

SUPPORTING INFORMATION

Supporting Information

Simple synthesis of [Ru(CO₃)(NHC)(*p*-cymene)] complexes and their use in transfer hydrogenation catalysis

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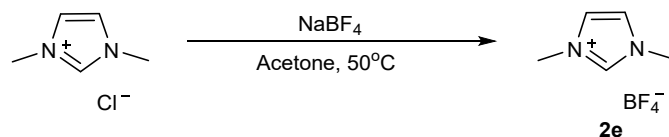
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Synthesis of IMe·HBF₄

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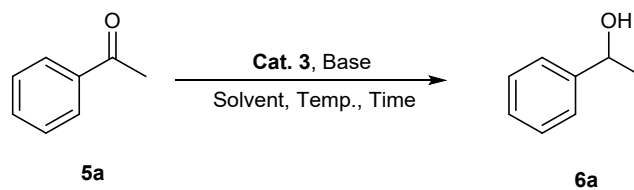
The synthesis process is similar to the literature.¹ In a 20 mL vial, 132.6 mg of IMe-HCl (1 mmol, 1 eq.) and 164.6 mg of NaBF₄ (1.2 mmol, 1.2 eq.) were stirred in 2 mL of acetone at 50 °C for 6 hours. The mixture was allowed to cool down to room temperature, filtered the suspension and washed the solid by 2 mL of acetone. The filtrate was collected and added extra 82.4 mg of NaBF₄ (0.75 mmol, 0.75 eq.) to the solvent, stirred at 50 °C for 3h. Filtered through a microfilter and washed with 6 mL of acetone (3 x 2 mL), and concentrated under reduced pressure. The product was dried in high vacuum overnight. Yield of product is 81 %.

¹H NMR (300 MHz, Acetone) δ 8.90 (s, 1H, N=CHN), 7.66 (d, *J* = 1.6 Hz, 2H, CH=CH), 4.02 (d, *J* = 0.5 Hz, 7H, NCH₃). ¹⁹F NMR: δ -153.24 (s, 4F, BF₄).

Selected Optimization of the Reaction Conditions

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Table S1 Optimization of the reaction conditions for hydrogenation.



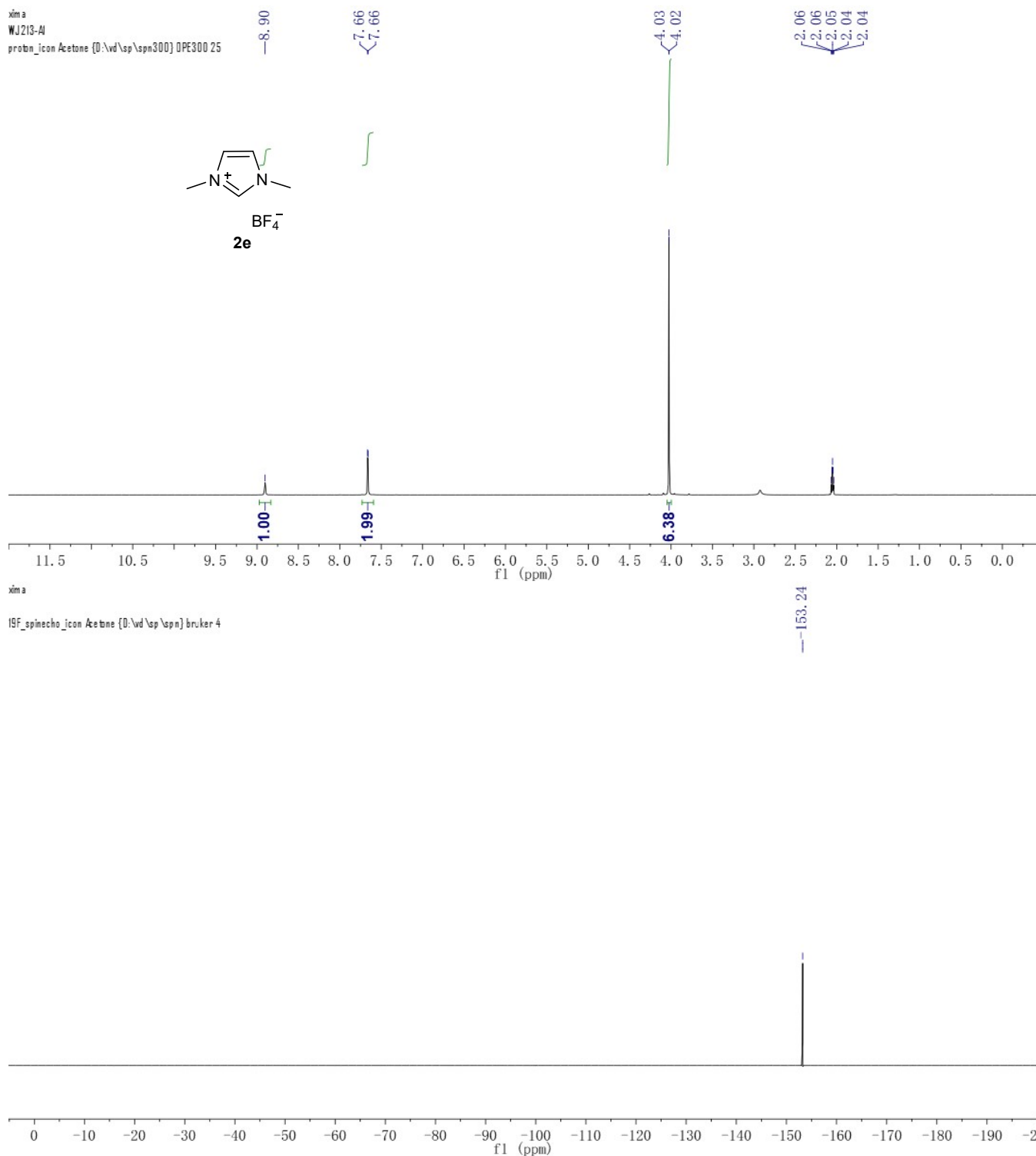
Entry	Ru-NHC	Base	Solvent	T(°C)	Time (h)	Yield (%) ^b
1	3a	NaOH (20%)	<i>i</i> PrOH	100	16	84
2	3a	K ^t BuO (20%)	<i>i</i> PrOH	100	16	91
3	3a	K ₃ PO ₄ (20 %)	<i>i</i> PrOH	100	16	trace
4	3a	K ₂ CO ₃ (20%)	<i>i</i> PrOH	100	16	47
5	3a	KOH (10%)	<i>i</i> PrOH	100	16	93
6	3a	KOH (5%)	<i>i</i> PrOH	100	16	76
7	3a	KOH (5%)	<i>i</i> PrOH	100	36	83
8	3a	KOH (2.5%)	<i>i</i> PrOH	100	16	56
9	-	KOH (10%)	<i>i</i> PrOH	80	4	25
10	3e	KOH (10%)	EtOH	80	4	-
11	3e	KOH (10%)	MeOH	80	4	-
12	3e	KOH (10%)	H ₂ O	80	4	-
13	3e(2 mol%)	KOH (10%)	<i>i</i> PrOH	80	2	94
14	3e(0.5 mol%)	KOH (10%)	<i>i</i> PrOH	80	2	61

^a Reaction conditions: **5a** (0.5 mmol, 1 eq.), **3** (1 mol%), Base, Solvent (1.5 mL) in a 4 mL vial. ^b NMR yields using 1,3,5-trimethoxybenzene as internal standard.

