

Electronic Supporting Information

Luminescence thermometry based on one-dimensional benzoato-bridged coordination polymers containing lanthanide ions

Alexandru Topor,^a Daniel Avram,^b Radu Dascalu,^c Catalin Maxim,^a Carmen Tiseanu,^{*b} Marius Andruh^{*a,d}

^a*University of Bucharest, Faculty of Chemistry, Inorganic Chemistry Laboratory, Str. Dumbrova Rosie nr. 23, 020464-Bucharest, Romania*

^b*National Institute for Laser, Plasma and Radiation Physics, RO, 76900, Bucharest-Magurele, Romania*

^c*National Institute for Research and Development in Electrical Engineering ICPE-CA, Splaiul Unirii 313, 030138 Bucharest, Romania*

^d*C. D. Nenitzescu Institute of Organic Chemistry of the Romanian Academy, Splaiul Independentei 202B, Bucharest, Romania*

Table S1. Crystallographic data, details of data collection and structure refinement parameters for **1** and **3**.

Compounds	1	3
Formula	C ₅₈ H ₇₀ O ₂₂ Tb ₂	C ₂₇ H ₃₁ O ₁₁ Nd
Molecular weight	1436.98	675.76
Temperature (K)	293(2)	293(2)
Crystal system	<i>triclinic</i>	<i>monoclinic</i>
Space group	<i>P</i> -1	<i>P</i> 2 ₁ /c
<i>a</i> (Å)	9.8516(2)	10.0703(3)
<i>b</i> (Å)	12.2547(3)	9.7655(3)
<i>c</i> (Å)	13.9649(3)	28.8888(9)
α (deg)	107.133(2)	90
β (deg)	98.272(5)	91.964(3)
γ (deg)	102.315(2)	90
<i>V</i> (Å ³)	1497.64(6)	2839.30(15)
<i>Z</i>	1	4
<i>D</i> _{calc} (g cm ⁻³)	1.593	1.581
<i>F</i> (000)	724.0	1364.0
Wavelength (Å)	0.71073	0.71073
Final <i>R</i> ₁ ^a , <i>wR</i> ₂ ^b [<i>I</i> > 2σ(<i>I</i>)]	0.0367, 0.0781	0.0619, 0.2022
<i>R</i> ₁ ^a , <i>wR</i> ₂ ^b (all data)	0.0415, 0.0810	0.0672, 0.2045
Goodness-of-fit on <i>F</i> ²	1.035	1.647
Largest peak in final difference (e Å ⁻³)	1.07, -1.09	2.33, -1.86

^a*R*₁ = $\sum ||F_o| - |F_c|| / \sum |F_o|$. ^b*wR*₂ = $[\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$; $w = 1 / [\sigma^2(F_o^2) + (aP)^2 + bP]$ where $P = [\max(F_o^2, 0) + 2F_c^2] / 3$.

Table S2. Continuous Shape Measures for the coordination polyhedron around the Ln^{III} ion

Geometry	Tb ^{III} (1)	Nd ^{III} (3)
OP-8	28.788	30.550
HPY-8	22.332	23.505
HBPY-8	12.277	16.124
CU-8	89.642	10.370
SAPR-8	2.172	1.288
TDD-8	2.246	1.508
JGBF-8	11.706	14.491
JETBPY-8	26.951	28.572
JBTPR-8	1.985	1.084
BTPR-8	1.801	0.909
JSD-8	2.720	3.302
TT-8	10.240	10.726
ETBPY-8	24.264	25.028

