Revealing Long-lived Electron-hole Migration in Core-Shell α/γ -Fe₂O₃/FCP for Efficient Photoelectrochemical Water Oxidation

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Fig. S2: XPS survey spectra of γ -Fe₂O₃/ α -Fe₂O₃/FCP and γ -Fe₂O₃/ α -Fe₂O₃.¹



Fig. S3: Raman spectra of γ -Fe₂O₃/ α -Fe₂O₃ and α -Fe₂O₃.



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Fig. S5. Ti-Fe₂O₃, 75Ti-Fe₂O₃/75Ti-Fe₂O₃, 175Ti-Fe₂O₃/75Ti-Fe₂O₃, and 225Ti-Fe₂O₃/75Ti-Fe₂O₃: (a) Current density-potential curves; (b) ABPE spectra; (c) IPCE plots.



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Fig. S7. LSV curves of α -Fe₂O₃ and phase junction measured without AM 1.5G irradiation.



Fig. S8. LSV plots of γ -Fe₂O₃/ α -Fe₂O₃/FCP and γ -Fe₂O₃/ α -Fe₂O₃/CoPi in KOH electrolyte.



Fig. S9. (a) UV-vis absorption spectra of α -Fe₂O₃ and phase junction. (b) Tauc's plots of α -Fe₂O₃ and phase junction.



Fig. S10: Work Function measurement of pure α -Fe₂O₃ and FCP.



Fig. S11: Equivalent circuit diagram and the fitted Nyquist plots for α -Fe₂O₃ and phase junction.

Samples	$\mathbf{R}_{1}\left(\Omega ight)$	$\mathbf{R}_{2}\left(\Omega ight)$	$R_3(\Omega)$
α-Fe ₂ O ₃	45.12	461.9	2036
α-Fe ₂ O ₃ /FCP	61.74	129.9	1096
γ -Fe ₂ O ₃ / α -Fe ₂ O ₃	46.62	111.9	564.8
γ -Fe ₂ O ₃ / α -Fe ₂ O ₃ /FCP	51.93	83.8	454.9

Table S1. The fitted results of Nyquist plots for α -Fe₂O₃ and phase junction. Here, we named FeCo Prussian Blue as FCP.

In this work, the surface photovoltage (SPV) spectroscopic measurement was performed based on the system consisted of a lock-in amplifier (SR830-D SP) with a light chopper (SR540), a computer, a sample cell together with a source of monochromatic light provided by a 500 W xenon lamp (CHF-XM-500 W, Global Xenon Lamp Power) coupled with a grating monochromator (Omni-5007, Zolix). A low chopping frequency of 24 Hz was used. Before measurement, the system was calibrated by a DSI200 UV enhanced silicon detector to eliminate the possible phase shift which was not correlated to the SPV response, so that any phase retardation reflected the kinetics of SPV response. The transient photovoltage (TPV) measurement was taken under a laser radiation pulse with the wavelength of 355 nm and pulse width of 5 ns from a third-harmonic Nd:YAG laser (Polaris II, New Wave128 Research, Inc.). The TPV signals were recorded by a 500 MHz digital phosphor oscilloscope (TDS 5054, Tektronix). Both SPV and TPV measurements are conducted in air at room temperature.

	\mathbf{A}_{1}	τ ₁ (ps)	A_2	τ ₂ (ps)
α-Fe ₂ O ₃	$\boldsymbol{0.81 \pm 0.020}$	$\boldsymbol{0.81 \pm 0.019}$	$\boldsymbol{0.22\pm0.002}$	21.00 ± 0.58
α-Fe ₂ O ₃ /FCP	0.90 ± 0.011	0.80±0.010	$\textbf{0.24} \pm \textbf{0.002}$	29.59 ± 0.59
γ -Fe ₂ O ₃ / α -Fe ₂ O ₃	0.66 ± 0.009	0.94 ± 0.014	$\boldsymbol{0.18\pm0.001}$	33.45 ± 0.50
γ- Fe ₂ O ₃ / α- Fe ₂ O ₃ / FCP	0.56 ± 0.009	1.31 ± 0.029	$\boldsymbol{0.18\pm0.004}$	50.64 ± 0.78

Table S2. Kinetic parameters of TAS decays of α -Fe₂O₃ and phase junction under 400 nm excitation with time profiles of absorption probed at 580 nm.³ Here, we named FeCo Prussian Blue as FCP.



Fig. S12. Experimental decay kinetics fitted under 485 nm: a) γ-Fe₂O₃/α-Fe₂O₃/FCP,
b) γ-Fe₂O₃/α-Fe₂O₃, c) α-Fe₂O₃/FCP, d) α-Fe₂O₃.



Fig. S13. Pseudocolor TA plots of α -Fe₂O₃-based photoanodes under femtosecond laser pulses excitation with ~0.8 mJ fluence at 1 kHz repetition rate: a) γ -Fe₂O₃/ α -Fe₂O₃/FCP, b) γ -Fe₂O₃/ α -Fe₂O₃, c) α -Fe₂O₃/FCP, d) α -Fe₂O₃.

Reference

- 1. S.-S. Yi, B.-R. Wulan, J.-M. Yan and Q. Jiang, *Adv. Funct. Mater.*, 2019, **29**, 1801902.
- F. S. Hegner, I. Herraiz-Cardona, D. Cardenas-Morcoso, N. Lopez, J. R. Galan-Mascaros and S. Gimenez, ACS Appl. Mater. Interfaces, 2017, 9, 37671-37681.
- 3. F. Liu, R. Shi, Z. Wang, Y. Weng, C. M. Che and Y. Chen, *Angew. Chem. Int. Ed.*, 2019, **58**, 11791-11795.