NMR spectra

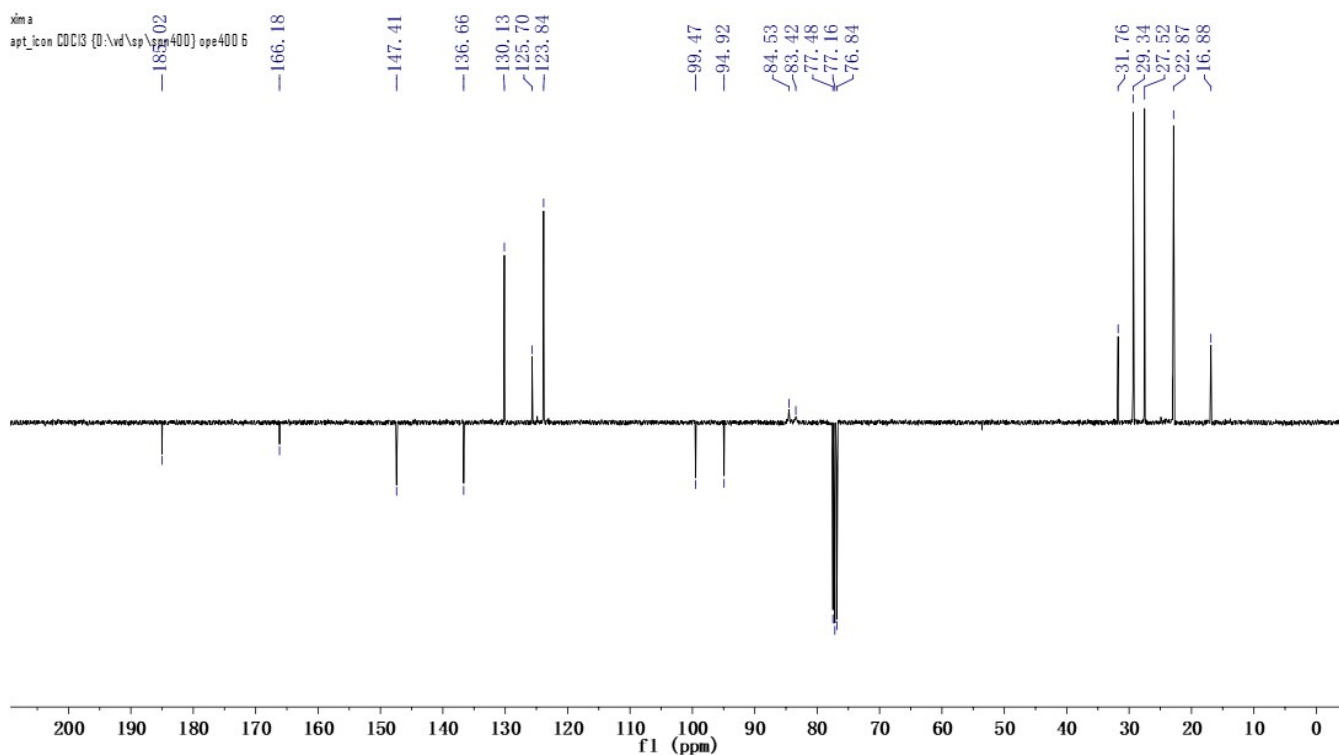
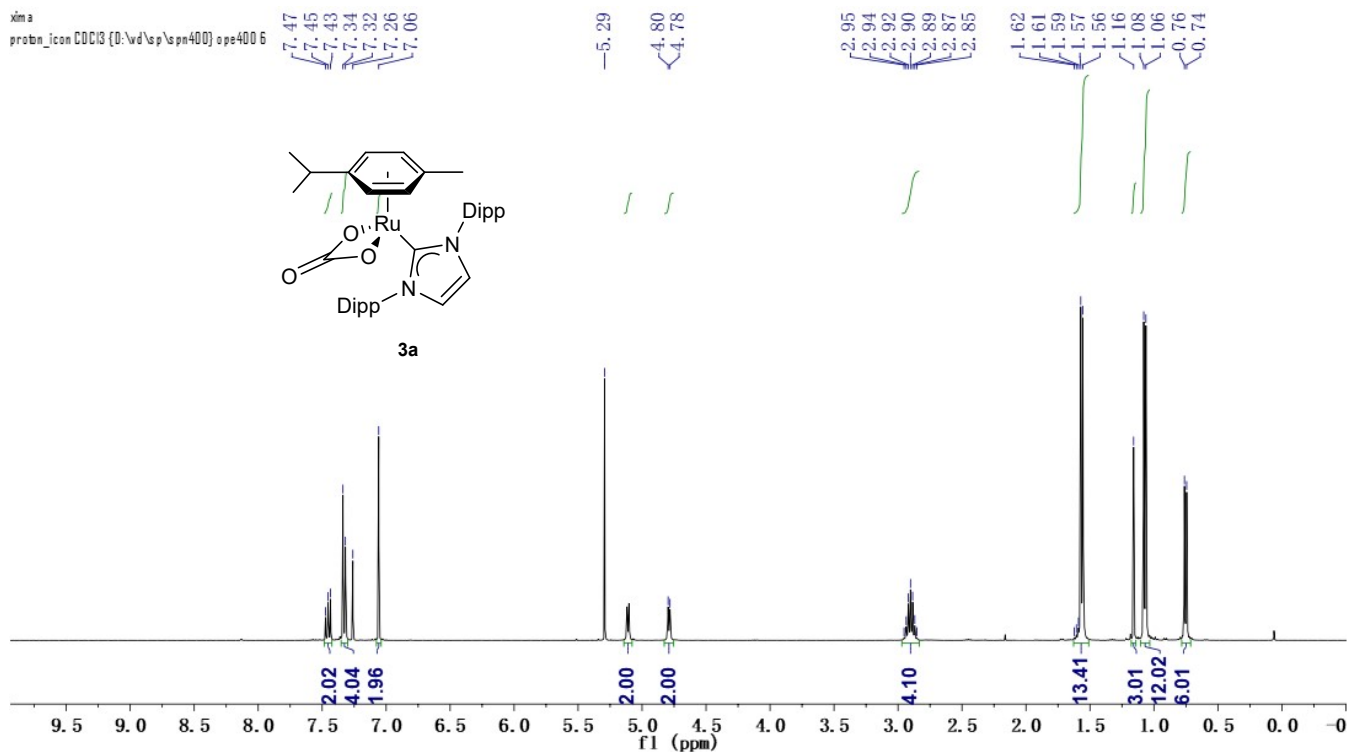
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^1H and ^{19}F NMR of $\text{IMe}\cdot\text{HBF}_4$ (**2e**)



^1H and ^{13}C { ^1H } apt NMR of $[\text{Ru}(\text{CO}_3)(\text{IPr})(p\text{-cymene})]$ (**3a**)

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^1H and ^{13}C $\{^1\text{H}\}$ apt NMR of $[\text{Ru}(\text{CO}_3)(\text{IMes})(p\text{-cymene})]$ (**3b**)

