

Electronic Supplementary Material (ESI) for Catalysis Science & Technology.

Electronic Supporting Information

A Novel Heterogeneous Catalyst NH₂-MIL-88/PMo₁₀V₂ for Photocatalytic Activity Enhancement of Benzene Hydroxylation

PanPan Xu¹, Liuxue Zhang^{*1}, Xu Jia^{*1}, Hao Wen¹, Xiulian Wang², Suqing Yang¹, Juxian Hui¹

¹ School of Materials and Chemical Engineering, Zhongyuan University of Technology, Zhengzhou, 450007, PR China

² School of Energy and Environment, Zhongyuan University of Technology, Zhengzhou, 450007, PR China

Correspondence: zlx100100@163.com, jjax@zut.edu.cn; Tel.: +86 -731-62506699; Fax: +86-731-62506095

- S1 The infrared partial spectra of (a) NH₂-MIL-88, (b) H₅PMo₁₀V₂O₄₀ and (c) NH₂-MIL-88/PMo₁₀V₂-3
- S2 SEM images of the (a) NH₂-MIL-88, (b) PMo₁₀V₂ and (c) NH₂-MIL-88/PMo₁₀V₂-3, EDS mapping of (d) NH₂-MIL-88, (e) PMo₁₀V₂ and (f) NH₂-MIL-88/PMo₁₀V₂-3
- S3 XPS spectra of the catalyst (a) V2p and (b) Fe2p
- S4 EPR spectra of fresh NH₂-MIL-88/PMo₁₀V₂-3 and recycled NH₂-MIL-88/PMo₁₀V₂-3
- S5 UV-vis spectra of the fresh NH₂-MIL-88/PMo₁₀V₂-3 catalyst and NH₄VO₃
- S6 UV-vis spectra of the recovered NH₂-MIL-88/PMo₁₀V₂-3 catalyst and VO(acac)₂
- S7 UV-Vis diffuse reflectance of the (a) NH₂-MIL-88, (b) PMo₁₀V₂ and (c) NH₂-MIL-88/PMo₁₀V₂-3
- S8 The infrared fractionated gain spectra of (a) recycled NH₂-MIL-88/PMo₁₀V₂-3 and (b) fresh NH₂-MIL-88/PMo₁₀V₂-3
-

S9 The apparent quantum yield (AQY)

S10: Photo of the Photocatalytic instrument (PCX-50C)

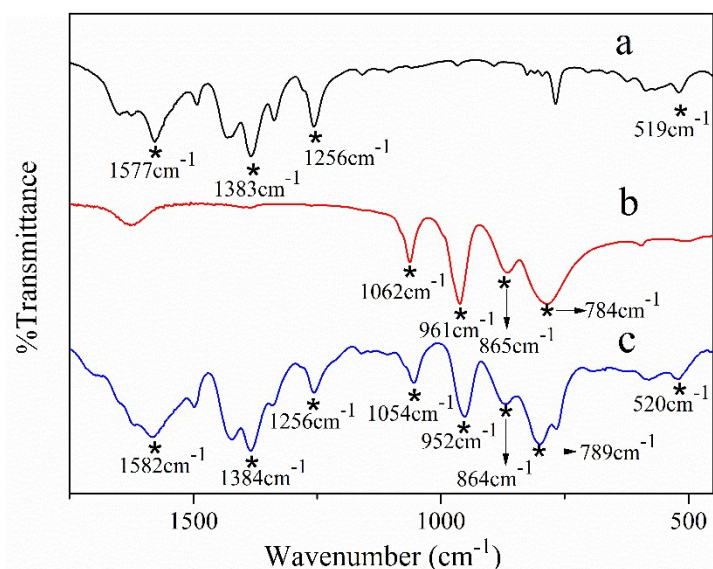


Fig S1. The infrared partial spectra of (a) NH₂-MIL-88, (b) H₅PMo₁₀V₂O₄₀ and (c) NH₂-MIL-88/PMo₁₀V₂-3

