

## Supporting Information (SI)

### **Efficient electrocatalytic CO<sub>2</sub> conversion into Formate by Al<sub>x</sub>Bi<sub>y</sub>O<sub>z</sub> nanorods in a wide potential window**

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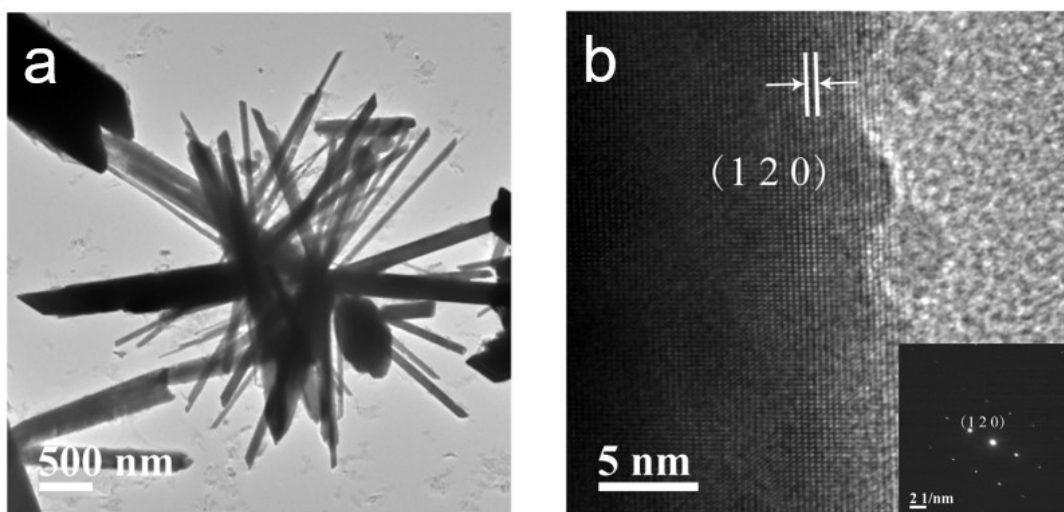
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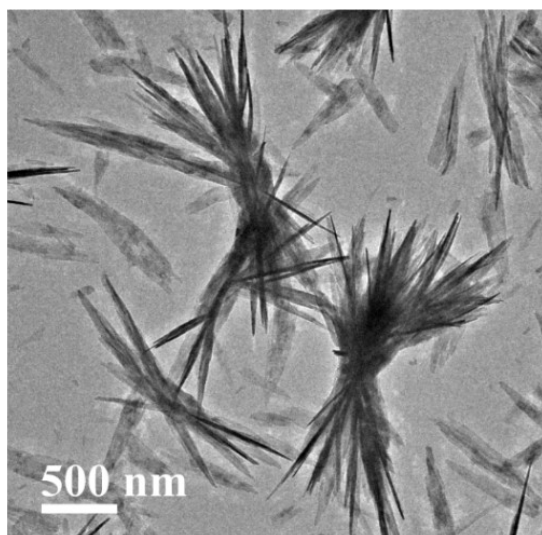
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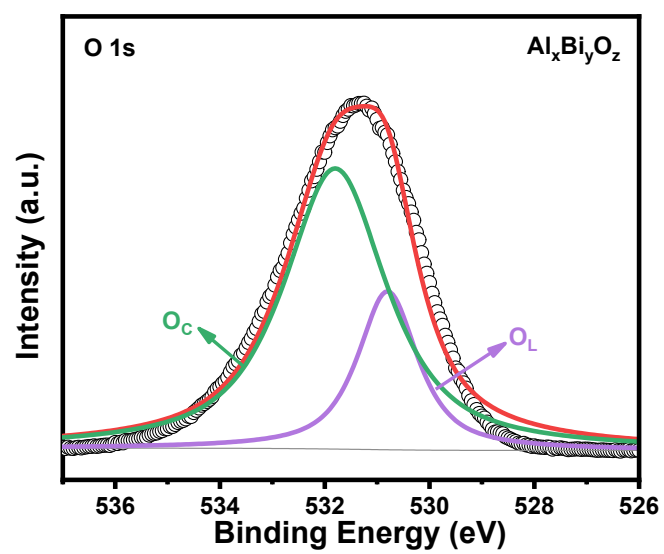
**Supplementary Figures:**



