

Figure S1: IRC curves for the  $S_N 2$  with zero to three explicit solvent molecules, at the M06-2X/6-31++G(d,p)/SMD(Water) level.



Figure S2: Extracted structure of butene with eight water molecules from the DICE3 simulation.



Figure S3: Molecular electrostatic potential for the reaction of 1-butene with HCl in the gas phase and using the implicit solvation model. The surfaces were prepared using an isovalue of 0.004 for the density, and  $E = \pm 0.09$  and  $E = \pm 0.13$ , for the gas phase and SMD respectively.

Table S1: TSs calculated with different amounts of explicitly added solvent molecules and the BSSE corrected energy variation between successive TSs. The energy variation corresponds to the next line minus the current line, plus the number of water molecules. The results were obtained at M06-2X/6-31++G(d,p)/SMD(Water) level of theory.

$\mathrm{TS}^{(a)}$	Energy (a.u.)	$\Delta E(TS^{\#}) (kJ.mol^{-1})$
$TS \cdot 0 H_2O$	-157.54006	2.76
$TS \cdot 1 H_2O$	-233.94883	-17.39
$TS \cdot 2 H_2O$	-310.36375	-10.00
$TS \cdot 3 H_2O$	-386.77665	-14.70
$TS \cdot 4 H_2O$	-463.19132	-16.28
$TS \cdot 5 H_2O$	-539.60659	-16.64
$TS \cdot 6 H_2O$	-616.02200	-18.49
$TS \cdot 7 H_2O$	-692.43811	-18.13
$TS \cdot 8 H_2O$	-768.85408	_

<sup>(a)</sup> TS =  $H_3CH_2C^{(+)}(OH_2)H-CH_3$  (Carbocation). <sup>(b)</sup>  $\Delta E(TS \cdot 1 H_2O) = E(TS \cdot 2 H_2O) - [E(TS \cdot 1 H_2O) + E(H_2O)]$ , where  $E(H_2O) = -76.40907$  a.u.



Figure S4: Molecular electrostatic potential for the  $S_N 2$  reaction TSs with (a) zero, (b) one, (c) two and (d) three explicit solvent molecules, obtained at M06-2X/6-31++G(d,p)/SMD(Water). The surfaces were prepared using an isovalue of 0.004 for the density, and  $E_{min} = \pm 0.101$ .



Figure S5: Representations of the different dihedrals the water molecule part of the hydronium ion adopts for TS  $\cdot$  nH<sub>2</sub>O with (a) n = 0 and (b) n = 7. An inversion of the angle of hH1 and hH2 in relation to the plane of the double bond can be clearly seen.

led explicit are always	WAT angle	1	175.12754	171 01065
h 0 to 7 add e first two	WAT dist 7	I	1.81346	1 06907
tion reaction witl 1 solvents and th	tH-hO-hH1-hH2	118.06048	111.21643	199 07100
of the hydra licitly adde	tH-hO-hH2	110.96811	102.40913	100 00105
s structures for the exp	tH-hO-hH1	107.19387	113.16253	110 01177
s of the TS nd angles	hH2-hO	0.96757	0.98024	0.0771c
ant atoms stances a addition.	hH1-hO	0.96757	0.96613	30340 0
the releva re the di order of	hO-tH	1.79656	1.88221	1 00010
nedrals for T angle" a rest are in	C1-tH-hO	174.43474	165.42531	161 01050
gles and dil " and "WA , while the	C2-C1-tH	81.25705	83.13558	00 11911
urces, ang VAT dist and hH2	C1-tH	1.19246	1.17422	1 1 7007
ond dista cules. "V <i>i</i> ith hH1	C1-C2	1.39194	1.39872	1 20015
Table S2: B solvent mole interacting w	Molecule	$TS \cdot 0 H_2O$	${ m TS} \cdot 1 \ { m H_2O}$	TC DI C
Table S2: Bon solvent molecu interacting with	Molecule C	$TS \cdot 0 H_2O 1.$	$TS \cdot 1 H_2O 1.$	

WAT angle		175.12754	174.04965	172.81391	177.7193	175.05152	176.73033	169.90601	167.10738	175.16637	174.98704	173.41491	174.41207	170.70510	177.47075	172.00419	168.73601	170.30637	178.11195	173.42334	173.33584	173.45521	162.90625	171.53061	177.07051	174.04576	171.95999	174.92089	172.46057
WAT dist	1	1.81346	1.85327	1.83376	1.82039	1.83191	1.82432	1.82396	1.81663	1.80705	1.81132	1.81855	1.85197	1.80684	1.84139	1.85060	1.80246	1.79976	1.81314	1.80338	1.77789	1.81821	1.80606	1.80636	1.81370	1.79863	1.76954	1.80759	1.78441
tH-hO-hH1-hH2	118.06048	111.21643	122.97498		126.01696			-100.84622				-116.12896					-99.25854						-96.45889						
tH-hO-hH2	110.96811	102.40913	108.83495		112.14527			99.07688				110.78176					96.7202						93.1793						
tH-hO-hH1	107.19387	113.16253	118.21117		116.74025			94.19132				101.92932					96.84559						99.71346						
hH2-hO	0.96757	0.98024	0.97716		0.97922			0.97982				0.9797					0.98152						0.9797						
hH1-hO	0.96757	0.96613	0.97686		0.9772			0.97955				0.97906					0.98155						0.98065						
hO-tH	1.79656	1.88221	1.88040		1.8548			1.84191				1.80134					1.70643						1.78037						
C1-tH-hO	174.43474	165.42531	164.21958		163.54692			171.14772				164.45459					173.32138						174.52714						
C2-C1-tH	81.25705	83.13558	83.11344		82.18465			83.19468				76.66185					84.7615						82.86834						
C1-tH	1.19246	1.17422	1.17207		1.17752			1.18317				1.2106					1.2002						1.19117						
C1-C2	1.39194	1.39872	1.39945		1.39742			1.39737				1.38718					1.39328						1.39502						
Molecule	$TS \cdot 0 H_2O$	${ m TS} \cdot 1 \; { m H_2^{-}O}$	${ m TS} \cdot 2 \; { m H_2^{-}O}$	I	$\mathrm{TS}\cdot 3~\mathrm{H_2O}$			$TS \cdot 4 H_2O$				${ m TS}\cdot 5~{ m H_2O}$	I				${ m TS} \cdot 6 \ { m H_2O}$	I					$\mathrm{TS}\cdot 7~\mathrm{H_2O}$	I					

