108 ->111 -0.27240  
                   109 ->113  0.11986  
                   110 ->114  0.23122

Excited State 8: Singlet-A 4.1748 eV 296.99 nm f=0.0946 <S\*\*2>=0.000  
                   107 ->111  0.11625  
                   108 ->111  0.10505  
                   108 ->112  0.19060  
                   109 ->113  0.63896  
                   110 ->113  0.11851

Excited State 9: Singlet-A 4.2408 eV 292.36 nm f=0.0504 <S\*\*2>=0.000  
                   107 ->111  0.24295  
                   110 ->114  0.64251

Excited State 10: Singlet-A 4.3444 eV 285.39 nm f=0.1182 <S\*\*2>=0.000  
                   108 ->112  0.65649  
                   109 ->113 -0.20745

Excited State 11: Singlet-A 4.3951 eV 282.10 nm f=0.0005 <S\*\*2>=0.000  
                   106 ->111  0.60940  
                   106 ->112 -0.31908

Excited State 12: Singlet-A 4.6496 eV 266.66 nm f=0.0601 <S\*\*2>=0.000  
                   104 ->111 -0.33284  
                   105 ->111  0.20514  
                   107 ->112  0.55126

Excited State 13: Singlet-A 4.6538 eV 266.41 nm f=0.0148 <S\*\*2>=0.000  
                   104 ->111  0.52661  
                   104 ->112  0.11578  
                   105 ->111 -0.15754  
                   107 ->112  0.37080

Excited State 14: Singlet-A 4.6843 eV 264.68 nm f=0.0193 <S\*\*2>=0.000  
                   110 ->115  0.59434  
                   110 ->116  0.32475  
                   110 ->120 -0.10968

Excited State 15: Singlet-A 4.7652 eV 260.19 nm f=0.0916 <S\*\*2>=0.000

103 ->111	0.29423
105 ->111	-0.16267
110 ->115	0.27893
110 ->116	-0.49363
110 ->117	0.13131

Excited State 16: Singlet-A 4.7850 eV 259.11 nm f=0.0338 <S\*\*2>=0.000

102 ->111	0.20198
103 ->111	-0.43573
105 ->111	0.30085
108 ->113	0.16271
110 ->115	0.14633
110 ->116	-0.24506
110 ->117	0.12624

Excited State 17: Singlet-A 4.7943 eV 258.61 nm f=0.1174 <S\*\*2>=0.000

105 ->111	-0.10980
109 ->116	0.11696
110 ->115	-0.11059
110 ->116	0.18938
110 ->117	0.61623

Excited State 18: Singlet-A 4.8750 eV 254.33 nm f=0.0104 <S\*\*2>=0.000

109 ->115	0.11524
110 ->117	0.10043
110 ->118	0.66326

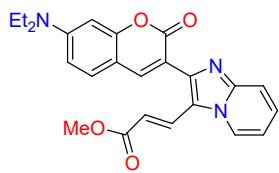
Excited State 19: Singlet-A 4.8886 eV 253.62 nm f=0.0048 <S\*\*2>=0.000

102 ->111	-0.13481
103 ->111	0.33989
104 ->111	0.21791
105 ->111	0.48057
108 ->113	0.19753

Excited State 20: Singlet-A 4.9787 eV 249.03 nm f=0.0445 <S\*\*2>=0.000

108 ->113	0.28155
109 ->114	0.61597
110 ->114	-0.10530

**9.- Equilibrium Geometry in Ground State, Equilibrium Geometry in Excited states,  
Excitation energies and oscillator strength of compound 7a' (B3LYP)**



**7a'**

B3LYP

Standard orientation:

Center	Atomic Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
				X	Y	Z
1	6	0	-4.347400	-0.159664	-0.317349	
2	6	0	-3.529237	-0.946293	0.530643	
3	6	0	-2.161164	-0.973037	0.340286	
4	6	0	-1.513834	-0.239777	-0.671300	
5	6	0	-2.333920	0.539650	-1.515930	
6	6	0	-3.697819	0.583632	-1.354030	
7	1	0	-3.938027	-1.523668	1.346030	
8	1	0	-1.871852	1.109836	-2.314460	
9	1	0	-4.280309	1.180685	-2.039542	
10	6	0	-0.104699	-0.349562	-0.777760	
11	6	0	0.632032	-1.132807	0.072036	
12	6	0	-0.040832	-1.870347	1.131860	
13	8	0	-1.424954	-1.752366	1.193825	
14	1	0	0.396908	0.192422	-1.572786	
15	6	0	3.019715	-0.193440	-0.065759	
16	6	0	4.006622	-2.190575	-0.341782	
17	7	0	2.700369	-2.452998	-0.220668	
18	6	0	2.094595	-1.256780	-0.052922	
19	7	0	4.257874	-0.815648	-0.260347	
20	6	0	5.083095	-3.077763	-0.526603	
21	6	0	6.360152	-2.577717	-0.606295	
22	1	0	7.199881	-3.246461	-0.747586	
23	6	0	6.585020	-1.183241	-0.494855	
24	1	0	7.586971	-0.778848	-0.542235	
25	6	0	5.538313	-0.322035	-0.322264	
26	1	0	5.668374	0.740937	-0.219275	
27	1	0	4.870347	-4.136437	-0.593693	
28	7	0	-5.702516	-0.114019	-0.155769	
29	6	0	-6.557784	0.770380	-0.957319	
30	1	0	-7.418524	1.029869	-0.337036	
31	1	0	-6.033521	1.708166	-1.148014	
32	6	0	-7.037090	0.140770	-2.268730	
33	1	0	-6.198245	-0.105586	-2.923663	
34	1	0	-7.601195	-0.775943	-2.078872	

35	1	0	-7.691772	0.837716	-2.799062
36	6	0	-6.393532	-0.946387	0.836337
37	1	0	-7.396241	-1.140541	0.449610
38	1	0	-5.901950	-1.919086	0.897207
39	6	0	-6.490399	-0.299986	2.221862
40	1	0	-7.035209	-0.959067	2.903229
41	1	0	-5.501533	-0.112809	2.646120
42	1	0	-7.024931	0.652079	2.171648
43	8	0	0.474071	-2.565748	1.983340
44	6	0	2.734549	1.191380	0.126810
45	6	0	3.508632	2.292985	-0.026646
46	6	0	3.030015	3.653513	0.230913
47	8	0	3.710285	4.651667	0.047910
48	8	0	1.762135	3.714195	0.691987
49	6	0	1.232200	5.028181	0.956508
50	1	0	0.221518	4.863695	1.322206
51	1	0	1.831697	5.536251	1.712461
52	1	0	1.213346	5.623416	0.042951
53	1	0	4.527412	2.269209	-0.382975
54	1	0	1.711075	1.364359	0.442470

