

## **Spin crossover phenomenon in a 2D heterometallic coordination polymer with $[\text{Pd}(\text{SCN})_4]^{2-}$ building blocks**

Kai-Ping Xie, Si-Guo Wu, Long-Fei Wang, Guo-Zhang Huang, Zhao-Ping Ni\* and Ming-Liang Tong

Key Laboratory of Bioinorganic and Synthetic Chemistry of Ministry of Education, School of Chemistry, Sun Yat-Sen University, Guangzhou, 510275 (P. R. China), Fax: (+)86 20 8411-2245, E-mail: nizhp@mail.sysu.edu.cn.

**Table S1.**  $\pi\cdots\pi$  interactions between anthracene units in **1** and **2**.

	<b>1</b>		<b>2</b>
	<b>120 K</b>	<b>298 K</b>	<b>150 K</b>
$Z^a$	5.7464(3)	5.7286(5)	5.8262(2)
$Z^b$	4.8131(2)	4.8992(4)	4.7646(2)
$I^c$	2.5762(1)	2.6528(2)	3.4516(2)
$r^d$	0.47628	0.48534	0.61034
$\theta^e$	67.106(113)	68.946(244)	36.719(163)
$\alpha^f$	156.097(2)	161.855(4)	108.942(2)

<sup>a</sup>The distance between the centroids of anthracene rings (Å); <sup>b</sup>The shortest distance between the centroids of benzene rings (Å); <sup>c</sup>The shortest distance of H-to-ring center (Å); <sup>d</sup>The offset distance between two aromatic rings (Å); <sup>e</sup>The angle between two aromatic rings (°); <sup>f</sup>The angle of  $\pi$ -H  $\cdots$   $\pi$  (°).

**Table S2.**  $\pi\cdots\pi$  interactions between anthracene and pyridine units in **1** and **2**.

		$Z^c$	$I^d$	$r^e$	$\theta^f$
<b>1_120 K</b>	anth1 and py1 <sup>a</sup>	5.1557(3)	2.9655(11)	0.5115	67.009(138)
<b>1_298 K</b>	anth1 and py1 <sup>a</sup>	5.2495(6)	3.0354(3)	0.5210	70.504(227)
<b>2_150 K</b>	anth1 and py1 <sup>b</sup>	4.6187(2)	2.9100(1)	1.3887	53.082(236)
	anth2 and py1 <sup>b</sup>	4.6627(2)	2.5864(1)	1.3887	53.082(241)

<sup>a</sup>anth1: C7-C12, x, 1+y, z; py1: C1-C5, N1, x, y, z.

<sup>b</sup>anth1: C7-C12, 3/2-x, -1/2+y, 1/2-z, z; py1: C1-C5, N1, x, y, z; anth2: C7-C12, x, y, z, py1: C1-C5, N1, 3/2-x, -1/2+y, 1/2-z.

<sup>c</sup>The shortest distance between the centroids of aromatic rings (Å); <sup>d</sup>The shortest distance of H-to-ring center (Å); <sup>e</sup>The offset distance between two aromatic rings (Å); <sup>f</sup>The angle between two aromatic rings (°)

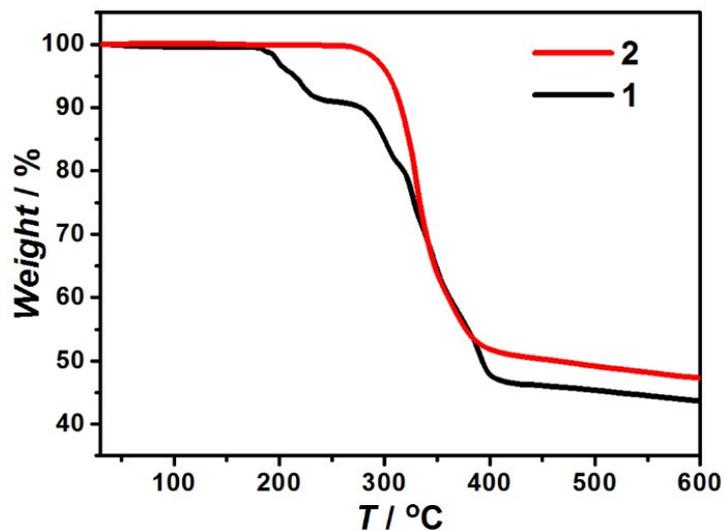


Figure S1. Thermogravimetric analyses of 1 and 2.