solvent eracting	AT angle	ı	6.94908	6.30455	6.44558	8.73195	7.69602	6.99743
plicit 's int	W/		17	17	16	17	16	17
added ex are alway	WAT dist		1.76297	1.80205	1.81362	1.76754	1.80884	1.81011
with 0 to 3 e first two a	C2-hO-hH2	106.69161	106.08392	100.80429		101.04009		
v2 reaction ents and the	C2-hO-hH1	108.29851	108.55595	108.04889		108.43710		
for the $S_N$ dded solve	hH2-hO	0.96971	0.96824	0.98060		0.98054		
tructures plicitly a	hH1-hO	0.96959	0.98570	0.98187		0.98454		
is of the TS s es for the ex	C3-C2-C1-hO	98.83627	100.40323	100.26192		100.13881		
celevant atom nces and angl on.	C3-C2-C1-Cl	-100.94425	-101.44172	-101.86493		-101.89334		
drals for the 1 are the distar rder of additi	Cl-C2-H(C2)	73.23624	74.01315	74.75206		75.07836		
ngles and dihe WAT angle'' a rest are in o	hO-C2-H(C2)	83.95312	81.41325	80.22848		79.91080		
ances, ar st" and " while the	Cl-C2	2.66267	2.62761	2.59060		2.57972		
Bond dist "WAT dii nd hH2, <sup>7</sup>	hO-C2	2.13963	2.16295	2.17561		2.17692		
Table S3: molecules. with hH1 a	Molecule	$TS \cdot 0 H_2O$	$TS \cdot 1 H_2O$	$TS \cdot 2 H_2O$		$TS \cdot 3 H_2O$		

atom	q	$\epsilon$	$\sigma$
С	-0.146844	0.066	3.500
$\mathbf{C}$	0.008368	0.066	3.500
$\mathbf{C}$	0.394383	0.066	3.500
$\mathbf{C}$	-0.428759	0.066	3.500
Ο	-0.834879	0.1521	3.1506
Η	0.059112	0.030	2.500
Η	0.101366	0.030	2.500
Η	0.062394	0.030	2.500
Η	0.084233	0.030	2.500
Η	0.074522	0.030	2.500
Η	0.104255	0.030	2.500
Η	0.196490	0.030	2.500
Η	0.191265	0.030	2.500
Η	0.261292	0.030	2.500
Η	0.439823	0.0000	0.0000
Η	0.432980	0.0000	0.0000

Table S4: Simulation parameters obtained for the TS structure of the reaction via  $\rm H_3O^+.$ 

Table S5: Optimized geometry for the TS of the reaction via the HCl in the gas-phase.

С	1.827799	-1.163083	0.099957
С	1.464478	0.252234	-0.334069
$\mathbf{C}$	0.404431	0.912922	0.450122
$\mathbf{C}$	-0.441943	1.911190	-0.063403
Η	2.045769	-1.201162	1.171478
Η	2.711811	-1.508125	-0.440406
Η	0.988757	-1.830998	-0.106191
Η	2.328357	0.928891	-0.186533
Η	1.216383	0.309624	-1.399393
Η	0.298938	0.626917	1.496063
Η	-0.942172	2.570469	0.640374
Η	-0.202644	2.320940	-1.042086
Η	-1.213405	0.983097	-0.244453
$\operatorname{Cl}$	-1.574140	-0.863484	-0.047323

Table S6: Optimized geometry for the TS of the reaction via the HCl with implicit solvent (SMD).

С	1.574688	-1.553710	0.120937
$\mathbf{C}$	1.598296	-0.116886	-0.392518
$\mathbf{C}$	0.898043	0.855135	0.432565
$\mathbf{C}$	0.184745	1.958369	-0.068884
Η	1.905014	-1.600067	1.161991
Н	2.240383	-2.172879	-0.483101
Η	0.560983	-1.955857	0.053913
Η	2.631540	0.286592	-0.315119
Η	1.315998	-0.013541	-1.444259
Η	0.876724	0.677929	1.509988
Η	-0.041588	2.762698	0.626803
Η	0.357055	2.227918	-1.109847
Η	-0.750657	1.260617	-0.071862
$\operatorname{Cl}$	-2.037064	-0.490050	-0.028300

Table S7: Optimized geometry for the TS of the reaction via the  $H_3O^+$  without explicit solvent molecules.

С	1.758160	-1.116792	0.101265
С	1.360841	0.283225	-0.371636
С	0.307420	0.901039	0.435826
С	-0.807346	1.556482	-0.079146
0	-2.031819	-1.166462	-0.061681
Η	2.013981	-1.110354	1.163879
Η	2.626539	-1.461008	-0.463037
Η	0.939552	-1.822278	-0.057858
Η	2.208116	0.974904	-0.205912
Η	1.113660	0.326527	-1.435037
Η	0.359992	0.757255	1.516903
Η	-1.421285	2.146061	0.596348
Η	-0.817749	1.806132	-1.138210
Η	-1.258293	0.454399	-0.015753
Η	-2.706748	-1.208532	-0.753701
Η	-2.517668	-1.275133	0.767979

Table S8: Optimized geometry for the TS of the reaction via the  $\rm H_3O^+$  with one explicit solvent molecule added.

