

# Supporting Information

## Experimental and Computational Details

In order to select an appropriate density functional for the  $[ReCl_x]^+/\text{CH}_4$  reaction system, a benchmark for the bond dissociation energy (BDE) of  $[Re-\text{Cl}]^+$ 、 $[\text{Cl}Re-\text{Cl}]^+$ 、 $[\text{Cl}_2\text{Re}-\text{Cl}]^+$ 、 $[\text{Re}-\text{H}]^+$ 、 $[\text{Re}-\text{CH}_2]^+$ 、 $[\text{Re}-\text{CH}_3]^+$  was carried out. The bond energy of  $[\text{M}-\text{L}]^+$  was calculated with the formulation of  $\text{BDE}([\text{M}-\text{L}]^+) = E(\text{M}^+) + E(\text{L}) - E([\text{M}-\text{L}]^+)$ , where  $E(X)$  is composed of the correction to the enthalpy and the single-point energy (SPE). We first optimized the structures of  $\text{M}^+$ 、 $\text{L}$ 、 $[\text{M}-\text{L}]^+$  at B3LYP<sup>1</sup>-<sup>2</sup>/cc-PVDZ(pp)<sup>3-6</sup> level and calculated the zero-point vibrational energy (ZPVE) and thermal corrections to the enthalpy. The ZPVE and thermal contribution to enthalpy at 298.15K were corrected with the scaling factors of 0.9689 and 0.9784, respectively, to get the thermal corrections to enthalpy.<sup>7</sup> Then, the SPEs were calculated at CCSD(T)<sup>8, 9</sup>/CBS(aug-cc-pV[T:Q]Z(pp))<sup>10, 11</sup> level of theory using B3LYP/cc-PVDZ(pp). The  $\text{BDE}([\text{M}-\text{L}]^+)$  calculated with the thermal correction to enthalpy gained by B3LYP and SPE gained by CCSD(T) acts as the criterion  $\text{BDE}([\text{M}-\text{L}]^+)^0$ .

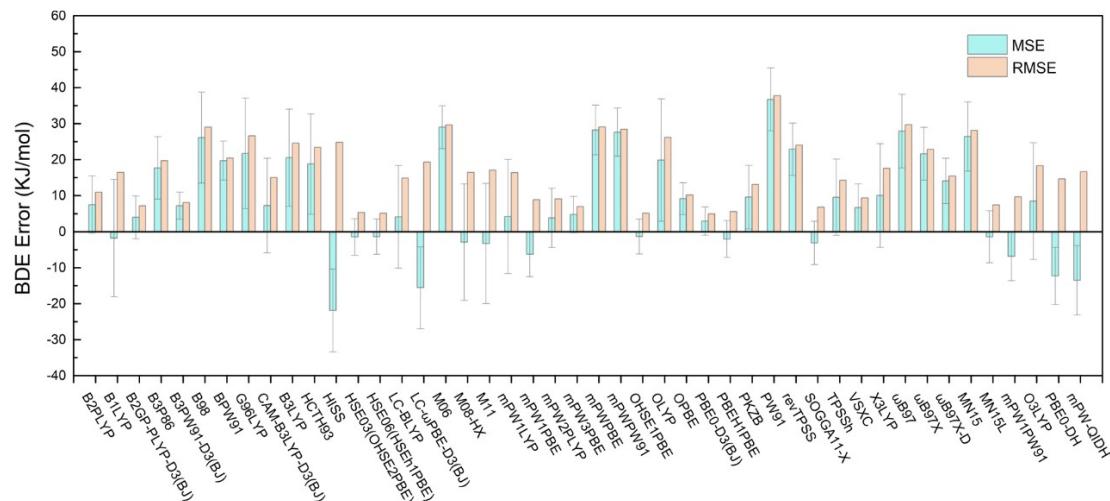
Next, the SPE calculations of  $\text{M}^+$ 、 $\text{L}$ 、 $[\text{M}-\text{L}]^+$  were performed by using the methods of B3LYP<sup>1, 2</sup>、B1LYP<sup>12</sup>、B2GP-PLYP<sup>13</sup>、B3P86<sup>14-16</sup>、B3PW91<sup>16, 17</sup>、B98<sup>18</sup>、BPW91<sup>15, 17</sup>、G96LYP<sup>19</sup>、CAM-B3LYP<sup>20</sup>、HCTH93<sup>21</sup>、HISS<sup>22, 23</sup>、HSE03(OHSE2PBE)<sup>24-26</sup>、HSE06(HSEh1PBE)<sup>24-26</sup>、LC-BLYP<sup>27</sup>、LC- $\omega$ pbe<sup>28</sup>、M06<sup>29</sup>、M08-HX<sup>30</sup>、M11<sup>31</sup>、mPW1PLYP<sup>32, 33</sup>、mPW1PBE<sup>32, 34</sup>、mPW2PLYP<sup>35</sup>、mPW3PBE<sup>32, 34</sup>、mPWPBE<sup>32, 34</sup>、mPWPW91<sup>17, 32</sup>、OHSE1PBE<sup>25, 26, 36-40</sup>、OLYP<sup>33, 41</sup>、OPBE<sup>34</sup>、PBE0<sup>42</sup>、PBEH1PBE<sup>43</sup>、PKZB<sup>44</sup>、PW91<sup>45</sup>、revTPSS<sup>46</sup>、SOGGA11-X<sup>47</sup>、TPSSh<sup>48</sup>、VSXC<sup>49</sup>、X3LYP<sup>50</sup>、 $\omega$ B97<sup>51</sup>、 $\omega$ B97X<sup>51</sup>、 $\omega$ B97X-D<sup>52</sup>、MN15<sup>53</sup>、MN15L<sup>54</sup>、mPW1PW91<sup>12</sup>、O3LYP<sup>55</sup>、PBE0-DH<sup>56</sup>、Mpw-QIDH<sup>57</sup> with the cc-PVDZ(pp) basis set and the ECP 60 pseudopotential<sup>58</sup> for Re atom. Besides, the methods of B2GP-PLYP, B3PW91, CAM-B3LYP, LC- $\omega$ pbe and PBE0 are performed with DFT-D3 dispersion correction.<sup>59, 60</sup> Once again, getting the energy of each elementary by combining the thermal correction to enthalpy gained by B3LYP with

SPE gained by each method, the BDE( $[M-L]^+$ ) of these six bonds were obtained. Taking BDE( $[M-L]^+$ )<sup>0</sup> as the benchmark, the average errors MSE and variances RMSE of six bonds' BDE for all computational methods are shown in Figure S1. Besides, Figure S2 demonstrates the normal distribution of BDE errors calculated by 15 methods with the small average errors.

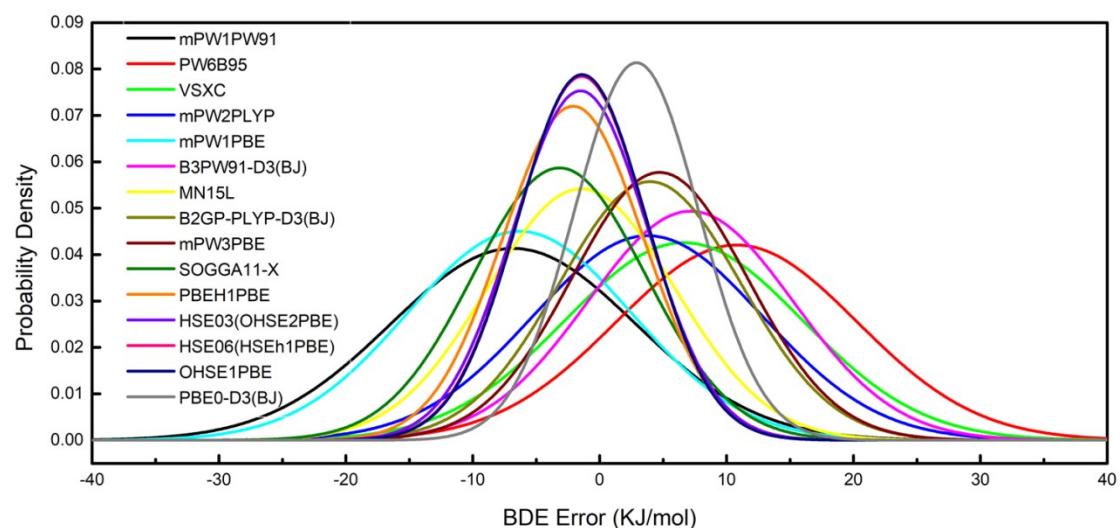
Above the analysis, considering the accuracy and precision of the computational method, PBE0-D3(BJ) is employed to optimize the structures. Natural bond orbital (NBO)<sup>61-66</sup> and quasi-restricted orbital (QRO)<sup>67, 68</sup> calculations were performed to obtain further information and the NBOs diagrams were painted by using Multifwn<sup>69</sup> and VMD<sup>70</sup> programs, respectively. The minimum energy crossing points (MECPs) were identified by means of the algorithm developed by Harvey et al.2 in ORCA 4.<sup>71</sup>

Deformation energy ( $\Delta E_{\text{def}}$ ) required for distorting the reactants to their geometries in TSs, is calculated as  $\Delta E_{\text{def}} = E(\text{CH}_4 + [\text{ReCl}_x]^+ \text{ at TS geometry}) - E(\text{EC})$ ;  $\Delta E_{\text{def}}$  and the respective reaction barrier ( $\Delta E^\ddagger$ ) were calculated at the CCSD(T)/BSII level. And the NPA charge of Re in the clusters are calculated at PBE0/def2-TZVP<sup>72</sup> level.

