Electronic Supplementary Information

Electronic Structure Study of Divanadium Complexes

with Rigid Covalent Coordination: Potential Molecular

Qubits with Slow Spin Relaxation

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Experimental Section

Synthesis of Complexes. All air-sensitive materials were manipulated using standard Schlenk techniques or a glovebox. The complexes $[V_2(\mu-S_2)_2(Et_2dtc)_4]$ (1-Et),¹ $[V_2(\mu-S_2)_2(^iBu_2dtc)_4]$ (1-Bu),² $[V_2(\mu-S_2)_2(Et_2dtc)_4]BF_4$ (2-Et),³ and $[V_2(\mu-S_2)_2(^iBu_2dtc)_4]BF_4$ (2-Bu),³ have been synthesised according to literature procedures.

Sulfur K-edge X-ray Absorption Spectroscopy. All data were measured at the Stanford Synchrotron Radiation Lightsource (SSRL) under ring conditions of 3.0 GeV and 400 mA. S K-edge data were measured using the 54-pole wiggler beam line 4-3 in a high-magnetic field mode of 10 kG with a Ni-coated harmonic rejection mirror and a fully tuned Si(111) double-crystal monochromators. Details of the optimisation of this setup for low-energy have been previously described.⁴ All samples were measured at room temperature as fluorescence spectra using a Lytle detector. Samples were ground finely and dispersed as thinly as possible on Mylar tape to minimise the possibility of fluorescence saturation effects. Data represent 2–3 scan averages. All samples were monitored for photoreduction throughout the course of data collection. The energy was calibrated using the S K-edge spectrum of Na₂S₂O₃·5H₂O, run at intervals between sample scans. The maximum of the first pre-edge feature in the spectrum was fixed at 2472.02 eV. A step size of 0.08 eV was used over the edge region. Data were averaged, and a smooth background was removed from all spectra by fitting a polynomial to the pre-edge region and subtracting this polynomial from the entire spectrum. Normalisation of the data was accomplished by fitting a flattened polynomial or straight line to the post-edge region and normalising the post-edge to 1.0.

Other Physical Methods. Electronic absorption spectra were recorded on a Shimadzu UVA 3600 spectrophotometer (range 200–1600 nm). Continuous wave X-band EPR spectra was recorded on a Bruker ELEXSYS E500 spectrometer. The spectra were simulated with the Bruker XSOPHE suite.⁵ Fluid solution spectra were simulated using a spin Hamiltonian of the form $\hat{H} = g \cdot \mu_{\rm B} \cdot B \cdot S + \Sigma a \cdot S \cdot I$, where the weighted summation is over all naturally occurring vanadium isotopes; the other parameters have their usual meanings. A satisfactory fit was achieved using a Lorentzian lineshape with molecular tumbling accommodated by the isotropic liquids model

given by $\sigma_v = a + bM_I + cM_I^2 + dM_I^{3.6}$ Randomly orientated EPR spectra were simulated following the spin Hamiltonian $\hat{H} = \mu_{\rm B} \cdot \mathbf{g} \cdot \mathbf{B} \cdot \mathbf{S} + \Sigma \mathbf{S} \cdot \mathbf{A} \cdot \mathbf{I}$, where \mathbf{g} and \mathbf{A} are the 3 × 3 electron Zeeman and magnetic hyperfine interaction matrices, respectively. A Gaussian lineshape and distribution of *g*- and *A*-values (strain) were employed to account for the linewidth variation.

For the unpaired electron located in an antibonding molecular orbital (ψ) is defined in eq S1,

$$\psi = \beta |3d\rangle - \beta' |\varphi_L\rangle \tag{S1}$$

where φ_L is an MO comprising the symmetry-adapted linear combination of S 3s and 3p orbitals, and β , β' are mixing coefficients that represent covalency.^{7,8} The mixing coefficient β , is calculated by evaluating the vanadium hyperfine structure tensor *A*, by means of eq. S2 and S3,

$$A_{\parallel} = -K - (\frac{4}{7})\beta^2 P + (g_{\parallel} - g_{\rm e})P + (\frac{3}{7})(g_{\perp} - g_{\rm e})P \tag{S2}$$

$$A_{\perp} = -K + {\binom{2}{7}}\beta^2 P + {\binom{11}{14}}(g_{\perp} - g_e)P$$
(S3)

where *K* is the isotropic hyperfine contact term arising from the polarisation of inner s electrons by the unpaired spin in the d orbital; $P = g_e g_n \beta_e \beta_n \langle r^{-3} \rangle_{3d}$, where g_e is the *g*-value of the free electron ($g_e = 2.0023$), g_n is the nuclear *g*-value, and β_e and β_n are the Bohr and nuclear magnetons, respectively. It has been found that the sign of *K* is the same as g_n , and $P = 128 \times 10^{-4}$ cm⁻¹ is used here.⁸

