

Supplementary Information

for

Effects of Geometrical Isomerism on Emissive Behaviour of Heteroleptic Cyclometalated Ir(III) Complexes

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S1. Chromatographic separation of *cis*-(N,N) and *trans*-(N,N) isomers of

[Ir(bzq)₂(PBO)]

The chromatogram for separating *cis*-(N,N) and *trans*-(N,N) isomers of [Ir(bzq)₂(PBO)] on a HPLC silica gel column is shown below.

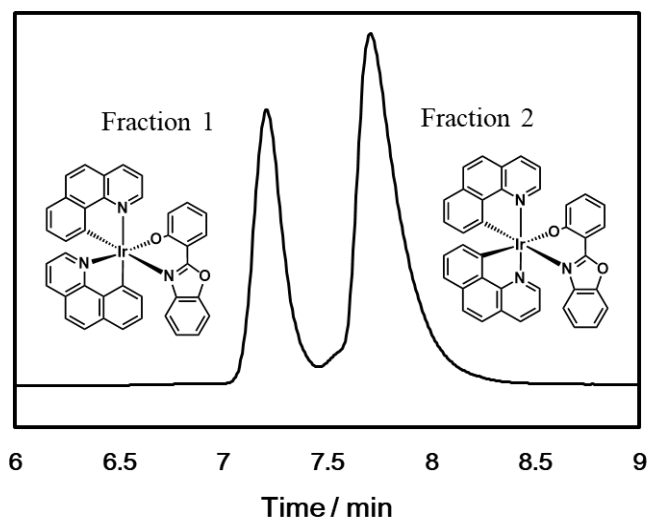


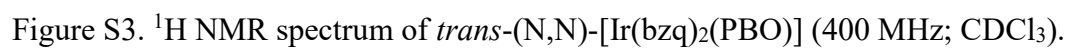
Figure S1. The chromatogram of [Ir(bzq)₂(PBO)] when it was upon the subjection to HPLC silica gel column. Fractions 1 and 2 correspond to *cis*-(N,N) and *trans*-(N,N) isomers, respectively, as described in the text. An eluting solvent was 10:1 (v/v) CH₂Cl₂ (MeOH 0.01%) and hexane. A flow rate was 2.5 mL min⁻¹ and the monitoring wavelength was 525 nm.

S2. ¹H NMR, ¹³C NMR, MS data and elemental analysis

[Ir(bzq)₂Cl]₂: ¹H (400 MHz; CDCl₃) 5.98 (2H, d, *J*=7.3 Hz), 6.81 (2H, t, *J*=7.6 Hz), 7.17 (2H, dd, *J*=8.0 and 5.5 Hz), 7.21 (2H, d, *J*=8.0 Hz), 7.69 (2H, d, *J*=8.8 Hz), 7.76 (2H, d, *J*=8.8 Hz), 8.28 (2H, d, *J*=7.1 Hz), 9.36 (2H, d, *J*=4.4 Hz)

cis-(N,N)-[Ir(bzq)₂(PBO)]: ¹H (400 MHz; CDCl₃) 5.46 (1H, d, *J*=8.0 Hz), 6.48 (1H, dd, *J*=7.0 and 1.0 Hz), 6.51 (1H, ddd, *J*=8.0 and 6.8 and 1.0 Hz), 6.63 (1H, td, *J*=8.0 and 1.0 Hz), 6.68 (1H, dd, *J*=8.8 and 1.0 Hz), 6.89 (1H, dd, *J*=8.0 and 5.5 Hz), 6.99 (1H, t, *J*=8.0 Hz), 7.03 (1H, td, *J*=8.0 and 1.0 Hz), 7.11 (1H, ddd, *J*=8.8, 6.8 and 1.8 Hz), 7.30 (1H, d, *J*=8.0 Hz), 7.41 (1H, d, *J*=8.0 Hz), 7.44 (1H, dd, *J*=8.0 and 5.0 Hz), 7.48 (1H, d, *J*=8.8 Hz), 7.54 (1H, t, *J*=8.0 Hz), 7.58 (1H, td, *J*=7.8 and 1.0 Hz), 7.64 (1H, d, *J*=8.8 Hz), 7.78 (1H, dd, *J*=5.5 and 1.0 Hz), 7.80 (1H, d, *J*=8.8 Hz), 7.83 (1H, d, *J*=8.8 Hz), 7.86 (1H, dd, *J*=8.0 and 1.0 Hz), 7.93 (1H, dd, *J*=7.8 and 1.0 Hz), 7.95 (1H, dd, *J*=8.0 and 1.8 Hz), 8.24 (1H, dd, *J*=8.0 and 1.0 Hz), 8.56 (1H, dd, *J*=5.0 and 1.0 Hz); ¹³C (125 MHz; CD₂Cl₂) 110.2, 110.9, 114.8, 118.9, 120.1, 120.4, 120.9, 121.2, 121.8, 123.4, 123.7, 123.8, 124.5, 124.6, 124.8, 126.7, 127.0, 127.7, 128.5, 128.8, 129.5, 130.1, 131.3, 133.1, 133.7, 134.4, 134.5, 135.4, 135.5, 136.6, 137.1, 141.9, 147.4, 148.3, 150.7, 153.4, 155.9, 161.1, 169.2; ESI-HRMS, Calcd for [IrC₃₉H₂₄N₃O₂]⁺ 759.1498, Found 759.1492; Anal. Calcd for IrC₃₉H₂₄N₃O₂•C₆H₁₄: C, 63.96; H, 4.53; N, 4.97. Found: C, 64.08; H, 4.89; N, 4.47.

