## Supporting Information

## Bimetallic persulfide nanoflake assembled by dealloying and sulfurization: a versatile electrocatalyst for overall water splitting and Zn-air batteries

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Fig. S1 XRD patterns of (a)  $Al_{90}Co_{2.5}Fe_{7.5}$ , (b)  $Al_{90}Co_5Fe_5$  and (c)  $Al_{90}Co_{7.5}Fe_{2.5}$  precursors.



Fig. S2 XRD patterns of (a)  $O-Co_{2.5}Fe_{7.5}$ , (b)  $O-Co_5Fe_5$  and (c)  $O-Co_{7.5}Fe_{2.5}$ .



Fig. S3 SEM images of (a) O-Co $_{2.5}$ Fe $_{7.5}$ , (b) O-Co $_{5}$ Fe $_{5}$  and (c) O-Co $_{7.5}$ Fe $_{2.5}$ .



**Fig. S4** EDX elemental mappings of Co, Fe, Al and S for (a) S-Co<sub>2.5</sub>Fe<sub>7.5</sub>, (b) S-Co<sub>2.5</sub>Fe<sub>7.5</sub> and (c) S-Co<sub>7.5</sub>Fe<sub>2.5</sub>.



Fig. S5 (a) TEM and (b) HRTEM images of S-Co<sub>2.5</sub>Fe<sub>7.5</sub>. (c) TEM and (b) HRTEM image of S-Co<sub>7.5</sub>Fe<sub>2.5</sub>.



**Fig. S6** XPS spectra of S-Co<sub>2.5</sub>Fe<sub>7.5</sub> sample. (a) Survey spectrum, (b) Co 2p, (c) Fe 2p, (d) Al 2p, (e) S 2p.



**Fig. S7** XPS spectra of S-Co<sub>7.5</sub>Fe<sub>2.5</sub> sample. (a) Survey spectrum, (b) Co 2p, (c) Fe 2p, (d) Al 2p, (e) S 2p.



Fig. S8 LSV curves of Pt/C measured before and after 5000 ADT continuous cycles.



**Fig. S9** (a) TEM image and (b-f) XPS spectra of S-Co<sub>5</sub>Fe<sub>5</sub> electrode after both ADT and subsequent CA test in 0.1 M KOH solution.



Fig. S10 Methanol tolerance comparison of (a) Pt/C and (b) S-Co<sub>5</sub>Fe<sub>5</sub>.



**Fig. S11** (a) OER polarization curves of S-Co<sub>2.5</sub>Fe<sub>7.5</sub>, S-Co<sub>5</sub>Fe<sub>5</sub>, S-Co<sub>7.5</sub>Fe<sub>2.5</sub> and RuO<sub>2</sub> in 1.0 M KOH. (b) HER polarization curves of S-Co<sub>2.5</sub>Fe<sub>7.5</sub>, S-Co<sub>5</sub>Fe<sub>5</sub>, S-Co<sub>7.5</sub>Fe<sub>2.5</sub> and Pt/C in 1.0 M KOH.



Fig. S12 CV curves of (a) S-Co<sub>2.5</sub>Fe<sub>7.5</sub>, (b) S-Co<sub>5</sub>Fe<sub>5</sub> and (c) S-Co<sub>7.5</sub>Fe<sub>2.5</sub> electrodes in the potential range of  $1.0 \sim 1.1$  V vs. RHE under different scan rates. (d) Capacitive currents at the middle of potential window as a function of scan rate.