Calculations. The program package ORCA was used for density functional theory (DFT) calculations.⁹ Geometry optimisation employed the TPSS meta-generalised gradient approximation functional;¹⁰ single-point calculations on optimised and crystallographic coordinates used the one-parameter hybrid variant TPSSh.¹¹ The scalar relativistically recontracted ZORA-def2-TZVP basis set as used for all atoms.¹² Auxiliary basis sets used to expand the electron density in the calculations were chosen to match the orbital basis. The RIJCOSX algorithm was used to speed the calculation of Hartree–Fock exchange.¹³ Increased integration accuracy was applied to the vanadium and sulfur (grid = 7) atoms. Calculations included the zeroth-order regular approximation (ZORA) for relativistic effects¹⁴ as implemented by van Wüllen.¹⁵ The self-consistent field calculations were tightly converged (1 × 10⁻⁸ E_h in energy, 1 × 10⁻⁷ E_h in the charge density, and 1 × 10⁻⁷ in the maximum element of the DIIS¹⁶ error vector). The geometry was converged with the following convergence criteria: change in energy <10⁻⁵ E_h , average force

 $<5 \times 10^{-4} E_{\rm h}$ Bohr⁻¹, and the maximum force $10^{-4} E_{\rm h}$ Bohr⁻¹. The geometry search for all complexes was carried out in redundant internal coordinates without imposing geometry constraints.

The broken symmetry (BS) approach to describe computational results for **1-Et** and **1-Bu**.¹⁷ The system divided into two fragments. The notation BS(m,n) refers then to a broken symmetry state with m unpaired α -spin electrons essentially on fragment 1 and n unpaired β -spin electrons localized on fragment 2. In each case, fragments 1 and 2 correspond to the two metal ions. In this notation the standard high spin, open-shell solution is written as BS(m+n,0). The BS(m,n) notation refers to the initial guess to the wavefunction. The variational process does, however, have the freedom to converge to a solution of the form BS(m-n,0) in which effectively the n β -spin electrons pair up with $n < m \alpha$ -spin electrons on the partner fragment. Such a solution is then a standard $M_5 \approx$ (m-n)/2 spin-unrestricted Kohn-Sham solution. As explained elsewhere,¹⁸ the nature of the solution is investigated from the corresponding orbital transformation (COT) which, from the corresponding orbital overlaps, displays whether the system should be described as a spin-coupled or a closed-shell solution. The exchange coupling constants J were obtained from broken symmetry solution using eq. S4,¹⁹ and assuming the spin-Hamiltonian eq. S5 is valid,

$$J = \frac{E_{HS} - E_{BS}}{\langle \hat{S}^2 \rangle_{HS} - \langle \hat{S}^2 \rangle_{BS}}$$
(S4)

$$\hat{H} = -2J\hat{S}_{\rm A}\cdot\hat{S}_{\rm B} \tag{S5}$$

where E_{BS} is the energy of the broken symmetry solution, E_{HS} is the energy of the high spin state, $\langle \hat{S}^2 \rangle_{HS}$ is the expectation value of \hat{S}^2 operator for the high spin state, $\langle \hat{S}^2 \rangle_{BS}$ is the expectation value of \hat{S}^2 operator for the broken symmetry solution, and $\langle \hat{S}^2 \rangle_{HS}$ is the expectation value of \hat{S}_{A}^2 and \hat{S}_{B}^2 are local spin operators.

The property calculations at the optimized geometries were done using dichloromethane as solvent.²⁰ In this case the same basis sets were used but with enhanced integration accuracy (SPECIALGRIDINTACC 10) for the vanadium and sulfur atoms. Calculation of the **g**-matrix included a larger the integration grid (Grid7) and fully decontracted basis sets.²¹ Vanadium hyperfine coupling values (**A**-matrix) were computed using Sauer's aug-ccpVTZ-J basis set.²² Electronic spectra were reproduced using the time-dependent (TD)-DFT protocol. The first 80 states were calculated, and the maximum dimension of the expansion space in the Davidson procedure (MAXDIM) was set to 800. The full width at half maximum (FWHM) was set to 3000 cm⁻¹. Molecular orbitals and spin density maps were visualised via the programme Molekel.²³

The multireference ground state composition of **1-Et** and **2-Et** was examined using the state-averaged complete active space self-consistent field (SA-CASSCF) method²⁴ with the def2-TZVP basis set for all atoms. A CASSCF(2,10) (two electrons in the ten active metal d-based orbitals) and CASSCF(1,10) was performed for **1-Et** and **2-Et**, respectively. For **2-Et**, the calculation was averaged over 10 doublet states; for **1-Et**, the calculation was averaged over 10 doublet states; for **1-Et**, the calculation was averaged over 10 triplet and 10 singlet states. The NEVPT2 calculations were performed on each reference space.²⁵ In the case of the CASSCF/NEVPT2 method, the matrix elements are obtained with the CASSCF wavefunctions and only the diagonal energies contain the dynamic correlation brought in by the NEVPT2 procedure.

