

Supporting Information

Ru doping induces the construction of the unique core-shell microflower self-supporting electrocatalyst for highly efficient overall water splitting

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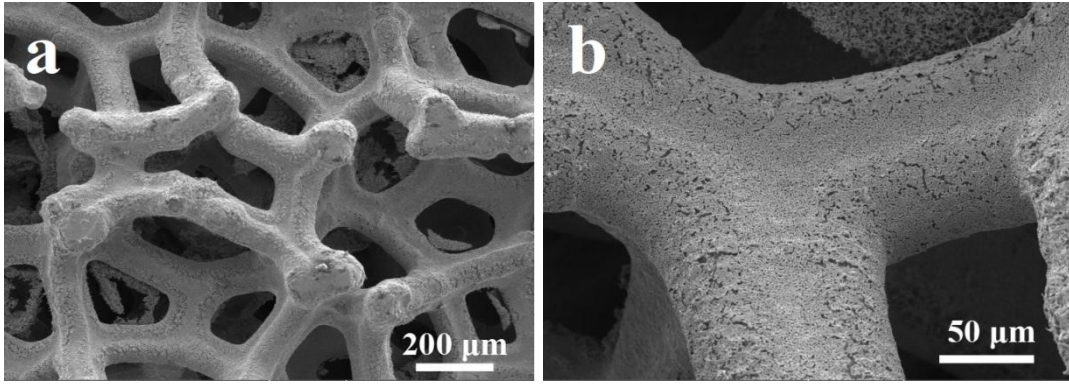


Fig. S1. SEM images of IF.

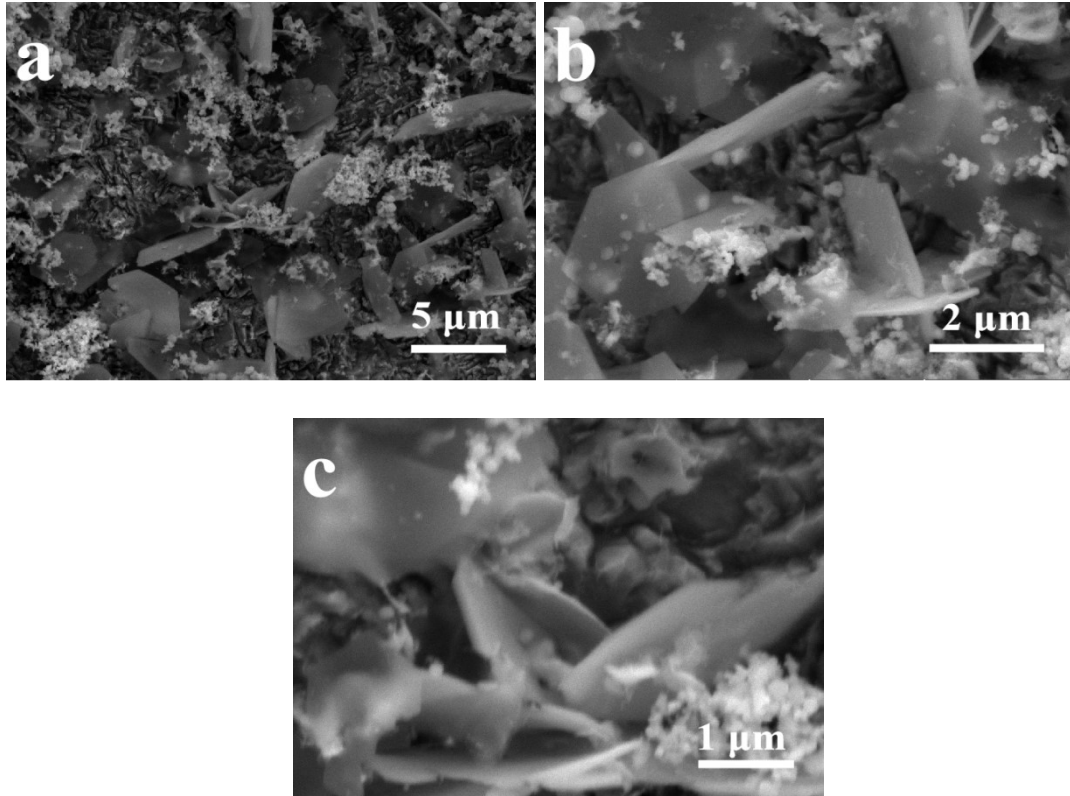


Fig. S2. SEM images of Fe-LDH/IF.