trans-(N,N)-[Ir(bzq)₂(PBO)]: ¹H (400 MHz; CDCl₃) 5.93 (1H, d, *J*=8.5 Hz), 6.16 (1H, brd, *J*=7.8 Hz), 6.39 (1H, brd, *J*=8.5 Hz), 6.48 (1H, ddd, *J*=8.0 and 6.8 and 1.0 Hz), 6.63 (1H, ddd, *J*=8.5, 7.5 and 1.0 Hz), 6.70 (1H, d, *J*=8.8 Hz), 6.99 (1H, t, *J*=8.0 Hz), 7.00 (1H, t, *J*=7.8 Hz), 7.03 (1H, ddd, *J*=8.5 and 7.5 and 1.0 Hz), 7.19 (1H, ddd, *J*=8.7 and 6.8 and 1.8 Hz), 7.27 (1H, dd, *J*=8.0 and 5.5 Hz), 7.32 (1H, d, *J*=8.0 Hz), 7.34 (1H, d, *J*=7.8 Hz), 7.39 (1H, d, *J*=8.0 Hz), 7.46 (1H, dd, *J*=7.8 and 5.5 Hz), 7.57 (1H, d, *J*=8.8 Hz), 7.61 (1H, d, *J*=8.8 Hz), 7.77 (1H, d, *J*=8.8 Hz), 7.78 (1H, d, *J*=8.8 Hz), 8.00 (1H, dd, *J*=8.0 and 1.8 Hz), 8.13 (1H, dd, *J*=8.0 and 1.0 Hz), 8.16 (1H, dd, *J*=7.8 and 1.0 Hz), 8.45 (1H, dd, *J*=5.5 and 1.0 Hz), 9.16 (1H, dd, *J*=5.5 and 1.0 Hz); ¹³C (125 MHz; CDCl₃) 109.8, 110.5, 113.7, 118.4, 118.8, 119.3, 120.8, 120.9, 123.0, 123.2, 124.2, 124.3, 125.5, 126.1, 126.7, 128.7, 128.8, 128.9, 129.4, 129.6, 130.7, 133.5, 133.8, 134.1, 135.7, 141.5, 141.9, 142.5, 144.2, 146.1, 147.6, 147.9, 148.3, 149.2, 158.3, 158.5, 160.1, 165.3, 169.5; ESI-HRMS, Calcd for [IrC₃₉H₂₄N₃O₂]⁺ 759.1498, Found 759.1492; Anal. Calcd for IrC₃₉H₂₄N₃O₂•H₂O: C, 60.30; H, 3.37; N, 5.41. Found: C, 60.67; H, 3.32; N, 5.41.



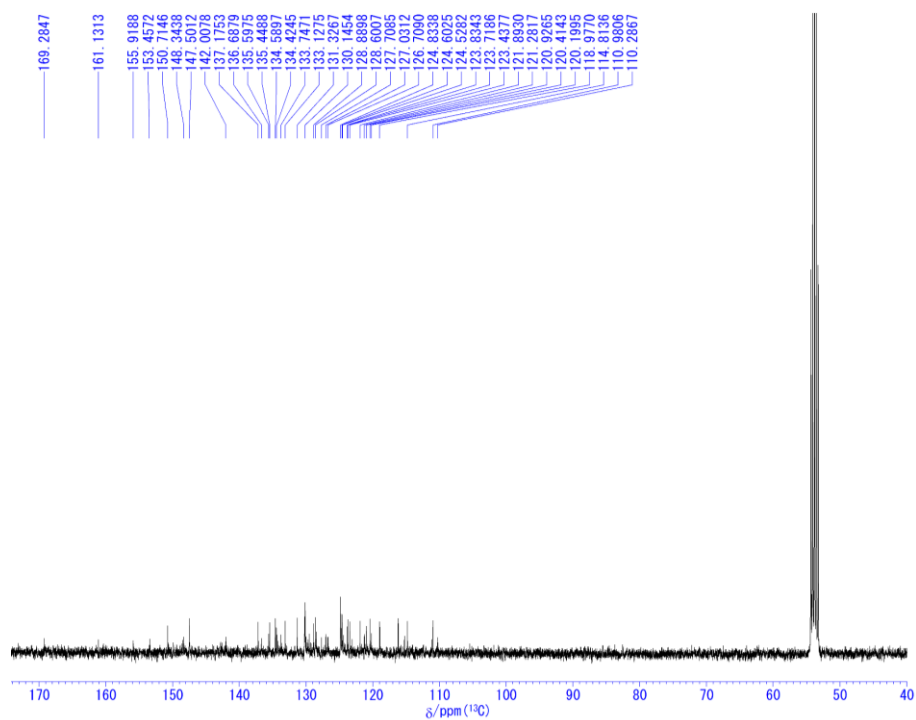


Figure S4. ¹³C NMR spectrum of *cis*-(N,N)-[Ir(bzq)₂(PBO)] (125 MHz; CD₂Cl₂).

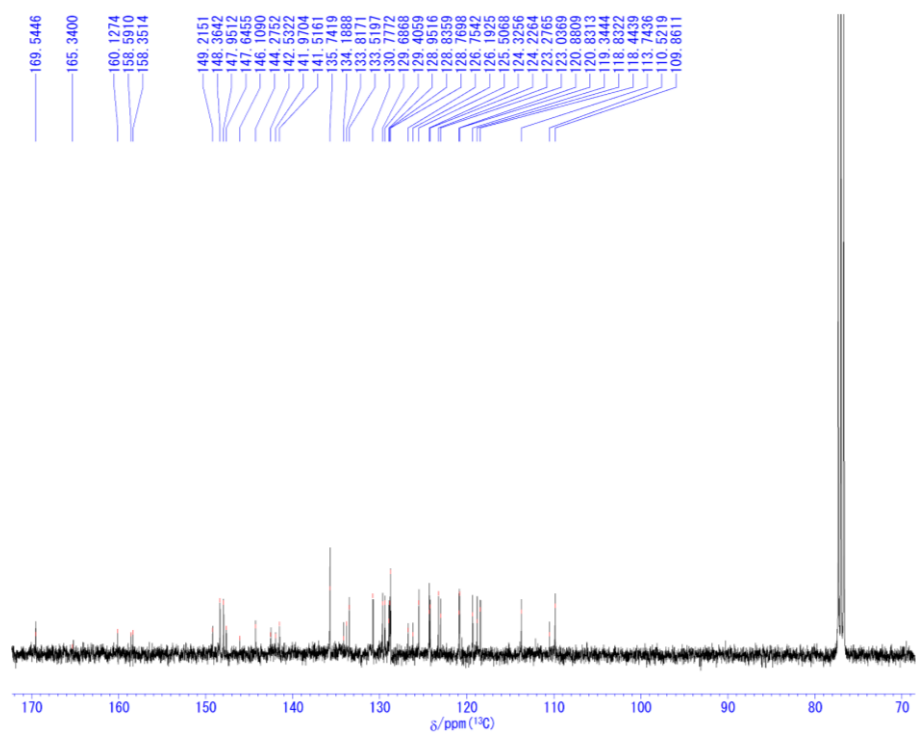


Figure S5. ¹³C NMR spectrum of *trans*-(N,N)-[Ir(bzq)₂(PBO)] (125 MHz; CDCl₃).