Geometry	CShM ^a
octagon	36.2
heptagonal pyramid	21.5
hexagonal bipyramid	16.1
cube	12.1
square antiprism	5.5
triangular dodecahedron	3.7
Johnson gyrobifastigium (J26)	13.1
Johnson elongated triangular bipyramid (J14)	27.6
biaugmented triangular prism (J50)	4.6
biaugmented trigonal prism	4.4
snub diphenoid (J84)	6.3
triakis tetrahedron	12.6
elongated trigonal bipyramid	24.3

 $\label{eq:stables} \textbf{Table S1} \quad \text{Continuous shape measures for eight-coordinate geometry of VS}_8 \text{ unit}$

^a Ref. 26



Fig. S1 Overlay of the electronic spectra of 1-Bu and 2-Bu recorded in dichloromethane solution at ambient temperature.



Fig. S2 X-band EPR spectrum of **2-Et** in CH₂Cl₂ at 293 K (experimental conditions: frequency, 9.4039 GHz; modulation, 0.8 mT; power, 0.063 mW). Experimental data are shown by the black line; simulation depicted by the red trace: $g_{iso} = 2.0066$; $A_{iso} = 29.6 \times 10^{-4}$ cm⁻¹.

Complex	g _{iso}	g_x	g_y	g_z	$\langle g \rangle^{\mathrm{b}}$	$A_{ m iso}$	A_{xx}	A_{yy}	A_{zz}	$\langle A \rangle^{c}$	ref
2-Et	2.0066	2.0021	2.0069	2.0152	2.0081	-29.6	-18.0	-18.0	-52.9	-29.6	this work
$[V_2(\mu\text{-}S_2)_2(^n\text{Bu}_2\text{dtc})_4]^+$	2.004	2.004	2.001	2.011	2.005	-29.1	-10.3	-24.3	-53.1	-29.2	27
$[VO(Et_2dtc)_2]$	1.9794	1.9871	1.9871	1.9640	1.9794	-83.5	-51.2	-51.2	-147.6	-83.3	28
$[V(Et_2dtc)_4]$	1.9746	1.9831	1.9831	1.9576	1.9746	-66.6	-37.8	-37.8	-123.5	-66.4	28
$[V(S_2CNR)_4]$											
R = pyrrole	1.976	1.986	1.986	1.961	1.978	-62.7	-36.9	-36.9	-118.3	-64.0	29
R = carbazole	1.976	1.986	1.986	1.964	1.979	-63.5	-36.4	-36.4	-116.0	-62.9	29
R = indole	1.979	1.986	1.986	1.966	1.979	-63.8	-37.1	-37.1	-118.6	-64.3	29
R = indoline	1.980	1.989	1.989	1.963	1.980	-63.3	-36.9	-36.9	-118.1	-64.0	29
$[V(S_2CMe)_4]$	1.9735	1.9817	1.9778	1.9610	1.9735	-61.2	-31.3	-43.0	-107.3	-60.5	30
$[V(S_2CPh)_4]$	1.976	1.9816	1.9814	1.9632	1.9754	-59.0	-34.45	-34.50	-104.9	-57.95	30
$[V(S_2CC_5H_3(NHR)_4]$											
$\mathbf{R} = \mathbf{H}$	1.971	1.985	1.985	1.931	1.967	-101.0	-74.1	-74.1	-174.9	-107.7	31
$\mathbf{R} = \mathbf{E}\mathbf{t}$	1.969	1.983	1.983	1.932	1.966	-102.8	-70.6	-70.6	-169.6	-103.6	31
$\mathbf{R} = {}^{n}\mathbf{P}\mathbf{r}$	1.968	1.984	1.984	1.929	1.966	-104.9	-71.8	-71.8	-173.8	-105.8	31
$R = {}^{n}Bu$	1.969	1.986	1.986	1.933	1.968	-105.0	-73.0	-73.0	-176.0	-107.3	31

Table S2 Comparison of g- and A-values^a for related compounds

^a Value in × 10⁻⁴ cm⁻¹; the sign is negative owing to the dominant Fermi-contact contribution. ^b $\langle g \rangle = (g_x + g_y + g_z)/3$. ^c $\langle A \rangle = (A_{xx} + A_{yy} + A_{zz})/3$.

