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## Efficient improvement in non-enzymatic glucose detection induced by the hollow prism-like NiCo<sub>2</sub>S<sub>4</sub> electrocatalyst

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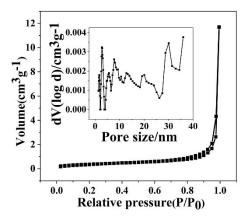


Fig. S1  $N_2$  adsorption-desorption isotherm of  $NiCo_2S_4$  nanoprisms, Inset: the pore size distribution curve of  $NiCo_2S_4$  HNPs.

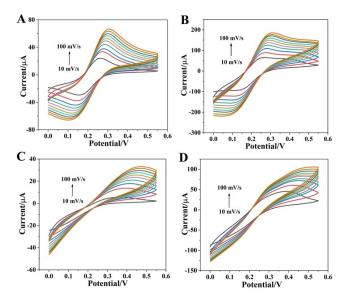


Fig. S2 The scan rate study of NiCo<sub>2</sub>S<sub>4</sub> electrodes annealed at 350 °C for 1, 2, 3 and 4h in 0.1 M KCl solution containing 5 mM K<sub>3</sub>[Fe(CN)<sub>6</sub>].

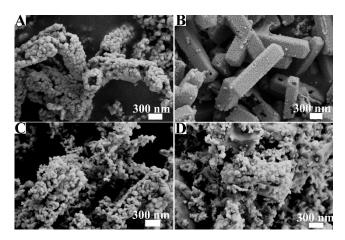
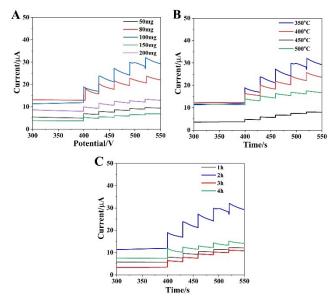
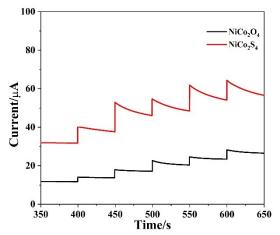


Fig. S3 SEM images of samples for different calcination time: (A) 1h, (B) 2h (C) 3h and (D) 4h.



**Fig. S4** Current-time (i-t) curves of (A) Co/Ni precursor with different amounts of TAA, (B) under different calcination temperature and (C) treated by different holding time at 0.55 V in 0.2 M NaOH with successive addition of 1 mM of glucose.



**Fig. S5** Current-time (i-t) curves of NiCo<sub>2</sub>O<sub>4</sub> and NiCo<sub>2</sub>S<sub>4</sub> modified electrodes at 0.55 V in 0.2 M NaOH with successive addition of 1 mM of glucose.

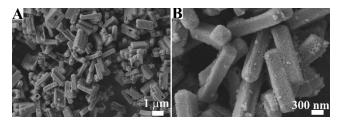
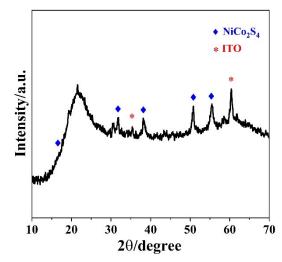


Fig.  $\mathbf{S6}$  SEM images of NiCo<sub>2</sub>S<sub>4</sub> modified electrodes after electrocatalysis.



 $\textbf{Fig. S7} \ XRD \ pattern \ of \ NiCo_2S_4 \ modified \ electrodes \ after \ electrocatalysis.$