

## Supporting Information

### Tailoring the activity and selectivity of Rh/SiO<sub>2</sub> in the selective hydrogenation of phenol by CoO<sub>x</sub> promotion

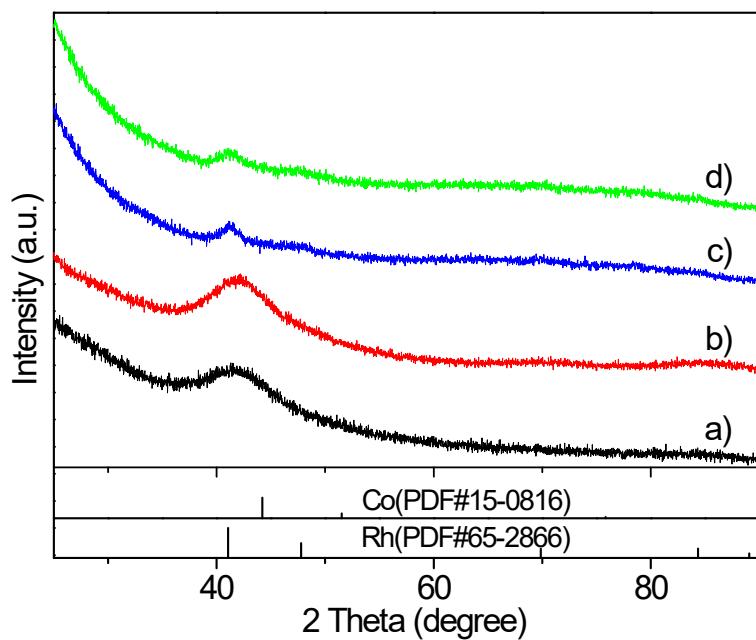
Fei Zhang,<sup>a, c</sup> Chunzheng Wu,<sup>b</sup> Shuibo Wang,<sup>a</sup> Shiwei Wang,<sup>a</sup> Tong Li,<sup>a</sup> Laixi Zou,<sup>c, \*</sup> Hongbo Yu,<sup>a, \*</sup> and Hongfeng Yin<sup>a, \*</sup>

<sup>a</sup> Ningbo Institute of Materials Technology and Engineering, Chinese Academy of Sciences, 1219 Zhongguan West Road, Ningbo, Zhejiang 315201, P. R. China