Table S3 Comparison of β^2 and *K*-values^a for related compounds

	CN ^b	Geometry	Symmetry	Ground State	d-orbital	β^2	<i>K</i> / 10 ⁻⁴ cm ⁻¹	$A_{\rm iso}$ / 10 ⁻⁴ cm ⁻¹
2-Et	8	bicapped trigonal prism	C_{2h}	ag	$x^2 - y^2$ °	0.66	30.4	-29.6
[V(Et ₂ dtc) ₄]	8	dodecahedral	D_{2d}	b_1	x^2-y^2	0.74	62.8	-66.6
$[VO(Et_2dtc)_2]$	5	square pyramidal	C_{2v}	a_1	x^2-y^2	0.84	80.4	-83.5
$[V(S_2C_2(CN)_2)_3]^{2-d}$	6	octahedral	D_3	a_1	z^2	0.65	55.2	-57.7

^a Calculated using eqs S2 and S3. ^b CN = coordination number. ^c Non-standard axis alignment with *x*-axis along V…V vector. ^d Values taken from ref. 32.



Fig. S3 Overlay of the normalised sulfur K-edge spectra of 1-Bu and 2-Bu

$Table \ S4 \quad Geometry-optimised \ coordinates \ of \ 1-Et$

V	1.08549101696758	10.76263712427527	-0.63527047216000
V	-1.08548098045553	9.72936539367769	0.63528079240439
S	1.10401653717003	8.73490121131349	0.65816126801282
S	0.90135572078665	10.44682341799428	1.70236427993384
S	1.61625007618664	13.08320128130520	0.15754577535688
S	3.53130841789361	10.99634238101503	-0.21333471014536
S	1.56229983538613	11.93212363082761	-2.78328556226583
S	2.07072998773385	9.16485298909911	-2.29331924370009
S	-1.10400638727720	11.75710191997463	-0.65815001226405
S	-0.90134592300606	10.04518011507447	-1.70235374741192
S	-1.61623858896166	7.40880162915587	-0.15753733465222
S	-3.53129829932384	9.49565919250676	0.21334394903234
S	-1.56229068449811	8.55987787833236	2.78329526421870
S	-2.07072239092037	11.32714822018429	2.29332892941278
Ν	4.24049082110737	13.44585259020327	0.66986387951042
Ν	2.65033659085337	10.32488956516683	-4.65893557062272
С	3.26218872319715	12.61330256504408	0.26407860000138
С	5.63234166845869	12.99649812381533	0.72347459889451
С	6.36644230959847	13.22667820782974	-0.59811118440917
С	3.94575904598671	14.82451705711189	1.06209535516064
С	3.64529584848974	14.95126625612815	2.55520147278616
С	2.15131907280462	10.46045667130097	-3.41456691108632
С	2.68271846288569	11.45923640721417	-5.58247516650369
C	1,41006033944643	11.55721646912871	-6.42358231888937
C	3.19561827020224	9.04644880636680	-5.11696109972369
C	4.68989681806690	8.91664406119643	-4.82108434970441
Н	6.12229943992572	13.53476095630728	1,54858416943585
Н	5.62893427607103	11.92669873810012	0.97801761174463
Н	7.40661567905639	12.87944657101047	-0.52238872679002
Н	6.37898961303855	14.29281430440508	-0.86639694651643
Н	5.87075350211857	12.66945889392521	-1.40386887667519
н	4.81309527903677	15.44070320584197	0.78053438039345
н	3.08778478533734	15.16827437084485	0.46620606183491
н	3.44146298108511	15,99952864035125	2.81561907708480
н	4.49426827715865	14.60702297535213	3.16370688643029
н	2.76317142024099	14.35122424843014	2.81552254702066
Н	3.57166235893854	11.34140699375133	-6.22023272556894
Н	2.81908180450116	12.37355332021327	-4,98653851616568
н	1,47177708256052	12,41306872546005	-7.11098646490171
н	0.53793160851007	11,69823197840666	-5.77117561669417
н	1.25536687931372	10.64781602575538	-7.02276920679471
н	2.63437635656338	8.24352941624239	-4.61762729347150
Н	2,99695000918755	8.96788923752876	-6.19658813681728
н	5.06734870883704	7,94693727767654	-5.17613604053956
н	4.86553297434332	8,98371900063136	-3.73886450740571
н	5.26621177153997	9,70933504264607	-5.31948123625157
С	-3.26217711066031	7.87870025157560	-0.26407286618359
Ċ	-2.15131668289363	10.03154300147631	3.41457442155574
N	-4.24047795924849	7.04615106155872	-0.66986283956509
N	-2.65034499210323	10.16710741450536	4.65893899364143
С	-5.63232866022725	7.49550547021587	-0.72347667365324
C	-3.94574428523042	5.66748829848474	-1.06209893553532
C.	-2.68273484848187	9.03275891116868	5.58247621770159
C.	-3.19563615900845	11.44554580975830	5.11695992273467
C	-6.36643417495428	7.26532037556962	0.59810554494780
H	-6.12228343127999	6.95724560661114	-1.54858995496942
Н	-5.62892048972576	8.56530578380130	-0.97801577844468
С	-3.64527678383599	5.54074550190491	-2.55520471754729
H	-4.81308088964859	5.05130039409766	-0.78054301663718
Н	-3.08777148810216	5.32372913057270	-0.46620861150365
С	-1.41008579116635	8.93477934703396	6.42359709101491
H	-3.57168583263014	9.15058584765435	6.22022441493981
Н	-2.81909051566878	8.11844282110568	4.98653651823354