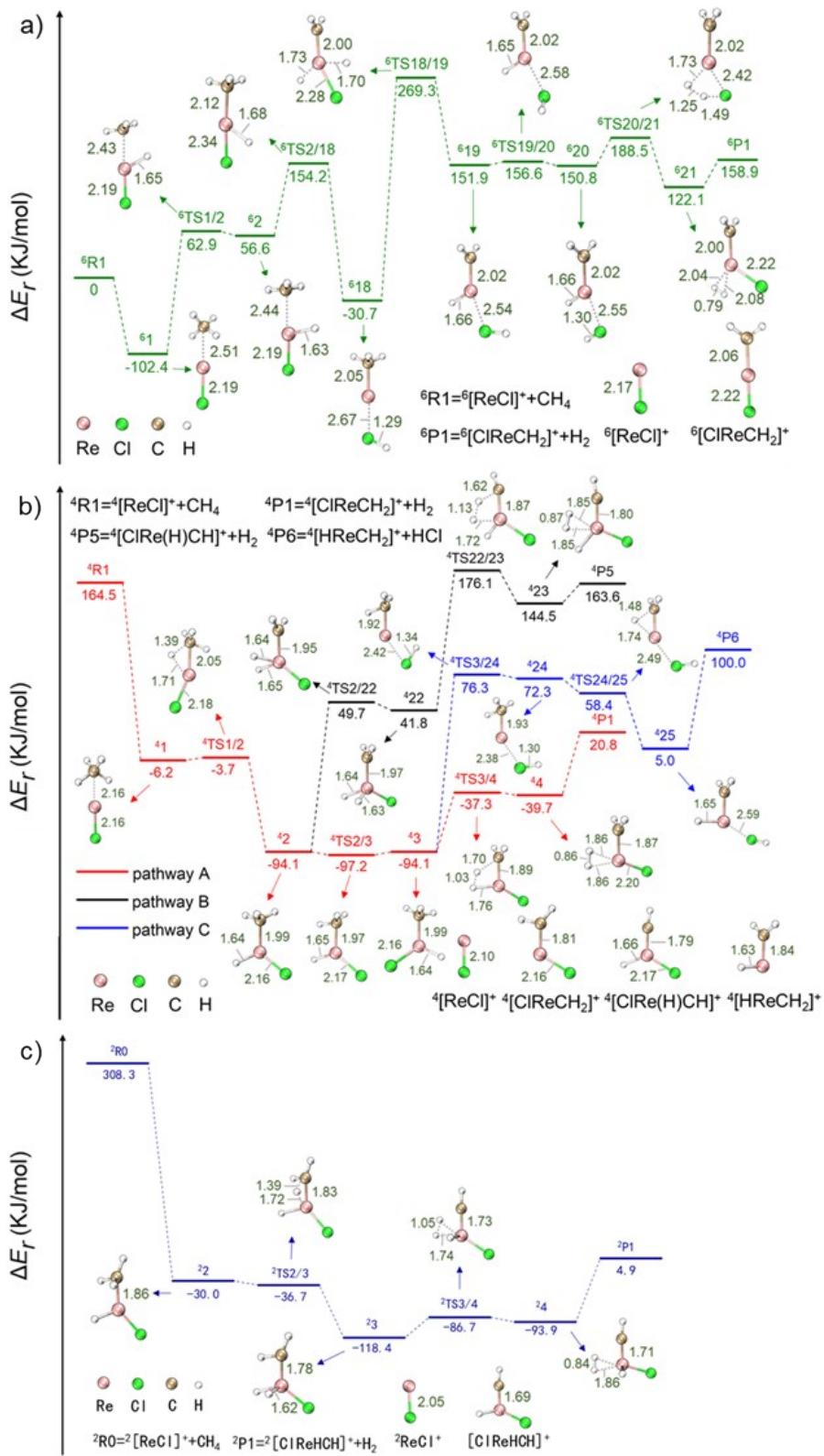
Different components of the total interaction energy of the ground state EC structures in  $[\text{ReCl}_x]^+$  ( $x = 1 - 3$ )/ $\text{CH}_4$  and  $[\text{ReCl}_2L]^+$  ( $L = \text{F}, \text{Cl}, \text{Br}$ )/ $\text{CH}_4$  systems are calculated at the PBE0-D3(BJ)/def2-TZVP level of theory by using GKS-EDA<sup>73</sup> method in GAMESS package and the input files for GAMESS are produced by MOKIT<sup>74</sup>. The ECP 60 pseudopotential for Re atom is modified to cut off the angular momentum, in which the pseudopotential only includes the definition of the part  $\geq g$ , s, d, f and the part  $\geq g$  is not affected by the pseudopotential. The total interaction energy ( $\Delta E^{\text{tot}}$ ) is decomposed to electrostatic energy ( $\Delta E^{\text{ele}}$ ), exchange energy ( $\Delta E^{\text{ex}}$ ), repulsion energy ( $\Delta E^{\text{rep}}$ ), polarization energy ( $\Delta E^{\text{pol}}$ ), grimme dispersion correction energy ( $\Delta E^{\text{disp}}$ ) and electron correlation energy ( $\Delta E^{\text{corr}}$ ), and Pauli repulsion energy ( $\Delta E^{\text{pauli}}$ ) is defined as the sum of exchange energy and repulsion energy.



**Figure S1.** The average errors MSE and variances RMSE of BDE( $[\text{Re}-\text{Cl}]^+$ ) $,$  BDE( $[\text{Cl}\text{Re}-\text{Cl}]^+$ ) $,$  BDE ( $[\text{Cl}_2\text{Re}-\text{Cl}]^+$ ) $,$  BDE ( $[\text{Re}-\text{H}]^+$ ) $,$  BDE ( $[\text{Re}-\text{CH}_2]^+$ ) $,$  BDE ( $[\text{Re}-\text{CH}_3]^+$ ) for each computational method

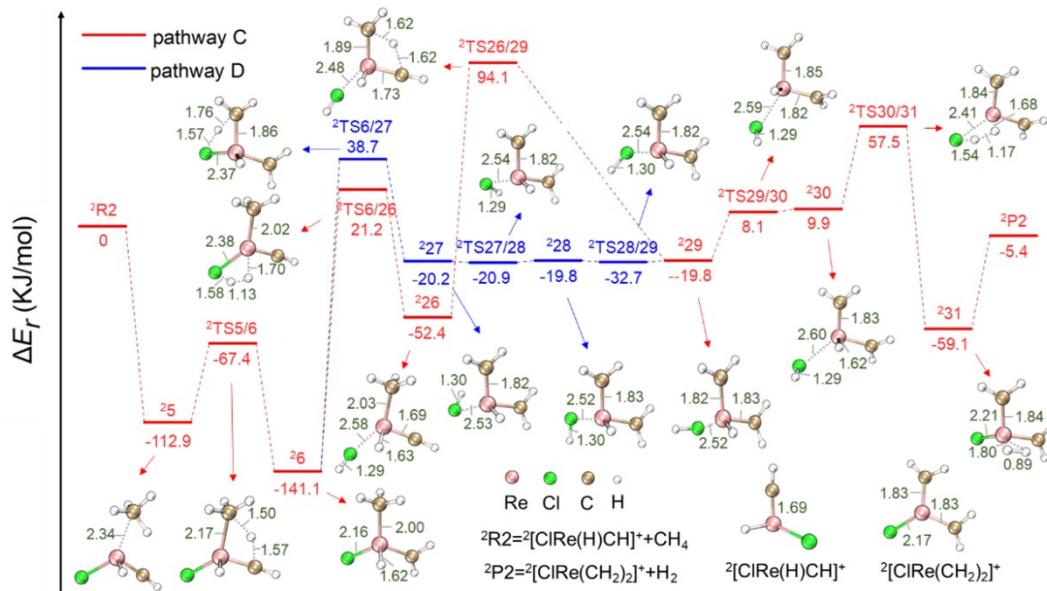


**Figure S2.** The normal distribution of BDE errors calculated by the methods of mPW1PW91、PW6B95、VSXC、mPW2PLYP、mPW1PBE、B3PW91-D3(BJ)、MN15L、B2GP-PLYP-D3(BJ)、PW3PBE、SOGGA11-X、PBEH1PBE、HSE03(OHSE2PBE)、HSE06(HSEh1PBE)、OHSE1PBE、PBE0-D3(BJ).

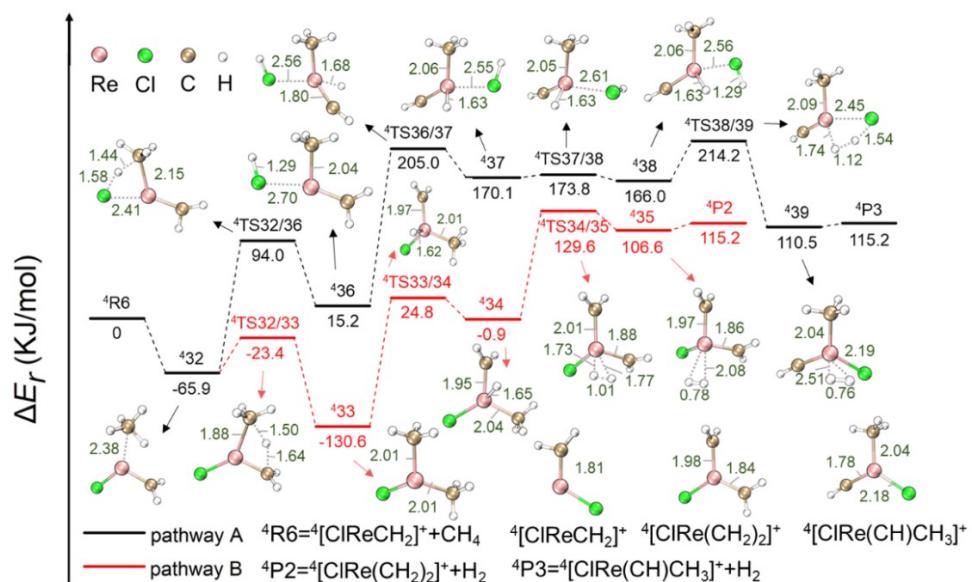


**Figure S3.** PES and selected structural information for the reaction of a)  ${}^6[\text{ReCl}]^+$ , b)  ${}^4[\text{ReCl}]^+$  and  ${}^2[\text{ReCl}]^+$  with  $\text{CH}_4$  as calculated at the CCSD(T)/BSII//PBE0-D3(BJ)/BSI level of theory. The relative energies of structures are given in KJ/mol and bond

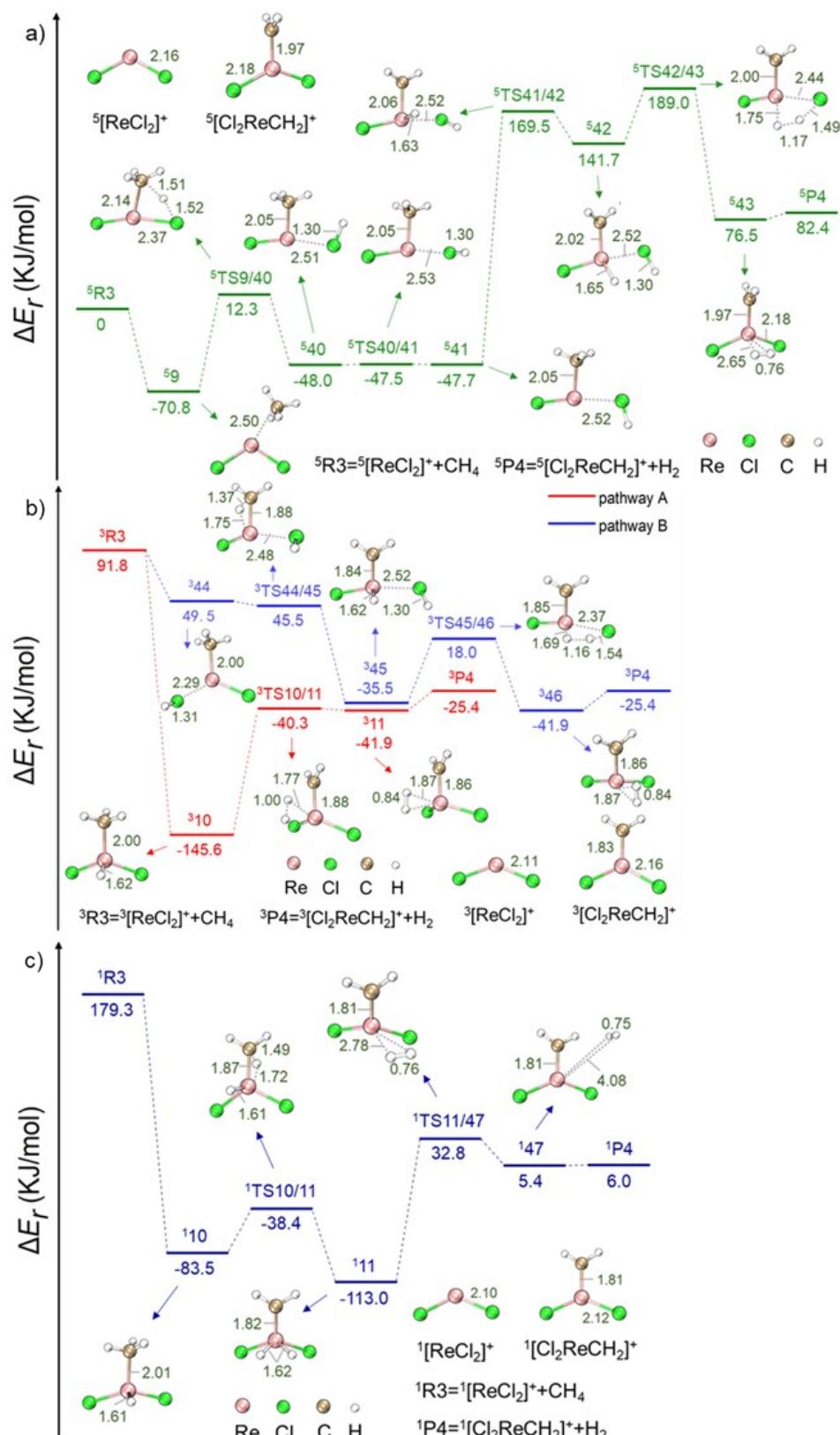
lengths are given in Å. Charges are omitted for the sake of clarity.