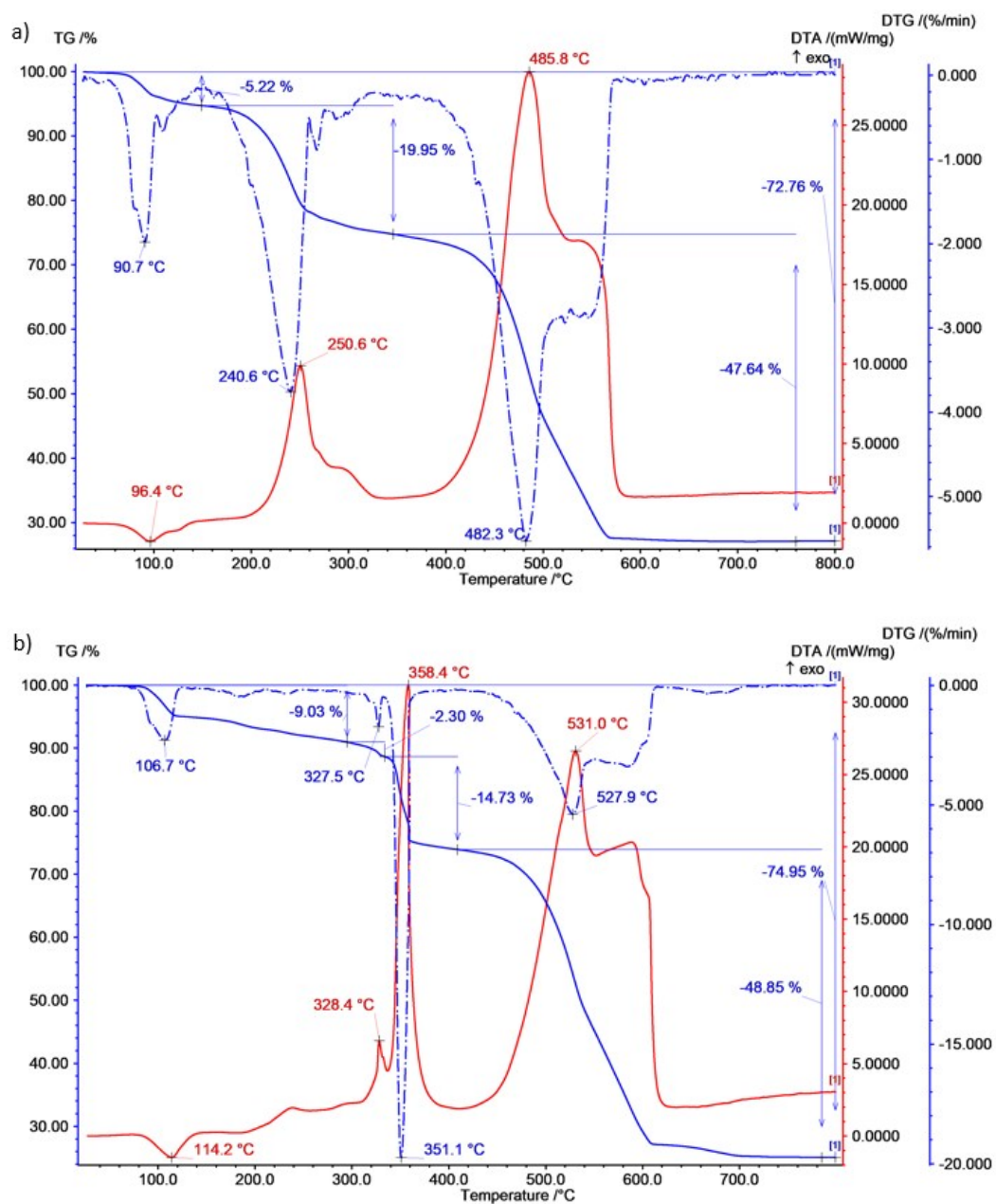


Figure S1. Thermogravimetry, Differential thermogravimetry and Differential Thermal Analysis plots of compounds **1** (a) and **3** (b).