С	-4.68991133127121	11.57534623424978	4.82106389773230
Н	-2.63439036644291	12.24846784358472	4.61763473387932
Н	-2.99698209096678	11.52410414129296	6.19658965631920
Н	-7.40660743369839	7.61255171401205	0.52238032780594
Н	-6.37898192197544	6.19918334380030	0.86638751506216
Н	-5.87074886496825	7.82253708471556	1.40386718235663
Н	-3.44144143098572	4.49248451705618	-2.81562595459633
Н	-4.49424810290638	5.88498979942080	-3.16371110677394
Н	-2.76315264066170	6.14079007532873	-2.81552078250331
Н	-1.47180863031381	8.07892549031038	7.11099867036090
Н	-0.53794953268509	8.79376681865539	5.77119982137428
Н	-1.25540051862924	9.84417874347561	7.02278763379606
Н	-5.06737098375037	12.54505133502997	5.17611193665210
Н	-4.86553299129397	11.50827205319807	3.73884161001496
Н	-5.26623018072849	10.78265269065187	5.31945220725152

Table S5 Geometry-optimised coordinates of 1-Bu

V	-3.51822499523900	6.00573936451236	2.60978607069722
V	-3.51760322645555	8.72715529195405	2.61007856710278
ċ	F 10007002004710	7 27510522452700	2 7200070070040
S	-5.1222/993684/12	/.3/518533453/88	3./3989/80/92049
S	-3.27976347819861	7.35184928196260	4.55564625559773
C	-3 98550920796735	1 16868115813839	1 22100727770217
5	5.90550920790755	4.10000140040009	4.22499727779247
S	-1.37505941203927	5.06661506999167	3.49543736358090
S	-5 61086828158417	9 65340863838636	1 62111536178980
5	5.01000020150417	5.00040000000000	1.02111000170000
S	-4.40752438599809	10.56492745471863	4.03121799628531
N	-1.75277118548002	3.06433069149951	5.26550853597827
	C 7040140C011CC2	11 5100001000000	2 170000000077140
IN	-6./9401486911663	11.51998218263594	3.1/9998298//140
С	-2.30353120653598	3.98073950329457	4.44517626461251
C	-2 600/1010/70502	2 22202620701021	6 1021/2202107/2
C	-2.00941010470302	2.22202020/01031	0.10314220310/43
С	-3.13242990431223	2.93354710120558	7.36256703101689
C	-1 99250399559524	3 44039869556763	8 24767253903861
~	1.99200399009021	5.11055005550705	0.21/0/2000001
С	-4.06599683865694	1.98/61495116839	8.12169132610/84
С	-0.29981078339501	2.88640072607231	5.30118771365698
C	0 0401070017000	2 00011505007200	4 101000007700
C	0.24913/201/6229	2.06611565997309	4.12102320907780
С	-0.37971598581942	0.67323048049537	4.04982544063449
C	1 77438559935955	1 99913656764488	4 22071768969636
C	1.77430339933933	1.99913030704400	4.220/1/00909050
С	-5.74365819546940	10.69642637667582	2.97713967250172
С	-7.96134188454593	11,49384285927259	2.30175590679836
^o	0 10040020214700	10 75150011001220	2 00024470542227
C	-9.10049030314/00	10./2120311001220	2.909244/034323/
С	-8.82835873422082	9.30156153962774	3.26543908556412
C	-10 34645957093761	10 83478224233314	1 9356/655188162
0	10.34043337033701	10.034/0224233314	1.95504055100102
С	-6.83126160776212	12.43482906919746	4.318/4/6138/1/4
С	-6.52453704772576	13.89408698786054	3.94159212219916
õ	E 1070C01E7070CC	14 0400000700244	2 22524150271507
C	-5.12/26015/9/966	14.04068058/68344	3.335341503/159/
С	-6.69894314244388	14.78049859935904	5.17844923545015
ц	-3 16191528712610	1 88407548010266	5 49203323100751
11	5.40194526742040	1.0040/040010200	5.49205525100751
Н	-2.02492295854513	1.33359850988638	6.38409403547933
Н	-3.71856678428849	3,80066066371959	7.01722703387156
	1 24552770075040	4 14405272006600	7 70(27727021(44
п	-1.34333770073040	4.14495572006500	1.1003/13/031044
Η	-1.36929343461447	2.60642434331945	8.61015753457689
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С	-4.42292617099651	2.21622421545935	-0.87842872457752
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С	0.92978585695070	11.48796784917364	2.92354557918458
С	-0.19959242053143	12.43464417913101	0.90876917869392
C	-3.90399344489978	2,92660356413263	-2.14018736781531
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Η	-3.50911918227537	1.08154587918496	-3.23829401863127
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Н	-7.03696702794973	0.10855501899958	2.04440063599890
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Η	-0.50777353989099	15.83837959193080	0.30412491297998
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Table S6Geometry-optimised coordinates of 2-Et

