

Boosted Dehydrogenation of Ethane in Porous Vanadium-based Single Crystals

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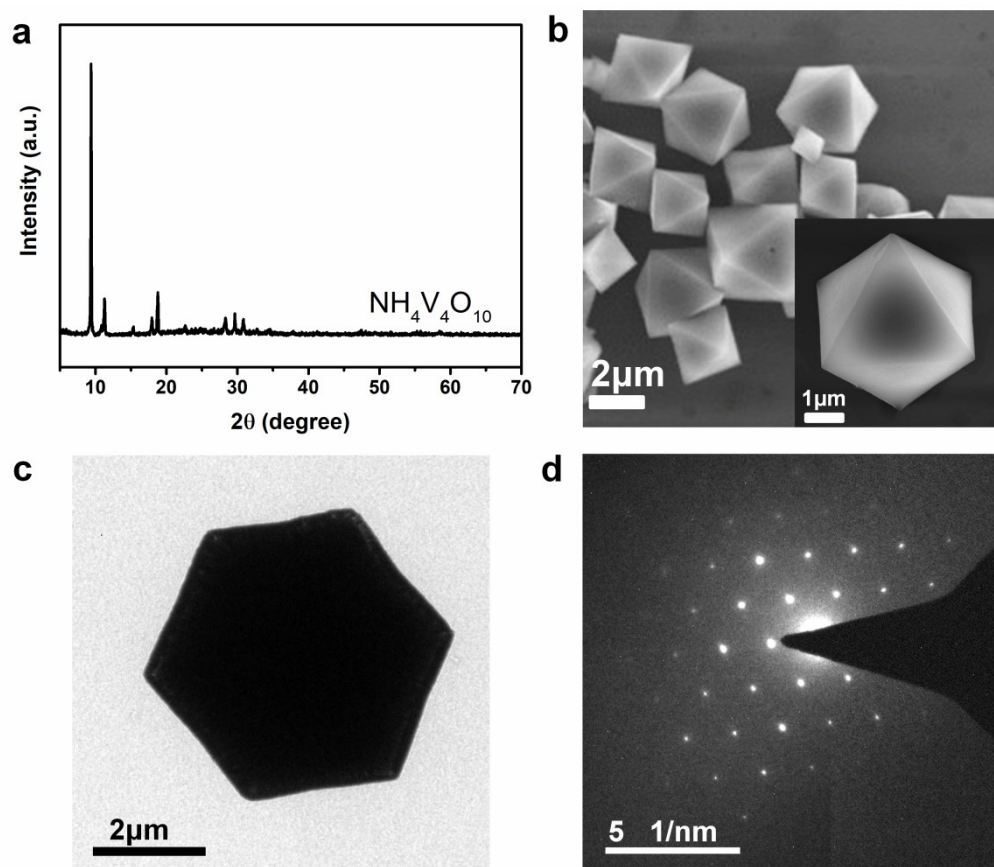


Fig. S1 (a) XRD, (b) SEM, (c) TEM and (d) HR-TEM images of $\text{NH}_4\text{V}_4\text{O}_{10}$ powder.

