Formation of plutonium (IV) silicate species in very alkaline reactive media

Paul Estevenon^{1,2,3,4}, Thomas Dumas^{1,*}, Pier Lorenzo Solari⁵, Eleonore Welcomme¹, Stephanie Szenknect², Adel Mesbah², Kristina O. Kvashnina^{3,4}, Philippe Moisy¹, Christophe Poinssot¹, Nicolas Dacheux^{2,*}

¹ CEA, DES, ISEC, DMRC, Univ Montpellier, Marcoule, France.

² ICSM, Univ Montpellier, CNRS, CEA, ENSCM, Bagnols-sur-Cèze, France

³ The Rossendorf Beamline at the ESRF, CS40220, 38043 Grenoble Cedex 9, France.

⁴ Helmholtz Zentrum Dresden-Rossendorf (HZDR), Institute of Resource Ecology, P.O. Box 510119, 01314, Dresden, Germany.

⁵ Synchrotron SOLEIL L'Orme des Merisiers, Saint-Aubin, BP 48, F-91192 Gif-sur-Yvette Cedex France

SUPPORTING INFORMATION

Table S1Results of the pH and UV-visible monitoring on a Na2SiO3 2 mol·L⁻¹ aqueous
solution depending on the addition of Pu(IV) nitric solution ($C_{HNO3} = 1.5 \text{ mol·L}^{-1}$;
 $C_{Pu} = 0.3 \text{ mol·L}^{-1}$).

V(Na ₂ SiO ₃ 2M)	V(Pu 0.3M in HNO ₃ 1.5M)	рН	UV-visible spectroscopy observations	
0.800 mL	0 mL	13.3	-	
	0.100 mL	13.3	Pu(IV)-silicate species	
	0.100 mL	13.2	Pu(IV)-silicate species	
	0.100 mL	13.1	Pu(IV)-silicate species	
	0.400 mL	13.0	Pu(IV)-silicate species	
	0.500 mL	12.5	Pu(IV)-silicate species (weak intensity)	
	0.600 mL	12.0	No Pu species observed	
	0.700 mL	11.4	No Pu species observed	
	0.900 mL	9.6	No Pu species observed	
	1.000 mL	2.5	No Pu species observed	

Table S2Interatomic distances for ThSiO4, USiO4, CeSiO4 and PuSiO4 (zircon type silicates,
space group I41/amd 1).

	Interatomic distances (ASiO ₄)				A. h	
	ThSiO ₄ ²	USiO ₄ ³	CeSiO ₄ ⁴	PuSiO ₄ ⁵	Abundance	
A-O ₁	2.46 Å	2.32 Å	2.27 Å	-	4	
A-O ₂	2.56 Å	2.51 Å	2.37 Å	-	4	
A-Si _{1 (bidentate)}	3.16 Å	3.13 Å	3.10 Å	3.11 Å	2	
A-Si _{2 (monodentate)}	3.91 Å	3.83 Å	3.81 Å	3.79 Å	4	
A-A	3.91 Å	3.83 Å	3.81 Å	3.79 Å	4	