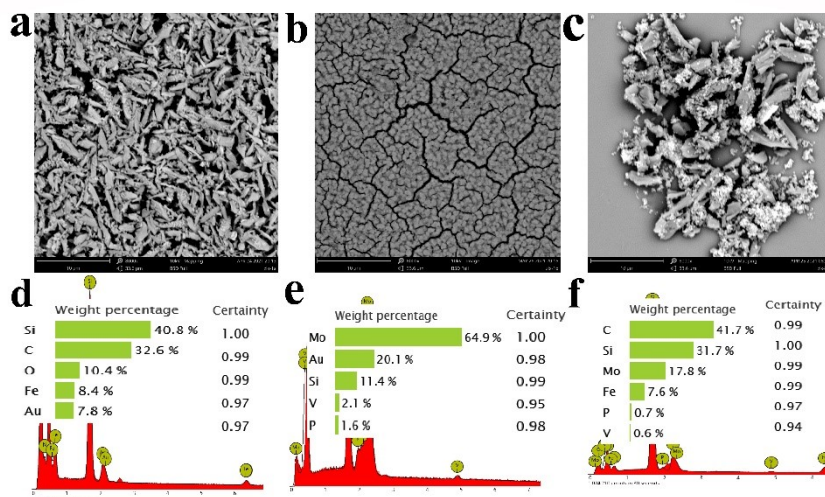


Figure S2 SEM images of the (a) NH₂-MIL-88, (b) PMo₁₀V₂ and (c) NH₂-MIL-88/PMo₁₀V₂-3, EDS mapping of (d) NH₂-MIL-88, (e) PMo₁₀V₂ and (f) NH₂-MIL-88/PMo₁₀V₂-3