С	-0.154620	1.833990	0.723307
С	-0.850153	1.087146	-0.416869
С	-1.652336	-0.051412	0.020252
С	-1.667152	-1.298726	-0.612520
Ο	0.908910	-1.499275	0.975929
Η	-0.879926	2.169985	1.468543
Η	0.363166	2.708597	0.324052
Η	0.578779	1.189046	1.214185
Η	-1.625607	1.742657	-0.860557
Η	-0.176671	0.800725	-1.229352
Η	-2.202637	0.053303	0.957698
Η	-2.444328	-2.004733	-0.333325
Η	-1.253870	-1.362799	-1.618037
Η	-0.754326	-1.511070	0.094899
Η	1.458995	-0.948571	0.380105
Η	1.335053	-2.366141	0.994912
Ο	2.345610	0.130058	-0.777099
Η	3.301039	-0.004140	-0.716305
H	2.209737	1.060883	-0.552478

С	-2.135158	0.942719	0.985285
$\mathbf{C}$	-2.166995	-0.051250	-0.175503
$\mathbf{C}$	-1.592220	-1.358087	0.126028
$\mathbf{C}$	-0.794869	-2.090308	-0.760844
Ο	1.131807	-0.069745	0.403930
Ο	3.904615	-0.489710	0.322631
Ο	0.229031	2.293235	-0.858271
Η	-2.644194	0.533716	1.861810
Η	-2.638105	1.865761	0.690344
Η	-1.104149	1.179593	1.259251
Η	-3.218221	-0.330557	-0.391721
Η	-1.757057	0.347054	-1.107729
Η	-1.721436	-1.745949	1.138624
Η	-0.594292	-3.132286	-0.527710
Η	-0.798618	-1.791693	-1.808033
Η	0.070343	-1.463886	-0.278394
Η	0.883400	0.763308	-0.042350
Η	2.102584	-0.150135	0.330599
Η	4.391235	0.156687	0.850973
Η	-0.202188	2.846652	-0.192875
Η	0.933290	2.844341	-1.224839
Η	4.309237	-0.451297	-0.554063

Table S9: Optimized geometry for the TS of the reaction via the  $\rm H_3O^+$  with two explicit solvent molecules added.

С	-1.651727	-0.503075	1.518039
С	-1.537171	-1.087647	0.111225
$\mathbf{C}$	-0.541646	-2.147025	-0.030067
$\mathbf{C}$	0.277692	-2.311141	-1.150126
Ο	1.586130	-0.097485	0.400301
Ο	-0.140983	2.081647	0.078691
Ο	-2.826109	1.832404	-0.684234
Ο	4.234279	0.740726	-0.002774
Η	-1.832127	-1.294215	2.250527
Η	-2.483756	0.201890	1.558363
Η	-0.736027	0.025425	1.793299
Η	-2.467742	-1.636923	-0.137403
Η	-1.410338	-0.335273	-0.673429
Η	-0.368593	-2.796889	0.830117
Η	0.838220	-3.236427	-1.245919
Η	0.008796	-1.776711	-2.060010
Η	0.962606	-1.556799	-0.559861
Η	0.977229	0.657203	0.264088
Η	2.488443	0.241695	0.239952
Η	4.490313	1.380842	0.674672
Η	0.231008	2.716802	-0.546930
Η	-1.072513	1.960191	-0.197705
Η	4.355706	1.203512	-0.842578
Η	-3.173448	2.718598	-0.853155
Η	-2.917212	1.362067	-1.524321

Table S10: Optimized geometry for the TS of the reaction via the  $\rm H_3O^+$  with three explicit solvent molecules added.

Table S11: Optimized geometry for the TS of the reaction via the  $\rm H_3O^+$  with four explicit solvent molecules added.

С	-0.284552	-0.839501	2.054976
Η	0.665314	-1.220130	2.439123
Η	-1.069553	-1.061902	2.780121
Η	-0.209095	0.245158	1.947522
С	-0.632024	-1.490757	0.715073
Η	-0.673835	-2.589878	0.847623
Η	-1.604473	-1.182875	0.320349
С	0.408302	-1.363633	-0.302268
Η	1.448132	-1.437848	0.026053
С	0.167692	-1.036723	-1.639379
Η	0.975704	-1.163445	-2.355530
Η	-0.854763	-1.114695	-2.006758
Η	0.251602	0.072660	-1.236726
Ο	0.150551	1.738316	-0.456986
Η	-0.696329	1.632554	0.024329
Η	0.870998	1.561479	0.183320
Ο	2.375573	1.036657	1.055645
Η	2.843765	0.448488	0.427621
Ο	-2.394441	1.361937	0.632630
Η	-2.856949	2.209826	0.610006
Η	-2.779290	0.837042	-0.099943
Η	2.952460	1.802935	1.170518
Ο	-3.354021	-0.120121	-1.520839
Η	-3.159137	-1.062001	-1.420417
Η	-4.318152	-0.066909	-1.569647
Ο	3.578767	-0.728538	-0.736529
Η	3.170565	-0.641423	-1.608964
Η	4.512537	-0.520723	-0.876718

Table S12: Optimized geometry for the TS of the reaction via the  $H_3O^+$  with five explicit solvent molecules added.

С	0.311978	-1.086899	-0.949123
Η	-0.645146	-0.651353	-1.247148
Η	0.514795	-1.950137	-1.586544
Η	1.098231	-0.344241	-1.111126
С	0.278752	-1.541894	0.517227
Η	-0.547447	-2.256616	0.649936
Η	1.205816	-2.033611	0.821951
С	-0.051431	-0.442664	1.439409
Η	-0.940734	0.144693	1.214524
С	0.770932	0.013507	2.459166
Η	0.374573	0.727807	3.176559
Η	1.645450	-0.570840	2.735482
Η	1.066044	0.723147	1.523824
Ο	1.135231	2.046042	0.303172
Η	1.781857	1.834963	-0.401908
Η	0.248540	1.988609	-0.107982
Ο	-1.491641	1.834619	-0.722679
Η	-2.058242	1.081204	-0.456833
Ο	3.093341	1.345977	-1.562953
Η	3.331889	0.522895	-1.087615
Η	2.651728	1.055822	-2.371923
Η	-2.008052	2.628443	-0.532383
Ο	-3.089089	-0.379499	-0.015903
Η	-2.869522	-1.064638	-0.661692
Η	-2.914383	-0.783636	0.857324
Ο	3.649972	-0.866847	0.022392
Η	3.206926	-1.637924	-0.358580
Ο	-2.358452	-1.422840	2.502646
Η	-2.465509	-0.754707	3.193959
	$3\ 171774$	-0 688746	0.844566
Н	0.111111	0.000110	0.011000

Table S13: Optimized geometry for the TS of the reaction via the  $\rm H_3O^+$  with six explicit solvent molecules added.

