

## Supporting Information

### Optical properties regularities in the family of $\{\text{Mo}_6\text{I}_8\}$ aquahydroxo complexes

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## Content

FTIR spectra, XRPD patterns, and TGA curves .....	S3
Crystallographic data and selected interatomic distances .....	S4
Representation of the layered structure in <b>1-NO<sub>3</sub></b> and <b>1-OTs</b> and absorption spectra of aqueous solutions of <b>1-NO<sub>3</sub></b> at different pH .....	S5
Selected parameters of crystal structure and emission of the clusters and deconvolution of the emission spectra of <b>1-NO<sub>3</sub></b> and <b>1-OTs</b> in solid state.....	S6
Deconvolution of the emission spectra of <b>2-h2</b> and <b>2-h12<sup>new</sup></b> in solid state.....	S7
Parameters of deconvolution of emission spectra in solid state and deconvolution of the emission spectra of <b>1-NO<sub>3</sub></b> in aqueous solution at pH = 3.2 and 2.1.....	S8
Deconvolution of the emission spectra of <b>1-NO<sub>3</sub></b> in aqueous solution at pH = 1.2 and 0.0 .....	S9
Parameters of deconvolution of emission spectra of <b>1-NO<sub>3</sub></b> in in aqueous solution at different pH.....	10

FTIR spectra, XRPD patterns, and TGA curves

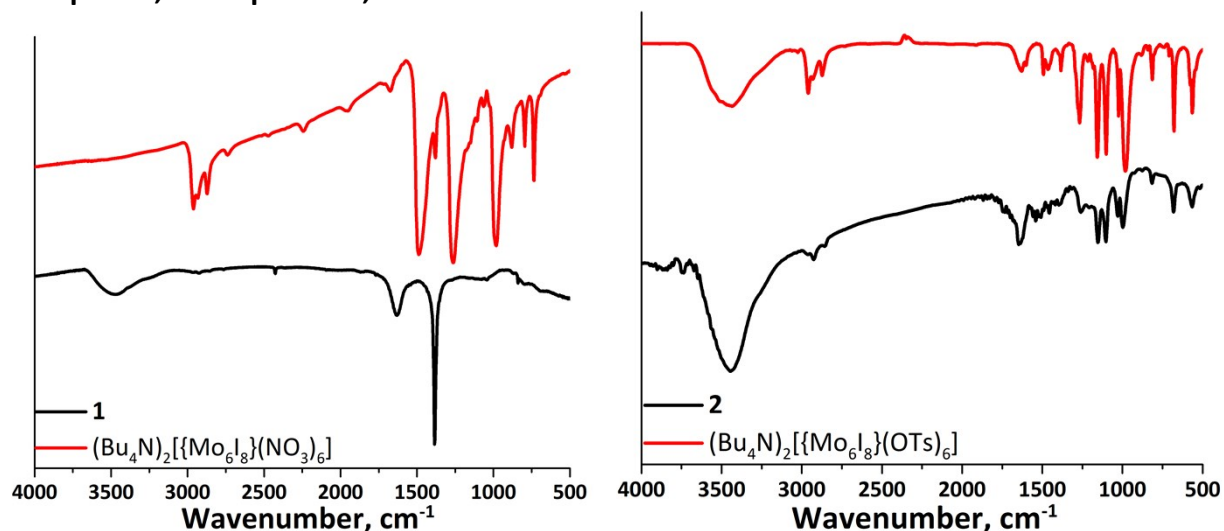


Figure 1S. FTIR spectra of 1-NO<sub>3</sub> (left) and 1-OTs (right) in comparison with initial clusters.

