

SUPPORTING INFORMATION

Vibrational energy relaxation of a diatomic molecule in a superfluid helium nanodroplet. Influence of the nanodroplet size, interaction energy and energy gap

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Table s1. Energy (E) contributions (in K) at $t=0$ for $I_2(v=1)@(^4He)_N$, as a function of N .^a

N	$E(He)_N$	E per atom $(He)_N$	Kinetic $E(He)_N$	Lennard-Jones $E(He)_N$
50	99.73	1.99	185.39	-141.26
100	-102.45	-1.02	214.45	-403.94
150	-338.14	-2.25	230.49	-674.32
200	-588.94	-2.94	241.10	-952.21

Interaction E				
N	Correlation $E(He)_N$	$I_2-(He)_N$	Vibrational $E(I_2)$	Total $E I_2@(^4He)_N$
50	55.30	-664.09	459.96	-104.40
100	86.45	-739.99	459.96	-382.48
150	104.98	-767.58	459.96	-645.76
200	121.41	-782.70	459.96	-911.68

^a $E(He)_N$ refers exclusively to the helium and includes several terms: kinetic energy, potential energy (Lennard-Jones potential energy function) and correlation energy.

Table s2. Inverse of the global relaxation times, lifetimes and transition times of the $\nu=1$ excited state and $\langle V_{01}^2 \rangle$ coupling terms for the five interaction potential energies considered ($V=xV_{I_2-He}$) and $N=100$.^a

x	$\langle V_{01}^2 \rangle$ (K ²)	1/Global relax.		1/Transition	
		time (ps ⁻¹)	1/Lifetime (ps ⁻¹)	time (ps ⁻¹)	1/Transition (ps ⁻¹)
0.50	0.15 (0.09)	$7.66 \cdot 10^{-5}$ (0.22)	$1.03 \cdot 10^{-4}$ (0.21)	$1.49 \cdot 10^{-4}$ (0.26)	
0.75	0.61 (0.39)	$1.96 \cdot 10^{-4}$ (0.56)	$2.73 \cdot 10^{-4}$ (0.54)	$3.50 \cdot 10^{-4}$ (0.60)	
1.00	1.56 (1.00)	$3.48 \cdot 10^{-4}$ (1.00)	$5.01 \cdot 10^{-4}$ (1.00)	$5.79 \cdot 10^{-4}$ (1.00)	
1.25	3.09 (1.98)	$8.50 \cdot 10^{-4}$ (2.44)	$1.27 \cdot 10^{-3}$ (2.53)	$1.34 \cdot 10^{-3}$ (2.31)	
1.50	5.98 (3.84)	$1.56 \cdot 10^{-3}$ (4.49)	$2.42 \cdot 10^{-3}$ (4.83)	$2.31 \cdot 10^{-3}$ (4.00)	

^a $\langle V_{01}^2 \rangle$ and the inverse of the relaxation times relative to the $x=1.00$ reference case are given between parentheses.

Table s3. Inverse of the global relaxation times, lifetimes and transition times of the $\nu=1$ excited state, $\langle V_{01}^2 \rangle$ coupling terms and \sqrt{x} for the four vibrational frequencies ν_e considered ($\nu_e = x\nu_{e,I_2}$) and $N=100$.^a

x	\sqrt{x}	$\langle V_{01}^2 \rangle$ (K ²)	1/Global relax.		1/Transition	
			time (ps ⁻¹)	1/Lifetime (ps ⁻¹)	time (ps ⁻¹)	1/Transition (ps ⁻¹)
0.75	0.87	1.16 (0.75)	$9.74 \cdot 10^{-4}$ (2.80)	$1.41 \cdot 10^{-3}$ (2.82)	$1.59 \cdot 10^{-3}$ (2.75)	
1.00	1.00	1.56 (1.00)	$3.48 \cdot 10^{-4}$ (1.00)	$5.01 \cdot 10^{-4}$ (1.00)	$5.79 \cdot 10^{-4}$ (1.00)	
1.25	1.12	1.93 (1.24)	$1.87 \cdot 10^{-4}$ (0.54)	$2.64 \cdot 10^{-4}$ (0.53)	$3.22 \cdot 10^{-4}$ (0.56)	
1.50	1.22	2.30 (1.48)	$1.28 \cdot 10^{-4}$ (0.37)	$1.65 \cdot 10^{-4}$ (0.33)	$2.23 \cdot 10^{-4}$ (0.38)	

^a $\langle V_{01}^2 \rangle$ and the inverse of the relaxation times relative to the $x=1.00$ reference case are given between parentheses.