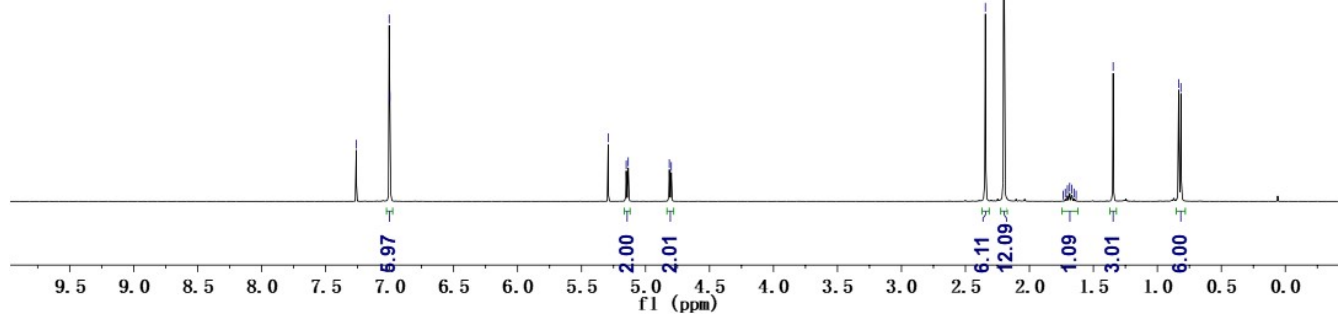
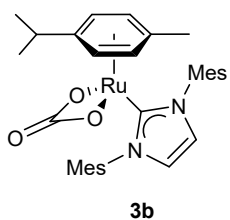
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xima
 protbn_icon CDCl3 (D:\vd\sp\spn400) ope400 5

7.26
 7.00
 7.00

5.29
 5.15
 5.14
 4.81
 4.80

2.34
 2.20
 1.72
 1.70
 1.69
 1.67
 1.65
 1.63
 0.83
 0.82



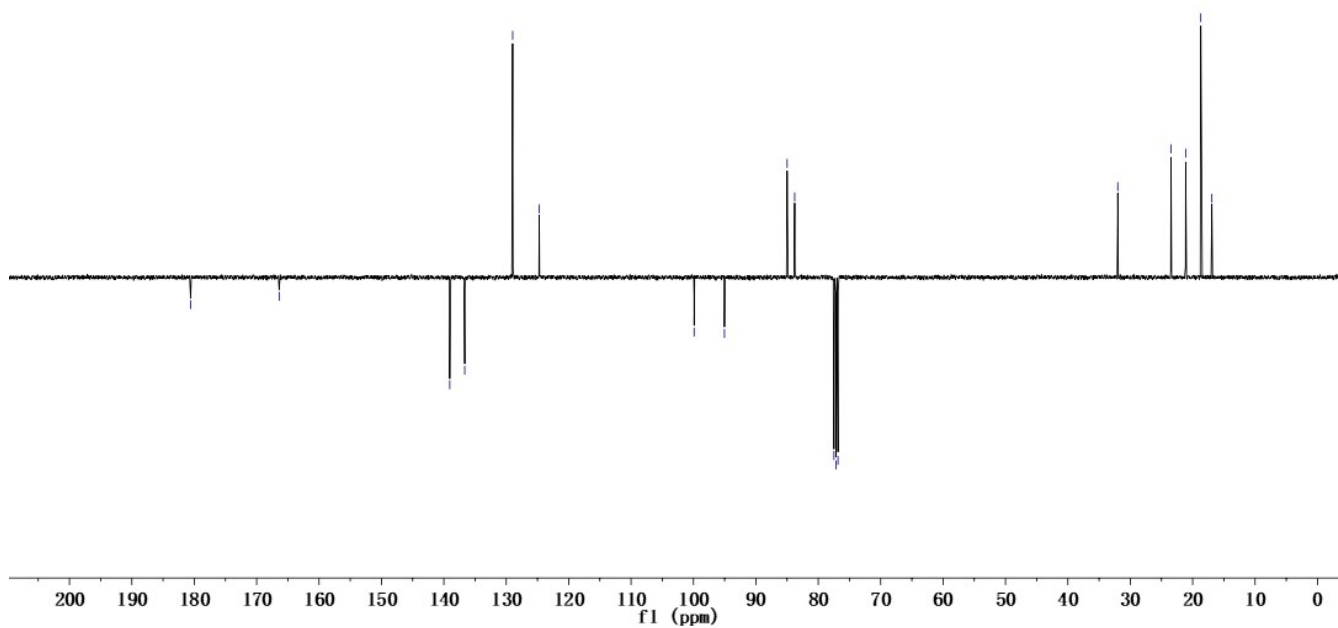
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180.57
 166.39

139.05
 136.65
 128.97
 124.71

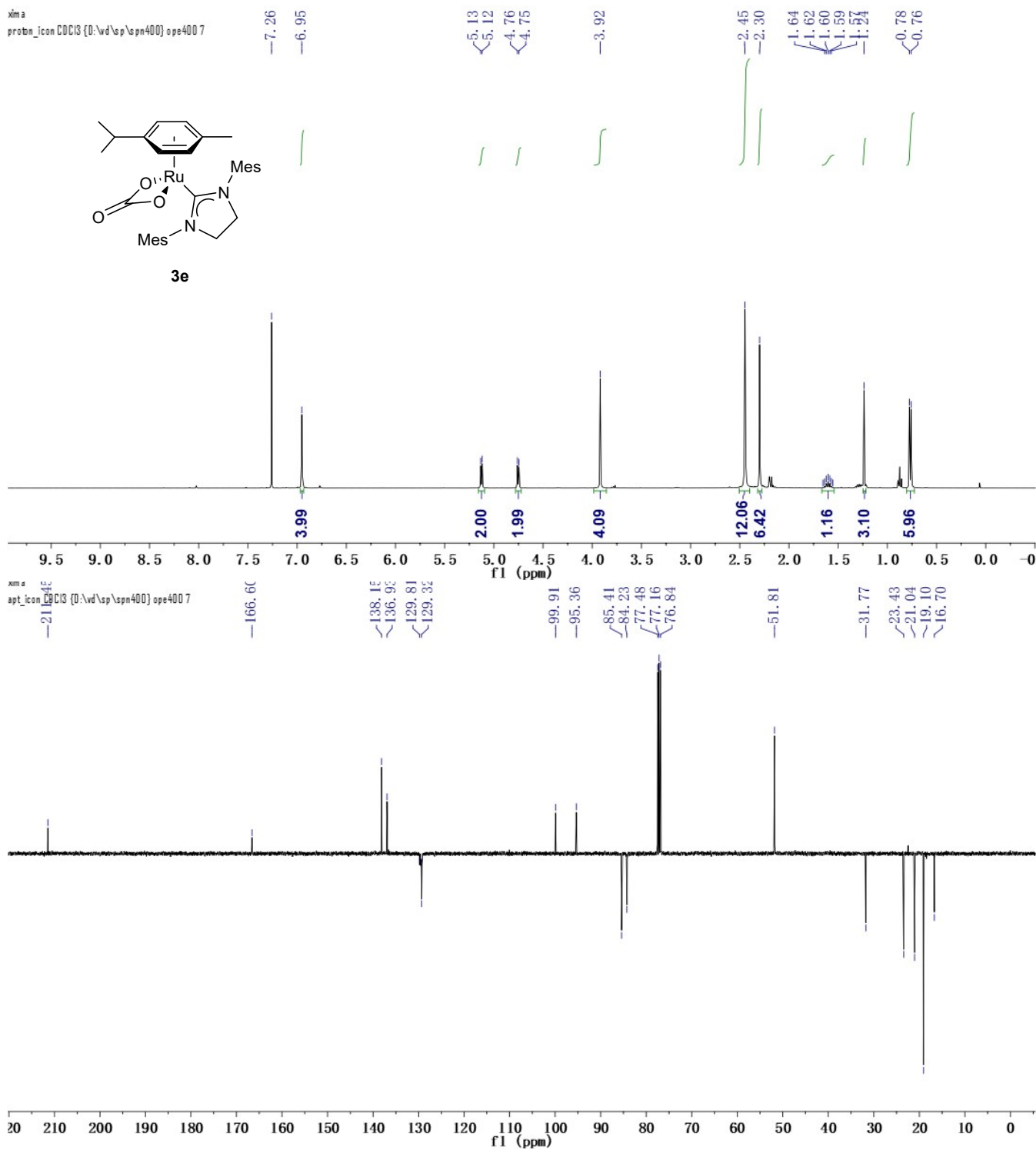
99.89
 95.03
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 83.78
 77.48
 77.16
 76.84