S3. Optical resolution of [Ir(bzq)₂(PBO)] isomers

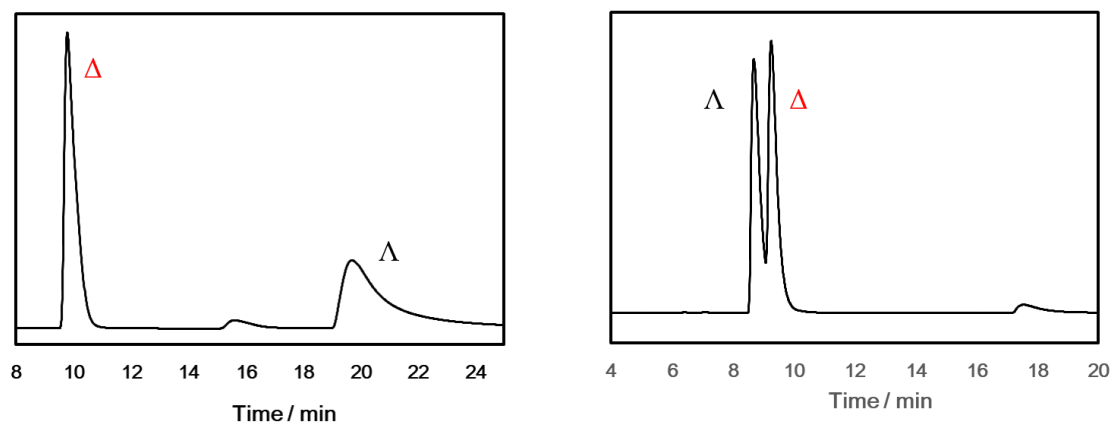


Figure S6. Chromatograms for optical resolution of [Ir(bzq)₂(PBO)] isomers: *cis*-(N,N) (left) and *trans*-(N,N) isomers (right). An eluting solvent was CH₂Cl₂. A flow rate was 0.5 mL min⁻¹. The monitoring wavelength was 350 nm. The used column was a CHIRALPACK IC (Daicel, Japan).

S4. Electronic circular dichroism spectra of enantiomeric isomers of [Ir(bzq)₂(PBO)]

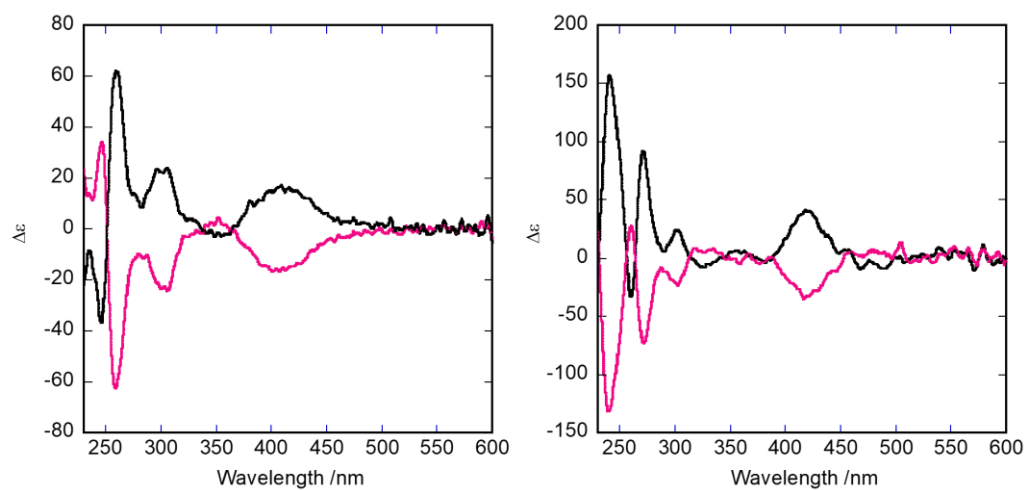


Figure S7. The electronic circular dichroism spectra of *cis*-(N,N) (left) and *trans*-(N,N) isomers (right) of [Ir(bzq)₂(PBO)]. The black and red lines correspond to Λ - and Δ -enantiomers, respectively. A solvent was CH₂Cl₂.

S5. Single crystal X-ray analyses of racemic and enantiomeric isomers of [Ir(bzq)₂(PBO)]

Single crystal X-ray analyses were performed for both the racemic and enantiomeric forms of [Ir(bzq)₂(PBO)] isomers. The preparation of single crystals, crystal data and molecular structures are given below.

Single crystals of Δ -*cis*-(N,N)-[Ir(bzq)₂(PBO)] and Λ -*cis*-(N,N)-[Ir(bzq)₂(PBO)] were prepared by slow evaporation of a mixed solution of dichloromethane and hexane. Crystallographic data: Δ -*cis*-(N,N)-[Ir(bzq)₂(PBO)]; C₃₉H₂₄IrN₃O₂, 1.18(CH₂Cl₂), 0.32(C₆H₁₄), FW 886.55, monoclinic, *P*2₁ (#4), *a* = 9.15927(16) Å, *b* = 27.4078(5) Å, *c* = 14.7784(3) Å, $\alpha=\gamma=90^\circ$, $\beta=102.475(2)^\circ$, *V* = 3622.31(12) Å³, *Z* = 4, $\mu=3.901\text{ mm}^{-1}$. *R*₁ = 0.0395 (*I* > 2σ(*I*)), *wR*₂ = 0.0923 (all data), GOF = 1.045 (*I* > 2σ(*I*)), Flack parameter = -0.013(3) (CCDC 2063613). Λ -*cis*-(N,N)-[Ir(bzq)₂(PBO)]; C₃₉H₂₄IrN₃O₂, 1.22(CH₂Cl₂), 0.28(C₆H₁₄), FW 886.58, monoclinic, *P*2₁ (#4), *a* = 9.15128(17) Å, *b* = 27.4566(6) Å, *c* = 14.7617(3) Å, $\alpha=\gamma=90^\circ$, $\beta=102.392(2)^\circ$, *V* = 3622.66(13) Å³, *Z* = 4, $\mu=3.907\text{ mm}^{-1}$. *R*₁ = 0.0434 (*I* > 2σ(*I*)), *wR*₂ = 0.0945 (all data), GOF = 1.039 (*I* > 2σ(*I*)), Flack parameter = -0.016(4) (CCDC 2063614).

Single crystals of Λ -*trans*-(N,N)-[Ir(bzq)₂(PBO)] was prepared by slow evaporation of a mixed solution of dichloromethane and hexane. Crystallographic data: Λ -*trans*-(N,N)-[Ir(bzq)₂(PBO)]; 3(C₃₉H₂₄IrN₃O₂), 1.5(CH₂Cl₂), FW 2403.82, orthorhombic, *P*2₁2₁2₁ (#19), *a* = 10.6678(2) Å, *b* = 18.2211(4) Å, *c* = 46.7278(8) Å, $\alpha=\beta=\gamma=90^\circ$, *V* = 9082.9(3) Å³, *Z* = 4, $\mu=4.541\text{ mm}^{-1}$. *R*₁ = 0.0607 (*I* > 2σ(*I*)), *wR*₂ = 0.1098 (all data), GOF = 1.032 (*I* > 2σ(*I*)), Flack parameter = -0.018(7) (CCDC 2063616).