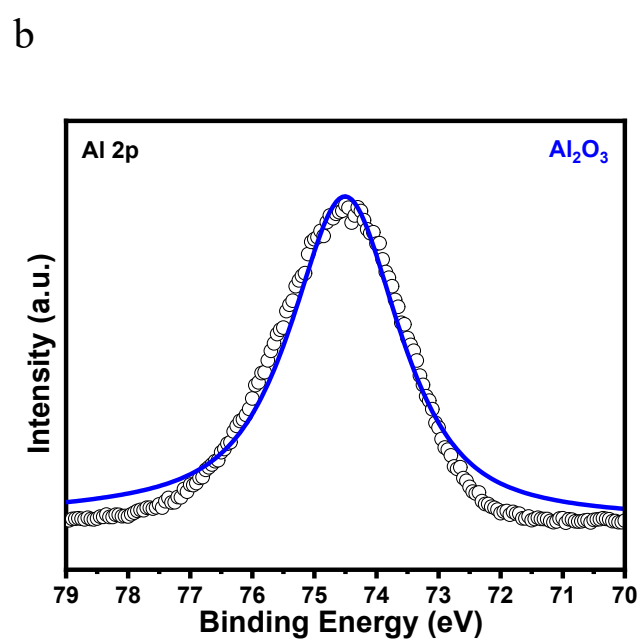
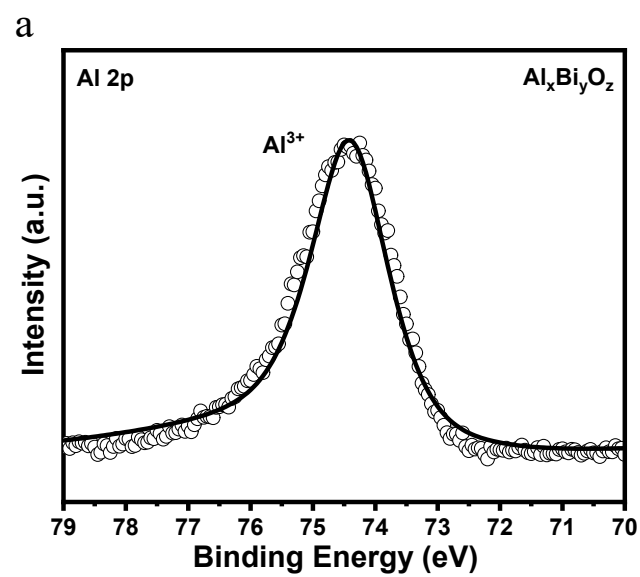
**Fig. S1** (a) Typical TEM image, (b) HRTEM image and SAED pattern (inset) of R-Bi<sub>2</sub>O<sub>3</sub>.



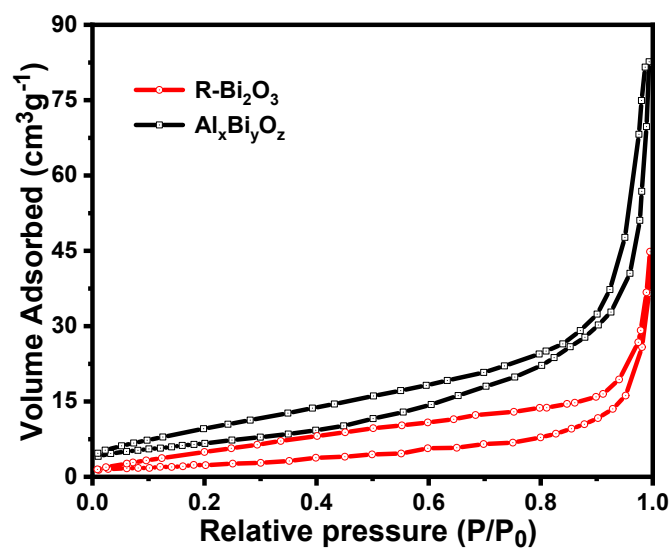
**Fig. S2** Typical TEM image of Al<sub>2</sub>O<sub>3</sub>.



