

Supplementary Material

An S-scheme heterojunction constructed from α -Fe₂O₃ and In-doped carbon nitride for high-efficiency CO₂ photoreduction

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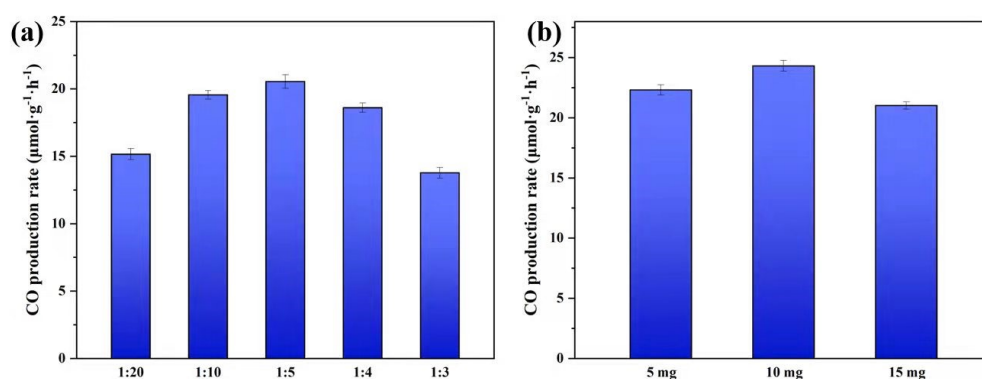


Fig. S1 Photocatalytic CO₂ reduction activity of (a) α -Fe₂O₃/PCN with different mass ratios of α -Fe₂O₃ to PCN and (b) α -Fe₂O₃/In-CN for which In-CN are prepared with different dosages of In-MOF. The corresponding In contents are 0.24 wt%, 0.41 wt% and 0.70 wt%, respectively.

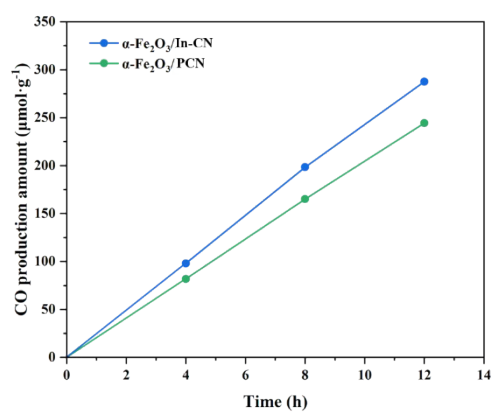


Fig. S2 Time-dependent CO yield of α -Fe₂O₃/PCN and α -Fe₂O₃/In-CN heterojunctions.

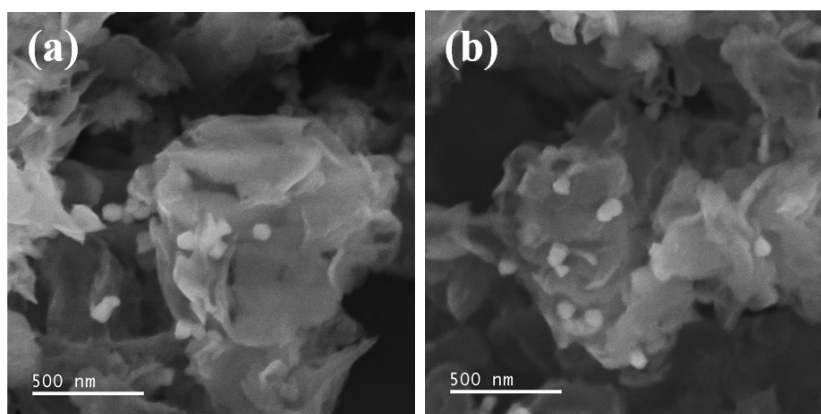


Fig. S3 SEM images of α -Fe₂O₃/In-CN (a) before reaction and (b) after continuous reaction for 12 h.