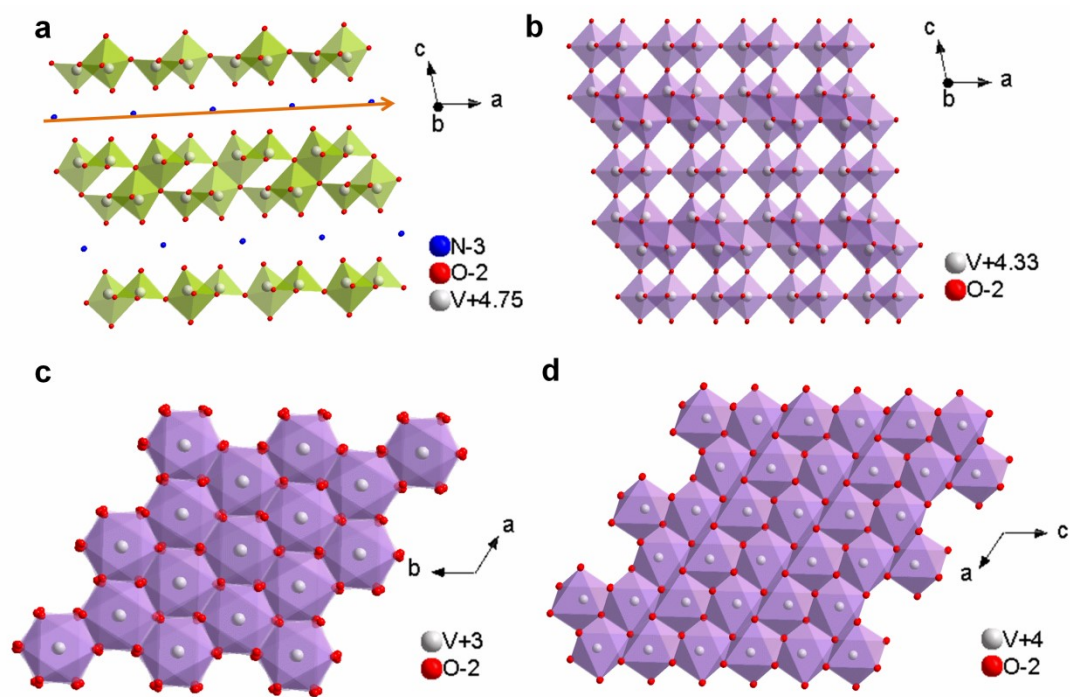


Fig. S2 (a) The lattice channel of atomic evaporation of $\text{NH}_4\text{V}_4\text{O}_{10}$ and the lattice structure of (b) V_6O_{13} , (c) VO_2 and (d) V_2O_3 .

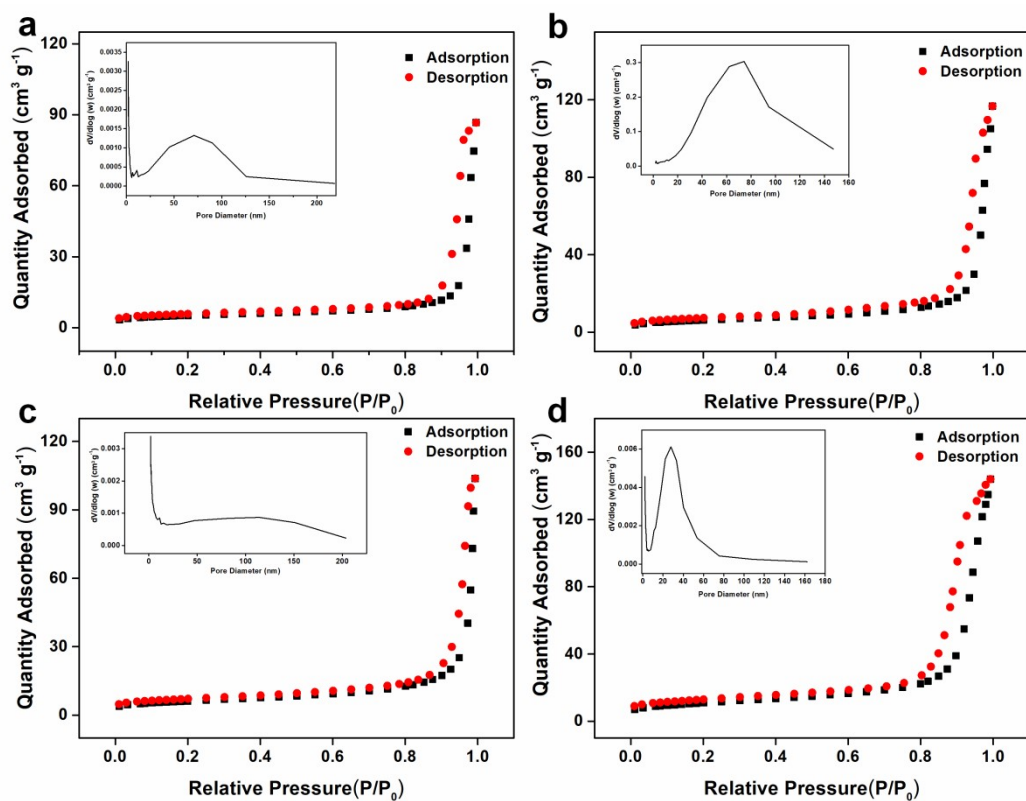


Fig. S3 N₂ adsorption-desorption isotherms and (inset) the corresponding pore diameter distribution of porous single crystal micron particles, (a) V_6O_{13} , (b) VO_2 , (c) V_2O_3 and (d) VN .

