

## Electronic supplementary information (ESI)

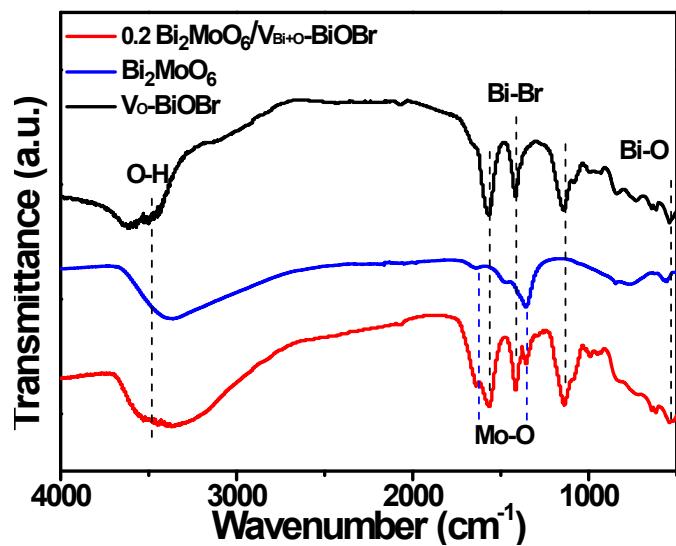
### Boosted photocatalytic nitrogen fixation by bismuth and oxygen vacancies in $\text{Bi}_2\text{MoO}_6/\text{BiOBr}$ composite structures

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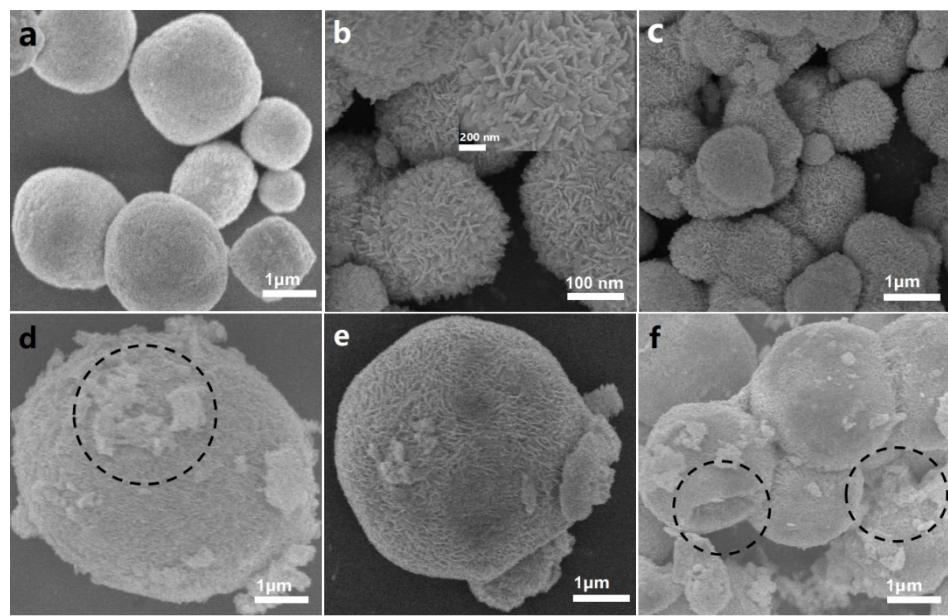
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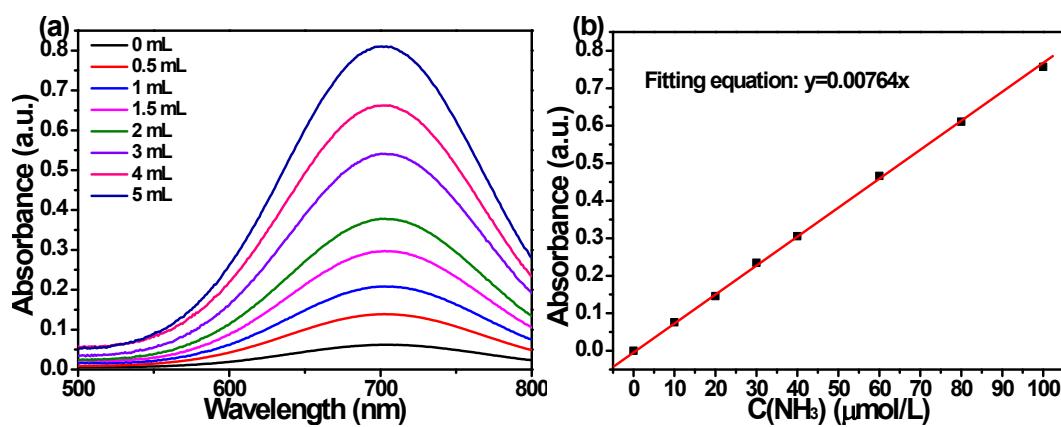
*<sup>†</sup> These authors contributed equally to this work.*