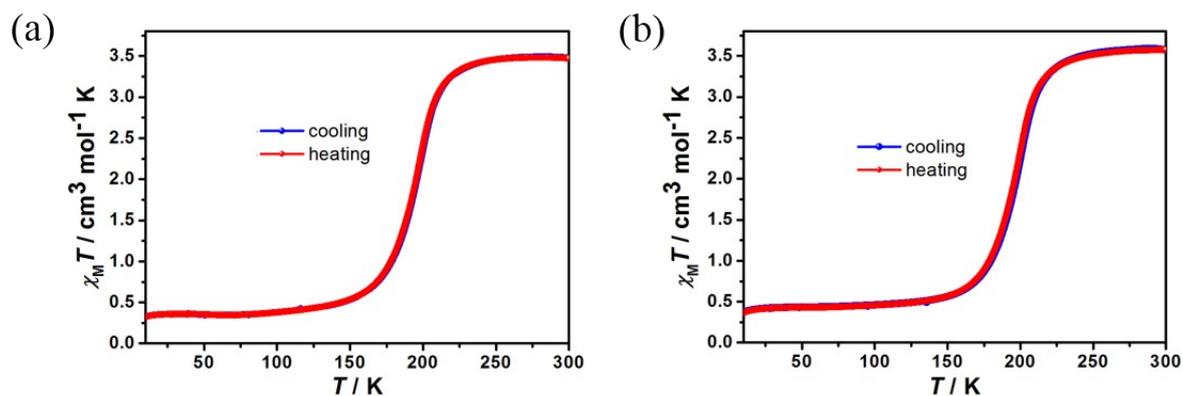


Figure S2. Magnetic susceptibility data for 1 at 2 K min<sup>-1</sup> on fresh samples (a) and 0.5 K min<sup>-1</sup> on old samples (aged about one year) (b) in the cooling and warming modes.

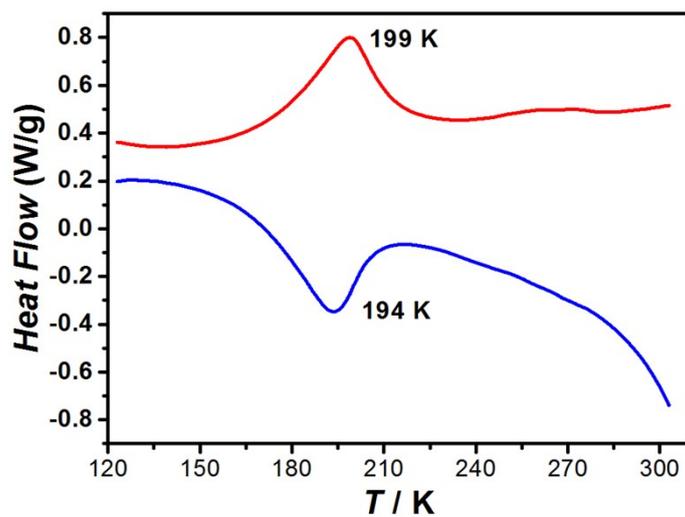


Figure S3. Differential scanning calorimetry (DSC) measurements of 1.

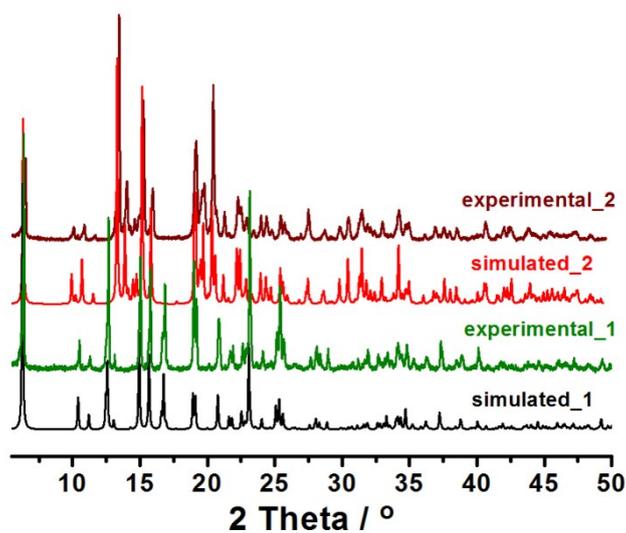


Figure S4. Powder X-ray diffraction data of 1 and 2.