32.00
 23.47
 21.12
 18.70
 16.92



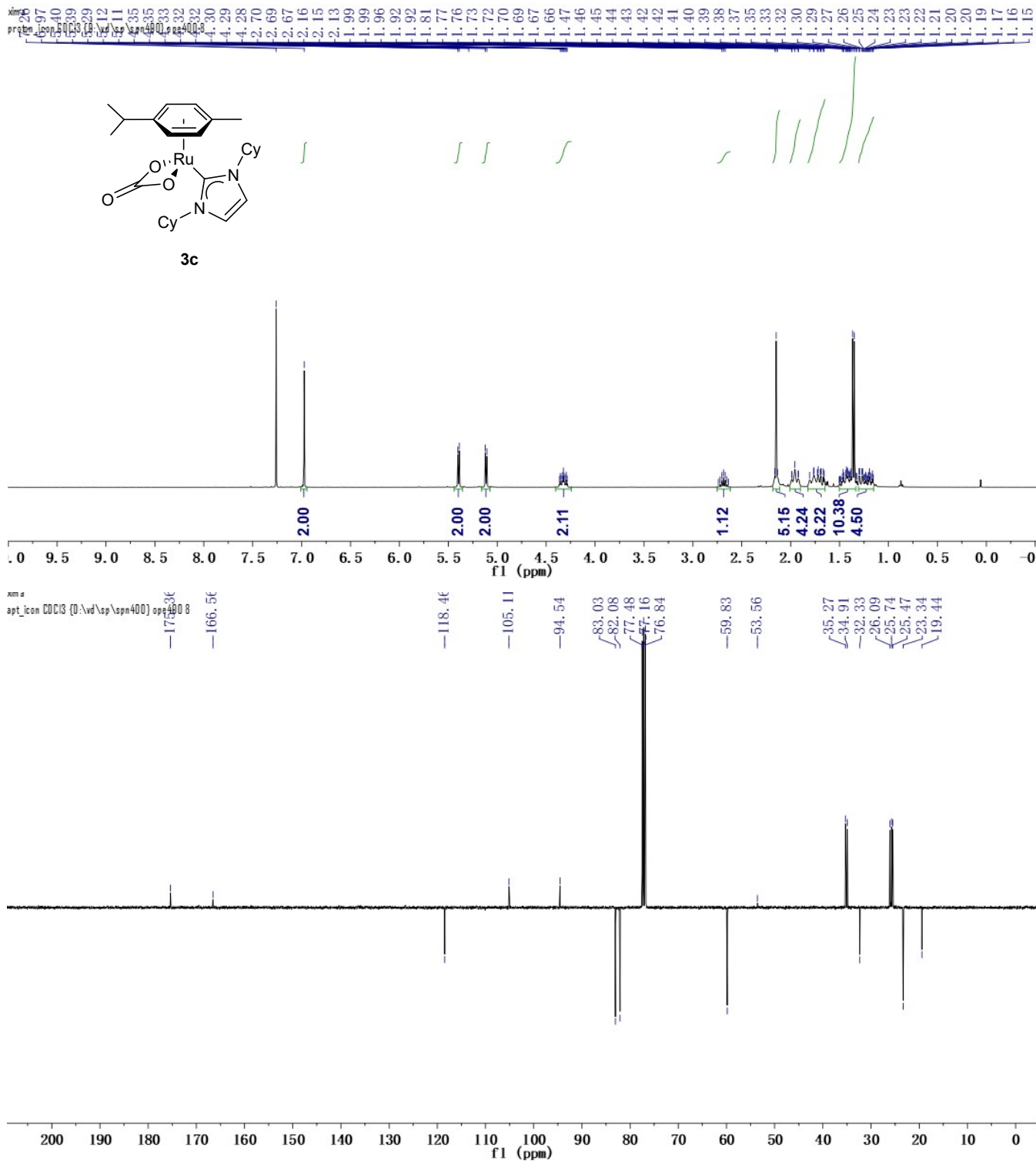
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^1H and ^{13}C $\{^1\text{H}\}$ apt NMR of $[\text{Ru}(\text{CO}_3)(\text{SiMes})(p\text{-cymene})]$ (**3c**)



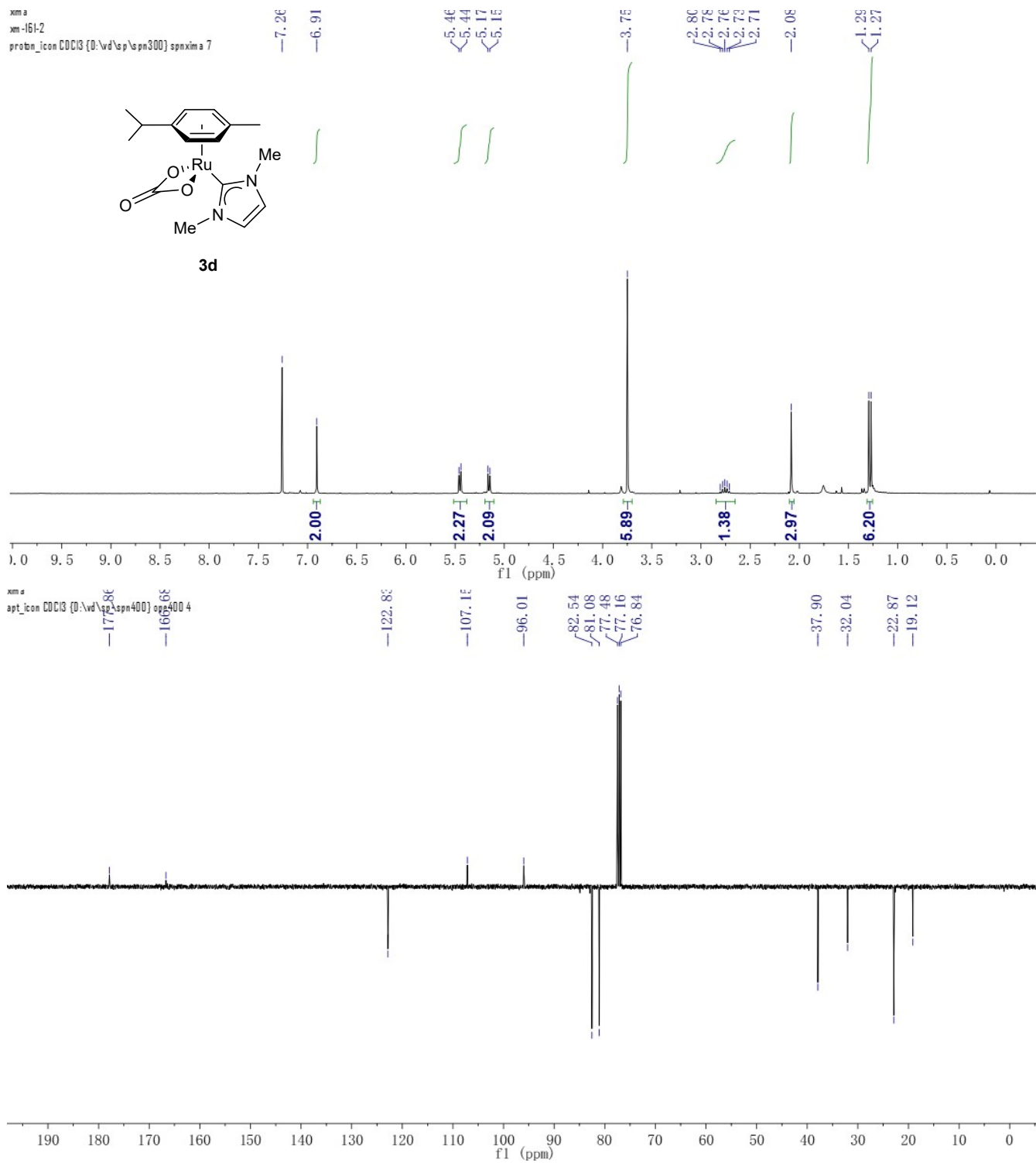
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^1H and ^{13}C { ^1H } apt NMR of $[\text{Ru}(\text{CO}_3)(\text{ICy})(p\text{-cymene})]$ (**3d**)