**Fig. S3** High-resolution O 1s XPS spectrum of Al<sub>x</sub>Bi<sub>y</sub>O<sub>z</sub> nanorods.



**Fig. S4** High-resolution Al 2p XPS spectra of (a)  $\text{Al}_x\text{Bi}_y\text{O}_z$  nanorods and (b)  $\text{Al}_2\text{O}_3$ .



**Fig. S5** Nitrogen adsorption-desorption isotherms of Al<sub>x</sub>Bi<sub>y</sub>O<sub>z</sub> and R-Bi<sub>2</sub>O<sub>3</sub>.

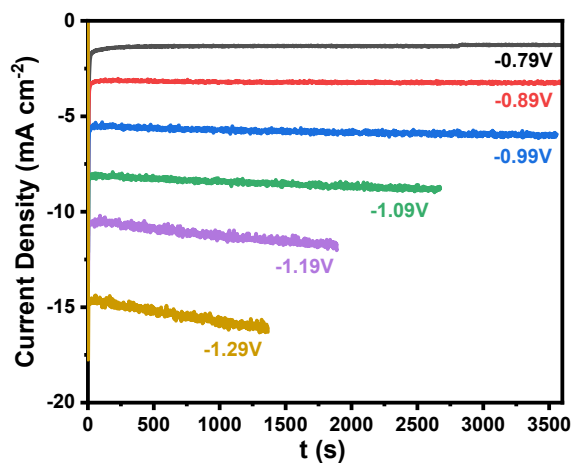


Fig. S6 Current density at six fixed potentials of  $\text{Al}_x\text{Bi}_y\text{O}_z$  nanorods.

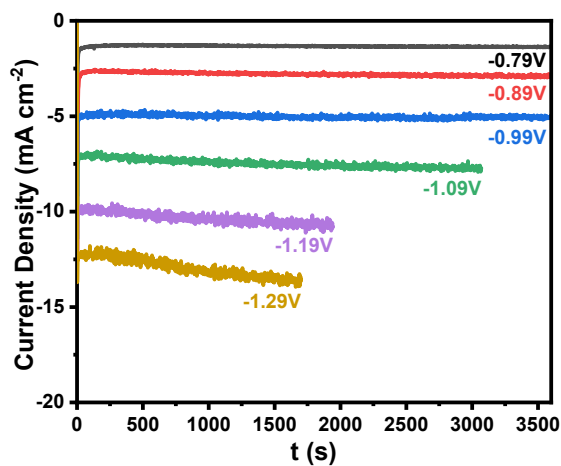


Fig. S7 Current density at six fixed potentials of R- $\text{Bi}_2\text{O}_3$ .