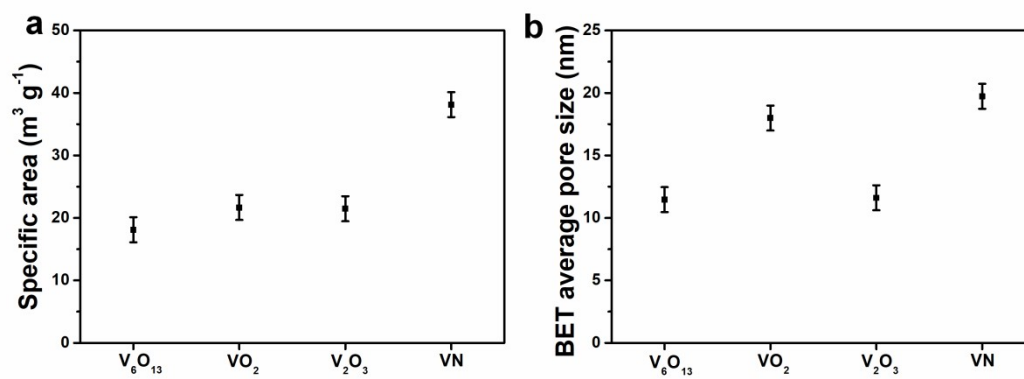


Fig. S4 The surface specific area (a) and average pore size (b) of the porous single crystal micro particles (The error bar indicates the standard deviation in repeated measurements).

