

## Electronic supplementary information (ESI)

### Determination of Acid Structures on the Surface of Sulfated Monoclinic and Tetragonal Zirconia through Experimental and Theoretical Approaches

Daofeng Huang,<sup>a</sup> Siyue Chen,<sup>b</sup> Sicong Ma,<sup>c</sup> Xin Chen,<sup>a</sup> Yuanhang Ren,<sup>a</sup> Meiyin Wang,<sup>a</sup> Lin Ye,<sup>a</sup> Li Zhang,<sup>a</sup> Xueying Chen,<sup>a</sup> Zhi-Pan Liu,<sup>b</sup> Bin Yue<sup>\*a</sup> and Heyong He<sup>\*a</sup>

<sup>a</sup> Department of Chemistry and Shanghai Key Laboratory of Molecular Catalysis and Innovative Materials, Fudan University, Shanghai 200438, P. R. China.

<sup>b</sup> Collaborative Innovation Center of Chemistry for Energy Material, Shanghai Key Laboratory of Molecular Catalysis and Innovative Materials, Key Laboratory of Computational Physical Science, Department of Chemistry, Fudan University, Shanghai 200438, China.

<sup>c</sup> Key Laboratory of Synthetic and Self-Assembly Chemistry for Organic Functional Molecules, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, Shanghai 200032, China.

\* Corresponding authors at: heyonghe@fudan.edu.cn (Heyong He); yuebin@fudan.edu.cn (Bin Yue)

**Table S1.** The acid models on SZ.

Proposed models	References	Proposed models	References
	Yamaguchi <sup>1</sup>		Ward <sup>2</sup>
	Arata <sup>3</sup>		Clearfield <sup>4, 5</sup>
	Lavallee <sup>6</sup>		Riemer <sup>7</sup>
	Morterra <sup>8</sup>		Kustov <sup>9</sup>

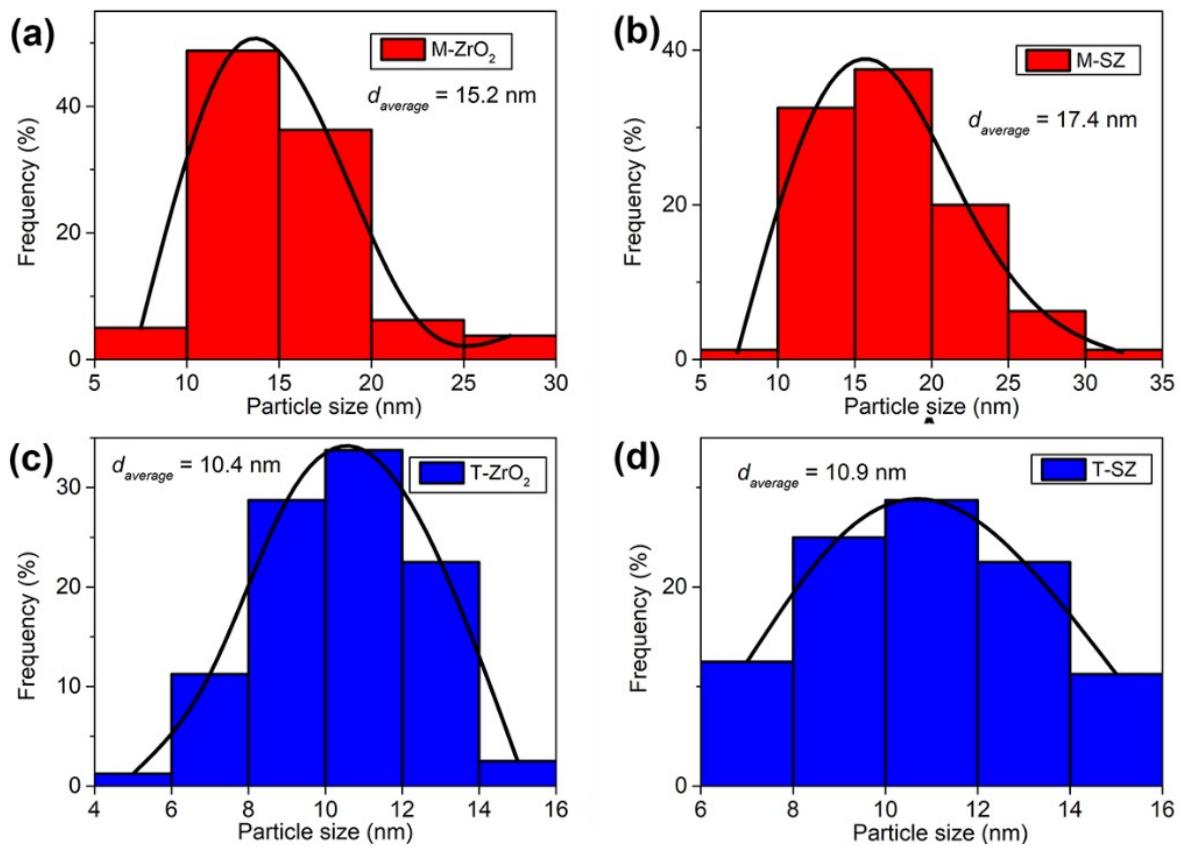
**Table S2.** The  $\Delta E_{\text{NH}_3}$  on different structures.

Samples	$\Delta E_{\text{NH}_3}$ (kJ·mol <sup>-1</sup> )		Origin
	Zr-(OH)-Zr	Zr-OH	
S-M(111)	-97, -74		Figure 3a
S-M(-111)	-117, -88		Figure 3b
S-M(001)	-73, -66, -66		Figure 3c
S-M(011)	-67		Figure 3d
S-T(101)	-79, -65		Figure 4a
S-T(001)	-71, -55		Figure 4b
S-T(100)	-76, -65		Figure 4c
S-T(111)-1	-61	-58	Figure 4d
S-T(111)-2	-76		Figure 4e
2S-T(111)	— <sup>a</sup>		Figure 4f