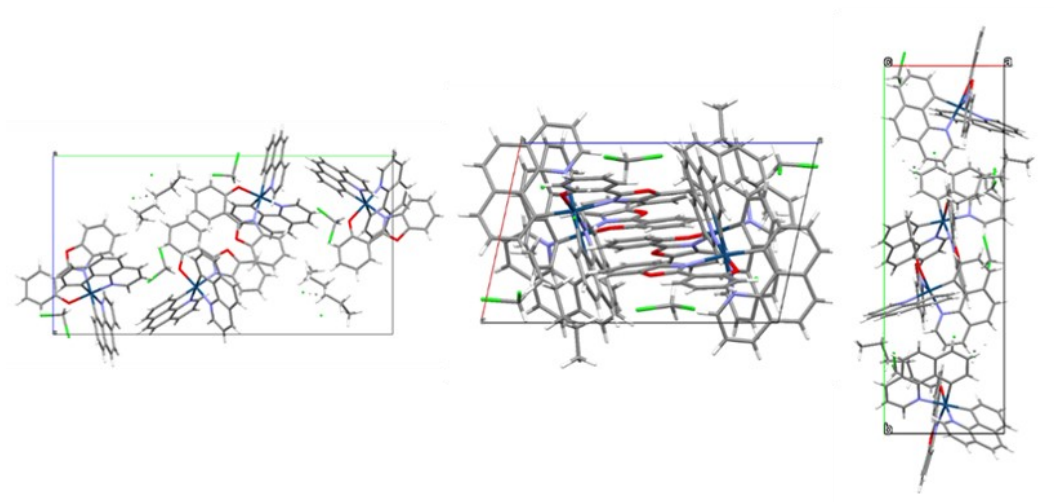


Figure S8. The views of Δ -cis-(N,N)-[Ir(bzq)₂(PBO)] along the a-axis (left), along the b-axis (middle) and along the c-axis (right), respectively. Solvent molecules are omitted for clarity.

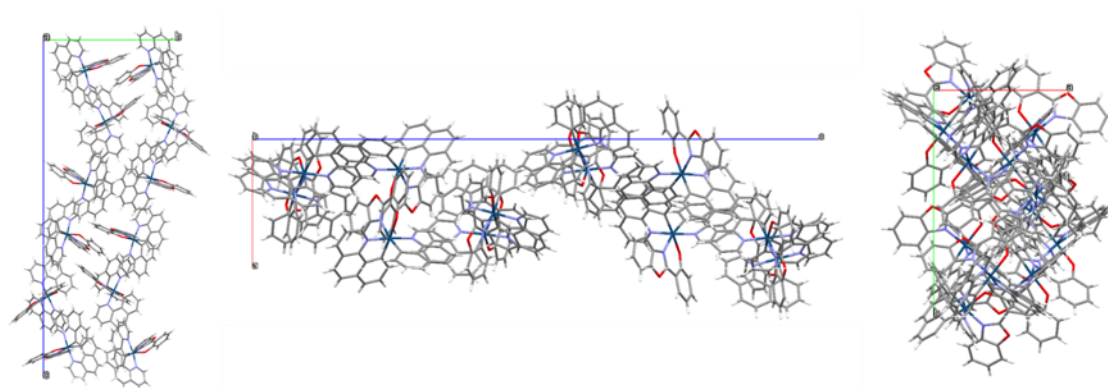


Figure S9. The views of Λ -trans-(N,N)-[Ir(bzq)₂(PBO)] along the a-axis (left), along the b-axis (middle) and along the c-axis (right), respectively. Solvent molecules are omitted for clarity.

Table S1. Structural information of racemic-*cis*, Δ -, and Λ -*cis*- (N,N) -[Ir(bzq)₂(PBO)]

Compound	Racemic- <i>cis</i> -(N,N) -[Ir(bzq) ₂ (PBO)]	Δ - <i>cis</i> -(N,N)-[Ir(bzq) ₂ (PBO)]	Λ - <i>cis</i> -(N,N)-[Ir(bzq) ₂ (PBO)]
Solvent	dichloromethane, hexane	dichloromethane, hexane	dichloromethane, hexane
Formula	C ₃₉ H ₂₄ IrN ₃ O ₂ , CH ₂ Cl ₂	C ₃₉ H ₂₄ IrN ₃ O ₂ , 0.32(C ₆ H ₁₄), 1.18(CH ₂ Cl ₂)	C ₃₉ H ₂₄ IrN ₃ O ₂ , 0.28(C ₆ H ₁₄), 1.22(CH ₂ Cl ₂)
FW	843.74	886.55	886.58
Crystal system	Triclinic	Monoclinic	Monoclinic
Space group	<i>P</i> -1	<i>P</i> 2 ₁	<i>P</i> 2 ₁
<i>a</i> [Å]	10.4642(3)	9.15927(16)	9.15128(17)
<i>b</i> [Å]	10.5088(2)	27.4078(5)	27.4566(6)
<i>c</i> [Å]	15.5137(4)	14.7784(3)	14.7617(3)
α [deg]	85.981(2)	90	90
β [deg]	76.654(2)	102.475(2)	102.392(2)
γ [deg]	72.389(2)	90	90
<i>V</i> [Å ³]	1582.12(7)	3622.31(12)	3622.66(13)
<i>Z</i>	2	4	4
μ [mm ⁻¹]	4.432	3.901	3.907
<i>R</i> ₁ (<i>I</i> > 2 σ (<i>I</i>))	0.0315	0.0395	0.0434
w <i>R</i> ₂ (all data)	0.0752	0.0923	0.0945
GOF (<i>I</i> > 2 σ (<i>I</i>))	1.051	1.045	1.039
Flack parameter	—	-0.013(3)	-0.016(4)
CCDC No.	2063612	2063613	2063614

Table S2. Structural information of racemic-*trans* and Λ -*trans* (N,N) -[Ir(bzq)₂(PBO)]