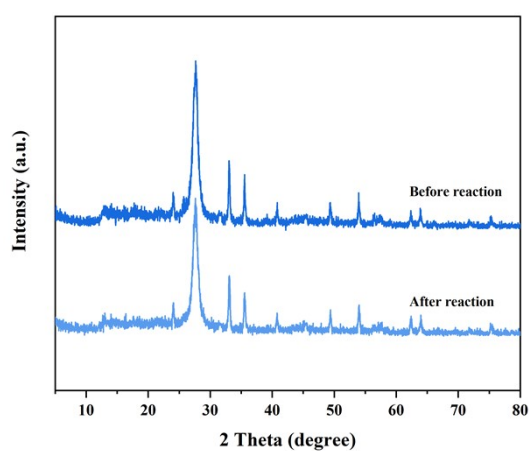


Fig. S4 XRD patterns of α -Fe₂O₃/In-CN before reaction and after continuous reaction for 12 h.

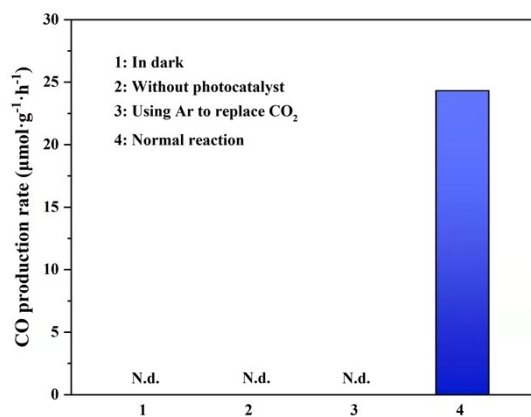


Fig. S5 CO production rate over $\alpha\text{-Fe}_2\text{O}_3/\text{In-CN}$ under various conditions (N.d.: not detected).