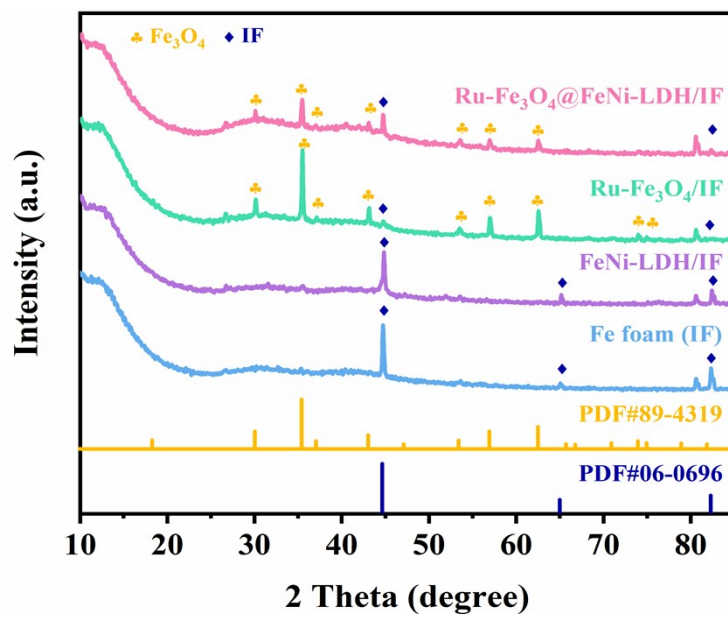


Fig. S3. XRD pattern of different samples.

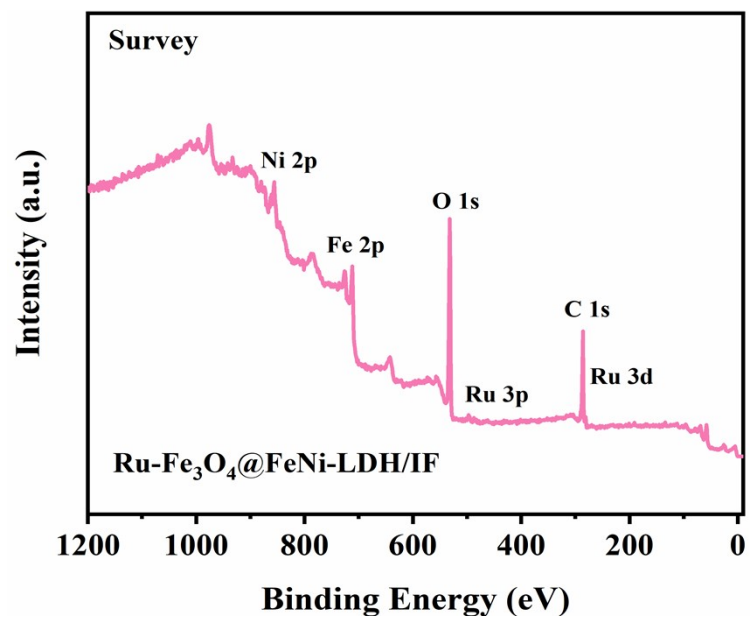


Fig. S4. XPS survey spectrum of Ru-Fe₃O₄@FeNi-LDH/IF.

