

One-pot Synthesis of Mesoporous Palladium/C Nanodendrites as High-performance Oxygen Reduction Eletrocatalysts through a Facile Dual Surface Protecting Agent-Assisted Strategy

Qiuyue Liu,^a Qiaoling Kang,^a Zhenhua Wang,^a Qingyi Lu,^{*a} Feng Gao^{*b}

^a State Key Laboratory of Coordination Chemistry, Coordination Chemistry Institute, Collaborative Innovation Center of Advanced Microstructures, School of Chemistry and Chemical Engineering, Nanjing University, Nanjing 210023, P. R. China. E-mail: qylu@nju.edu.cn

^b Department of Materials Science and Engineering, Jiangsu Key Laboratory of Artificial Functional Materials, Collaborative Innovation Center of Advanced Microstructures, College of Engineering and Applied Sciences, Nanjing University, Nanjing 210093, P. R. China. E-mail: fgao@nju.edu.cn

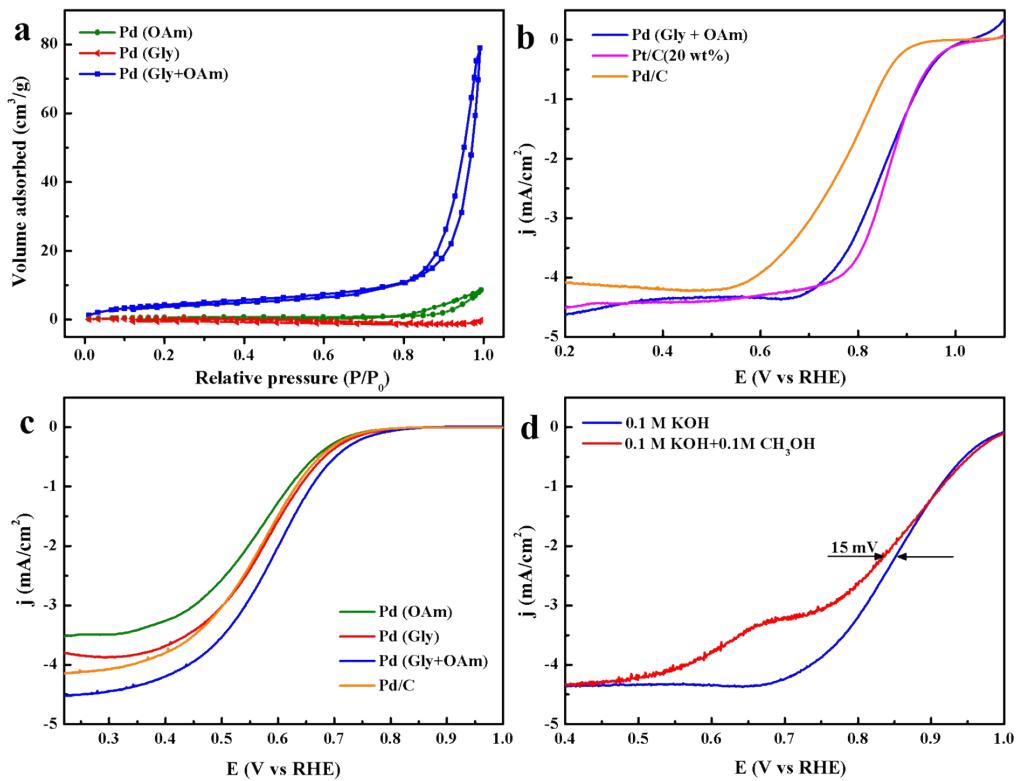


Figure S1 (a) Nitrogen adsorption-desorption isotherms of the as-synthesized Pd catalysts at 77 K; (b) ORR polarization curves in 0.1 M O₂-saturated KOH of the as-synthesized Pd/C nanodendrites and commercial Pt/C and Pd/C catalysts; (c) ORR polarization curves in 0.5 M O₂-saturated H₂SO₄ of the as-synthesized Pd catalysts and commercial Pd/C; (d) ORR polarization curves of mesoporous Pd nanodendrites in 0.1 M KOH and the mixed electrolyte containing 0.1 M KOH and 0.1 M CH₃OH.

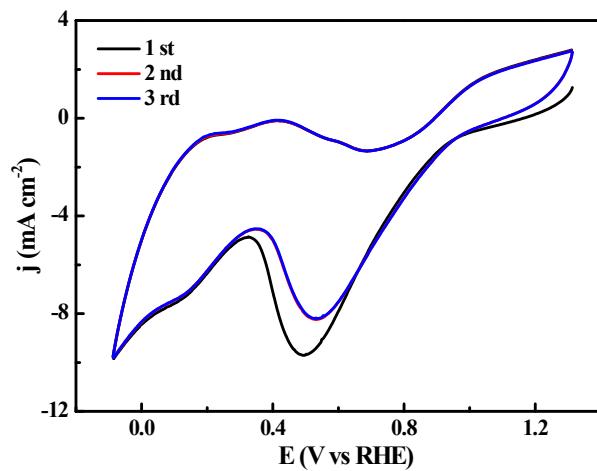


Figure S2 CV curves of the obtained the as-synthesized mesoporous Pd/C nanodendrites.