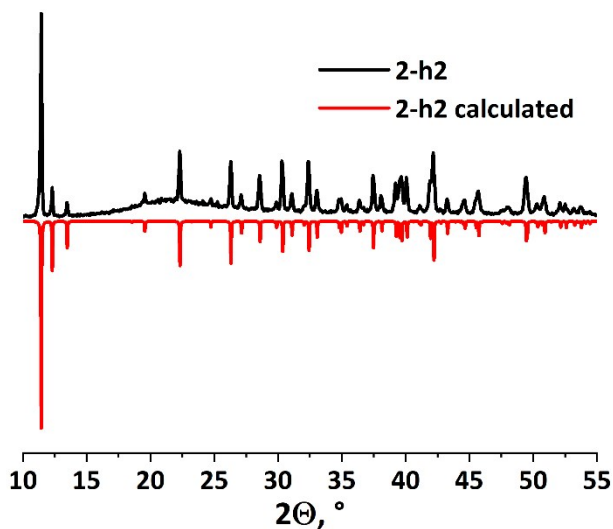


Figure 2S. XRPD pattern of 2-h2 in comparison with calculated from the crystal structures of  $[\{\text{Mo}_6\text{I}_8\}(\text{H}_2\text{O})_2(\text{OH})_4]\cdot 2\text{H}_2\text{O}$ .

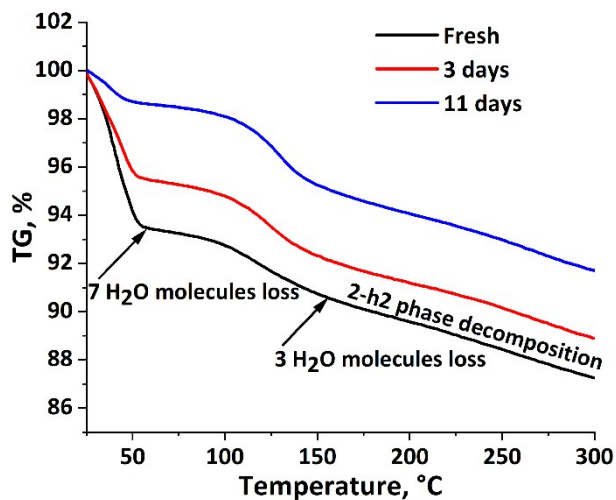


Figure 3S. TGA curves of 2-h2 and 2-h12<sup>new</sup> (fresh and 3 and 11 days old). Heating rates of 10 °C min<sup>-1</sup>.

