

1 Deoxygenation of Stearic Acids using Alkaline Treated Beta Molecular Sieves

2 Assisted by Microwave Irradiation

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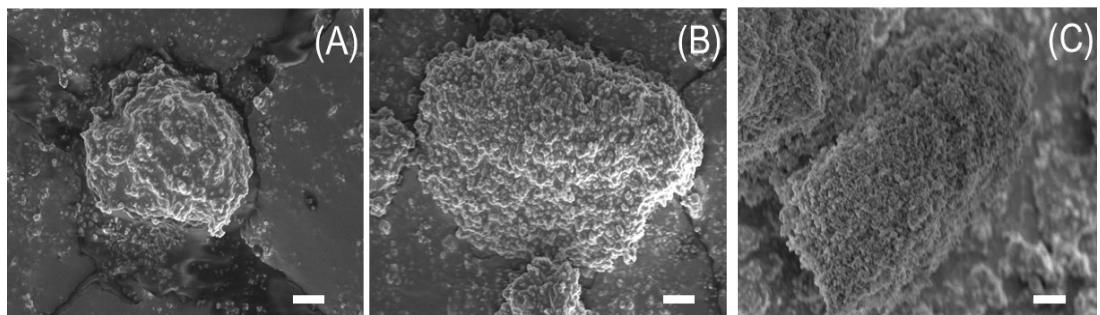
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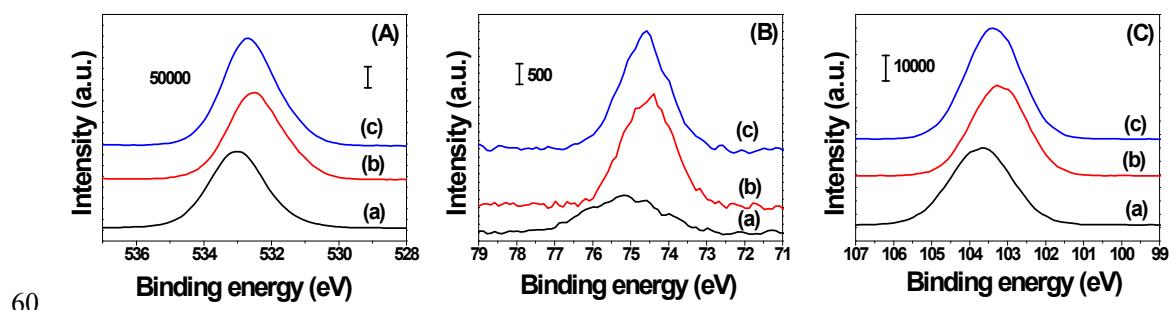
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60 **Figure S2** O_{1s} (A), Al_{2p} (B), and Si_{2p} (C) XPS spectra of Beta-1 (a), Beta-2 (b), and Beta-3 (c)

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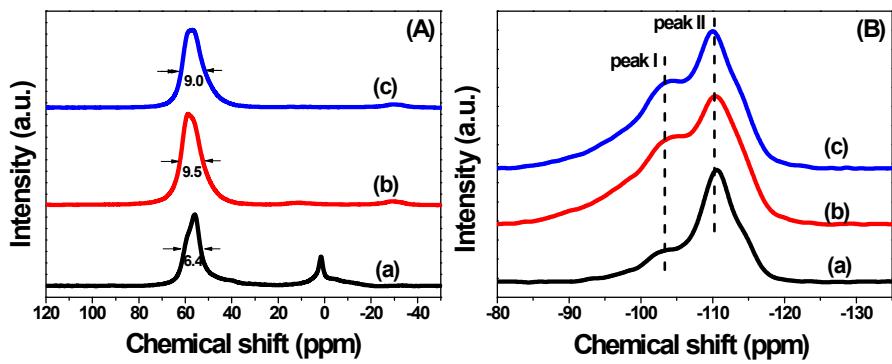
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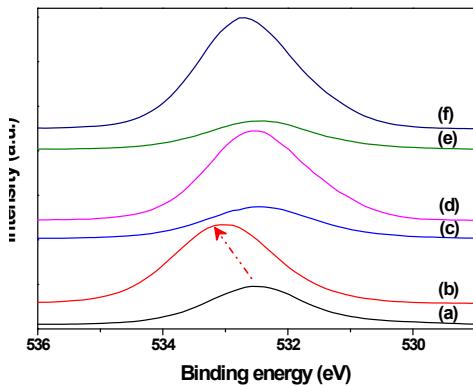
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91 **Figure S3** ^{27}Al and ^{29}Si MAS NMR spectra of Beta-1 (a), Beta-2 (b), and Beta-3 (c)
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107 **Figure S4 XPS spectra of the O 1s core level for Beta-1 (a), Ni/Beta-1 (b), Beta-2 (c), Ni/Beta-
108 2 (d), Beta-3 (e), and Ni/Beta-3 (f) samples recorded at room temperature. Notes: Spectra of
109 (a), (c), and (e) are cited from Figure S2 (C). Spectra (a, c, and e) and (b, d, and f) were
110 detected in two tests. Therefore, the intensity of spectra is different and spectra are
111 employed to compare the binding energy.**

