

**Electronic Supplementary Material**

**For**

**Solid Phase Behavior of Mixture Systems Based on Tripalmitoyl Glycerol and  
Monounsaturated Triacylglycerols Forming a Molecular Compound**

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**Table S1.** Melting temperatures (°C) of stabilized PPP/PPO mixtures when heated at 2 °C·min<sup>-1</sup>.

<b>Polymorph</b>	PPP	95PPP	90PPP	85PPP	80PPP	75PPP	70PPP	65PPP
$\beta'$ 1-3L (PPO)	-	-	-	-	-	-	-	34.1 ± 0.5
$\beta$ -2L (ss)	67.9 ± 0.7	66.4 ± 1.0	66.5 ± 0.7	65.8 ± 1.1	66.3 ± 2.0	64.3 ± 0.4	64.1 ± 0.7	63.2 ± 0.4
<b>Polymorph</b>	60PPP	50PPP	40PPP	30PPP	20PPP	10PPP	5PPP	PPO
$\beta'$ 1-3L (PPO)	35.8 ± 0.6	36.0 ± 1.0	35.2 ± 0.6	35.5 ± 0.8	35.5 ± 0.3	35.6 ± 0.6	36.0 ± 0.3	36.1 ± 1.3
$\beta$ -2L (ss)	62.9 ± 0.5	61.9 ± 0.8	60.3 ± 0.9	58.3 ± 0.6	55.5 ± 0.4	51.8 ± 0.6	45.4 ± 0.4	-

ss: solid solution

**Table S2.** Peak top temperatures (°C) of the main polymorphic events of PPP/PPO mixtures when heating at 2 °C·min<sup>-1</sup> soon after a melt crystallization process at the same rate.

<b>PPP/PPO ratio</b>		$\alpha$ (PPP) + $\beta'$ (PPP) → $\beta$ (PPP)			$\beta$ (PPP) melting
90/10		44.5 ± 0.4			65.6 ± 0.6
		$\beta'_2$ (PPO) melting			$\beta'$ (PPP) → $\beta$ (PPP)
80/20		31.2 ± 0.8		41.6 ± 0.3	65.0 ± 0.4
70/30		30.9 ± 0.5		39.5 ± 1.2	63.9 ± 0.6
60/40		30.5 ± 1.1		39.0 ± 0.7	62.7 ± 0.9
		$\beta'_2$ (PPO) melting	$\beta'$ (PPP) → melt → $\beta$ (PPP)	$\beta$ (PPP) melting	
50/50		30.6 ± 0.8	42.4 ± 0.4	61.7 ± 0.6	
40/60		30.4 ± 0.6	40.5 ± 0.4	60.0 ± 0.4	
30/70		30.3 ± 0.6	39.3 ± 0.6	57.6 ± 0.8	
20/80		31.2 ± 0.7	39.4 ± 0.4	55.3 ± 0.4	
		$\alpha_1$ (PPO) melting	$\beta'_2$ (PPO) crystallization	$\beta'_2$ (PPO) melting	$\beta'$ (PPP) → melt → $\beta$ (PPP)
10/90		18.8 ± 0.3	19.7 ± 0.4	31.4 ± 0.3	41.5 ± 0.6
					49.6 ± 0.5

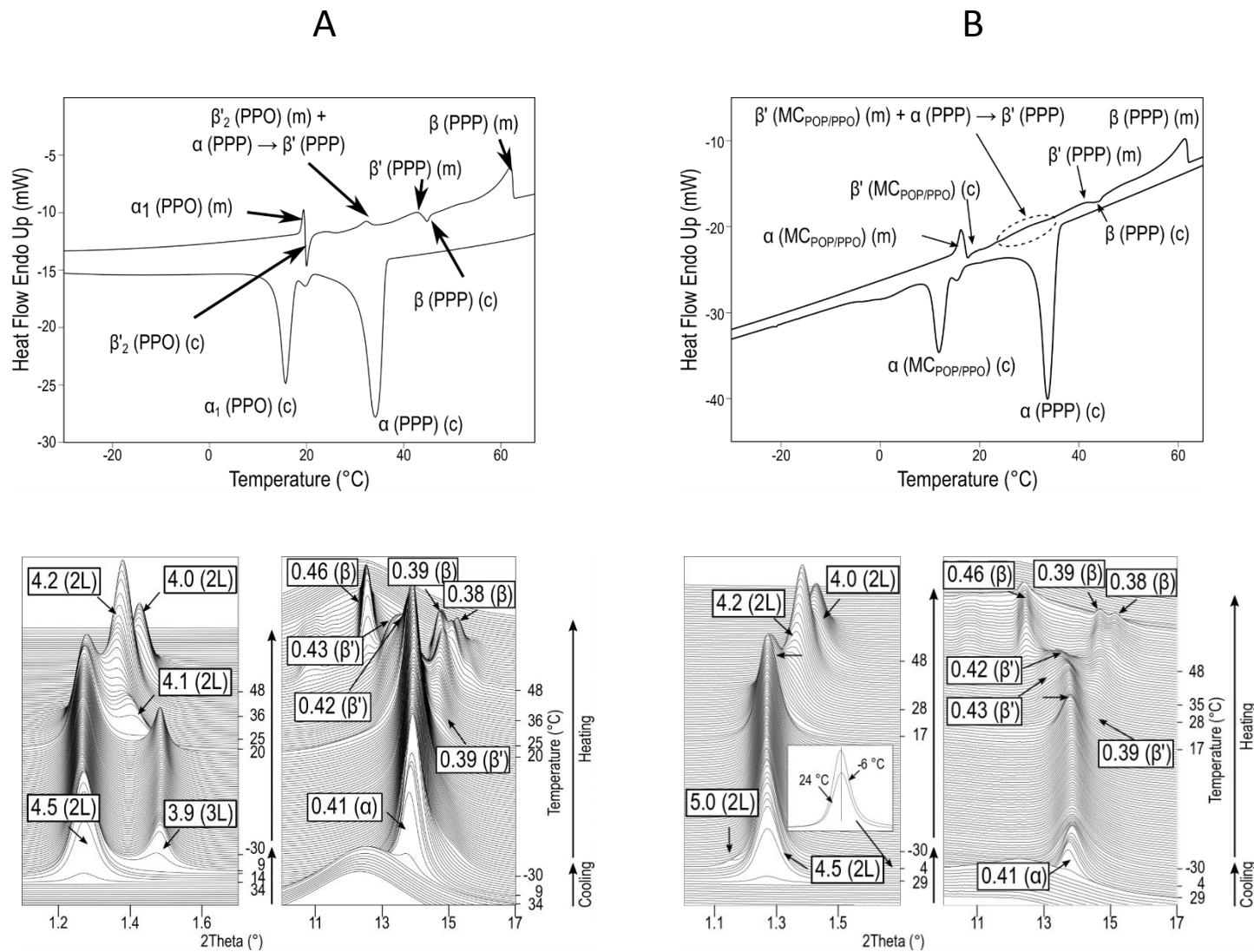
**Table S3.** Melting temperatures (°C) of stabilized PPP/MC<sub>POP/PPO</sub> mixtures when heated at 2 °C·min<sup>-1</sup>.