**Table 1S.** Crystallographic data, data collection and refinement parameters.

	<b>1-NO<sub>3</sub></b>	<b>1-OTs</b>	<b>2-h12<sup>new</sup></b>
Empirical formula	H <sub>16</sub> I <sub>8</sub> Mo <sub>6</sub> N <sub>2</sub> O <sub>15</sub>	C <sub>14</sub> H <sub>28</sub> I <sub>8</sub> Mo <sub>6</sub> O <sub>14</sub> S <sub>2</sub>	H <sub>32</sub> I <sub>8</sub> Mo <sub>6</sub> O <sub>18</sub>
Formula weight	1874.99	2075.32	1911.09
Temperature (K)	150(2)	150(2)	150(2)
Crystal size (mm <sup>3</sup> )	0.120 × 0.100 × 0.030	0.160 × 0.130 × 0.040	0.085 × 0.085 × 0.035
Crystal system	Triclinic	Triclinic	Triclinic
Space group	P-1	P-1	P-1
Z	2	1	1
Unit cell dimensions			
a (Å)	10.5553(3)	9.447(1)	9.4284(9)
b (Å)	11.5145(3)	9.459(1)	9.799(1)
c (Å)	12.9195(3)	11.504(1)	10.008(1)
α (°)	78.837(1)	77.300(3)	73.929(4)
β (°)	78.265(1)	79.348(3)	67.566(4)
γ (°)	67.511(1)	73.852(3)	88.598(4)
Volume (Å <sup>3</sup> )	1408.80(6)	954.8(2)	817.8(1)
D <sub>calcd.</sub> (g·cm <sup>-3</sup> )	4.420	3.609	3.815
μ (mm <sup>-1</sup> )	11.411	8.539	9.837
θ range (°)	1.623 – 27.619	1.831 – 27.157	2.300 – 36.312
Index ranges	-12 ≤ h ≤ 13 -11 ≤ k ≤ 14 -13 ≤ l ≤ 16	-12 ≤ h ≤ 12 -12 ≤ k ≤ 12 -14 ≤ l ≤ 14	-15 ≤ h ≤ 12 -16 ≤ k ≤ 14 -16 ≤ l ≤ 13
Reflections collected	12016	7699	22858
Independent reflections	6450 (R <sub>int</sub> = 0.0293)	4143 (R <sub>int</sub> = 0.0337)	7293 (R <sub>int</sub> = 0.0411)
Data / restraints / parameters	6450 / 19 / 342	4143 / 6 / 210	7293 / 0 / 159
R[F <sup>2</sup> > 2σ(F <sup>2</sup> )]	R <sub>1</sub> = 0.0185 wR <sub>2</sub> = 0.0466	R <sub>1</sub> = 0.0634 wR <sub>2</sub> = 0.1967	R <sub>1</sub> = 0.0357 wR <sub>2</sub> = 0.0645
R(F <sup>2</sup> ) (all data)	R <sub>1</sub> = 0.0342 wR <sub>2</sub> = 0.0492	R <sub>1</sub> = 0.0680 wR <sub>2</sub> = 0.2010	R <sub>1</sub> = 0.0504 wR <sub>2</sub> = 0.0692
Goodness-of-fit on F <sup>2</sup>	1.037	1.063	1.018
Δρ <sub>max</sub> , Δρ <sub>min</sub> (e·Å <sup>-3</sup> )	1.135, -0.866	2.813, -7.017	1.462, -2.278

**Table 2S.** Selected interatomic distances (Å).

	<b>1-NO<sub>3</sub></b>	<b>1-OTs</b>	<b>2-h12<sup>new</sup></b>
Mo-Mo	2.6509(4)–2.6751(4)	2.647(1)–2.662(1)	2.6566(5)–2.6670(5)
Mo-I	2.7599(4)–2.8049(4)	2.781(1)–2.801(1)	2.787(5)–2.8057(5)
Mo-O	2.113(2)–2.189(3)	2.053(7)–2.175(8)	2.089(3)–2.192(3)

Representation of the layered structure in 1-NO<sub>3</sub> and 1-OTs and absorption spectra of aqueous solutions of 1-NO<sub>3</sub> at different pH