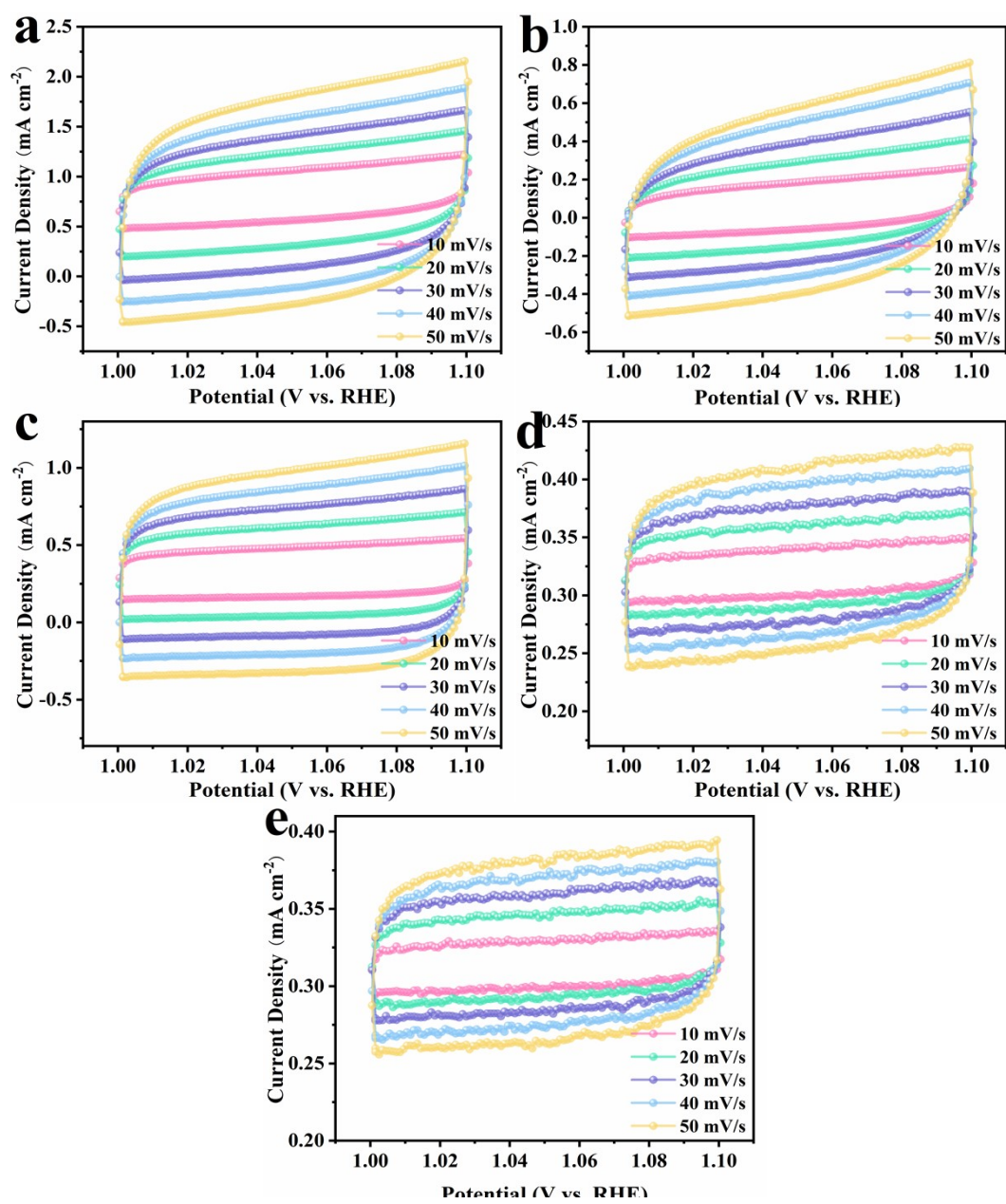


Fig. S5. Cyclic voltammetry curves of (a) Ru-Fe₃O₄@FeNi-LDH/IF, (b) Ru-Fe₃O₄/IF, (c) FeNi-LDH/IF, (d) Fe-LDH/IF and (e) IF toward OER

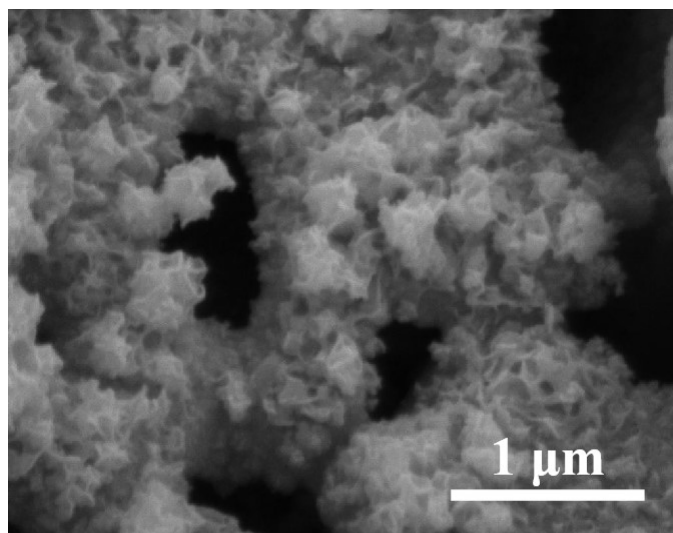
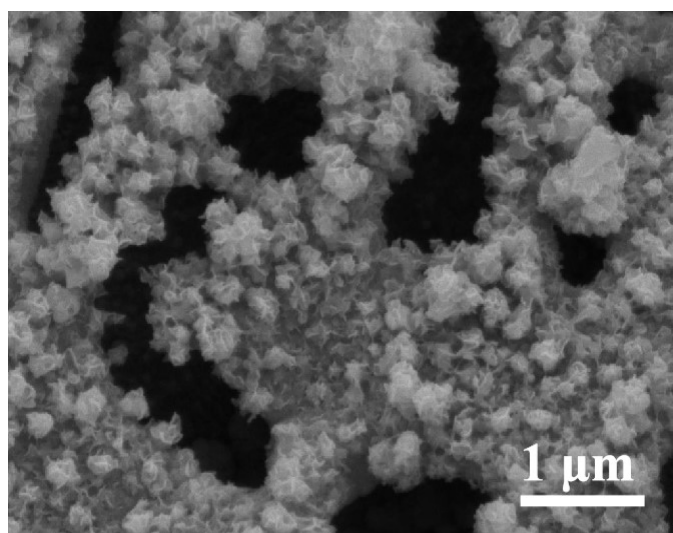


Fig. S6. SEM images of Ru-Fe₃O₄@FeNi-LDH/IF after 25 hours OER stability measurements.