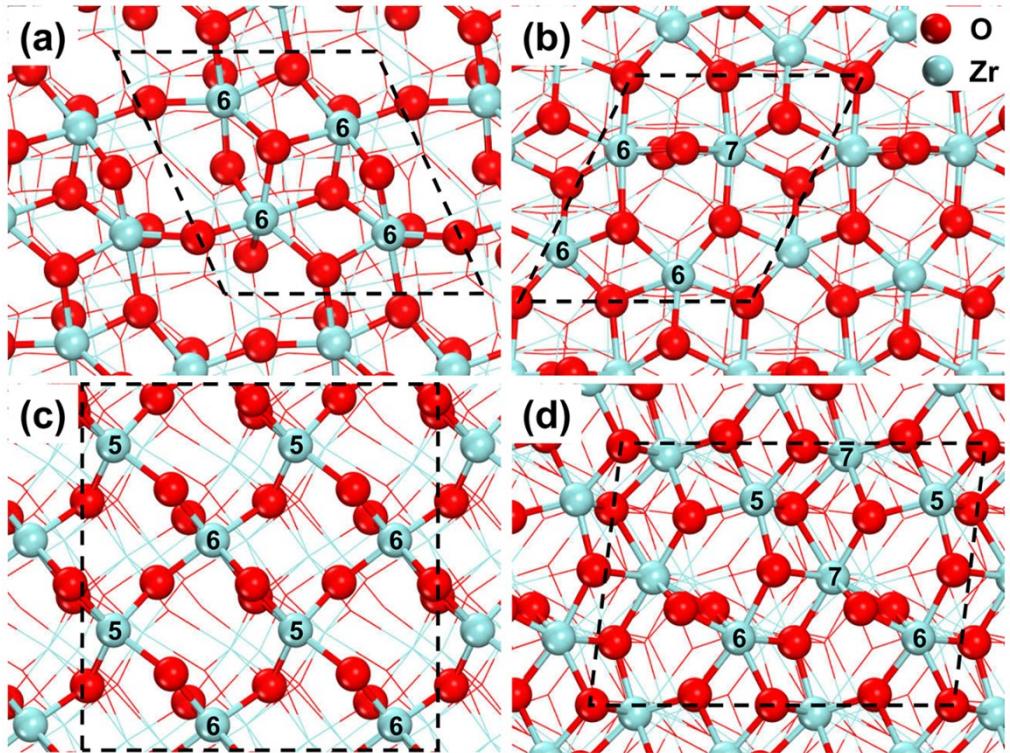
<sup>a</sup> The  $\Delta E_{\text{NH}_3}$  on the hydroxyl groups on 2S-T(111) is out of detection because of the nearby highly acidic protons on Zr-(OH)-SO<sub>2</sub>.

**Table S3.** The unit parameters of the structures on different planes.

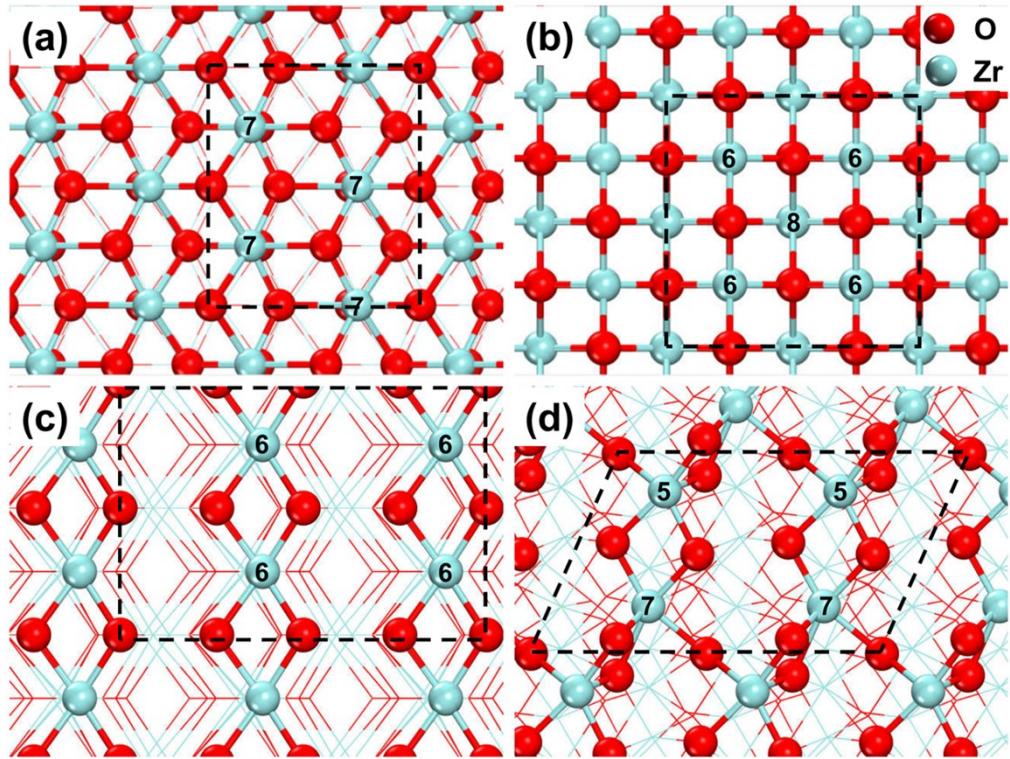
Structural models	a (Å)	b (Å)	Angle (degree)	Unit area per H <sub>2</sub> SO <sub>4</sub> (Å <sup>2</sup> /H <sub>2</sub> SO <sub>4</sub> )
S-M(111)	7.39	7.51	114.6	50.5
S-M(-111)	6.80	7.39	63.73	45.1
S-M(001)	10.37	10.53	90	109.2
S-M(011)	10.37	7.51	83.12	77.3
S-T(101)	6.37	7.26	90	46.2
S-T(001)	7.25	7.25	90	52.6
S-T(100)	10.48	7.26	90	76.1
S-T(111)-1	10.26	6.37	66.28	59.8
S-T(111)-2	10.26	6.37	66.28	59.8
2S-T(111)	10.26	6.37	66.28	29.9



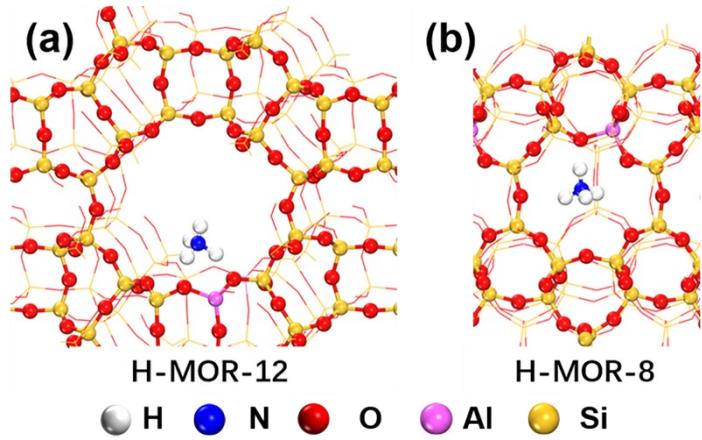
**Figure S1.** The particle size distributions of (a) M-ZrO<sub>2</sub>, (b) M-SZ, (c) T-ZrO<sub>2</sub> and (d) T-SZ.