С	0.634912	0.193769	-1.781157
Η	1.694622	0.422579	-1.634181
Η	0.334016	0.548553	-2.769128
Η	0.505607	-0.890246	-1.749449
С	-0.226436	0.880557	-0.712721
Η	-0.093950	1.970553	-0.798157
Η	-1.291711	0.662940	-0.831328
С	0.217867	0.566671	0.654263
Η	1.291448	0.600803	0.847010
С	-0.606971	0.038053	1.644931
Η	-0.220696	-0.017977	2.659169
Η	-1.684048	0.135324	1.516407
Η	-0.347665	-1.020361	1.141951
Ο	-0.170339	-2.533670	0.373602
Η	-0.781011	-2.355261	-0.373810
Η	0.743446	-2.391491	0.044628
Ο	2.485612	-2.066384	-0.268914
Η	2.838937	-1.166756	-0.100700
Ο	-2.094016	-1.856169	-1.503325
Η	-2.614073	-2.638819	-1.728328
Η	-2.698372	-1.283396	-0.986828
Η	3.008449	-2.653363	0.292850
Ο	3.651379	0.393745	0.295461
Η	4.584167	0.371667	0.045535
Η	3.302213	1.247990	-0.041734
Ο	2.573129	2.811817	-0.470434
Η	2.115893	2.742576	-1.319294
Η	1.868039	2.967167	0.190309
Ο	-3.781246	-0.230671	0.016474
Η	-3.470560	0.685095	0.022790
Η	-4.665140	-0.192767	-0.373369
Ο	0.537318	3.049426	1.426557
Η	-0.203278	3.606508	1.149281
Η	0.783473	3.368802	2.305526

Table S14: Optimized geometry for the TS of the reaction via the  $H_3O^+$  with seven explicit solvent molecules added.

С	0.409634	0.004482	-1.232263
С	0.405767	-0.674093	0.142743
С	-0.683219	-0.201780	1.007946
С	-0.516585	0.249666	2.317344
0	0.112335	2.807963	0.949746
0	2.663884	1.765136	0.870767
0	3.666748	-0.014003	-1.033746
0	-1.626722	2.586532	-1.207909
0	-3.281460	0.357660	-1.170483
0	-2.286300	-2.200728	-1.251563
Η	-0.591845	-0.007880	-1.673036
Η	1.090372	-0.524407	-1.902309
Η	0.740342	1.043423	-1.149075
Η	0.171383	-1.745978	0.017039
Η	1.369380	-0.609956	0.656874
Η	-1.662258	-0.064064	0.544229
Η	-1.404465	0.403550	2.925293
Η	0.416981	0.017663	2.826375
Η	-0.329599	1.294071	1.775910
Η	1.016503	2.493894	0.740804
Η	-0.425442	2.714904	0.134995
Η	-2.157845	1.761724	-1.204400
Η	3.324661	2.467250	0.931441
Η	2.994861	1.153318	0.180927
Η	-2.269489	3.300275	-1.102303
Η	-3.986174	0.457401	-1.823540
Η	-2.931531	-0.553420	-1.287837
Η	-1.429514	-2.243201	-1.697191
Η	-2.101939	-2.368699	-0.304699
Η	3.232673	-0.872434	-0.931212
Η	4.594414	-0.175649	-0.814535
0	-1.905210	-2.615387	1.475137
Η	-1.034623	-2.766973	1.901086
Η	-2.454562	-2.165353	2.130365
0	0.640701	-2.925167	2.494710
Η	1.224639	-3.120493	1.748748
Η	0.746329	-3.671481	3.099830

$\mathbf{C}$	1.569705	-1.557947	0.118542
С	1.876564	-0.140299	-0.371648
С	1.142152	0.900697	0.420621
С	0.355073	1.848722	-0.100949
Η	1.808976	-1.662319	1.181703
Η	2.155770	-2.296895	-0.434674
Η	0.510171	-1.799641	-0.013222
Η	2.954138	0.042401	-0.271915
Η	1.626676	-0.037348	-1.433364
Η	1.270328	0.855706	1.503627
Η	-0.145730	2.579432	0.528432
Η	0.205200	1.928173	-1.176575
Η	-1.157869	0.276884	0.090029
$\operatorname{Cl}$	-2.287566	-0.364320	-0.021967

Table S15: Optimized geometry for the reactants of the reaction via the HCl.

Table S16: Optimized geometry for the product of the reaction via the HCl.

С	2.189207	-0.224083	0.112439
$\mathbf{C}$	1.066123	0.711529	-0.318328
$\mathbf{C}$	-0.266054	0.450256	0.364773
С	-1.337220	1.452649	-0.015102
Η	2.320225	-0.199337	1.199541
Η	3.135042	0.072703	-0.348561
Η	1.981252	-1.256920	-0.180748
Η	1.332655	1.746108	-0.068113
Η	0.922185	0.672756	-1.404773
Η	-0.135071	0.409023	1.448551
Η	-1.010424	2.451027	0.293249
Η	-1.494175	1.458283	-1.097941
Η	-2.285664	1.231842	0.479499
$\operatorname{Cl}$	-0.863433	-1.231035	-0.069612

С	1.498771	-1.257696	-0.023077
С	1.520550	0.266922	-0.167393
С	0.321014	0.904654	0.467211
С	-0.604103	1.622101	-0.185032
0	-2.058646	-0.999111	-0.044391
Η	1.404494	-1.549641	1.027916
Н	2.419710	-1.697594	-0.415048
Н	0.655365	-1.687935	-0.571540
Η	2.418617	0.659845	0.324524
Η	1.572609	0.549971	-1.223757
Η	0.207291	0.748058	1.541788
Η	-1.445097	2.070259	0.337943
Η	-0.522790	1.803968	-1.254880
Η	-1.415866	-0.224915	-0.045905
Η	-2.733478	-0.861982	-0.739259
Н	-2.509078	-1.033036	0.823092

Table S17: Optimized geometry for the reactants of the reaction via the  $\rm H_3O^+$  without explicit solvent molecules added.