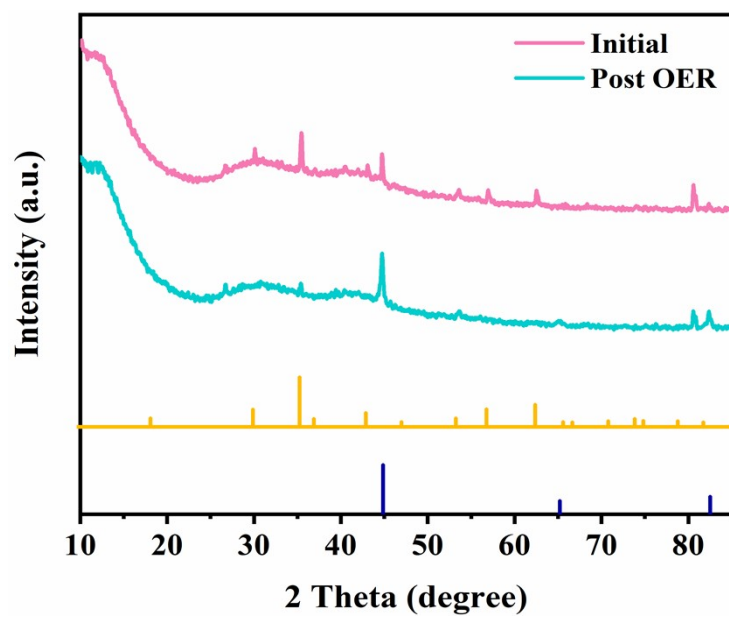


Fig. S7. XRD pattern of Ru-Fe₃O₄@FeNi-LDH/IF before and after OER stability test.

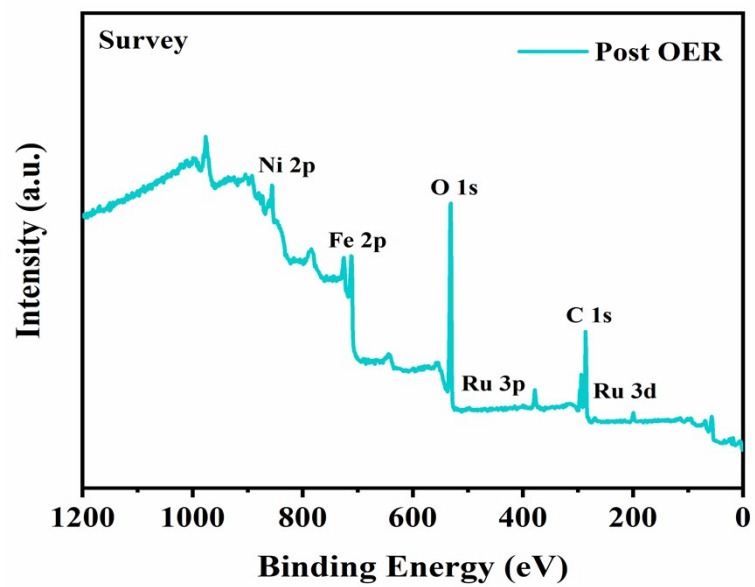


Fig. S8. XPS survey spectrum of Ru-Fe₃O₄@FeNi-LDH/IF after OER stability test.