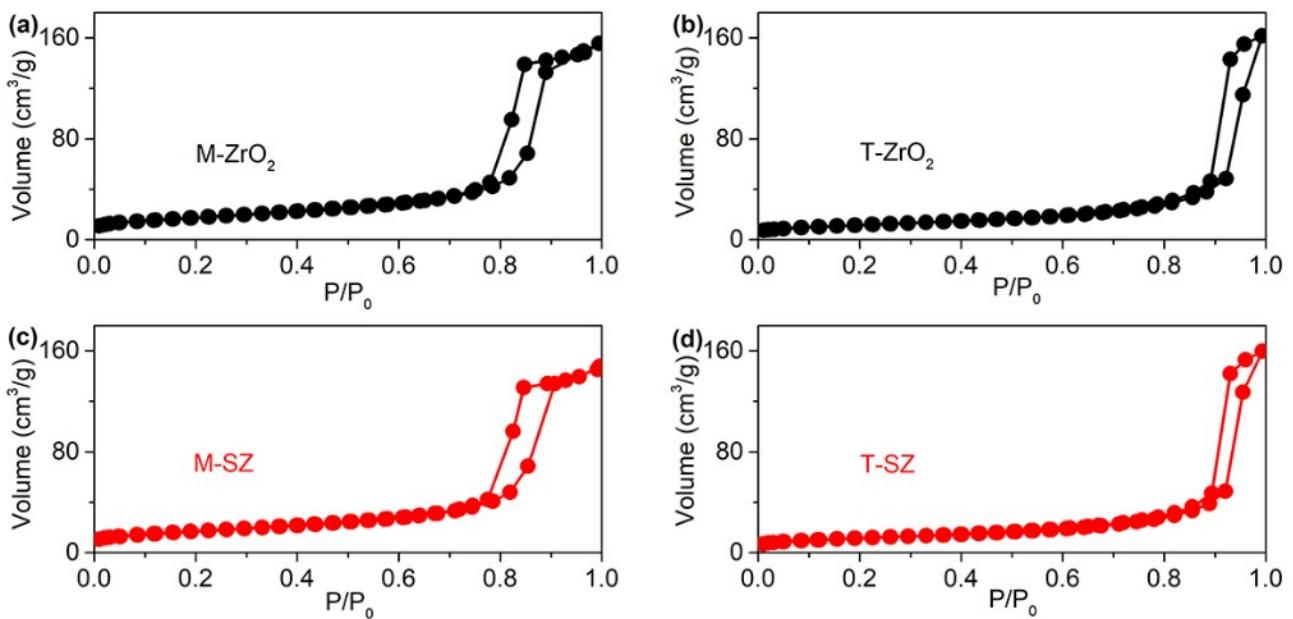
**Figure S2.** The clean planes of monoclinic zirconia. (a) M(111), (b) M(-111), (c) M(001) and (d) M(011). The numbers on zirconium cations represent their coordination number. The area within the dotted lines corresponds to the unit in Table S3.



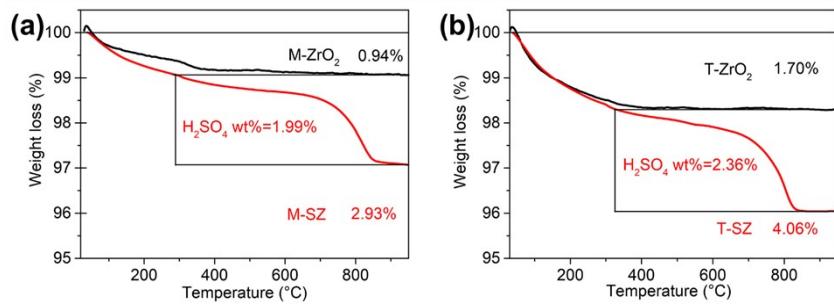
**Figure S3.** The clean planes of tetragonal zirconia. (a) T(101) (b) T(001), (c) T(100) and (d) T(111). The numbers on the zirconium cations represent their coordination number. The area within the dotted lines corresponds to the unit in Table S3.



**Figure S4.** The structures of  $\text{NH}_3$  adsorbed on (a) H-MOR-12 and (b) H-MOR-8.



**Figure S5.**  $\text{N}_2$  adsorption-desorption isotherms of (a) M-ZrO<sub>2</sub>, (b) T-ZrO<sub>2</sub>, (c) M-SZ and (d) T-SZ.



**Figure S6.** The TG curves of (a) M-ZrO<sub>2</sub> and M-SZ and (b) T-ZrO<sub>2</sub> and T-SZ.

The third step of the weight loss could be corresponded to the chemical dissociation of SO<sub>3</sub> to SO<sub>2</sub> and O<sub>2</sub> species on zirconia.<sup>10</sup> The sulfur content in form of sulfuric acid may be calculated by

$$\text{H}_2\text{SO}_4 \text{ wt\%} = \text{SO}_3 \text{ wt\%} \times \frac{98}{80}$$

as 1.99 wt% for M-SZ and 2.36% for T-SZ.