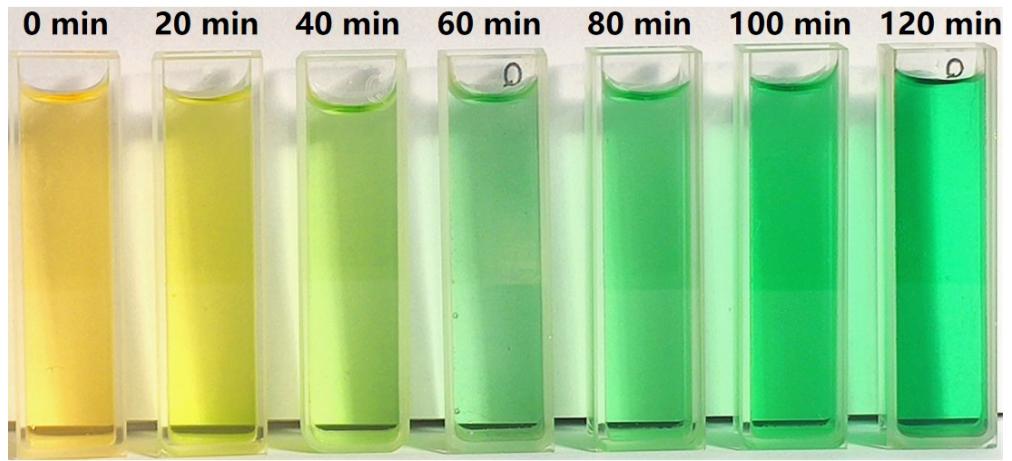
**Fig. S1** FT-IR spectrum of  $\text{Bi}_2\text{MoO}_6$ ,  $\text{V}_0\text{-BiOBr}$  and  $0.2\text{ Bi}_2\text{MoO}_6/\text{V}_{\text{Bi}+\text{O}}\text{-BiOBr}$  samples, respectively.



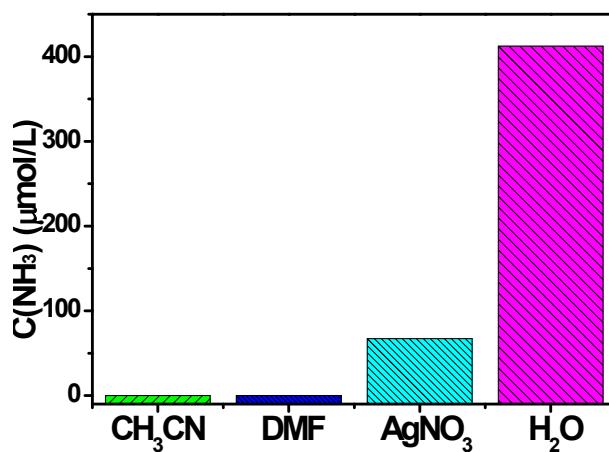
**Fig. S2** (a) SEM images of Vo-BiOBr, (b-c)  $\text{Bi}_2\text{MoO}_6$ , (d-e) 0.1  $\text{Bi}_2\text{MoO}_6/\text{V}_{\text{Bi}+\text{O}}\text{-BiOBr}$  and (f) 0.3  $\text{Bi}_2\text{MoO}_6/\text{V}_{\text{Bi}+\text{O}}\text{-BiOBr}$  composite.



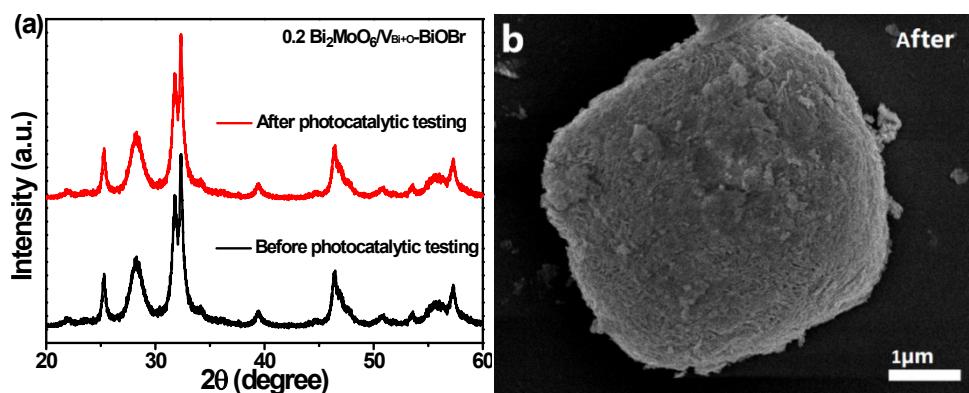
**Fig. S3** Calibration for  $\text{NH}_4^+$  determination. (a) Absorption spectra of ammonia chloride solutions with different concentrations through the indophenol blue method and (b) Linear relationship between the peak absorbance at 700 nm and the  $\text{NH}_4^+$  concentration.



