

Supporting Information

New Ln-MOFs based on mixed organic ligands: synthesis, structure and efficient luminescence sensing of the Hg²⁺ ion in aqueous solutions

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Table S1 Crystal data and structure refinement for complexes **1-4**.

Complex	1	2	3	4
Formula	C ₁₅₆ H ₁₂₀ Ce ₄ N ₁₀ O ₂₈ S ₆	C ₁₅₆ H ₁₁₆ Pr ₄ N ₁₀ O ₂₆ S ₆	C ₁₅₆ H ₁₂₀ Eu ₄ N ₁₀ O ₂₈ S ₆	C ₁₅₆ H ₁₂₀ Tb ₄ N ₁₀ O ₂₈ S ₆
Fw	3335.45	3302.58	3382.81	3410.65
Cryst system	triclinic	triclinic	triclinic	triclinic
Space group	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a</i> (Å)	11.58490(10)	11.6366(10)	11.49550(10)	11.47990(10)
<i>b</i> (Å)	22.0977(2)	22.1801(3)	22.0977(3)	22.0925(2)
<i>c</i> (Å)	29.3064(3)	29.5908(3)	29.0980(3)	29.0010(3)
α (°)	71.0700(10)	70.7540(10)	108.9420(10)	108.9390(10)
β (°)	89.0760(10)	89.2320(10)	90.1040(10)	90.4640(10)
γ (°)	80.4440(10)	80.4290(10)	98.5400(10)	98.1850(10)
<i>V</i> (Å) ³	6993.10(12)	7103.17(18)	6903.91(14)	6874.58(12)
<i>Z</i>	2	2	2	2
ρ_{calc} (g·cm ⁻³)	1.585	1.544	1.624	1.648
μ/mm^{-1}	11.353	11.779	14.315	11.432
<i>F</i> (000)	3358.0	3324.0	3384.0	3412.0
Reflns collected/ independent	135072/27833	79542/28000	82789/27158	131914/27405
GOF	1.054	1.063	1.049	1.106
R indexes (<i>I</i> > 2 σ (<i>I</i>))	<i>R</i> ₁ = 0.0356 <i>wR</i> ₂ = 0.0964	<i>R</i> ₁ = 0.0630 <i>wR</i> ₂ = 0.1825	<i>R</i> ₁ = 0.0398 <i>wR</i> ₂ = 0.1047	<i>R</i> ₁ = 0.0455 <i>wR</i> ₂ = 0.1246
R indexes (all data)	<i>R</i> ₁ = 0.0383 <i>wR</i> ₂ = 0.0979	<i>R</i> ₁ = 0.0809 <i>wR</i> ₂ = 0.1963	<i>R</i> ₁ = 0.0454 <i>wR</i> ₂ = 0.1077	<i>R</i> ₁ = 0.0528 <i>wR</i> ₂ = 0.1285

^A*R*₁ = $\sum ||\text{Fo}| - |\text{Fc}|| / \sum |\text{Fo}|$. ^B*wR*₂ = $\{\sum [w(\text{Fo}^2 - \text{Fc}^2)^2] / \sum [w(\text{Fo}^2)^2]\}^{1/2}$.