Compound	Racemic- <i>trans</i> -(N,N) -[Ir(bzq) ₂ (PBO)]	Λ - <i>trans</i> -(N,N)-[Ir(bzq) ₂ (PBO)]
Solvent	dichloromethane, hexane	dichloromethane, hexane
Formula	C ₃₉ H ₂₄ IrN ₃ O ₂ , CH ₂ Cl ₂	3(C ₃₉ H ₂₄ IrN ₃ O ₂), 1.5(CH ₂ Cl ₂)
FW	843.74	2403.82
Crystal system	Triclinic	orthorhombic
Space group	<i>P</i> -1	<i>P</i> 2 ₁ 2 ₁ 2 ₁
<i>a</i> [Å]	12.2740(2)	10.6678(2)
<i>b</i> [Å]	16.9938(3)	18.2211(4)
<i>c</i> [Å]	18.1515(5)	46.7278(8)
α [deg]	106.596(2)	90
β [deg]	109.535(2)	90
γ [deg]	103.187(2)	90
<i>V</i> [Å ³]	3192.42(13)	9082.9(3)
<i>Z</i>	4	4
μ [mm ⁻¹]	4.392	4.541
<i>R</i> ₁ (<i>I</i> > 2 σ (<i>I</i>))	0.0426	0.0607
w <i>R</i> ₂ (all data)	0.0912	0.1098
GOF (<i>I</i> > 2 σ (<i>I</i>))	1.022	1.032
Flack parameter	—	-0.018(7)
CCDC No.	2063615	2063616

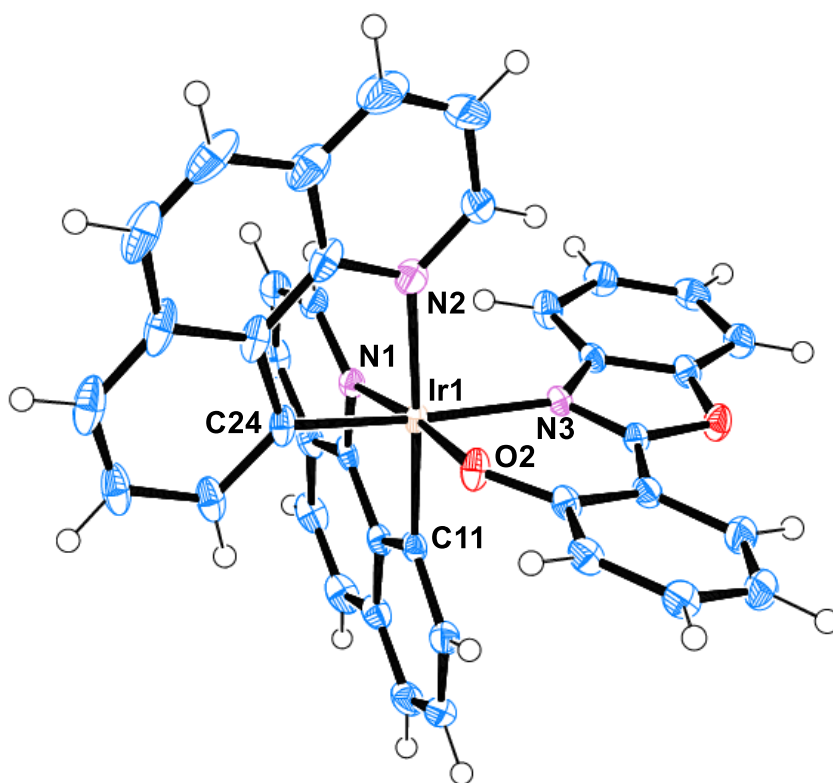


Figure S10. An ORTEP drawing of racemic-*cis*(N,N)-[Ir(bzq)₂(PBO)] with a numbering scheme of atoms around a metal ion. The thermal ellipsoids are scaled to the 50% probability level.

Table S3. Selected bond distances (Å) and angles (deg) of racemic-*cis*(N,N)-[Ir(bzq)₂(PBO)].

<u>Distance (Å)</u>			
Ir1—O2	2.042 (3)	Ir1—N1	2.025 (3)
Ir1—N2	2.131 (3)	Ir1—N3	2.147 (3)
Ir1—C11	2.007 (4)	Ir1—C24	2.009 (4)
<u>Angle (deg)</u>			
O2—Ir1—N2	86.77 (12)	O2—Ir1—N3	87.54 (11)
N1—Ir1—N2	96.68 (13)	N1—Ir1—N3	97.36 (12)
N2—Ir1—N3	93.47 (12)	C11—Ir1—O2	94.84 (13)
C11—Ir1—N1	81.5 (14)	C11—Ir1—N3	88.97 (13)
C11—Ir1—C24	97.39 (15)	C24—Ir1—O2	88.16 (12)
C24—Ir1—N1	87.36 (13)	C24—Ir1—N2	80.28 (15)

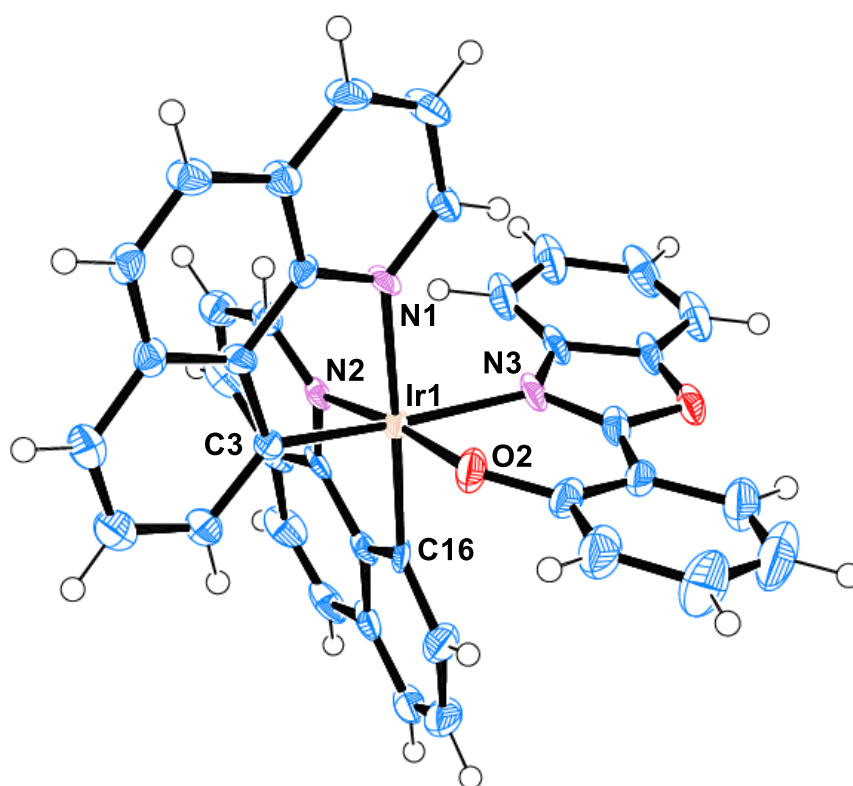


Figure S11. An ORTEP drawing of Δ -cis(N,N)-[Ir(bzq)₂(PBO)] with a numbering scheme of atoms around a metal ion. The thermal ellipsoids are scaled to the 50% probability level.

