

**Rational synthesis of CoFeP@nickel-manganese sulfide core-shell nanoarrays  
for hybrid supercapacitor**

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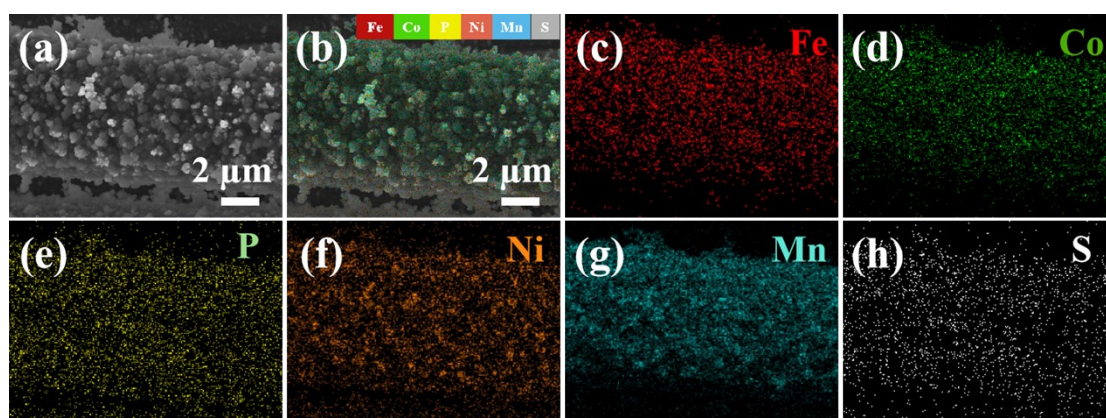


Fig. S1 (a,b) SEM images for the corresponding EDX mapping image of CoFeP@NiMnS/CC, (c) Fe elemental mapping, (d) Co elemental mapping, (e) P elemental mapping, (f) Ni elemental mapping, (g) Mn elemental mapping, and (h) S elemental mapping.

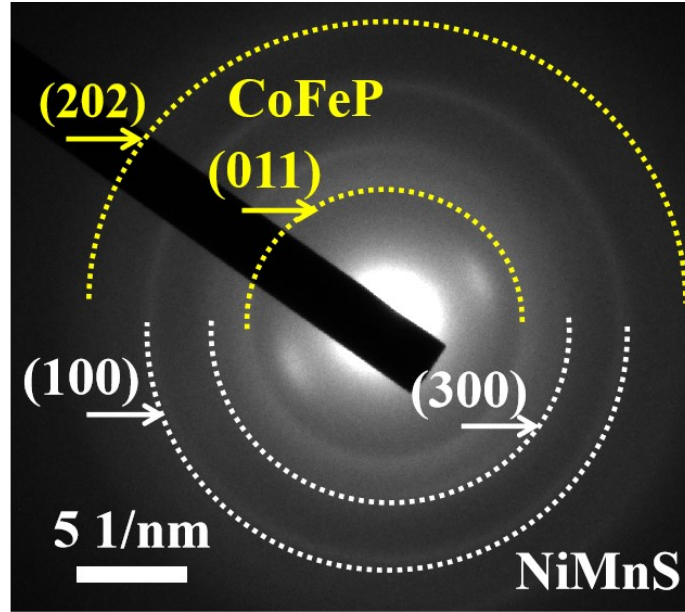


Fig. S2 The selected area electron diffraction (SEAD) pattern of CoFeP@NiMnS.