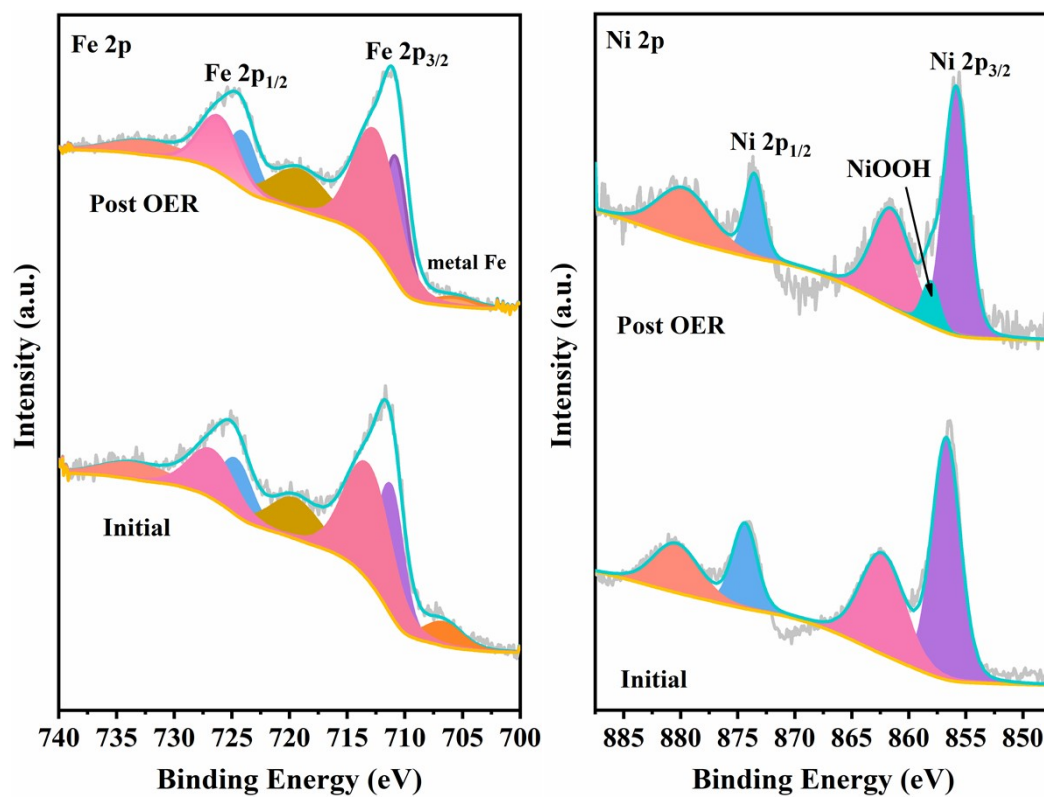


Fig. S9. Comparison of XPS spectra before and after OER stability test.