-----  
Rotational constants (GHZ): 0.2194130 0.0680333 0.0566899

#### Excitation energies and oscillator strengths:

Excited State 1: Singlet-A 2.9950 eV 413.97 nm f=0.2820 <S\*\*2>=0.000  
 110 ->111 0.69613

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -1393.85193139

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A 3.1782 eV 390.10 nm f=0.4993 <S\*\*2>=0.000  
 109 ->111 -0.19910  
 110 ->112 0.66022

Excited State 3: Singlet-A 3.4726 eV 357.03 nm f=0.1361 <S\*\*2>=0.000  
 109 ->111 0.42083  
 109 ->112 0.52434  
 110 ->112 0.19119

Excited State 4: Singlet-A 3.6092 eV 343.52 nm f=0.5724 <S\*\*2>=0.000  
 109 ->111 0.50476  
 109 ->112 -0.45108  
 110 ->113 0.13116

Excited State 5: Singlet-A 3.7604 eV 329.71 nm f=0.0423 <S\*\*2>=0.000  
 109 ->111 -0.10785  
 109 ->113 -0.12787  
 110 ->113 0.66700

Excited State 6: Singlet-A 4.0234 eV 308.16 nm f=0.0057 <S\*\*2>=0.000

	107 ->111	-0.19320
	108 ->111	-0.16929
	109 ->113	0.61336
	110 ->113	0.16770
Excited State 7:	Singlet-A	4.1442 eV 299.18 nm f=0.0106 <S**2>=0.000
	107 ->111	0.20256
	107 ->112	0.19286
	108 ->111	-0.37194
	108 ->112	-0.31203
	110 ->114	0.41411
Excited State 8:	Singlet-A	4.3475 eV 285.19 nm f=0.1022 <S**2>=0.000
	107 ->112	0.10873
	108 ->111	0.49380
	108 ->112	0.11238
	109 ->113	0.14275
	110 ->114	0.43550
Excited State 9:	Singlet-A	4.4429 eV 279.06 nm f=0.0574 <S**2>=0.000
	107 ->111	0.21393
	108 ->111	-0.24655
	108 ->112	0.59170
Excited State 10:	Singlet-A	4.4747 eV 277.08 nm f=0.0484 <S**2>=0.000
	107 ->111	0.53900
	107 ->112	0.18465
	108 ->111	0.10108
	108 ->112	-0.10333
	109 ->113	0.17417
	110 ->114	-0.30443
	110 ->117	0.10451
Excited State 11:	Singlet-A	4.5513 eV 272.42 nm f=0.1594 <S**2>=0.000
	107 ->111	-0.23459
	107 ->112	0.61583
	109 ->113	-0.15995
	110 ->114	-0.11604
Excited State 12:	Singlet-A	4.6202 eV 268.35 nm f=0.0028 <S**2>=0.000
	104 ->111	0.22478
	104 ->112	-0.11786
	105 ->111	0.54268
	105 ->112	-0.22100
	106 ->111	0.23685
Excited State 13:	Singlet-A	4.7560 eV 260.69 nm f=0.0127 <S**2>=0.000
	110 ->115	0.66669
	110 ->118	-0.18060
Excited State 14:	Singlet-A	4.8105 eV 257.73 nm f=0.0009 <S**2>=0.000

103 ->112	-0.14543
104 ->111	0.32837
104 ->112	0.44689
104 ->113	0.10354
105 ->111	-0.10247
105 ->112	-0.22584
106 ->111	-0.15272
106 ->112	-0.20121

Excited State 15: Singlet-A 4.8892 eV 253.59 nm f=0.0916 <S\*\*2>=0.000

107 ->113	0.10515
109 ->116	-0.17417
110 ->116	0.49895
110 ->117	0.38688

Excited State 16: Singlet-A 4.9208 eV 251.96 nm f=0.1131 <S\*\*2>=0.000

106 ->111	-0.11079
107 ->113	-0.15308
108 ->113	-0.15535
109 ->116	0.10652
110 ->116	-0.35744
110 ->117	0.52082

Excited State 17: Singlet-A 5.0094 eV 247.50 nm f=0.0195 <S\*\*2>=0.000

103 ->111	0.12488
104 ->111	0.15977
104 ->112	-0.13548
106 ->111	-0.26616
109 ->114	0.41973
110 ->118	0.34843
110 ->119	-0.11605

Excited State 18: Singlet-A 5.0178 eV 247.09 nm f=0.0043 <S\*\*2>=0.000

103 ->111	-0.10262
104 ->111	-0.13601
104 ->112	0.10841
106 ->111	0.20059
109 ->114	-0.23841
110 ->115	0.12904
110 ->118	0.51445
110 ->119	-0.16507

Excited State 19: Singlet-A 5.0303 eV 246.47 nm f=0.0294 <S\*\*2>=0.000

103 ->111	-0.17060
104 ->111	-0.24862
104 ->112	0.17739
106 ->111	0.28835
106 ->112	-0.11252
109 ->114	0.49822

Excited State 20: Singlet-A 5.0586 eV 245.10 nm f=0.0589 <S\*\*2>=0.000

104 ->111	0.11749
107 ->113	0.24178
108 ->113	0.46712
109 ->116	-0.28401
110 ->116	-0.27497

SavETr: write IOETrn= 770 NScale= 10 NData= 16 NLR=1 LETran= 370.