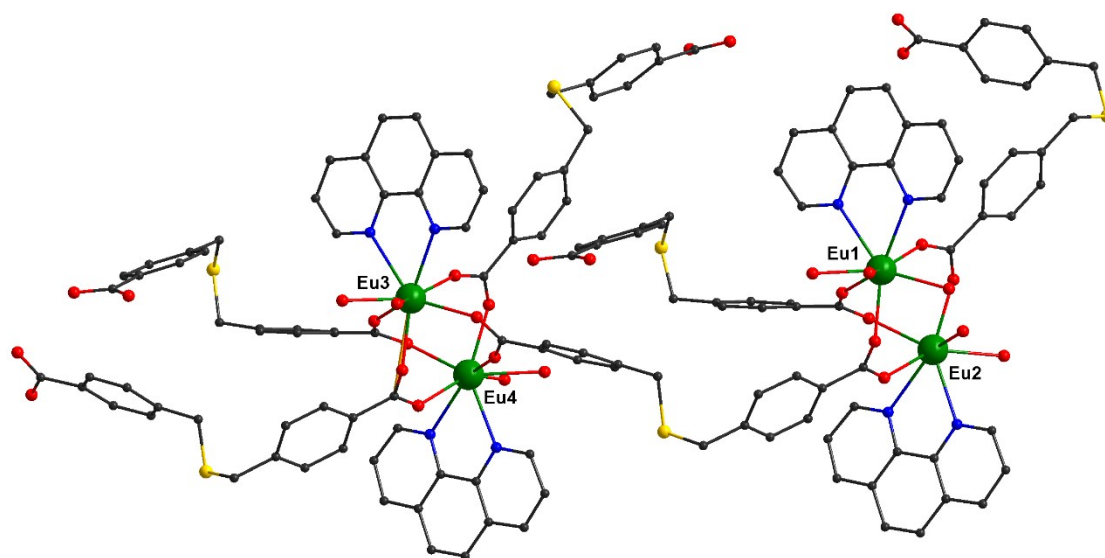


Fig. S1 The coordination environments of Eu^{3+} ions in compound **3**. The hydrogen atoms, free 1,10-phenanthroline and free water molecules were omitted for clarity.