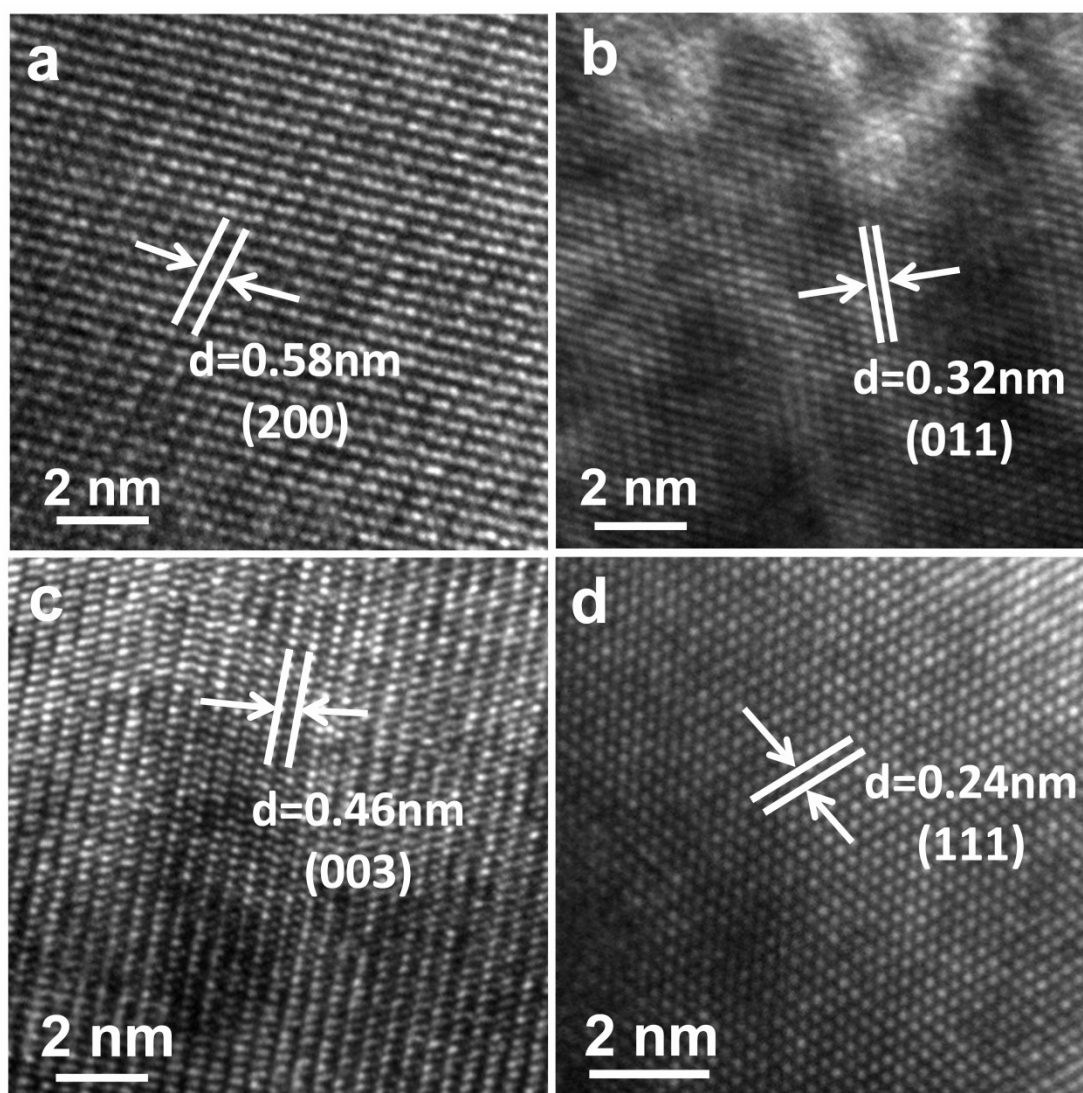


Fig. S5 HR-TEM images of (a) V_6O_{13} , (b) VO_2 , (c) V_2O_3 and (d) VN porous single crystal micro particles.

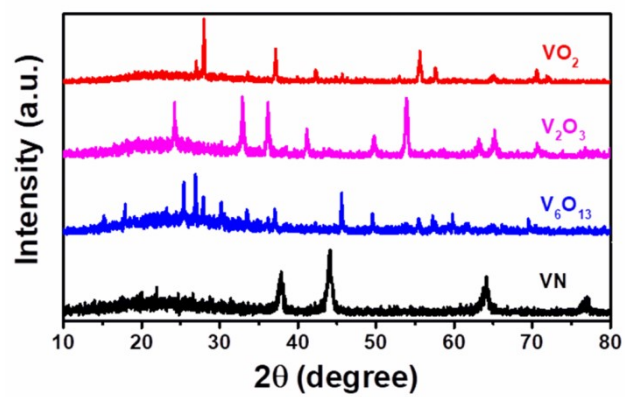


Fig. S6 XRD of V_6O_{13} , VO_2 , V_2O_3 and VN single crystal powders after ethane dehydrogenation.