\*\*\*\*\*

Excited State 5: Singlet-A 3.9152 eV 316.68 nm f=0.1530 <S\*\*2>=0.000

99 ->104	0.10166
100 ->103	0.15302
100 ->104	-0.31073
101 ->104	-0.14061
102 ->105	0.57426

Excited State 6: Singlet-A 4.0080 eV 309.34 nm f=0.0497 <S\*\*2>=0.000

101 ->103	0.10329
101 ->104	0.67847
102 ->105	0.10456

Excited State 7: Singlet-A 4.1436 eV 299.22 nm f=0.0124 <S\*\*2>=0.000

99 ->103	0.20258
100 ->104	0.59085
102 ->105	0.30570

Excited State 8: Singlet-A 4.4361 eV 279.49 nm f=0.0085 <S\*\*2>=0.000

99 ->103	0.62904
99 ->104	-0.11501
100 ->104	-0.12300
101 ->105	-0.11804
101 ->106	0.12855
102 ->105	-0.14001

Excited State 9: Singlet-A 4.5620 eV 271.77 nm f=0.0002 <S\*\*2>=0.000

98 ->103	0.49825
98 ->104	0.47510

Excited State 10: Singlet-A 4.6563 eV 266.27 nm f=0.0521 <S\*\*2>=0.000

99 ->103	0.10040
101 ->105	0.67740

Excited State 11: Singlet-A 4.7110 eV 263.18 nm f=0.0035 <S\*\*2>=0.000

95 ->103	0.17882
96 ->103	0.21550
97 ->103	0.58154
97 ->104	-0.16276

Excited State 12: Singlet-A 4.7660 eV 260.14 nm f=0.1354 <S\*\*2>=0.000  
                   100 ->105      0.64899  
                   102 ->107      -0.14278  
                   102 ->108      0.11278

Excited State 13: Singlet-A 4.8196 eV 257.25 nm f=0.1297 <S\*\*2>=0.000  
                   99 ->104      0.47650  
                   102 ->106      0.48526

Excited State 14: Singlet-A 4.9004 eV 253.01 nm f=0.1685 <S\*\*2>=0.000  
                   99 ->103      0.11082  
                   99 ->104      0.47126  
                   102 ->106      -0.45767

Excited State 15: Singlet-A 4.9611 eV 249.91 nm f=0.0112 <S\*\*2>=0.000  
                   95 ->103      -0.13543  
                   96 ->103      0.58885  
                   97 ->103      -0.16551  
                   98 ->103      -0.19250  
                   98 ->104      0.19847

Excited State 16: Singlet-A 5.1227 eV 242.03 nm f=0.1911 <S\*\*2>=0.000  
                   99 ->105      -0.19596  
                   100 ->106      0.16453  
                   101 ->106      0.43981  
                   102 ->106      -0.15034  
                   102 ->107      -0.17720  
                   102 ->108      0.32570  
                   102 ->109      -0.11755

Excited State 17: Singlet-A 5.1565 eV 240.44 nm f=0.0059 <S\*\*2>=0.000  
                   96 ->103      -0.20335  
                   97 ->103      0.13046  
                   98 ->103      -0.40341  
                   98 ->104      0.41062  
                   101 ->106      0.14177  
                   102 ->107      0.21832

Excited State 18: Singlet-A 5.1661 eV 240.00 nm f=0.0103 <S\*\*2>=0.000  
                   96 ->103      0.11561  
                   98 ->103      0.19612  
                   98 ->104      -0.19269  
                   99 ->105      0.19858  
                   100 ->105      0.13526  
                   101 ->106      0.21207  
                   102 ->107      0.48329

Excited State 19: Singlet-A 5.2275 eV 237.18 nm f=0.0046 <S\*\*2>=0.000  
                   99 ->105      -0.11432  
                   101 ->106      -0.15409  
                   102 ->108      0.41337

102 ->109      0.37135  
 102 ->110      0.31802  
 Excited State 20: Singlet-A    5.3686 eV 230.95 nm f=0.0632 <S\*\*2>=0.000  
 96 ->104      -0.12113  
 97 ->104      0.29231  
 101 ->106      0.14406  
 102 ->107      -0.27284  
 102 ->108      -0.25432  
 102 ->109      0.41206  
 102 ->110      0.14264  
 SavETr: write IOETrn= 770 NScale= 10 NData= 16 NLR=1 LETran= 370.

## 10. Benchmark of both vertical excitation and emission energy

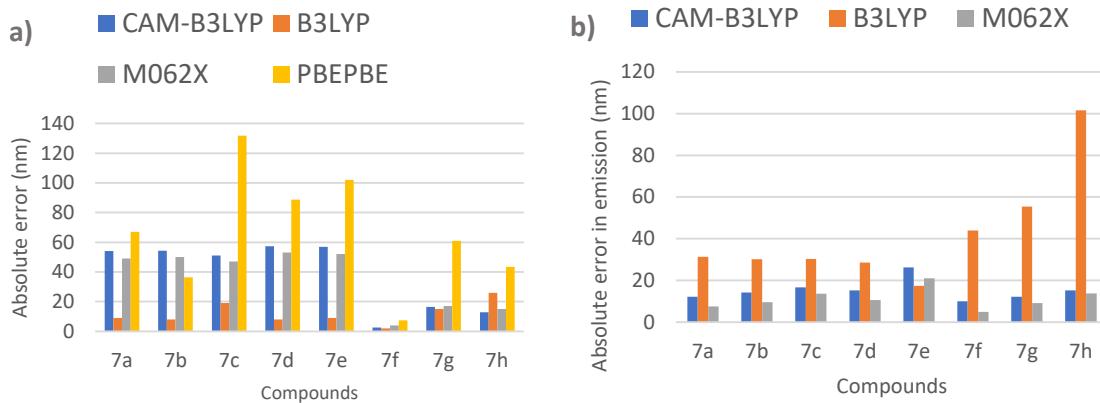


Figure SI-133. Absolute error in nm between the experimental data and theoretical data of compounds 7a-h. a) absorption. b) emission.

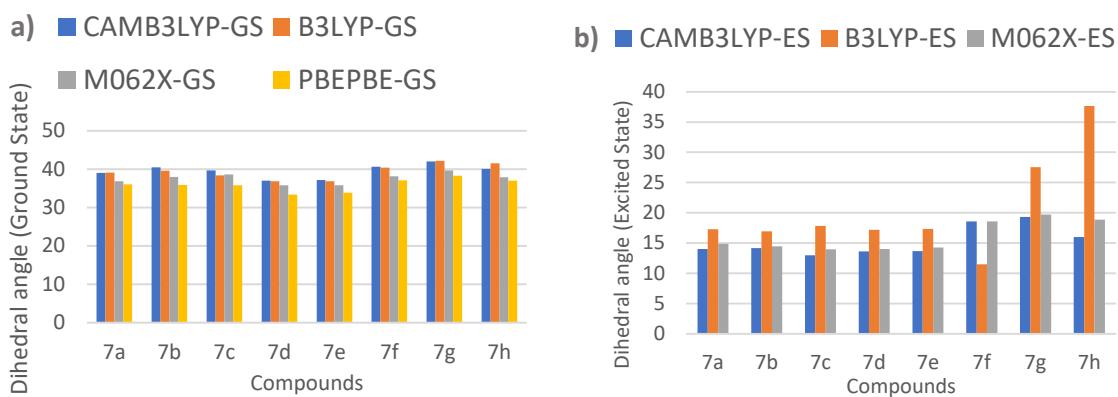


Figure SI-134. Dihedral angles of compound 7a-h. a) Ground State and b) Excited State.