Table S4. Selected bond distances (Å) and angles (deg) of Δ -*cis*(N,N)-[Ir(bzq)₂(PBO)]

<u>Distance (Å)</u>			
Ir1—O2	2.047 (7)	Ir1—N1	2.132 (7)
Ir1—N2	2.024 (8)	Ir1—N3	2.164 (7)
Ir1—C3	2.011 (8)	Ir1—C16	2.020 (9)
<u>Angle (deg)</u>			
O2—Ir1—N1	87.1 (3)	O2—Ir1—N3	88.3 (3)
N1—Ir1—N3	93.2 (3)	N2—Ir1—N1	96.9 (3)
N2—Ir1—N3	96.4 (3)	C3—Ir1—O2	86.6 (3)
C3—Ir1—N1	80.9 (3)	C3—Ir1—N2	89.1 (3)
C3—Ir1—C16	97.3 (4)	C16—Ir1—O2	94.0 (3)
C16—Ir1—N2	81.9 (3)	C16—Ir1—N3	88.8 (3)

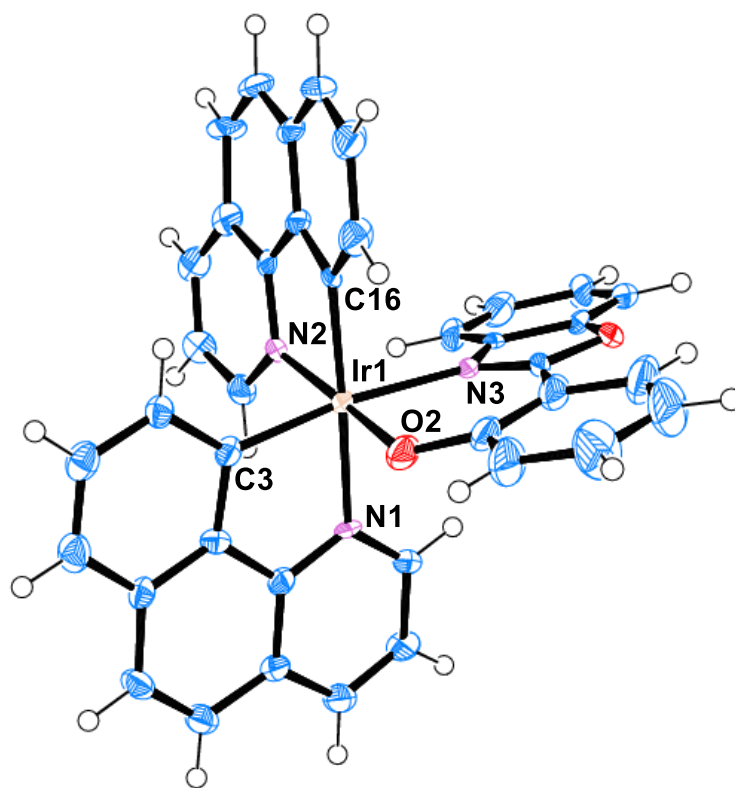


Figure S12. An ORTEP drawing of Λ -*cis*(N,N)-[Ir(bzq)₂(PBO)] with a numbering scheme of atoms around a metal ion. The thermal ellipsoids are scaled to the 50% probability level.

Table S5. Selected bond distances (Å) and angles (deg) of Λ -*cis*(N,N)-[Ir(bzq)₂(PBO)]

<u>Distance (Å)</u>			
Ir1—O2	2.047 (7)	Ir1—N1	2.132 (7)
Ir1—N2	2.024 (8)	Ir1—N3	2.164 (7)
Ir1—C3	2.011 (8)	Ir1—C16	2.020 (9)
<u>Angle (deg)</u>			
O2—Ir1—N1	87.1 (3)	O2—Ir1—N3	88.3 (3)
N1—Ir1—N3	93.2 (3)	N2—Ir1—N1	96.9 (3)
N2—Ir1—N3	96.4 (3)	C3—Ir1—O2	86.6 (3)
C3—Ir1—N1	80.9 (3)	C3—Ir1—N2	89.1 (3)
C3—Ir1—C16	97.3 (4)	C16—Ir1—O2	94.0 (3)
C16—Ir1—N2	81.9 (3)	C16—Ir1—N3	88.8 (3)

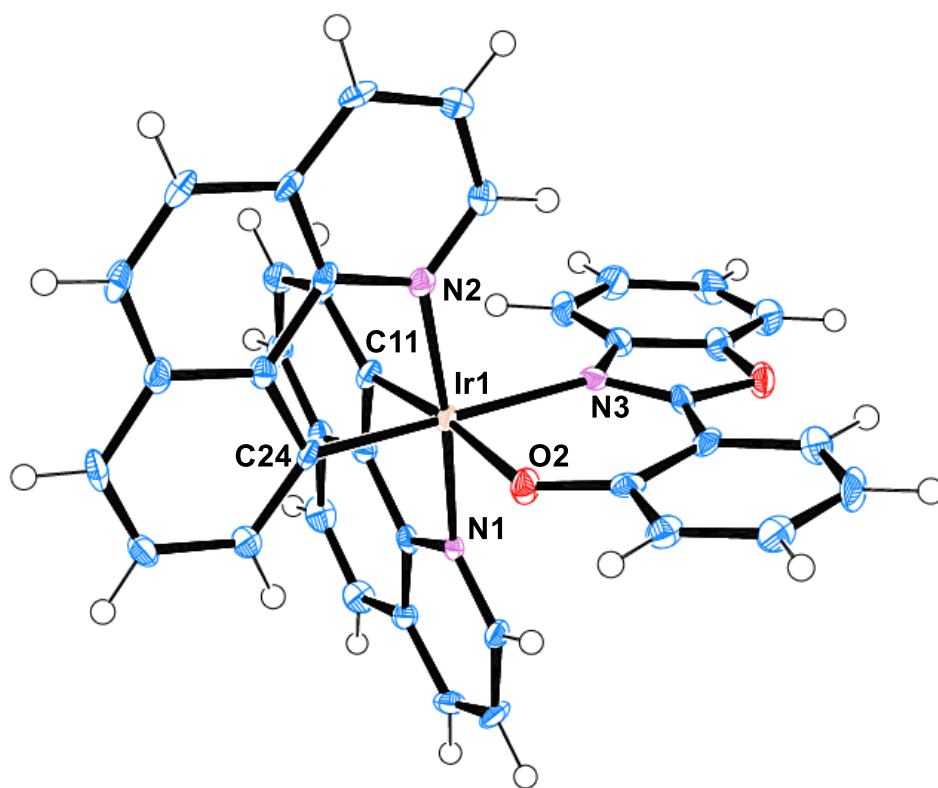


Figure S13. An ORTEP drawing of *racemic-trans*-(N,N)-[Ir(bzq)₂(PBO)] with a numbering scheme of atoms around a metal ion. The thermal ellipsoids are scaled to the 50% probability level.