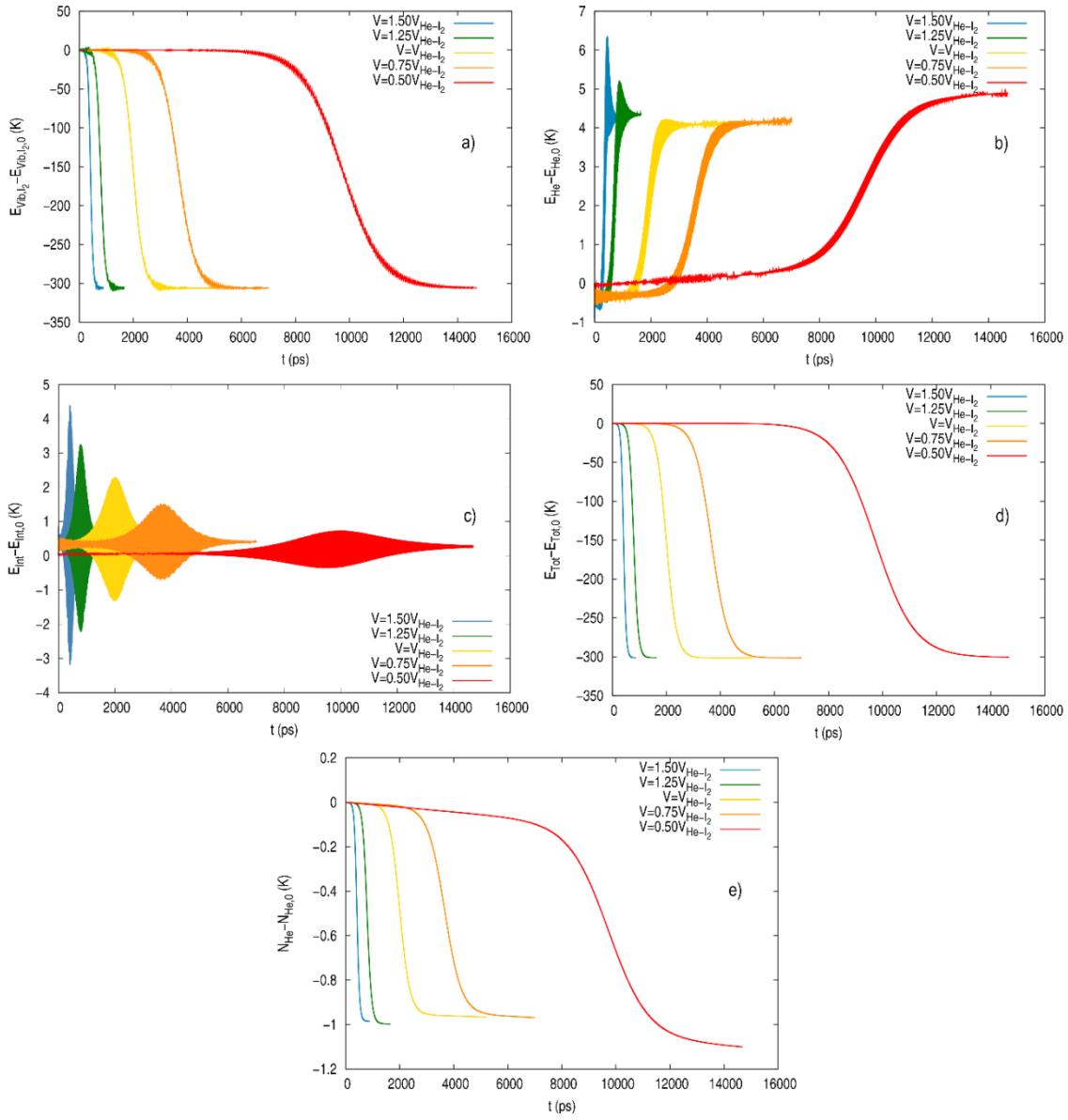


Figure s1. Energy increments involved in the VER from $\nu=1$ for the five selected “I₂-HeND” interaction potential energies and number of ⁴He atoms of the nanodroplet, as a function of time for $N=100$: a) vibrational energy of the molecule; b) helium energy of the nanodroplet; c) molecule-nanodroplet interaction energy; d) total energy of the system (I₂@HeND); e) number of ⁴He atoms of the HeND.