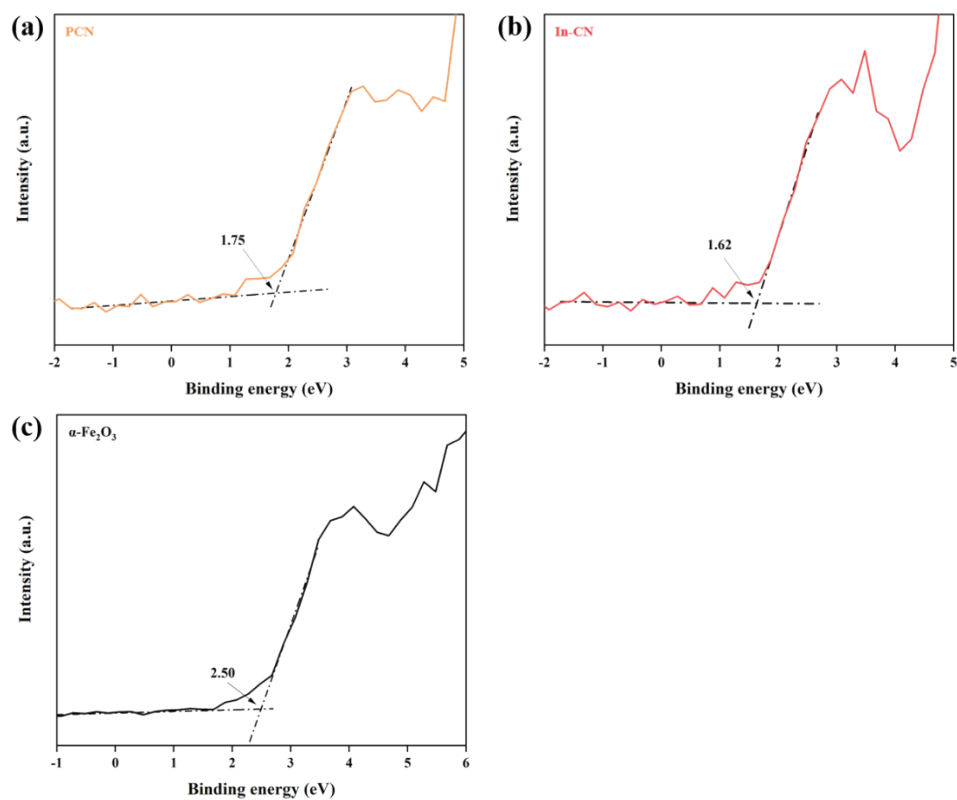


Fig. S6 Valence band (VB)-XPS spectra of (a) PCN, (b) In-CN and (c) $\alpha\text{-Fe}_2\text{O}_3$.

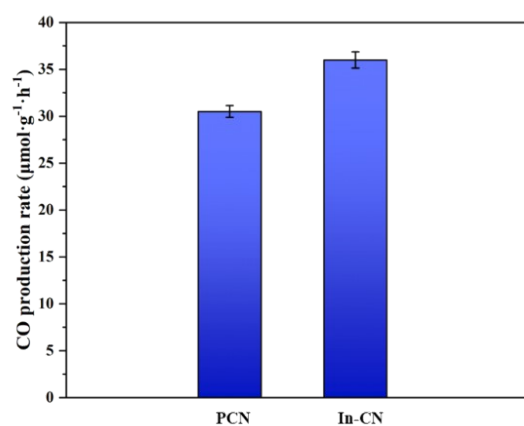


Fig. S7 Comparison of the photocatalytic activity of PCN and In-CN in the presence of TEOA as hole sacrificial agent.

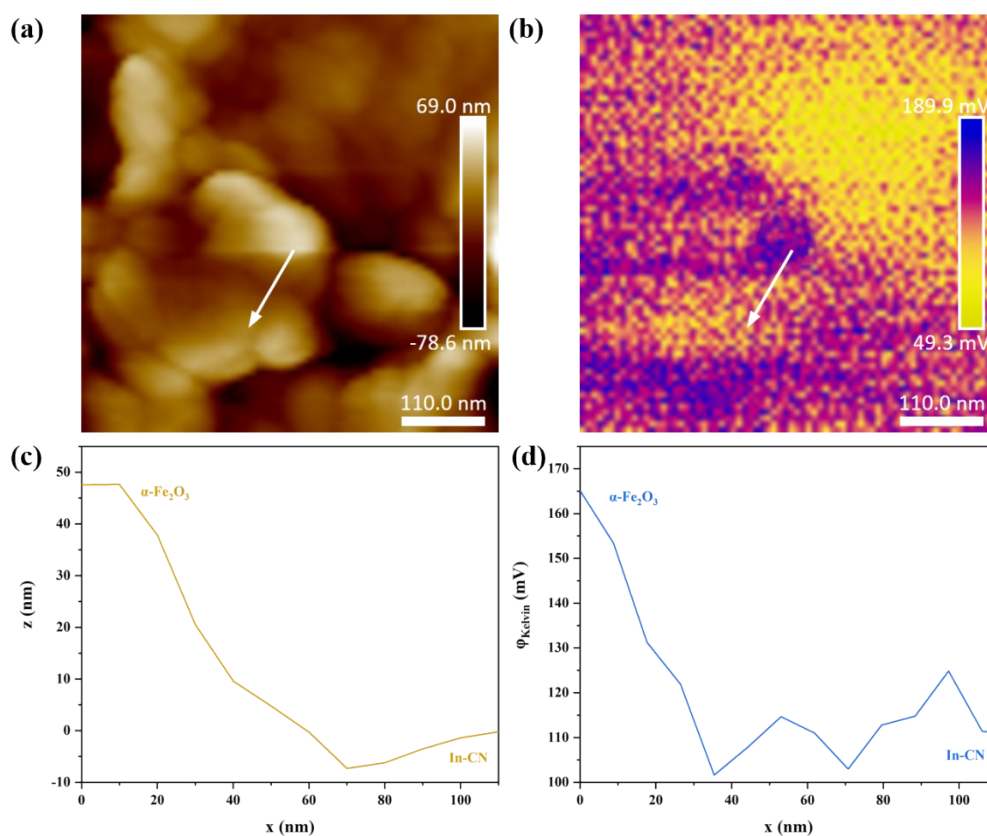


Fig. S8 (a) Topography image and (b) surface potential of $\alpha\text{-Fe}_2\text{O}_3/\text{In-CN}$, (c,d) line-scanning profile and the corresponding Kelvin probe signal through one $\alpha\text{-Fe}_2\text{O}_3$ nanoparticle.