**Fig. S13** (a) HER polarization curves of S-Co<sub>2.5</sub>Fe<sub>7.5</sub>, S-Co<sub>5</sub>Fe<sub>5</sub>, S-Co<sub>7.5</sub>Fe<sub>2.5</sub> and Pt/C in 0.1 M KOH. (b) Tafel plots. (c) Nyquist plots for S-Co<sub>2.5</sub>Fe<sub>7.5</sub>, S-Co<sub>3</sub>Fe<sub>5</sub> and S-Co<sub>7.5</sub>Fe<sub>2.5</sub> at -0.1 V vs. RHE (The inset is the equivalent circuit for fitting). (d) LSV curves of S-Co<sub>5</sub>Fe<sub>5</sub> measured before and after 5000 ADT continuous cycles (The inset is the *j*-*t* profile recorded at a -0.15 V for 40 h). (e) Polarization curves of S-Co<sub>5</sub>Fe<sub>5</sub> " S-Co<sub>5</sub>Fe<sub>5</sub> S-Co<sub>5</sub>Fe<sub>5</sub> in S-Co<sub>5</sub>Fe<sub>5</sub> " S-Co<sub>5</sub>Fe<sub>5</sub> water splitting in 0.1 M KOH. (f) Stability curves of S-Co<sub>5</sub>Fe<sub>5</sub> " S-Co<sub>5</sub>Fe<sub>5</sub> and S-Co<sub>5</sub>Fe<sub>5</sub> under constant currents of 10 and 20 mA cm<sup>-2</sup> (The inset is the photograph of overall water splitting).



**Fig. S14** (a) Photograph of the Zn-air battery device. (b) Photographs of the anode and cathode materials.



**Fig. S15** The discharge specific capacity plots of ZnAB (Zn plate  $_{11}^{11}$  S-Co<sub>5</sub>Fe<sub>5</sub>) and ZnAB (Zn plate  $_{11}^{11}$  Pt/C) at a current density of 10 mA cm<sup>-2</sup>.



Fig. S16 A schematic of the rechargeable Zn-air battery.

Catalysts		Element pro	portion (at. %)	
	Al	Co	Fe	S
S-Co <sub>2.5</sub> Fe <sub>7.5</sub>	3.97	11.81	38.07	46.15
S-Co <sub>5</sub> Fe <sub>5</sub>	2.74	16.12	15.94	65.2
S-Co <sub>7.5</sub> Fe <sub>2.5</sub>	1.93	37.35	12.74	47.98

Table S1 ICP analysis for S-Co<sub>x</sub>Fe<sub>10-x</sub> (x=2.5, 5, 7.5).

**Table S2** Comparison of ORR performance in 0.1 M KOH for the as-prepared catalystsin this study with the other reported catalysts in literatures.

Catalysta	Eonset	E <sub>1/2</sub>	Tafel slope	Deference
Catalysis	(V)	(V)	$(mV dec^{-1})$	Kelelence
S-Co <sub>2.5</sub> Fe <sub>7.5</sub>	0.84	0.62	169	This work
S-Co <sub>5</sub> Fe <sub>5</sub>	0.91	0.79	64	This work
S-Co <sub>7.5</sub> Fe <sub>2.5</sub>	0.86	0.72	85	This work
Pt/C	0.99	0.87	61	This work
NiCo <sub>2</sub> O <sub>4</sub> @NiCoFe-OH	0.89	0.77		1
NiO/NiCo <sub>2</sub> O <sub>4</sub>	0.89	0.73	85.4	2
Co/PCNF		0.78		3
CoSe <sub>2</sub>	0.82	0.75	107	4

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Cotolysta	$\eta_{onset}$	$\eta_{10}$	Tafel slope	Deference
Catalysis	(mV)	(mV)	(mV dec <sup>-1</sup> )	Kelefelice
S-Co <sub>2.5</sub> Fe <sub>7.5</sub>	150	390	114	This work
S-Co <sub>5</sub> Fe <sub>5</sub>	70	300	79	This work
S-Co <sub>7.5</sub> Fe <sub>2.5</sub>	130	330	86	This work
RuO <sub>2</sub>	200	370	96	This work
$W-N/C-4@Co_9S_8@WS_2$		560	36	5
LaMnNiCoO <sub>3</sub> (1:2:3)		370		6
ZnCoMnO <sub>4</sub> /N-rGO	340	480	158	7
Fe/Ni-N-C		322	69	8

**Table S3** Comparison of OER performance in 0.1 M KOH for the as-prepared catalysts in this study with the other reported catalysts in literatures.