Theoretical lineal and non-lineal properties analysis.

The total static dipole moment  $\mu$  was determined by the square root of the quadratic sum of the components of the dipole moment. (Ec. 1)

$$\mu = \sqrt{\mu_x^2 + \mu_y^2 + \mu_z^2} \quad (1).$$

The isotropic polarizability  $\alpha$  was calculated from the trace of the polarization tensor (Ec. 2)

$$\alpha = (\alpha_{xx} + \alpha_{yy} + \alpha_{zz}) \quad (2).$$

The mean/static first hyperpolarizability  $\beta_0$  is expressed by (Ec. 3 and Ec. 4):

$$\beta_0 = (\beta_x^2 + \beta_y^2 + \beta_z^2) \quad (3).$$

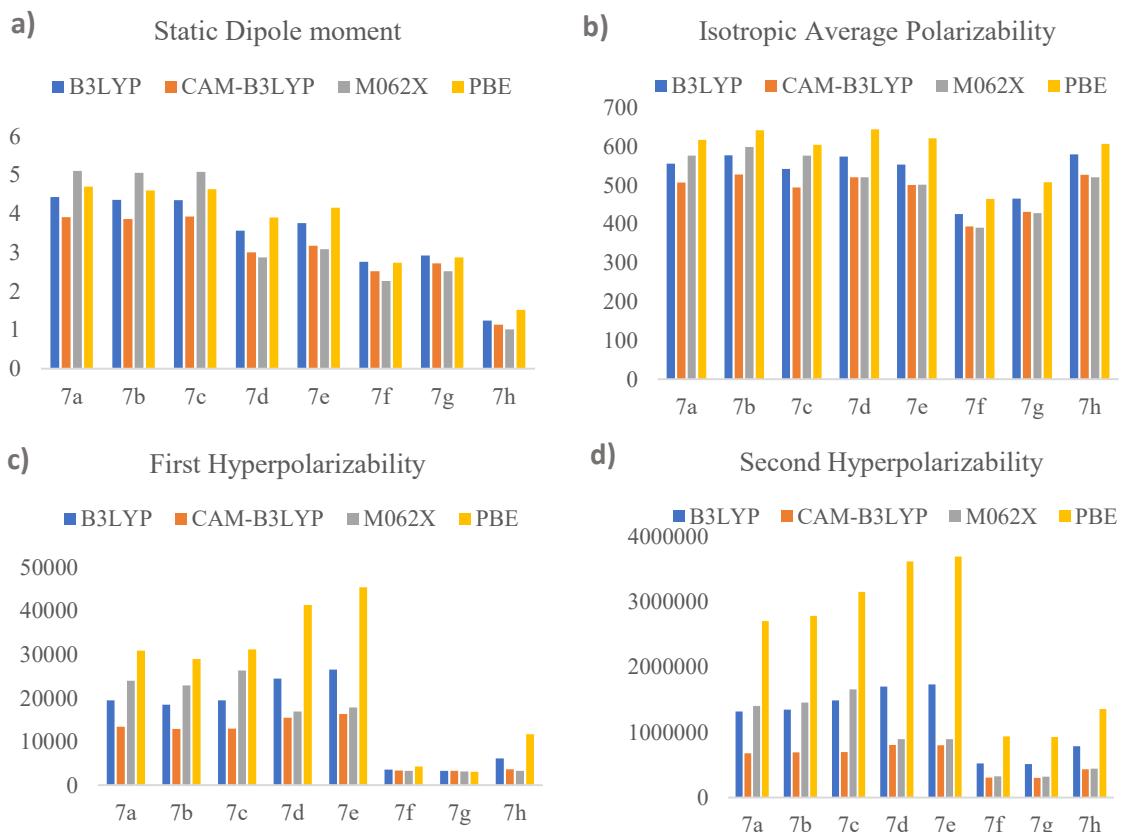
$$\beta_0 = [(\beta_{xxx} + \beta_{xyy} + \beta_{xzz})^2 + (\beta_{yxx} + \beta_{yyy} + \beta_{yzz})^2 + (\beta_{zxx} + \beta_{zyy} + \beta_{zzz})^2]^{1/2} \quad (4).$$

$\beta_x, \beta_y, \beta_z$  are the components of the second order polarizability tensor along the x, y and z axes.

The mean second hyperpolarizability  $\gamma$  is expressed by (Ec. 5):

$$\gamma = \frac{1}{5}[(\gamma_{xxxx} + \gamma_{yyyy} + \gamma_{zzzz}) + 2(\gamma_{xxyy} + \gamma_{yyzz} + \gamma_{zzxx})] \quad (5)$$

Where,  $\gamma_{xxxx} + \gamma_{yyyy} + \gamma_{zzzz}$  are the second order tensor components.



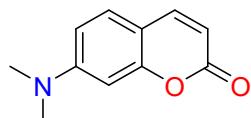
**Figure SI-135.** Theoretical calculations of the nonlinear properties of compounds 7a-h. a) static dipole moment, b) isotropic average polarizability, c) first hyperpolarizability, d) second hyperpolarizability.

