

Electronic Supplementary Material (ESI) for Chemical Society Reviews.
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Electronic Supplementary Information (ESI) of “Hydrovoltaic Technology: from Mechanism to Applications”

Various hydrovoltaic devices can harvest mechanical energy and latent heat from the earth water cycle, including bulk-water waving and flowing, droplets dragging and impinging, water evaporation and moisture absorption. And for different water form, there are also differences in material system, device structure and power generation capacity, and we here summarize the main achievements in Table S1. Note that careful evaluations of chemical interactions should be taken into account when metal alligator clips were employed in their experiments.

Table S1 Overview of water-energy harvestings and performance comparisons

Water form	Materials	V_{oc}	I_{sc}	Solution	Electrode	Marks	Reference
Bulk water motion	CVD Gr/PET	0.1 V	~11 μ A	NaCl	Ag emulsion	Waving potential	Ref S1
	CVD Gr/PET pair	~1 V	~2 μ A	NaCl	Ag emulsion	One Gr pair fixed	Ref S2
	Gr-CB/PU	0.02 V	~10 μ A	NaCl	Cu	Ocean wave energy	Ref S3
	ZnO nanofilms/PET	~0.05 V	1.3 μ A	NaCl	Ag emulsion	--	Ref S4
	Metal oxide nanolayers	millivolt	~0.2 μ A	NaCl	--	Heterogeneous salinity	Ref S5
	MoS ₂ -encapsulated C fibers	0.54 V	~2 mA	NaCl	Cu wire	DC	Ref S6
	PTFE tube	~904 V	509 μ A	Rain water	Al and Cu	Volume effect	Ref S7
Water droplet	Gr/PET	<1 mV	< 1 μ A	NaCl	Ag emulsion	Drawing potential	Ref S8
	Gr/PVDF	~0.5 V	~25 μ A	NaCl	Silver epoxy	Pre-polarized PVDF	Ref S9
	MoS ₂ /PEN	~5 V	~5 nA	NaCl	Silver paste	--	Ref S10
	Silicon	~0.3 V	0.64 μ A	Water	Ti/Au	Droplet between PN junction	Ref S11
	Silicon/Gr	~0.3 V	~0.7 μ A	Water	Ti/Au	Droplet between PN junction	Ref S12
	Silica/PV4P/Cu _x O	~0.03 V	~6.4 nA	NaCl	--	Electrolyte-insulator-	Ref S13