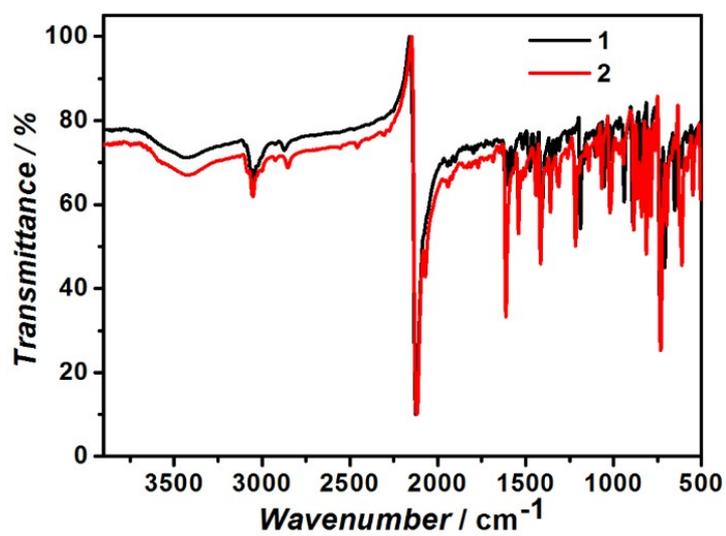
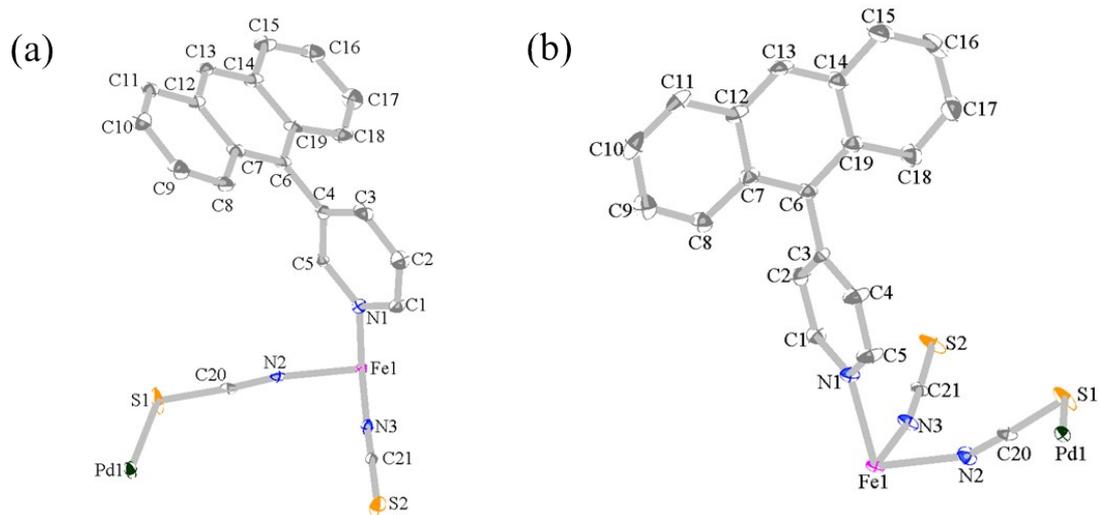
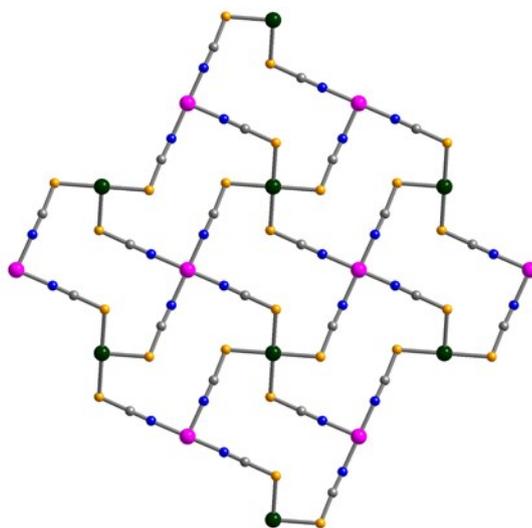


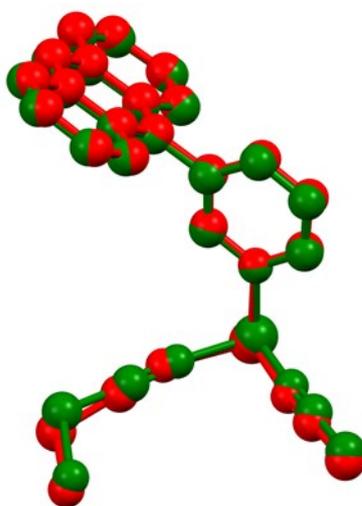
Figure S5. IR spectra of 1 and 2.



**Figure S6.** Asymmetric units of **1** (a) and **2** (b). Hydrogen atoms are omitted for clarity.



**Figure S7.** Top view of the two-dimensional coordination layer in **1** at 120 K.



**Figure S8.** The overlaid structures of **1**: green (LS), red (HS). Hydrogen atoms are omitted for clarity.