**Table S4** EIS parameters obtained by fitting the Nyquist plots to the equivalent circuitmodel in 0.1 M KOH at 1.5 V vs. RHE.

Catalysts	$R_{s}\left(\Omega ight)$	$R_{ct}\left(\Omega\right)$	$Q_1$ (F cm <sup>-2</sup> )	<b>n</b> <sub>1</sub>
S-Co <sub>2.5</sub> Fe <sub>7.5</sub>	2.662	12.59	0.0101	0.918
S-Co <sub>5</sub> Fe <sub>5</sub>	1.104	1.290	0.1078	0.785
S-Co <sub>7.5</sub> Fe <sub>2.5</sub>	1.820	8.887	0.0229	0.957

		-	-		
		OER	ORR	$\Delta E = E_{10} - E_{1/2}$	ЪĆ
Catalysts	Electrolyte	$E_{10}(V)$	E <sub>1/2</sub> (V)	(V)	Reference
S-Co <sub>2.5</sub> Fe <sub>7.5</sub>	0.1 M KOH	1.62	0.62	1.00	This work
S-Co <sub>5</sub> Fe <sub>5</sub>	0.1 M KOH	1.53	0.79	0.74	This work
S-Co <sub>7.5</sub> Fe <sub>2.5</sub>	0.1 M KOH	1.56	0.72	0.84	This work
RuO <sub>2</sub>	0.1 M KOH	1.60	0.31	1.29	This work
Pt/C	0.1 M KOH	1.73	0.87	0.86	This work
FeCo/Co <sub>2</sub> P@NPCF	0.1 M KOH	1.56	0.79	0.77	9
NiCo <sub>2</sub> O <sub>4</sub>	0.1 M KOH	1.64	0.77	0.87	10
CoDNG900	0.1 M KOH	1.613	0.864	$\sim$ 0.75	11
NiO/NiCo <sub>2</sub> O <sub>4</sub>	0.1 M KOH	1.587	0.73	0.857	2
Co@Co <sub>3</sub> O <sub>4</sub> /NC-1	0.1 M KOH	1.65	0.80	0.85	12
Co(OH) <sub>2</sub> +N-rGO	0.1 M KOH	1.66	0.79	0.87	13
CoO@Co/N-rGO	0.1 M KOH	1.64	0.73	0.91	14
NiFe-LDH/Fe-N-C (1:1)	0.1 M KOH	1.515	0.728	0.787	15
Fe@N-C-700	0.1 M KOH	1.71	0.83	0.88	16
$Co_3O_4/2.7Co_2MnO_4$	0.1 M KOH	1.77	0.68	1.09	17
Co <sub>3</sub> FeS <sub>1.5</sub> (OH) <sub>6</sub>	0.1 M KOH	1.588	0.721	0.867	18
CuS/NiS <sub>2</sub>	0.1 M KOH	1.52	0.73	0.79	19
FeCo-Co <sub>4</sub> N/N-C	0.1 M KOH	1.51	0.76	0.75	20
PPy/FeTCPP/Co	0.1 M KOH	1.61	0.86	0.75	21
Ni/NiO/NiCo <sub>2</sub> O <sub>4</sub> /		1.60	0.74	0.96	22
N-CNT-As	0.1 M KOH	1.00	0./4	0.80	22

**Table S5** Comparison of OER/ORR bi-functional activities for the as-preparedcatalysts in this study with the other reported catalysts in literatures.