**Figure S4.** Reaction pathway B and C and selected structural information for the reaction of  $^2[\text{ClReHCH}]^+$  with  $\text{CH}_4$  as calculated at the CCSD(T)/BSII//PBE0-D3(BJ)/BSI level of theory. The relative energies of structures are given in KJ/mol and bond lengths are given in Å. Charges are omitted for the sake of clarity.

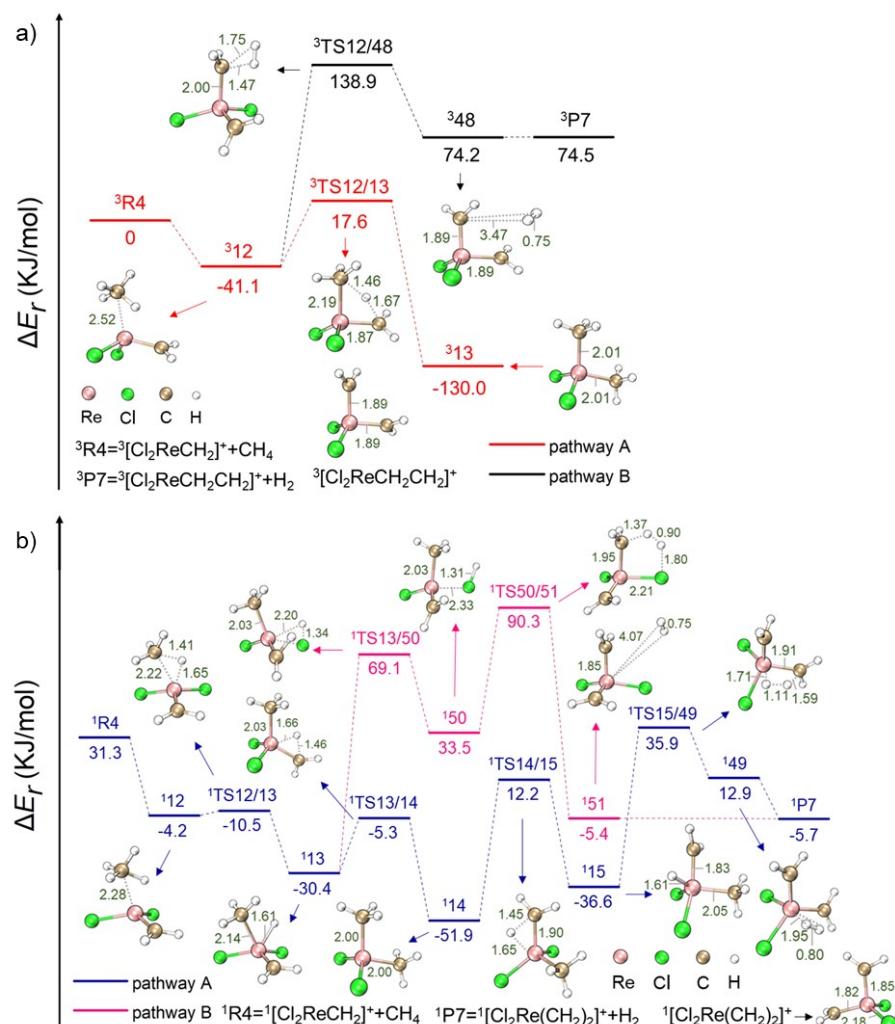


**Figure S5.** PES and selected structural information for the reaction of  $^4[\text{ClReCH}_2]^+$  with  $\text{CH}_4$  as calculated at the CCSD(T)/BSII//PBE0-D3(BJ)/BSI level of theory. The relative energies of structures are given in KJ/mol and bond lengths are given in Å. Charges are omitted for the sake of clarity.

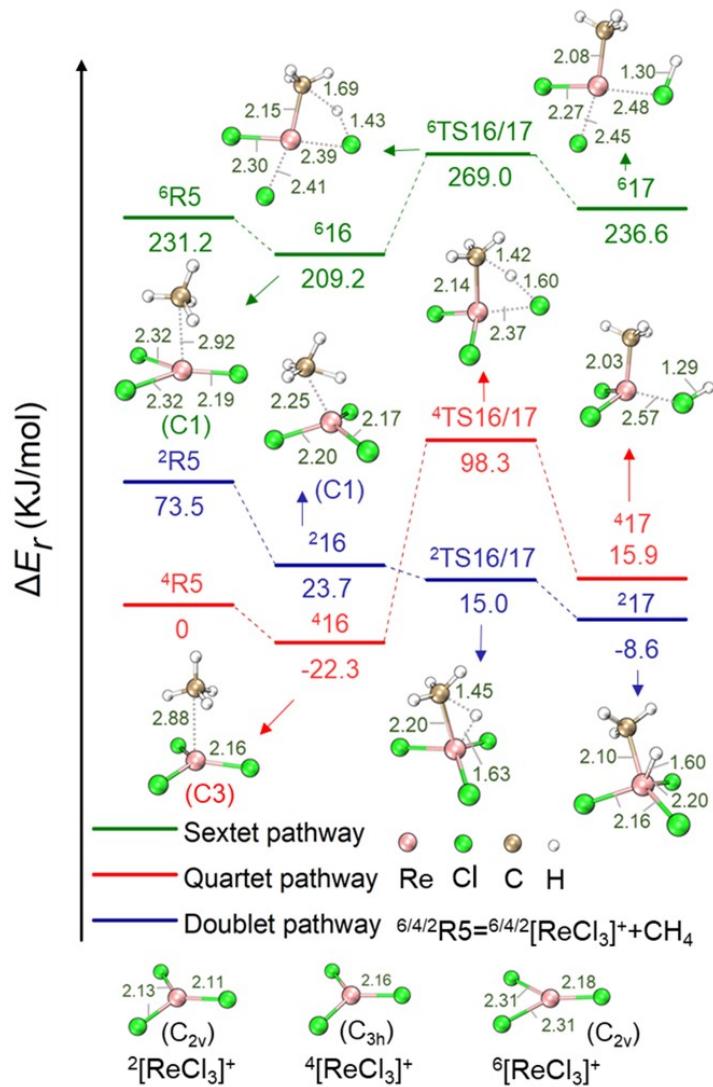


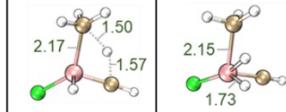
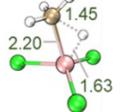
**Figure S6.** PES and selected structural information for the reaction of a)  ${}^5[\text{ReCl}_2]^+$ , b)  ${}^3[\text{ReCl}_2]^+$  and  ${}^1[\text{ReCl}_2]^+$  with  $\text{CH}_4$  as calculated at the CCSD(T)/BSII//PBE0-D3(BJ)/BSI level of theory. The relative energies of structures are given in KJ/mol

and bond lengths are given in Å. Charges are omitted for the sake of clarity.



**Figure S7.** PES and selected structural information for the reaction of a)  ${}^3[\text{ReCl}_2\text{CH}_2]^+$  and b)  ${}^1[\text{ReCl}_2\text{CH}_2]^+$  with  $\text{CH}_4$  as calculated at the CCSD(T)/BSII//PBE0-D3(BJ)/BSI level of theory. The relative energies of structures are given in KJ/mol and bond lengths are given in Å. Charges are omitted for the sake of clarity.



TS	<sup>4</sup> TS1/2	<sup>2</sup> TS5/6	<sup>2</sup> TS5/8	MECP3	<sup>1</sup> TS12/13	<sup>2</sup> TS16/17
structure						
$\rho_{\text{Me}}$	-0.01	0.02	0.08	-0.04	-	-0.01
$Q_{\text{Me}}$	0.03	0.02	0.04	0.00	0.08	0.11
$Q_{\text{H}}$	0.20	0.31	0.24	0.25	0.27	0.27
$Q_{\text{c}}$	0.77	0.68	0.72	0.65	0.65	0.62
$Q_{\text{CT}}$	0.23	0.32	0.28	0.35	0.35	0.38
$\Delta E_{\text{def}}$	371.2	325.6	367.4	211.5	254.9	279.3
$\Delta E^{\ddagger}$	98.7	45.5	65.1	61.5	35.8	37.3

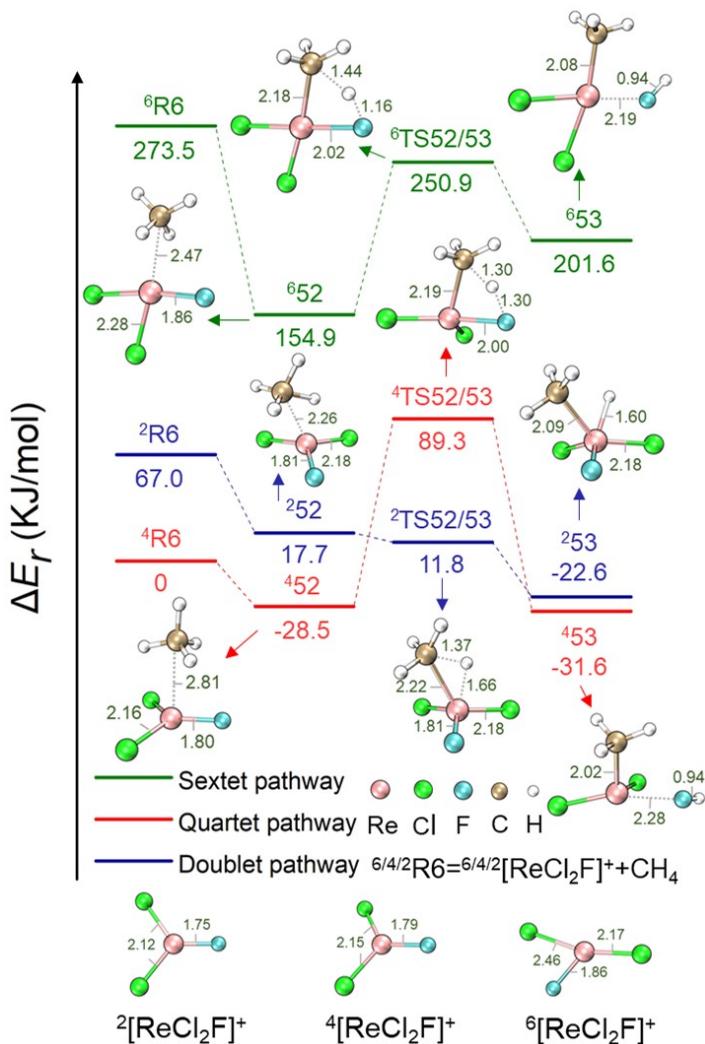
**Figure S9.** Schematic representations of the TSs together with NPA charges ( $Q$ ) and spin densities ( $\rho$ ) on the methyl group (Me), the charge on the transfer–hydrogen (H), cluster ( $Q_c$ ), the amount of charge transferred ( $Q_{\text{CT}}$ ) from methane to the cluster ion and the deformation energies (KJ/mol) of the reactants in the TSs ( $\Delta E_{\text{def}}$ ), the corresponding barriers ( $\Delta E^{\ddagger}$  ).

**Table S1.** Energy gap (eV) between the electron–donation orbital of methane and the electron-acception orbital of the cluster ( $\Delta E_{\text{DO}(\text{methane}) \rightarrow \text{AO}(\text{cluster})}$ ) for different systems.

