## **Supplementary Materials**

## Comparative studies on the VOCs sorption performances over hierarchical and conventional ZSM-5 zeolites

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			CZ	HZ
Hexane –	Peak2	2 K·min <sup>-1</sup>	333	_a
		3 K·min <sup>-1</sup>	341	324
		4 K·min <sup>-1</sup>	348	333
		5 K·min <sup>-1</sup>	354	338
		E(kJ·mol <sup>-1</sup> )	37.0	27.1
		<b>R</b> <sup>2</sup>	0.9939	0.9710
	Peak3	2 K·min <sup>-1</sup>	365	_a
		3 K·min <sup>-1</sup>	373	_a
		4 K·min <sup>-1</sup>	381	_a
		5 K·min <sup>-1</sup>	387	_a
		E(kJ·mol <sup>-1</sup> )	42.1	-
		$\mathbb{R}^2$	0.9899	-
Toluene	Peak2	0.5 K · min <sup>-1</sup>	_a	_a
		1 K·min <sup>-1</sup>	333	_a
		2 K·min <sup>-1</sup>	347	_a
		3 K·min <sup>-1</sup>	358	_a
		E(kJ·mol <sup>-1</sup> )	38.1	-
		$\mathbb{R}^2$	0.9899	-

 Table S1 Desorption peak temperature (Tp) and binding energy (E) of hexane and toluene over CZ and HZ.

<sup>a</sup>The desorption peak temperature cannot be directly obtained from the TPD profiles due to poor resolution.



Fig.S1 Schematic image of HZ procedure.



Fig.S2 SEM images of CZ (A) and HZ (B).



**Fig.S3** Temperature-programmed desorption profiles of hexane over CZ (A) and HZ (B) and toluene over CZ (C) and HZ (D) at different gas flow rates (heating rate=5 K·min<sup>-1</sup>).



Fig. S4 Toluene evolution curves (The samples were first saturated with toluene and at time t = 0, toluene was cut off. And the solid lines were fitted with an exponential).



Fig. S5 Desorption curves of hexane over CZ (A) and HZ (B) and toluene over CZ (C) and HZ (D) in cycle adsorption-desorption tests.