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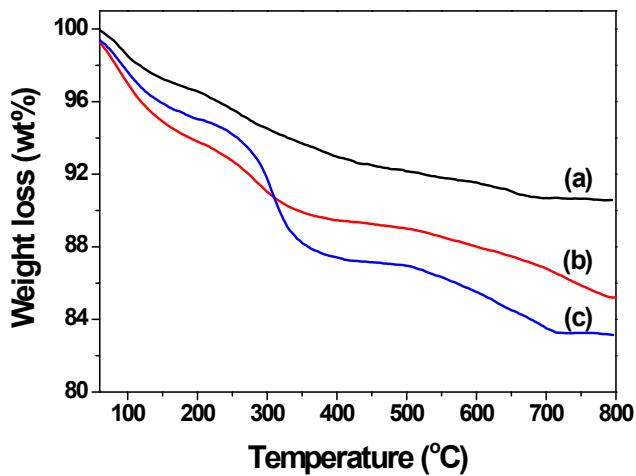
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121 **Figure S5 TG curves of spent Ni/Beta-1 (a), Ni/Beta-2 (b), and Ni/Beta-3 (c) catalysts.**

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Table S1 The composition of gas products

	methane(wt%)	ethane(wt%)	ethylene(wt%)	propane(wt%)
Ni/Beta-1	62%	12%	22%	4%
Ni/Beta-2	33%	12%	22%	33%
Ni/Beta-3	32%	13%	21%	34%

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Table S2 Catalytic performance of catalysts in references

	Catalyst	Reaction conditions	Conversion (%)	C ₁₅₋₁₈ selectivity (%)	Iso-paraffin selectivity (%)	Ref
1	5% Pt/ γ -Al ₂ O ₃	T=375 °C, P=2 MPa, t=5 h	100	97	<1	1
2	4% Pd/C	T=300 °C, P=1.7MPa, t=5h	100	98	<1	2
3	10% Ni/ γ -Al ₂ O ₃	T=330 °C, P=5MPa,	100	88	<1	3
4	7% Ni/SiO ₂		49	67	2	4
5	7% Ni/ γ -Al ₂ O ₃		91	88	<1	4
6	7% Ni/SAPO-11	T=220 °C, P=2MPa, t=6 h	99	93	3	4
7	7% Ni/H-ZSM-5		99	53	11	4
8	7% Ni/H-Y		99	30	13	4
9	Ni/Al ₂ O ₃	T=360 °C, P=2 MPa, t=0.75h	100	58	1	5
10	4% Pd/C	T=375 °C, P=1.5MPa, t=5 h	90	43	<1	6

- 159 [1] A. T. Madsen, E. H. Ahmed, C. H. Christensen, R. Fehrman and A. Riisager, *Fuel*, 2011, **90**,
160 3433-3438.
- 161 [2] S. Lestari, I. Simakova, A. Tokarev, M. A. Päivi, E. Kari and D. Y. Murzin, *Catal. Lett.*, 2008,
162 **122**, 247-251.
- 163 [3] R. Kaewmeesri, A. Srifa, V. Itthibenchapong and K. Faungnawakij, *Energ. Fuel.*, 2015, **29**,
164 833-840.
- 165 [4] H. Zuo, Q. Liu, T. Wang, L. Ma, Q. Zhang and Q. Zhang, *Energ. Fuel.*, 2012, **26**, 3747-3755
- 166 [5] Q. B. Ayodele, Z. Gholami and Y. Uemura, *J. Energy. Chem.*, 158-168.
- 167 [6] B. H. Susanto, M. Nasikin, Sukirno and A. Wiyo, *Procedia. Chem.*, 2014, **9**, 139-150.