Table S18: Optimized geometry for the product of the reaction via the  $H_3O^+$  without explicit solvent molecules added.

С	-2.069108	-0.073789	0.096909
$\mathbf{C}$	-0.739984	-0.717781	-0.285854
$\mathbf{C}$	0.469915	-0.078112	0.360751
$\mathbf{C}$	1.790373	-0.708536	-0.001047
Ο	0.584544	1.364969	-0.077042
Η	-2.187970	-0.038691	1.184336
Η	-2.900359	-0.649977	-0.316315
Η	-2.157068	0.945709	-0.292626
Η	-0.725348	-1.764195	0.037801
Η	-0.598757	-0.716703	-1.373352
Η	0.345806	0.029427	1.439656
Η	1.784733	-1.742340	0.352235
Η	1.930565	-0.713791	-1.086331
Η	2.620499	-0.181583	0.473070
Η	0.664814	1.457356	-1.048760
Η	-0.160445	1.924335	0.222071

Table S19: Optimized geometry for the reactants of the reaction via the  $H_3O^+$  with one explicit solvent molecule added.

С	-0.551565	1.760848	0.637694
С	-1.225512	0.928483	-0.455819
С	-1.710331	-0.396401	0.056981
$\mathbf{C}$	-1.444026	-1.578530	-0.510374
Ο	1.356916	-1.167940	0.858345
Η	-1.251478	1.965414	1.454218
Η	-0.207035	2.719790	0.240372
Η	0.311106	1.237497	1.062001
Η	-2.086443	1.488318	-0.843219
Η	-0.542485	0.766891	-1.297209
Η	-2.321645	-0.366740	0.960609
Η	-1.830852	-2.506360	-0.098469
Η	-0.844779	-1.644451	-1.417421
Η	0.403495	-1.293809	0.653348
Η	1.799514	-0.431693	0.135770
Η	1.793834	-2.033884	0.781776
Ο	2.261894	0.442749	-0.703032
Η	3.193878	0.276259	-0.915534
Η	2.221009	1.337898	-0.329635

Table S20: Optimized geometry for the products of the reaction via the  $\rm H_3O^+$  with one explicit solvent molecule added.

С	-1.605762	1.520887	0.075528
С	-1.133728	0.251138	-0.626290
С	-0.554291	-0.792095	0.314779
С	-0.047000	-2.030463	-0.388818
Ο	0.606165	-0.231928	1.046314
Η	-2.330280	1.285661	0.862084
Η	-2.085059	2.197348	-0.636369
Η	-0.770264	2.062686	0.531675
Η	-1.974022	-0.229231	-1.139123
Η	-0.385074	0.487631	-1.393172
Η	-1.268905	-1.046703	1.101070
Η	-0.874849	-2.501491	-0.924313
Η	0.728974	-1.766142	-1.114149
Η	0.360357	-2.747879	0.326970
Η	1.330220	0.284950	0.378166
Η	0.326010	0.382354	1.747776
Ο	2.170690	0.867281	-0.444991
Η	2.988952	1.130877	0.004990
Η	1.783792	1.680315	-0.807379

С	-2.240886	0.957392	0.603498
С	-2.075761	-0.092706	-0.498277
С	-1.627832	-1.415269	0.054154
С	-0.593656	-2.127511	-0.406166
Ο	1.189620	0.208722	0.762737
Ο	3.382147	-0.676302	0.000586
Ο	0.518387	2.202552	-0.585560
Η	-3.004965	0.645874	1.322736
Η	-2.545756	1.920262	0.182727
Η	-1.305062	1.103369	1.153480
Η	-3.040273	-0.227954	-1.004047
Η	-1.359774	0.248151	-1.254867
Η	-2.200054	-1.799046	0.900473
Η	-0.317574	-3.079374	0.038315
Η	-0.004841	-1.783570	-1.255677
Η	0.516167	-0.492931	0.641072
Η	0.901701	1.048444	0.208773
Η	2.116904	-0.153631	0.426100
Η	4.122301	-0.302465	0.501313
Η	-0.187181	2.714569	-0.162337
Η	1.254304	2.819705	-0.710228
Η	3.561692	-0.472620	-0.929193

Table S21: Optimized geometry for the reactants of the reaction via the  $H_3O^+$  with two explicit solvent molecules added.

2.357086	-0.297241	-1.113486
1.726547	-0.548750	0.252295
0.247266	-0.885702	0.186246
-0.397550	-1.047953	1.548008
-0.428847	0.188411	-0.552674
-2.821903	-0.443207	-0.808322
-0.003730	2.388027	0.565444
2.181005	-1.142943	-1.786485
3.437217	-0.159371	-1.018461
1.945510	0.600897	-1.582909
2.217025	-1.394376	0.746035
1.864068	0.321911	0.906409
0.088741	-1.773282	-0.432740
0.080291	-1.874584	2.080624
-0.274024	-0.134919	2.139130
-1.464705	-1.269142	1.453905
-0.273412	1.114857	-0.098829
-1.448767	-0.019485	-0.656204
-3.324620	0.141816	-1.394012
0.948638	2.487423	0.716072
-0.260334	3.138454	0.008856
-3.280886	-0.425226	0.044638
	$\begin{array}{c} 2.357086\\ 1.726547\\ 0.247266\\ -0.397550\\ -0.428847\\ -2.821903\\ -0.003730\\ 2.181005\\ 3.437217\\ 1.945510\\ 2.217025\\ 1.864068\\ 0.088741\\ 0.080291\\ -0.274024\\ -1.464705\\ -0.273412\\ -1.448767\\ -3.324620\\ 0.948638\\ -0.260334\\ -3.280886\end{array}$	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$

Table S22: Optimized geometry for the products of the reaction via the  $H_3O^+$  with two explicit solvent molecules added.