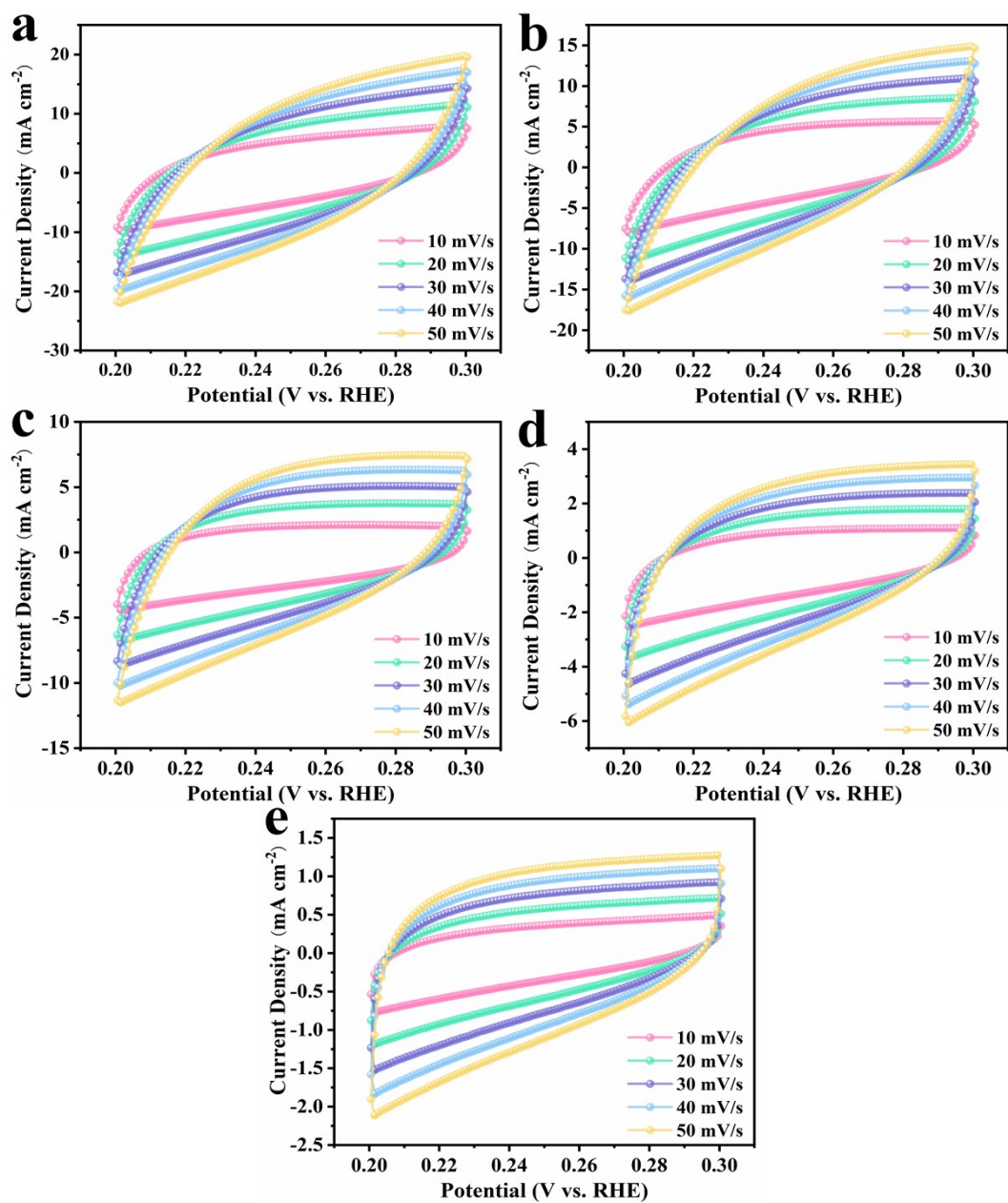


Fig. S10. Cyclic voltammety curves of (a) Ru-Fe₃O₄@FeNi-LDH/IF, (b) Ru-Fe₃O₄/IF, (c) FeNi-LDH/IF, (d) Fe-LDH/IF and (e) IF toward HER.

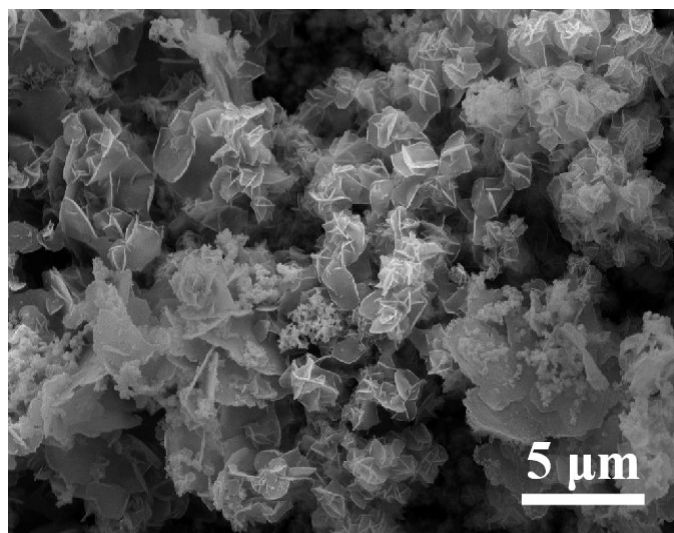
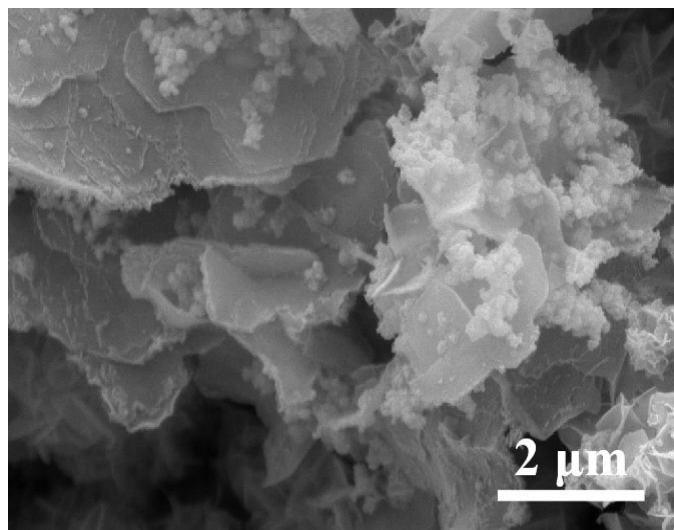


Fig. S11. SEM images of Ru-Fe₃O₄@FeNi-LDH/IF after 25 hours HER stability measurements.