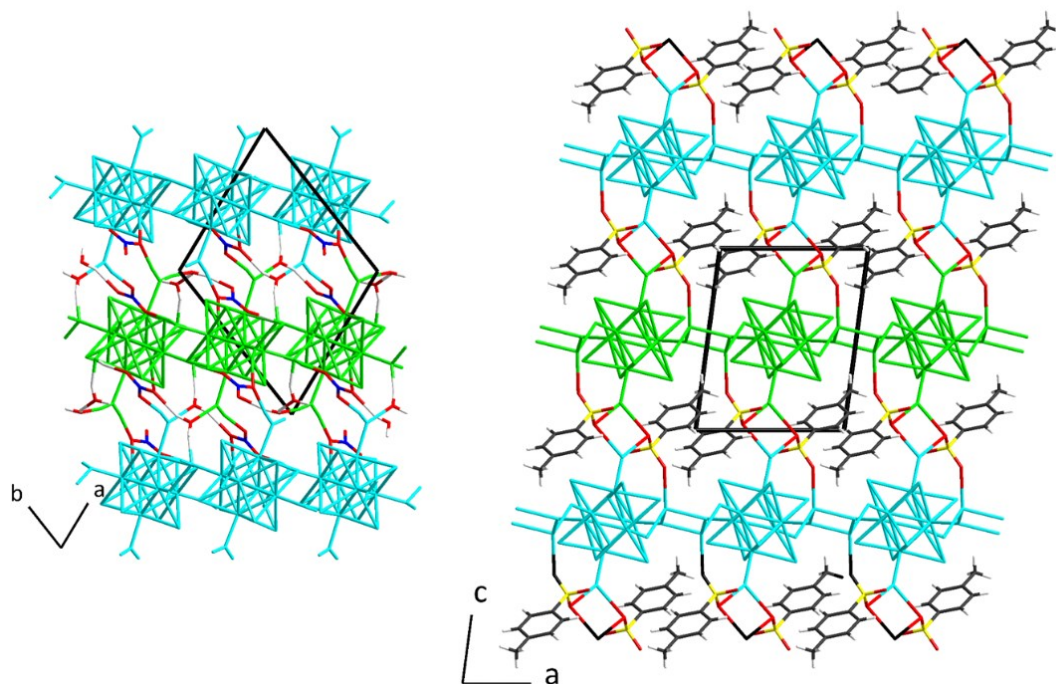


Figure 4S. The side view on the layers in 1-NO<sub>3</sub> (left) and 1-OTs (right).

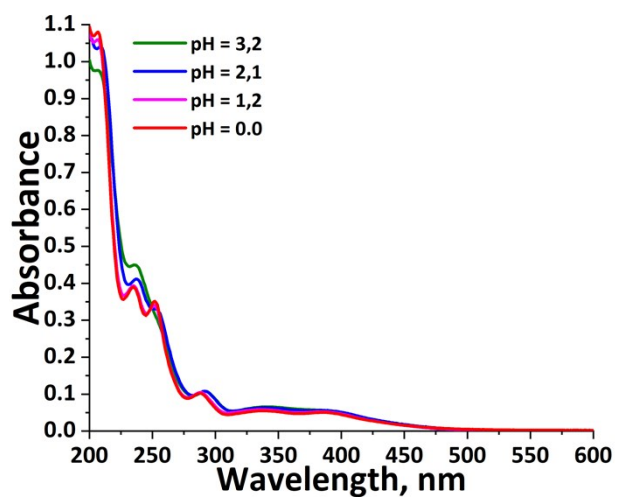


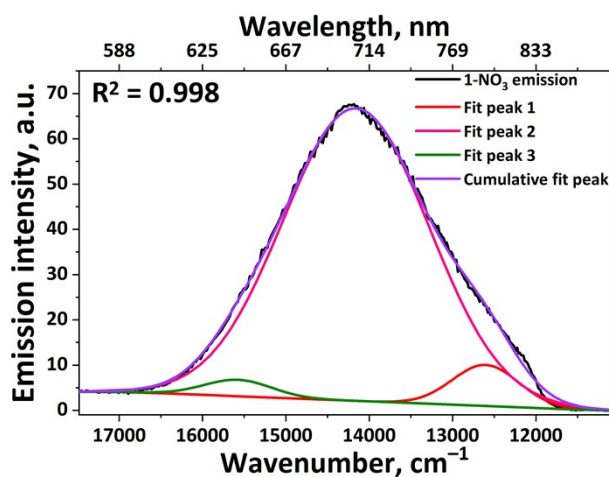
Figure 5S. Absorption spectra of aqueous solutions of 1-NO<sub>3</sub> at different pH (3.1, 2.1, 1.2 and 0).

## Selected parameters of crystal structure and emission of the clusters and deconvolution of the emission spectra of 1-NO<sub>3</sub> and 1-OTs in solid state

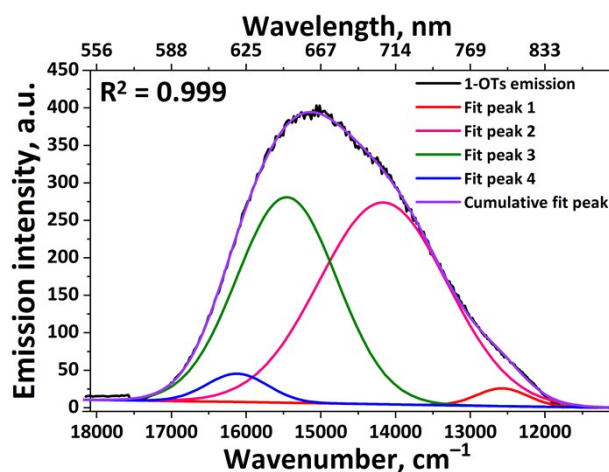
**Table 3S.** Selected parameters of crystal structure and emission of the clusters.