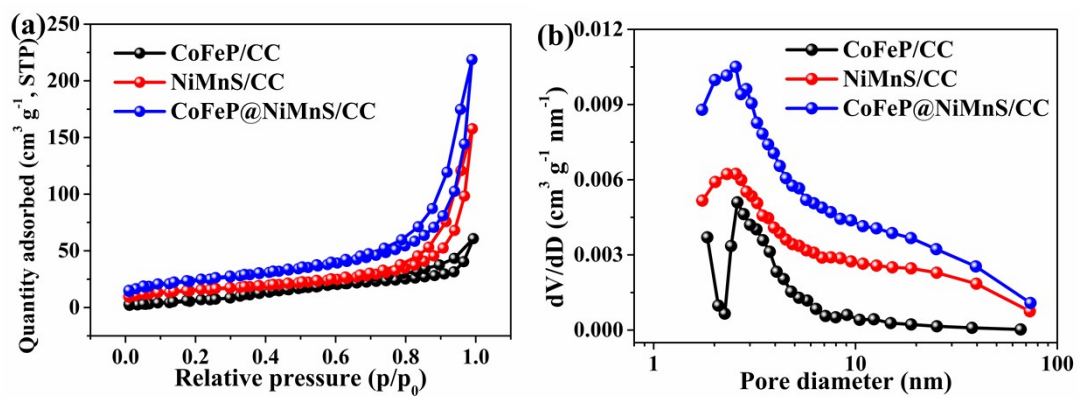


Fig. S3 (a) N<sub>2</sub> adsorption-desorption isotherms, and (b) pore size distributions of the CoFeP/CC, NiMnS/CC, and CoFeP@NiMnS/CC.

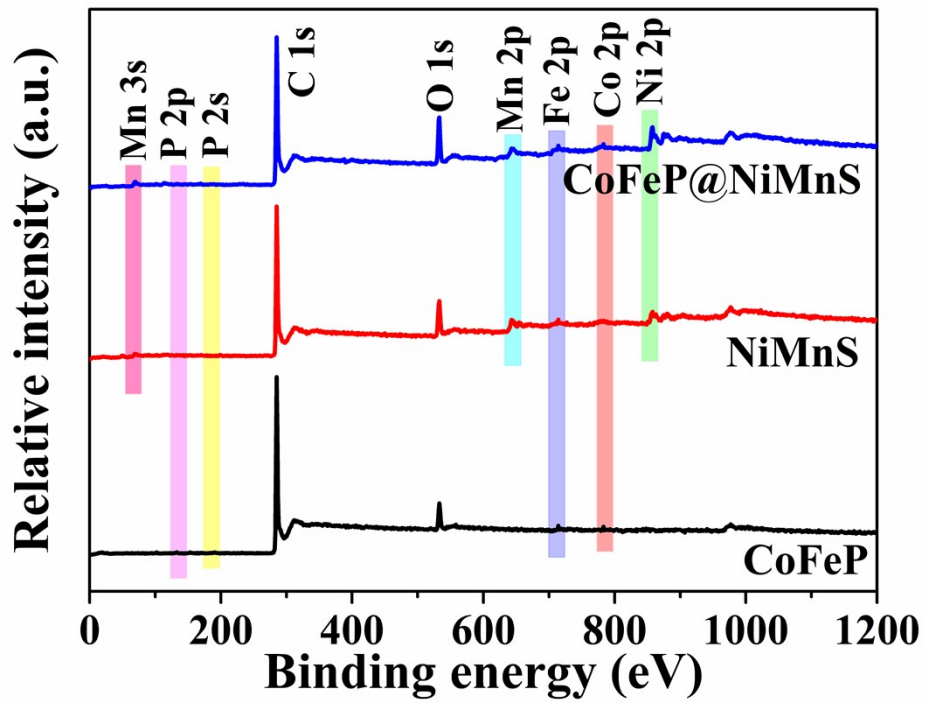


Fig. S4 XPS surveys of CoFeP/CC, NiMnS/CC, and CoFeP@NiMnS/CC.

