

Supplementary Information

Dual Functions of pH-sensitive Cation Zr-MOF for 5-Fu: Large Drug Loading Capacity and High Sensitivity Fluorescence Detection

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1. Experimental procedures

Synthesis of ligand 9H-carbazole-2,7 dicarboxylic acid (9H-2,7-CDC)

The ligand 9H-carbazole-2,7 dicarboxylic acid (9H-2,7-CDC) was synthesized according to a literature report.^[i] Summarily, dimethyl 4,4'-biphenyldicarboxylate (Me₂BPDC) (5.03g) and 45ml of 98% concentrated sulfuric acid was mixed and stirred mechanically at room temperature until to green solution. Reduce the temperature of the solution to 5° C and continue stirring for one hour. Then, 1.8ml 56% HNO₃ and 5ml 98% concentrated sulfuric acid was added dropwise. The reaction solution was poured onto crushed ice, and a milky white solid was precipitated immediately. The obtained solid dimethyl 2-nitro-4,4'-bipthalate (Me₂BPDC-NO₂) was mixed with triphenylphosphonium (DPh₃) (8.2782 g), adding 26 ml o-dichlorobenzene, heating at 180° C, and refluxing for 48h. The mixture was dissolved in tetrahydrofuran and reflux at 60° C for 19 hours. The reaction solution was evaporated under a vacuum rotary to obtain a solid powder, dissolved in water, and filtered to remove the insoluble Ph₃PO. The filtrate was adjusted to acid with dilute hydrochloric acid, filtered, and dried to obtain a brown powder (9H-carbazole-2,7 dicarboxylic acid, 9H-2,7-CDC) 1.105g (yield 23.26%).

2. Characterization

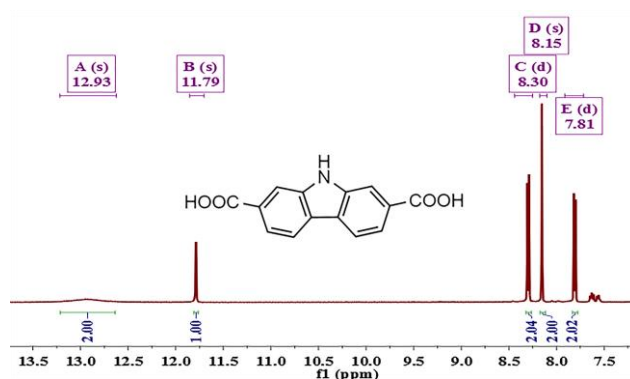


Figure S1 ¹H NMR spectrum of ligand 9H-carbazole-2,7 dicarboxylic acid (9H-2,7-CDC)

¹H NMR (500 MHz, DMSO) δ 12.93 (s, 1H), 11.79 (s, 1H), 8.30 (d, *J* = 8.2 Hz, 1H), 8.15 (s, 1H), 7.81 (d, *J* = 8.2 Hz, 1H).

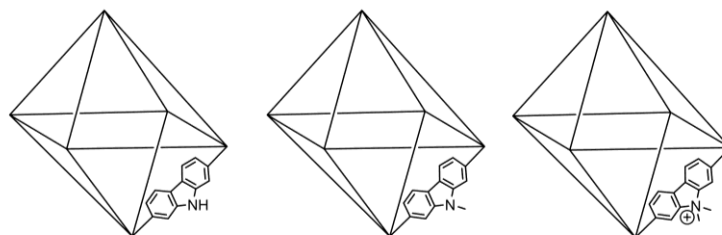


Figure S2 The three products after MeOTf methylation, from left to right, UiO-67-CDC, UiO-67-CDC-CH₃, UiO-67-CDC-(CH₃)₂

Table S1 The yield of the three products after methylation

product	UiO-67-CDC-(CH ₃) ₂	UiO-67-CDC-CH ₃	UiO-67-CDC
Yield (%)	49.1	32.1	18.8

3. Standard curve of 5-Fu

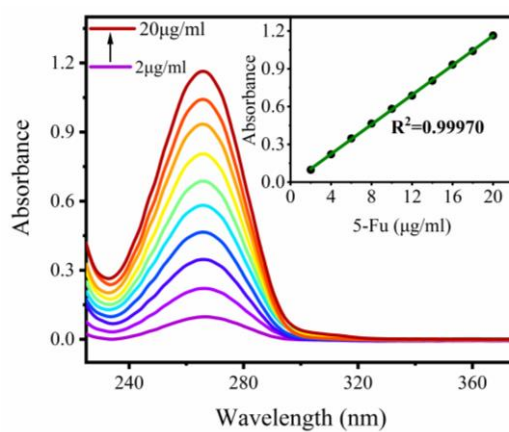


Figure S3 Standard curve of 5-Fu in ethanol

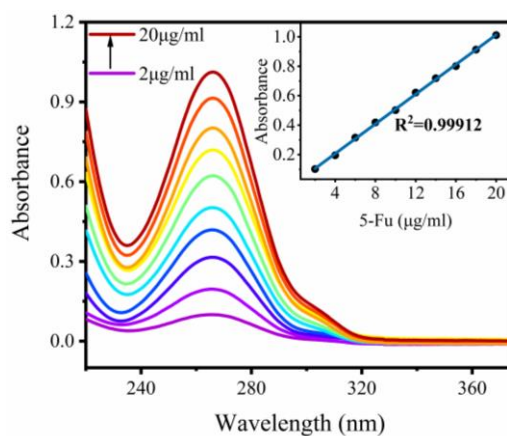


Figure S4 Standard curve of 5-Fu in PBS solution

4. Zeta potential

Table S2 Zeta potential of UiO-67-CDC, UiO-67-CDC-(CH₃)₂ and 5-Fu@UiO-67-CDC-(CH₃)₂