Cluster	Number of direct hydrogen contacts	$d_{\text{calc}}$ , g cm <sup>-3</sup>	Cluster-cluster contacts, Å	$\lambda_{\text{em}}$ , nm	Relative intensity, a.u.	$S_{\text{int}}$ , a.u. <sup>#</sup>
1-OTs	0	3.609	9.447; 9.459; 11.504	661	395	50239
2-h12 <sup>new</sup>	0	3.815	9.428; 9.799; 10.008	693	274	7887
1-NO <sub>3</sub>	4	4.420	8.885; 8.945; 10.555	705	66	29261
2-h2	6	4.932	9.138; 9.634	706	18	1889

<sup>#</sup> $S_{\text{int}}$  is integrated emission intensity (area under the curve)



**Figure 6S.** Deconvolution of the emission spectra of 1-NO<sub>3</sub> in solid state.



**Figure 7S.** Deconvolution of the emission spectra of 1-OTs in solid state.

## Deconvolution of the emission spectra of 2-h2 and 2-h12<sup>new</sup> in solid state

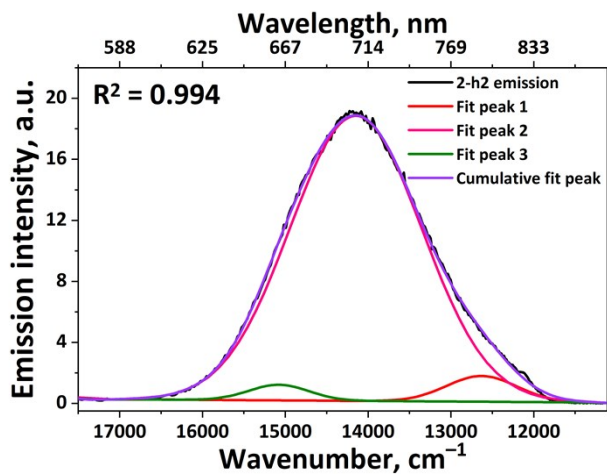


Figure 8S. Deconvolution of the emission spectra of 2-h2 in solid state.

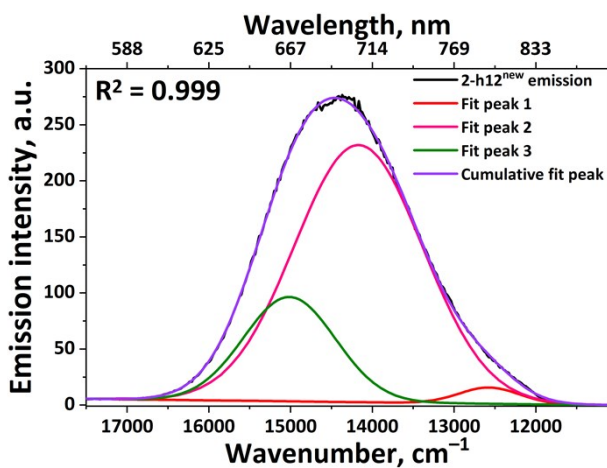
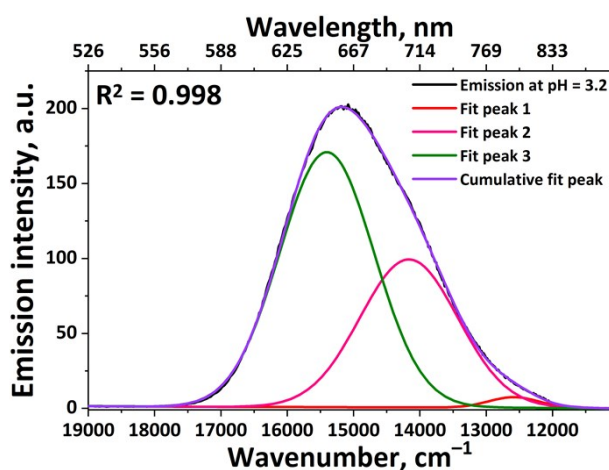