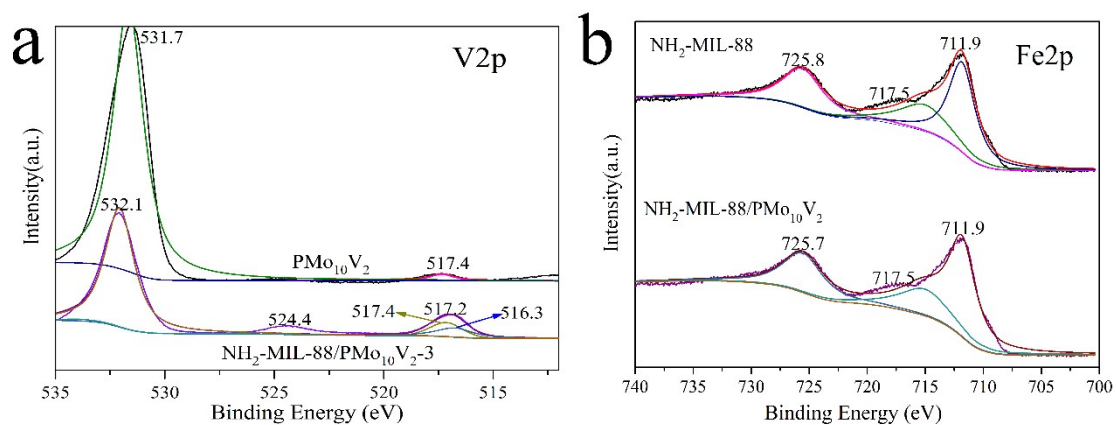


Figure S3 XPS spectra of the catalyst (a) V2p and (b) Fe2p

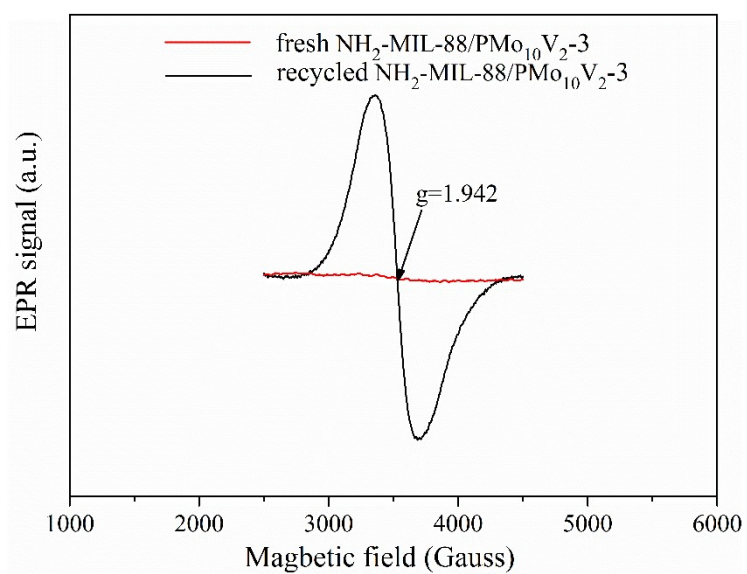


Figure S4 EPR spectra of fresh $\text{NH}_2\text{-MIL-88/PMo}_{10}\text{V}_2\text{-3}$ and recycled $\text{NH}_2\text{-MIL-88/PMo}_{10}\text{V}_2\text{-3}$

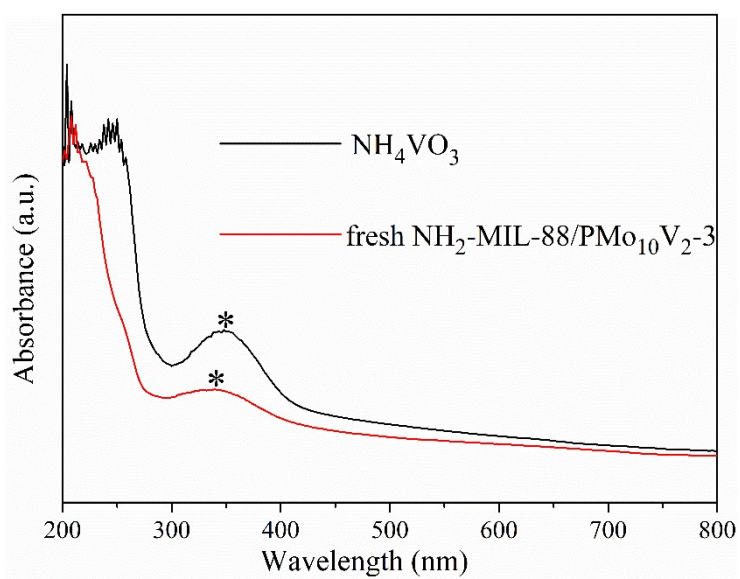


Fig S5 UV-vis spectra of the fresh $\text{NH}_2\text{-MIL-88/PMo}_{10}\text{V}_2\text{-3}$ catalyst and NH_4VO_3

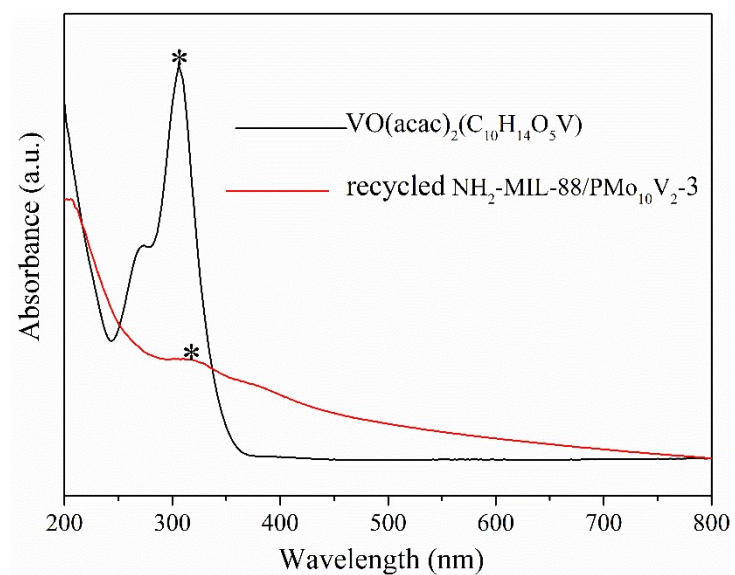


Fig S6 UV-vis spectra of the recovered $\text{NH}_2\text{-MIL-88/PMo}_{10}\text{V}_2\text{-3}$ catalyst and $\text{VO}(\text{acac})_2$