						semiconductor	
	Silica gel+APTES+POTS	~1 V	~0.55 μ A	NaCl	ITO	Surface functionality control	Ref S14
	CYTOP and PTFE	~42 V	--	Water	Silicon	Planar structure, alternate contacts	Ref S15
	PTFE	~4 V	~0.5 μ A	Water	ITO	Press-releasing mode	Ref S16
	PTFE	~150 V	~250 μ A	Tap water	Al/ITO	Transient pulse	Ref S17
	FEP	~200 V	~400 μ A	Tap water	Al/Cu	Superhydrophobic	Ref S18
Water evaporation	Carbon black	~1 V	~150 nA	DI water	MWCNTs	DC output for 8 days	Ref S19
	All-printed CB slurry	~0.5 V	~300 nA	DI water	MWCNTs	Additives of terpineol, ethyl cellulose and ethanol	Ref S20
	Cotton fabric/CB	0.53 V	3.91 μ A	Salt solutions	Alligator clips	KCl, NaCl, LiCl, HCl	Ref S21
	Carbon black	0.79 V	~120 nA	DI water	CNTs	Al ₂ O ₃ substrate	Ref S22
	3D G membrane	~0.28 V	~10 μ A	NaCl	Cu	Cu/3DG/Cu, vertical	Ref S23
	Porous GO sponges	0.63 V	27.4 μ A	DI water	SACNT bucky papers	Partially-reduced	Ref S24
	rGO film	~1 V	~10 μ A	NaCl	CNTs	Honeycomb structured	Ref S25
	Metal oxides film	~2.5 V	0.18 μ A	DI water	Carbon paste	PET substrate	Ref S26
	Ni-Al LDH film	~0.7 V	~1.3 μ A	DI water	CNTs	PET substrate	Ref S27
	Porous ZnO film	~0.4 V	~20 nA	DI water	Ag/AgCl	--	Ref S28
	CuO nanowire film	~0.45 V	0.23 μ A	NaCl	Alligator clips	Droplet Infiltrations	Ref S29
	TiO ₂ nanoparticle film	~0.2 V	~120 nA	DI water	MWCNTs	Atmosphere monitoring	Ref S30
	AlOOH/UIO-66 MOF	1.63 V	~500 nA	DI water	Silver slurry	--	Ref S31
	CBAP/PVDF	~0.2 V	~500 nA	NaCl	Pt	Solar steam desalination	Ref S32
	paper-based CuCAT-1 MOF	~0.55 V	~30 μ A	NaCl	Alligator clips	Cogenerating freshwater	Ref S33
	Cu(BDC-OH) MOF	~0.6 V	~120 nA	DI Water	CB/Cu paste	Hierarchical Oriented	Ref S34
	PEDOT:PSS/GO sponge	2.13 V	~8 μ A/cm ²	DI water	Cu	Thermoelectric	Ref S35
	Wasted tree gum	~0.4 V	~3 μ A	DI water	--	Carbonization	Ref S36
	Carbon nanoparticles/ TiO ₂ nanowires	~1.3 V	~350 nA	DI water	Carbon ink	Photo-response performance	Ref S37
	Gr / α -FeOOH film	~0.4 V	~3 μ A	NaCl	CNTs	H ₂ O ₂ boosting	Ref S38
MoS ₂ /SiO ₂ film	~0.8 V	0.25 μ A	NaCl	Pt	--	Ref S39	
3D wood	0.3 V	10 μ A	DI water	Carbon paste	Natural directional channels	Ref S40	

	Cellulose fabrics	~0.7 V	~3 μ A	NaCl	Cu	--	Ref S41
	Silicon nanowire arrays	0.4 V	55 μ A/cm ²	DI water	Graphite and Ag paint	Directional nanochannels	Ref S42
	Silicon nanowire arrays	0.55 V	22 μ A/cm ²	DI water	Hierarchical Nanofabric	Graphite/PEDOT:PSS/fabric and silver paint	Ref S43
	Carbon black-glass fiber hybrid film	~5.0 V	~1.5 μ A	DI water	CNT ink	Two films with opposite surface charges	Ref S44
	PVA/FCB/3DS film	0.7 V	~63 μ A	Hydrogel	Cu	PVA skeleton	Ref S45
	d-Al ₂ O ₃ /gelatin	6.4 V	~0.7 μ A	DI water	Conductive carbon paste	Combination with thermoelectric	Ref S46
	Carbon slurry	1.2 V	~0.5 μ A	DI water	CNT ink	--	Ref S47
	Ethanol carbon black	~1.0 V	~100 nA	DI water	Ag	Memristor	Ref S48
	Toluene carbon black	~1.2 V	475 nA	DI water	CNTs	Gas monitor	Ref S49
(Solar enhancement)	CNTs/cellulose paper	0.47-0.6 V	5-22 μ A	Water	Ni/Cu	Solar-energy enhancement	Ref S50
	CB/PVA+wood	~1.2 V	~4.5 μ A	Water	Ag	Asymmetric evaporation	Ref S51
	Ti ₃ C ₂ TX film	~40 mV	~12 μ A	DI water	Ag	Asymmetric irradiation	Ref S52
	CNT film	~0.48 V	~40 mA/g	Water	CNTs	Asymmetric irradiation	Ref S53
Moisture	3-layers g-GO film	~20 mV	~5 μ A/cm ²	Δ RH=30 %	Au	Breath sensor	Ref S54
	3D g-GO framework	0.26 V	3.2 mA/cm ²	Δ RH=75 %	Al	--	Ref S55
	Graphene quantum dots	0.15 V	~16.8 mA/cm ²	Δ RH=50 %	Au	High specific surface area	Ref S56
	GO/Paper (r=1mm)	~0.7 V	~38 μ A/cm ²	Δ RH=60 %	Ag paste	Printable, flexible	Ref S57
	Pristine Graphene-Oxide	~0.4 V	~1.5 μ A/cm ²	Δ RH=70 %	Au	--	Ref S58
	Flexible GO film	~0.2 V	~1.2 μ A	Δ RH=80 %	rGO	Direct laser reduction	Ref S59
	g-rGO/GO	1.5 V	~43.3 nA/cm ²	Δ RH=80 %	Au	Schottky junction	Ref S60
	In-plane graphene oxide	70 mV	12 mA/cm ²	Δ RH=60 %	rGO	Direct laser reduction	Ref S61

