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Supporting Information

Vacuum ultraviolet photochemistry of sulfuric acid vapor: A combined experimental and theoretical study

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Table S1. Relative isotopic distribution in the photoionization mass spectra of Figure 1 and their assignments.

| Group | Mass | Relative intensity (%) | | C |
|-------|------|------------------------|-------------------------|--|
| | | Experimental | Literature ^a | - Species |
| 1 | 64 | 100 | 100 | ³² SO ₂ ⁺ |
| | 65 | 2 | 0.88 | $^{33}\mathrm{SO}_2^+$ |
| | 66 | 6 | 4.9 | $^{34}\mathrm{SO_2}^+$ |
| 2 | 80 | 100 | 100 | $^{32}\mathrm{SO_3}^+$ |
| | 81 | 28 | 0.91 | $H^{32}SO_3^+$ |
| | 82 | 5 | 5.1 | $^{34}\mathrm{SO_3}^+$ |
| 3 | 98 | 100 | 100 | $\mathrm{H2^{32}SO_{4}}^{+}$ |
| | 99 | 65 | 0.98 | $H2^{32}SO_4H^+$ |
| | 100 | 8 | 5.3 | $H_2^{34}SO_4^+$ |

a. J. Anal. At. Spectrom., 1999, 14, 5N-24N; Molecular Weight Calculator version 3, www.wsearch.com.au.

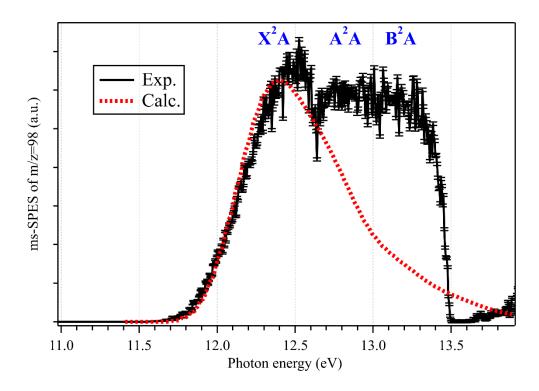


Figure S1. Slow photoelectron spectrum of H_2SO_4 (in black) and the calculated photoelectron spectrum corresponding to the X^2A ground electronic state of $H_2SO_4^+$ (in red).