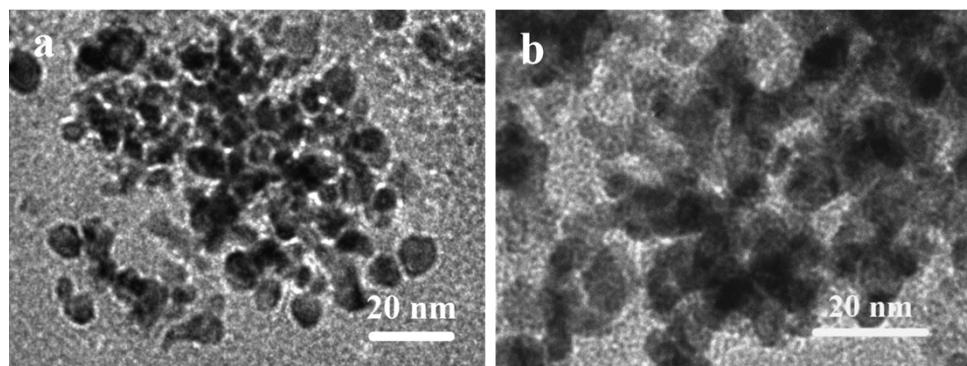


Figure S3 TEM images of the as-synthesized Pd/C nanodendrites after 24 h ORR test.

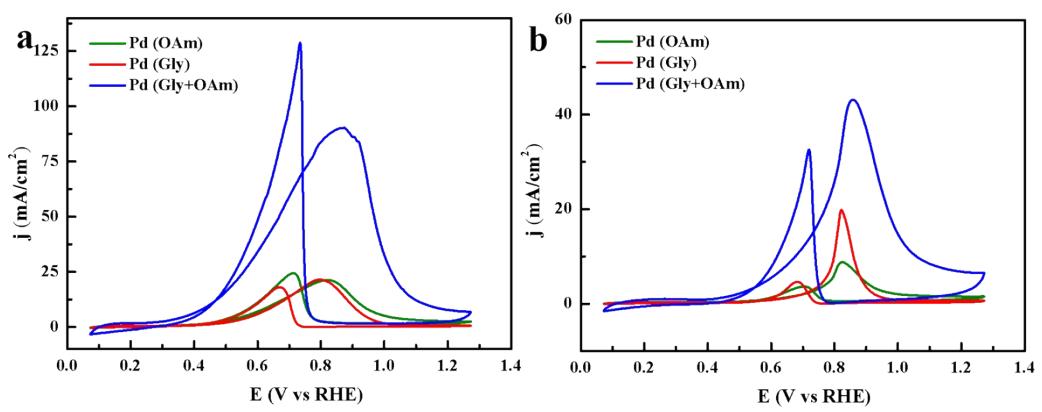


Figure S4 (a) EOR performance in N_2 -saturated 1 M KOH and 1 M $\text{C}_2\text{H}_5\text{OH}$ and (b) MOR performance in N_2 -saturated 1 M KOH and 1 M CH_3OH of the as-synthesized Pd catalysts at the rate of 0.05 V/s.

Table S1 Comparison of the pore structural parameters of the as-synthesized Pd catalysts.

Catalyst	Pd (OAm)	Pd (Gly)	Pd (Gly + OAm)
BET surface area (m ² /g)	2.78	1.02	17.21
Pore volume (cm ³ /g)	\	\	0.122

Table S2 The ORR activity comparison of the Pd/C nanodendrites and the reported Pd nanostructures.

Catalyst	E _{onset} (V vs RHE)	E _{1/2} (V vs RHE)	References
Pd porous nanosheets	~	0.837	<i>Appl. Catal. B Environ.</i> 2019, 243 , 86-93
Multipod Pd	0.643	~	<i>ACS Sustain. Chem. Eng.</i> 2020, 8 , 9217-9225
Pd-PDA-coated CNFs	0.84	0.73	<i>J. Mater. Chem. A</i> 2019, 7 , 7396-7405
PdNC-Pt	0.87	0.78	<i>Electrochim. Acta</i> 2020, 340 , 135840
18Pd-OMC-900	0.963	0.872	<i>Electrochim. Acta</i> 2016, 191 , 355-363
Pd nanodendrites	0.975	0.851	this work
PdH _{0.33} NDs	~	0.911	<i>Electrochim. Commun.</i> 2019, 102 , 67-71
PdBp MSs	1.01	0.92	<i>ACS Nano</i> 2019, 13 , 12052-12061
Pd metallene/C	1.02	0.90	<i>Angew. Chem. Int. Ed.</i> 2021, 60 , 10.1002/anie.202101019