Table S6. Selected bond distances (Å) and angles (deg) of racemic-*trans*(N,N)-[Ir(bzq)₂(PBO)]

<u>Distance (Å)</u>			
Ir1—O2	2.123 (3)	Ir1—N1	2.035 (4)
Ir1—N2	2.045 (4)	Ir1—N3	2.167 (4)
Ir1—C11	1.994 (5)	Ir1—C24	1.994 (5)
<u>Angle (deg)</u>			
O2—Ir1—N3	87.14 (14)	N1—Ir1—O2	93.20 (15)
N1—Ir1—N3	89.12 (15)	N2—Ir1—O2	90.25 (15)
N2—Ir1—N3	96.07 (15)	C11—Ir1—N1	81.89 (18)
C11—Ir1—N2	94.24 (18)	C11—Ir1—N3	97.34 (18)
C24—Ir1—O2	90.36 (16)	C24—Ir1—N1	93.00 (18)
C24—Ir1—N2	81.95 (18)	C24—Ir1—C11	85.3 (2)

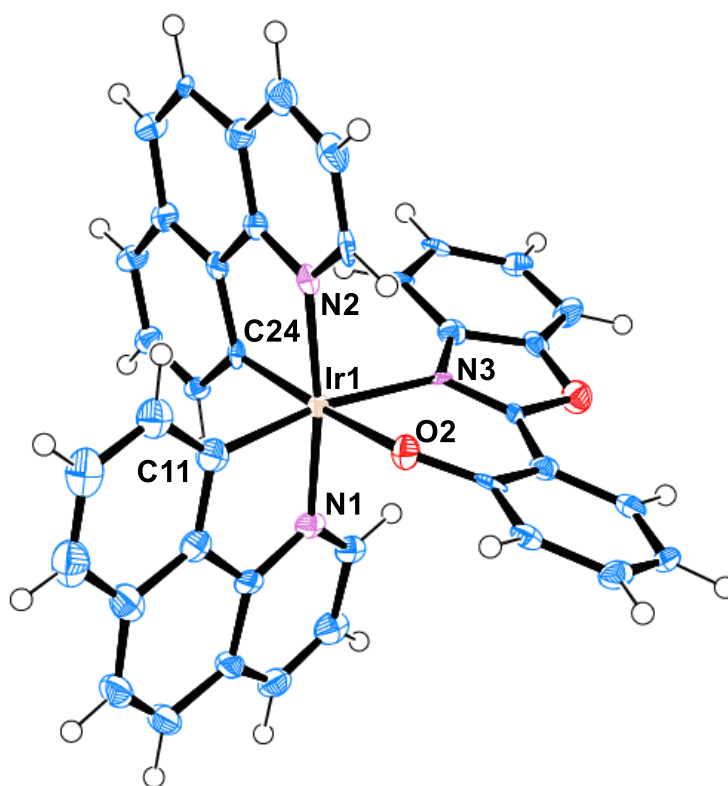


Figure S14. An ORTEP drawing of Λ -*trans*(N,N)-[Ir(bzq)₂(PBO)] with a numbering scheme of atoms around a metal ion. The thermal ellipsoids are scaled to the 50% probability level.

Table S7. Selected bond distances (Å) and angles (deg) of Λ -*trans*(N,N)-[Ir(bzq)₂(PBO)]

<u>Distance (Å)</u>			
Ir1—O1	2.155 (10)	Ir1—N1	2.031 (12)
Ir1—N2	2.048 (12)	Ir1—N3	2.166 (11)
Ir1—C11	2.008 (17)	Ir1—C24	2.008 (14)
<u>Angle (deg)</u>			
O1—Ir1—N3	84.6 (4)	N1—Ir1—O1	94.8 (4)
N1—Ir1—N3	91.5 (5)	N2—Ir1—O1	91.6 (4)
N2—Ir1—N3	93.9 (4)	C11—Ir1—O1	87.6 (5)
C11—Ir1—N1	81.9 (6)	C11—Ir1—N2	93.7 (5)
C24—Ir1—N1	91.9 (5)	C24—Ir1—N2	81.5 (5)
C24—Ir1—N3	98.5 (5)	C24—Ir1—C11	90.2 (6)

S6. Thermal isomerization of *trans*-(N,N)-[Ir(bzq)₂Cl]₂

The stereochemically pure *trans*-(N,N)-[Ir(bzq)₂Cl]₂ which was obtained by a reported procedure was heated in 2-ethoxyethanol/water (1:1) at 110 °C for 12 h under a nitrogen atmosphere. A mixture of iridium complexes was quantitatively recovered by precipitation from the reaction mixture and analyzed by using ¹H NMR and ESI-MS.

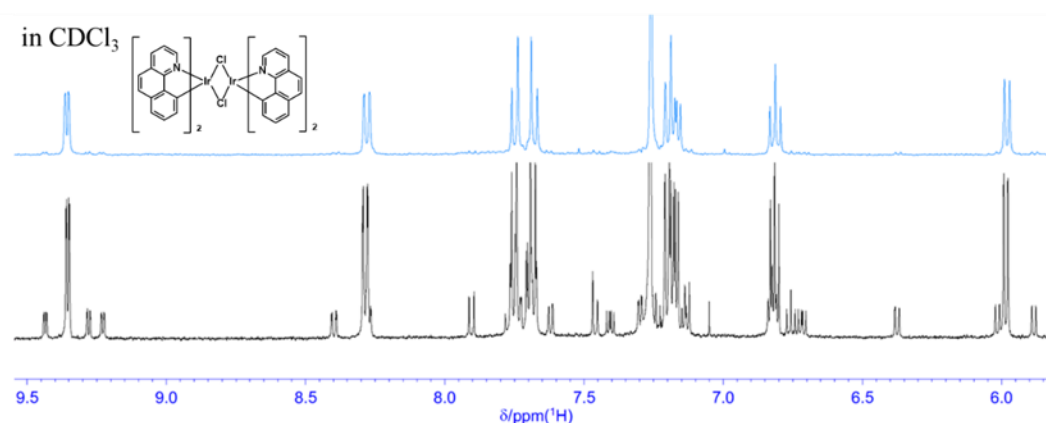


Figure S15. ¹H NMR (500 MHz, CDCl₃) spectra of synthetic [Ir(bzq)₂Cl]₂ (top) and a dimeric mixture obtained in 45% of recovery after washing the precipitate with methanol (bottom).