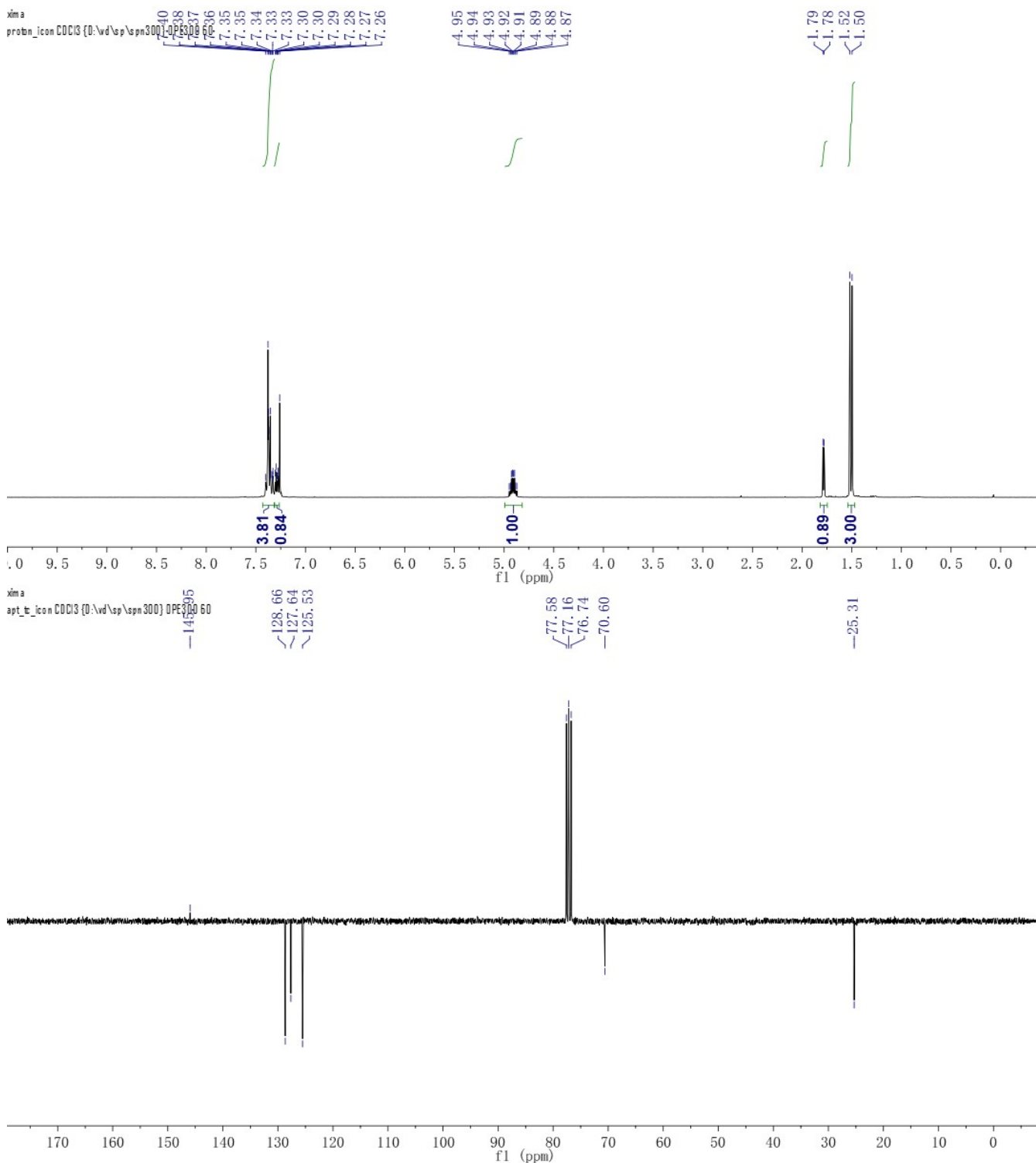
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^1H and ^{13}C { ^1H } apt NMR of $[\text{Ru}(\text{CO})_3(\text{IME})(p\text{-cymene})]$ (**3e**)



