## SUPPORTING INFORMATION

## MESO-TETRA-(4-PYRIDYL)PORPHYRIN/PALLADIUM(II) COMPLEXES AS ANTICANCER AGENTS

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CCDC code	2088291
Empirical formula	$B_4C_{154}Cl_8F_{16}H_{146}N_8O_2P_8Pd_4\\$
Formula weight	3445.271
Temperature/K	150.03
Crystal system	monoclinic
Space group	P2 <sub>1</sub> /a
a/Å	21.2707(11)
b/Å	23.0586(12)
c/Å	31.8123(19)
α/°	90
β/°	93.226(4)
γ/°	90
Volume/Å <sup>3</sup>	15578.3(15)
Z	4
$\rho_{calc}g/cm^3$	1.469
µ/mm <sup>-1</sup>	6.312
F(000)	7029.2
Crystal size/mm <sup>3</sup>	0.113  imes 0.08  imes 0.012
Radiation	$Cu K\alpha (\lambda = 1.54178)$
$2\Theta$ range for data collection/°	4.74 to 90
Index ranges	$-20 \le h \le 27, -28 \le k \le 28, -38 \le l \le 38$
Reflections collected	185018
Independent reflections	12550 [ $R_{int} = 0.3483$ , $R_{sigma} = 0.2043$ ]
Data/restraints/parameters	12550/1824/1477
Goodness-of-fit on F <sup>2</sup>	1.477
Final R indexes [I>= $2\sigma$ (I)]	$R_1 = 0.1466, wR_2 = 0.3819$
Final R indexes [all data]	$R_1 = 0.1951, wR_2 = 0.4253$
Largest diff. peak/hole / e Å <sup>-3</sup>	2.59/-1.21

**Table S1.** Crystal data and structure refinement for the complex<br/> ${TPyP[PdCl(dppb)]_4}(PF_6)_4.$ 

Bonds (Å)	lengths (Å)	Bonds (Å)	lengths (Å)	Atoms	Angles (°)
Pd2-Cl2	2.35(1)	N1-C1	1.37(3)	P4-Pd2-Cl2	178.5(3)
Pd2-P3	2.26(1)	N1-C4	1.45(3)	P4-Pd2-P3	92.3(3)
Pd2-P4	2.25(1)	N2-C6	1.33(3)	P3-Pd2-Cl2	89.2(3)
P3-C231	1.83(1)	N2-C9	1.44(3)	N8-Pd2-P4	92.3(6)
P3-C241	1.85(2)	N3-C11	1.44(3)	N8-Pd2-P3	175.3(6)
P4-C221	1.83(2)	N3-C14	1.44(3)	C6-N2-C9	112.0(2)
P4-C211	1.85(2)	N4-C19	1.47(3)	C1-N1-C4	105.0(2)
Pd2-N8	2.08(2)	N4-C16	1.40(3)	C14-N3-C11	104.0(2)

**Table S2.** Selected bond lengths (Å) and angles (°) for the complex ${TPyP[PdCl(dppp)]_4}(BF_4)_4.$ 



Figure S1.FT-IR spectra (KBr pellets) of the metalloporphyrins (a) $\{TPyP[PdCl(dppe)]_4\}(PF_6)_4$ , (b) $\{TPyP[PdCl(dppp)]_4\}(PF_6)_4$ , (c) $\{TPyP[PdCl(dppb)]_4\}(PF_6)_4$  and (d) $\{TPyP[PdCl(dppf)]_4\}(PF_6)_4$ .



Figure S2. <sup>1</sup>H NMR spectra of the {TPyP[PdCl(dppe)]<sub>4</sub>}(PF<sub>6</sub>)<sub>4</sub> complex, in CDCl<sub>3</sub>.



**Figure S3.** <sup>1</sup>H NMR spectra of the {TPyP[PdCl(dppp)]<sub>4</sub>}(PF<sub>6</sub>)<sub>4</sub> complex, in CDCl<sub>3</sub>.



Figure S4. <sup>1</sup>H NMR spectra of the {TPyP[PdCl(dppb)]<sub>4</sub>}(PF<sub>6</sub>)<sub>4</sub> complex, in CDCl<sub>3</sub>.



Figure S5. <sup>1</sup>H NMR spectra of the {TPyP[PdCl(dppf)]<sub>4</sub>}(PF<sub>6</sub>)<sub>4</sub> complex, in CDCl<sub>3</sub>.



**Figure S6.** <sup>1</sup>H-<sup>1</sup>H COSY NMR contour map of the  $\{TPyP[PdCl(dppe)]_4\}(PF_6)_4 \text{ complex}, in CDCl_3.$ 



Figure S7.  $^{1}H^{-1}H COSY NMR$  contour map of the  $\{TPyP[PdCl(dppp)]_{4}\}(PF_{6})_{4}$  complex, in CDCl<sub>3</sub>.



**Figure S8.**  $^{1}H^{-1}H COSY NMR$  contour map of the  $\{TPyP[PdCl(dppb)]_{4}\}(PF_{6})_{4}$  complex, in CDCl<sub>3</sub>.



**Figure S9.**  $^{1}H^{-1}H \text{ COSY NMR contour map of the } {TPyP[PdCl(dppf)]_{4}}(PF_{6})_{4} \text{ complex, in CDCl}_{3}.$ 



Figure S10. UV-Vis spectrua for the complexes A -  $\{TPyP[PdCl(dppe)]_4\}(PF_6)_4, B - \{TPyP[PdCl(dppp)]_4\}(PF_6)_4, C - \{TPyP[PdCl(dppb)]_4\}(PF_6)_4, D - \{TPyP[PdCl(dppf)]_4\}(PF_6)_4, and E - free TPyP, in DMSO, at 298 K.$ 



**Figure S11.** Fluorescence emission spectra of the palladium/TPyP complexes in DMSO, at 298 K.



**Figure S12.** UV/Vis spectra of the complexes A - {TPyP[PdCl(dppe)]<sub>4</sub>}(PF<sub>6</sub>)<sub>4</sub>, B - {TPyP[PdCl(dppp)]<sub>4</sub>}(PF<sub>6</sub>)<sub>4</sub>, C - {TPyP[PdCl(dppb)]<sub>4</sub>}(PF<sub>6</sub>)<sub>4</sub> and D - {TPyP[PdCl(dppf)]<sub>4</sub>}(PF<sub>6</sub>)<sub>4</sub>, (initially 2  $\mu$ M) in DMSO (5%) / Tris–HCl solution (pH 7.4), with increasing amount of CT-DNA, solutions.



**Figure S13.** Full interaction map obtained for the crystal structure for the  $\{TPyP[PdCl(dppb)]_4\}(BF_4)_4$  complex, displaying the regions with probability to perform aromatic interactions.