	1	5		
Catalysts	$\eta_{onset}$	$\eta_{10}$	Tafel slope	Deference
	(mV)	(mV)	(mV dec <sup>-1</sup> )	Kelefence
S-Co <sub>2.5</sub> Fe <sub>7.5</sub>	160	280	132	This work
S-Co <sub>5</sub> Fe <sub>5</sub>	57	161	109	This work
S-Co <sub>7.5</sub> Fe <sub>2.5</sub>	108	238	122	This work
Pt/C	16	106	44	This work
PPy/FeTCPP/Co		240	83	21
FeNi/NPC		260	112	23
Co@Co-N-C	78	314	59	24
SHG	230	310	112	25

**Table S6** Comparison of HER performance in 0.1 M KOH for the as-prepared catalysts in this study with the other reported catalysts in literatures.

**Table S7** EIS parameters obtained by fitting the Nyquist plots to the equivalent circuitmodel in 0.1 M KOH at -0.1 V vs. RHE.

Catalysts	$R_{s}\left(\Omega ight)$	$R_{ct}\left(\Omega ight)$	$Q_1$ (F cm <sup>-2</sup> )	<b>n</b> <sub>1</sub>
S-Co <sub>2.5</sub> Fe <sub>7.5</sub>	8.250	31.350	0.0307	0.714
S-Co <sub>5</sub> Fe <sub>5</sub>	2.451	15.970	0.0337	0.622
S-Co <sub>7.5</sub> Fe <sub>2.5</sub>	5.253	31.560	0.0220	0.766

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Catalysts	Flectrolyte	Cell voltage	Pafaranca	
Catalysis	Electrolyte	E <sub>10</sub> (V)		
S-C0 <sub>2.5</sub> Fe <sub>7.5</sub> "S-C0 <sub>2.5</sub> Fe <sub>7.5</sub>	0.1 M KOH	1.84	This work	
S-Co <sub>5</sub> Fe <sub>5</sub> "S-Co <sub>5</sub> Fe <sub>5</sub>	0.1 M KOH	1.62	This work	
S-C07.5Fe2.5    S-C07.5Fe2.5	0.1 M KOH	1.76	This work	
RuO <sub>2 II</sub> Pt/C	0.1 M KOH	1.69	This work	
CoFe@NC/NCHNSs-700		1 ( ( 5	26	
CoFe@NC/NCHNSs-700	ТМКОН	1.005	20	
CoFe@N-GCNCs-700		1 (2	27	
CoFe@N-GCNCs-700	ТМКОН	1.63	27	
CoFe-N-CNTs/CNFs-900		1.66	20	
CoFe-N-CNTs/CNFs-900	Т М КОН	1.66	28	
$Ni_3S_2 \parallel Ni_3S_2$	1 M KOH	1.63	29	
Co <sub>0.85</sub> Se/NF <sup>11</sup> <sub>11</sub> Co <sub>0.85</sub> Se/NF	1 M KOH	1.63	30	

**Table S8** Comparison of HER/OER bi-functional activities for the as-preparedcatalysts in this study with the other reported catalysts in literatures.

ZaAD	OCD(M)	Peak power density	Defenence
ZIIAD	OCP(V)	(mW cm <sup>-2</sup> )	Kelerence
Zn plate    S-Co <sub>5</sub> Fe <sub>5</sub>	1.46	179	This work
Zn plate    CoFe/N-HCSs	1.387	96.5	31
Zn plate    CoFe/FeNC		154.1	32
Zn plate    CoFe@NO-CNT	1.45	142	33
Zn plate    NPSC-Co <sub>2</sub> Fe <sub>1</sub>	1.44	174.6	34
Zn plate    CoFe@N-CNWF	1.46	90	35
Zn plate    CoO <sub>x</sub> @NOC	1.44	141.65	36
Zn plate    Co-MOF-800	1.38	144	37
Zn plate    NCFPO-350	1.36	74.6	38
Zn plate    AlFeCoNiMn	1.44	136	39
Zn plate $_{II}^{II}$ CuCo <sub>2</sub> S <sub>4</sub>	1.38	123.9	40

**Table S9** Comparison of ZnAB performance using  $S-Co_5Fe_5$  as cathode catalyst with the other reported catalysts in literatures.

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