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S	-2.16903321281488	11.31421438129195	2.31393809275293
Ν	4.29459274788943	13.49131407433796	0.70951716025006
Ν	2.67210519805991	10.30662161688893	-4.72368926673985
С	3.32618719472864	12.67321120961524	0.28400944993752
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С	1.42909972653176	11.48654191759959	-6.54608714847074
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н	5 67570058183136	11 95143081900398	1 01041676697799
ц Ц	7 5581/586171029	13 015/0808/3011/	-0 26688999097379
11 U	6 55247904227261	14 43740700946455	-0 61670730425209
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п	0.12032307094949	16 40002661020642	-1.30302340020302
H	4.8/245/9/080625	15.49003551939542	0.74725219799174
Н	3.14111645106947	15.22150757025593	0.49112849193891
Н	3.5/982895462593	16.13490386019983	2./890/80463821/
Н	4.62103425806010	14.74914268095059	3.1/252326/618/3
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Η	3.58643359070605	11.33480163077260	-6.28389584190742
Η	2.78259149453628	12.36261770850037	-5.08618916070170
Н	1.50018533658586	12.33382949170642	-7.24234349098425
Η	0.52707686407493	11.62054205971156	-5.93437682658920
Η	1.31365315421849	10.57107877722976	-7.14388044493825
Η	2.68518097513202	8.21889933285932	-4.64079868713762
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C	-3./38/3649382589	5.419208/8349039	-2.5/244580626964
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Н	-7.55813239648264	7.47650404181180	0.26694103477098
Н	-6.55345854201779	6.05459999058469	0.61676071818692
Н	-6.12029980903112	7.64253034029566	1.30305533138277
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Table S7 Geometry-optimised coordinates of 2-Bu

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Н	1.22541975195832	7.46183237467335	2.00382615903908
Η	3.29761555033214	6.90734792571528	0.71245728169954
Н	2.91442693704314	5.63182544263919	1.88855179639695
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Н	1.36121344532285	1.93665520086254	4.4//804696625//
Η	3.04348594589860	7.36824315954181	4.48284169580204
Η	1.72172836189019	6.23948451439773	4.10091695353259