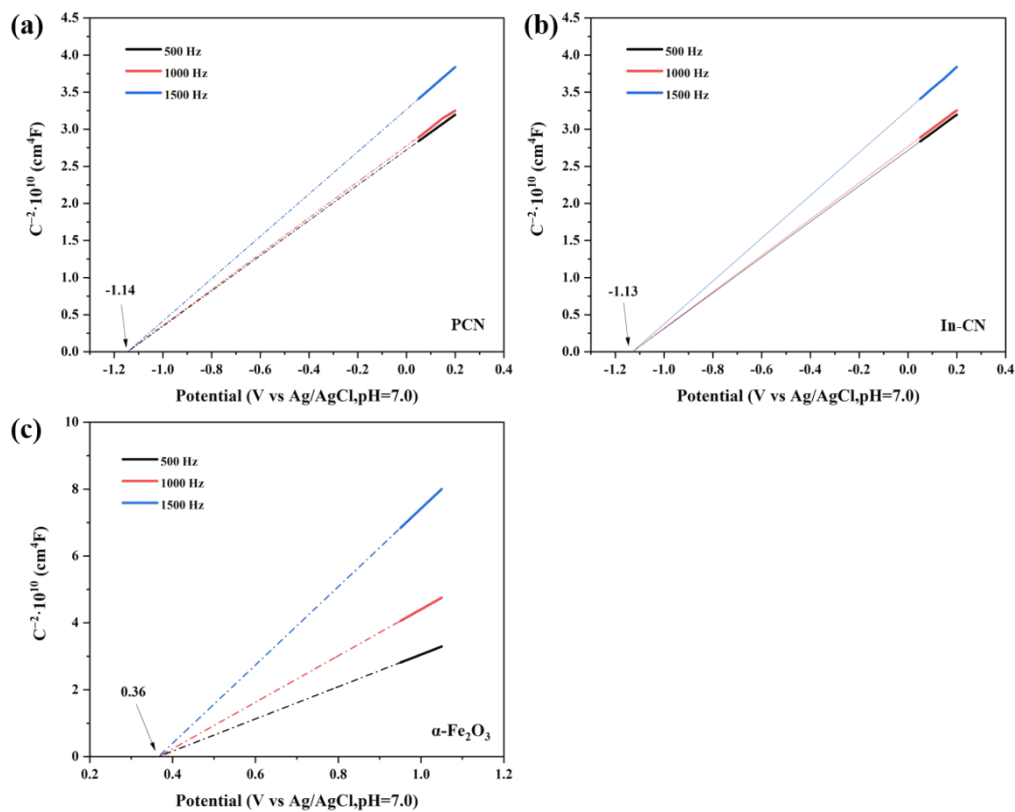


Fig. S9 Mott-Schottky plots of (a) PCN, (b) In-CN and (c) α -Fe₂O₃.