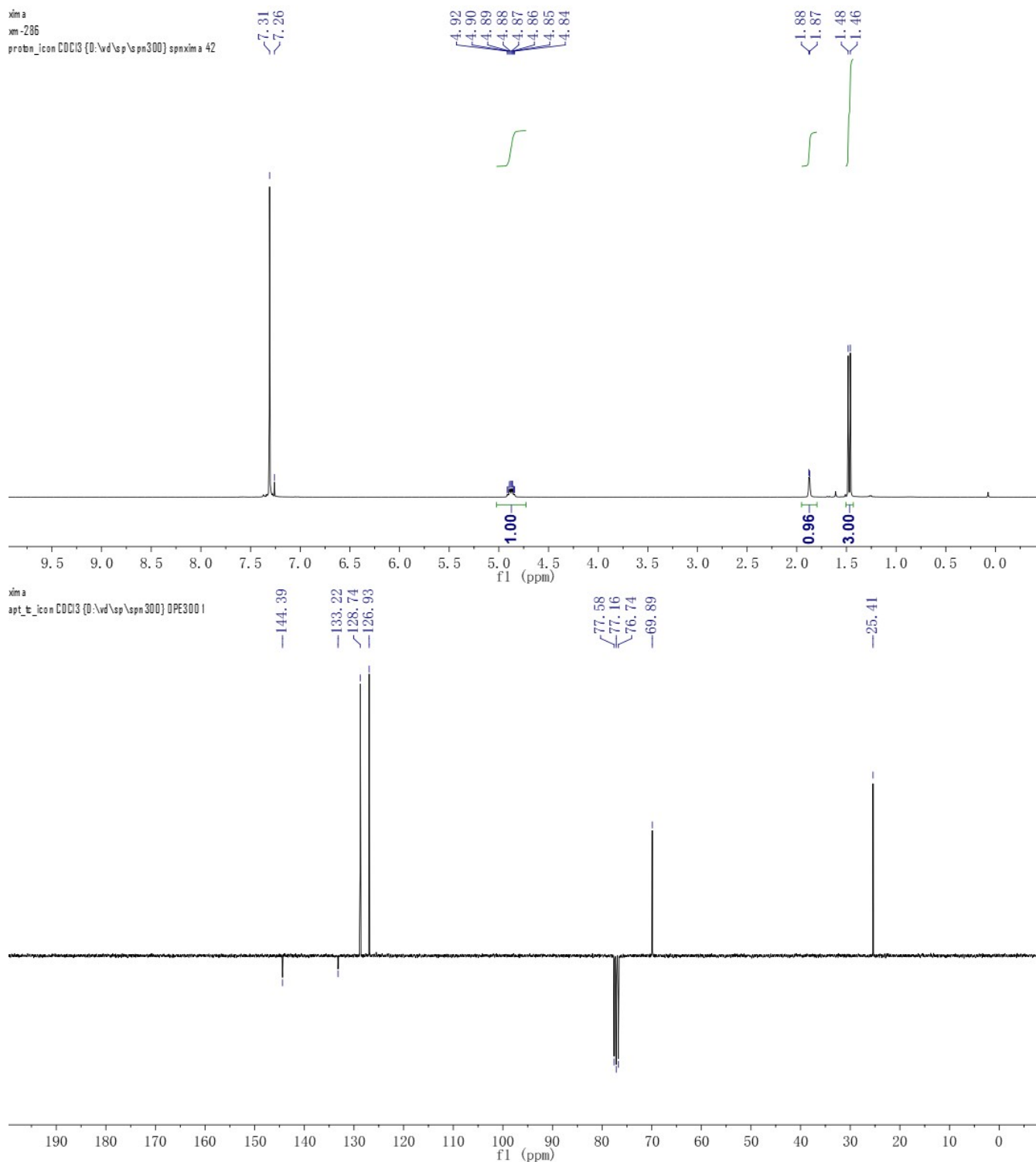
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^1H and ^{13}C { ^1H } apt NMR of α -Methylbenzenemethanol (**6a**)



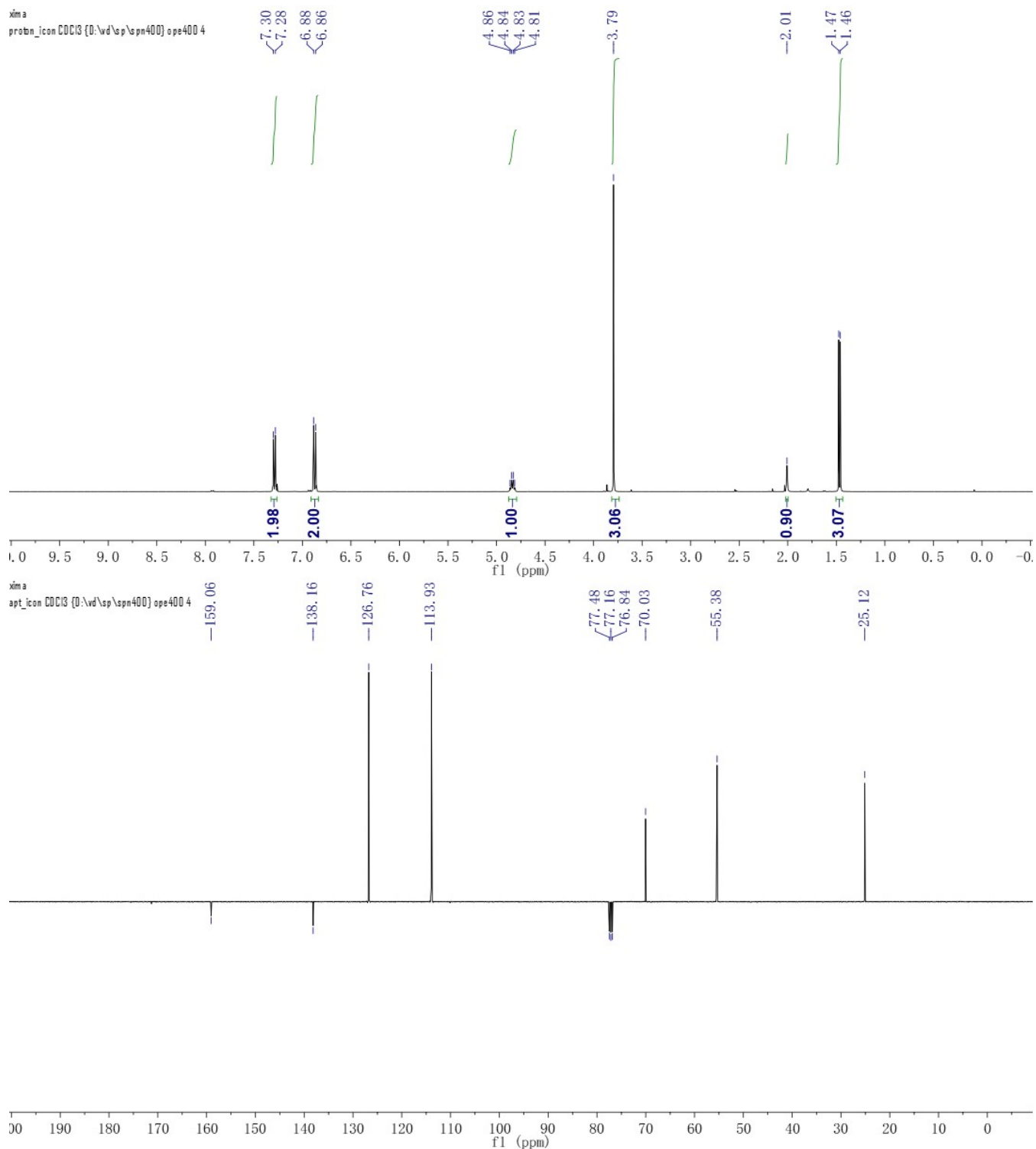
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^1H and ^{13}C $\{^1\text{H}\}$ apt NMR of 4-chlorobenzenemethanol (**6b**)