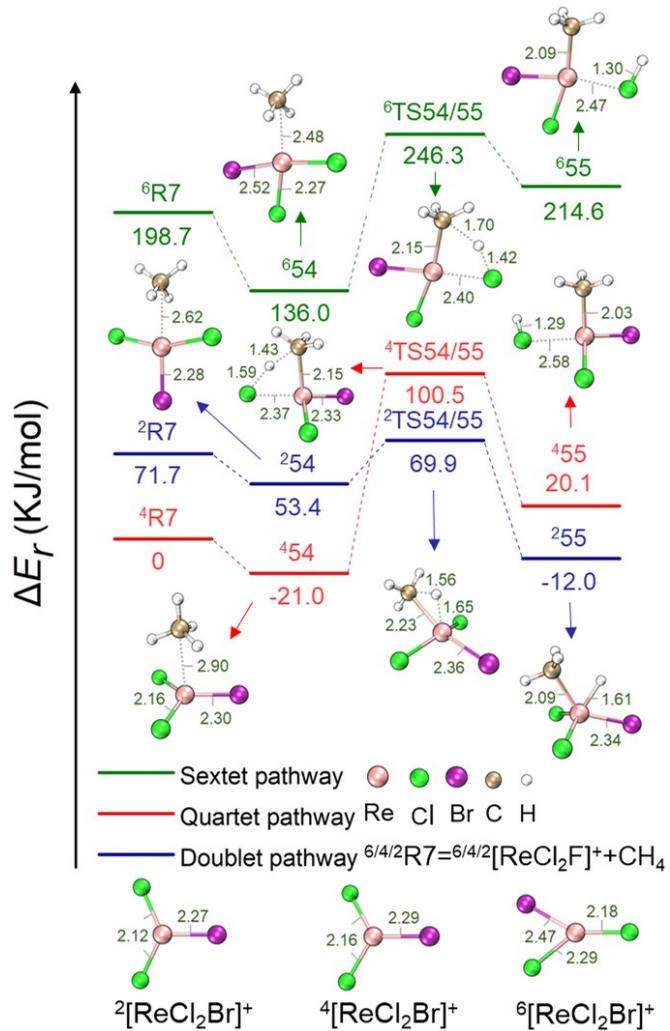
System	$\Delta E_{\text{DO}(\text{methane}) \rightarrow \text{AO}(\text{cluster})}$
<sup>6</sup> [ReCl] <sup>+</sup> /CH <sub>4</sub>	4.8
<sup>4</sup> [ReCl] <sup>+</sup> /CH <sub>4</sub>	1.4
<sup>5</sup> [ReCl <sub>2</sub> ] <sup>+</sup> /CH <sub>4</sub>	0.8
<sup>3</sup> [ReCl <sub>2</sub> ] <sup>+</sup> /CH <sub>4</sub>	0.6
<sup>4</sup> [ReCl <sub>3</sub> ] <sup>+</sup> /CH <sub>4</sub>	5.7
<sup>2</sup> [ReCl <sub>3</sub> ] <sup>+</sup> /CH <sub>4</sub>	0.6

**Table S2.** Decomposition of interaction energies (KJ/mol) in the ground state EC structures for [ReCl<sub>x</sub>]<sup>+</sup> ( $x = 1 - 3$ )/CH<sub>4</sub> systems as calculated at the PBE0-D3(BJ)/def2-TZVP level of theory.

System	$\Delta E^{\text{ele}}$	$\Delta E^{\text{pauli}}$	$\Delta E^{\text{pol}}$	$\Delta E^{\text{disp}}$	$\Delta E^{\text{corr}}$	$\Delta E^{\text{tot}}$
$[\text{ReCl}]^+/\text{CH}_4$	-53.2	165.8	-167.9	-6.3	-49.4	-110.5
$[\text{ReCl}_2]^+/\text{CH}_4$	-80.4	243.6	-195.1	-8.8	-49.8	-90.4
$[\text{ReCl}_3]^+/\text{CH}_4$	-33.9	82.9	-44.4	-10.9	-28.5	-34.3



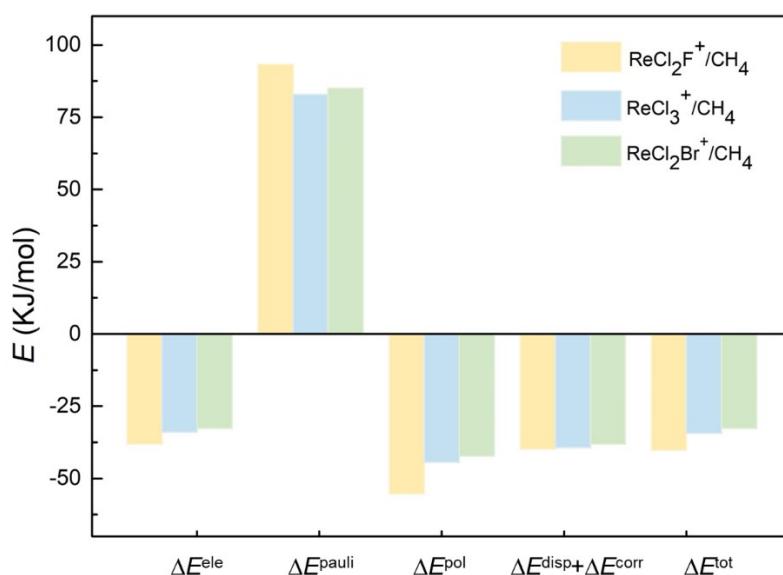
**Figure S10.** PES and selected structural information for the reaction of  $[\text{ReCl}_2\text{F}]^+$  with  $\text{CH}_4$  as calculated at the CCSD(T)/BSII//PBE0-D3(BJ)/BSI level of theory. The relative energies of structures are given in kJ/mol and bond lengths are given in Å. Charges are omitted for the sake of clarity.



**Figure S11.** PES and selected structural information for the reaction of  $[\text{ReCl}_2\text{Br}]^+$  with  $\text{CH}_4$  as calculated at the CCSD(T)/BSII//PBE0-D3(BJ)/BSI level of theory. The relative energies of structures are given in KJ/mol and bond lengths are given in Å. Charges are omitted for the sake of clarity.

**Table S3.** NPA charge of Re in  $[\text{ReCl}_2L]^+$  ( $L = \text{F}, \text{Cl}, \text{Br}$ ) as calculated at the PBE0/def2-TZVP level of theory.

Clusters	Spin state	Charge of Re
	6	1.25
$[\text{ReCl}_2\text{F}]^+$	4	1.29
	2	1.02
	6	0.93
$[\text{ReCl}_3]^+$	4	0.82
	2	0.50
	6	0.84
$[\text{ReCl}_2\text{Br}]^+$	4	0.69
	2	0.35



**Figure S12.** Different components of the total interaction energy of the ground state EC structures in  $[\text{ReCl}_2L]^+$  ( $L = \text{F}, \text{Cl}, \text{Br}$ )/ $\text{CH}_4$  systems as calculated at the PBE0-D3(BJ)/def2-TZVP level of theory using GKS-EDA method.

**Table S4.** Decomposition of interaction energies (KJ/mol) in the ground state EC structures for  $[\text{ReCl}_2L]^+$  ( $L = \text{F}, \text{Cl}, \text{Br}$ )/ $\text{CH}_4$  systems as calculated at the PBE0-D3(BJ)/def2-TZVP level of theory.

System	$\Delta E^{\text{ele}}$	$\Delta E^{\text{pauli}}$	$\Delta E^{\text{pol}}$	$\Delta E^{\text{disp}}$	$\Delta E^{\text{corr}}$	$\Delta E^{\text{tot}}$
$[\text{ReCl}_2\text{F}]^+/\text{CH}_4$	-38.1	93.3	-55.3	-9.6	-30.1	-40.2
$[\text{ReCl}_3]^+/\text{CH}_4$	-33.9	82.9	-44.4	-10.9	-28.5	-34.3
$[\text{ReCl}_2\text{Br}]^+/\text{CH}_4$	-32.7	85.0	-42.3	-10.9	-27.2	-32.7

## Cartesian coordinates

### <sup>6</sup>TS1/2

1 6 (charge multiplicity)

C	2.56381100	-0.05027100	-0.00080100
H	2.88144300	0.41799000	-0.92598100
H	0.97610700	1.41579800	0.00012200
H	2.43397200	-1.13726800	-0.00300200
H	2.88152900	0.41394100	0.92628400
Re	0.13833900	-0.01155000	0.00020500
Cl	-2.05478600	0.00337600	-0.00047000

### <sup>4</sup>TS1/2

1 4

Cl	-1.99057600	0.21728200	0.00012300
Re	0.16452400	-0.10608200	-0.00004900
C	2.15727400	0.39538600	0.00030500
H	1.11358400	1.31864900	-0.00244300
H	2.70935900	0.67287600	-0.89980800
H	2.70698900	0.67365400	0.90163800
H	2.02688400	-0.77510600	0.00036200

### <sup>4</sup>TS2/3

1 4

Cl	1.88923400	0.40424300	-0.00000300
Re	-0.19394100	-0.18921200	0.00000200
C	-1.76804400	0.99814400	-0.00000600
H	-0.61627700	-1.78689300	-0.00005900
H	-2.00340700	1.55281000	0.91155400
H	-2.00340300	1.55281200	-0.91156400
H	-2.34007300	0.01114200	-0.00000800

### <sup>2</sup>TS2/3

1 2

Cl	1.90093200	0.30152400	0.02147300
Re	-0.20274100	-0.13900200	-0.02465700
C	-1.76260300	0.81041400	-0.01217000
H	-0.64948400	-1.52608500	0.70607600
H	-2.41755400	1.66586600	0.11561200
H	-2.21374300	-0.12664500	-0.48220900
H	-1.25384400	0.42364500	1.21781700

### <sup>4</sup>TS3/4

1 4

Cl	1.93198200	0.32804200	-0.00009400
Re	-0.21093100	-0.17617600	0.00006300

C	-1.60286700	1.10045800	-0.00002600
H	-1.42240200	-1.44982900	-0.00134600
H	-2.01599400	1.52362900	0.92123200
H	-2.01469900	1.52438100	-0.92153500
H	-1.95356000	-0.56445700	-0.00128700

### **<sup>2</sup>TS3/4**

1 2			
Cl	1.91910000	0.24793500	-0.00040300
Re	-0.23652100	-0.13848500	-0.01445200
C	-1.44668100	1.09420200	0.01485300
H	-1.46285500	-1.32904900	-0.31485400
H	-2.16546000	1.90947100	0.05076400
H	-1.96091500	-0.40358600	-0.23441400
H	-0.61633900	-0.57055000	1.50016200

### **<sup>2</sup>TS5/6**

1 2			
Cl	2.05927700	-0.26367600	-0.11057600
Re	-0.11437600	0.05661300	0.04842100
C	-0.93457900	1.55050900	-0.21636700
H	-1.39849700	2.51737100	-0.38655700
H	-0.39234800	0.21466100	1.63679700
C	-1.83778500	-1.23704200	-0.16412000
H	-1.56423000	-1.94909000	0.62265000
H	-1.70068900	-1.66546100	-1.16287400
H	-2.90221200	-1.02749500	-0.03065800
H	-1.83731200	0.26574800	-0.14820100

### **<sup>2</sup>TS6/7**

1 2			
Cl	1.99933900	-0.12568400	0.21449700
Re	-0.17797800	0.00147200	-0.14884900
C	-0.82726300	1.58868800	0.50376800
H	-0.38459100	2.53906200	0.18332500
H	-1.58535900	1.66336200	1.28738000
C	-1.27705500	-1.30355200	0.63045400
H	-1.64435700	-0.54494500	-0.97297300
H	-1.21596900	-2.36839000	0.38108800
H	-1.17731900	0.08724600	-1.57463700
H	-2.00689500	-1.06089100	1.40768000

### **<sup>2</sup>TS5/8**

1 2			
Cl	2.00589100	-0.35337000	-0.19435900
Re	-0.13602800	0.07809200	0.13521400