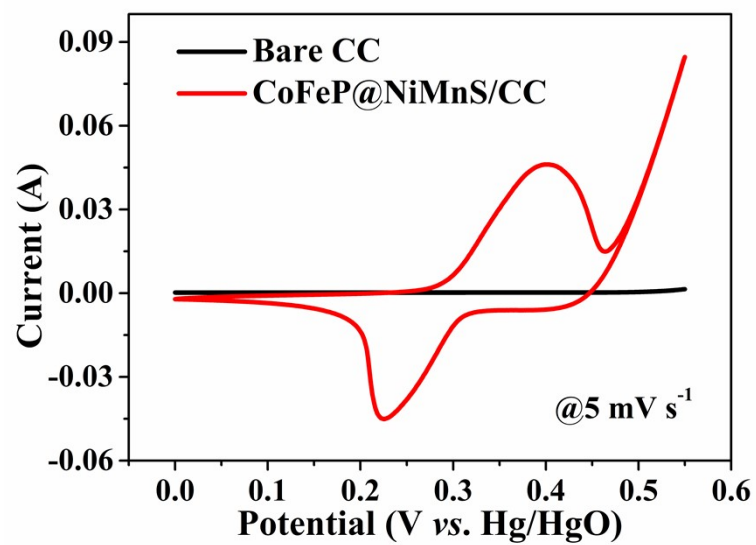


Fig. S5 Comparison of CV curves of bare carbon cloth and CoFeP@NiMnS/CC electrodes at a scan rate of 5 mV s<sup>-1</sup>.

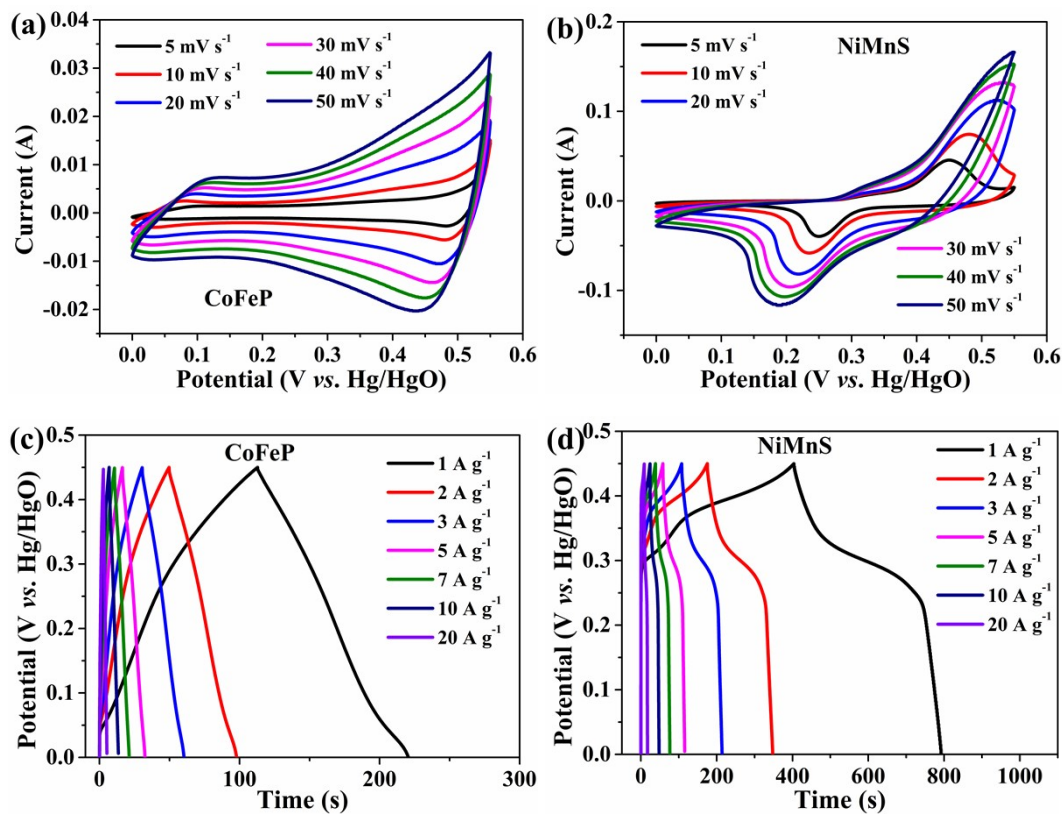


Fig. S6 Electrochemical performance tested in a three-electrode system in 2.0 M KOH electrolyte: CV curves of (a) CoFeP/CC electrode and (b) NiMnS/CC electrode, and GCD curves of (c) CoFeP/CC electrode and (d) NiMnS/CC electrode.