С	1.856274	0.433136	1.367576
С	1.900972	1.112330	-0.004774
С	0.810231	2.133005	-0.154327
С	-0.100482	2.154340	-1.132340
Ο	-1.498643	-0.111599	0.612418
Ο	-0.232110	-1.481523	-0.963735
Ο	2.063024	-2.496000	-0.066174
Ο	-3.988067	-0.335498	0.366013
Η	1.864434	1.181878	2.166552
Η	2.721177	-0.220024	1.514190
Η	0.949122	-0.168983	1.478820
Η	2.869402	1.616278	-0.117526
Η	1.826404	0.368637	-0.807728
Η	0.777741	2.906235	0.615388
Η	-0.861914	2.927989	-1.180913
Η	-0.100781	1.404280	-1.921999
Η	-1.245406	0.813615	0.428698
Η	-0.936879	-0.733394	-0.076732
Η	-2.521288	-0.218825	0.487324
Η	-4.357905	-0.960932	1.006086
Η	0.000395	-0.933606	-1.728497
Η	0.626095	-1.835154	-0.605364
Η	-4.256014	-0.663309	-0.504615
Η	2.312170	-3.234540	-0.639044
Η	2.777649	-1.850046	-0.159632

Table S23: Optimized geometry for the reactants of the reaction via the  $\rm H_3O^+$  with three explicit solvent molecules added.

С	1.353190	-1.669752	-1.079106
С	0.688460	-1.500841	0.283309
$\mathbf{C}$	-0.752904	-1.027016	0.199398
$\mathbf{C}$	-1.398751	-0.813215	1.554316
Ο	-0.769810	0.230226	-0.552976
Ο	0.704232	1.876564	0.515917
Ο	3.167491	1.089958	-0.148222
Ο	-3.164533	0.919577	-0.855569
Η	0.784348	-2.363658	-1.706798
Η	2.365506	-2.068417	-0.967467
Η	1.425985	-0.715869	-1.609605
Η	0.676672	-2.455749	0.819434
Η	1.256079	-0.794841	0.903836
Η	-1.341993	-1.718834	-0.409939
Η	-1.418653	-1.757370	2.105143
Η	-0.828464	-0.081362	2.136364
Η	-2.426227	-0.454226	1.444698
Η	-0.152084	0.971047	-0.077982
Η	-1.739946	0.567420	-0.675737
Η	-3.295376	1.601151	-1.530829
Η	0.602684	1.827917	1.478211
Η	1.638535	1.600294	0.318044
Η	-3.540767	1.279911	-0.039166
Η	3.868330	1.546008	0.337589
Η	3.286372	0.150926	0.053500

Table S24: Optimized geometry for the products of the reaction via the  $H_3O^+$  with three explicit solvent molecules added.

С	-0.985093	2.073201	0.045161
С	-0.789578	0.758586	0.795795
С	-0.746580	-0.464331	-0.090530
С	-0.549298	-1.764217	0.643614
0	-2.112212	-0.568667	-0.759284
Η	-0.238878	2.184198	-0.748439
Η	-0.877447	2.918005	0.729883
Η	-1.980989	2.146745	-0.404431
Η	0.179543	0.766242	1.305348
Η	-1.564200	0.618233	1.558021
Η	-0.037556	-0.345110	-0.911123
Η	0.455194	-1.747414	1.073173
Η	-1.282482	-1.876648	1.446900
Η	-0.615111	-2.618578	-0.036253
Η	-2.264172	0.140531	-1.417409
Η	-2.216249	-1.420741	-1.230376
Cl	2.691961	0.009673	-0.215015

Table S25: Optimized geometry for the reactants of the  $S_N 2$  reaction without explicit solvent molecules added.

Table S26: Optimized geometry for the TS of the  $S_N 2$  reaction without explicit solvent molecules added.

С	-0.660643	2.109381	0.022024
$\mathbf{C}$	-0.552925	0.817961	0.819288
$\mathbf{C}$	-0.333271	-0.393255	-0.012396
$\mathbf{C}$	-0.259492	-1.728932	0.629235
Ο	-2.378009	-0.580002	-0.614210
Η	0.254451	2.276397	-0.553719
Η	-0.807010	2.961068	0.690391
Η	-1.505611	2.074714	-0.672431
Η	0.280783	0.858275	1.529068
Η	-1.449172	0.632047	1.422905
Η	-0.150051	-0.289025	-1.071801
Η	0.542462	-1.716066	1.371865
Η	-1.193945	-1.914260	1.168323
Η	-0.077154	-2.523228	-0.094120
Η	-2.554759	0.081072	-1.301117
Η	-2.459035	-1.435414	-1.063692
$\operatorname{Cl}$	2.293006	-0.070382	-0.309465

Table S27: Optimized geometry for the products of the  $S_N 2$  reaction without explicit solvent molecules added.

С	-0.343887	2.091309	0.166059
$\mathbf{C}$	-0.208936	0.764589	0.903399
$\mathbf{C}$	0.237619	-0.394431	0.027260
$\mathbf{C}$	0.202113	-1.729193	0.743075
Ο	-3.193701	-0.486418	-0.634541
Η	0.619057	2.427715	-0.229165
Η	-0.722690	2.867866	0.835940
Η	-1.042333	2.001658	-0.673693
Η	0.481783	0.856591	1.749988
Η	-1.182845	0.472745	1.317066
Η	-0.354915	-0.429024	-0.890718
Η	0.829497	-1.705715	1.639335
Η	-0.829226	-1.935973	1.049625
Η	0.539349	-2.541020	0.094881
Η	-2.612634	0.269737	-0.793023
Η	-2.595777	-1.203256	-0.384073
$\operatorname{Cl}$	1.946993	-0.093154	-0.569329

Table S28: Optimized geometry for the reactants of the  $S_N 2$  reaction with one explicit solvent molecule added.