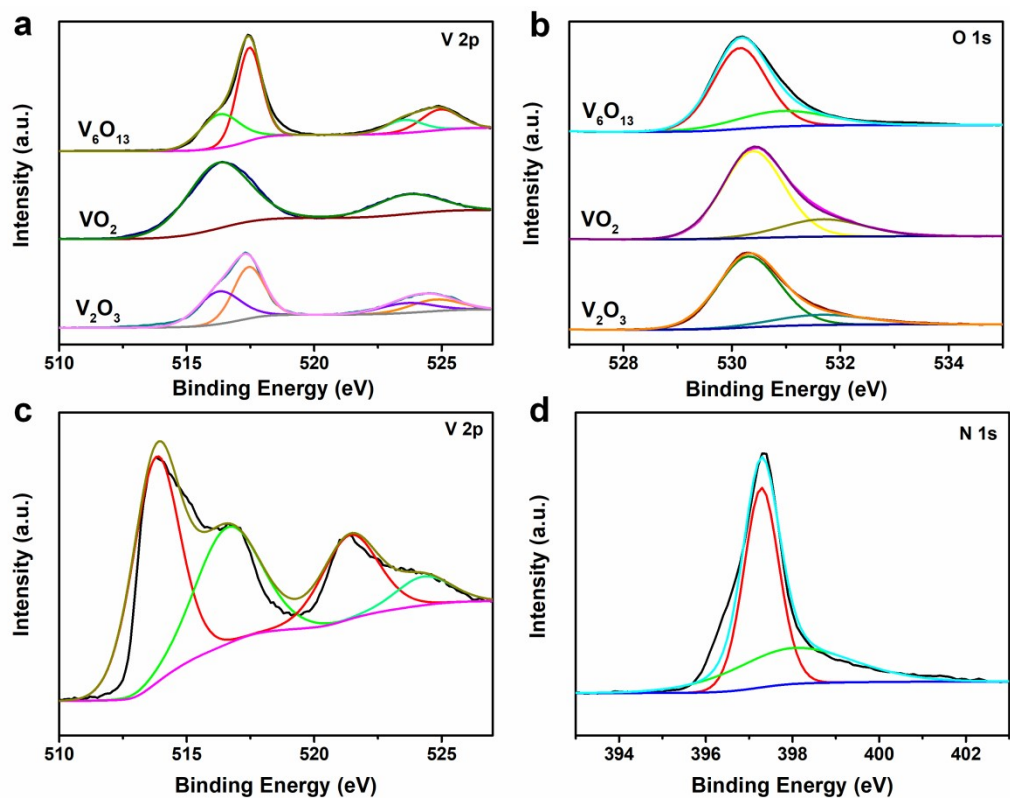


Fig. S7 XPS spectra of PSC V_6O_{13} , VO_2 , V_2O_3 and VN powders after ethane dehydrogenation. (a) V 2p peaks of V_6O_{13} , VO_2 and V_2O_3 . (b) O 1s peaks of V_6O_{13} , VO_2 and V_2O_3 . (c) V 2p peaks of VN. (d) N 1s peaks of VN.

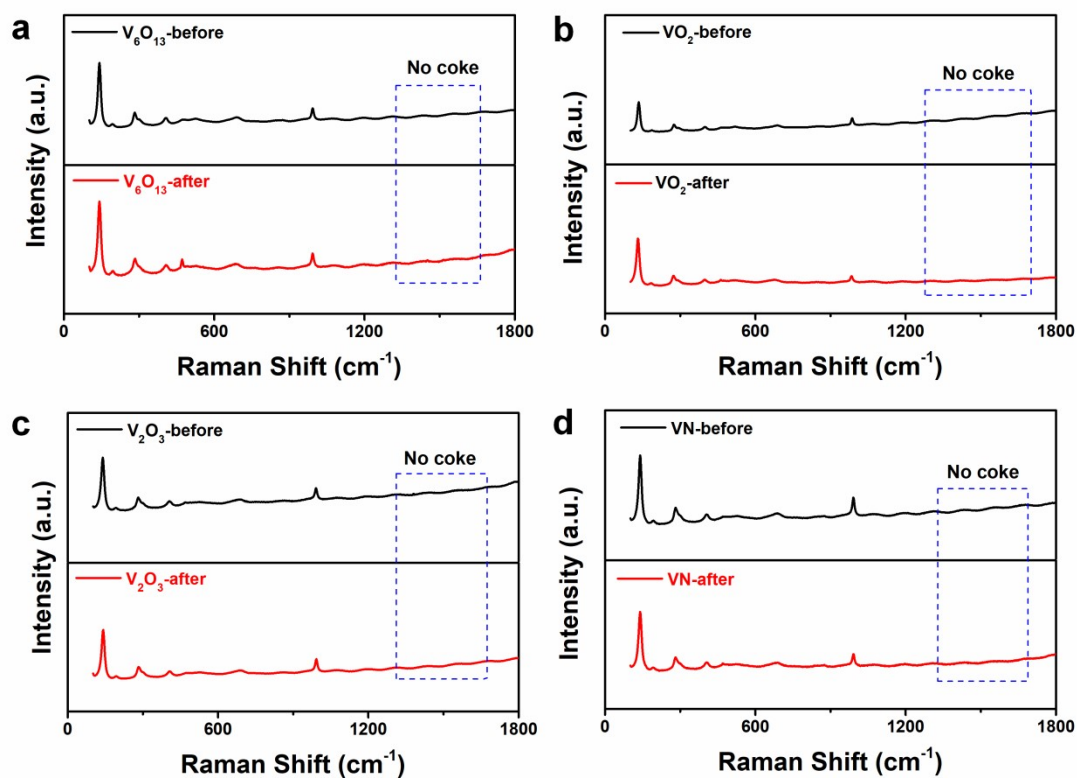


Fig. S8 Raman images of (a) V_6O_{13} , (b) VO_2 , (c) V_2O_3 and (d) VN powders before (black line) and after (red line) the dehydrogenation of ethane reaction.

Table S1 The ICP and EA results of vanadium-based materials.

Samples	Contents of V (wt)	Contents of O or N (wt)
V ₆ O ₁₃	59.64%	10.62%
VO ₂	44.87%	20.06%
V ₂ O ₃	58.02%	20.87%
VN	74.82%	21.08%