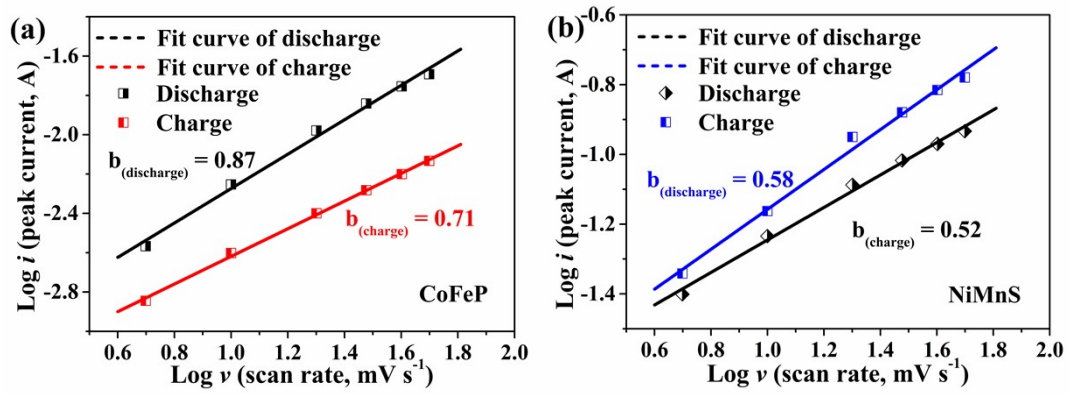


Fig. S7 The linear relationship between the peak current and the scan rate during charge and discharge for (a) the CoFeP/CC and (b) NiMnS/CC electrodes.



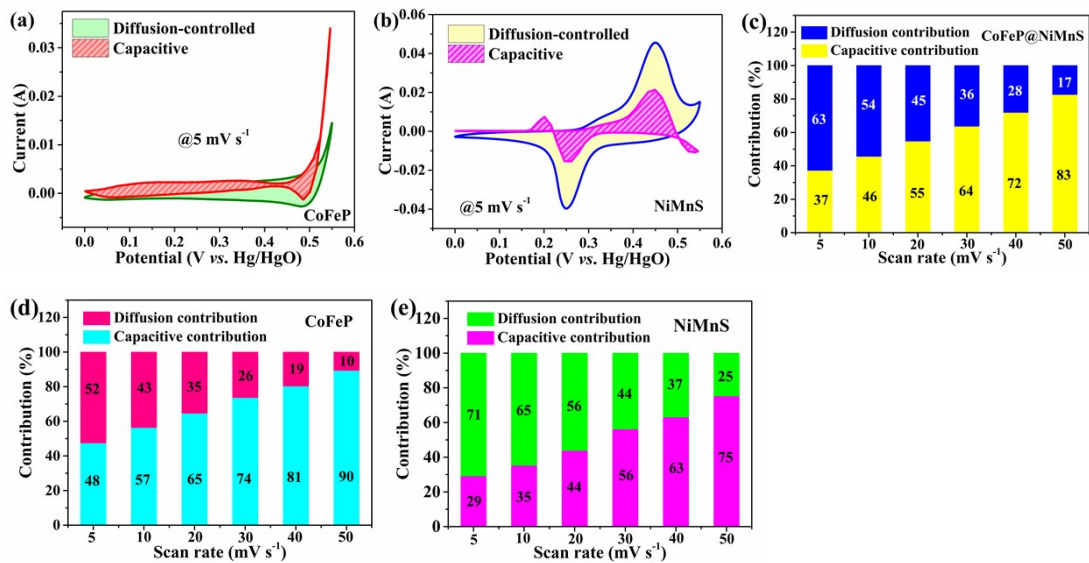


Fig. S8 The diffusion-controlled and capacitive-controlled CV response at 5 mV s<sup>-1</sup> of (a) the CoFeP/CC and (b) NiMnS/CC electrode, and the percentages of capacitive contribution at different scan rates of 5-50 mV s<sup>-1</sup> for (c) CoFeP@NiMnS/CC, (d) CoFeP/CC, and (e) NiMnS/CC electrodes.