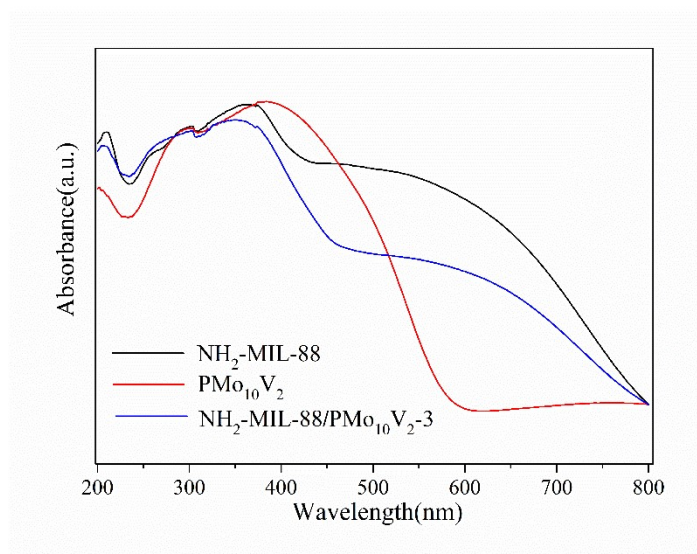


Figure S7 UV-Vis diffuse reflectance of the (a) NH₂-MIL-88, (b) PMo₁₀V₂ and (c) NH₂-MIL-88/PMo₁₀V₂-3

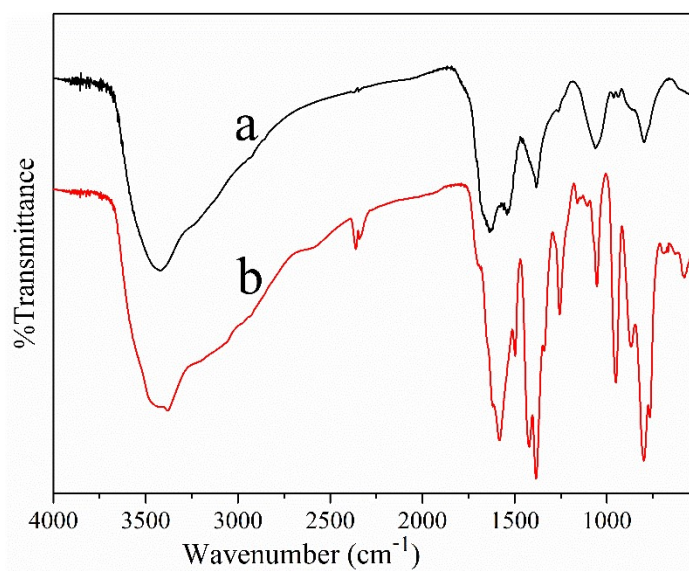


Fig S8. The infrared fractionated gain spectra of (a) recycled NH₂-MIL-88/PMo₁₀V₂-3 and (b) fresh NH₂-MIL-88/PMo₁₀V₂-3

Fig S9

Since the light source (white light and wavelength range from 320 nm to 780 nm) used in this experiment is fixed inside the PCX-50C instrument, the AQY can't be measured at a single wavelength. Hence, under optimal conditions, the Apparent Quantum Yield (AQY) of the photocatalytic hydroxylation of benzene to phenol at the minimum wavelength ($\lambda=320$ nm) and the maximum wavelength ($\lambda=780$ nm) was determined to be 3.55% and 1.45%, respectively according to the following equation, where R_1 (mol s^{-1}) is the hydroxylation of benzene to phenol rate in 3 h and R_2 (mol s^{-1}) is the rate of incident photon.

$$\text{AQY (\%)} = (R_1 / R_2) \times 100\%$$

As one electron produce one hydroxyl radical, hence one photon is required for this system. Assuming that all photons are absorbed by the catalyst, the resulting quantum yield can be calculated as the apparent quantum yield.

Fig S10

Photo of the Photocatalytic instrument (PCX-50C):