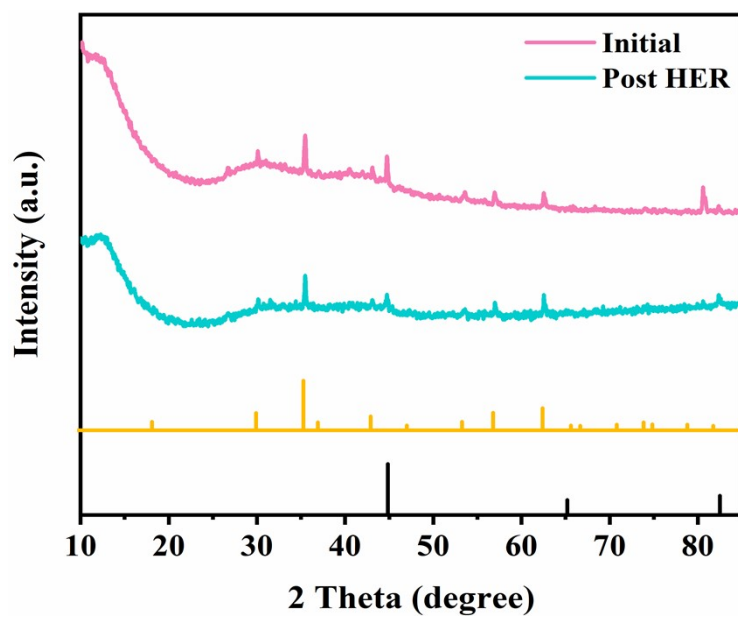


Fig. S12. XRD pattern of Ru-Fe₃O₄@FeNi-LDH/IF before and after HER stability test.

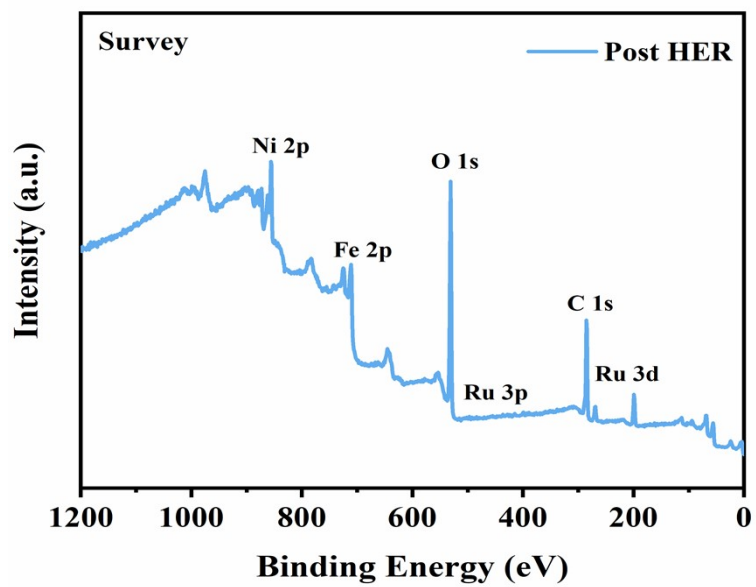


Fig. S13. XPS survey spectrum of Ru-Fe₃O₄@FeNi-LDH/IF after HER stability test.