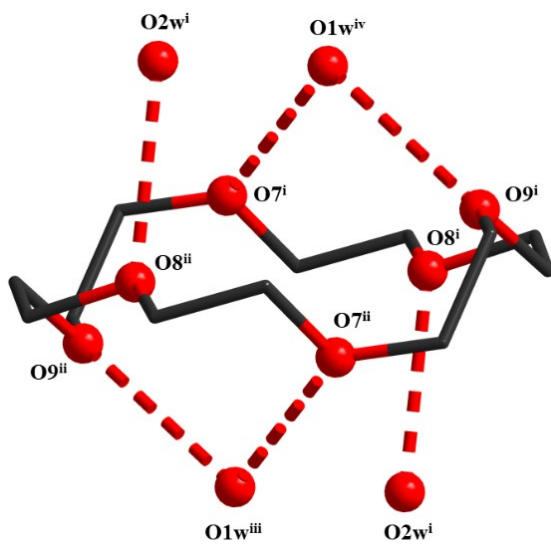


Figure S2. Hydrogen bonds formed via 18C6 molecules in compound **3**

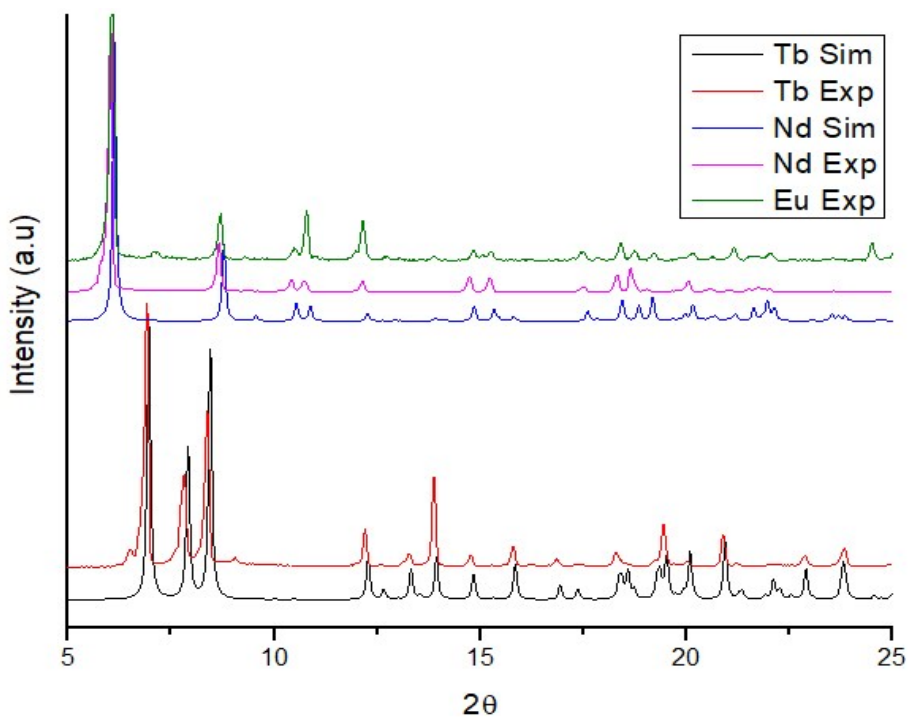


Figure S3. Simulated PXRD patterns for compounds **1** (black) and **3** (blue), experimental PXRD patterns for compounds **1** (red), **2** (green) and **3** (violet)

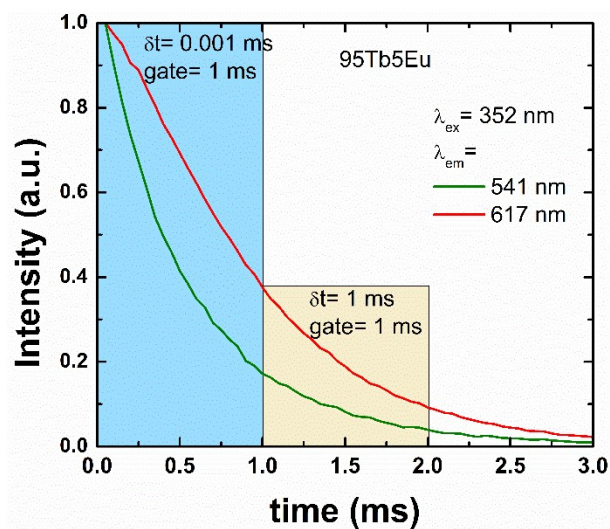


Figure S4. Selection of time delay and gate width used in PL thermometry measurements.

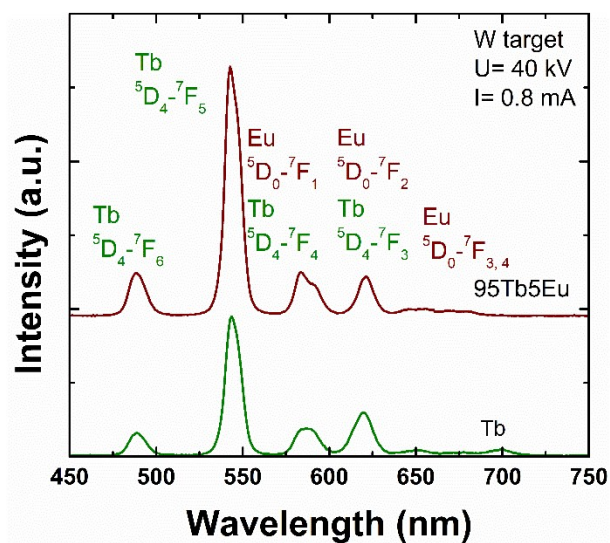


Figure S5. XEOL spectra of 1 (Tb) and 4 (95Tb5Eu) measured upon X-ray irradiation with a W target at $U = 40 \text{ kV}$ and $I = 0.8 \text{ mA}$.