С	0.315591	2.110374	-0.024882
$\mathbf{C}$	0.249692	0.776928	-0.762583
$\mathbf{C}$	0.187133	-0.426614	0.161440
С	0.185642	-1.751223	-0.565081
Ο	1.372851	-0.452465	1.056962
Ο	3.442503	-0.183301	-0.195525
Η	-0.534327	2.219747	0.657543
Η	0.290713	2.943510	-0.731806
Η	1.237222	2.201976	0.560090
Η	-0.658232	0.732417	-1.373633
Η	1.105981	0.664198	-1.439589
Η	-0.666707	-0.346111	0.837225
Η	-0.696083	-1.798363	-1.209120
Η	1.078963	-1.849727	-1.190256
Η	0.146375	-2.584903	0.139947
Η	2.319569	-0.318857	0.518185
Η	1.306469	0.222839	1.755609
Η	3.577810	0.744008	-0.447574
Η	4.227325	-0.446086	0.310169
Cl	-3.346250	-0.091489	0.109903

Table S29: Optimized geometry for the TS of the  $S_N 2$  reaction with one explicit solvent molecule added.

С	0.522672	1.848153	0.219959
$\mathbf{C}$	0.176642	0.658847	-0.663435
С	-0.377758	-0.503722	0.080533
$\mathbf{C}$	-0.647482	-1.774692	-0.635389
Ο	1.497563	-1.153857	0.940097
Ο	3.489872	0.194840	-0.387288
Η	-0.372636	2.215632	0.730450
Η	0.936282	2.664387	-0.377335
Η	1.261416	1.569251	0.978144
Η	-0.572403	0.928200	-1.415829
Η	1.045117	0.298583	-1.228604
Η	-0.633999	-0.399762	1.123649
Η	-1.313378	-1.572300	-1.477943
Η	0.293733	-2.149421	-1.048245
Η	-1.093780	-2.525718	0.015487
Η	2.231752	-0.673176	0.491195
Η	1.501645	-0.828721	1.852108
Η	3.255536	1.132285	-0.437519
Η	4.336653	0.171481	0.078877
Cl	-2.871756	0.321759	0.134299

Table S30: Optimized geometry for the products of the  $S_N 2$  reaction with one explicit solvent molecule added.

С	-0.397025	-1.711336	0.270032
$\mathbf{C}$	0.007452	-0.565551	-0.649460
С	0.961111	0.435421	-0.017841
С	1.218207	1.644594	-0.893316
Ο	-2.198900	1.403838	1.148830
Ο	-3.454007	-0.375155	-0.633575
Η	0.471057	-2.306320	0.569053
Η	-1.099474	-2.384234	-0.231136
Η	-0.876554	-1.332469	1.179954
Η	0.458039	-0.947390	-1.573325
Η	-0.882810	0.005704	-0.942061
Η	0.604531	0.739612	0.968968
Η	1.602601	1.340220	-1.871568
Η	0.271798	2.175187	-1.041267
Η	1.930183	2.330897	-0.429643
Η	-2.611425	0.772514	0.527346
Η	-1.586225	0.872096	1.672680
Η	-3.037853	-1.243728	-0.545252
Η	-4.375576	-0.512766	-0.376456
$\operatorname{Cl}$	2.565676	-0.385739	0.336128
-			

Table S31: Optimized geometry for the reactants of the  $S_N 2$  reaction with two explicit solvent molecules added.

$\begin{array}{cccccccccccccccccccccccccccccccccccc$	255 470 823 507
C -0.967513 -0.920151 1.2002 C -0.656050 -0.992318 -0.2874 C -1.459503 -2.045663 -1.0188	255 470 823 507
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	470 823 507
C -1.459503 -2.045663 -1.0188	823 507
	507
O -0.923339 0.298835 -0.932	
O 1.067659 1.702285 -0.6985	239
O -3.052672 1.348165 -0.0592	243
Н 0.837736 0.153433 1.7541	107
Н -0.428836 0.166623 2.9968	388
Н -0.585768 1.192539 1.5675	538
Н -0.673838 -1.885227 1.6277	729
Н -2.053836 -0.832340 1.3337	779
Н 0.418964 -1.141781 -0.4363	333
Н -1.262250 -3.026932 -0.5789	950
Н -2.530439 -1.835767 -0.9292	260
Н -1.190853 -2.076878 -2.0773	372
Н -1.787374 0.740088 -0.5799	990
Н -0.075276 0.957769 -0.8374	493
Н 1.311086 2.119384 -1.5379	917
Н -3.072272 1.256728 0.9054	165
Н -3.084526 2.301101 -0.2299	932
Н 1.822073 1.089225 -0.460	549
Cl 3.269639 -0.206277 0.0068	807

Table S32: Optimized geometry for the TS of the  $S_N 2$  reaction with two explicit solvent molecules added.

С	-0.418011	0.103957	1.827133
$\mathbf{C}$	0.061970	-0.932831	0.822257
С	0.541703	-0.355265	-0.463653
С	0.896898	-1.260213	-1.586242
Ο	-1.424783	0.165668	-1.234899
Ο	-1.205789	2.772869	-0.305857
Ο	-3.242988	-1.248308	0.326399
Η	0.377645	0.824368	2.043446
Η	-0.709011	-0.375658	2.764972
Η	-1.284632	0.650675	1.441481
Η	0.891688	-1.521761	1.226941
Η	-0.724239	-1.657689	0.578937
Η	0.660700	0.712951	-0.562293
Η	1.641214	-1.982430	-1.242400
Η	0.004836	-1.823961	-1.874287
Η	1.283376	-0.714170	-2.446219
Η	-2.096811	-0.314056	-0.703569
Η	-1.486314	1.105447	-0.961773
Η	-0.797353	3.374007	-0.943454
Η	-3.069436	-1.025614	1.251738
Η	-4.165572	-1.000778	0.177617
Η	-0.535745	2.641094	0.379736
Cl	2.970751	0.132461	0.293353