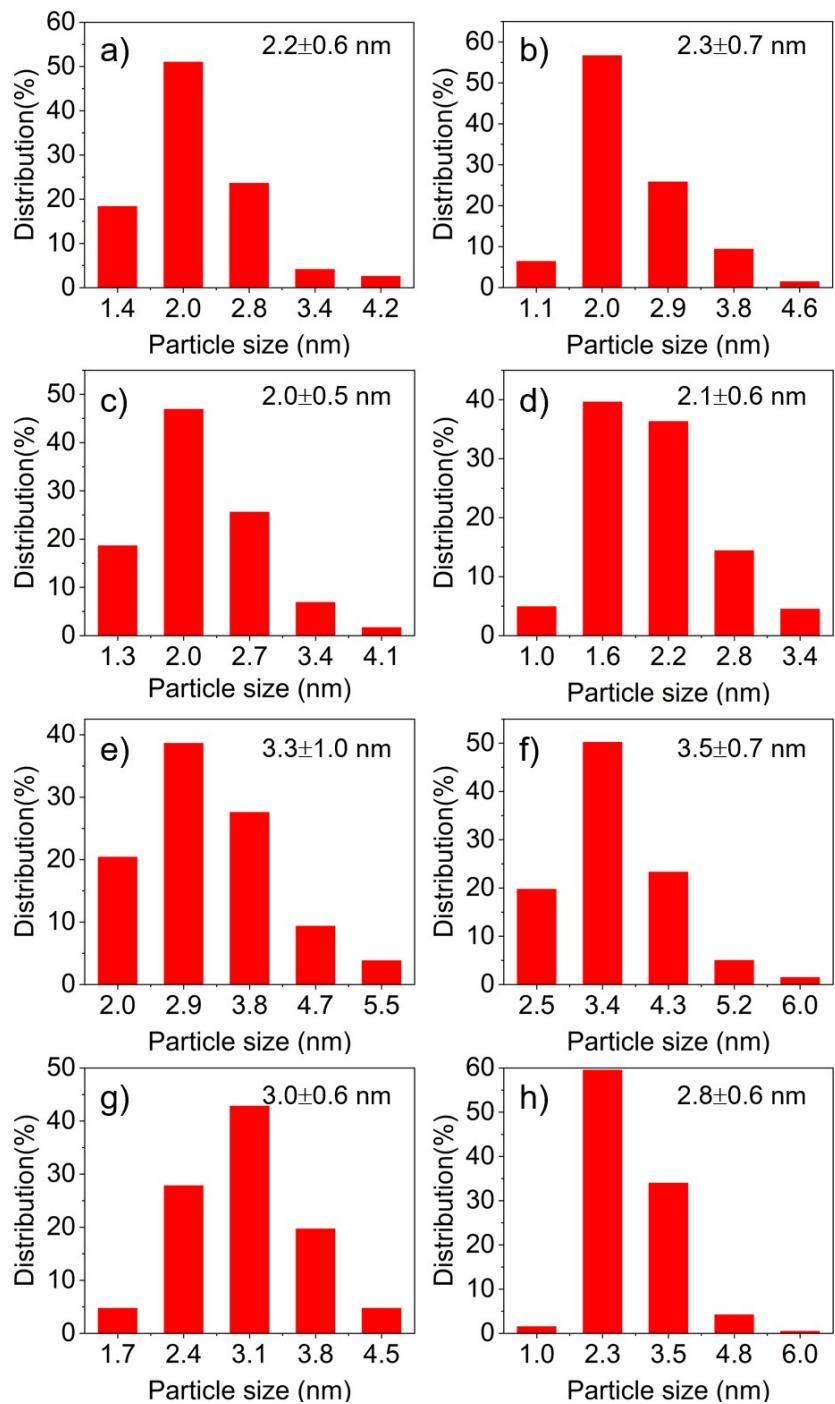
<sup>b</sup> Chemistry and Materials Engineering, Zhejiang A&F University, Hangzhou, Zhejiang 311300, P.R. China

<sup>c</sup> School of Metallurgy and Chemical Engineering, Jiangxi University of Science and Technology, Ganzhou, Jiangxi 341000, P.R. China

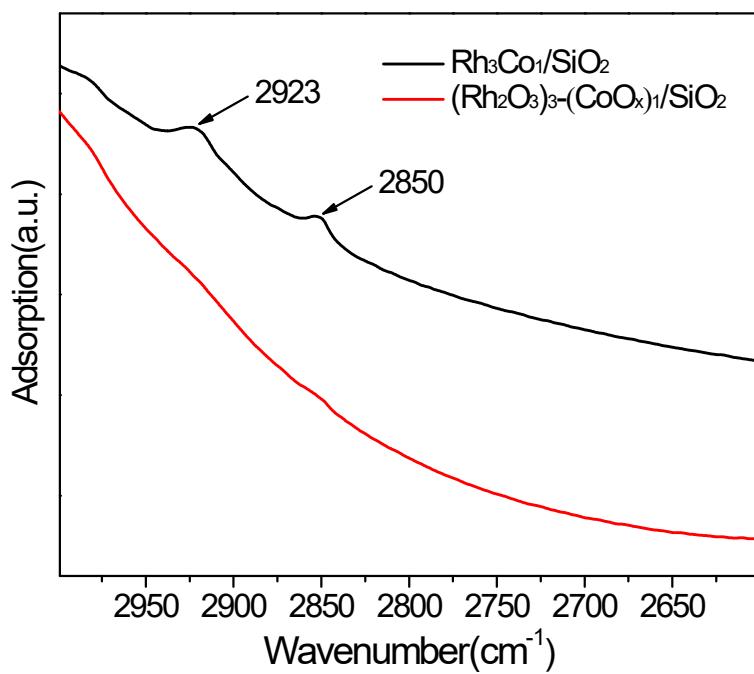
\* Corresponding author: E-mail: yinhf@nimte.ac.cn; Fax: +86 0574 8668 5043