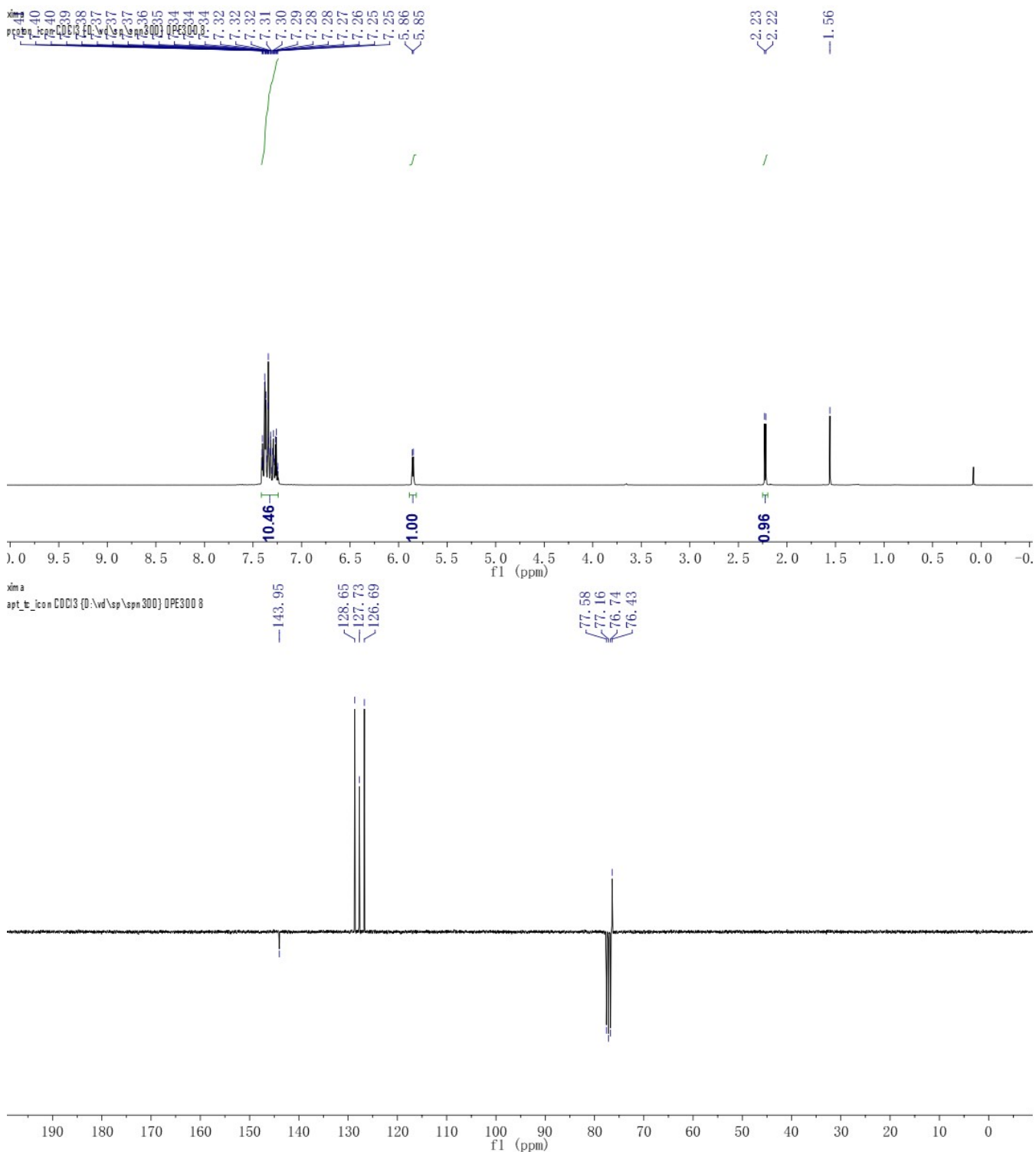
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^1H and ^{13}C $\{^1\text{H}\}$ apt NMR of 4-Methoxy- α -methylbenzyl alcohol (**6c**)



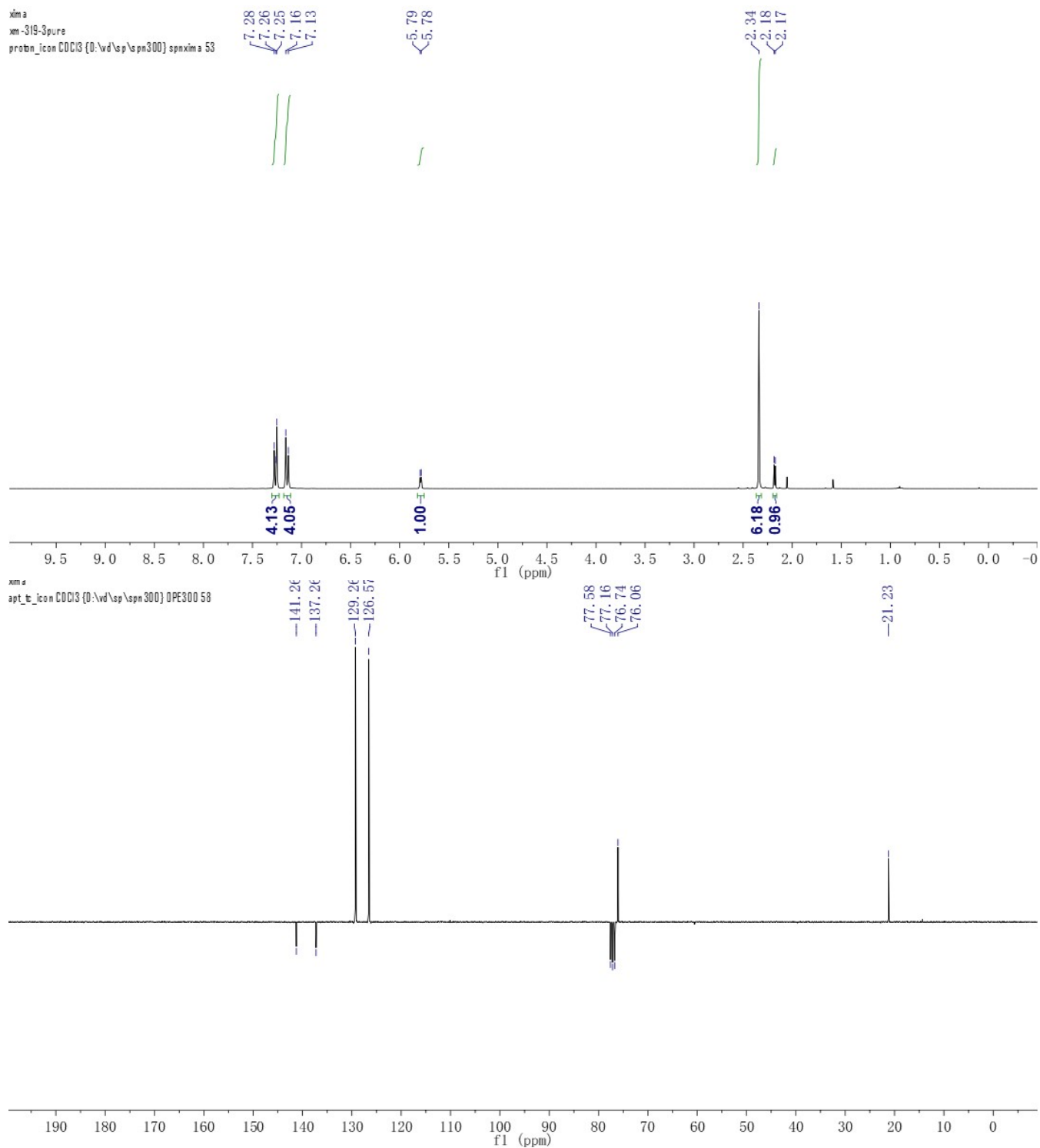
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^1H and ^{13}C $\{^1\text{H}\}$ apt NMR of Diphenylmethanol (**6d**)