GO fiber	0.35 V	1.06 mA cm ⁻²	$\Delta_{RH}=65\%$	rGO	Laser-reduced GO as electrodes	Ref S62
3D polypyrrole (PPy) framework	60 mV	$\sim 12 \mu\text{A}/\text{cm}^2$	$\Delta_{RH}=85\%$	Au	Anion gradients	Ref S63
Polypyrrole (PPy) nanowire array	~ 80 mV	~ 150 nA	$\Delta_{RH}=80\%$	Au	Sodium ion gradient doped	Ref S64
1D polypyrrole nanoarray	~ 0.15 V	~ 500 nA	$\Delta_{RH}=75\%$	Au	High-valent metal carrier transport	Ref S65
PSSA membrane	~ 0.8 V	$0.1 \text{ mA}/\text{cm}^2$	$\Delta_{RH}=80\%$	Au	Various polymers, flexible	Ref S66
PSS/PVA film	~ 0.6 V	$13.2 \mu\text{A}/\text{cm}^2$	$\Delta_{RH}=85\%$	Ag NWs	Transparent, self-healing, arbitrary tailorable	Ref S67
Porous polydopamine layer	0.52 V	$3.1 \text{ mA}/\text{cm}^2$	$\Delta_{RH}=90\%$	Ag	Pulse/DC, Wearable sensor	Ref S68
Thin gelatin film	0.71 V	$\sim 7.7 \mu\text{A}/\text{cm}^2$	$\Delta_{RH}=90\%$	Cu/Al	Cu/protein/Al	Ref S69
Print paper	0.25 V	$\sim 10 \text{ nA}/\text{cm}^2$	$\Delta_{RH}=70\%$	ITO/Au	Asymmetry electrode	Ref S70
TiO ₂ nanowire networks	~ 0.6 V	$\sim 9 \mu\text{A}/\text{cm}^2$	$\Delta_{RH}=85\%$	Al	Breathing sensor	Ref S71
Bulk rGO	~ 0.45 V	$4.48 \mu\text{A}/\text{cm}^2$	RH=85%	Au	DC, flexible integration	Ref S72
GO/PAAS	~ 0.6 V	$> 1 \mu\text{A}/\text{cm}^2$	RH=85%	Au/Ag	DC, > 160h	Ref S73
PSSA/PDDA	1.38 V	$\sim 4 \mu\text{A}/\text{cm}^2$	RH=85%	Carbon tape	DC, heterogenous structure	Ref S74
Toluene CB	~ 65 mV	$0.6 \text{ nA}/\text{cm}^2$	RH>95%	Cu	DC, Asymmetric functionalization	Ref S75
Carbon-coated cotton fiber	0.74 V	$0.83 \mu\text{A}/\text{cm}^2$	RH=37%	Alligator clips	DC, CaCl ₂ modification	Ref S76

Waste corn stalk	~0.25 V	~0.5 $\mu\text{A}/\text{cm}^2$	RH=75%	Cu	DC, KCl modification	Ref S77
Cellulose acetate membranes	0.32 V	~80 nA/cm^2	RH~80%	Ag gauze	DC or pulse, electrostatic spinning	Ref S78
Biological nanofibrous	0.1 V	~30 nA/cm^2	RH=99%	Pt nets	DC or pulse, Cellulose, chitin, silk fibroin, and amyloid	Ref S79
Protein wires	~0.5 V	17 $\mu\text{A}/\text{cm}^2$	RH~50%	Au	DC, > 2month, a spontaneous water gradient	Ref S80

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