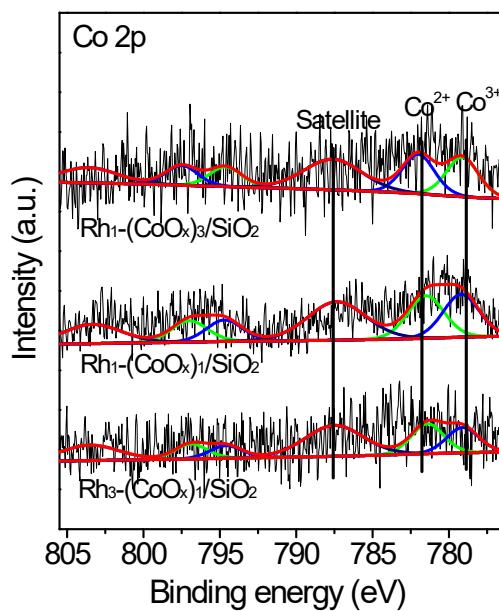
**Fig. S1** XRD patterns showing: a)  $\text{Rh}_1\text{Co}_3$  NPs; b)  $\text{Rh}_1\text{Co}_1$  NPs;  $\text{Rh}_1\text{-}(\text{CoO}_x)_3/\text{SiO}_2$  nanocatalysts;  $\text{Rh}_1\text{-}(\text{CoO}_x)_1/\text{SiO}_2$  nano catalysts.



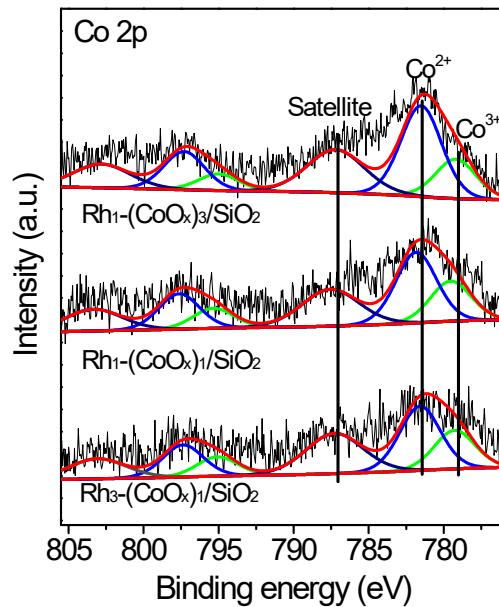
**Fig. S2** Size distributions of as-synthesized nanomaterials showing: a), Rh NPs; b),  $Rh_3Co_1$  NPs; c),  $Rh_1Co_1$  NPs; d),  $Rh_1Co_3$  NPs; e)  $Rh/SiO_2$ ; f),  $Rh_3-(CoO_x)_1/SiO_2$ ; g),  $Rh_1-(CoO_x)_1/SiO_2$ ; and h),  $Rh_1-(CoO_x)_3/SiO_2$ .



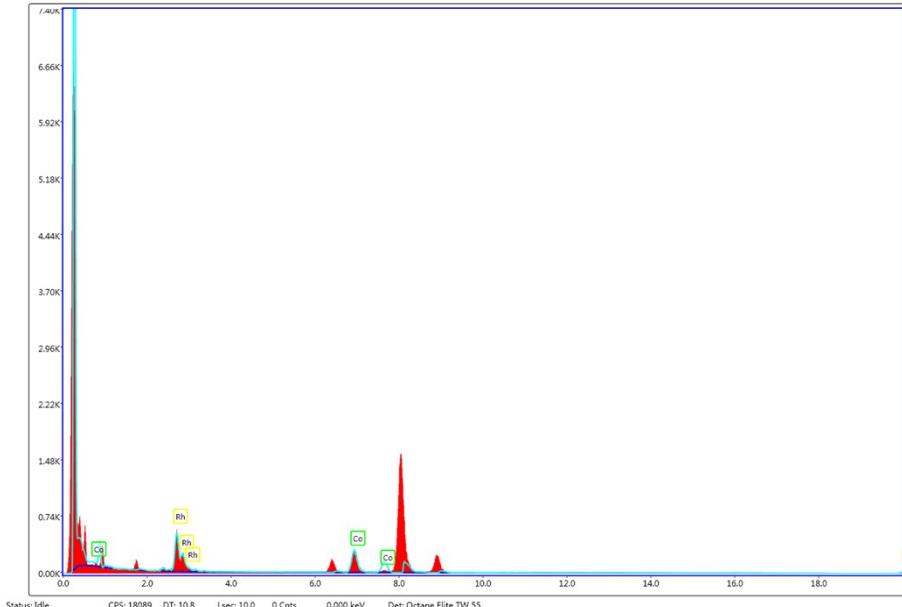
**Fig. S3** FT-IR spectra of  $\text{Rh}_3\text{Co}_1/\text{SiO}_2$  sample (without calcination) and  $(\text{Rh}_2\text{O}_3)_3-(\text{CoO}_x)_1/\text{SiO}_2$  after calcination treatment.