C	-0.62766100	1.55062900	-0.55733700
H	-0.85043700	2.49720500	-1.04288200
H	-1.25386300	0.33926100	1.43135700
C	-1.74595600	-1.21933600	-0.45511100
H	-1.43161900	-2.20364000	-0.09369200
H	-1.56690600	-1.09409700	-1.53202600
H	-2.82270600	-1.10917400	-0.30753600
H	-1.73083500	-0.26689000	0.78247900

### <sup>3</sup>TS10/11

1 3			
C	-0.36776100	1.77646300	-0.72265600
H	-1.16228400	1.85535500	-1.46893200
H	-0.97629500	1.42524800	0.90345800
H	-0.78837600	0.72767900	1.59668700
H	0.20471500	2.69454400	-0.55078500
Re	0.01672500	0.16350700	0.16558000
Cl	1.98368400	-0.70687600	-0.25703800
Cl	-1.76754200	-1.03575100	-0.24666900

### <sup>1</sup>TS10/11

1 1			
H	-0.57119500	0.48892100	1.51199700
Re	0.01145200	0.19506700	0.03870600
Cl	1.65816300	-1.20836900	-0.06571700
Cl	-1.94811800	-0.75933400	-0.10489200
C	0.46156600	1.97740200	-0.29278300
H	0.78861700	1.36563500	1.03020200
H	-0.34399100	2.71507900	-0.18769500
H	1.42753500	2.38688300	-0.60039500

### <sup>3</sup>TS12/13

1 3			
C	0.36810000	1.01685000	-1.62334900
H	-0.07748000	1.98123700	-1.88615700
H	1.04997200	0.59722500	-2.36583000
Re	-0.05073200	0.10086500	-0.04366900
Cl	1.59526100	-1.35648800	0.07075500
Cl	-2.14559300	-0.52166600	0.22130500
C	0.94933400	1.52510200	1.28330900
H	0.24009700	2.33644100	1.45283500
H	1.95974000	1.93302200	1.18191300
H	0.99222100	0.84092200	2.13612700
H	1.09138800	1.42316200	-0.16852500

### <sup>1</sup>TS12/13

1 1

C	-0.07250300	1.57780400	-1.21312000
H	-1.03322300	2.07860900	-1.38591000
H	0.78998200	1.97376900	-1.75392000
Re	-0.05129100	0.15110000	-0.07634700
Cl	1.57179400	-1.32010600	-0.30316400
Cl	-1.99448500	-0.74692000	0.32544400
C	1.22664900	0.99488500	1.53766000
H	0.97649200	1.84871200	2.17343800
H	2.07390800	1.28095500	0.91131600
H	1.45988900	0.12971200	2.15451700
H	-0.15933100	1.05904400	1.30058600

**<sup>1</sup>TS13/14**

1 1

C	-0.10225300	0.91329600	-1.75440900
H	0.30710600	1.93010800	-1.66339100
H	-0.55881500	0.70254500	-2.72238500
Re	0.01873600	-0.04558400	-0.12820900
Cl	-1.87639500	0.52216200	0.82475700
Cl	1.95211500	0.38309800	0.75137300
C	-0.21438500	-2.06524600	-0.06210200
H	0.56695600	-2.63820500	-0.56329700
H	-1.15429200	-2.22652100	-0.61525200
H	-0.34518400	-2.36600100	0.98216100
H	0.39160300	-0.46087900	-1.69732800

**<sup>1</sup>TS14/15**

1 1

C	0.22958600	1.50904500	-1.43455000
H	0.47601600	0.99667700	-2.36984400
H	0.97980200	2.26518900	-1.17982600
Re	-0.00460300	0.15260100	0.04236000
Cl	1.90923900	-0.90751100	0.13002800
Cl	-1.63384700	-1.27581500	-0.28993800
C	-0.79025300	1.22451700	1.39472100
H	-1.47837500	1.96482900	0.95439700
H	0.59577100	0.81544000	1.43069600
H	-0.79682600	1.22679800	2.48481700
H	-0.74880700	2.00112900	-1.53977300

**<sup>6</sup>TS16/17**

1 6

Re	0.10808600	-0.09445900	0.00006200
Cl	-2.00846700	-0.99961700	-0.00015700
Cl	-1.57130200	1.63093300	0.00001500

Cl	2.28627600	0.89937600	-0.00014100
C	1.24632300	-1.91518400	0.00001100
H	2.34937500	-2.01150800	0.00022600
H	0.92789000	-2.45589400	-0.89687100
H	0.92773800	-2.45574300	0.89694100
H	2.20001200	-0.52306400	-0.00023000

#### **<sup>4</sup>TS16/17**

14			
Re	0.09795300	-0.08303900	-0.08715400
Cl	2.25758300	-0.39894800	-0.34938400
Cl	-0.22006800	2.02131200	0.39823300
Cl	-2.18088900	-0.46404700	-0.61042300
C	-0.38626400	-1.31717000	1.59796800
H	-1.37420400	-1.63677600	1.98476000
H	0.04718900	-0.71505300	2.40081300
H	0.17865300	-2.23215100	1.39979400
H	-1.44320000	-0.97649300	0.71013500

#### **<sup>2</sup>TS16/17**

12			
Re	0.00066900	-0.07356400	0.02249700
Cl	1.89059600	-1.09540900	-0.21634700
Cl	-0.01842900	1.96222300	-0.81931100
Cl	-1.87148400	-1.12825100	-0.21421600
C	-0.00667000	1.06357100	1.90776300
H	-0.91182400	1.65983100	1.79039900
H	0.88862800	1.67410100	1.78793900
H	-0.00162200	0.60948400	2.90443900
H	0.00309500	-0.36315500	1.63118900

#### **<sup>6</sup>TS2/18**

16			
C	2.31732600	0.07802600	0.00000400
H	2.64343700	0.58377200	-0.91181400
H	-0.69782000	1.38978400	-0.00000500
H	2.55593500	-0.99868400	0.00007100
H	2.64342400	0.58388300	0.91176600
Re	0.20370300	-0.02837100	-0.00000100
Cl	-2.13686300	0.00593600	0.00000200

#### **<sup>6</sup>TS18/19**

16			
C	2.14940500	0.47310000	0.00001300
H	2.72341700	0.67524400	0.90827500
H	-0.48058400	-1.68814100	0.00006100

H	-0.24476100	1.52932000	-0.00006600
H	2.72344400	0.67517100	-0.90824800
Re	0.22552400	-0.10872200	0.00000800
Cl	-2.03130900	0.24258700	-0.00004000

#### **<sup>6</sup>TS19/20**

1 6			
C	2.21056900	-0.56599200	-0.01286000
H	2.80567800	-0.62320000	-0.92569500
H	-0.10335900	1.69633100	0.08152100
H	-2.55847700	-0.39548800	1.16462300
H	2.80013800	-0.69721700	0.89577700
Re	0.29657200	0.09431300	0.00349800
Cl	-2.26178200	-0.21517300	-0.08243700

#### **<sup>6</sup>TS20/21**

1 6			
C	2.16705000	-0.55757100	-0.00009400
H	2.75275100	-0.71334300	-0.90858700
H	-0.47189600	1.67156100	-0.00096400
H	-1.53587100	1.01449500	0.00008600
H	2.75314800	-0.71264700	0.90826200
Re	0.26165300	0.10030200	0.00002900
Cl	-2.12496600	-0.31983900	-0.00002400

#### **<sup>4</sup>TS2/22**

1 4			
Cl	-1.88478300	0.38833500	-0.03271200
Re	0.21768700	-0.19280500	0.02800600
C	1.62073000	1.15494800	-0.01516500
H	0.80514700	-1.72311900	-0.19473600
H	1.88434200	1.74773200	-0.89144000
H	2.14686000	1.41996800	0.90887800
H	1.15402000	-0.51556800	-1.27604300

#### **<sup>4</sup>TS22/23**

1 4			
Cl	-1.91863200	0.30574200	-0.02266500
Re	0.22921900	-0.13746500	-0.00097800
C	1.62149300	1.10509700	0.01566800
H	1.89395900	-0.39349200	-0.54784600
H	2.53156800	1.60772900	0.34634800
H	1.32671200	-1.36945300	-0.49362800
H	-0.05586100	-1.36309200	1.05977700

#### **<sup>4</sup>TS3/24**

1 4

Cl	-2.13514900	0.23515500	-0.00017700
Re	0.25591900	-0.15780800	0.00004500
C	1.93590000	0.76662800	-0.00019300
H	-1.42021100	1.36631400	0.00201400
H	2.39239800	1.16189200	-0.90968400
H	2.39285300	1.16101400	0.90945800
H	2.12316800	-0.45100700	-0.00096600

#### **<sup>4</sup>TS24/25**

1 4

Cl	-2.18560600	0.18502200	0.07794000
Re	0.28236700	-0.13767300	-0.01789400
C	1.92030500	0.76345200	0.08815400
H	-2.34564900	1.19287200	-0.72142800
H	2.40110500	1.14418800	-0.81794200
H	2.47632200	0.97753900	1.00288200
H	1.92417300	-0.71521600	0.02464900

#### **<sup>2</sup>TS6/26**

1 2

Cl	2.15106400	0.27960100	0.17058500
Re	-0.20211600	-0.02573900	-0.05567700
C	-0.72777700	-1.60665200	0.26729500
H	-0.96115600	-2.64539800	0.48492100
H	0.36870100	-0.52863500	-1.57805100
C	-1.89683700	1.05323800	0.16050200
H	-1.78172100	1.98244900	-0.41496900
H	-1.73610300	1.29424600	1.24065000
H	-2.89807000	0.64644200	0.05825800
H	1.34665500	-0.25143500	-1.08179700

#### **<sup>2</sup>TS6/27**

1 2

Cl	-2.06735900	-0.26911600	-0.18345100
Re	0.26370200	-0.02177000	0.17419200
C	1.40364900	-1.10519300	-0.74957100
H	1.40850400	-2.16966100	-0.47758800
H	1.32346000	0.08815700	1.44129900
C	0.40061800	1.71834100	-0.47039000
H	0.06635400	2.58845400	0.10508300
H	-1.14297800	0.90431800	-0.66597600
H	0.83911000	1.93421700	-1.44759400
H	2.04736200	-0.81664800	-1.58121800

#### **<sup>2</sup>TS27/28**

1 2

Cl	2.23269900	0.09764200	0.16125400
Re	-0.28716200	0.00566500	-0.17486400
C	-0.95901600	-1.50662000	0.59568700
H	-0.86295600	-2.47607700	0.09217000
H	-1.43861600	-0.02198800	-1.34455100
C	-1.10570900	1.39384500	0.67879900
H	-1.06015700	2.39606900	0.23421500
H	2.46955100	-1.10531800	0.57433600
H	-1.64674800	1.31514100	1.62237500
H	-1.49144700	-1.51595300	1.54803000

### <sup>2</sup>TS28/29

1 2

Cl	2.21432900	-0.00008600	-0.26403500
Re	-0.28445200	0.00002500	0.17418800
C	-1.04070500	1.45803300	-0.61851900
H	-0.91737400	2.45604100	-0.18195400
H	-1.43770300	0.00014200	1.34310100
C	-1.04024400	-1.45832000	-0.61819200
H	-1.63165600	-1.40790100	-1.53364000
H	-0.91720000	-2.45609000	-0.18070500
H	-1.63268600	1.40682100	-1.53354000
H	2.71259300	0.00232000	0.93153700

### <sup>2</sup>TS26/29

1 2

Cl	2.24579300	0.10059600	0.09597300
Re	-0.22798500	-0.00875200	0.00725200
C	-1.03842300	-1.53198200	0.01359900
H	-1.62353700	-2.44329500	-0.04207300
H	-0.09304600	0.00132200	-1.62999100
C	-1.66772400	1.21500700	-0.03255400
H	-2.73773900	1.24576600	-0.20139700
H	-1.23034400	2.17415900	0.29764100
H	-1.78572000	-0.24527200	0.65926400
H	2.62763100	0.11545200	-1.14513600