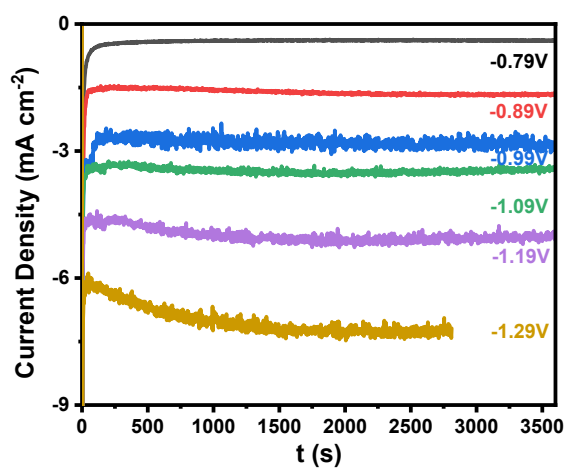
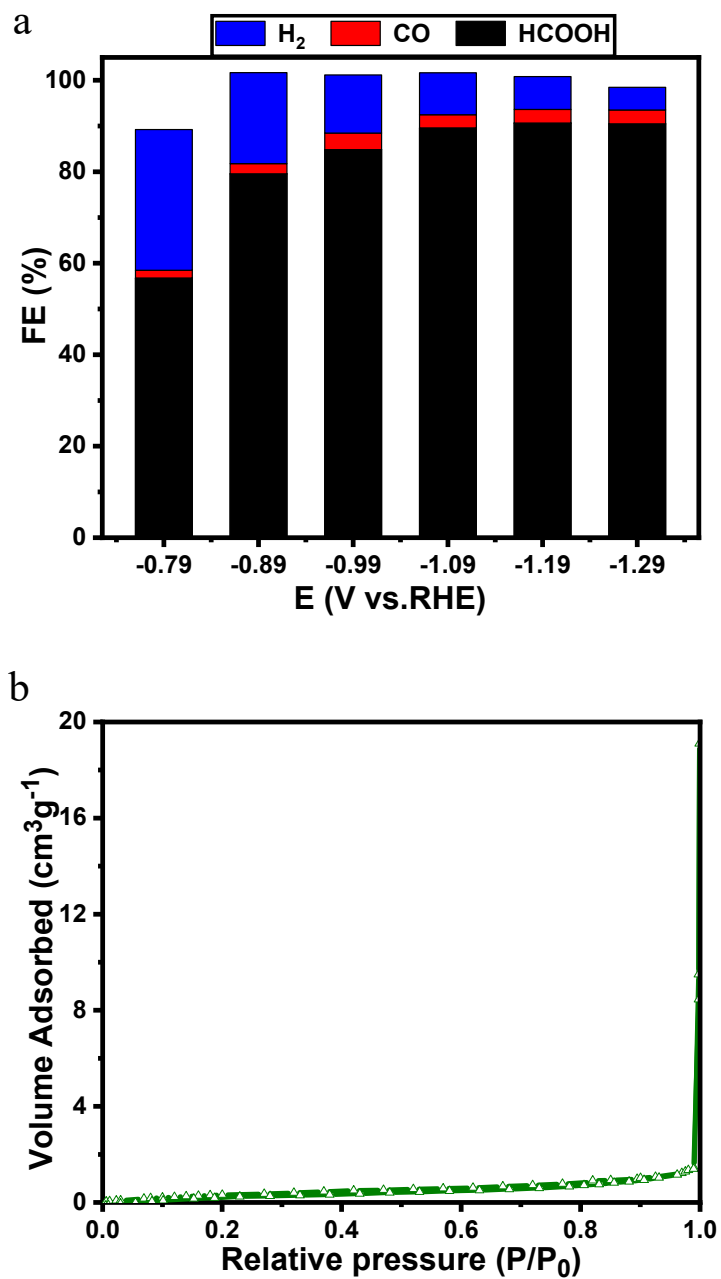
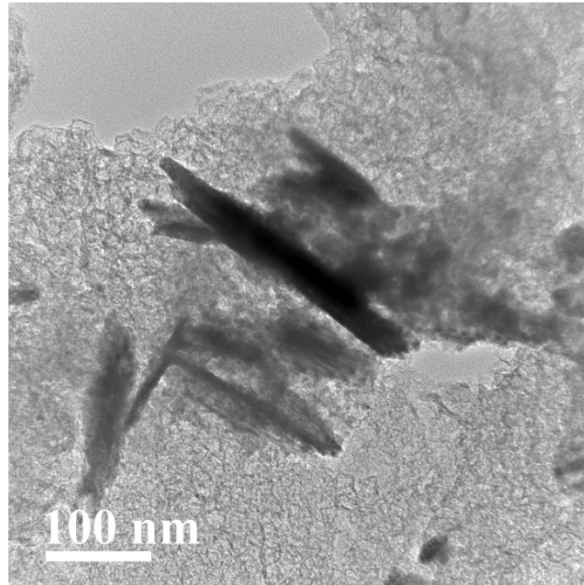


Fig. S8 Current density at six fixed potentials of  $\text{Al}_2\text{O}_3$ .



**Fig. S9** (a) Faradaic efficiencies at different potentials, (b) Nitrogen adsorption-desorption isotherm of commercial Bi<sub>2</sub>O<sub>3</sub>.