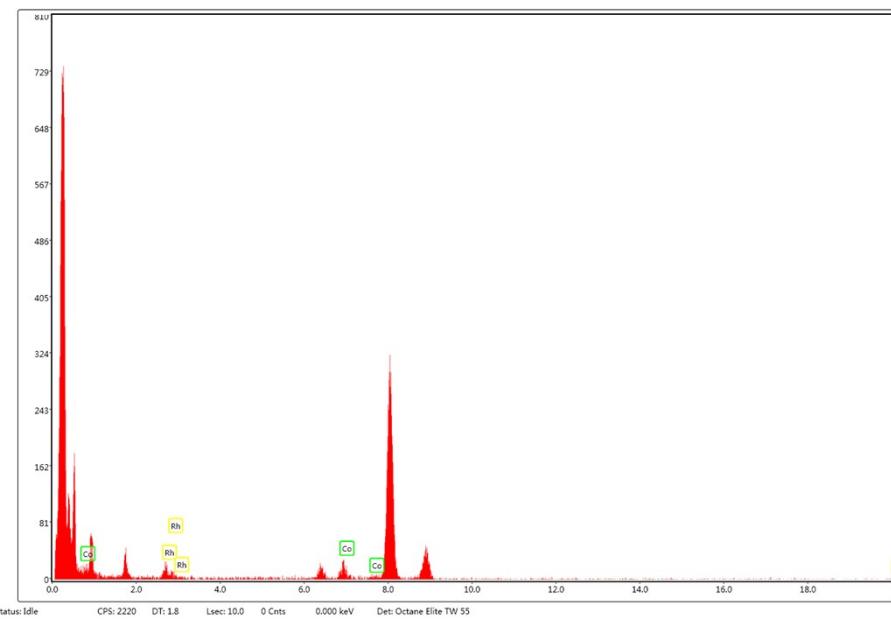
**Figure S4** Co 2p XPS spectra of  $\text{SiO}_2$  supported Rh-based nanocatalysts. (Rh loading, theoretical value 1 wt%).



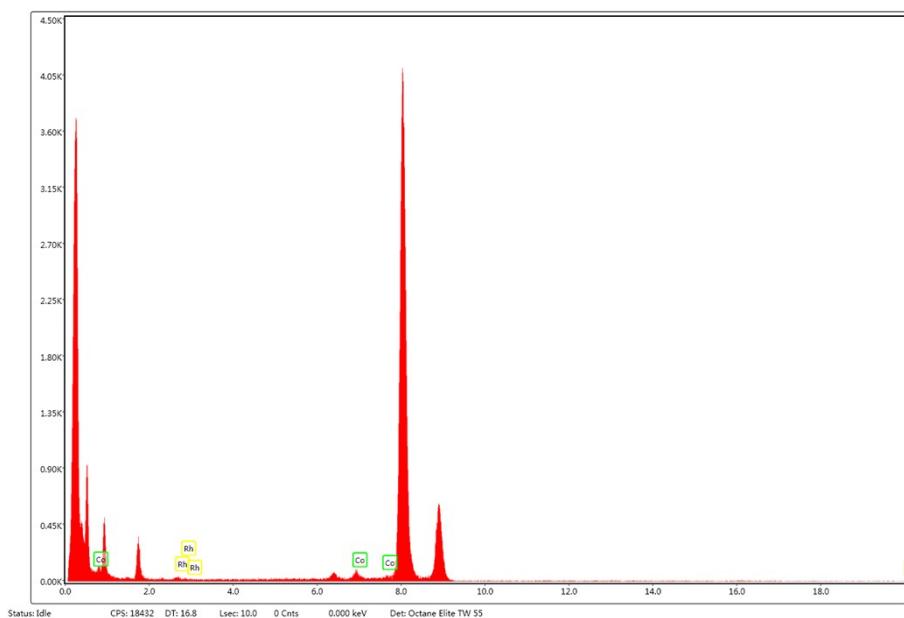
**Figure S5** Co 2p XPS spectra of  $\text{SiO}_2$  supported Rh-based nanocatalysts. (Rh loading, theoretical value 3 wt%).