Plutonium (IV)-silicate species (this study)							
Reference	R (Å)	N	σ^2 (Å ²)	$\Delta E_{k=0}$	F		
Pu-O ₁	2.23 (2) Å	4 *	0.007 (3)	- 1.2 eV	0.02		
Pu-O ₂	2.34 (3) Å	4 *	0.007 (3)				
Pu-Si ₁	3.21 (6) Å	1.1 (9)	0.009*				
Pu-Si ₂	3.46 (3) Å	5.4 (22)	0.012*				
Pu-Pu	3.77 (3) Å	6.0 (21)	0.020 (26)				
Filtered 0.45 µm							
Pu-O ₁	2.20 (1) Å	4 *	0.005 (2)	-2.8 eV	0.03		
Pu-O ₂	2.33 (2) Å	4 *	0.005 (2)				
Pu-Si ₁	3.04 (7) Å	2.1 (6)	0.009*				
Pu-Si ₂	3.49 (3) Å	5.3 (19)	0.012*				
Pu-Pu	3.73 (3) Å	5.1 (39)	0.017 (8)				
Filtered 100 kDa							
Pu-O ₁	2.24 (37) Å	4 *	0.007 (3)	- 2.7 eV	0.02		
Pu-O ₂	2.30 (38) Å	4 *	0.007 (3)				
$Pu-Si_1$	3.26 (14) Å	0.7 (18)	0.009*				
Pu-Si ₂	3.47 (5) Å	6.1 (39)	0.012*				
Pu-Pu	3.73 (4) Å	6.4 (32)	0.020(1)				
1 year-old							
Pu-O ₁	2.23 (17) Å	4 *	0.007 (3)	-2.3 eV	0.04		
Pu-O ₂	2.29 (18) Å	4 *	0.007 (3)				
Pu-Si ₁	3.26 (8) Å	1.1 (6)	0.009*				
Pu-Si ₂	3.46 (3) Å	8.8 (19)	0.012*				
Pu-Pu	3.75 (3) Å	3.9 (39)	0.013 (7)				
	Thorium	(IV)-silicat	e colloids ⁶				
Th/Si = 1.4	R (Å)	N	σ^2 (Å ²)	$\Delta E_{k=0}$	F		
Th-O	2.39 Å	8.9	0.016	- 0.9 eV	0.10		
Th-Si	3.26 Å	1.3	0.011				
Th/Si = 1.7							
Th-O	2.38 Å	8.6	0.016	- 1.6 eV	0.07		
Th-Si	3.25 Å	1.6	0.013				
Th/Si = 5.8							
Th-O	2.36 Å	8.3	0.018	- 2.1 eV	0.09		
Th-Si	3.23 Å	2.5	0.011				
	Uranium (IV)-silicate colloids ⁷						
Si/U = 0.83	R (Å)	N	σ^2 (Å ²)	$\Delta E_{k=0}$	F		
U-O ₁	2.25 Å	4.1	0.0082	- 19 eV	0.19		
U-O ₂ /Si	2.81 Å	2.7	0.0082				
Ū-Ū	3.85 Å	2.6	0.0067	-6 eV			
Si/U = 1.68							
U-O ₁	2.23 Å	2.7	0.0081	- 16 eV	0.16		
U-O ₂ /Si	2.83 Å	3.5	0.0081				

Table S3Structural parameters for the Pu(IV)-silicate species determined in this study and
reported in the litterature for Th(IV)-, U(IV)- and Np(IV)-silicates colloids. The
values fixed for the simulations have been marked by an asterisk.