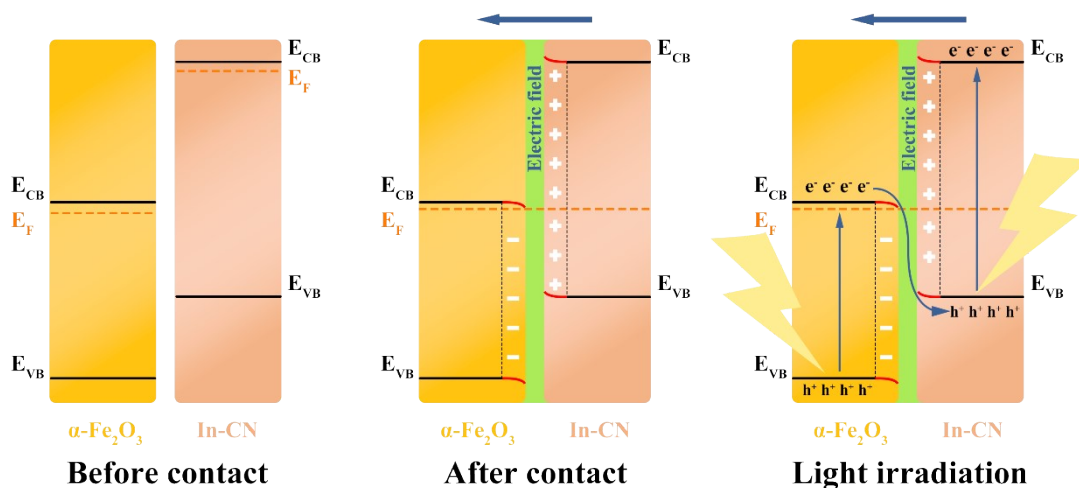


Fig. S10 Energy band positions of α -Fe₂O₃ and In-CN (a) before contact, (b) after contact, and (c) S-scheme charge transfer mechanism under light irradiation.

Table S1 Comparison of the CO₂ photoreduction performance of α -Fe₂O₃/In-CN with those of other photocatalysts reported in the literature.

Photocatalyst	Light source	Reaction medium	Product	Product generation rate ($\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$)	Ref.
α -Fe ₂ O ₃ /In-CN	300W Xe lamp	Gas-solid, H ₂ O	CO	24.3	This work
MoS ₂ /TiO ₂	300W Xe lamp	Gas-solid, H ₂ O	CO CH ₄	2.83 1.45	[1]
NH ₂ -MIL-101(Fe)/g-C ₃ N ₄	300W Xe lamp $\lambda > 400$ nm	Gas-solid, TEOA	CO	22.13	[2]
Bi ₃ TiNbO ₉	300W Xe lamp with AM1.5 G filter	Gas-solid, H ₂ O	CO	20.91	[3]
Mo/C ₃ N ₄	300W Xe lamp $\lambda > 420$ nm	Gas-solid, H ₂ O	CO H ₂	18.0 37.0	[4]
Urchin-like α -Fe ₂ O ₃ /g-C ₃ N ₄	300W Xe lamp	Gas-solid, H ₂ O	CO	27.2	[5]
InVO ₄ / β -AgVO ₃	300W Xe lamp $\lambda > 420$ nm	Gas-solid, H ₂ O	CO	12.61	[6]
ZnFe ₂ O ₄ /RGO/In ₂ O ₃	300W Xe lamp	Gas-solid, H ₂ O	CO CH ₄	8.85 1.95	[7]
Co ₁ -C ₃ N ₄ @ α -Fe ₂ O ₃	300W Xe lamp	Gas-solid, H ₂ O	CO	25.2	[8]
NiAl-LDH/Ti ₃ C ₂	300W Xe lamp	Gas-solid, H ₂ O	CO	11.82	[9]
LaPO ₄ /g-C ₃ N ₄	300W Xe lamp	Gas-solid, H ₂ O	CO	14.4	[10]

Table S2 The fitted parameters of time-resolved PL spectra.

Sample	τ_1 (ns)	f_1 (%)	τ_2 (ns)	f_2 (%)	τ (ns)
α -Fe ₂ O ₃ /In-CN	3.87	72.35	16.50	27.65	11.69
In-CN	4.03	81.13	17.44	19.77	10.76
α -Fe ₂ O ₃ /PCN	2.05	82.35	8.50	17.65	5.09
PCN	1.90	80.24	7.53	19.76	4.68

References

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