## 11. Static dipole moment, isotropic average polarizability, first hyperpolarizability and second hyperpolarizability of compound 7a-h

Molecule	Method	$\mu^{[a]}$	$\alpha_0^{[b]}$	$\beta_0^{[c]}$	$\gamma_0^{[d]}$
----------	--------	-------------	------------------	-----------------	------------------

		a.u.	a.u.	a.u.	a.u.
7a	B3LYP	4.432623	555.63933	19543.4411	1321884
	CAM-B3LYP	3.914257	507.35733	13443.4485	680356
	M062X	5.104336	576.77467	24036.2611	1407060
	PBE	4.700977	616.73933	30914.4481	2710851
7b	B3LYP	4.354041	577.35967	18479.5701	1347974
	CAM-B3LYP	3.862711	527.47567	12915.161	690913
	M062X	5.057873	598.66267	22914.8533	1458239
	PBE	4.602749	642.04333	28973.818	2787777
7c	B3LYP	4.353766	542.43967	19527.8854	1490650
	CAM-B3LYP	3.926845	494.01667	13051.1336	696898
	M062X	5.081003	576.205	26380.3665	1658576
	PBE	4.630039	604.64	31186.907	3157246
7d	B3LYP	3.565883	573.94967	24516.1865	1703297
	CAM-B3LYP	3.004084	520.55433	15506.4538	804267
	M062X	2.875138	520.797	16935.3404	897191
	PBE	3.906009	644.51933	41423.9897	3622888
7e	B3LYP	3.759327	553.42	26601.8673	1737046
	CAM-B3LYP	3.17446	501.02	16338.8322	801496
	M062X	3.079435	501.34867	17861.6618	897517
	PBE	4.156943	620.923	45514.066	3697776
7f	B3LYP	2.759275	426.01	3559.797	524577
	CAM-B3LYP	2.514013	393.61833	3413.61965	303890
	M062X	2.265561	390.99467	3323.69175	323841
	PBE	2.733683	464.72633	4317.22768	939318
7g	B3LYP	2.918466	465.27867	3283.88806	514166
	CAM-B3LYP	2.714823	431.443	3289.8928	302766
	M062X	2.513504	428.53033	3187.23534	318663
	PBE	2.869619	507.986	3077.76374	928800
7h	B3LYP	1.241644	579.221	6172.80369	786579
	CAM-B3LYP	1.136127	526.82333	3654.19197	434853
	M062X	1.011368	520.54533	3329.57635	444456
	PBE	1.516401	607.08767	11720.1685	1359374
<b>Coumarin</b>	B3LYP	4.269110	263.95367	7500.13424	207626
<b>Imidazo</b>	B3LYP	0.923603	240.06700	4884.65540	170203

## 12. Equilibrium Geometry in Ground State, Excitation energies and oscillator strength of coumarin fragment (B3LYP)



Coumarin  
B3LYP  
ABSORPTION

Excitation energies and oscillator strengths:

Excited State 1: Singlet-A 3.3739 eV 367.48 nm f=0.5454 <S\*\*2>=0.000  
58 -> 59 0.69967

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -709.706366173

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A 4.2085 eV 294.60 nm f=0.0181 <S\*\*2>=0.000  
57 -> 59 0.41631  
58 -> 60 0.55516

Excited State 3: Singlet-A 4.4967 eV 275.72 nm f=0.0552 <S\*\*2>=0.000  
57 -> 59 0.56148  
58 -> 60 -0.41520

Excited State 4: Singlet-A 4.7538 eV 260.81 nm f=0.0133 <S\*\*2>=0.000  
58 -> 61 0.69482

Excited State 5: Singlet-A 4.8825 eV 253.93 nm f=0.0005 <S\*\*2>=0.000  
55 -> 59 0.69032  
55 -> 62 0.12182

Excited State 6: Singlet-A 4.9091 eV 252.56 nm f=0.1407 <S\*\*2>=0.000  
56 -> 59 -0.18291  
58 -> 62 0.66426

Excited State 7: Singlet-A 5.0593 eV 245.06 nm f=0.0061 <S\*\*2>=0.000  
58 -> 63 0.69700

Excited State 8: Singlet-A 5.2139 eV 237.79 nm f=0.0347 <S\*\*2>=0.000  
56 -> 59 -0.38800  
58 -> 64 0.57477

Excited State 9: Singlet-A 5.2975 eV 234.04 nm f=0.0372 <S\*\*2>=0.000  
56 -> 59 0.37490

		58 -> 64	0.22879
		58 -> 65	0.52308
Excited State 10:	Singlet-A	5.3889 eV 230.07 nm f=0.0963 <S**2>=0.000	
		56 -> 59	-0.36790
		58 -> 62	-0.15460
		58 -> 64	-0.29676
		58 -> 65	0.45725
		58 -> 67	-0.10574
Excited State 11:	Singlet-A	5.4754 eV 226.44 nm f=0.0012 <S**2>=0.000	
		58 -> 66	0.68575
Excited State 12:	Singlet-A	5.6555 eV 219.23 nm f=0.0079 <S**2>=0.000	
		57 -> 60	-0.10205
		58 -> 67	0.67129
Excited State 13:	Singlet-A	5.8526 eV 211.84 nm f=0.0037 <S**2>=0.000	
		58 -> 67	-0.11033
		58 -> 68	0.34019
		58 -> 69	0.58380
Excited State 14:	Singlet-A	5.9180 eV 209.50 nm f=0.0182 <S**2>=0.000	
		56 -> 60	-0.10057
		57 -> 62	0.17209
		58 -> 68	0.57137
		58 -> 69	-0.31883
Excited State 15:	Singlet-A	5.9854 eV 207.14 nm f=0.3450 <S**2>=0.000	
		56 -> 60	0.26332
		57 -> 60	0.58389
		57 -> 62	-0.14658
		58 -> 68	0.16004
Excited State 16:	Singlet-A	6.0995 eV 203.27 nm f=0.0146 <S**2>=0.000	
		57 -> 61	0.62651
		57 -> 63	0.15764
		57 -> 66	-0.12056
		58 -> 70	0.18022
		58 -> 71	-0.10156
Excited State 17:	Singlet-A	6.1123 eV 202.84 nm f=0.5304 <S**2>=0.000	
		56 -> 60	-0.24477
		57 -> 60	0.27064
		57 -> 62	0.54671
		58 -> 68	-0.13627

Excited State 18: Singlet-A 6.1584 eV 201.33 nm f=0.0028 <S\*\*2>=0.000  
 57 -> 61 0.14126  
 58 -> 70 -0.12316  
 58 -> 71 0.66363

Excited State 19: Singlet-A 6.1835 eV 200.51 nm f=0.0046 <S\*\*2>=0.000  
 57 -> 61 -0.13097  
 58 -> 70 0.57038  
 58 -> 71 0.12969  
 58 -> 72 -0.32991