**Fig. S6** EDS spectra of  $\text{Rh}_3\text{-}(\text{CoO}_x)_1/\text{SiO}_2$  nano catalyst.



**Fig. S7** EDS spectra of  $\text{Rh}_1\text{-}(\text{CoO}_x)_1/\text{SiO}_2$  nano catalyst.



**Fig. S8** EDS spectra of  $\text{Rh}_1\text{-}(\text{CoO}_x)_3/\text{SiO}_2$  nano catalyst.

**Table S1.** The actual loadings of Rh and Co in Rh-CoO<sub>x</sub>/SiO<sub>2</sub> catalysts analyzed by ICP-OES.

Sample	Rh loading (wt%)	Co loading (wt%)
Rh/SiO <sub>2</sub>	0.99%	-
Rh <sub>3</sub> -(CoO <sub>x</sub> ) <sub>1</sub> /SiO <sub>2</sub>	1.02%	0.18%
Rh <sub>1</sub> -(CoO <sub>x</sub> ) <sub>1</sub> /SiO <sub>2</sub>	1.08%	0.56%
Rh <sub>1</sub> -(CoO <sub>x</sub> ) <sub>3</sub> /SiO <sub>2</sub>	1.04%	1.65%

**Table S2.** The catalytic performance for hydrogenations of phenol over CoO<sub>x</sub>/SiO<sub>2</sub> catalyst.

Catalyst <sup>a</sup>	Conversion (%)	Selectivity (%)	
		Cyclohexanol	Others
0.1000 g CoO <sub>x</sub> /SiO <sub>2</sub>	0	N/A	N/A

<sup>a</sup>Reaction conditions: phenol, 0.1000 g; temperature, 60 °C; ethanol, 10.0 mL; H<sub>2</sub>, 1.0 MPa; reaction time, 100 min; speed of agitation, 600 rpm/min.

### Synthesis of CoO<sub>x</sub>/SiO<sub>2</sub> catalyst.

The CoO<sub>x</sub>/SiO<sub>2</sub> catalyst was prepared by a wet impregnation method. A certain amount of Co(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O aqueous solution was impregnated on 1.0 g of SiO<sub>2</sub> to prepare the 1.0 wt % CoO<sub>x</sub>/SiO<sub>2</sub> catalyst. After 4h of immersion, the sample was dried for 5h at 110 °C, and followed by calcining at 500 °C for 3h in air to obtain CoO<sub>x</sub>/SiO<sub>2</sub> catalyst.

**Table S3.** Comparison of catalytic performance of  $\text{Rh}_3\text{-}(\text{CoO}_x)_1/\text{SiO}_2$  and previously reported heterogeneous catalysts for the selective hydrogenation of phenol.

Catalysts	Conditions			Conv. (%)	Sel. (%)	TON <sup>a</sup> / h	Ref.
	T (°C)	P (MPa)	t (h)				
$\text{Rh}_3\text{-}(\text{CoO}_x)_1/\text{SiO}_2$	60	1.0	1.7	98.1	99.4	213	This work
$\text{Rh}/\text{SiO}_2$	25	0.1	9.0	100	18	18	1
$\text{Rh@HMSNs}$	45	0.5	3.0	90.6	96.6	28	2
$\text{Rh/CNF}$	300	2.0	0.2	51	40	89	3
$\text{Rh-PAA}$	80	4.0	0.5	40	100	40	4
FFSiRh	75	0.6	2.0	100	52	10.	5
$\text{Rh@S-MIL-101}$	50	0.5	2.0	80	35	33	6

<sup>a</sup> Turnover number (TON) is measured as moles of products per total molar Rh atoms.

## Reference

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