**Fig. S10** TEM image of  $\text{Al}_x\text{Bi}_y\text{O}_z$  after chronoamperometric electrolysis for 10 h.



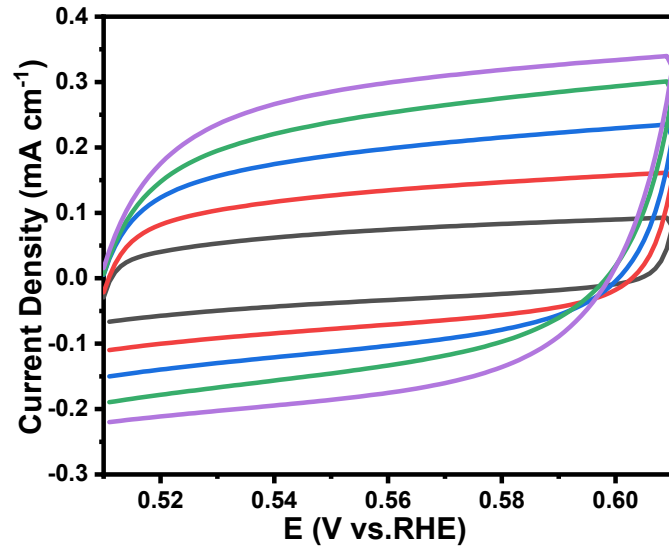


Fig. S11 CV curves at different scan rates of  $\text{Al}_x\text{Bi}_y\text{O}_z$ .

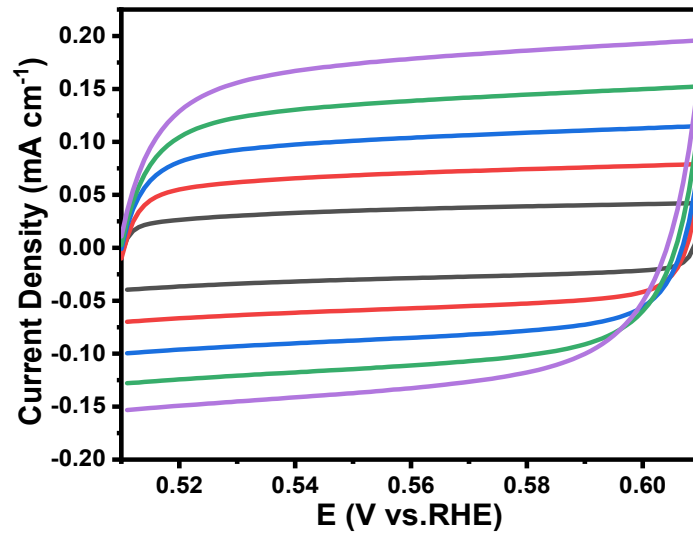


Fig. S12 CV curves at different scan rates of  $\text{R-Bi}_2\text{O}_3$ .

**Table. S1** Comparison of the performances between the reported Bi-based catalysts and this work.

Catalyst	Electrolyte	FE <sub>HCOO-</sub>	Formation rate	Ref
Bi(btbb) derived Bi/organic matrix	0.5M KHCO <sub>3</sub>	67% (-0.77V)	0.1 mmol·h <sup>-1</sup> cm <sup>-1</sup> (-0.97 V)	1
Bi-MOF (CAU-17)	0.1M KHCO <sub>3</sub>	85% (-0.8V)	/	2
Bi/Bi <sub>2</sub> O <sub>3</sub> nanosheets	0.5M KHCO <sub>3</sub>	90.4% (-0.87V)	1.12 mmol·mg <sup>-1</sup> h <sup>-1</sup> cm <sup>-2</sup> (-1.17V)	3
BiPO <sub>4</sub> derived Bi/Bi <sub>2</sub> O <sub>3</sub> nanosheets	0.5M KHCO <sub>3</sub>	~75% (-0.8V)	/	4
Bi nanosheets on Ag nanowire	0.5M KHCO <sub>3</sub>	~90% (-0.8V)	/	5
Bi nanosheets (flow cell)	0.5M KHCO <sub>3</sub>	91.6% (-0.8V)	/	6
Bi nanoparticles on carbon nanosheets	0.5M KHCO <sub>3</sub>	86% (-0.83V)	0.46 mmol·h <sup>-1</sup> cm <sup>-1</sup> (-0.97 V)	7
Bi nanorods@N-C nanotubes	0.1M KHCO <sub>3</sub>	~65% (-0.8V)	/	8
Al <sub>x</sub> Bi <sub>y</sub> O <sub>z</sub> nanorods	0.1M KHCO <sub>3</sub>	91.5(-0.79V)	0.1 mmol·h <sup>-1</sup> cm <sup>-1</sup> (-0.99 V)	this work

All potentials in this table are measured against the RHE reference.

## References

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