Excited State 20: Singlet-A 6.1896 eV 200.31 nm f=0.0045 <S\*\*2>=0.000  
 58 -> 70 0.30620  
 58 -> 72 0.61437

Standard orientation:

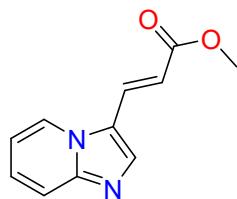
Center	Atomic	Atomic	Coordinates (Angstroms)		
	Number	Type	X	Y	Z
1	6	0	-1.135459	-0.228016	-0.061026
2	6	0	-0.040185	0.649861	0.124883
3	6	0	1.254569	0.168681	0.056631
4	6	0	1.555276	-1.182723	-0.199396
5	6	0	0.460237	-2.051662	-0.385482
6	6	0	-0.838276	-1.602757	-0.319391
7	1	0	-0.177760	1.705531	0.304413
8	1	0	0.652317	-3.101601	-0.578298
9	1	0	-1.638209	-2.315810	-0.451852
10	6	0	2.925275	-1.565119	-0.251566
11	6	0	3.917651	-0.652470	-0.062990
12	6	0	3.613049	0.731886	0.195745
13	8	0	2.266443	1.077893	0.243544
14	1	0	3.172393	-2.603763	-0.446637
15	7	0	-2.426145	0.218586	0.005710
16	6	0	-3.571352	-0.651367	-0.287618
17	1	0	-4.361749	-0.014407	-0.691051
18	1	0	-3.306458	-1.342458	-1.089462
19	6	0	-4.096116	-1.413374	0.933321
20	1	0	-3.335905	-2.081341	1.344707
21	1	0	-4.404193	-0.722253	1.722117
22	1	0	-4.964549	-2.015787	0.652726
23	6	0	-2.749528	1.602749	0.368163
24	1	0	-3.729567	1.587984	0.850063
25	1	0	-2.047197	1.951109	1.127866
26	6	0	-2.777310	2.563260	-0.825383
27	1	0	-3.048486	3.567879	-0.489450

28	1	0	-1.803624	2.619386	-1.317159
29	1	0	-3.514634	2.243299	-1.566077
30	8	0	4.410899	1.634218	0.377850
31	1	0	4.964911	-0.918457	-0.098382

---

Rotational constants (GHZ): 1.1060997 0.2905825 0.2396559

### 13. Equilibrium Geometry in Ground State, Excitation energies and oscillator strength of imidazo fragment (B3LYP)



Imidazo

B3LYP

ABSOPRTION

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	3.980758	0.176590	0.000032
2	6	0	4.062846	-1.195364	0.000032
3	6	0	2.882014	-1.977729	0.000014
4	6	0	1.652806	-1.380841	-0.000002
5	6	0	2.716010	0.790382	0.000015
6	1	0	4.858829	0.808767	0.000046
7	1	0	5.029256	-1.683542	0.000045
8	1	0	2.934361	-3.058031	0.000015
9	1	0	0.732745	-1.938832	-0.000014
10	7	0	1.565486	-0.010476	-0.000002
11	7	0	2.397931	2.095358	0.000015
12	6	0	1.053889	2.133583	-0.000013
13	1	0	0.509481	3.067786	-0.000021
14	6	0	0.471534	0.862944	-0.000014
15	6	0	-0.928200	0.586611	-0.000030
16	1	0	-1.520847	1.496603	-0.000037
17	6	0	-1.614474	-0.581104	-0.000039
18	1	0	-1.147329	-1.554412	-0.000033
19	6	0	-3.077898	-0.653983	-0.000060
20	8	0	-3.699273	-1.705800	-0.000016
21	8	0	-3.689292	0.549934	-0.000004
22	6	0	-5.130567	0.544647	0.000047

23	1	0	-5.510512	0.044036	0.891178
24	1	0	-5.423117	1.591875	0.000094
25	1	0	-5.510574	0.044099	-0.891093

-----  
Rotational constants (GHZ): 1.5654195 0.2803146 0.2381013

Excitation energies and oscillator strengths:

Excited State 1: Singlet-A 3.6384 eV 340.76 nm f=0.6873 <S\*\*2>=0.000  
                   53 -> 54      0.69900

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -685.201122417

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A 3.9443 eV 314.34 nm f=0.0267 <S\*\*2>=0.000  
                   52 -> 54      0.27018  
                   53 -> 55      0.64457

Excited State 3: Singlet-A 4.6482 eV 266.74 nm f=0.0001 <S\*\*2>=0.000  
                   51 -> 54      0.68424  
                   51 -> 56      -0.10735

Excited State 4: Singlet-A 4.6790 eV 264.98 nm f=0.3301 <S\*\*2>=0.000  
                   52 -> 54      0.63920  
                   53 -> 55      -0.25919

Excited State 5: Singlet-A 5.0662 eV 244.73 nm f=0.1370 <S\*\*2>=0.000  
                   52 -> 55      -0.48198  
                   53 -> 56      0.49989

Excited State 6: Singlet-A 5.2135 eV 237.81 nm f=0.0001 <S\*\*2>=0.000  
                   50 -> 54      0.68910

Excited State 7: Singlet-A 5.2274 eV 237.18 nm f=0.0010 <S\*\*2>=0.000  
                   53 -> 57      0.68208

Excited State 8: Singlet-A 5.5356 eV 223.98 nm f=0.2040 <S\*\*2>=0.000  
                   49 -> 54      -0.38802  
                   52 -> 55      0.41116  
                   53 -> 56      0.40119