Sample	UiO-67-CDC	UiO-67-CDC-(CH ₃) ₂	5-Fu@UiO-67-CDC-(CH ₃) ₂
ζ (mV)	0.229	22.017	-0.106

5. The fitting of kinetic models

Table S3 Summary of four types of kinetic model fitting tables for drug sustained-release behavior under different pH conditions

PH	Model	Equation	Correlation coefficients (R ²)
PH=7.4	Korsmeyer-Peppas	$y=3297 \times x^{5.32054E-5} - 3297$	0.78103
	Higuchi	$y=0.14373 \times x^{0.5} + 0.2756$	0.60220
	Weibull	$y=0.87973 \times (1 - e^{-1.18077x})^{0.44827}$	0.99114
	First order	$y=0.87973 \times (1 - e^{-0.52931x})$	0.99213
PH=6.8	Korsmeyer-Peppas	$y=2397 \times x^{6.71748E-5} - 2397$	0.86138
	Higuchi	$y=0.13127 \times x^{0.5} + 0.20544$	0.68094
	Weibull	$y=0.76247 \times (1 - e^{-1.39451x})^{0.32206}$	0.98931
	First order	$y=0.76247 \times (1 - e^{-0.44911x})$	0.99050
PH=5	Korsmeyer-Peppas	$y=1803 \times x^{7.15013E-5} - 1803$	0.87101
	Higuchi	$y=0.09999 \times x^{0.5} + 0.09093$	0.70426
	Weibull	$y=0.5323 \times (1 - e^{-0.38331x})^{0.87146}$	0.97321
	First order	$y=0.5323 \times (1 - e^{-0.33404x})$	0.97618
PH=3	Korsmeyer-Peppas	$y=0.02064 \times x^{0.32101} + 0.00364$	0.55848
	Higuchi	$y=0.0103 \times x^{0.5} + 0.0121$	0.67984
	Weibull	$y=0.06253 \times (1 - e^{-0.32729x})^{0.64221}$	0.6323
	First order	$y=0.06253 \times (1 - e^{-0.21019x})$	0.67316

6. Scanning electron micrograph

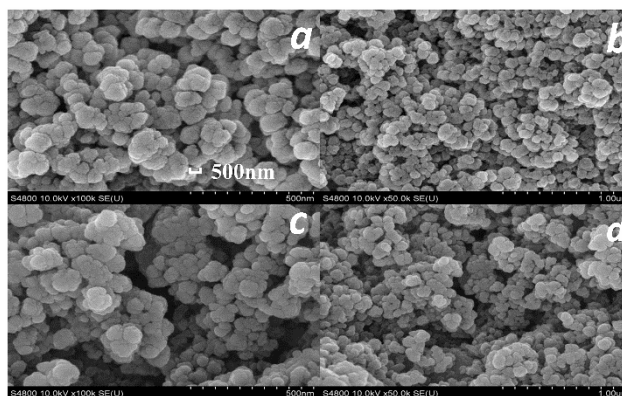


Figure S5 Scanning electron micrograph of UiO-67-CDC (a) (b), and UiO-67-CDC-(CH₃)₂ (c) (d).

7. Cell viability test

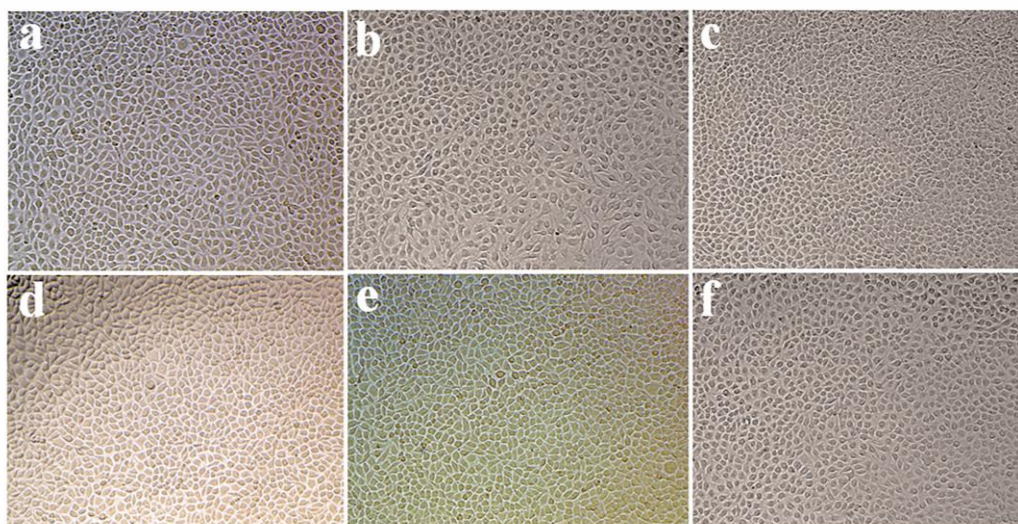


Figure S6 Morphology of L929 cells under microscope after treatment with different concentrations of UiO-67-CDC-(CH₃)₂, (a) control group, (b) 20 µg/ml, (c) 40µg/ml, (d) 60 µg/ml, (e) 80 µg/ml, (f) 100 µg/ml

^[i] Xiu-Chun Yi, Fu-Gui Xi, Kun Wang, Zhao Su, En-Qing Gao, Synthesis, structure and properties of zinc(II) coordination polymers with 9H-carbazole-2,7-dicarboxylic acid, *J. Solid State Chem*, 206 (2013) 293–299.