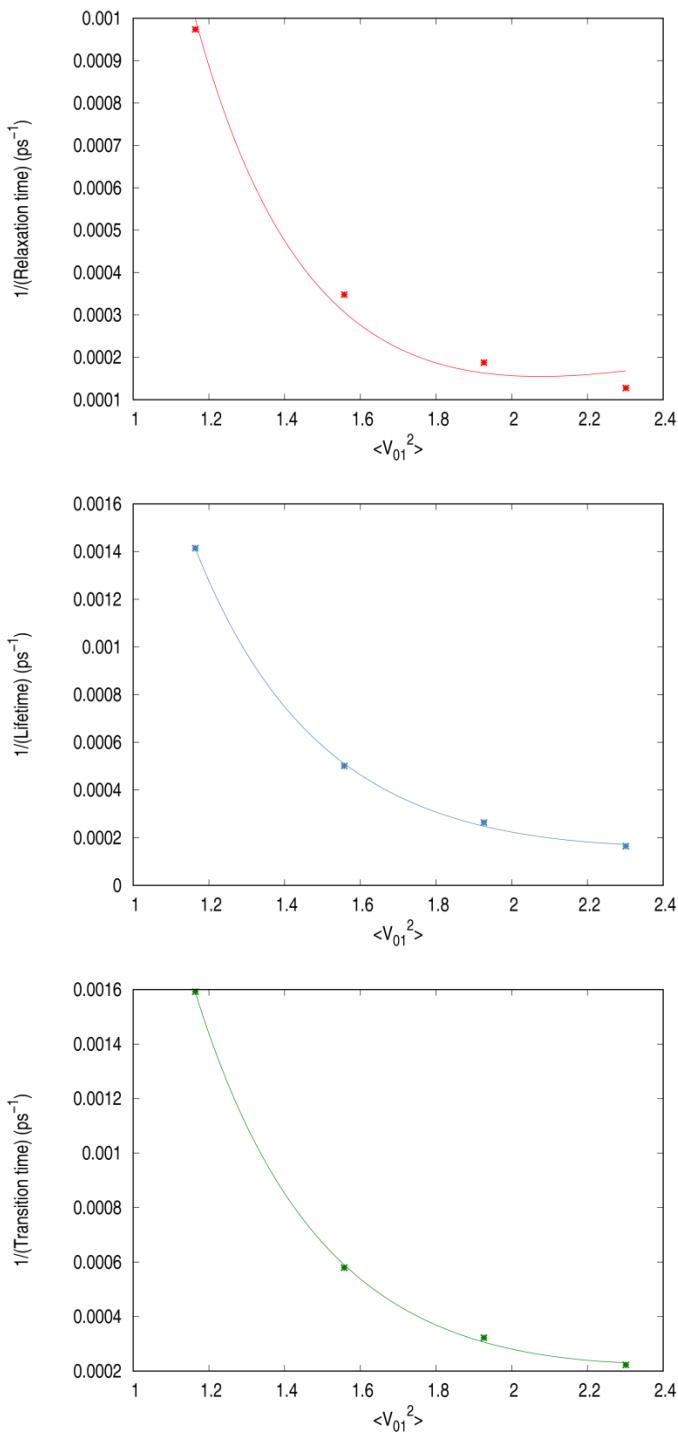


Figure s2. Inverse of the global relaxation time (top), lifetime (middle) and transition time (bottom) of the $\nu=1$ excited state, as a function of the $\langle V_{01}^2 \rangle$ coupling term for the vibrational frequencies selected ($\nu_e = x\nu_{e,I_2}$, with $x=0.75, 1.00, 1.25$ and 1.50) and $N=100$. The coupling term increases with x .