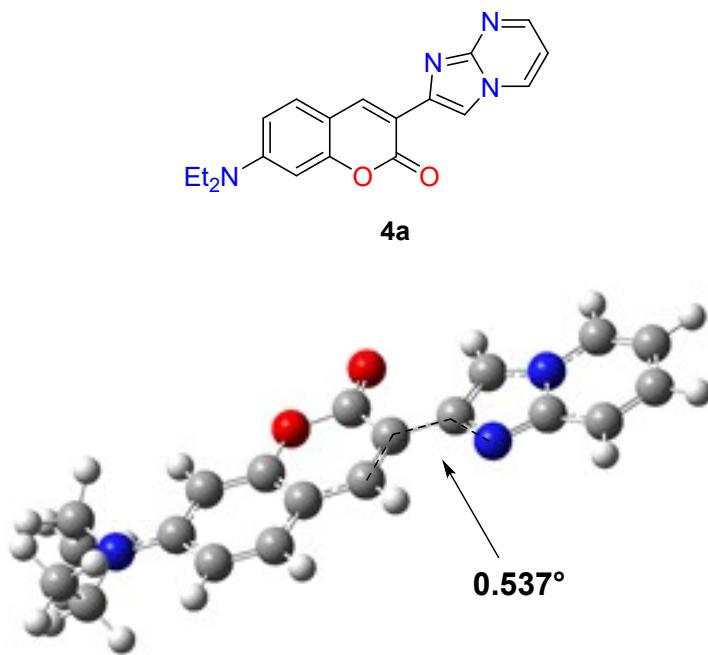
Excited State 9: Singlet-A 5.5672 eV 222.71 nm f=0.0009 <S\*\*2>=0.000  
                   53 -> 58      0.67844  
                   53 -> 59      -0.12009  
                   53 -> 60      0.10180

Excited State 10: Singlet-A 5.6753 eV 218.46 nm f=0.0062 <S\*\*2>=0.000

		53 -> 58	0.10502
		53 -> 59	0.68657
Excited State 11:	Singlet-A	5.7042 eV 217.36 nm f=0.0004 <S**2>=0.000	
		50 -> 55	0.60363
		51 -> 55	-0.35137
Excited State 12:	Singlet-A	5.8283 eV 212.73 nm f=0.0100 <S**2>=0.000	
		48 -> 54	0.57118
		49 -> 54	-0.36324
		52 -> 55	-0.13463
Excited State 13:	Singlet-A	5.9410 eV 208.69 nm f=0.0024 <S**2>=0.000	
		53 -> 57	0.12091
		53 -> 60	0.67668
Excited State 14:	Singlet-A	6.0028 eV 206.54 nm f=0.0000 <S**2>=0.000	
		50 -> 55	0.35123
		51 -> 55	0.60473
Excited State 15:	Singlet-A	6.1239 eV 202.46 nm f=0.1697 <S**2>=0.000	
		48 -> 54	0.36924
		49 -> 54	0.40474
		52 -> 55	0.21413
		52 -> 56	0.16368
		53 -> 56	0.19734
		53 -> 63	-0.10855
		53 -> 65	0.17915
Excited State 16:	Singlet-A	6.1652 eV 201.10 nm f=0.0010 <S**2>=0.000	
		53 -> 61	0.67692
		53 -> 64	0.15342
Excited State 17:	Singlet-A	6.2067 eV 199.76 nm f=0.0047 <S**2>=0.000	
		48 -> 55	0.11054
		49 -> 55	0.52314
		52 -> 56	0.43980
Excited State 18:	Singlet-A	6.3439 eV 195.44 nm f=0.0060 <S**2>=0.000	
		52 -> 57	0.60775
		52 -> 58	0.20696
		53 -> 62	-0.25394
Excited State 19:	Singlet-A	6.3882 eV 194.08 nm f=0.0001 <S**2>=0.000	
		52 -> 57	0.15897
		53 -> 62	0.53833
		53 -> 64	-0.40002

Excited State 20: Singlet-A 6.4025 eV 193.65 nm f=0.0019 <S\*\*2>=0.000  
 53 -> 63 0.62761  
 53 -> 67 -0.29728

#### 14. Equilibrium Geometry in Ground State and Dihedral angles of compound 4a



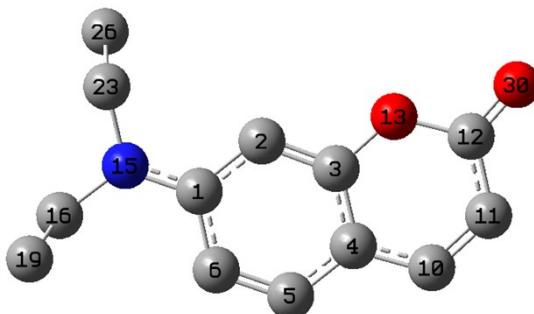
Standard orientation:

Center	Atomic Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
				X	Y	Z
1	6	0	3.740944	-0.315076	0.045668	
2	6	0	2.851468	0.766559	-0.099312	
3	6	0	1.491022	0.543357	-0.076026	
4	6	0	0.926653	-0.722617	0.091625	
5	6	0	1.819570	-1.798305	0.237187	
6	6	0	3.176096	-1.612291	0.214872	
7	1	0	3.196482	1.782854	-0.210486	
8	1	0	1.418796	-2.797682	0.362100	
9	1	0	3.818071	-2.474413	0.311164	
10	6	0	-0.487771	-0.829106	0.103495	
11	6	0	-1.295159	0.255570	-0.038531	
12	6	0	-0.691950	1.566656	-0.207903	
13	8	0	0.679408	1.632157	-0.216894	
14	1	0	-0.944557	-1.803531	0.230652	
15	6	0	-3.708580	1.111227	-0.140156	

16	1	0	-3.631890	2.174385	-0.267940
17	6	0	-4.642046	-0.890921	0.102121
18	7	0	-3.337984	-1.103576	0.126392
19	6	0	-2.753926	0.127695	-0.022724
20	7	0	-4.918979	0.461073	-0.060355
21	6	0	-5.729119	-1.790122	0.211223
22	6	0	-7.000577	-1.305611	0.154069
23	1	0	-7.842775	-1.980674	0.235992
24	6	0	-7.238972	0.087606	-0.013386
25	1	0	-8.248448	0.471442	-0.058542
26	6	0	-6.198925	0.949304	-0.118028
27	1	0	-6.301855	2.017289	-0.246290
28	1	0	-5.517767	-2.843316	0.337636
29	7	0	5.090699	-0.131927	0.022857
30	6	0	6.027454	-1.218959	0.292482
31	1	0	6.920834	-0.770655	0.730736
32	1	0	5.615636	-1.873142	1.061717
33	6	0	6.409705	-2.017304	-0.949455
34	1	0	5.538406	-2.498039	-1.397616
35	1	0	6.863701	-1.369485	-1.702314
36	1	0	7.132748	-2.793774	-0.690256
37	6	0	5.684755	1.170471	-0.260766
38	1	0	6.654692	0.987500	-0.726526
39	1	0	5.085348	1.685924	-1.012471
40	6	0	5.864264	2.041835	0.978301
41	1	0	6.330274	2.991476	0.706595
42	1	0	4.907176	2.256299	1.456794
43	1	0	6.505837	1.547143	1.710565
44	8	0	-1.280195	2.616793	-0.343393