### The calculation of the adsorbed sulfuric acid molecules on M-SZ and T-SZ. (Section 3.5.1)

$$S_{M/T(hkl)} = D(H_2SO_4) \times N_A \times U_{S-M/T(hkl)} \quad (1)$$

According to the equation (1), for M-SZ, if the strong BA only relates to S-M(001)

$$S_{M(001)} = D(H_2SO_4) \times N_A \times U_{S-M(001)} = 23.9 \text{ } \mu\text{mol}\cdot\text{g}^{-1} \times 6.022 \times 10^{23} \text{ mol}^{-1} \times 109.2 \text{ } \text{\AA}^2 = 15.7 \text{ m}^2\cdot\text{g}^{-1}.$$

If the strong BA only relates to S-M(011),

$$S_{M(011)} = D(H_2SO_4) \times N_A \times U_{S-M(011)} = 23.9 \text{ } \mu\text{mol}\cdot\text{g}^{-1} \times 6.022 \times 10^{23} \text{ mol}^{-1} \times 77.3 \text{ } \text{\AA}^2 = 11.1 \text{ m}^2\cdot\text{g}^{-1}.$$

So, the total specific surface area of M(001) and M(011) is in the range of  $11.1 \sim 15.7 \text{ m}^2\cdot\text{g}^{-1}$  and the specific surface area of the rest crystal planes is  $44.5 \sim 49.1 \text{ m}^2\cdot\text{g}^{-1}$ .

If M(111) represents the rest planes, the sulfuric acid density on M(111) is calculated as

$$\begin{aligned} D(H_2SO_4) &= S_{M(111)} / (N_A \times U_{S-M(111)}) = (44.5 \sim 49.1) \text{ m}^2\cdot\text{g}^{-1} / (6.022 \times 10^{23} \text{ mol}^{-1} \times 50.5 \text{ } \text{\AA}^2) \\ &= 146 \sim 161 \text{ } \mu\text{mol}\cdot\text{g}^{-1}. \end{aligned}$$

If M(-111) represents the rest planes, the sulfuric acid density on M(-111) is calculated as

$$\begin{aligned} D(H_2SO_4) &= S_{M(-111)} / (N_A \times U_{S-M(-111)}) = (44.5 \sim 49.1) \text{ m}^2\cdot\text{g}^{-1} / (6.022 \times 10^{23} \text{ mol}^{-1} \times 45.1 \text{ } \text{\AA}^2) \\ &= 164 \sim 181 \text{ } \mu\text{mol}\cdot\text{g}^{-1}. \end{aligned}$$

As a result, the sulfuric acid density on M(111) and M(-111) is in the range of  $146 \sim 181 \text{ } \mu\text{mol}\cdot\text{g}^{-1}$ . The percentage of the sulfuric acid molecules related to these strong BA sites is  $23.9 / (23.9 + 181) \times 100\% \sim 23.9 / (23.9 + 146) \times 100\% = 12 \sim 14\%$ .

For T-SZ, the specific area for T(111) is calculated by equation (1) as

$$\begin{aligned} S_{T(111)} &= D(H_2SO_4) \times N_A \times U_{S-T(111)-1,2} = 29.2 \text{ } \mu\text{mol}\cdot\text{g}^{-1} \times 6.022 \times 10^{23} \text{ mol}^{-1} \times 59.8 \text{ } \text{\AA}^2 = 10.5 \text{ m}^2\cdot\text{g}^{-1}, \text{ and} \\ S_{T(111)} &= D(H_2SO_4) \times N_A \times U_{2S-T(111)} = 29.2 \text{ } \mu\text{mol}\cdot\text{g}^{-1} \times 6.022 \times 10^{23} \text{ mol}^{-1} \times 29.9 \text{ } \text{\AA}^2 = 5.3 \text{ m}^2\cdot\text{g}^{-1}. \end{aligned}$$

So, the specific area of T(111) is in the range of  $5.3 \sim 10.5 \text{ m}^2\cdot\text{g}^{-1}$  and the total area of the rest planes is  $29.9 \sim 35.1 \text{ m}^2\cdot\text{g}^{-1}$ .

If T(101) represents the rest planes, the sulfuric acid density on T(101) is calculated as

$$\begin{aligned} D(H_2SO_4) &= S_{T(101)} / (N_A \times U_{S-T(101)}) = (29.9 \sim 35.1) \text{ m}^2\cdot\text{g}^{-1} / (6.022 \times 10^{23} \text{ mol}^{-1} \times 46.2 \text{ } \text{\AA}^2) \\ &= 107 \sim 126 \text{ } \mu\text{mol}\cdot\text{g}^{-1} \end{aligned}$$

If T(001) represents the rest planes, the sulfuric acid density on T(001) is calculated as

$$\begin{aligned} D(H_2SO_4) &= S_{T(001)} / (N_A \times U_{S-T(001)}) = (29.9 \sim 35.1) \text{ m}^2\cdot\text{g}^{-1} / (6.022 \times 10^{23} \text{ mol}^{-1} \times 52.6 \text{ } \text{\AA}^2) \\ &= 94.4 \sim 111 \text{ } \mu\text{mol}\cdot\text{g}^{-1} \end{aligned}$$

If T(100) represents the rest planes, the sulfuric acid density is calculated as

$$\begin{aligned} D(H_2SO_4) &= S_{T(100)} / (N_A \times U_{S-T(100)}) = (29.9 \sim 35.1) \text{ m}^2\cdot\text{g}^{-1} / (6.022 \times 10^{23} \text{ mol}^{-1} \times 76.1 \text{ } \text{\AA}^2) \\ &= 65.2 \sim 76.6 \text{ } \mu\text{mol}\cdot\text{g}^{-1} \end{aligned}$$

So, the sulfuric acid density on the rest planes is in the range of  $65.2 \sim 126 \text{ } \mu\text{mol}\cdot\text{g}^{-1}$ . The percentage of the sulfuric acid molecules related to the strong BA is  $29.2 / (29.2 + 126) \times 100\% \sim 29.2 / (29.2 + 65.2) \times 100\% = 19 \sim 31\%$

### The density ratio of the newly generated LA to strong BA sites on T-SZ (3.5.2)

$$D(LA) = S_{0-T} \times \text{Area percentage} \times X / (N_A \times U_{S-T(hkl)}) \quad (2)$$

According to the equation (2) in Section 3.5.2, for S-T(101),  $X = 0$ ,  $D(LA) = 0 \mu\text{mol}\cdot\text{g}^{-1}$ .