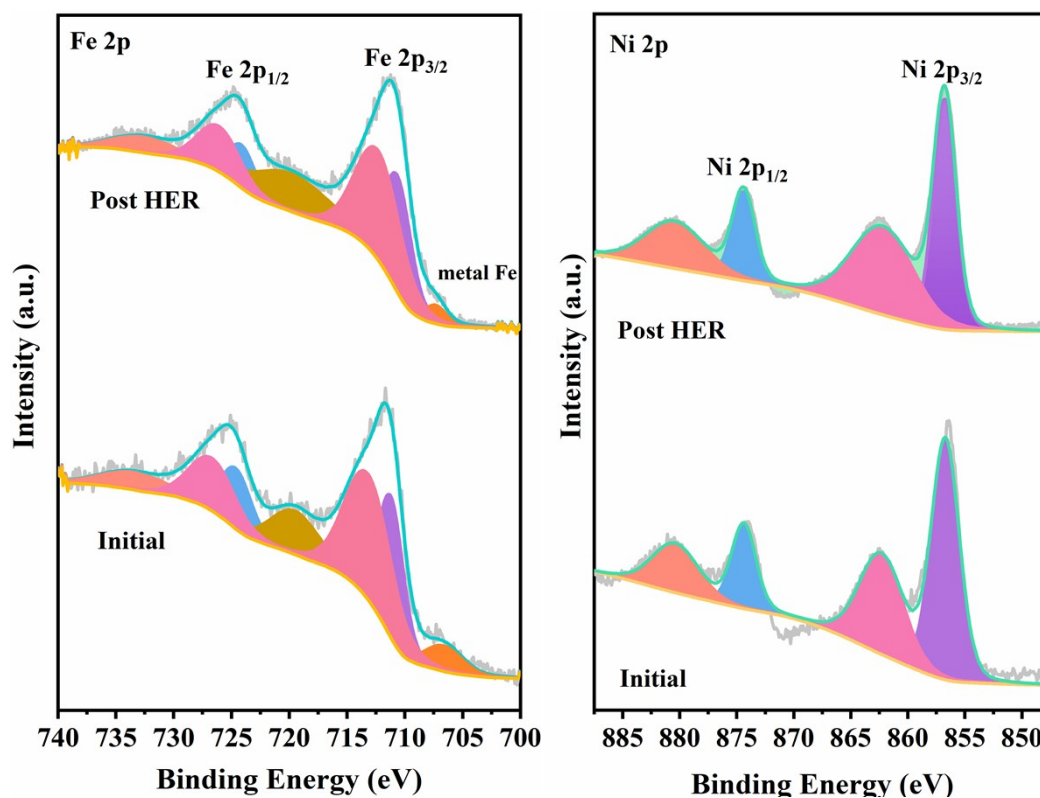


Fig. S14. Comparison of XPS spectra before and after HER stability test.

Fig. S15. Digital photograph of the experimental measurement of the gas production of Ru-Fe₃O₄@FeNi-LDH/IF electrocatalyst at a current density of 100 mA·cm⁻².

Table S1 Comparison of OER activity between Ru-Fe₃O₄@FeNi-LDH/IF and other recently reported non noble metal catalysts.

Catalysts	Overpotential (V vs. RHE)	Reference
Ru-Fe ₃ O ₄ @FeNi-LDH/IF	189 mV@10 mA cm ⁻²	This work
Ru SAs/AC-FeCoNi	205 mV@10 mA cm ⁻²	1
Ni(OH) ₂ @NiSe ₂	225 mV@10 mA cm ⁻²	2
NiFe LDH@NiCoP/NF	220 mV@10 mA cm ⁻²	3
Co(OH) ₂ /CNTs-IrCl- 0.075	230 mV@10 mA cm ⁻²	4
NiFeCo LDH	200 mV@10 mA cm ⁻²	5
CoFe@NiFe-100/NF	220 mV@10 mA cm ⁻²	6
NiFeP@N-CS	216 mV@10 mA cm ⁻²	7
Co-C@NiFe LDH	249mV@10 mA cm ⁻²	8

Table S2 Comparison of HER activity between Ru-Fe₃O₄@FeNi-LDH/IF and other recently reported non noble metal catalysts.

Catalysts	Overpotential (V vs. RHE)	Reference
Ru-Fe ₃ O ₄ @FeNi-LDH/IF	104 mV@10 mA cm ⁻²	This work
NiFe LDH@NiCoP/NF	120mV@10 mA cm ⁻²	3
NiFeP@N-CS	186 mV@10 mA cm ⁻²	7
Fe ₂ O ₃ /NiSe ₂	132 mV@10 mA cm ⁻²	9
Ni-GF/VC	128 mV@10 mA cm ⁻²	10
CoV/ CF-CWs	118 mV@10 mA cm ⁻²	11
Ni-P/Ni/NF	129 mV@10 mA cm ⁻²	12
WP2 NW/NF	130 mV@10 mA cm ⁻²	13
0.05Mn-MoP NSs	198 mV@10 mA cm ⁻²	14

Table S3. Comparison of overall water splitting performance for the Ru-Fe₃O₄@FeNi-LDH/IF catalyst with other reported electrocatalysts in 1 M KOH.

Catalysts	Overpotential (V vs. RHE)	Reference
Ru-Fe ₃ O ₄ @FeNi-LDH/IF	1.52 V@10 mA cm ⁻²	This work
U-Fe-β-Ni(OH) ₂ @/NF-2	1.58 V@10 mA cm ⁻²	15
Fe-FVO-60-act	1.58 V@10 mA cm ⁻²	16
Fe-Ni ₂ P/MoS _x /NF	1.61 V@10 mA cm ⁻²	17
Ni/Ni(OH) ₂	1.68 V@10 mA cm ⁻²	18
Ni@NiFe LDH	1.53 V@10 mA cm ⁻²	19
Ni ₂ P-Fe ₂ P/NF	1.56 V@10 mA cm ⁻²	20
V-CoP@ a-CeO ₂	1.56 V@10 mA cm ⁻²	21
Fe-Ni ₃ P ₄ /NiFeOH-350	1.55 V@10 mA cm ⁻²	22
NiFeP@N-CS	1.63 V@10 mA cm ⁻²	23
CoFe@NiFe/NF	1.59 V@10 mA cm ⁻²	24
Ru-NiFeP/NF	1.47 V@10 mA cm ⁻²	25
WP ₂ NW/NF	1.65 V@10 mA cm ⁻²	26
NiFe-LDH/NiCo ₂ O ₄	1.60 V@10 mA cm ⁻²	27
Ni ₁ Fe ₁₀ -LDH@Ni ₃ S ₂ /NF	1.65 V@10 mA cm ⁻²	28

Supplementary References

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