### <sup>2</sup>TS29/30

1 2

Cl	2.33214500	-0.14453000	-0.04744400
Re	-0.26090100	-0.04320000	-0.02896600
C	-0.83035200	1.68271900	0.01291900
H	-0.61770800	2.32874800	-0.85137000
H	-0.11121000	-0.28632700	1.56918500
C	-1.83701800	-1.00844100	0.02151100

H	-1.63724100	-2.08319900	-0.13170500
H	-2.87788500	-0.78823100	0.24218500
H	-1.40952000	2.13519400	0.81936900
H	2.57888200	0.34519400	1.12473300

#### **<sup>2</sup>TS30/31**

1 2			
Cl	2.18931700	-0.00006600	0.01721700
Re	-0.21775100	-0.00000700	-0.05739700
C	-1.31798300	1.46737700	0.06789600
H	-0.91670200	2.37773100	-0.40537900
H	0.08128400	-0.00004900	1.59076100
C	-1.31826600	-1.46718400	0.06793900
H	-0.91803100	-2.37756100	-0.40613800
H	-2.24851700	-1.61025100	0.61761700
H	-2.24869000	1.61077300	0.61668600
H	1.18105800	-0.00015700	1.18354400

#### **<sup>4</sup>TS32/33**

1 4			
Cl	-2.10254700	-0.19411500	-0.00009200
Re	0.10121700	0.02087300	0.00000700
C	1.06311000	1.63678800	0.00011800
H	1.32720900	2.16580900	-0.92016400
H	1.32718500	2.16569800	0.92047000
C	1.84477500	-1.29633300	0.00000300
H	1.63450300	-1.87527500	0.90521200
H	2.91833800	-1.09070400	-0.00016400
H	1.63427400	-1.87541400	-0.90506700
H	1.86318000	0.20166400	0.00004600

#### **<sup>4</sup>TS33/34**

1 4			
Cl	-2.03904800	0.05888500	0.01063600
Re	0.15296100	-0.02677900	-0.00048600
C	1.28373700	1.66663200	0.03034900
H	1.24016300	2.21138800	-0.91762300
H	0.84580400	2.27024300	0.84032200
C	1.16371100	-1.69758700	0.08028400
H	2.32910300	1.44030900	0.26356000
H	1.16636400	-2.45626400	-0.70142700
H	1.74907400	-1.94632800	0.97151300
H	1.17657400	-0.32621500	-1.26450700

#### **<sup>4</sup>TS34/35**

1 4

Cl	-1.98156500	0.26658600	-0.21419500
Re	0.19226500	0.00462800	0.11744900
C	0.72833900	-1.69100000	-0.50722500
H	0.08248200	-2.22025900	-1.21449600
H	1.62079500	-2.23980300	-0.19362300
C	1.56887200	1.38023700	-0.39816700
H	0.02210900	-1.47816300	1.07431800
H	1.87040900	2.16355600	0.30169500
H	2.06642000	1.44580400	-1.36532700
H	-0.17876000	-0.68565600	1.66243200

#### **<sup>4</sup>TS32/36**

1 4			
Cl	2.05881500	-0.58028200	0.00003700
Re	-0.30200200	-0.07483200	-0.00002400
C	-2.12942700	-0.41295600	0.00009900
H	-3.12270800	0.02991500	0.00023400
H	-2.16408600	-1.52851900	0.00001800
C	0.46502600	1.93255800	0.00002800
H	-0.00256100	2.32319200	-0.90883900
H	1.49199900	2.34240500	-0.00003400
H	-0.00249400	2.32318000	0.90892900
H	1.43655800	0.86941100	0.00007200

#### **<sup>4</sup>TS36/37**

1 4			
Cl	-2.20459320	0.44949599	-0.11037597
Re	0.30940280	0.01019799	0.09333203
C	-0.07512920	-1.97521801	-0.30875597
H	-0.66559720	-2.13045301	-1.21753797
H	-0.60359120	-2.40302201	0.55501403
C	1.58632280	1.02239399	-0.67118097
H	0.89719180	-2.46900401	-0.41786997
H	1.71254980	0.26022099	0.97426003
H	2.48829080	1.27585599	-1.22476097
H	-2.72442520	-0.67657401	-0.48491497

#### **<sup>4</sup>TS37/38**

1 4			
Cl	2.25084700	-0.35033900	0.01513300
Re	-0.33943000	-0.06518700	0.02168800
C	-0.05027800	1.96763800	-0.04519500
H	-0.28211700	2.17781600	-1.10589000
H	0.93822400	2.35700000	0.19906900
C	-1.93302300	-0.89466000	-0.20499400
H	-0.82281000	2.41223400	0.59251300

H	-0.28874600	-0.69374100	1.52564700
H	-2.95229700	-1.24110300	-0.36622600
H	2.50040200	-0.60526300	-1.22781900

**<sup>4</sup>TS38/39**

1 4			
Cl	2.13031100	-0.25053700	-0.19411400
Re	-0.29873600	-0.07639400	0.08423900
C	-0.17893200	2.00760500	-0.08900400
H	-1.13825900	1.93185700	-0.65331800
H	0.63032700	2.41866900	-0.68723400
C	-1.79582700	-0.83864000	-0.47712200
H	-0.30108000	2.48236900	0.88558200
H	0.22279000	-1.35016500	1.15200400
H	-2.58841600	-1.37985300	-0.99076900
H	1.21308000	-1.12800400	0.67247400

**<sup>5</sup>TS9/40**

1 5			
C	-0.66174300	1.92913400	0.00000000
H	-0.20754500	2.33679800	0.90805800
H	-0.20760900	2.33678700	-0.90809600
H	-1.68906300	0.82180600	0.00001200
H	-1.70670200	2.29304300	0.00003600
Re	0.08936400	-0.07174300	-0.00000100
Cl	-2.22011700	-0.60528500	0.00000100
Cl	2.28359100	-0.21721500	0.00000100

**<sup>5</sup>TS40/41**

1 5			
C	0.04612800	1.96439500	-0.00239500
H	1.08013600	2.29883600	-0.11070500
H	-0.56049300	2.34180200	-0.83124700
H	2.76963400	-0.19666000	1.18002300
H	-0.37600000	2.32828700	0.94006200
Re	-0.08247400	-0.07865300	-0.00702900
Cl	2.43178800	-0.36814100	-0.05948100
Cl	-2.25558200	-0.37654300	0.02203600

**<sup>5</sup>TS41/42**

1 5			
C	0.07764200	1.98156400	-0.10301200
H	-0.77854000	2.58435400	0.18306300
H	1.05432800	2.45620900	-0.14127100
H	2.72174700	-0.57322200	1.11355400
Re	-0.08561500	-0.06644800	0.01403700

Cl	2.41461900	-0.34140700	-0.12411800
Cl	-2.24855100	-0.35535300	-0.05982400
H	0.13457500	0.47179800	1.53696500

### **<sup>5</sup>TS42/43**

1 5			
C	0.28857000	1.91998400	0.04674500
H	-0.09114600	2.56278900	0.84240400
H	1.02663400	2.37033700	-0.61788000
H	1.39110400	-1.17054900	0.71177600
Re	-0.07801200	-0.04066300	0.01625000
Cl	2.33533700	-0.36196200	-0.11019300
Cl	-2.24793500	-0.27865100	-0.09859600
H	0.30704400	-1.34235700	1.11386500

### **<sup>3</sup>TS44/45**

1 3			
C	0.27025100	1.83870300	-0.15979100
H	-0.51835700	2.59347300	-0.14029400
H	0.27139000	1.34511700	1.12102500
H	2.64988800	-0.79571600	1.06498800
H	1.26803300	2.22677200	-0.37262500
Re	-0.09744200	0.00320700	0.05219200
Cl	2.32383600	-0.51165300	-0.15966900
Cl	-2.20526500	-0.46730900	-0.11261000

### **<sup>3</sup>TS45/46**

1 3			
C	0.06086500	1.82324000	-0.18024500
H	-0.77568300	2.48553700	0.05706600
H	0.53255800	0.19172300	1.64866900
H	1.51900900	-0.09167200	1.10004000
H	0.95202700	2.30710500	-0.58917800
Re	-0.05141500	-0.00123500	0.07063600
Cl	2.25089300	-0.46714100	-0.20735900
Cl	-2.17659900	-0.45871200	-0.17104500

### **<sup>1</sup>TS11/47**

1 1			
C	-0.09215900	1.85862000	-0.49186000
H	-1.09599900	2.30347200	-0.45877100
H	1.74634100	1.24770800	1.89660700
H	1.02135800	1.10946200	2.06862200
H	0.71773000	2.48749600	-0.86625000
Re	-0.01526500	0.13324300	0.05806600
Cl	1.87982300	-0.86261200	-0.20552500

Cl -1.92050600 -0.80168700 -0.03235900

**<sup>3</sup>TS12/48**

1 3  
C 0.30434200 1.71024900 -0.99343700  
H -0.22722600 2.62488200 -0.73552300  
H 0.52091900 1.60655800 -2.05878000  
Re -0.02773300 0.03141800 0.04727500  
Cl 1.89687200 -1.04535000 -0.17877500  
Cl -2.10078900 -0.60604600 -0.31408600  
C -0.00091300 0.77038300 1.73219600  
H -0.73664000 0.46133700 2.48108000  
H 1.65967000 1.97493500 -0.49431800  
H 0.75069400 1.48080900 2.08484800  
H 1.75854200 2.68507400 -0.87682800

**<sup>1</sup>TS15/49**

1 1  
C 0.10079700 1.38560000 1.45568900  
H -0.65034300 1.16861700 2.22054100  
H -0.59543600 1.82429200 0.09782800  
Re 0.10234800 0.20536000 -0.04105100  
Cl -2.03749900 -0.42949100 -0.24003000  
Cl 0.90724500 -1.84570200 0.26286500  
C 1.38200100 1.00206300 -1.09026800  
H 2.31247500 0.70016000 -0.56620000  
H -0.67775700 1.43739100 -0.93730200  
H 1.49729500 1.59450200 -1.99079700  
H 0.75519000 2.22532800 1.67404900

**<sup>1</sup>TS13/50**

1 1  
C 0.06047100 1.52660500 -1.24389400  
H -0.24687300 2.14639600 -0.31511900  
H 0.09116100 2.04216800 -2.19696400  
Re 0.15795800 0.19435900 -0.07002200  
Cl 0.73413400 -1.92111800 -0.22216100  
Cl -2.19286500 -0.24443700 0.33333500  
C 1.49045700 0.90721200 1.27817600  
H 1.45980400 1.95532500 1.56336900  
H 2.40972000 0.73206900 0.67598900  
H 1.50626900 0.26418000 2.16136500  
H -1.57406400 0.49443800 1.26734300

**<sup>1</sup>TS50/51**

1 1

C	-0.06260500	1.83219100	-0.74803400
H	-1.06192500	2.28216000	-0.76468400
H	0.75358200	2.42675900	-1.16419400
Re	0.07694700	0.16399100	-0.03628500
Cl	2.00912100	-0.81329200	-0.00459000
Cl	-1.76086600	-0.95924300	-0.53436800
C	-0.66512900	0.34728500	1.76279900
H	-1.89941100	-0.22980400	1.87001300
H	-0.96031900	1.29535000	2.20823100
H	-0.19026000	-0.34346100	2.46840300
H	-2.26664600	-0.67410800	1.17733400

#### **<sup>6</sup>TS52/53**

1 6

Re	0.24860900	-0.05304000	-0.00001900
Cl	-1.86143700	-1.15735500	-0.00000900
Cl	-1.47036200	1.47020300	0.00004900
F	1.79775100	-1.35610700	0.00010000
C	2.06968200	1.14201700	-0.00002200
H	2.00873800	1.75587100	0.90162000
H	2.00881200	1.75574600	-0.90175500
H	3.12758500	0.79080300	0.00004700
H	2.25190600	-0.28997900	0.00006700