For S-T(001),

$$D(LA) = 40.4 \text{ m}^2\cdot\text{g}^{-1} \times 4\% \times 4 / (6.022 \times 10^{23} \text{ mol}^{-1} \times 52.6 \text{ \AA}^2) = 20.4 \mu\text{mol}\cdot\text{g}^{-1}$$

For S-T(100),

$$D(LA) = 40.4 \text{ m}^2\cdot\text{g}^{-1} \times 18\% \times 2 / (6.022 \times 10^{23} \text{ mol}^{-1} \times 76.1 \text{ \AA}^2) = 31.7 \mu\text{mol}\cdot\text{g}^{-1}$$

For structures on T(111), when S-T(111)-1 is the only structure on T(111),

$$D(LA) = 0 \mu\text{mol}\cdot\text{g}^{-1} \text{ and the total LA density on T-SZ } (D(LA)_{\text{total}}) \text{ is } 20.4 + 31.7 = 52.1 \mu\text{mol}\cdot\text{g}^{-1}.$$

When S-T(111)-2 is the only structure on T(111),

$$D(LA) = 40.4 \text{ m}^2\cdot\text{g}^{-1} \times 17\% \times 2 / (6.022 \times 10^{23} \text{ mol}^{-1} \times 59.8 \text{ \AA}^2) = 38.1 \mu\text{mol}\cdot\text{g}^{-1} \text{ and } D(LA)_{\text{total}} = 20.4 + 31.7 + 38.1 = 90.2 \mu\text{mol}\cdot\text{g}^{-1}.$$

When 2S-T(101) represents the structure on T(111),

$$D(LA) = 0 \mu\text{mol}\cdot\text{g}^{-1} \text{ and } D(LA)_{\text{total}} = 20.4 + 31.7 + 0 = 52.1 \mu\text{mol}\cdot\text{g}^{-1}$$

$$D(BA) = S_{0-T} \times \text{area percentage} / (N_A \times U_{S-T(hkl)}) \quad (3)$$

According to equation (3) in Section 3.5.2, only structures on T(111) generate strong BA.

When S-T(111)-1 is the only structure on T(111),

$$D(BA) = 40.4 \text{ m}^2\cdot\text{g}^{-1} \times 17\% / (6.022 \times 10^{23} \text{ mol}^{-1} \times 59.8 \text{ \AA}^2) = 19.1 \mu\text{mol}\cdot\text{g}^{-1} \text{ and } D(LA)_{\text{total}} :$$

$$D(BA) = 52.1 \mu\text{mol}\cdot\text{g}^{-1} : 19.1 \mu\text{mol}\cdot\text{g}^{-1} = 2.7:1.0.$$

When S-T(111)-2 is the only structure on T(111),

$$D(BA) = 40.4 \text{ m}^2\cdot\text{g}^{-1} \times 17\% / (6.022 \times 10^{23} \text{ mol}^{-1} \times 59.8 \text{ \AA}^2) = 19.1 \mu\text{mol}\cdot\text{g}^{-1} \text{ and } D(LA)_{\text{total}} : D(BA) = 90.2 \mu\text{mol}\cdot\text{g}^{-1} : 19.1 \mu\text{mol}\cdot\text{g}^{-1} = 4.7:1.0.$$

When 2S-T(101) represents the structure on T(111),

$$D(BA) = 40.4 \text{ m}^2\cdot\text{g}^{-1} \times 17\% / (6.022 \times 10^{23} \text{ mol}^{-1} \times 29.9 \text{ \AA}^2) = 38.1 \mu\text{mol}\cdot\text{g}^{-1} \text{ and } D(LA)_{\text{total}} : D(BA) = 52.1 \mu\text{mol}\cdot\text{g}^{-1} : 38.1 \mu\text{mol}\cdot\text{g}^{-1} = 1.4:1.0.$$

So the density ratio of the newly generated LA to strong BA sites is in the range of ca. 1.4:1.0 ~ 4.7:1.0.

**The calculation detail of the optimized XYZ position for typical acid structures.**

Optimized XYZ position for S-M(001)							
Lengths Å	a 10.3746	b 10.5298	c 20.3527	Angles degree	α 90	β 90	γ 90
Atom		x		y		z	
O		9.39879		1.02668		3.23194	
O		4.21154		6.29152		3.23187	
O		9.39912		6.29148		3.23207	
O		1.33819		3.58505		3.6553	
O		6.52507		3.58538		3.65535	
O		1.3381		8.85002		3.65533	
O		6.52516		8.85021		3.65537	
Zr		2.98234		2.33424		4.1716	
Zr		8.17007		2.33453		4.17143	
Zr		2.98226		7.59939		4.17161	
Zr		8.17025		7.59949		4.17136	
Zr		0.1567		5.03383		4.45089	
Zr		5.34452		5.03406		4.45094	
Zr		0.15651		10.29883		4.45081	
Zr		5.34445		10.29874		4.45096	
O		1.73887		0.97484		4.8953	
O		6.92553		0.97423		4.89541	
O		1.73879		6.23957		4.89535	
O		6.92543		6.23902		4.89543	
O		3.72545		3.92097		5.39537	
O		8.91268		3.92093		5.39541	
O		3.72544		9.18587		5.39539	
O		8.91267		9.18583		5.39543	
O		4.12464		1.38091		5.83204	
O		9.31196		1.38091		5.83218	
O		4.12466		6.64578		5.83205	
O		9.31196		6.64579		5.83221	
O		0.84573		4.38801		6.46166	
O		6.03294		4.38806		6.46154	
O		0.84571		9.65292		6.46167	
O		6.03296		9.65295		6.46158	
Zr		2.6019		0.22651		6.7658	
Zr		7.7892		0.22651		6.7658	
Zr		2.6019		5.49141		6.7658	
Zr		7.7892		5.49141		6.7658	
Zr		4.83292		2.85897		7.20298	
Zr		10.02022		2.85897		7.20298	
Zr		4.83292		8.12387		7.20298	
Zr		10.02022		8.12387		7.20298	
O		1.41002		1.77661		7.46637	
O		6.59732		1.77661		7.46637	
O		1.41002		7.04151		7.46637	
O		6.59732		7.04151		7.46637	