С	0.439264	-0.450936	-1.653803
$\mathbf{C}$	-0.131401	-1.213235	-0.463424
$\mathbf{C}$	-1.007333	-0.378023	0.456456
$\mathbf{C}$	-1.424966	-1.113887	1.712755
Ο	2.077284	0.852814	1.312101
Ο	0.575230	2.775090	-0.190516
Ο	3.315354	-1.211473	-0.202902
Η	-0.355529	-0.015545	-2.267311
Η	1.023095	-1.117796	-2.296096
Η	1.096558	0.359985	-1.320076
Η	-0.703630	-2.086623	-0.798476
Η	0.690461	-1.590671	0.158793
Η	-0.515257	0.564680	0.703598
Η	-1.950699	-2.040713	1.463977
Η	-0.525471	-1.366844	2.283748
Η	-2.070333	-0.496493	2.341544
Η	2.455349	0.167149	0.731182
Η	1.567041	1.448552	0.734024
Η	-0.015010	3.200673	0.446175
Η	2.930727	-1.243444	-1.089595
Η	4.244813	-0.985813	-0.342089
Η	-0.014317	2.340496	-0.822272
$\operatorname{Cl}$	-2.520831	0.145144	-0.446382

Table S33: Optimized geometry for the products of the  $S_N 2$  reaction with two explicit solvent molecules added.

Table S34: Optimized geometry for the reactants of the  $S_N 2$  reaction with three explicit solvent molecules added.

С	0.314495	-0.574920	-1.536466
$\mathbf{C}$	0.060365	-1.063148	-0.113463
$\mathbf{C}$	-0.426186	0.030544	0.824488
$\mathbf{C}$	-0.603326	-0.441261	2.250924
Ο	0.528149	1.148476	0.847479
Ο	2.868979	0.381211	0.900914
Ο	3.486375	-1.461733	-0.970619
Ο	-0.057294	2.729606	-1.013739
Η	-0.580754	-0.096057	-1.949244
Η	0.578479	-1.408663	-2.194243
Η	1.134432	0.151948	-1.569298
Η	-0.715924	-1.836006	-0.116562
Η	0.967153	-1.519498	0.306467
Η	-1.355679	0.460357	0.440310
Η	-1.328513	-1.259401	2.274387
Η	0.346342	-0.808181	2.654328
Η	-0.971419	0.367588	2.886895
Η	1.544152	0.822078	0.849663
Η	0.349363	1.804498	0.069291
Η	0.439800	3.560325	-1.000638
Η	3.492884	1.108737	0.764349
Η	3.072971	-0.296611	0.207616
Η	0.102411	2.339285	-1.885727
Η	2.700438	-1.845032	-1.384602
Η	3.940261	-2.206551	-0.552606
$\operatorname{Cl}$	-3.788127	-0.554505	-0.380327

Table S35: Optimized geometry for the TS of the  $S_N 2$  reaction with three explicit solvent molecules added.

С	0.400433	-0.395668	-1.484317
$\mathbf{C}$	-0.045931	-0.919801	-0.126785
$\mathbf{C}$	-0.967743	-0.006787	0.604807
$\mathbf{C}$	-1.339771	-0.313621	2.009956
Ο	0.538223	1.483849	1.103856
Ο	3.036455	0.358753	0.846740
Ο	3.232856	-1.993295	-0.639304
Ο	-0.063989	3.059808	-1.099067
Н	-0.466988	-0.095558	-2.081420
Н	0.939715	-1.168293	-2.038057
Η	1.062709	0.468411	-1.372646
Η	-0.572896	-1.874453	-0.225956
Н	0.807275	-1.119505	0.534116
Η	-1.381661	0.854859	0.104612
Η	-1.730498	-1.332594	2.064253
Η	-0.435370	-0.278102	2.624271
Η	-2.078392	0.385725	2.400430
Η	1.427682	1.074616	1.000375
Η	0.441454	2.107163	0.353139
Η	0.366937	3.921168	-1.179238
Η	3.640398	0.973364	0.409660
Η	3.072802	-0.465623	0.318345
Η	0.185611	2.577324	-1.899370
Н	2.404038	-2.484865	-0.720034
Н	3.839518	-2.597295	-0.190192
$\operatorname{Cl}$	-3.161919	-0.847294	-0.460121

Table S36: Optimized geometry for the Product of the  $S_N 2$  reaction with three explicit solvent molecules added.

С	0.432177	-0.244783	-1.413452
$\mathbf{C}$	0.073119	-0.942896	-0.107273
$\mathbf{C}$	-1.226606	-0.474336	0.526383
$\mathbf{C}$	-1.501974	-1.123598	1.866934
Ο	0.956643	2.109881	1.256111
Ο	3.174716	0.836127	0.046978
Ο	3.240924	-1.941148	-0.288374
Ο	-1.179632	2.855242	-0.504629
Η	-0.323229	-0.427897	-2.183264
Η	1.392681	-0.606250	-1.792710
Η	0.517531	0.836938	-1.263179
Η	0.027150	-2.031054	-0.247425
Η	0.854588	-0.748413	0.640043
Η	-1.237525	0.613996	0.612624
Η	-1.527507	-2.213407	1.772367
Η	-0.696021	-0.853764	2.557426
Η	-2.448780	-0.781100	2.290199
Η	1.657852	1.613001	0.793010
Η	0.289079	2.333749	0.583109
Η	-1.058913	3.705497	-0.948278
Η	3.180635	1.204137	-0.846519
Η	3.144158	-0.134728	-0.076923
Η	-1.306986	2.216082	-1.219519
Η	2.360393	-2.338987	-0.244673
Η	3.728007	-2.332261	0.449521
$\operatorname{Cl}$	-2.632625	-0.836629	-0.599532