-----  
 Rotational constants (GHZ):    0.7725879    0.0793274    0.0728829

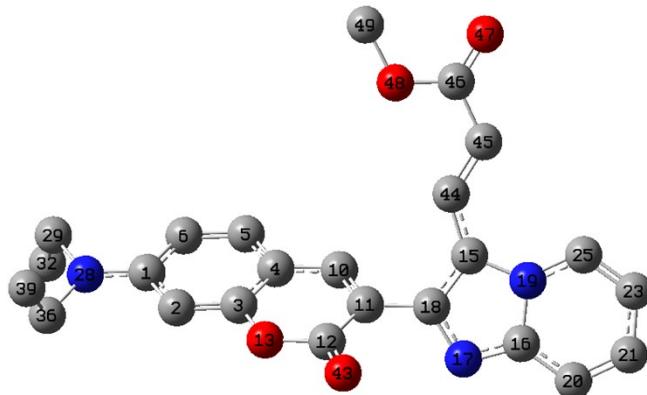
### Mulliken and Hirshfeld charges for the coumarin and imidazo[1,2-a]pyridine moieties



Coumarin moiety

Number	Atom	Charge*	
		Mulliken	Hirshfeld
1	C	-0.014534	0.066341
2	C	-0.347231	-0.033362
3	C	-1.495054	0.079459
4	C	1.512279	-0.042438
5	C	0.203971	0.024677
6	C	0.006306	-0.018571
10	C	-0.005721	0.063879
11	C	0.156157	-0.020614
12	C	0.16769	0.184711
13	O	-0.160629	-0.12186
15	N	0.238834	-0.070163
16	C	0.070358	0.107409
19	C	-0.007207	0.022907
23	C	0.087413	0.107233
26	C	-0.011412	0.023069
30	O	-0.401219	-0.372356

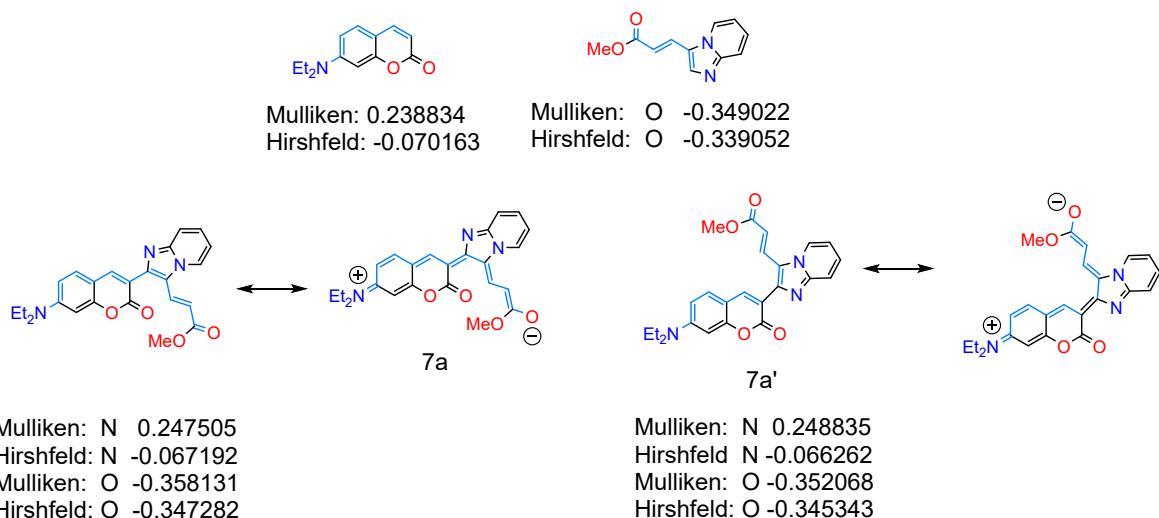
\*Hydrogens charges summed into heavy atoms



	Atom	7a		7a'	
		Charge*		Charge*	
		Mulliken	Hirshfeld	Mulliken	Hirshfeld
1	C	-0.174924	0.069881	0.014796	0.069088
2	C	-0.487775	-0.029374	-0.392624	-0.03
3	C	-1.043379	0.08248	-1.000349	0.08181
4	C	0.908072	-0.040431	0.725183	-0.040741
5	C	-0.088182	0.030438	0.037988	0.029593
6	C	0.167453	-0.013922	0.117326	-0.015042
10	C	0.414935	0.057506	0.206209	0.062593

11	C	0.63653	-0.038746	0.212857	-0.035184
12	C	-0.201706	0.193732	0.18416	0.190963
13	O	-0.122641	-0.11682	-0.147803	-0.118446
15	C	0.047747	0.014117	0.189537	0.016497
16	C	0.379125	0.100394	0.544243	0.100456
17	N	-0.037396	-0.252668	-0.048593	-0.256217
18	C	-0.12706	0.044478	-0.125209	0.0442
19	N	0.237738	-0.002802	0.24777	-0.002864
20	C	0.101523	0.026456	-0.02953	0.027236
21	C	-0.394565	0.037992	-0.345713	0.039083
23	C	0.062369	0.023529	0.001833	0.024274
25	C	0.004743	0.107857	0.016487	0.108161
28	N	0.248835	-0.066262	0.247505	-0.067192
29	C	0.081937	0.110137	0.085623	0.109515
32	C	-0.000301	0.02543	-0.002232	0.024834
36	C	0.080079	0.10996	0.084605	0.109419
39	C	-0.004009	0.025776	-0.003341	0.025022
43	O	-0.375488	-0.339651	-0.36087	-0.35796
44	C	0.070057	0.012359	-0.27982	0.023446
45	C	-0.107701	-0.038254	0.103909	-0.033608
46	C	-0.046217	0.199018	-0.055513	0.199992
47	O	-0.358131	-0.347282	-0.352068	-0.345343
48	O	-0.136235	-0.133012	-0.140274	-0.13273
49	C	0.264567	0.147413	0.263904	0.149186

\*Hydrogens charges summed into heavy atoms



**Scheme S1.** Resonances structures of the ICT, The N charges refers to Nitrogen in diethylamino in coumarin and the O chargers refers to carbonylic oxygen in acrylate.

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