Uranium (IV)-silicate colloids ⁸						
$4 \text{ mM}_{\text{Si}} \text{ pH} = 9$	R (Å)	Ν	σ^2 (Å ²)	$\Delta E_{k=0}$	F	
U-O ₁	2.26 (1) Å	4.3	0.009(1)	4.1 (10) eV	0.0124	
$U-O_2$	2.44 (1) Å	3.8	0.009(1)			
U-Si ₁	3.17 (2) Å	1.1	0.010 (3)			
U-Si ₂	3.70 (3) Å	2.5	0.013 (4)			
U-U	3.78(5) Å	1	0.013(7)			
$4 \text{ mM}_{\text{Si}} \text{ pH} = 10.5$						
U-O ₁	2.28 (1) Å	4.5	0.006 (1)	5.5 (9) eV	0.0076	
$U-O_2$	2.46 (1) Å	3.5	0.006(1)			
U-Si ₁	3.18 (2) Å	1.6	0.011 (3)			
U-Si ₂	3.73 (4) Å	2	0.014 (5)			
U-U	3.79 (4) Å	1.8	0.014 (5)			
$4 \text{ mM}_{\text{Si}} \text{ pH} = 12$						
U-O ₁	2.29 (1) Å	4.4	0.007(1)	6.1 (6) eV	0.0042	
$U-O_2$	2.46 (1) Å	3.0	0.007(1)			
U-Si ₁	3.19 (2) Å	1.6	0.013 (3)			
U-Si ₂	3.71 (5) Å	0.8	0.012 (7)			
U-U	3.84 (2) Å	2.8	0.013 (2)			
$2 \text{ mM}_{\text{Si}} \text{ pH} = 9$						
U-O ₁	2.28 (1) Å	3.7	0.008 (1)	6.5 (5) eV	0.0024	
$U-O_2$	2.47 (1) Å	3.9	0.008(1)			
U-Si ₁	3.19 (1) Å	1	0.009 (2)			
U-Si ₂	3.76 (3) Å	1	0.013 (2)			
U-U	3.85 (1) Å	2.5	0.013 (4)			
$2 \text{ mM}_{\text{Si}} \text{ pH} = 10.5$						
U-O ₁	2.28 (1) Å	4.4	0.007(1)	4.5 (8) eV	0.0124	
$U-O_2$	2.45 (1) Å	3.5	0.007(1)			
U-Si ₁	3.17 (2) Å	0.9	0.007 (3)			
U-Si ₂	3.74 (4) Å	1.5	0.011 (5)			
U-U	3.80 (3) Å	2.6	0.013 (3)			
$2 \text{ mM}_{\text{Si}} \text{ pH} = 12$						
$U-O_1$	2.30 (1) Å	4.7	0.005 (1)	6.2 (8) eV	0.0121	
$U-O_2$	2.46 (1) Å	3.3	0.005 (1)			
U-Si ₁	3.18 (3) Å	1	0.009 (5)			
U-Si ₂	3.75 (5) Å	1.8	0.014 (7)			
U-U	3.86 (2) Å	5	0.015 (2)			
Neptunium (IV)-silicate colloids ⁹						
	R (Å)	Ν	σ^2 (Å ²)	$\Delta E_{k=0}$	F	
Np-O	2.28(1) Å	7.1(2)	0.017(4)	-9.3 eV	0.07	
Np-Si	3.11(1) Å	1.3(1)	0.0091(2)			
Np-Np	3.75(2) Å	1.1(1)	0.0098(1)			



Figure S1 pH monitoring of a 2 mol·L⁻¹ aqueous Na₂SiO₃ solution depending on the addition of Pu(IV) mother solution in nitric acid media ($C_{HNO_3} = 1.5 \text{ mol·L}^{-1}$; $C_{Pu} = 0.3 \text{ mol·L}^{-1}$), according to **Table S1**. The measured plutonium concentration was determined from the UV-visible spectra by Beer-Lambert law and the expected concentration is corresponding to the plutonium amount added to the system. The expected plutonium concentration is corresponding to the concentration expected by dilution of the mother solution (without precipitation).



Figure S2 UV-visible spectra of a Pu(IV)-silicate solution (pH = 13.2, $C_{Na_2SiO_3} \approx 2 \text{ mol}\cdot\text{L}^{-1}$, $C_{Pu} = 3 \times 10^{-2} \text{ mol}\cdot\text{L}^{-1}$) freshly prepared and after a 3 months ageing.



Figure S3 Pu(IV)-silicate solution (pH = 13.2, $C_{Na_2SiO_3} \approx 2 \text{ mol} \cdot \text{L}^{-1}$) filtrated at 0.1 µm (a.) and 3 kDa (b.).



Figure S4 EXAFS spectra 2 Å < k < 11 Å (a) and respective Fourier Transform (b) at plutonium L_{III} edge for a Pu(IV)-silicate solution filtered at 0.45 μ m.



Figure S5 EXAFS spectra 2 Å < k < 11 Å (a) and respective Fourier Transform (b) at plutonium L_{III} edge for a Pu(IV)-silicate solution filtered at 100 kDa.



Figure S6 EXAFS spectra 2 Å < k < 11 Å (a) and respective Fourier Transform (b) at plutonium L_{III} edge for a Pu(IV)-silicate solution filtered at 100 kDa and 1 year old aged.



(a)

(b)



Figure S7 Alpha spectrometer measurements for the 100 kDa filtered Pu-silicate colloids samples (10 μ L of sample after its dilution by a factor of 3200) (a) and corresponding platelets with the deposited sample (b).

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