	1-Et		1-F	Bu	2-Et	2-E	Bu
	exptl ^a	calcd	exptl ^b	calcd	calcd	exptl ^c	calcd
V1–S1	2.446(5)	2.405	2.402(2)	2.395	2.400	2.373(1)	2.374
V1-S2	2.393(5)	2.366	2.402(1)	2.383	2.446	2.447(2)	2.441
V1-S3	2.426(6)	2.405	2.402(2)	2.395	2.401	2.424(2)	2.410
V1–S4	2.386(6)	2.367	2.402(1)	2.383	2.444	2.401(2)	2.408
V1–S5	2.518(6)	2.505	2.492(1)	2.494	2.507	2.479(2)	2.470
V1–S6	2.491(6)	2.492	2.501(2)	2.488	2.453	2.413(1)	2.426
V1–S7	2.495(5)	2.493	2.492(1)	2.494	2.510	2.435(2)	2.439
V1–S8	2.496(6)	2.509	2.501(2)	2.488	2.454	2.505(2)	2.484
V2–S1	2.426(6)	2.405	2.413(1)	2.379	2.401	2.384(2)	2.360
V2–S2	2.386(6)	2.367	2.403(2)	2.392	2.444	2.449(2)	2.441
V2S3	2.446(5)	2.405	2.413(1)	2.378	2.400	2.422(1)	2.409
V2–S4	2.393(5)	2.366	2.403(2)	2.392	2.446	2.403(2)	2.401
V2–S9	2.518(6)	2.509	2.500(2)	2.502	2.507	2.473(2)	2.494
V2–S10	2.491(6)	2.493	2.487(1)	2.490	2.453	2.443(1)	2.445
V2–S11	2.495(5)	2.492	2.500(2)	2.502	2.510	2.436(2)	2.423
V2-S12	2.496(6)	2.505	2.487(1)	2.490	2.454	2.467(2)	2.470
$V1\cdots V2$	2.900(6)	2.719	2.851(1)	2.721	2.958	2.935(1)	2.885
S1–S2	2.00(1)	2.015	1.997(4)	2.015	2.025	2.007(2)	2.023
S3–S4	2.00(1)	2.015	1.997(4)	2.015	2.025	2.003(2)	2.020
S5-C1	1.737(2)	1.715	1.703(8)	1.716	1.722	1.715(6)	1.719
S6-C1	1.716(2)	1.706	1.705(6)	1.707	1.722	1.718(7)	1.719
C1–N1	1.326(2)	1.348	1.343(5)	1.350	1.337	1.321(5)	1.335
S7–C2	1.701(2)	1.715	1.703(8)	1.716	1.722	1.723(6)	1.719
S8–C2	1.717(2)	1.707	1.705(6)	1.707	1.723	1.713(7)	1.719
C2-N2	1.304(2)	1.347	1.343(5)	1.350	1.337	1.310(9)	1.337
S9–C3	1.737(2)	1.715	1.721(5)	1.716	1.722	1.718(5)	1.718
S10–C3	1.716(2)	1.706	1.699(5)	1.707	1.722	1.712(7)	1.716
C3–N3	1.326(2)	1.348	1.325(9)	1.348	1.337	1.338(6)	1.335
S11–C4	1.701(2)	1.715	1.721(5)	1.716	1.722	1.726(7)	1.718
S12–C4	1.717(2)	1.707	1.699(5)	1.707	1.723	1.718(6)	1.718
C4–N4	1.304(2)	1.347	1.325(9)	1.348	1.337	1.303(9)	1.336

Table S8 Comparison of experimental and calculated bond distances (Å)

^a Data taken from ref. 1; ^b Data taken from ref. 2; ^c Data taken from ref. 3.

	1-Et		1-Bu		2-Et	2-Bu	
	exptl ^a	calcd	exptl ^b	calcd	calcd	exptl ^c	calcd
S1-V1-S2	49.2(2)	49.97	49.14(8)	49.90	49.39	49.18(5)	49.66
S3-V1-S4	48.9(2)	49.96	49.14(8)	49.90	49.39	49.05(5)	49.59
S5-V1-S6	69.3(2)	69.72	69.27(7)	69.63	70.10	71.04(5)	71.07
S7–V1–S8	69.2(2)	69.65	69.27(7)	69.63	70.10	70.00(5)	70.51
S1-V2-S2	48.9(2)	49.96	49.00(9)	49.97	49.39	49.05(5)	49.80
S3-V2-S4	49.2(2)	49.97	49.00(9)	49.97	49.39	49.06(5)	49.67
S9-V2-S10	69.3(2)	69.72	69.26(7)	69.78	70.10	70.24(5)	70.56
S11-V2-S12	69.2(2)	69.65	69.26(7)	69.77	70.10	70.72(5)	71.11
V1-S1-V2	73.1(2)	70.14	72.61(6)	69.52	76.06	76.20(5)	75.10
V1-S2-V2	74.8(2)	68.65	72.79(6)	69.49	74.45	73.67(5)	72.45
V1-S3-V2	73.1(2)	68.65	72.61(6)	69.53	74.45	74.55(4)	73.55
V1-S4-V2	74.8(2)	70.14	72.79(6)	69.47	76.06	75.30(4)	73.72
V1-S5-C1	88.4(2)	88.24	88.70(7)	88.70	88.23	87.50(5)	87.84
V1-S6-C1	89.7(2)	88.86	89.05(8)	89.10	90.04	89.58(5)	89.28
S5-C1-S6	111.1(2)	113.17	112.74(8)	112.38	111.65	111.81(5)	111.77
V1-S7-C2	89.5(2)	88.24	88.70(7)	88.70	88.23	90.52(5)	89.41
V1-S8-C2	89.1(2)	88.94	89.05(8)	89.09	90.04	87.97(5)	87.91
S7–C2–S8	112.1(2)	113.17	112.74(8)	112.37	111.66	111.11(5)	111.51
V2-S9-C3	88.4(2)	88.86	88.96(7)	88.28	88.26	88.58(5)	87.63
V2-S10-C3	89.7(2)	88.24	89.89(8)	88.85	90.00	89.72(5)	89.30
S9-C3-S10	111.1(2)	113.17	111.89(9)	113.10	111.65	111.08(5)	112.31
V2-S11-C4	89.5(2)	88.24	88.96(7)	88.28	88.23	89.58(5)	89.32
V2-S12-C4	89.1(2)	88.94	89.89(8)	88.85	90.04	88.73(5)	87.79
S11-C4-S12	112.1(2)	113.17	111.89(9)	113.09	111.66	110.98(5)	111.74

 Table S9
 Comparison of experimental and calculated bond angles(°)

^a Data taken from ref. 1; ^b Data taken from ref. 2; ^c Data taken from ref. 3.