O	3.24022	3.98873	8.20343
O	8.42752	3.98873	8.20343
O	3.24022	9.25363	8.20343
O	8.42752	9.25363	8.20343
O	3.7451	1.27621	8.40356
O	8.9324	1.27621	8.40356
O	3.7451	6.54111	8.40356
O	8.9324	6.54111	8.40356
O	0.388	3.48832	9.14062
O	5.5753	3.48832	9.14062
O	0.388	8.75322	9.14062
O	5.5753	8.75322	9.14062
Zr	2.1524	2.40597	9.40401
Zr	7.3397	2.40597	9.40401
Zr	2.1524	7.67087	9.40401
Zr	7.3397	7.67087	9.40401
Zr	4.38342	5.03843	9.8412
Zr	9.57072	5.03843	9.8412
Zr	4.38342	10.30333	9.8412
Zr	9.57072	10.30333	9.8412
O	0.9335	0.90118	10.15415
O	6.11984	0.87926	10.13091
O	0.95571	6.13394	10.15096
O	6.14026	6.16117	10.20711
O	2.90131	3.90193	10.79453
O	8.02372	3.90418	10.78489
O	2.79514	9.17739	10.78818
O	8.01504	9.1413	10.78817
O	3.27428	1.36722	11.23329
O	8.43655	1.32677	11.16421
O	3.25049	6.63301	11.13217
O	8.49223	6.59704	11.0793
O	0.02083	4.32724	11.73014
O	5.27244	4.33847	11.74636
O	0.00862	9.58672	11.75775
O	5.17107	9.54771	11.74746
Zr	1.61129	0.22617	12.13668
Zr	6.79968	0.18527	12.12939
Zr	1.63016	5.53924	12.12179
Zr	7.01673	5.42677	12.17056
Zr	4.11418	2.92206	12.49707
Zr	9.21981	2.85157	12.39001
Zr	3.82511	8.20901	12.44233
Zr	9.08397	8.13455	12.47538
O	0.5088	1.7337	13.01444
O	5.62195	1.64452	12.96046
O	0.48364	6.98513	12.87963
O	5.63945	7.03175	12.90425
O	2.2787	3.72419	13.32362
O	7.94108	3.97943	13.38766

O	2.67821	9.53509	13.4038
O	7.93021	9.58901	13.36331
O	8.05428	6.6604	13.71282
H	8.53645	6.029	14.27748
H	6.06205	7.25863	13.76191
O	4.61127	4.46085	14.37798
O	2.9713	6.48508	13.88895
S	3.96543	5.47278	15.22444
O	4.84298	6.51256	15.69185
O	2.8669	5.04997	16.04366
H	2.23205	6.88139	14.40903
H	1.56204	3.01525	13.32476

Optimized XYZ position for S-M(011)

Lengths Å	a 10.3746	b 7.5079	c 28.575	Angles degree	$\alpha$ 90	$\beta$ 90	$\gamma$ 83.1229
Atom	x	y	z				
O	0.83105	6.40113	4.9427				
O	6.01813	6.40143	4.94296				
O	2.48057	1.71543	5.1065				
O	7.66793	1.71519	5.106				
Zr	3.82682	3.2554	5.50681				
Zr	9.01368	3.25544	5.50661				
O	3.33336	5.28935	5.26574				
O	8.52073	5.2894	5.26576				
Zr	4.34806	7.24231	5.59451				
Zr	9.53483	7.24247	5.59429				
O	4.70762	1.426	6.16596				
O	9.89485	1.42625	6.16591				
O	0.56997	3.75585	6.02391				
O	5.75728	3.7558	6.0239				
Zr	1.62817	5.50066	6.59217				
Zr	6.81542	5.50084	6.59213				
O	2.73507	7.36984	7.08388				
O	7.92245	7.36991	7.08416				
Zr	1.24775	1.96318	7.08875				
Zr	6.43532	1.96312	7.089				
O	2.82329	3.51747	7.22918				
O	8.01061	3.51749	7.2291				
O	10.28121	6.40457	7.38395				
O	5.09368	6.40459	7.38397				
O	10.41537	2.6227	8.68456				
O	5.22798	2.62274	8.68468				
O	2.3183	5.439	8.86132				
O	7.50562	5.439	8.8614				
Zr	3.87598	6.94746	8.99096				
Zr	9.06322	6.94761	8.99077				
O	2.36847	1.58747	9.01973				
O	7.55579	1.58752	9.01989				
Zr	3.45704	3.56223	9.30706				
Zr	8.64436	3.56223	9.30706				
O	4.64892	5.15305	9.9095				
O	9.83623	5.15305	9.9095				
O	4.77196	0.09338	9.93861				
O	9.95928	0.09338	9.93861				
Zr	0.77652	1.6842	10.54105				
Zr	5.96383	1.6842	10.54105				
O	1.86434	3.65202	10.81344				
O	7.05166	3.65202	10.81344				
Zr	1.22601	5.7311	10.86172				
Zr	6.41333	5.7311	10.86172				
O	2.81872	7.23729	10.9548				

O	8.00604	7.23729	10.9548
O	4.19942	2.63515	11.12118
O	9.38673	2.63515	11.12118
O	10.40878	6.33007	12.45384
O	5.22146	6.33007	12.45384
O	1.41484	1.72794	12.62023
O	6.60215	1.72794	12.62023
Zr	3.00754	3.23412	12.71332
Zr	8.19486	3.23412	12.71332
O	2.36922	5.31321	12.76159
O	7.55654	5.31321	12.76159
Zr	3.45704	7.28103	13.03398
Zr	8.64436	7.28103	13.03398
O	3.74991	1.41798	13.63642
O	8.93723	1.41798	13.63642
O	4.77196	3.81218	13.66553
O	9.95927	3.81218	13.66553
Zr	0.77651	5.403	14.26797
Zr	5.96383	5.403	14.26797
O	1.86775	7.36792	14.55463
O	7.03764	7.36347	14.53945
Zr	0.35004	2.00342	14.59033
Zr	5.52274	1.99868	14.55724
O	1.90018	3.51532	14.71573
O	7.10912	3.54195	14.70847
O	4.21471	6.37077	14.89965
O	9.3925	6.36722	14.8584
O	4.23784	2.47457	16.13941
O	9.4722	2.48787	16.22687
O	1.39233	5.43706	16.39878
O	6.57451	5.4326	16.35602
Zr	2.97459	6.9233	16.52177
Zr	8.16752	7.00599	16.45419
O	1.47419	1.56304	16.49377
O	6.67704	1.62318	16.40814
Zr	2.54703	3.50814	16.96682
Zr	7.71811	3.49324	16.80817
O	3.77093	5.13663	17.46098
O	8.89042	5.13119	17.3469
O	3.70414	0.02914	17.64908
O	8.94057	0.09117	17.50529
Zr	10.32136	1.64082	17.96536
Zr	4.9942	1.72844	17.90411
O	0.83265	3.62415	18.28312
O	6.13602	3.59375	18.22995
Zr	10.65743	5.68881	18.23447
Zr	5.65517	5.71521	18.16507
O	1.74698	7.2253	18.42902
O	6.98626	7.31856	18.23951
O	3.3816	2.62635	18.61093