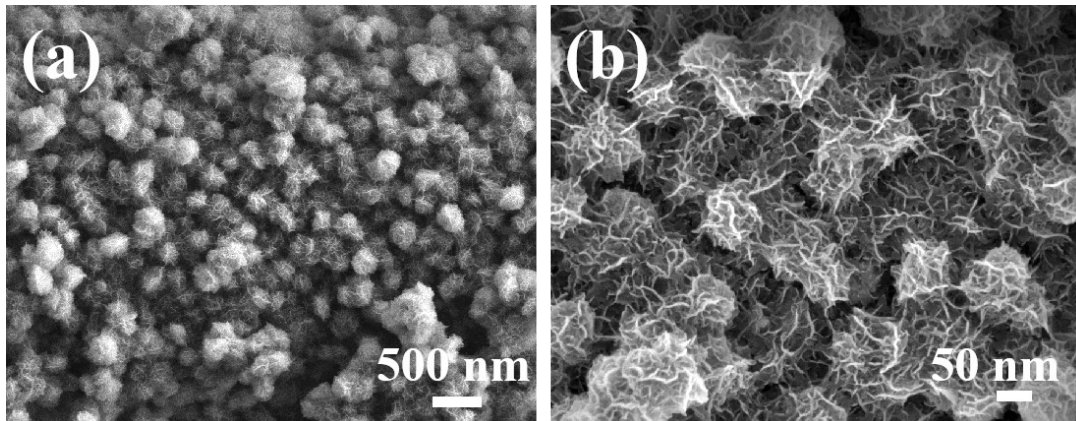


Fig. S9 SEM images of CoFeP@NiMnS/CC electrode after 5000 cycles at 5 A g<sup>-1</sup> at (a) low and (b) high magnifications.

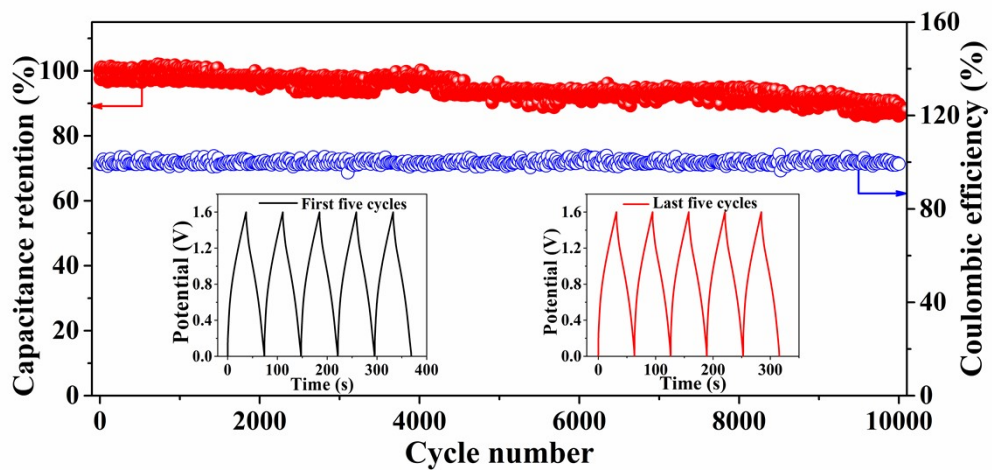


Fig. S10 The cyclic performance and coulombic efficiency of the assembled CoFeP@NiMnS/CC//LPC-Cu HSC device over 10000 cycles at  $5 \text{ A g}^{-1}$  (inset shows the first and last five GCD cycles).

**Table S1** Specific surface areas and pore parameters of CoFeP/CC, NiMnS/CC, and

CoFeP@NiMnS/CC.

Samples	$S_{\text{BET}}$ ( $\text{m}^2 \text{g}^{-1}$ ) <sup>a</sup>	$S_{\text{micro}}$ ( $\text{m}^2 \text{g}^{-1}$ ) <sup>b</sup>	$V_{\text{total}}$ ( $\text{cm}^3 \text{g}^{-1}$ ) <sup>c</sup>	$V_{\text{micro}}$ ( $\text{cm}^3 \text{g}^{-1}$ )
CoFeP/CC	38.6	1.02	0.102	0.008
NiMnS/CC	61.3	4.73	0.137	0.031
CoFeP@NiMnS/CC	89.7	6.36	0.169	0.082

<sup>a</sup> Specific surface area ( $S_{\text{BET}}$ ) was calculated via Brunauer-Emmett-Teller (BET) method.