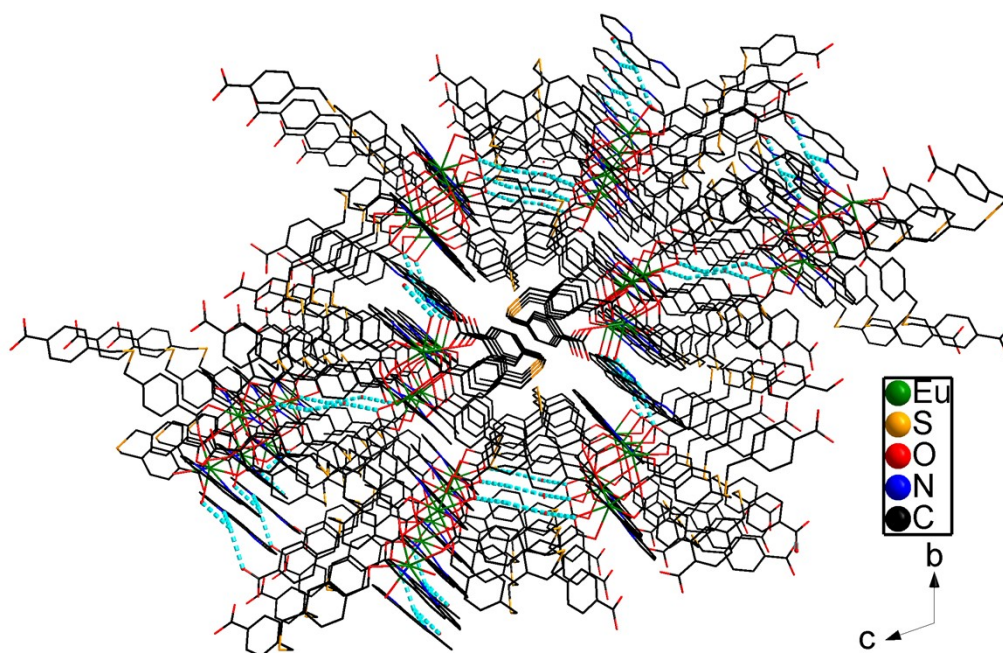
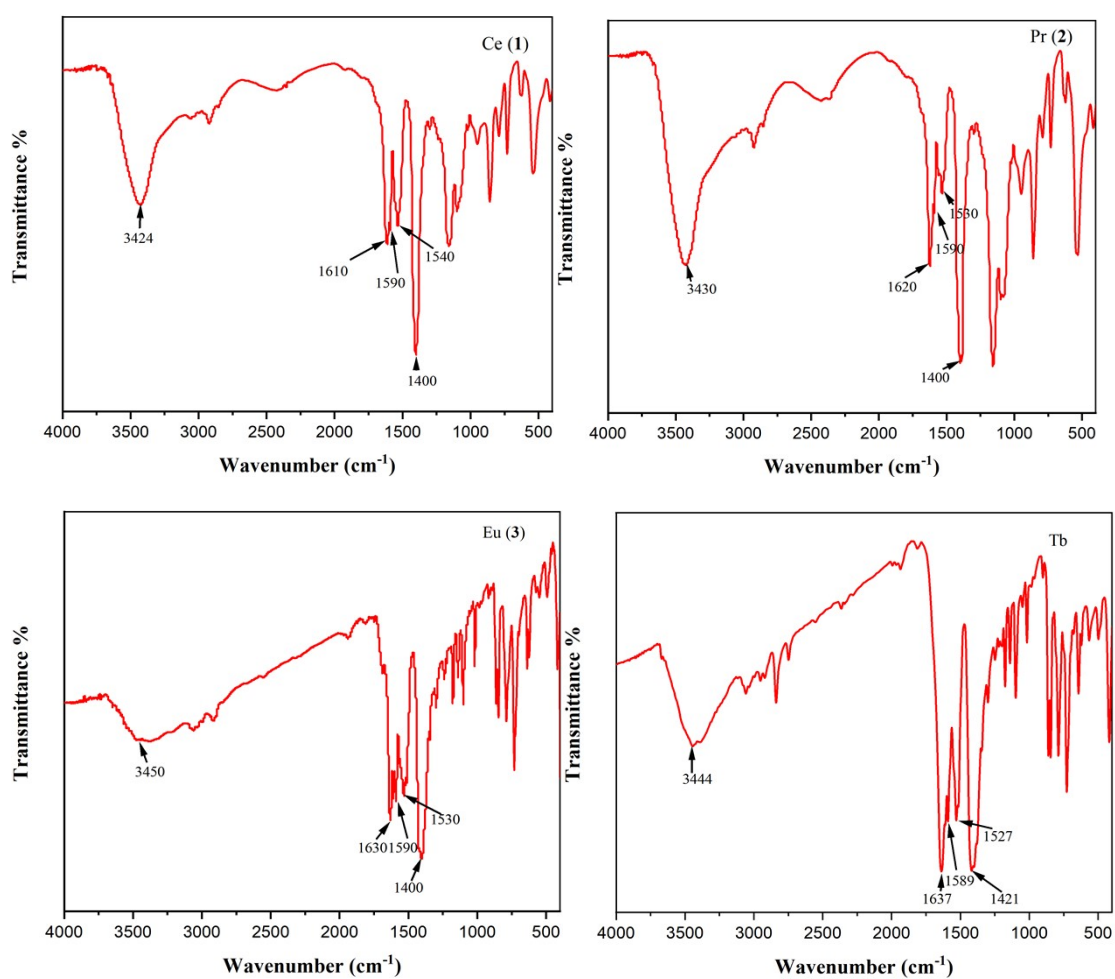


Fig. S2 Hydrogen bonding interactions in **3**.

Table S2 Selected hydrogen bonds for complex **3**.

D-H...A	<i>d</i> (D-H)/nm	<i>d</i> (H...A)/nm	(D...A)/nm	D-H...A/(°)
O(25)-H(25A)...O(19)	0.085	0.202	0.2854(7)	166
O(25)-H(25B)...O(25)	0.085	0.247	0.3198(9)	144
O(26A)-H(26C)...O(7)	0.085	0.202	0.2860(13)	171
O(26A)-H(26D)...O(26A)	0.085	0.246	0.3049(16)	127
O(27)-H(27A)...N(1)	0.085	0.208	0.2870(5)	154
O(27)-H(27B)...O(13)	0.085	0.201	0.2853(4)	172
O(28)-H(28A)...O(4)	0.085	0.193	0.2737(4)	157
O(28)-H(28B)...O(27)	0.097	0.220	0.2734(4)	113