Figure 9S. Deconvolution of the emission spectra of 2-h12<sup>new</sup> in solid state.

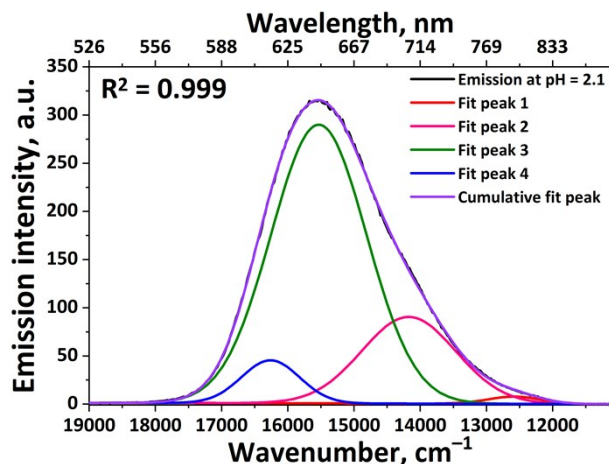
**Parameters of deconvolution of emission spectra in solid state and deconvolution of the emission spectra of 1-NO<sub>3</sub> in aqueous solution at pH = 3.2 and 2.1**

**Table 4S.** Overall emission maximum ( $\lambda_{em}$ ), integrated emission intensity ( $S_{int}$ ), peak positions, FWHM, relative contributions of the peaks integrated area to the area of cumulative peak in %, and coefficient of determination of cumulative peak ( $R^2$ ) for clusters in solid state.

Parameter	1-OTs	1-NO <sub>3</sub>	2-h12 <sup>new</sup>	2-h2
$\lambda_{em}$ , nm; (cm <sup>-1</sup> )	661; (15135)	705; (14180)	693; (14416)	706, (14164)
$\lambda_{max1}$ , nm; (cm <sup>-1</sup> )	620; (16118)	–	–	–
$\lambda_{max2}$ , nm; (cm <sup>-1</sup> )	647; (15455)	642 (15578)	666; (15013)	663; (15077)
$\lambda_{max3}$ , nm; (cm <sup>-1</sup> )	706; (14165)	706; (14159)	706; (14168)	707; (14147)
$\lambda_{max4}$ , nm; (cm <sup>-1</sup> )	795; (12571)	793; (12605)	796; (12571)	792 (12629)
FWHM 1	958	–	–	–
FWHM 2	1575	948	1321	851
FWHM 3	2016	2062	1848	1868
FWHM 4	795	936	865	998
Cont. 1, %	3.51	–	–	–
Cont. 2, %	41.96	2.37	21.98	2.32
Cont. 3, %	52.70	91.79	75.78	93.19
Cont. 4, %	1.83	5.84	2.24	4.49
$R^2$	0.999	0.998	0.999	0.994



**Figure 10S.** Deconvolution of the emission spectra of 1-NO<sub>3</sub> in aqueous solution at pH = 3.2.



**Figure 11S.** Deconvolution of the emission spectra of 1-NO<sub>3</sub> in aqueous solution at pH = 2.1.



Deconvolution of the emission spectra of 1-NO<sub>3</sub> in aqueous solution at pH = 1.2 and 0.0