**Fig. S4** Photograph of the aliquots obtained during N<sub>2</sub> photofixation with the 0.2 Bi<sub>2</sub>MoO<sub>6</sub>/V<sub>Bi+O</sub>-BiOBr sample and subjected to the indophenol-blue test. The colors of the solutions change gradually with the irradiation time.



**Fig. S5** Photocatalytic  $\text{N}_2$  fixation with using aprotic solvents ( $\text{CH}_3\text{CN}$  and  $\text{DMF}$ ) instead of water or added  $\text{AgNO}_3$  as the electron scavenger.



**Fig. S6** (a) XRD patterns and (b) SEM image of 0.2 Bi<sub>2</sub>MoO<sub>6</sub>/V<sub>Bi+O</sub>-BiOBr before and after five cycles of photocatalytic nitrogen fixation tests.

**Table S1.** XPS surface element analysis of the V<sub>O</sub>-BiOBr and 0.2 Bi<sub>2</sub>MoO<sub>6</sub>/V<sub>Bi+O</sub>-BiOBr samples.

Sample	Bi atom%	O atom%	Br atom%	Mo atom%	Bi :O
V <sub>O</sub> -BiOBr	<b>24.38</b>	<b>31.25</b>	<b>24.82</b>	<b>0</b>	<b>0.78:1</b>
<b>0.2 Bi<sub>2</sub>MoO<sub>6</sub>/V<sub>Bi+O</sub>-BiOBr</b>	<b>21.52</b>	<b>29.83</b>	<b>23.16</b>	<b>4.15</b>	<b>0.72:1</b>

**Table S2.** Elemental composition of the V<sub>O</sub>-BiOBr and 0.2 Bi<sub>2</sub>MoO<sub>6</sub>/V<sub>Bi+O</sub>-BiOBr samples by inductively coupled plasma mass spectrometry (ICP-MS).

Sample	Bi atom%	O atom%	Br atom%	Mo atom%	Bi :O
V <sub>O</sub> -BiOBr	<b>25.64</b>	<b>30.16</b>	<b>28.94</b>	<b>0</b>	<b>0.85:1</b>
<b>0.2 Bi<sub>2</sub>MoO<sub>6</sub>/V<sub>Bi+O</sub>-BiOBr</b>	<b>24.77</b>	<b>30.98</b>	<b>29.15</b>	<b>4.24</b>	<b>0.80:1</b>

**Table S3.** Comparison of the photocatalytic performance of relevant catalysts towards N<sub>2</sub> fixation.

Catalyst	Reaction medium	Scavenger	Light source	Ammoniageneration rate	Reference
<b>0.2 Bi<sub>2</sub>MoO<sub>6</sub>/V<sub>Bi+O-</sub> BiOBr</b>	H <sub>2</sub> O(l), 25 °C	No	<b>300 W Xe lamp, Full range light</b>	<b>206 μmol/L/h</b>	<b>This work</b>
<b>V<sub>0</sub>-BiOBr nanosheets</b>	H <sub>2</sub> O(l), 25 °C	No	<b>300 W Xe lamp, Full range light</b>	<b>54.7 μmol/g/h</b>	<b>S10</b>
<b>BiOBr-001-OV</b>	H <sub>2</sub> O(l), 25 °C	No	<b>300 W Xe lamp, λ &gt; 420 nm</b>	<b>104.2 μmol/g/h</b>	<b>S11</b>
<b>5 nm Bi<sub>5</sub>O<sub>7</sub>Br-NT</b>	H <sub>2</sub> O(l), 25 °C	No	<b>300 W Xe lamp, λ &gt; 400 nm</b>	<b>1.38 mmol/g/h</b>	<b>S39</b>
<b>defect-rich SUC Bi<sub>3</sub>O<sub>4</sub>Br</b>	H <sub>2</sub> O(l), 25 °C	No	<b>300 W Xe lamp, λ &gt; 400 nm</b>	<b>25.4 μmol/L/h</b>	<b>S15</b>
<b>p-n Bi<sub>2</sub>MoO<sub>6</sub>/OV-BiOBr</b>	H <sub>2</sub> O(l), 25 °C	No	<b>300 W Xe lamp, λ &gt; 400 nm</b>	<b>90.7 μmol/g/h</b>	<b>S21</b>
<b>0.5% Fe-Bi<sub>2</sub>MoO<sub>6</sub></b>	H <sub>2</sub> O(l), 25 °C	No	<b>300 W Xe lamp, λ &gt; 400 nm</b>	<b>106.5 μmol/g/h</b>	<b>S25</b>
<b>Bi<sub>5</sub>O<sub>7</sub>Br-40°C</b>	H <sub>2</sub> O(l), 25 °C	No	<b>300 W xenon lamp (200-800 nm)</b>	<b>12.72 mmol/g/h</b>	<b>S32</b>
<b>BiOBr-Fe-S-1</b>	H <sub>2</sub> O(l), 25 °C	No	<b>300 W Xe lamp, λ &gt; 400 nm</b>	<b>46.1 μmol/g/h</b>	<b>S12</b>