**Fig. S3** Infrared spectra of **1-4**.

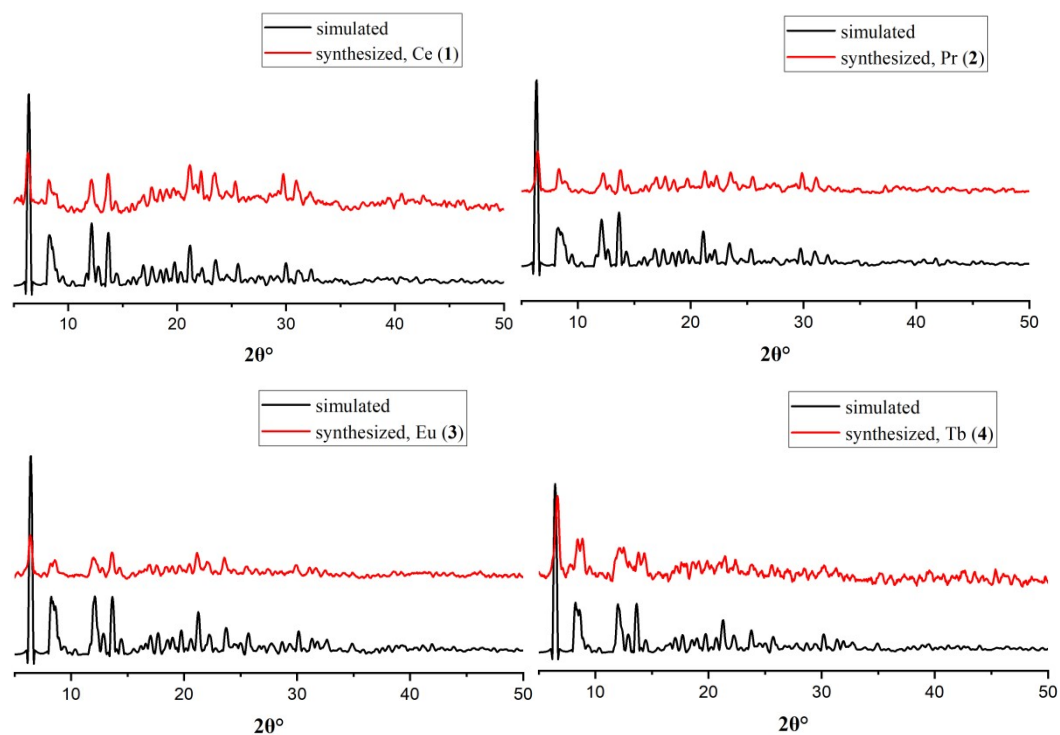


Fig. S4 PXRd patterns of 1-4.

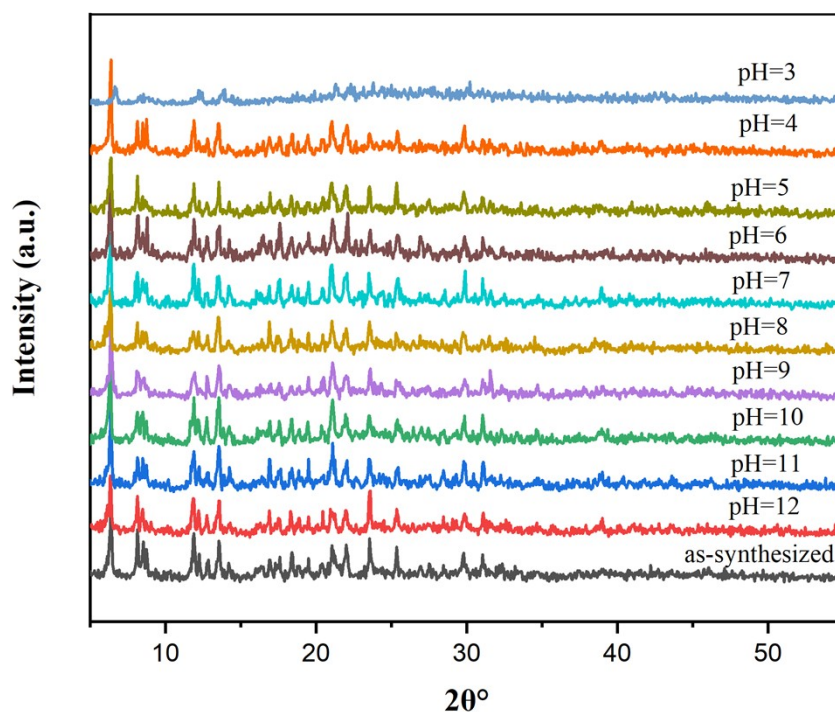


Fig. S5 PXRd patterns of as-synthesized 3 (black line) and 3 soaked in solutions with different pH.