O	8.5721	2.55716	18.76959
H	7.9645	1.92423	19.19987
H	7.24395	4.37118	20.14237
S	8.3402	6.1413	20.95061
O	9.55626	5.87399	20.10126
O	7.07336	5.33933	20.08677
O	7.93645	7.51868	20.91998
O	8.39805	5.42557	22.18976

Optimized XYZ position for S-T(111)-1

Lengths Å	a 10.2592	b 6.3733	c 27.6706	Angles degree	$\alpha$ 90	$\beta$ 90	$\gamma$ 66.2762
Atom		x		y		z	
O	3.51158		2.19731		6.33699		
O	8.64118		2.19731		6.33699		
Zr	4.66312		3.7109		7.1297		
Zr	9.79272		3.7109		7.1297		
O	3.71395		5.54856		7.30903		
O	8.84355		5.54856		7.30903		
O	6.18734		5.20706		7.56457		
O	11.31694		5.20706		7.56457		
Zr	2.58219		1.23317		7.88118		
Zr	7.71179		1.23317		7.88118		
O	3.50199		3.24628		8.68191		
O	8.63159		3.24628		8.68191		
O	6.01755		2.61725		8.3085		
O	11.14715		2.61725		8.3085		
Zr	2.46454		5.03609		9.07933		
Zr	7.59414		5.03609		9.07933		
O	3.80856		0.94265		9.6003		
O	8.93816		0.94265		9.6003		
O	1.25681		0.57227		9.84712		
O	6.38641		0.57227		9.84712		
Zr	5.13075		2.45192		10.29317		
Zr	10.26035		2.45192		10.29317		
O	6.41286		4.41661		10.75941		
O	11.54246		4.41661		10.75941		
O	3.84805		3.96931		10.97838		
O	8.97765		3.96931		10.97838		
Zr	2.566		0.09922		11.44465		
Zr	7.6956		0.09922		11.44465		
O	1.28332		2.06329		11.91117		
O	6.41292		2.06329		11.91117		
O	3.84808		1.61529		12.13037		
O	8.97768		1.61529		12.13037		
Zr	2.56536		3.57939		12.59689		
Zr	7.69496		3.57939		12.59689		
O	3.84748		5.54408		13.06313		
O	8.97708		5.54408		13.06313		
O	6.41226		5.09675		13.28211		
O	11.54186		5.09675		13.28211		
Zr	5.1302		1.22665		13.74837		
Zr	10.2598		1.22665		13.74837		
O	3.85578		3.13191		14.20806		
O	8.98641		3.12502		14.20247		
O	1.30089		2.75206		14.37398		
O	6.43303		2.76077		14.39452		
Zr	5.17078		4.58484		14.9749		

Zr	10.3071	4.58219	14.93739
O	4.20626	1.02924	15.68442
O	9.30736	1.01033	15.66616
O	1.60555	0.48291	15.49008
O	6.68933	0.47014	15.46591
Zr	2.58829	2.4708	15.99798
Zr	7.73945	2.48848	15.99817
O	3.94247	3.95414	16.70576
O	9.09269	3.95599	16.68385
O	6.61939	4.29603	16.43793
O	11.73079	4.27599	16.38272
Zr	2.83931	5.70696	17.04524
Zr	8.02567	5.72794	17.0277
O	1.52609	1.66041	17.67378
O	6.8923	1.54275	17.7105
H	11.67579	2.14724	18.50833
O	12.42336	5.71334	18.94002
H	12.75018	6.04794	19.78917
O	18.75915	5.0173	19.27663
S	17.79918	6.24092	20.01126
O	18.75089	7.23984	20.44314
O	17.0946	6.58582	18.72385
O	16.96018	5.60833	20.97927
H	19.67944	5.06467	19.61345
H	19.62712	7.73208	18.62089

Optimized XYZ position for S-T(111)-2

Lengths Å	a 10.2592	b 6.3733	c 27.6706	Angles degree	α 90	β 90	γ 66.2762
	Atom	x	y	z			
O	3.51158	2.19731	6.33699				
O	8.64118	2.19731	6.33699				
Zr	4.66312	3.7109	7.1297				
Zr	9.79272	3.7109	7.1297				
O	3.71395	5.54856	7.30903				
O	8.84355	5.54856	7.30903				
O	6.18734	5.20706	7.56457				
O	11.31694	5.20706	7.56457				
Zr	2.58219	1.23317	7.88118				
Zr	7.71179	1.23317	7.88118				
O	3.50199	3.24628	8.68191				
O	8.63159	3.24628	8.68191				
O	6.01755	2.61725	8.3085				
O	11.14715	2.61725	8.3085				
Zr	2.46454	5.03609	9.07933				
Zr	7.59414	5.03609	9.07933				
O	3.80856	0.94265	9.6003				
O	8.93816	0.94265	9.6003				
O	1.25681	0.57227	9.84712				
O	6.38641	0.57227	9.84712				
Zr	5.13075	2.45192	10.29317				
Zr	10.26035	2.45192	10.29317				
O	6.41286	4.41661	10.75941				
O	11.54246	4.41661	10.75941				
O	3.84805	3.96931	10.97838				
O	8.97765	3.96931	10.97838				
Zr	2.566	0.09922	11.44465				
Zr	7.6956	0.09922	11.44465				
O	1.28332	2.06329	11.91117				
O	6.41292	2.06329	11.91117				
O	3.84808	1.61529	12.13037				
O	8.97768	1.61529	12.13037				
Zr	2.56536	3.57939	12.59689				
Zr	7.69496	3.57939	12.59689				
O	3.84748	5.54408	13.06313				
O	8.97708	5.54408	13.06313				
O	6.41226	5.09675	13.28211				
O	11.54186	5.09675	13.28211				
Zr	5.1302	1.22665	13.74837				
Zr	10.2598	1.22665	13.74837				
O	3.85647	3.12388	14.21215				
O	9.01055	3.13096	14.20255				
O	1.323	2.73453	14.4183				
O	6.45346	2.74076	14.42192				
Zr	5.26875	4.49254	14.96545				