#### **<sup>4</sup>TS52/53**

1 4

Re	-0.01121400	-0.05990700	-0.10556900
Cl	1.50737200	1.41025700	0.40134000
Cl	-2.03887900	0.74319800	-0.06641600
F	1.44629900	-1.28479900	-0.72851100
C	-0.37633400	-1.97035100	0.89072100
H	-0.37206700	-1.66364400	1.93818200
H	0.20553000	-2.92253300	0.88246600
H	-1.36436800	-2.24357500	0.51704500
H	0.64888600	-1.90068600	0.09853500

#### **<sup>2</sup>TS52/53**

1 2

Re	-0.02027300	0.14383600	-0.09462200
Cl	-1.99050200	-0.64796700	0.39555300
Cl	1.36056600	-1.53535000	-0.22197500
F	-0.07948900	1.64831700	-1.09054700
C	1.34211700	1.13736600	1.34434700
H	1.96116600	0.37841600	1.81357700
H	1.02518300	1.88137000	2.08338400
H	1.86637300	1.67438400	0.54641200

H	0.03934300	0.73549100	1.45126000
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**<sup>6</sup>TS54/55**

1 6

Re	-0.36515000	-0.05352600	0.00003400
Cl	0.82254800	2.01906700	0.00030600
Cl	-2.71455900	0.42706200	-0.00035100
Br	2.04912100	-0.47419400	-0.00019700
C	-1.07045700	-2.08623200	0.00052400
H	-2.32847900	-0.93659400	-0.00020400
H	-2.12661500	-2.41972800	0.00050900
H	-0.64530700	-2.55001200	-0.89511000
H	-0.64568600	-2.54920800	0.89676400

**<sup>4</sup>TS54/55**

1 4

Re	0.18724700	-0.04737000	-0.04562000
Cl	2.36659100	-0.69209200	-0.72598800
Cl	0.75853100	2.02561500	0.33997700
Br	-2.13967900	-0.16610600	-0.18274200
C	0.66875300	-1.29045800	1.63487900
H	0.35393500	-0.63024700	2.44735200
H	1.65065400	-1.68783400	1.95967500
H	0.01053900	-2.15384700	1.51118800
H	1.69050200	-1.08873600	0.65216400

**<sup>2</sup>TS54/55**

1 2

Re	-0.32337400	-0.00008200	0.08243400
Cl	-0.67898700	2.02216600	-0.58850000
Cl	-0.67435300	-2.02316200	-0.58862500
Br	2.03447400	0.00129900	0.13211900
C	-2.42608500	-0.00221400	0.83901000
H	-1.04572700	-0.00116200	1.56523200
H	-2.73282700	-0.89469300	1.38278100
H	-2.73512400	0.88956500	1.38261400
H	-2.87659500	-0.00277700	-0.16026600

**MECP1**

Cl	1.96854232476555	0.00049894808393	0.00045372836349
Re	-0.19571489175736	-0.00032551772021	-0.00018499735677
C	-2.29217305816218	0.00047774296825	0.00074483546155
H	-1.71109873546328	1.07354590136882	-0.00000462509079
H	-2.91196057663834	0.00076126236746	0.90130929905938
H	-2.91371389811522	0.00095130426531	-0.89861487486178
H	-1.71226916462916	-1.07301564133357	-0.00022436557508

**MECP2**

Cl	-1.90594009738659	0.01080618904801	-0.13213454275195
Re	0.21928160374776	0.03997680278177	0.10713251644261
C	1.67635765679387	1.15915198653018	-0.28002058586407
H	0.86884766930523	-1.45615564706665	0.04600018745790
H	2.12236991766513	2.05234912103496	-0.70489492532026
H	2.10892338616069	0.81687224106884	0.71212619141210
H	1.70911586371391	0.24223130660289	-1.04722784137633

**MECP3**

Re	0.18393478279192	0.03604668422349	-0.07616560340119
Cl	0.75760697856150	-2.01769360579709	-0.45138550280784
Cl	-1.89006199698657	0.42152166826786	0.41846558011423
C	2.04645414949012	1.35692400229574	-0.05602290510490
H	1.53934109668534	2.29249987353103	-0.30934670025285
H	1.60094758525629	0.74883869316529	0.86375553317651
H	2.37618091070324	0.80121797071004	-0.94532772248174
H	2.94627531349815	1.59235439360362	0.52419032075778

**MECP4**

C	-0.27642087958554	1.83032272317001	-0.86732892437538
H	-1.18801161359458	2.09592965131223	-1.40986785218922
H	-1.32493755356004	0.59842887789577	1.48603428047510
H	-0.80707026445462	1.24997353233315	1.53291063134094
H	0.56870018579274	2.52397813647735	-0.97131466723254
Re	-0.07834218368139	0.29191854350107	0.12785351601558
Cl	1.88717910189271	-0.56588848204368	-0.26636313214267
Cl	-1.74165567280926	-1.05718702264590	-0.36336605189181

**MECP5**

C	-0.12456940016376	1.88008057141923	-0.50450710120633
H	-1.07653218830973	2.42645123468666	-0.47706054589346
H	1.27081274496378	0.57039031239533	1.61447689722268
H	0.65647422207481	1.11222328859409	1.76999669274254
H	0.69265975222464	2.36145260497539	-1.04950871980034
Re	0.00407840736277	0.21999251183882	0.28612824466165
Cl	1.83110220918158	-0.74944014104390	-0.45534982007078
Cl	-1.81489974733409	-0.90910338286563	-0.12815764765595

**MECP6**

C	0.23718081209684	0.87387385247443	-1.79561004990709
H	-0.62866586853194	1.31156522226943	-2.30975320359143
H	1.18855034037853	0.90552613094565	-2.33522106360713
Re	0.02938526635165	0.18239237519472	-0.11110019933446
Cl	1.64706940800490	-1.24875498403233	0.33085101651383
Cl	-1.94837727486226	-0.64393268466639	0.22473814434001

C	1.13340566385335	1.60982533800719	1.45522205152367
H	0.99735081187438	2.63721742344934	1.80855912893955
H	2.18526196603874	1.34392987537193	1.40875218126868
H	0.56903169359419	0.97126480070923	2.14237301889077
H	0.68341118120163	1.86280965027678	0.41336697496359

### MECP7

Re	0.09590461507516	-0.27448802581189	0.05519518340479
Cl	1.73775769280192	-0.48038699500734	-1.33813726524439
Cl	0.35669510120570	1.61189002373020	1.12751226233242
Cl	-1.71349529595841	-1.10674218700277	-0.83070887046005
C	-0.44317434359248	-1.27253022456279	2.17599450467503
H	-0.74608580927270	-2.30823417507742	2.33905433716635
H	0.12125521550311	-0.85943209790818	3.00781956702917
H	0.37891471226456	-1.51028933273220	1.35970313642173
H	-1.34924888802687	-0.68652098562758	1.99522414467493

### Reference

- Becke, A. D., A new mixing of Hartree–Fock and local density - functional theories. *J. Chem. Phys.* **1993**, *98* (2), 1372-1377.
- Stephens, P. J.; Devlin, F. J.; Chabalowski, C. F.; Frisch, M. J., Ab Initio Calculation of vibrational absorption and circular dichroism spectra using density functional force fields. *J. Phys. Chem.* **1994**, *98* (45), 11623-11627.
- Jr., T. H. D., Gaussian basis sets for use in correlated molecular calculations. I. The atoms boron through neon and hydrogen. *J. Chem. Phys.* **1989**, *90* (2), 1007-1023.
- Woon, D. E.; Jr., T. H. D., Gaussian basis sets for use in correlated molecular calculations. III. The atoms aluminum through argon. *J. Chem. Phys.* **1993**, *98* (2), 1358-1371.
- Wilson, A. K.; Woon, D. E.; Peterson, K. A.; Jr., T. H. D., Gaussian basis sets for use in correlated molecular calculations. IX. The atoms gallium through krypton. *J. Chem. Phys.* **1999**, *110* (16), 7667-7676.
- Figgen, D.; Peterson, K. A.; Dolg, M.; Stoll, H., Energy-consistent pseudopotentials and correlation consistent basis sets for the 5d elements Hf–Pt. *J. Chem. Phys.* **2009**, *130* (16), 164108.
- Sinha, P.; Boesch, S. E.; Gu, C.; Wheeler, R. A.; Wilson, A. K., Harmonic vibrational frequencies: Scaling factors for HF, B3LYP, and MP2 methods in combination with correlation consistent basis sets. *J. Phys. Chem. A* **2004**, *108* (42), 9213-9217.
- III, G. D. P.; Bartlett, R. J., A full coupled - cluster singles and doubles model: The inclusion of disconnected triples. *J. Chem. Phys.* **1982**, *76* (4), 1910-1918.
- Pople, J. A.; Head - Gordon, M.; Raghavachari, K., Quadratic configuration interaction. A general technique for determining electron correlation energies. *J. Chem. Phys.* **1987**, *87* (10), 5968-5975.

10. Halkier, A.; Helgaker, T.; Jørgensen, P.; Klopper, W.; Koch, H.; Olsen, J.; Wilson, A. K., Basis-set convergence in correlated calculations on Ne, N<sub>2</sub>, and H<sub>2</sub>O. *Chem. Phys. Lett.* **1998**, 286 (3), 243-252.
11. Halkier, A.; Helgaker, T.; Jørgensen, P.; Klopper, W.; Olsen, J., Basis-set convergence of the energy in molecular Hartree–Fock calculations. *Chem. Phys. Lett.* **1999**, 302 (5), 437-446.
12. Adamo, C.; Barone, V., Toward reliable adiabatic connection models free from adjustable parameters. *Chem. Phys. Lett.* **1997**, 274 (1), 242-250.
13. Karton, A.; Tarnopolsky, A.; Lamère, J.-F.; Schatz, G. C.; Martin, J. M. L., Highly accurate first-principles benchmark data sets for the parametrization and validation of density functional and other approximate methods. Derivation of a robust, generally applicable, double-hybrid functional for thermochemistry and thermochemical kinetics. *J. Phys. Chem. A* **2008**, 112 (50), 12868-12886.
14. Perdew, J. P., Density-functional approximation for the correlation energy of the inhomogeneous electron gas. *Phys. Rev. B* **1986**, 33 (12), 8822-8824.
15. Becke, A. D., Density-functional exchange-energy approximation with correct asymptotic behavior. *Phys. Rev. A* **1988**, 38 (6), 3098-3100.
16. Becke, A. D., Density - functional thermochemistry. III. The role of exact exchange. *J. Chem. Phys.* **1993**, 98 (7), 5648-5652.
17. Perdew, J. P.; Wang, Y., Accurate and simple analytic representation of the electron-gas correlation energy. *Phys. Rev. B* **1992**, 45 (23), 13244-13249.
18. Schmider, H. L.; Becke, A. D., Optimized density functionals from the extended G2 test set. *J. Chem. Phys.* **1998**, 108 (23), 9624-9631.
19. Gill, P. M. W., A new gradient-corrected exchange functional. *Mol. Phys.* **1996**, 89 (2), 433-445.
20. Yanai, T.; Tew, D. P.; Handy, N. C., A new hybrid exchange–correlation functional using the Coulomb-attenuating method (CAM-B3LYP). *Chem. Phys. Lett.* **2004**, 393 (1), 51-57.
21. Hamprecht, F. A.; Cohen, A. J.; Tozer, D. J.; Handy, N. C., Development and assessment of new exchange-correlation functionals. *J. Chem. Phys.* **1998**, 109 (15), 6264-6271.
22. Henderson, T. M.; Izmaylov, A. F.; Scuseria, G. E.; Savin, A., The importance of middle-range Hartree-Fock-type exchange for hybrid density functionals. *J. Chem. Phys.* **2007**, 127 (22), 221103.
23. Henderson, T. M.; Izmaylov, A. F.; Scuseria, G. E.; Savin, A., Assessment of a Middle-Range Hybrid Functional. *J. Chem. Theory Comput.* **2008**, 4 (8), 1254-1262.
24. Heyd, J.; Scuseria, G. E.; Ernzerhof, M., Hybrid functionals based on a screened Coulomb potential. *J. Chem. Phys.* **2003**, 118 (18), 8207-8215.
25. Heyd, J.; Scuseria, G. E., Efficient hybrid density functional calculations in solids: Assessment of the Heyd–Scuseria–Ernzerhof screened Coulomb hybrid functional. *J. Chem. Phys.* **2004**, 121 (3), 1187-1192.
26. Krukau, A. V.; Vydrov, O. A.; Izmaylov, A. F.; Scuseria, G. E., Influence of the exchange screening parameter on the performance of screened hybrid functionals. *J. Chem. Phys.* **2006**, 125 (22), 224106.