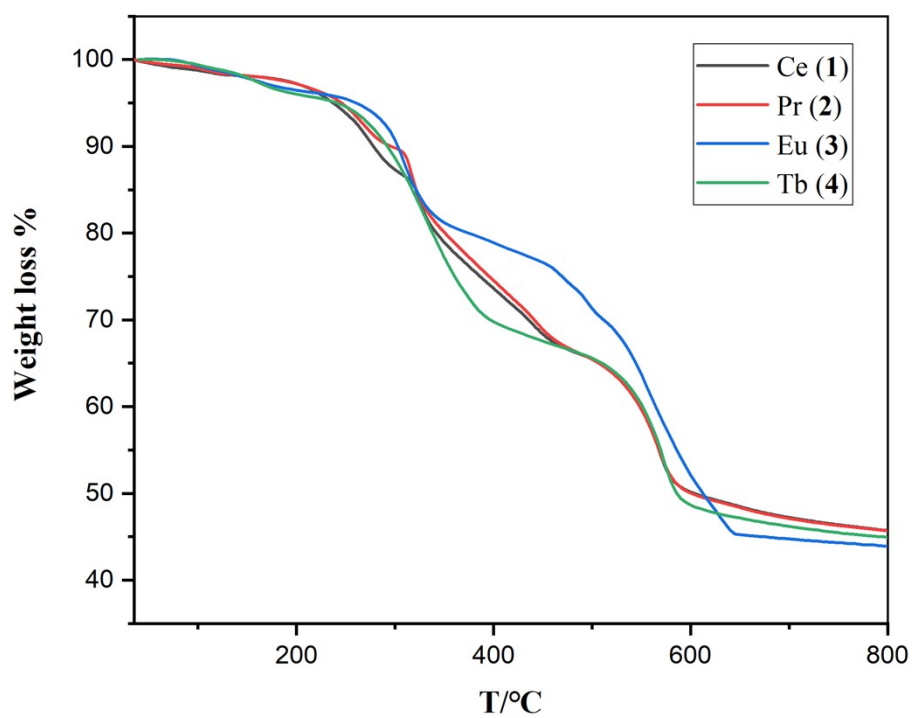


Fig. S6 Thermogravimetric analysis (TGA) curves of coordination polymer **1-4**.

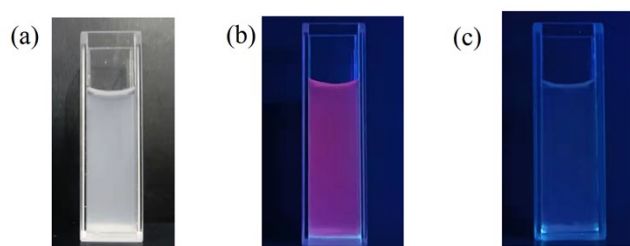


Fig. S7 The photographs of **3** dispersion in water before (a) and after (b) the excitation, (c) the sample soaking in Hg^{2+} solution upon excitation.