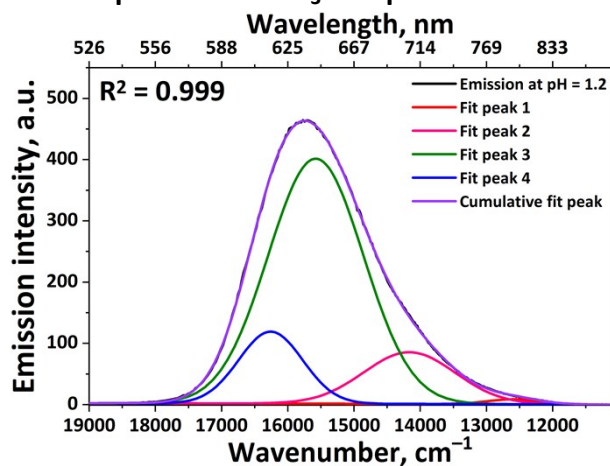


Figure 12S. Deconvolution of the emission spectra of 1-NO<sub>3</sub> in aqueous solution at pH = 1.2.

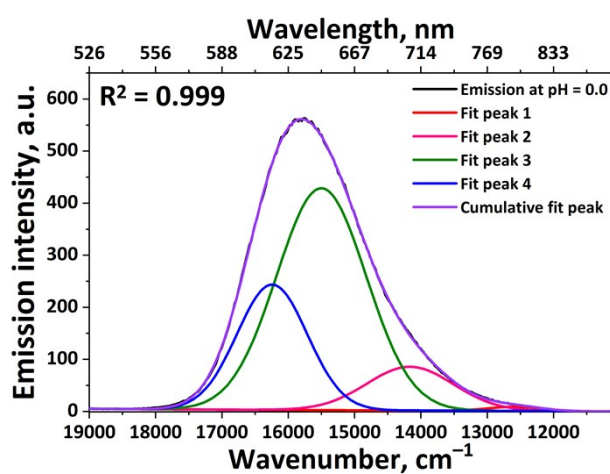


Figure 13S. Deconvolution of the emission spectra of 1-NO<sub>3</sub> in aqueous solution at pH = 0.0.

## Parameters of deconvolution of emission spectra of 1-NO<sub>3</sub> in aqueous solution at different pH

**Table 5S.** Overall emission maximum ( $\lambda_{em}$ ), integrated emission intensity ( $S_{int}$ ), peak positions, FWHM, relative contributions of the peaks integrated area to the area of cumulative peak in %, and coefficient of determination of cumulative peak ( $R^2$ ) for **1-NO<sub>3</sub>** in aqueous solution at different pH.

Parameter	pH = 3.2	pH = 2.1	pH = 1.2	pH = 0.0
$\lambda_{em}$ , nm; ( $cm^{-1}$ )	658; (15198)	643; (15552)	636; (15723)	633, (15798)
$S_{int}$ , a.u.#	22324	31797	42967	49345
$\lambda_{max1}$ , nm; ( $cm^{-1}$ )	-	615; (16261)	615; (16258)	616; (16242)
$\lambda_{max2}$ , nm; ( $cm^{-1}$ )	649; (15402)	644; (15532)	642; (15577)	645; (15500)
$\lambda_{max3}$ , nm; ( $cm^{-1}$ )	706; (14166)	706; (14173)	706; (14167)	706; (14164)
$\lambda_{max4}$ , nm; ( $cm^{-1}$ )	795; (12583)	795; (12577)	795; (12579)	795; (12575)
FWHM 1	-	1031	1137	1245
FWHM 2	1676	1696	1690	1588
FWHM 3	1738	1673	1628	1541
FWHM 4	884	935	934	947
Cont. 1, %	-	6.61	13.99	26.91
Cont. 2, %	61.60	70.70	70.83	60.80
Cont. 3, %	37.06	21.67	14.40	11.61
Cont. 4, %	1.34	1.01	0.78	0.68
$R^2$	0.998	0.999	0.999	0.999

# $S_{int}$  is integrated emission intensity (area under the curve)