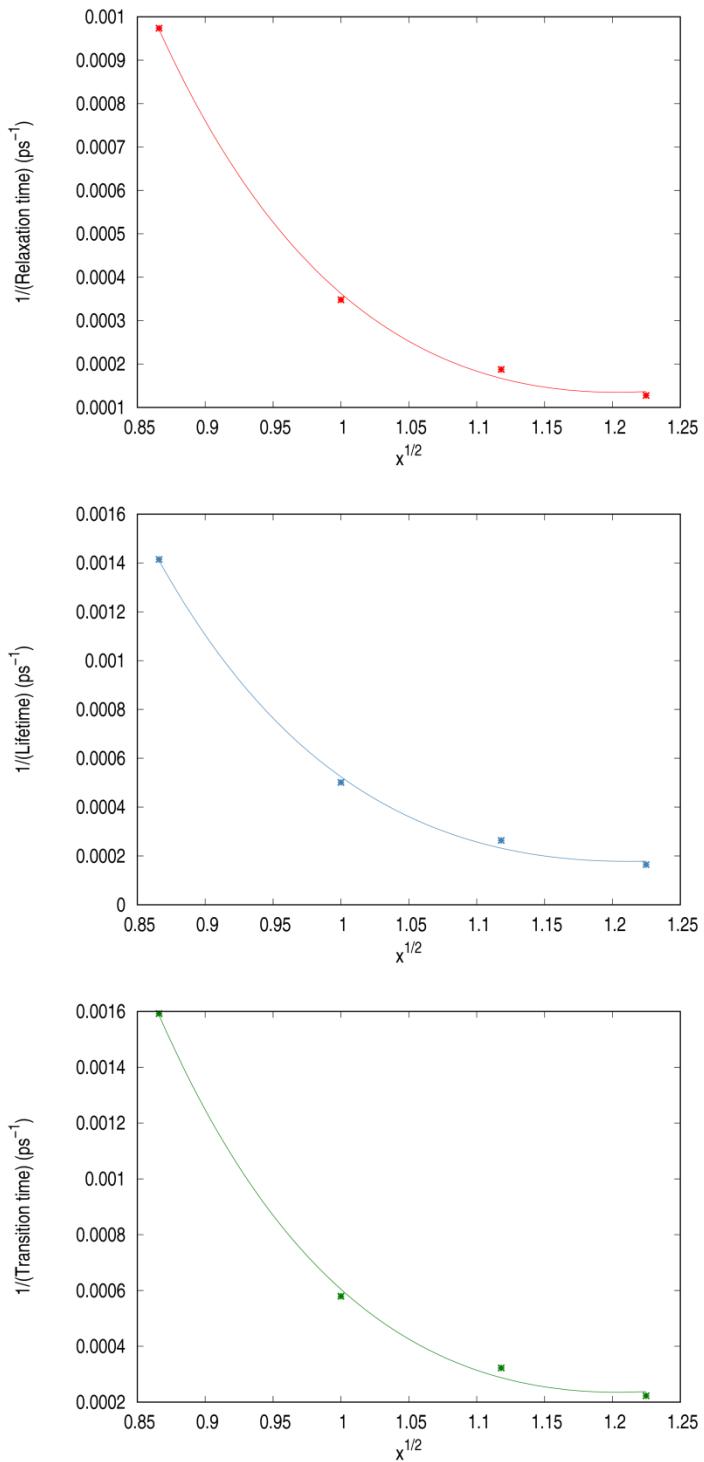


Figure s3. Inverse of the global relaxation time (top), lifetime (middle) and transition time (bottom) of the $\nu=1$ excited state, as a function of the $x^{1/2}$ term for the vibrational frequencies selected ($\nu_e = x\nu_{e,I_2}$, with $x=0.75, 1.00, 1.25$ and 1.50) and $N=100$.