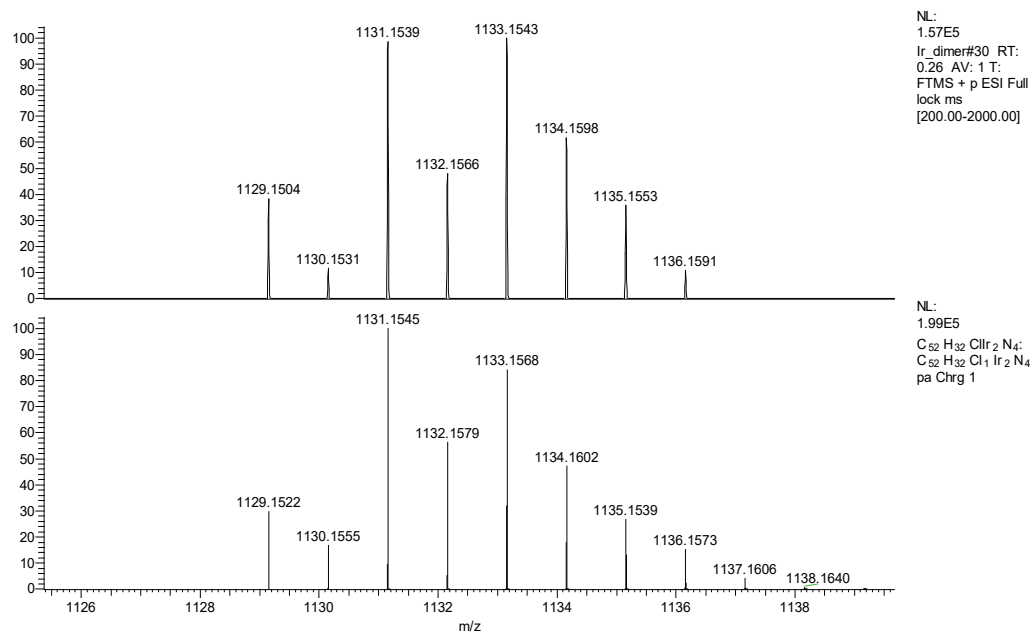


Figure S16. ESI-mass spectra of a crude sample (top) obtained by a thermal experiment and its calculated simulation result (bottom).

S7. Emission properties of [Ir(bzq)₂(PBO)]

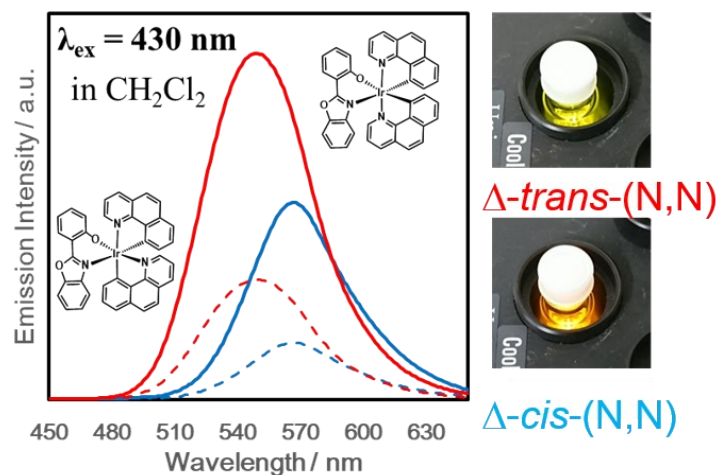


Figure S17. Emission spectra of (blue) *cis*-(N,N) isomer ($5.6 \times 10^{-5} \text{ M}$) and (red) *trans*-(N,N) isomer ($5.5 \times 10^{-5} \text{ M}$) of [Ir(bzq)₂(PBO)] at room temperature. A solvent was CH_2Cl_2 . The excitation wavelength was 430 nm. The atmosphere was N_2 (solid line) and air (dotted line). The photo shows the color of emission for each isomer.

S8. Theoretical calculation of energy diagrams of [Ir(bzq)₂(PBO)]

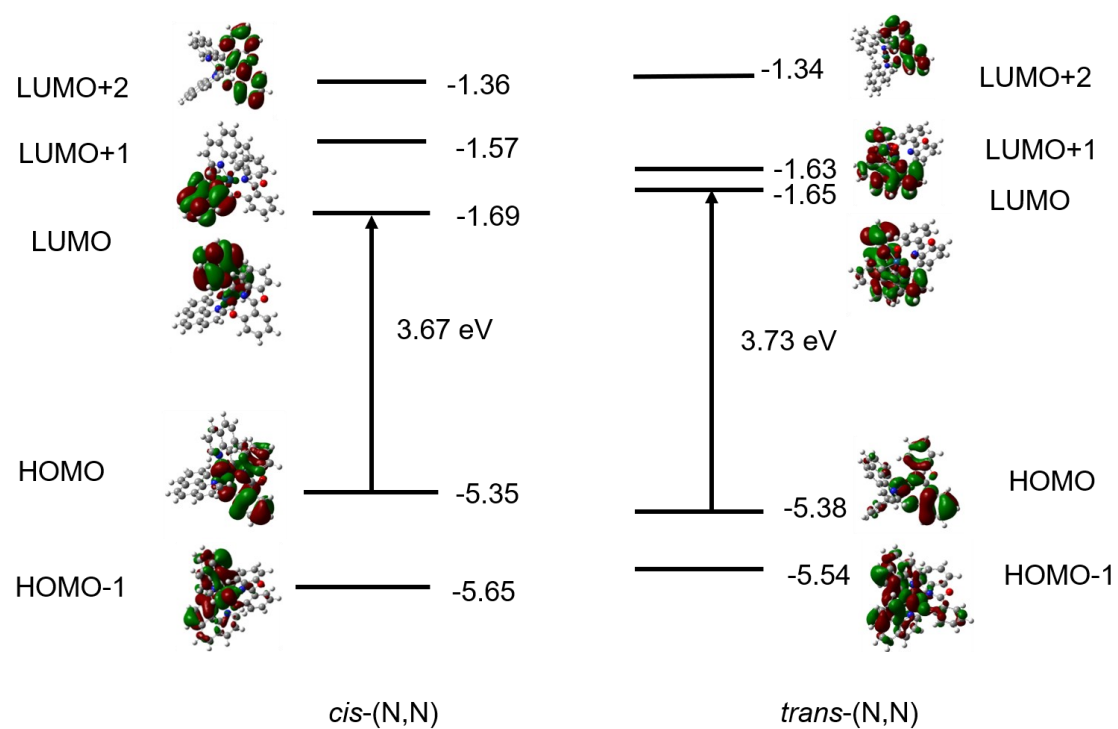


Figure S18. Energy diagrams and molecular orbitals for HOMO-1 to LUMO+2 based on the DFT ground state calculation: (left) *cis*-(N,N)- and (right) *trans*-(N,N) isomers, respectively.