## Supplementary Information

## A cyanide-bridged Fe-Co pearl-like single chain magnet containing

## 4-coordinate cobalt(II) ions

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			1
Fe1		Co1	
Fe1–C1	1.908(7)	Co1–N1	2.026(6)
Fe1–C2	1.923(6)	Co1–N3 <sup>2</sup>	2.042(6)
Fe1–C3	1.926(7)	Co1–Cl1	2.233(3)
Fe1–N4	1.963(5)	Co1–Cl2	2.250(3)
Fe1–N6	1.949(5)		
Fe1–N8	1.962(5)		
Co2			
Co2–N2	2.103(6)		
Co2–N2 <sup>1</sup>	2.103(6)		
Co2-01	2.079(6)		
Co2–O2	2.075(7)		
Co2-01 <sup>1</sup>	2.079(6)		
Co2-O2 <sup>1</sup>	2.075(7)		
Fe1		Co1	
C1–Fe1–C2	90.1(3)	N1-Co1-N3 <sup>2</sup>	104.5(3)
C1–Fe1–C3	84.1(3)	N1–Co1–Cl1	107.0(2)
C1-Fe1-N6	91.0(3)	N1–Co1–Cl2	112.5(3)
C1–Fe1–N8	95.9(2)	N3 <sup>2</sup> –Co1–Cl1	109.7(2)
C2–Fe1–C3	88.8(3)	N3 <sup>2</sup> –Co1–Cl2	107.4(3)
C2–Fe1–N4	89.3(2)	Cl1–Co1–Cl2	115.18(14)
C2–Fe1–N8	91.1(2)		
C3–Fe1–N4	91.6(2)		
C3-Fe1-N6	92.2(3)		
N4–Fe1–N6	89.6(2)		
N4–Fe1–N8	88.4(2)		
N6–Fe1–N8	87.9(2)		
Co2			
N2-Co2-O1	87.4(3)		
N2-Co2-O11	91.7(2)		
N2-Co2-O2	91.6(3)		
N2-Co2-O2 <sup>1</sup>	88.4(3)		
01–Co2–O2	87.4(3)		
01–Co2–O2 <sup>1</sup>	92.6(3)		

 Table S1 Selected Bond lengths [Å] and angles [deg] for 1.

			-
Fe1		Co1	
Fe1–C1	1.928(7)	Co1–N1	1.987(6)
Fe1–C2	1.931(8)	Co1–N3 <sup>1</sup>	1.989(7)
Fe1–C3	1.923(8)	Co1–Cl1	2.294(5)
Fe1–N6	2.004(7)	Co1–Cl2	2.195(4)
Fe1–N8	2.002(6)		
Fe1-N10	1.979(7)		
Co2			
Co2–N2	2.107(6)		
Co2–N2 <sup>2</sup>	2.107(6)		
Co2-01	2.100(7)		
Co2–O2	2.061(7)		
Co2-01 <sup>2</sup>	2.100(7)		
Co202 <sup>2</sup>	2.061(7)		
Fe1		Co1	
C1–Fe1–C2	90.9(3)	N1-Co1-N3 <sup>1</sup>	103.2(3)
C1–Fe1–C3	87.1(3)	N1–Co1–Cl1	109.0(3)
C1-Fe1-N6	90.7(3)	N1–Co1–Cl2	111.9(3)
C1-Fe1-N10	90.0(3)	N3 <sup>1</sup> –Co1–Cl1	105.1(2)
C2–Fe1–C3	89.8(3)	N3 <sup>1</sup> –Co1–Cl2	115.0(2)
C2-Fe1-N6	90.2(3)	Cl1–Co1–Cl2	112.0(2)
C2–Fe1–N8	88.7(3)		
C3–Fe1–N8	92.5(3)		
C3-Fe1-N10	90.5(3)		
N6–Fe1–N8	89.7(3)		
N8-Fe1-N10	90.3(3)		
N6-Fe1-N10	89.6(3)		
Co2			
N2-Co2-O1	88.4(3)		
N2-Co2-O1 <sup>2</sup>	91.6(3)		
N2-Co2-O2	88.4(3)		
N2-Co2-O2 <sup>2</sup>	91.6(3)		
01–Co2–O2	89.5(3)		
01–Co2–O2 <sup>2</sup>	90.5(3)		

 Table S2 Selected Bond lengths [Å] and angles [deg] for 2.

Compound	Metal ion	Symmetry	CShM
1	Co1	T <sub>d</sub>	0.493
	Co2	$O_h$	0.068
	Fe1	$O_h$	0.140
2	Co1	T <sub>d</sub>	0.550
	Co2	$O_h$	0.050
	Fe1	$O_h$	0.047

**Table S3.** Continuous Shape Measures (CShM) calculated by *SHAPE*<sup>1</sup> Software for **1** and **2**.

1. SHAPE, version 2.0: continuous shape measures calculation; Electronic Structure Group, Universiat de Barcelona: Barcelona, Spain, 2010.



Fig. S1 IR spectrum of compound 1.



Fig. S2 Experimental and simulated PXRD patterns of 1.



Fig. S3 IR spectrum of compound 2.



Fig. S4 Thermo-gravimetric curve of 1.



Fig. S5 Thermo-gravimetric curve of 2.



**Fig. S6** Crystal structure of **1** drawn with ellipsoids at the 50% probability level. The hydrogen atoms have been omitted for the sake of clarity.



**Fig. S7** Crystal structure of **2** drawn with ellipsoids at the 50% probability level. The hydrogen atoms have been omitted for the sake of clarity.



Fig. S8 The packing diagram of 1 view along the crystallographic b axis.



Fig. S9 The packing diagram of 2 view along the crystallographic b axis.



Fig. S10 Zero-field cooled and field-cooled magnetization data of 1 at 2–20 K.



Fig. S11 Zero-field cooled and field-cooled magnetization data of 2 at 2–20 K.



Fig. S12 Temperature dependence of the in-phase ( $\chi'$ ) and out-of-phase ( $\chi''$ ) ac magnetic susceptibilities for 1 under 1 kOe dc field.



**Fig. S13** Temperature (left) and frequency (right) dependence of the in-phase  $(\chi')$  and out-of-phase  $(\chi'')$  ac magnetic susceptibilities for **2** under zero dc field.