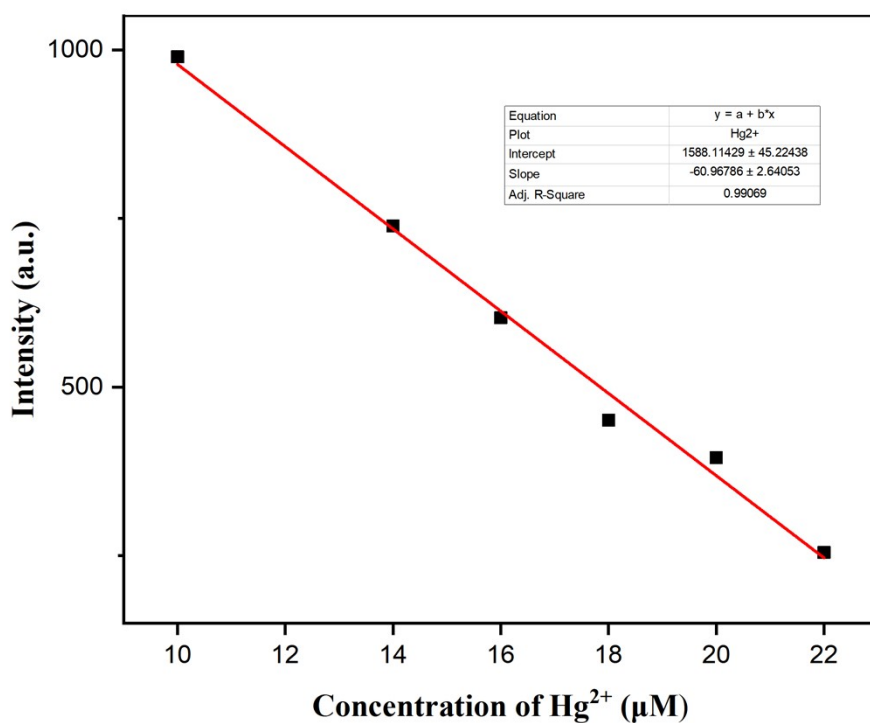


Fig. S8 Linear region of fluorescence intensity of **3** in water upon addition of Hg²⁺ solution.

Table S3 Calculation of Detection Limit.

Blank readings	Fluorescence Intensity
1	2035
2	1988
3	2031
4	2015
5	1998
Standard Deviation (σ)	20.38
Slope (m)	60.96786
Detection limit ($3\sigma/m$)	1.00 μ M

Table S4 Comparison of the sensitivities of **3** with previously reported probes for Hg²⁺ ions.

Probe	Target ion	Detection Limit (μ M)	Reference
[Cd(L)(NTA)] _n		3.05	15a
[Ni(L)(NPTA)·H ₂ O] _n		2.29	15a

S, N-GQDs	9.14	15b
AuNPs@CNF	0.001	15c
SiO ₂ -AuNCs	0.004	15d
[Co(NPDC)(bpee)]·DMF·2H ₂ O	4.1	15e
[PCN-221]	0.01	15f
{[Eu ₄ (tmba) ₆ (phen) ₄]·3(H ₂ O)(phen)} _n	1.00	this work