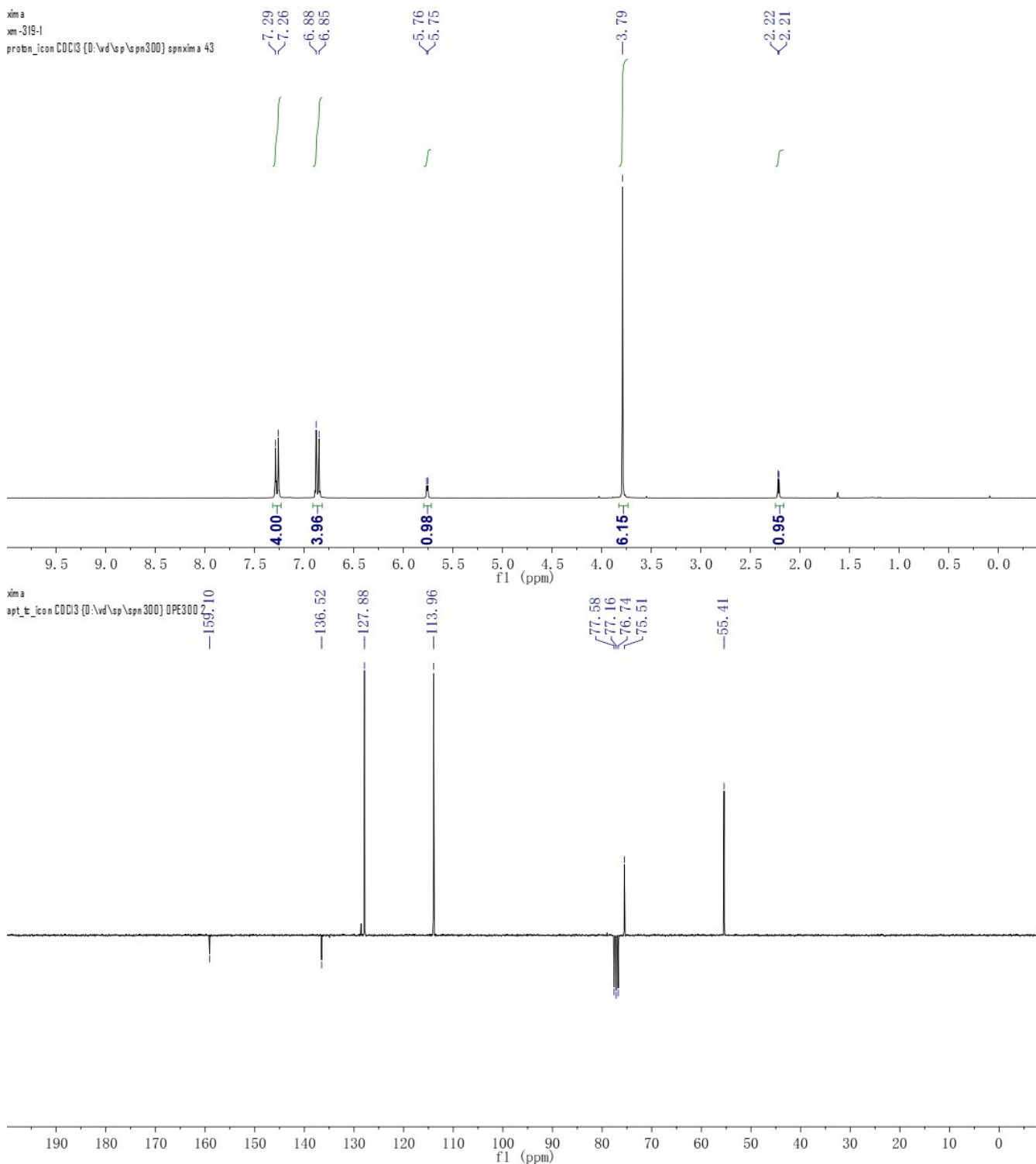
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^1H and ^{13}C $\{^1\text{H}\}$ apt NMR of 4,4'-Dimethylbenzhydrol (**6e**)



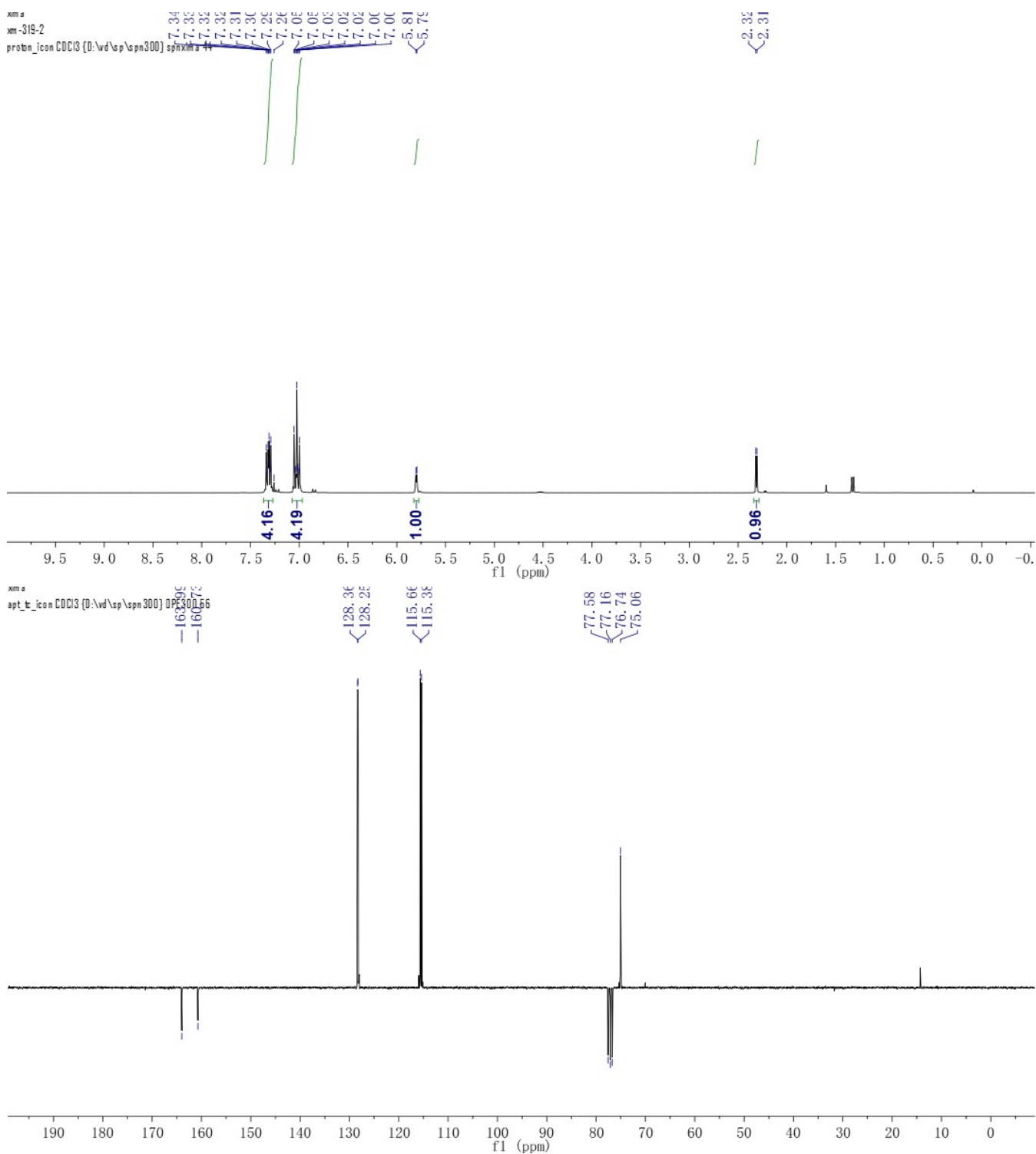
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^1H and ^{13}C $\{^1\text{H}\}$ apt NMR of 4,4'-Dimethoxybenzhydrol (**6f**)



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^1H and ^{13}C { ^1H } apt NMR of 4, 4'-Difluorobenzhydrol (**6g**)



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References

1. O. D. Bakulina, M. Yu. Ivanov, S. A. Prikhod'ko, S. Pylaeva, I. V. Zaytseva, N. V. Surovtsev, N. Y. Adonin and M. V. Fedin, *Nanoscale*, **2020**, *12*, 19982–19991.