<b>Polymorph</b>	PPP	95PPP	90PPP	85PPP	80PPP	75PPP	70PPP	65PPP	60PPP	55PPP	50PPP
β-2L (MC <sub>POP/PPO</sub> )	-	-	-	-	-	-	-	-	32.8 ± 0.4	33.4 ± 1.2	33.2 ± 0.5
β-2L (ss)	67.5 ± 0.6	66.4 ± 0.6	66.0 ± 0.5	65.7 ± 0.5	65.6 ± 1.5	65.0 ± 0.6	64.7 ± 1.0	63.4 ± 0.9	62.7 ± 0.4	62.0 ± 0.5	61.4 ± 0.8
<b>Polymorph</b>	45PPP	40PPP	35PPP	30PPP	25PPP	20PPP	15PPP	10PPP	5PPP	MC <sub>POP/PPO</sub>	
β-2L (MC <sub>POP/PPO</sub> )	33.3 ± 0.2	33.7 ± 0.4	33.1 ± 0.3	33.6 ± 0.7	33.5 ± 0.6	33.6 ± 0.5	33.6 ± 0.8	33.2 ± 0.8	33.6 ± 0.6	33.6 ± 0.4	
β-2L (ss)	60.4 ± 0.6	59.7 ± 0.4	58.2 ± 0.3	58.2 ± 0.6	56.4 ± 0.6	54.5 ± 0.8	52.9 ± 0.7	49.6 ± 1.0	43.7 ± 0.6	-	

ss: solid solution

**Table S4.** Peak top temperatures (°C) of the main polymorphic events of PPP/MC<sub>POP/PPO</sub> mixtures when heating at 2 °C·min<sup>-1</sup> soon after a melt crystallization process at the same rate.

<b>PPP/MC<sub>POP/PPO</sub> ratio</b>				
	$\alpha$ (PPP) + $\beta'$ (PPP) → $\beta$ (PPP)		$\beta$ (PPP) melting	
90/10	44.1 ± 0.3		65.5 ± 0.5	
80/20	40.6 ± 0.3		64.5 ± 0.3	
	$\alpha$ (MC <sub>POP/PPO</sub> ) melting	$\beta'$ (MC <sub>POP/PPO</sub> ) melting	$\beta'$ (PPP) → $\beta$ (PPP)	$\beta$ (PPP) melting
70/30	14.6 ± 0.3	28.6 ± 1.1	39.4 ± 0.6	63.5 ± 0.6
60/40	14.7 ± 0.3	28.5 ± 0.4	38.7 ± 0.4	62.6 ± 0.4
	$\alpha$ (MC <sub>POP/PPO</sub> ) melting	$\beta'$ (MC <sub>POP/PPO</sub> ) melting	$\beta'$ (PPP) → melt → $\beta$ (PPP)	$\beta$ (PPP) melting
50/50	14.7 ± 0.5	28.0 ± 0.3	39.8 ± 0.2	61.1 ± 0.5
40/60	14.7 ± 0.3	28.0 ± 0.4	40.3 ± 0.4	59.7 ± 0.3
30/70	14.9 ± 0.3	27.2 ± 0.6	37.0 ± 0.9	57.3 ± 0.5
20/80	15.1 ± 0.4	27.2 ± 0.5	34.0 ± 0.5	54.6 ± 0.7
	$\alpha$ (MC <sub>POP/PPO</sub> ) → melt → $\beta'$ (MC <sub>POP/PPO</sub> )	$\beta'$ (MC <sub>POP/PPO</sub> ) melting	$\beta'$ (PPP) → melt → $\beta$ (PPP)	$\beta$ (PPP) melting
10/90	15.5 ± 0.5	28.0 ± 0.5	34.6 ± 1.2	48.9 ± 0.6



**Figure S1.** DSC thermograms (up) and SR-XRD patterns (down) obtained when equimolecular (A) PPP/PPO and (B) PPP/MC<sub>POP</sub>/PPO mixtures were cooled to -30 °C at 15 °C·min<sup>-1</sup> and subsequently heated at 2 °C·min<sup>-1</sup>. (c): crystallization; (m): melting. *d*-spacing values are given in nm.