<sup>b</sup> Micropore surface area ( $S_{\text{mic}}$ ) was calculated from t-plot method.

<sup>c</sup> Total pore volume ( $V_{\text{total}}$ ) was obtained from the adsorbed amount at a relative pressure of 0.99.

**Table S2** Comparison of the electrochemical performances of the as-fabricated CoFeP@NiMnS/CC electrode with previously reported core-shell electrode materials in literature.

Electrode	Synthesis method	Specific capacity (mAh g <sup>-1</sup> )	Rate performance	Maximum energy density (W kg <sup>-1</sup> )	Cycling stability	Ref.
NiCoP@CoS	Hydrothermal method and phosphorization	1796 F g <sup>-1</sup> (2 A g <sup>-1</sup> )	68.3% (20 A g <sup>-1</sup> )	35.8 Wh kg <sup>-1</sup> at 748.9 W kg <sup>-1</sup>	86.1% after 10000 cycles	1
NiCoP@MoSe <sub>2</sub>	Hydrothermal, phosphorization and CVD	2245.4 F g <sup>-1</sup> (1 mA cm <sup>-2</sup> )	91.6% (10 A g <sup>-1</sup> )	55.1 Wh kg <sup>-1</sup> at 799.8 W kg <sup>-1</sup>	95.8% after 8000 cycles	2
CoP@NiCoP	Hydrothermal method and phosphorization	265.5 (2 A g <sup>-1</sup> )	61.68% (25 A g <sup>-1</sup> )	37.16 Wh kg <sup>-1</sup> at 875 W kg <sup>-1</sup>	50% after 1000 cycles	3
NiCoP/NiCo-OH	Hydrothermal method and phosphorization	1100 (1 A g <sup>-1</sup> )	60% (10 A g <sup>-1</sup> )	34 Wh kg <sup>-1</sup> at 775 W kg <sup>-1</sup>	92% after 1000 cycles	4
CuCo-P@Ni(OH) <sub>2</sub>	Solvothermal, phosphorization and electrodeposition	230.6 (2 mA cm <sup>-2</sup> )	86.3% (20 mA cm <sup>-2</sup> )	40 Wh kg <sup>-1</sup> at 319.6 W kg <sup>-1</sup>	73% after 6000 cycles	5
ZnCo <sub>2</sub> O <sub>4</sub> @Ni-Co-S	Hydrothermal method and electrodeposition	1396.9 C g <sup>-1</sup> (1 A g <sup>-1</sup> )	52.8% (10 A g <sup>-1</sup> )	53.1 Wh kg <sup>-1</sup> at 3375 W kg <sup>-1</sup>	85.5% after 10000 cycles	6
Co-CH@NiCoMn-CH	Solvothermal method	3224 F g <sup>-1</sup> (1 A g <sup>-1</sup> )	83.97% (5 A g <sup>-1</sup> )	20.31 Wh kg <sup>-1</sup> at 748.46 W kg <sup>-1</sup>	90.4% after 6000 cycles	7
Co-MoS <sub>2</sub> @Cu <sub>2</sub> MoS <sub>4</sub>	Hydrothermal method	220 (1 A g <sup>-1</sup> )	64% (10 A g <sup>-1</sup> )	41.6 Wh kg <sup>-1</sup> at 6240 W kg <sup>-1</sup>	92.8% after 5000 cycles	8
<b>CoFeP@NiMnS/CC</b>	<b>Hydrothermal, phosphorization and electrodeposition</b>	<b>260.7 (1 A g<sup>-1</sup>)</b>	<b>65.9% (20 A g<sup>-1</sup>)</b>	<b>60.1 Wh kg<sup>-1</sup> at 371.8 W kg<sup>-1</sup></b>	<b>85.7% after 10000 cycles</b>	<b>This work</b>

## References

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