Fig. S4 MO energy level scheme of frontier Kohn-Sham orbitals for **1-Et** with $C_{2\nu}$ symmetry labels. α -spin and β -spin magnetic orbitals (SOMOs) are highlighted red and blue, respectively, with overlap integral (S) specified.



Fig. S5 MO energy level scheme of frontier Kohn-Sham orbitals for **1-Bu** with $C_{2\nu}$ symmetry labels. α -spin and β -spin magnetic orbitals (SOMOs) are highlighted red and blue, respectively, with overlap integral (S) specified.



Fig. S6 MO energy level scheme of frontier Kohn-Sham orbitals for **2-Et** with C_{2h} symmetry labels. The magnetic orbital (SOMO) is highlighted red.



Fig. S7 MO energy level scheme of frontier Kohn-Sham orbitals for **2-Bu** with C_{2h} symmetry labels. The magnetic orbital (SOMO) is highlighted red.



Fig. S8 Mulliken spin density map for **1-Et** (red: α -spin; yellow: β -spin)



Fig. S9 Mulliken spin density map for **1-Bu** (red: α -spin; yellow: β -spin)



Fig. S10 Mulliken spin density map for **2-Et** (red: α -spin; yellow: β -spin)



Fig. S11 Mulliken spin density map for **2-Bu** (red: α -spin; yellow: β -spin)



Fig. S12 Corresponding orbitals of $[V_2(\mu-S_2)_2(CS_3)_4]^{4-}$ determined from BS(1,1) DFT calculations on the crystallographic coordinates of the complex, with $C_{2\nu}$ symmetry labels. The orbital overlap integral (S) is specified.



Fig. S13 Visualisation of the occupied MOs of **1-Et** from the SA-CASSCF/NEVPT2 CAS(2,10) calculation. The singlet ground state is comprised of an admixture 60.6% of a_g and 39.3% a_u . This configurational interaction estimates a 20% V–V σ -bond contribution to the ground state of the {V^{IV}₂(µ-S₂)₂} core unit. The triplet excited state, which comprises 1:1 admixture of these singly-occupied orbitals, $(a_g)^1(a_u)^1$, resides 1204.5 cm⁻¹ above the ground state.



Fig. S14 Visualisation of the singly-occupied MOs of that comprise the ground state of **2-Et** from the SA-CASSCF/NEVPT2 CAS(1,10) calculation. The remaining 10% comes from an admixture of other d orbitals which results in the slightly deviation of these MOs from true C_{2h} symmetry.



Fig. S15 Visualisation of the unoccupied MOs of that comprise the first excited state of **2-Et** from the SA-CASSCF/NEVPT2 CAS(1,10) calculation, which define the IVCT transition.



Fig. S16 Overlay of the experimental and TD-DFT calculated electronic spectrum for **1-Et**. Transition assignment is detailed in Table S10.



Fig. S17 Overlay of the experimental and TD-DFT calculated electronic spectrum for **2-Et**. Transition assignment is detailed in Table S11.



 Table S10
 Assignment of electronic transitions in 1-Et

^{*a*} Recorded in CH_2Cl_2 solution at ambient temperature, energy in cm^{-1} ; sh = shoulder. ^{*b*} From TPSSh/ZORA-def2-TZVP+CPCM(CH₂Cl₂) TD-DFT calculations.

Exptl ^a	Calcd ^b	Transition
9120	9157	$a_g \rightarrow a_u (IVCT)$
	11779	$a_g \rightarrow a_u (LMCT)$
13400	14295	$a_u \rightarrow a_g (LMCT) \longrightarrow $
14250	15836	$a_u \rightarrow a_g (LMCT)$

 Table S11
 Assignment of electronic transitions in 2-Et



^{*a*} Recorded in CH₂Cl₂ solution at ambient temperature, energy in cm⁻¹; sh = shoulder. ^{*b*} From TPSSh/ZORA-def2-TZVP+CPCM(CH₂Cl₂) TD-DFT calculations.



Fig. S18 Orientation of the principal *g*-values superimposed on the optimized structure of **2-Et**: (a) the view along g_z which is orthogonal to the V····V (defined as the *x*-axis in this non-standard alignment of the molecular axes in C_{2h} symmetry) but parallel to the S–S bonds on the bridging disulfide ligands; and (b) the view along the V····V bond showing g_z parallel and g_y orthogonal to the S–S bonds of the bridging disulfide ligands.

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