Zr	10.2998	4.5638	14.95206
O	4.22092	1.04241	15.66386
O	9.35518	1.01537	15.68269
O	1.6878	0.42169	15.52575
O	6.67408	0.44267	15.37289
Zr	2.60726	2.46267	16.02179
Zr	7.70408	2.4408	16.14696
O	3.98236	3.89292	16.70372
O	9.11859	4.0045	16.77827
O	6.65096	4.28834	16.45078
O	11.79549	4.34313	16.4319
Zr	2.97919	5.72412	16.97003
Zr	8.0862	5.83711	17.0216
O	1.15433	1.88519	17.55459
O	6.8418	1.4715	17.68043
O	4.27326	5.33692	18.87752
S	5.63574	6.15396	19.61513
O	5.02551	7.26471	20.27401
O	6.25513	5.08298	20.34546
O	6.44572	6.56953	18.41209
H	3.18699	8.48239	17.85281
H	4.57183	4.40463	18.75497

Optimized XYZ position for 2S-T(111)

Lengths Å	a 10.2592	b 6.3733	c 27.6706	Angles degree	$\alpha$ 90	$\beta$ 90	$\gamma$ 66.2762
Atom		x		y		z	
O	3.51158		2.19731		6.33699		
O	8.64118		2.19731		6.33699		
Zr	4.66312		3.7109		7.1297		
Zr	9.79272		3.7109		7.1297		
O	3.71395		5.54856		7.30903		
O	8.84355		5.54856		7.30903		
O	6.18734		5.20706		7.56457		
O	11.31694		5.20706		7.56457		
Zr	2.58219		1.23317		7.88118		
Zr	7.71179		1.23317		7.88118		
O	3.50199		3.24628		8.68191		
O	8.63159		3.24628		8.68191		
O	6.01755		2.61725		8.3085		
O	11.14715		2.61725		8.3085		
Zr	2.46454		5.03609		9.07933		
Zr	7.59414		5.03609		9.07933		
O	3.80856		0.94265		9.6003		
O	8.93816		0.94265		9.6003		
O	1.25681		0.57227		9.84712		
O	6.38641		0.57227		9.84712		
Zr	5.13075		2.45192		10.29317		
Zr	10.26035		2.45192		10.29317		
O	6.41286		4.41661		10.75941		
O	11.54246		4.41661		10.75941		
O	3.84805		3.96931		10.97838		
O	8.97765		3.96931		10.97838		
Zr	2.566		0.09922		11.44465		
Zr	7.6956		0.09922		11.44465		
O	1.28332		2.06329		11.91117		
O	6.41292		2.06329		11.91117		
O	3.84808		1.61529		12.13037		
O	8.97768		1.61529		12.13037		
Zr	2.56536		3.57939		12.59689		
Zr	7.69496		3.57939		12.59689		
O	3.84748		5.54408		13.06313		
O	8.97708		5.54408		13.06313		
O	6.41226		5.09675		13.28211		
O	11.54186		5.09675		13.28211		
Zr	5.1302		1.22665		13.74837		
Zr	10.2598		1.22665		13.74837		
O	3.89294		3.11551		14.19106		
O	9.02053		3.11206		14.18759		
O	1.35062		2.71911		14.39467		
O	6.47785		2.71824		14.3959		
Zr	5.23316		4.54021		14.96146		

Zr	10.36429	4.52963	14.96169
O	4.25798	0.99895	15.68751
O	9.38996	1.00895	15.69475
O	1.65302	0.42114	15.51507
O	6.80414	0.41941	15.53762
Zr	2.65599	2.42594	16.00606
Zr	7.77604	2.42501	16.00428
O	4.03499	3.88513	16.68342
O	9.16103	3.87731	16.67249
O	6.69813	4.27943	16.40675
O	11.82491	4.26839	16.40987
Zr	2.97778	5.64317	17.05995
Zr	8.11444	5.63513	17.06887
O	1.60838	1.64046	17.68615
O	6.61548	1.71327	17.652
H	8.38213	11.66579	19.14908
H	3.17052	11.75926	19.04011
O	9.30706	11.9137	18.91719
O	4.07283	12.1067	18.81972
O	11.47794	10.99789	19.10273
O	6.15887	11.00309	19.13949
O	9.82506	10.30655	20.8104
O	4.51339	10.8711	20.98969
O	10.80906	12.60186	20.83992
O	5.72056	12.99992	20.48698
S	10.44084	11.40993	20.1392
S	5.19542	11.72695	20.07446
H	8.97371	8.04144	18.46818
H	14.27796	7.85903	18.57485

## REFERENCE

1. T. Jin, T. Yamaguchi and K. Tanabe, *J. Phys. Chem.*, 1986, **90**, 4794-4796.
2. D. A. Ward and E. I. Ko, *J. Catal.*, 1994, **150**, 18-33.
3. K. Arata and M. Hino, *Mater. Chem. Phys.*, 1990, **26**, 213-237.
4. A. Clearfield, G. P. D. Serrette and A. H. Khazisyyed, *Catal. Today*, 1994, **20**, 295-312.
5. J. H. Lunsford, H. Sang, S. M. Campbell, C. H. Liang and R. G. Anthony, *Catal. Lett.*, 1994, **27**, 305-314.
6. M. Bensitel, O. Saur, J. C. Lavalle and B. A. Morrow, *Mater. Chem. Phys.*, 1988, **19**, 147-156.
7. T. Riemer, D. Spielbauer, M. Hunger, G. A. H. Mekhemer and H. Knozinger, *J. Chem. Soc.-Chem. Commun.*, 1994, **10**, 1181-1182.
8. C. Morterra, G. Cerrato, F. Pinna and M. Signoretto, *J. Phys. Chem.*, 1994, **98**, 12373-12381.
9. L. M. Kustov, V. B. Kazansky, F. Figueras and D. Tichit, *J. Catal.*, 1994, **150**, 143-149.
10. R. Srinivasan, R. A. Keogh, D. R. Milburn and B. H. Davis, *J. Catal.*, 1995, **153**, 123-130.