27. Iikura, H.; Tsuneda, T.; Yanai, T.; Hirao, K., A long-range correction scheme for generalized-gradient-approximation exchange functionals. *J. Chem. Phys.* **2001**, *115* (8), 3540-3544.
28. Vydrov, O. A.; Scuseria, G. E., Assessment of a long-range corrected hybrid functional. *J. Chem. Phys.* **2006**, *125* (23), 234109.
29. Zhao, Y.; Truhlar, D. G., The M06 suite of density functionals for main group thermochemistry, thermochemical kinetics, noncovalent interactions, excited states, and transition elements: Two new functionals and systematic testing of four M06-class functionals and 12 other functionals. *Theor. Chem. Acc.* **2008**, *120* (1), 215-241.
30. Zhao, Y.; Truhlar, D. G., Exploring the limit of accuracy of the global hybrid meta density functional for main-group thermochemistry, kinetics, and noncovalent interactions. *J. Chem. Theory Comput.* **2008**, *4* (11), 1849-1868.
31. Peverati, R.; Truhlar, D. G., Improving the accuracy of hybrid meta-gga density functionals by range separation. *J. Phys. Chem. Lett.* **2011**, *2* (21), 2810-2817.
32. Adamo, C.; Barone, V., Exchange functionals with improved long-range behavior and adiabatic connection methods without adjustable parameters: The mPW and mPW1PW models. *J. Chem. Phys.* **1998**, *108* (2), 664-675.
33. Lee, C.; Yang, W.; Parr, R. G., Development of the Colle-Salvetti correlation-energy formula into a functional of the electron density. *Phys. Rev. B* **1988**, *37* (2), 785-789.
34. Perdew, J. P.; Burke, K.; Ernzerhof, M., Generalized gradient approximation made simple. *Phys. Rev. Lett.* **1996**, *77* (18), 3865-3868.
35. Schwabe, T.; Grimme, S., Towards chemical accuracy for the thermodynamics of large molecules: New hybrid density functionals including non-local correlation effects. *Phys. Chem. Chem. Phys.* **2006**, *8* (38), 4398-4401.
36. Heyd, J.; Scuseria, G. E., Assessment and validation of a screened Coulomb hybrid density functional. *J. Chem. Phys.* **2004**, *120* (16), 7274-7280.
37. Heyd, J.; Peralta, J. E.; Scuseria, G. E.; Martin, R. L., Energy band gaps and lattice parameters evaluated with the Heyd-Scuseria-Ernzerhof screened hybrid functional. *J. Chem. Phys.* **2005**, *123* (17), 174101.
38. Heyd, J.; Scuseria, G. E.; Ernzerhof, M., Erratum: "Hybrid functionals based on a screened Coulomb potential" [J. Chem. Phys. 118, 8207 (2003)]. *J. Chem. Phys.* **2006**, *124* (21), 219906.
39. Izmaylov, A. F.; Scuseria, G. E.; Frisch, M. J., Efficient evaluation of short-range Hartree-Fock exchange in large molecules and periodic systems. *J. Chem. Phys.* **2006**, *125* (10), 104103.
40. Henderson, T. M.; Izmaylov, A. F.; Scalmani, G.; Scuseria, G. E., Can short-range hybrids describe long-range-dependent properties? *J. Chem. Phys.* **2009**, *131* (4), 044108.
41. Handy, N. C.; Cohen, A. J., Left-right correlation energy. *Mol. Phys.* **2001**, *99* (5), 403-412.
42. Adamo, C.; Barone, V., Toward reliable density functional methods without adjustable parameters: The PBE0 model. *J. Chem. Phys.* **1999**, *110* (13), 6158-6170.
43. Ernzerhof, M.; Perdew, J. P., Generalized gradient approximation to the angle-

- and system-averaged exchange hole. *J. Chem. Phys.* **1998**, *109* (9), 3313-3320.
44. Perdew, J. P.; Kurth, S.; Zupan, A.; Blaha, P., Accurate density functional with correct formal properties: A step beyond the generalized gradient approximation. *Phys. Rev. Lett.* **1999**, *82* (12), 2544-2547.
45. Perdew, J. P.; Chevary, J. A.; Vosko, S. H.; Jackson, K. A.; Pederson, M. R.; Singh, D. J.; Fiolhais, C., Atoms, molecules, solids, and surfaces: Applications of the generalized gradient approximation for exchange and correlation. *Phys. Rev. B* **1992**, *46* (11), 6671-6687.
46. Perdew, J. P.; Ruzsinszky, A.; Csonka, G. I.; Constantin, L. A.; Sun, J., Workhorse semilocal density functional for condensed matter physics and quantum chemistry. *Phys. Rev. Lett.* **2009**, *103* (2), 026403.
47. Peverati, R.; Truhlar, D. G., Communication: A global hybrid generalized gradient approximation to the exchange-correlation functional that satisfies the second-order density-gradient constraint and has broad applicability in chemistry. *J. Chem. Phys.* **2011**, *135* (19), 191102.
48. Staroverov, V. N.; Scuseria, G. E.; Tao, J.; Perdew, J. P., Comparative assessment of a new nonempirical density functional: Molecules and hydrogen-bonded complexes. *J. Chem. Phys.* **2003**, *119* (23), 12129-12137.
49. Voorhis, T. V.; Scuseria, G. E., A novel form for the exchange-correlation energy functional. *J. Chem. Phys.* **1998**, *109* (2), 400-410.
50. Xu, X.; Goddard, W. A., The X3LYP extended density functional for accurate descriptions of nonbond interactions, spin states, and thermochemical properties. *Proc. Natl. Acad. Sci. U. S. A.* **2004**, *101* (9), 2673-2677.
51. Chai, J.-D.; Head-Gordon, M., Systematic optimization of long-range corrected hybrid density functionals. *J. Chem. Phys.* **2008**, *128* (8), 084106.
52. Chai, J.-D.; Head-Gordon, M., Long-range corrected hybrid density functionals with damped atom–atom dispersion corrections. *Phys. Chem. Chem. Phys.* **2008**, *10* (44), 6615-6620.
53. Yu, H. S.; He, X.; Li, S. L.; Truhlar, D. G., MN15: A Kohn–Sham global-hybrid exchange–correlation density functional with broad accuracy for multi-reference and single-reference systems and noncovalent interactions. *Chem. Sci.* **2016**, Medium: X; Size: p. 5032-5051.
54. Yu, H. S.; He, X.; Truhlar, D. G., MN15-L: A new local exchange-correlation functional for Kohn–Sham density functional theory with broad accuracy for atoms, molecules, and solids. *J. Chem. Theory Comput.* **2016**, *12* (3), 1280-1293.
55. Cohen, A. J.; Handy, N. C., Dynamic correlation. *Mol. Phys.* **2001**, *99* (7), 607-615.
56. Brémond, E.; Adamo, C., Seeking for parameter-free double-hybrid functionals: The PBE0-DH model. *J. Chem. Phys.* **2011**, *135* (2), 024106.
57. Brémond, É.; Sancho-García, J. C.; Pérez-Jiménez, Á. J.; Adamo, C., Communication: Double-hybrid functionals from adiabatic-connection: The QIDH model. *J. Chem. Phys.* **2014**, *141* (3), 031101.
58. Andrae, D.; Häußermann, U.; Dolg, M.; Stoll, H.; Preuß, H., Energy-adjusted ab initio pseudopotentials for the second and third row transition elements.

- Theor. Chim. Acta* **1990**, *77* (2), 123-141.
59. Grimme, S.; Ehrlich, S.; Goerigk, L., Effect of the damping function in dispersion corrected density functional theory. *J. Comput. Chem.* **2011**, *32* (7), 1456-1465.
60. Grimme, S.; Antony, J.; Ehrlich, S.; Krieg, H., A consistent and accurate ab initio parametrization of density functional dispersion correction (DFT-D) for the 94 elements H-Pu. *J. Chem. Phys.* **2010**, *132* (15), 154104.
61. Reed, A. E.; Curtiss, L. A.; Weinhold, F., Intermolecular interactions from a natural bond orbital, donor-acceptor viewpoint. *Chem. Rev.* **1988**, *88* (6), 899-926.
62. Carpenter, J. E.; Weinhold, F., Analysis of the geometry of the hydroxymethyl radical by the “different hybrids for different spins” natural bond orbital procedure. *J. Mol. Struct.: THEOCHEM* **1988**, *169*, 41-62.
63. Reed, A. E.; Weinstock, R. B.; Weinhold, F., Natural population analysis. *J. Chem. Phys.* **1985**, *83* (2), 735-746.
64. Reed, A. E.; Weinhold, F., Natural localized molecular orbitals. *J. Chem. Phys.* **1985**, *83* (4), 1736-1740.
65. Reed, A. E.; Weinhold, F., Natural bond orbital analysis of near - Hartree-Fock water dimer. *J. Chem. Phys.* **1983**, *78* (6), 4066-4073.
66. Foster, J. P.; Weinhold, F., Natural hybrid orbitals. *J. Am. Chem. Soc.* **1980**, *102* (24), 7211-7218.
67. Neese, F., Importance of direct spin–spin coupling and spin-flip excitations for the zero-field splittings of transition metal complexes: A case study. *J. Am. Chem. Soc.* **2006**, *128* (31), 10213-10222.
68. Ye, S.; Geng, C.-Y.; Shaik, S.; Neese, F., Electronic structure analysis of multistate reactivity in transition metal catalyzed reactions: the case of C–H bond activation by non-heme iron(IV)-oxo cores. *Phys. Chem. Chem. Phys.* **2013**, *15* (21), 8017-8030.
69. Lu, T.; Chen, F., Multiwfn: A multifunctional wavefunction analyzer. *J. Comput. Chem.* **2012**, *33* (5), 580-592.
70. Humphrey, W.; Dalke, A.; Schulten, K., VMD: Visual molecular dynamics. *J. Mol. Graphics* **1996**, *14* (1), 33-38.
71. Harvey, J. N.; Aschi, M.; Schwarz, H.; Koch, W., The singlet and triplet states of phenyl cation. A hybrid approach for locating minimum energy crossing points between non-interacting potential energy surfaces. *Theor. Chem. Acc.* **1998**, *99* (2), 95-99.
72. Weigend, F.; Ahlrichs, R., Balanced basis sets of split valence, triple zeta valence and quadruple zeta valence quality for H to Rn: Design and assessment of accuracy. *Phys. Chem. Chem. Phys.* **2005**, *7* (18), 3297-3305.
73. Su, P.; Jiang, Z.; Chen, Z.; Wu, W., Energy decomposition scheme based on the generalized Kohn–Sham scheme. *J. Phys. Chem. A* **2014**, *118* (13), 2531-2542.
74. Zou, J., Molecular Orbital Kit (MOKIT), <https://gitlab.com/jxzou/mokit> (accessed 5 29, 2021)