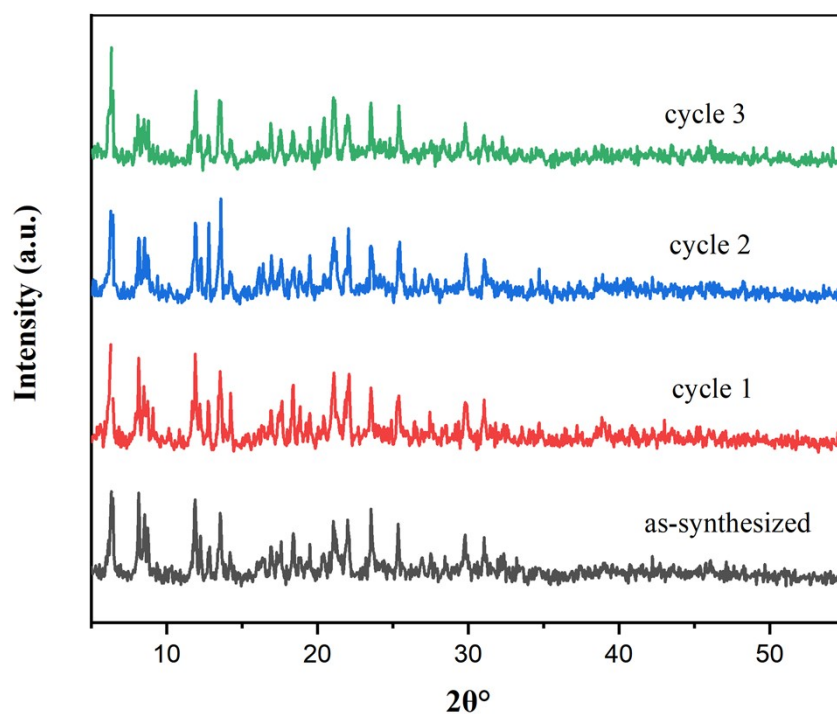


Fig. S9 PXRD patterns of compound **3**, as-synthesized (black line) and after three recycling experiments of soaking into Hg^{2+} solutions.

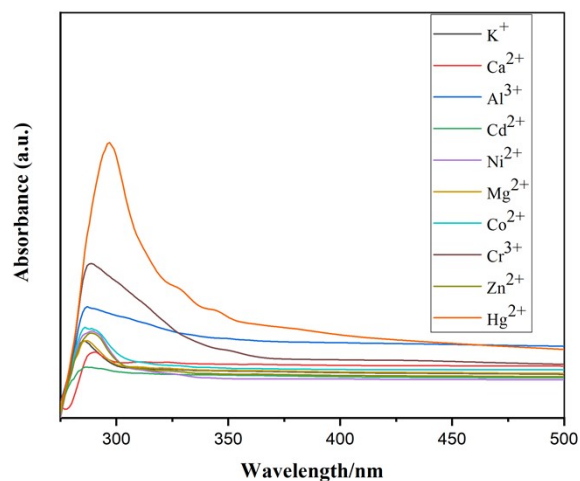


Fig. S10 UV-Vis absorption spectra of different measured ions.

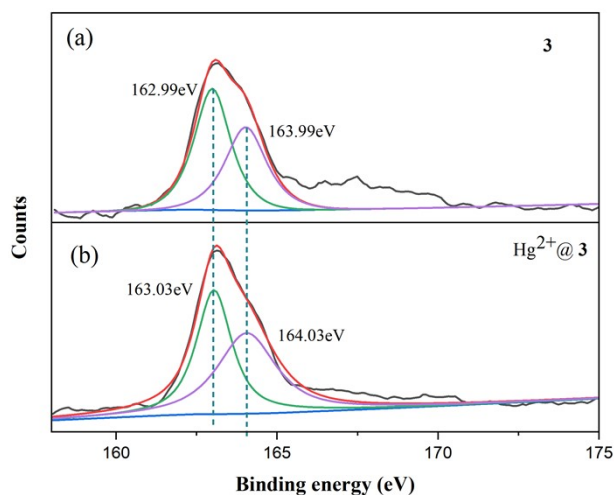


Fig. S11 (a) The XPS spectrum for the S2p region of **3**; (b) The XPS spectrum for the S2p region of **3** after soaking into a Hg^{2+} solution.

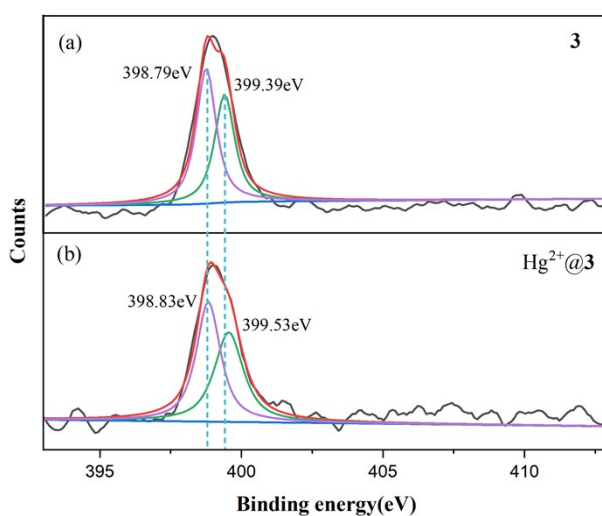


Fig. S12 (a) The XPS spectrum for the N1s region of **3**; (b) The XPS spectrum for the N1s region of **3** after soaking into a Hg^{2+} solution.